

Supporting Information

Formamide Iodide: A new cation additive for inhibiting δ -phase formation of formamidinium lead iodide perovskite

AUTHOR NAMES

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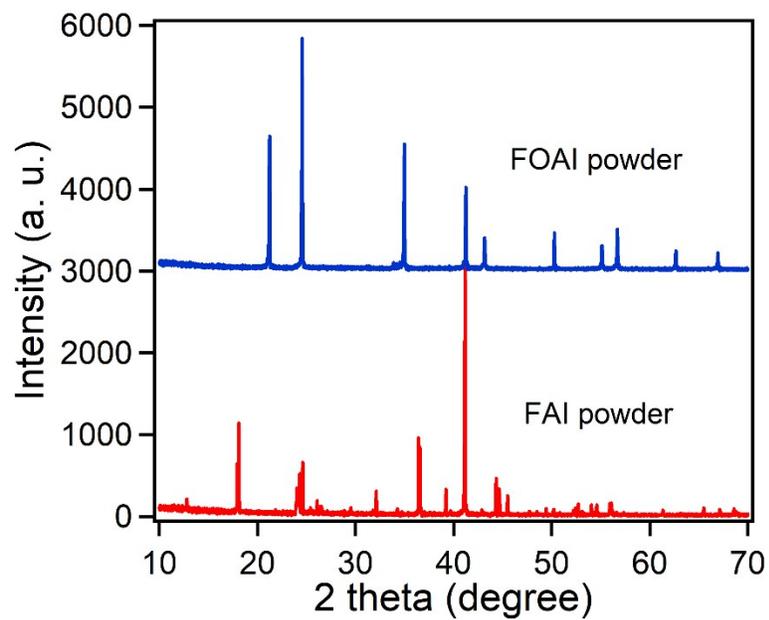


Figure S1 Powder XRD patterns of formamide iodide (FoAI) and formamidine iodide (FAI).

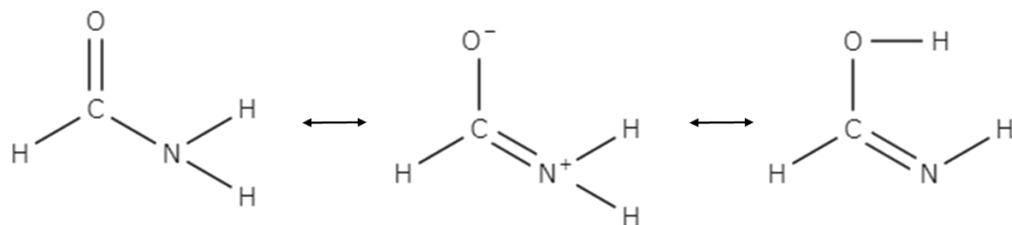


Figure S2 Resonance structures of formamide.

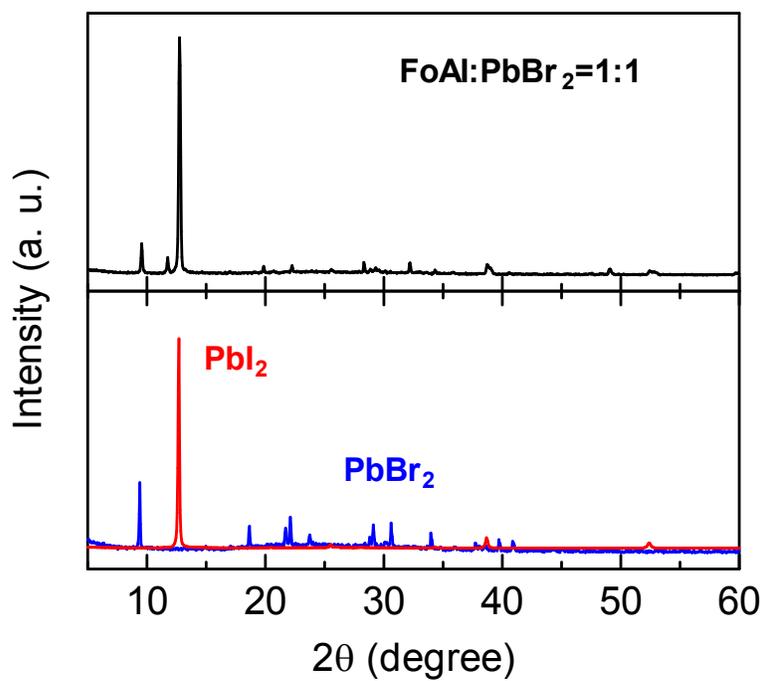


Figure S3 XRD patterns of PbI_2 , PbBr_2 films and a film prepared from a precursor solution including PbBr_2 and formamide iodide (FoAI) with the molar ratio of 1:1.

