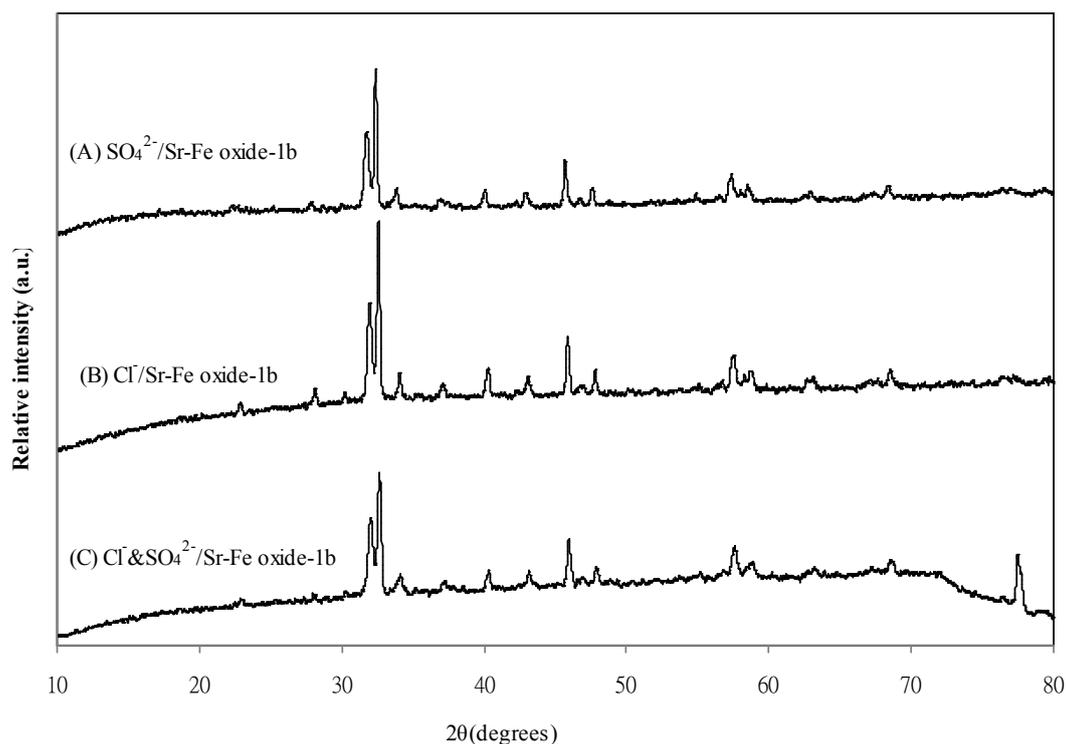


## SUPPLEMENTARY INFORMATION

### Sulfated/chlorinated Sr-Fe composite oxide as a novel solid and reusable superacid catalyst for oleic acid esterification

Chien-Chang Huang<sup>\*a</sup>, Shih-Han Ho<sup>a</sup>, Jo-Shu Chang<sup>b,c,d</sup> and Pei-Jyuan Gao<sup>a</sup>



