

Supporting Information

Defect clustering in an Eu-doped NaMgF₃ compound and its influence on luminescent properties

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(A) Cubic NaMgF₃

Table S1- Lattice parameters and interatomic distance for cubic NaMgF₃.

Lattice parameters	Ref. 1	This work	%
a(A)	3.876	3.885	0.22
V(A ³)	58.2306	58.6139	0.66
Interatomic distance	Ref.1	This work	%
Na-F (x12)	2.741	2.747	0.22
Mg-F (x6)	1.938	1.942	0.20
F-F (x8)	2.741	2.747	0.22

Table S2- Elastic constants for cubic NaMgF₃.

Elastic Constants	Ref. 2	This work
C ₁₁	136.9082	151.4231
C ₁₂	31.8473	41.7655
C ₄₄	51.0104	41.7655

(B) Fluoride precursors (NaF and MgF₂)

Table S3- Lattice parameters for NaF.

Lattice parameters	Ref. 3	This work	%
a (Å)	4.65	4.6599	0.2129
V (Å ³)	100.5450	101.1895	0.6410

Table S4- Elastic constants for NaF.

Elastic constants	Ref. 4	This work	%
C_{11}	108.5	98.7431	-8.99
C_{12}	22.90	28.3125	23.64
C_{44}	28.99	28.325	-2.34

Table S5- Dielectric constants for NaF

Dielectric constants	Ref. 4	This work	%
ϵ_0	4.73	4.83	2.1
ϵ_∞	1.75	1.91	9.1

Table S6- Lattice parameters for MgF₂.

Lattice parameters	Ref. 5	This work	%
a (Å)	4.6213	4.5447	-1.66
b (Å)	3.0519	3.1540	3.35
V (Å ³)	65.1776	65.1456	-0.05

Table S7- Elastic constants for MgF₂.

Elastic Constants	Ref. 6	Ref. 7	This work
C_{11}	140.80	142.70	120.89
C_{12}	90.00	92.20	94.50
C_{13}	63.50	64.10	68.90
C_{33}	205.30	204.00	213.66
C_{44}	56.70	56.70	52.02
C_{66}	95.70	93.50	95.14

(C) Defect calculations

Table S8- Lattice energy for orthorhombic NaMgF₃, fluorides precursors and dopants.

Materials	Lattice energies (eV)
NaMgF ₃	-39.784
NaF	-9.573
MgF ₂	-29.998
EuF ₃	-51.717
EuF ₂	-25.430

Table S9- Defect formation energy for selected defects.

Defect Energy (eV)				
V_{Na}^{\cdot}	$V_{Mg}^{\bullet\bullet}$	V_F^{\cdot}	Na_{Mg}^{\cdot}	Mg_{Na}^{\bullet}
6.209	24.445	5.821	17.264	-15.016
Interstitial Position		Na _i	Mg _i	F _i
(1/2,1/4,1/4)		-2.848	-16.860	-2.171
(1/2,1/8,1/4)		-2.847	-16.865	-2.472
(1/2,1/2,1/4)		-2.848	-16.851	-2.433
(1/4,1/4,1/4)		-2.848	-16.854	-2.228

Table S10- Defect formation energy of Eu³⁺ and Eu²⁺ isolated ions doped orthorhombic NaMgF₃.

Impurity (Eu ³⁺)	Energy defects (eV)
$Eu_{Na}^{\bullet\bullet}$	-31.047
Eu_{Mg}^{\bullet}	-15.276
Impurity (Eu ²⁺)	Energy defects (eV)
Eu_{Na}^{\bullet}	-10.460
Eu_{Mg}^{\times}	6.564

Table S11- Solution energies of Eu³⁺ doped NaMgF₃.

Schemes	Solution Energies (bound defects) (in eV)
Incorporating into Na site	
$Eu_{Na}^{\bullet\bullet} + 2F_i'$	3.256
$Eu_{Na}^{\bullet\bullet} + V_{Mg}''$	2.758
$Eu_{Na}^{\bullet\bullet} + 2V_{Na}'$	2.914
$Eu_{Na}^{\bullet\bullet} + 2Na_{Mg}'$	2.580
$2Eu_{Na}^{\bullet\bullet} + V_{Mg}'' + 2F_i'$	6.105
Incorporating into Mg site	
$Eu_{Mg}^{\bullet} + F_i'$	2.449
$Eu_{Mg}^{\bullet} + V_{Na}'$	2.129
$2Eu_{Mg}^{\bullet} + V_{Mg}''$	5.641
$Eu_{Mg}^{\bullet} + Na_{Mg}'$	2.843
$2Eu_{Mg}^{\bullet} + Mg_{Na}^{\bullet} + 3F_i'$	5.173
Incorporating into Na and Mg sites	
$Eu_{Mg}^{\bullet} + Eu_{Na}^{\bullet\bullet} + 2V_{Na}' + F_i'$	5.310
$Eu_{Mg}^{\bullet} + Eu_{Na}^{\bullet\bullet} + V_{Mg}'' + Na_{Mg}'$	7.247

Table S12- Solution energies of Eu²⁺ doped NaMgF₃.

Schemes	Solution Energy (bound defects) (in eV)
Incorporating into Na site	
$Eu_{Na}^{\bullet} + F_i'$	2.623
$2Eu_{Na}^{\bullet} + V_{Mg}''$	2.952
$Eu_{Na}^{\bullet} + V_{Na}'$	1.572
$2Eu_{Na}^{\bullet} + V_{Na}' + F_i'$	3.729
$Eu_{Na}^{\bullet} + V_F' + 2F_i'$	2.438
$Eu_{Na}^{\bullet} + Na_{Mg}'$	1.523
$Eu_{Na}^{\bullet} + V_{Mg}'' + V_F'$	3.224
$Eu_{Na}^{\bullet} + 2V_{Na}' + V_F'$	3.589
Incorporating into Na site	
Eu_{Mg}^{\times}	1.996

Reference

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