

Electronic Supplementary Information for:

An All-Inorganic, Fully Dense, Stretchable Ceramic Magnetic Film

Muchun Liu^{1,2*}, Lijuan Qian³, Chao Yu¹, Gang Xiao^{3*}, Robert H. Hurt^{1*}

¹ School of Engineering, Brown University; ² Department of Civil and Environmental Engineering, Massachusetts Institute of Technology; ³ Department of Physics, Brown University

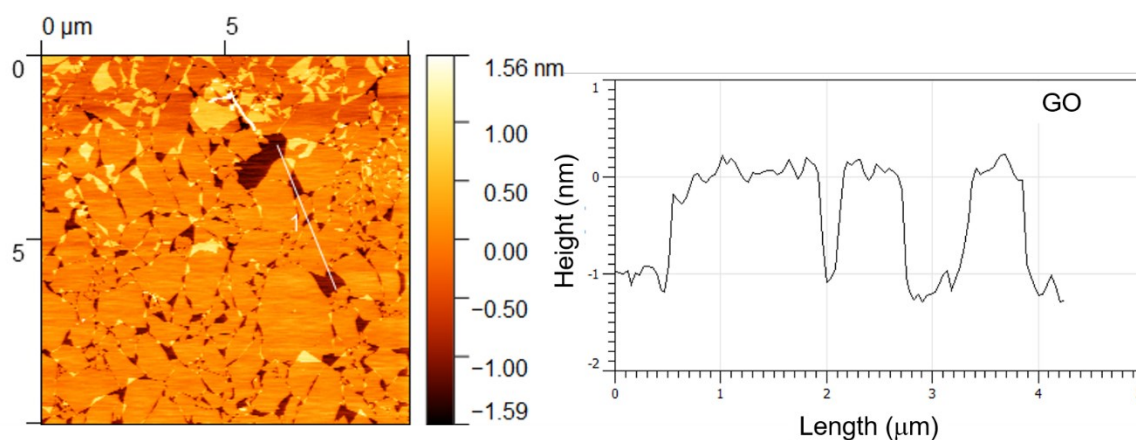


Fig. S1. Morphologies of GO nanosheets. AFM image and accompanying height profile of GO nanosheets drop-cast from diluted suspension onto plasma treated mica. GO nanosheets show lateral size of $\sim 1 \mu\text{m}$, and thickness of $\sim 1 \text{ nm}$.

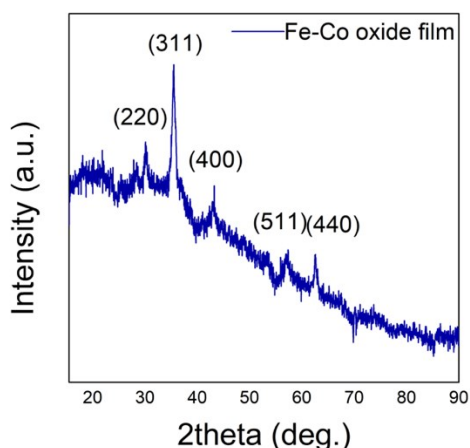


Fig. S2. XRD spectrum of cobalt ferrite pre-folded films (SMOF).

Fig. S3 shows results of thermal gravimetric analysis of GO-Fe(III)/Co(II) films, GO and $\text{Fe}(\text{NO}_3)_3/\text{Co}(\text{NO}_3)_2$ salts carried out in air at a heating rate of $10\text{ }^\circ\text{C min}^{-1}$. From the TGA curves we can see the decomposition of $\text{Fe}(\text{NO}_3)_3/\text{Co}(\text{NO}_3)_2$ is complete at $\sim 200\text{ }^\circ\text{C}$ with weight retention of 32.8%, which is close to theoretical value ($\text{CoFe}_2\text{O}_4/[2*\text{Fe}(\text{NO}_3)_3+\text{Co}(\text{NO}_3)_2] = 35.1\%$). During the annealing of GO, the first weight loss at $\sim 200\text{ }^\circ\text{C}$ is caused by the thermal decomposition and deoxygenation GO, while the second weight loss at $\sim 450\text{ }^\circ\text{C}$ is rGO oxidation. In GO-Fe(III)/Co(II) films, the weight retention (25.1%) is also close to theoretical value ($\text{CoFe}_2\text{O}_4/[2*\text{Fe}(\text{NO}_3)_3+\text{Co}(\text{NO}_3)_2+\text{GO}] = 20.1\%$). Since GO is fully oxidized over $500\text{ }^\circ\text{C}$, the formation of CoFe_2O_4 must be occurring while the graphene-based scaffold is intact, which enables the 2D assembly and replication of wrinkled topographies.

