

Electronic Supplementary Information (ESI)

Relating X-Ray Photoelectron Spectroscopy Data to Chemical Bonding in MXenes

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Table S1. Experimental results of XPS analysis in MXenes. Bold numbers are selected to compare and discuss our theoretical predictions.

| MXene | Region | Binding energy (eV) | Asignation | Ref |
|---|--------|---|---|-----|
| Nb ₂ CT _z | O (1s) | 530.0 531.8 532.4 | O in Nb ₂ O ₅ O/HO-Nb₂C Water trapped between MXene layer | [1] |
| | C (1s) | 281.8 282.0 283.6 | C in Nb₂C Non-stoichiometric C in Nb ₂ C Adventitious contamination | |
| V ₂ CT _z | O (1s) | 529.9 531.3 532.9 | Mixed VO _x and surface oxide groups Mixed VO_x and surface oxide groups Water trapped between MXene layer | [1] |
| | C (1s) | 281.8 286.1 284.5 288.3 | C in V₂C Adventitious contamination Adventitious contamination Adsorbed methanol | |
| Ti ₂ CT _z before sputtering | O (1s) | 531.1 532.0 533.1 | C-Ti-O_z C-Ti-(OH) _z Water trapped between MXene layer | [2] |
| | C (1s) | 281.9 286.2 289.0 | C-Ti-T_z C-O C-OO | |
| Ti ₂ CT _z after sputtering | O (1s) | 531.3 532.2 533.5 | C-Ti-O_z C-Ti-(OH) _z Water trapped between MXene layer | |
| | C (1s) | 281.9 286.2 | C-Ti-T_z C-O | |
| Mo ₂ CT _z | O (1s) | 531.1 532.2 533.4 | Mo₂CO_x (-O terminated) Mo ₂ C(OH) _x (-OH terminated) Water trapped between MXene layer | [3] |
| | C (1s) | 283.1 | Mo-C species | |
| MultiLayered Cr ₂ CT _z | O (1s) | 530.2 531.2 | O-Cr-C (-O terminated) HO-Cr-C | [4] |
| | C (1s) | 283.2 | C-Cr-T_z | |
| FewLayered Cr ₂ CT _z | O (1s) | 530.4 531.6 | O-Cr-C (-O terminated) HO-Cr-C | [4] |
| | C (1s) | 283.5 | C-Cr-T_z | |

Table S2. Kohn-Sham orbital energy (ϵ_{1s}), Fermi energy (ϵ_F), initial state C(1s) CLBE relative to the Fermi energy of each system (see Eq. 2) and net charge on the C atom (Q_C) of partial relaxed O-terminated M_2CO_2 MXenes. The number of d electrons in the corresponding metal (d^n) is also provided. The units of energetic parameters (ϵ_{1s} , ϵ_F , and C(1s) CLBE) are in eV, whereas Q_C units are |e|. Note that the O layers are relaxed maintaining the M_2C fixed at the clean MXene.

| MXene | d^n | ϵ_{1s} | ϵ_F | CLBE | Q_C |
|---------------------------------|-------------------------|-----------------------------------|--------------------------------|-------------|-------------------------|
| Ti ₂ CO ₂ | d ² | -267.8 | -3.6 | 264.2 | -1.9 |
| Zr ₂ CO ₂ | | -265.6 | -2.0 | 263.6 | -1.9 |
| Hf ₂ CO ₂ | | -265.2 | -1.5 | 263.7 | -1.9 |
| V ₂ CO ₂ | d ³ | -268.0 | -2.7 | 265.3 | -1.6 |
| Nb ₂ CO ₂ | | -267.6 | -2.0 | 265.6 | -1.8 |
| Ta ₂ CO ₂ | | -267.4 | -1.6 | 265.8 | -2.1 |
| Cr ₂ CO ₂ | d ⁴ | -267.7 | -2.2 | 265.5 | -1.3 |
| Mo ₂ CO ₂ | | -268.3 | -1.7 | 266.6 | -1.4 |
| W ₂ CO ₂ | | -267.3 | -1.5 | 265.8 | -1.5 |

Table S3. Kohn-Sham orbital energy (ϵ_{1s}), Fermi energy (ϵ_F), initial state O(1s) CLBE relative to the Fermi energy of each system (see Eq. 2) and net charge on the O atom (Q_O) of partial relaxed O-terminated M_2CO_2 MXenes. The number of d electrons in the corresponding metal (d^n) is also provided. The units of energetic parameters (ϵ_{1s} , ϵ_F , and O(1s) CLBE) are in eV, whereas Q_O units are |e|. Note that the O layers are relaxed maintaining the M_2C fixed at the clean MXene.

| MXene | d^n | ϵ_{1s} | ϵ_F | CLBE | Q_o |
|---------------------------------|-------------------------|-----------------------------------|--------------------------------|-------------|-------------------------|
| Ti ₂ CO ₂ | d ² | -508.8 | -3.6 | 505.2 | -1.0 |
| Zr ₂ CO ₂ | | -507.4 | -2.0 | 505.4 | -1.2 |
| Hf ₂ CO ₂ | | -507.0 | -1.5 | 505.5 | -1.3 |
| V ₂ CO ₂ | d ³ | -508.8 | -2.7 | 506.1 | -1.0 |
| Nb ₂ CO ₂ | | -508.3 | -2.0 | 506.3 | -1.1 |
| Ta ₂ CO ₂ | | -508.2 | -1.6 | 506.6 | -1.2 |
| Cr ₂ CO ₂ | d ⁴ | -508.7 | -2.2 | 506.5 | -0.9 |
| Mo ₂ CO ₂ | | -508.6 | -1.7 | 506.9 | -0.9 |
| W ₂ CO ₂ | | -508.2 | -1.5 | 506.7 | -1.0 |

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