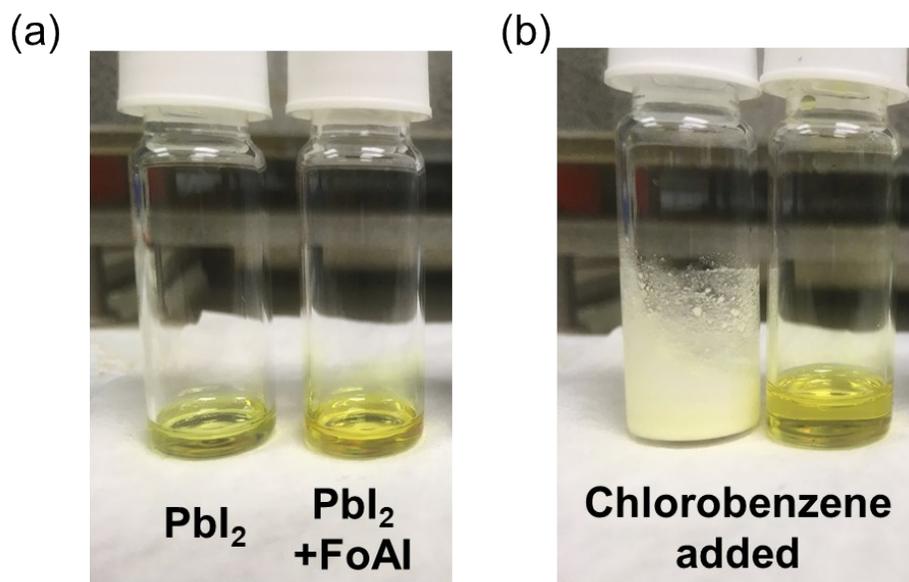


Figure S4 Photographs of PbI_2 and $\text{PbI}_2 + \text{FoAI}$ solution (a) before and (b) after adding chlorobenzene.

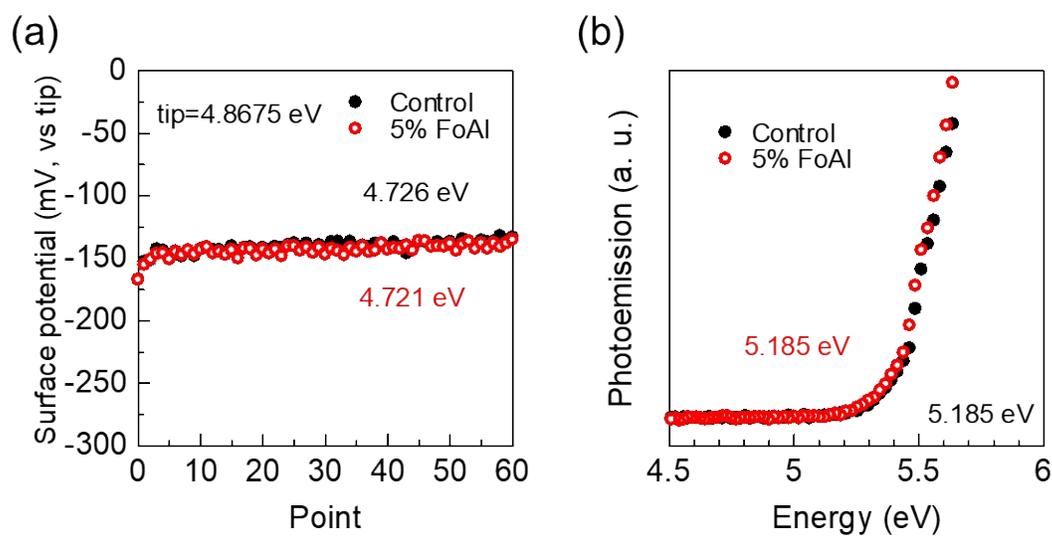


Figure S5 (a) Surface potential of FAPbI_3 films with and without FoAI additive. Values in the graph are work function of both films. (b) Photoemission spectra of FAPbI_3 films with and without FoAI additive. Values in the graph are ionization potential of both films.