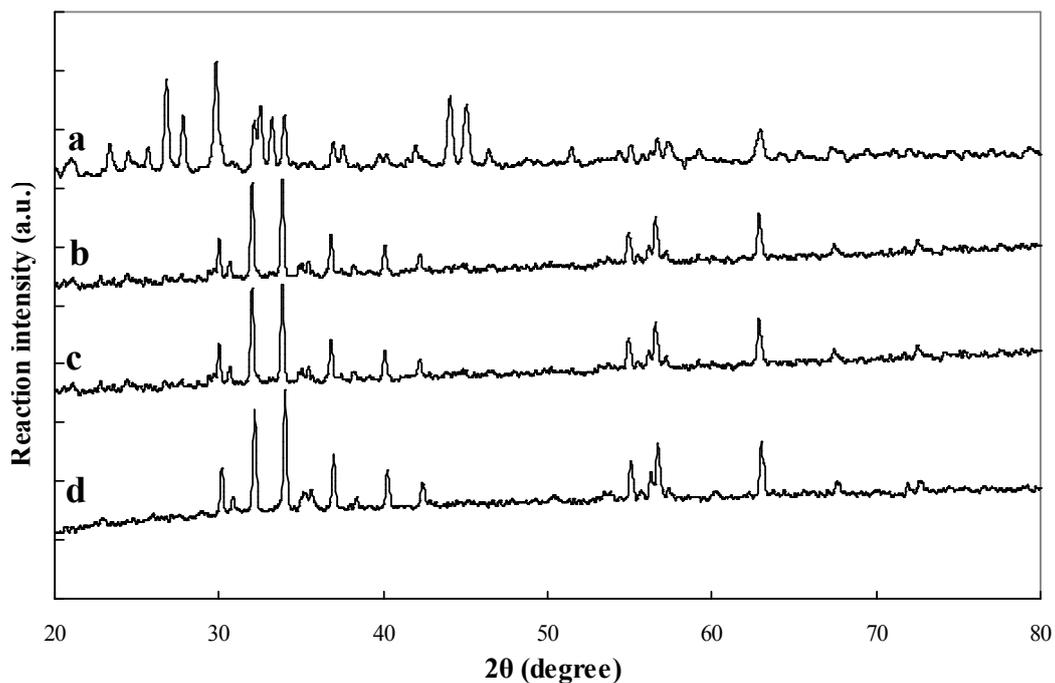
S1. Powder XRD pattern of the Sr-Fe oxide-1 modified with (A) Na<sub>2</sub>SO<sub>4</sub> (SO<sub>4</sub><sup>2-</sup>/Sr-Fe oxide-1b) (B) KCl (Cl<sup>-</sup>/Sr-Fe oxide-1b) and Na<sub>2</sub>SO<sub>4</sub> and KCl (Cl<sup>-</sup>&SO<sub>4</sub><sup>2-</sup>/Sr-Fe oxide-1b).

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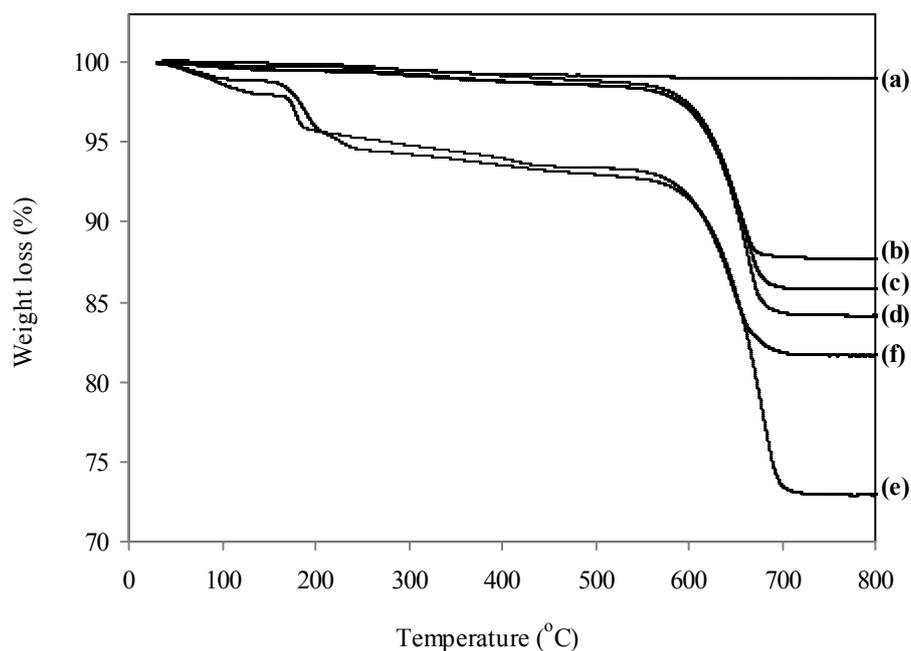
<sup>b</sup> Department of Chemical and Materials Engineering, College of Engineering, Tunghai University, Taichung 40704, Taiwan

