Bandgap engineering via Boron and Sulphur doped carbon modified Anatase TiO₂: A

Visible light stimulated Photocatalyst for Photo-Fixation of N₂ and TCH Degradation

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Figure. S1 XPS survey spectrum of (a) CT, (b) S-CT and (c) B-CT.



Figure. S2 Bandgap energy of (a) CT and (b) S-CT

Urbach energy calculation

Following the mathematics $\alpha = \alpha_0 \exp[E/E_u]$, where α stands for absorption coefficient, E as the energy of photon and E_u represents Urbach energy respectively. ^[1,2] As the absorption coefficient is equal to absorbance (A or F(R)) for solid samples, so α can be substituted by A. ^[2] The Urbach energy was calculated by taking the reciprocal of the slope of the graph that is plotted between lnA vs E.



Figure. S3 Urbach energy (a) CT, (b) S-CT and (c) B-CT.



Figure. S4 EDX of (a) CT, (b) S-CT and (c) B-CT.



Figure. S5 Elemental colour mapping of (a) CT and (b) B-CT.

Catalytic system	Antibiotic type	Degradation%	Light	Time	Reference
			source	period	
Au/B-TiO ₂ /rGO	15ppm TCH	100	300W Xe	60min	3
			lamp		
Cu-	30ppm	100	Open	180min	4
TiO2@functionalized	sulfamethazine		sunlight		
SWCNT					
Al-doped TiO ₂	$(2*10^{-4}M)$	93	(103800	120min	5

Nanoflakes	Fluoroquinolone		lux as per HTC LX102A lux meter		
Co-TNs/rGO	30ppm TCH	60	Halogen, ECO OSRAM 500 W lamp	180min	6
N-doped TiO ₂	$1.0 \times 10^{-2} \text{ mol}$ L ⁻¹ cefazolin	80	5 × 8W blacklight fluorescent lamps	30min	7
Co-doped TiO ₂	10ppm	>90	Xe lamp	240 min of irradiation under UV-C and 300 min under visible irradiation	8
Carbon-sensitized and nitrogen-doped TiO ₂	5ppm	95	LED flexible strip (SMD 5016 water research 45 (2011) 5015 - 5026 5050, 15W)	5h	9
Nitrogen-doped TiO ₂ /diatomite	20ppm	91	150W Xe lamp	90min	10
B-CT	10ppm	95	250W Xe- lamp	60min	Present work



Table. T1 comparison table representing different doped TiO₂ towards antibiotic degradation.

Figure. S6 LC-MS chromatogram of TCH over B-CT.

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