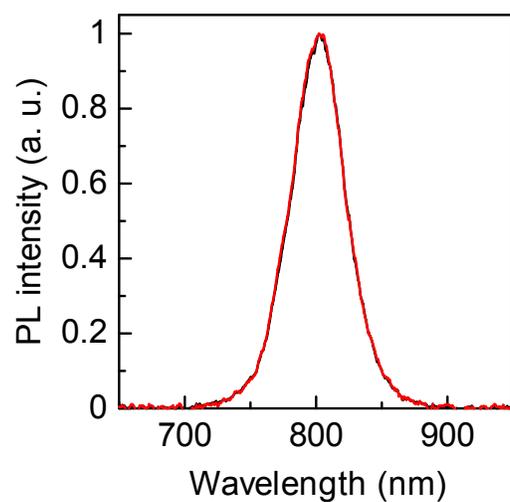


Figure S6 PL spectra of pure FAPbI₃ and FoAI added FAPbI₃ films. The black and red lines represent PL spectra of pure and FoAI added FAPbI₃ films, respectively.

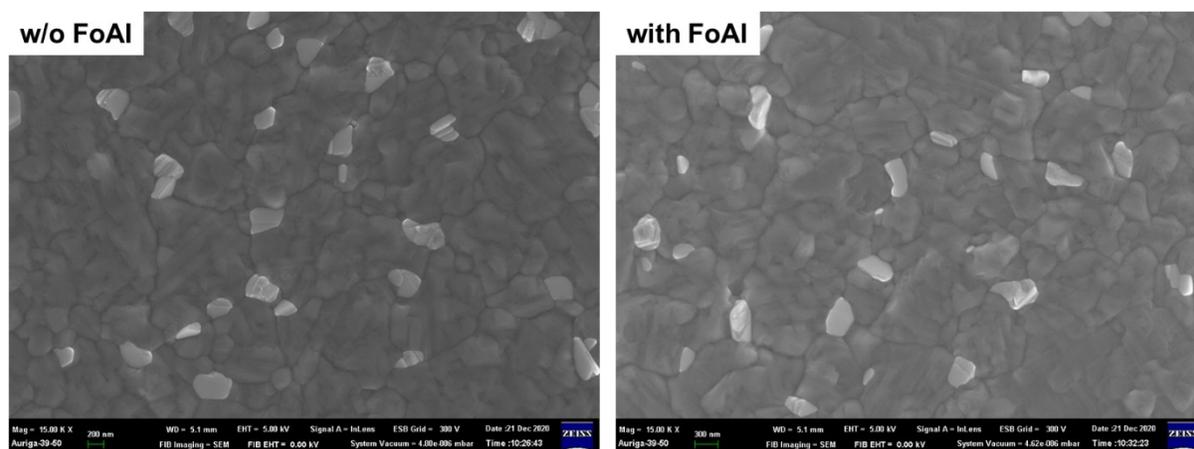


Figure S7 SEM images of pure FAPbI₃ and FoAI added FAPbI₃ films.