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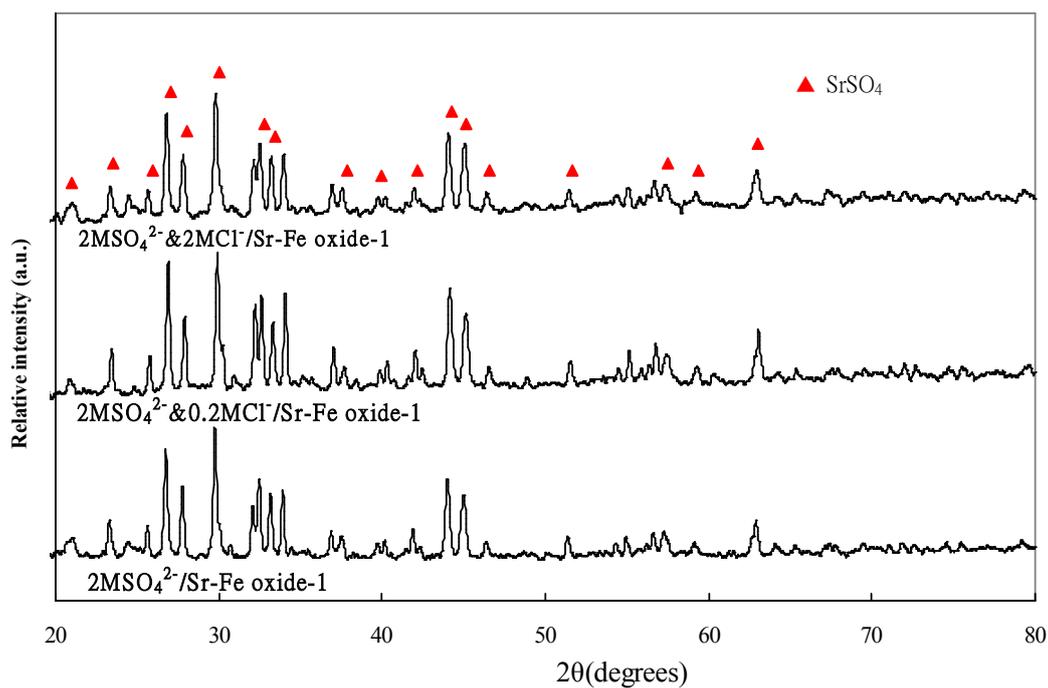
<sup>d</sup> Department of Chemical Engineering, National Cheng Kung University, 1 University Road, Tainan 701, Taiwan.. Tel: +886 6275 7575