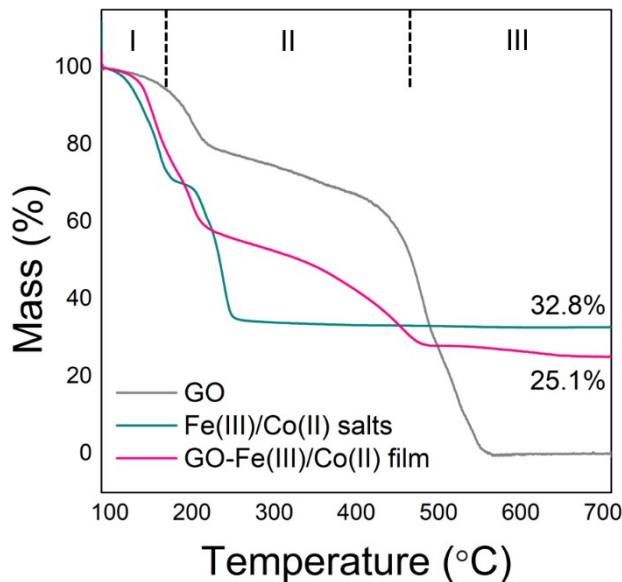


Fig. S3. TGA curves of GO-Fe(III)/Co(II) films, GO and $\text{Fe}(\text{NO}_3)_3/\text{Co}(\text{NO}_3)_2$ salts in air. The salts start to decompose at $\sim 180\text{ }^\circ\text{C}$ (phase I). The thermal reduction of GO to rGO takes place between 150 to $200\text{ }^\circ\text{C}$ (phase II) and the combustion of rGO starts at $450\text{ }^\circ\text{C}$ (phase III). The transformation of $\text{Fe}(\text{NO}_3)_3/\text{Co}(\text{NO}_3)_2$ salts to CoFe_2O_4 is completed at $\sim 200\text{ }^\circ\text{C}$, while GO is fully oxidized over $500\text{ }^\circ\text{C}$. Heating rate: $10\text{ }^\circ\text{C min}^{-1}$.

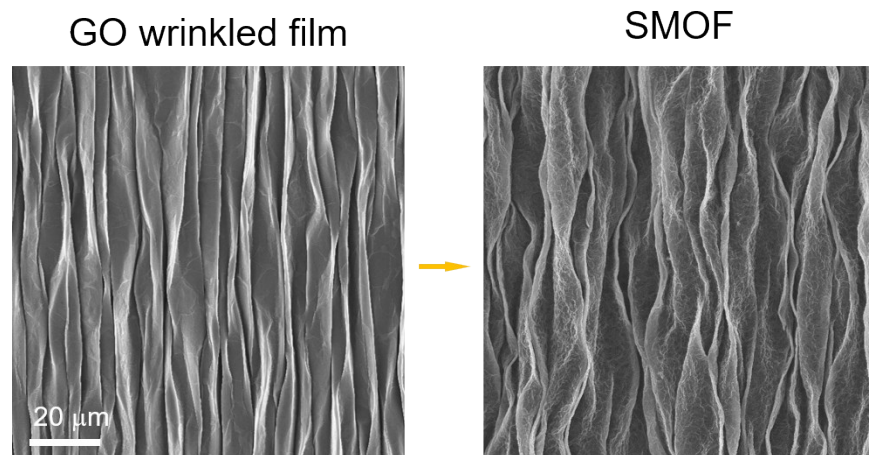


Fig. S4. Surface morphologies of GO wrinkled film and SMOF. Scale bar, 20 μm .

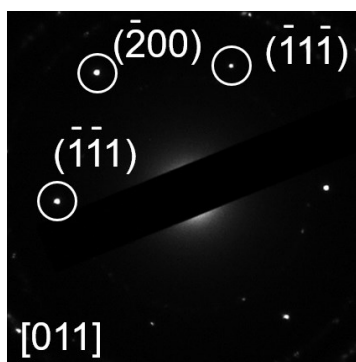


Fig. S5. SAED of CoFe_2O_4 single nanoplatelet. The resulting nanoplatelet with basal surfaces that are primarily (011) planes.

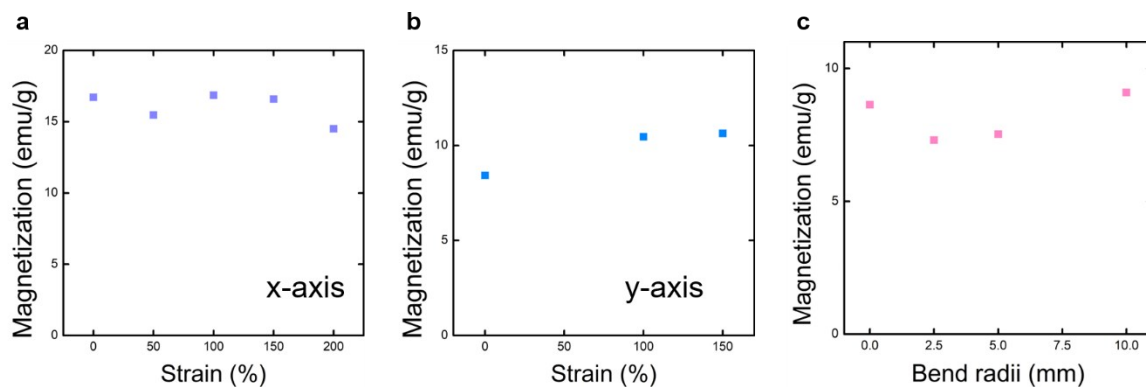


Fig. S6. Remnant magnetization of SMOFs (a) long x-axis during stretching, (b) along y-axis during stretching and (c) along x-axis during bending.