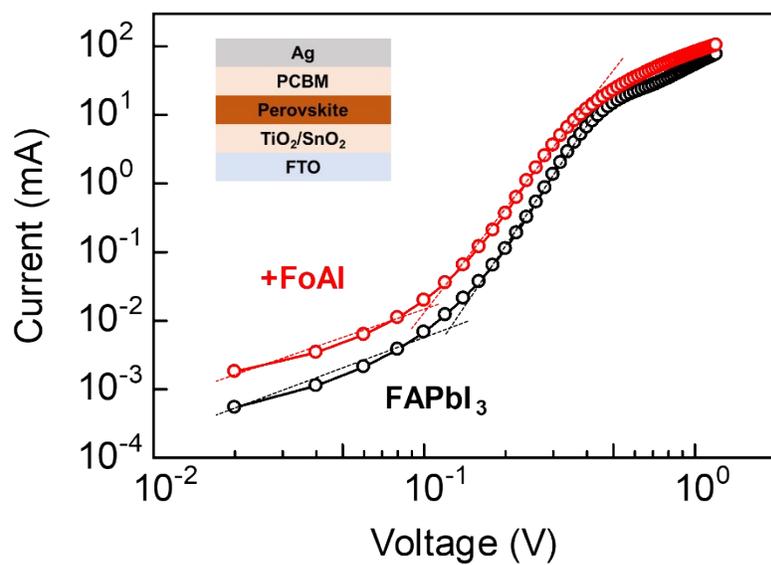


Figure S8 Dark $I-V$ curves of electron-only devices employing pure and FoAl added FAPbI₃ films.

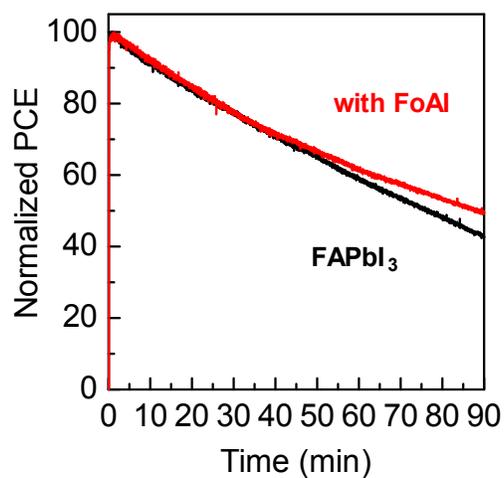


Figure S9 Stability of FAPbI₃ perovskite solar cells kept at maximum power point under 1 sun irradiation without encapsulation.

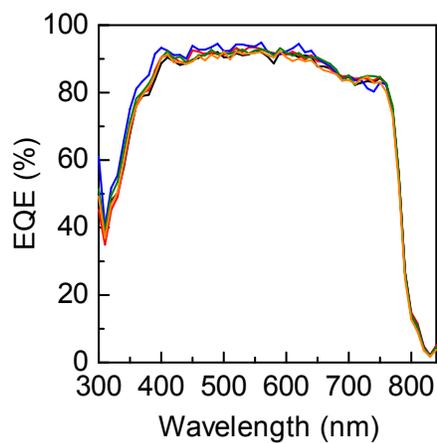


Figure S10 EQE spectra of FoAI added triple-cation perovskite solar cells. The black, red, blue, green, and orange lines represent 0, 0.625, 1.25, 2.5 and 5 mol% FoAI added samples, respectively.

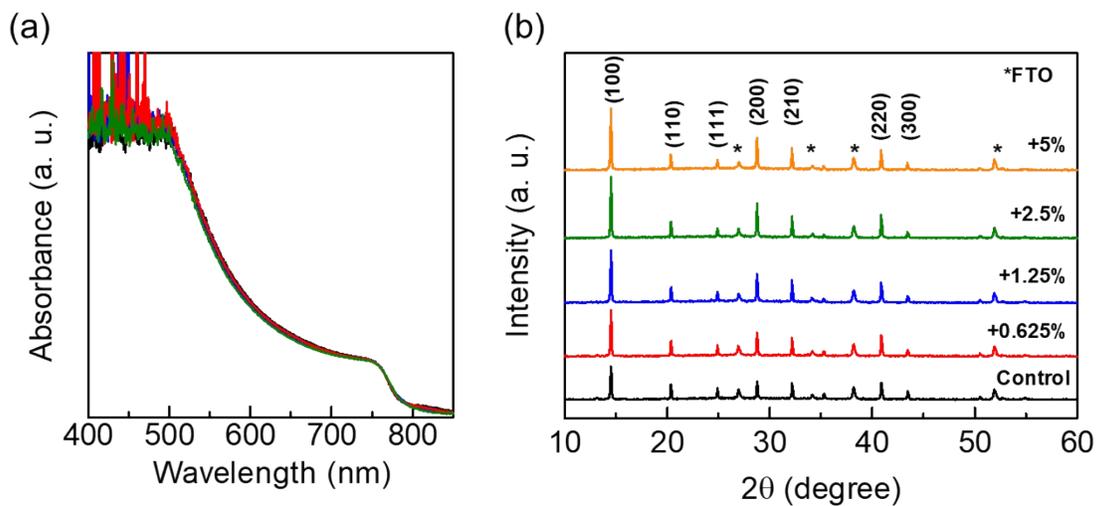


Figure S11 Absorbance spectra and (b) XRD patterns of FoAI added triple-cation perovskite films. The black, red, blue, green and orange line represent 0, 0.625, 1.25, 2.5 and 5 mol% FoAI added films.

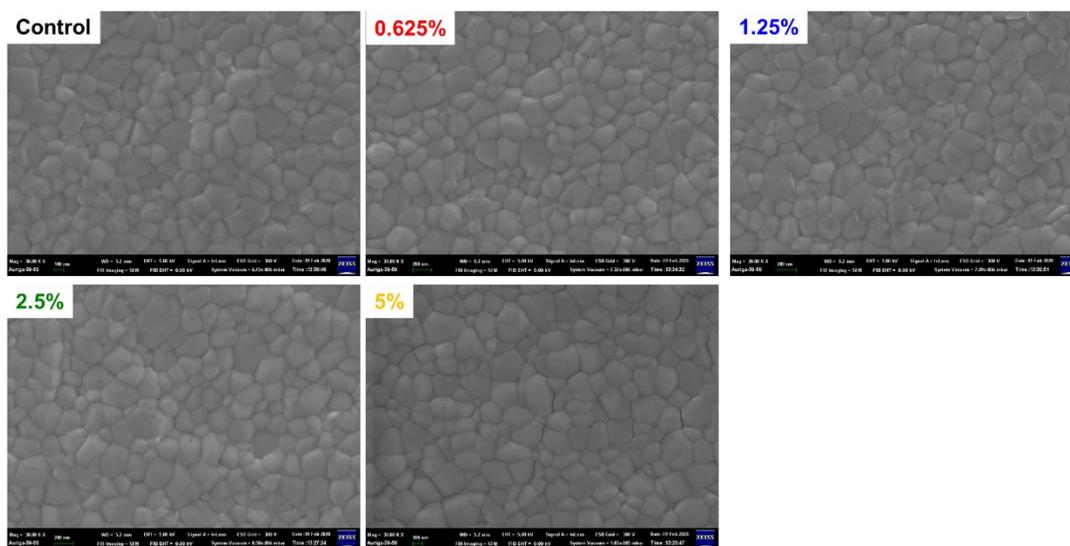


Figure S12 SEM images of FoAI added triple-cation perovskite films.

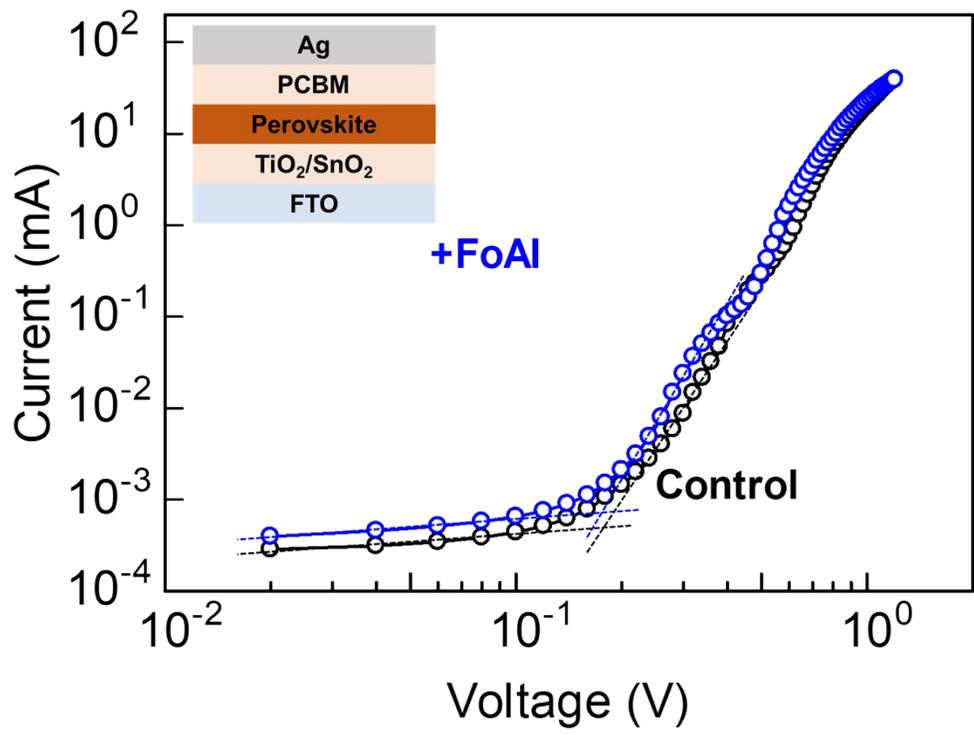


Figure S13 Dark $I-V$ curves of electron-only devices employing pure and FoAl added triple cation perovskite films.

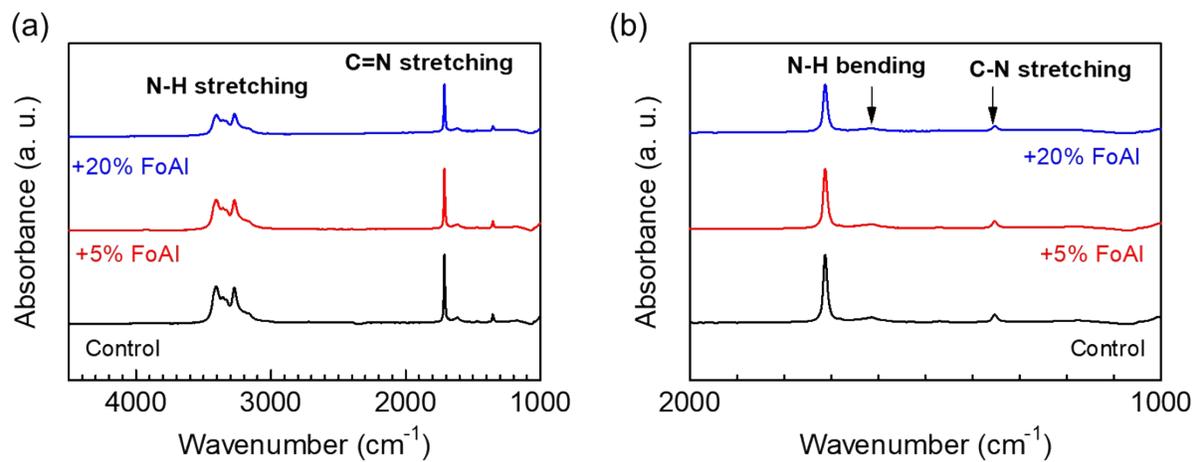


Figure S14 FT-IR spectra of FoAI added triple-cation perovskite films. (a) from 4500 cm^{-1} to 1000 cm^{-1} and (b) from 2000 cm^{-1} to 1000 cm^{-1} .

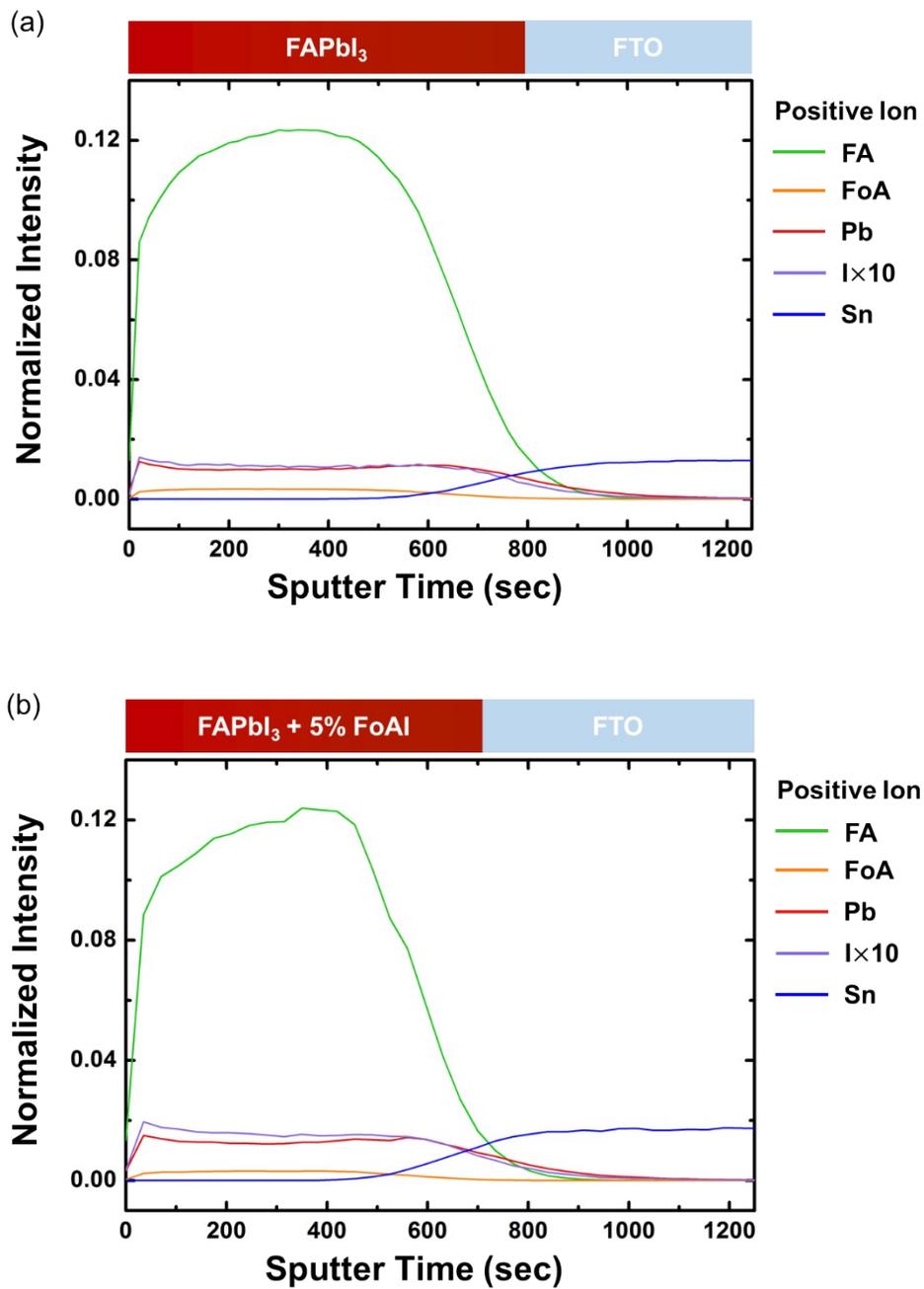


Figure S15 ToF-SIMS profile (positive ion) of (a) pure FAPbI₃ and (b) FoAl added FAPbI₃ films.

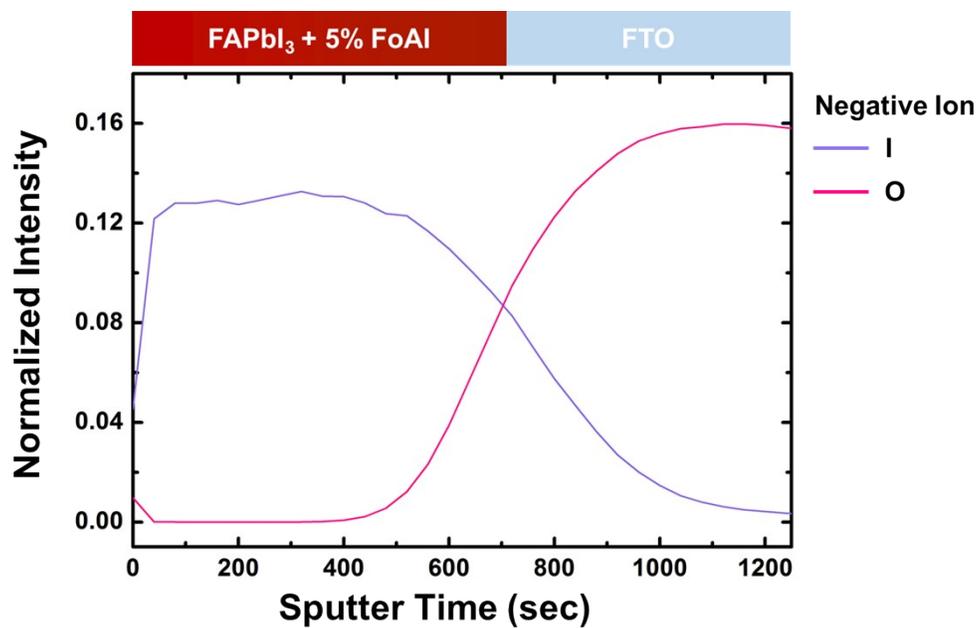


Figure S16 ToF-SIMS profile (negative ion) of FoAl added FAPb₃ films.

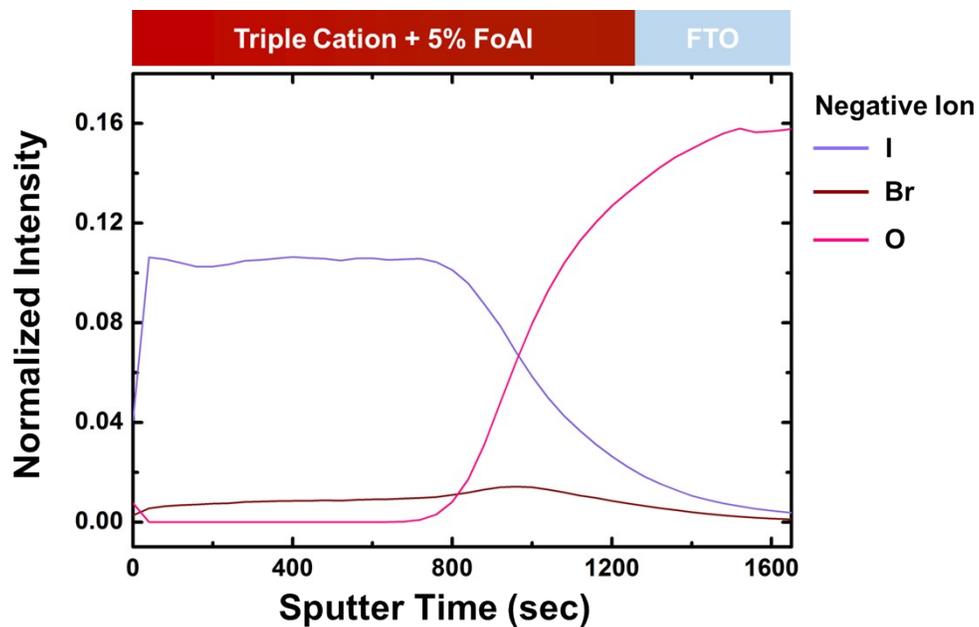


Figure S17 ToF-SIMS profile (negative ion) of 5% FoAl added triple cation perovskite films.

Table S1 The peak positions and assumed bonding in the FT-IR spectrum of formamidine iodide (FAI).

Wavenumber (cm ⁻¹)	Signal
3400-3000	N-H stretching
2800	C-H stretching
1700	C=N stretching
1600	N-H bending
1400-1200	C-N stretching

Table S2 The peak positions and assumed bonding in the FT-IR spectrum of formamide iodide (FoAI).

Wavenumber (cm ⁻¹)	Signal
3300-2900	N-H stretching
2800	C-H stretching
	C-N stretching
1400	C-OH bending

Table S3 Hysteresis index of FAPbI₃ PSCs

Sample	Hysteresis index (%)
Pure FAPbI ₃	6.16±1.67
FoAI added	4.34±1.06

Table S4 Hysteresis index of triple cation PSCs

FoAI concentration (mol%)	Hysteresis index (%)
0	3.94±0.57
0.625	3.81±0.87
1.25	4.23±0.62
2.5	4.28±0.69
5	5.78±0.37