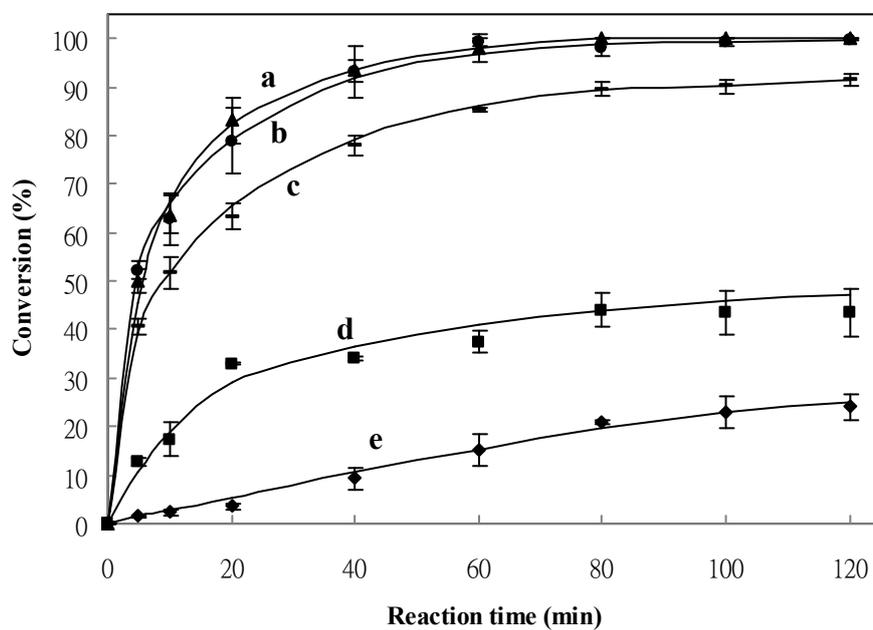
S2. The diffractogram of (a).  $2\text{MSO}_4^{2-}$  &  $2\text{MCl}$  / Sr-Fe oxide-1, (b).  $2\text{MSO}_4^{2-}$  -  $2\text{MCl}$  / Sr-Fe oxide-1, (c).  $2\text{MCl}$  -  $2\text{MSO}_4^{2-}$  / Sr-Fe oxide-1 and (d).  $2\text{MCl}$  / Sr-Fe oxide-1



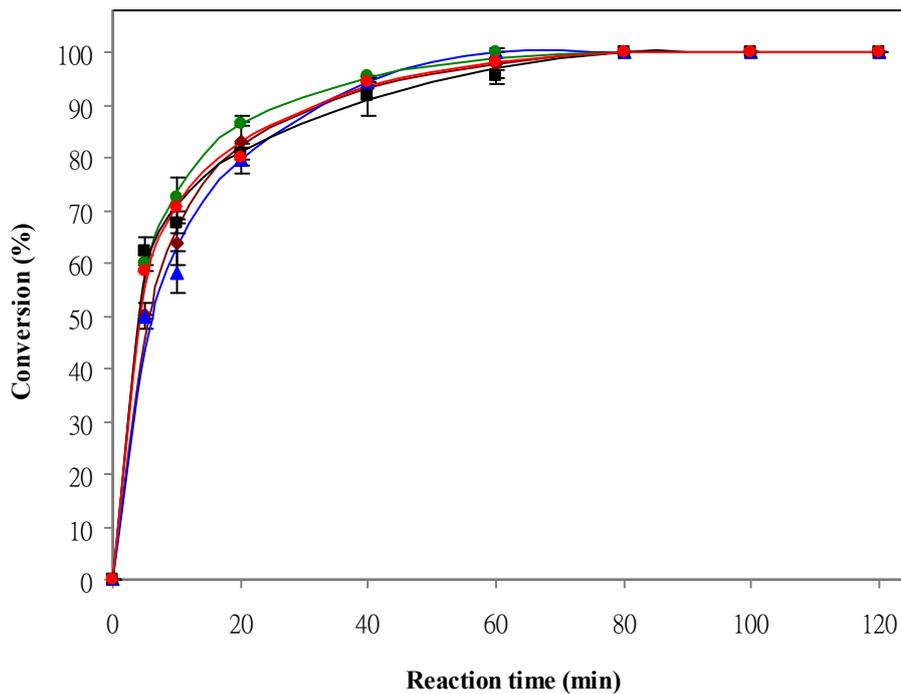
S3. TGA profiles of (a) Sr-Fe oxide-1 (b).  $2\text{MSO}_4^{2-}$  &  $0.1\text{M Cl}$  / Sr-Fe oxide-1, (c).  $2\text{MSO}_4^{2-}$  &  $0.2\text{MCl}$  / Sr-Fe oxide-1, (d).  $2\text{MSO}_4^{2-}$  &  $0.5\text{MCl}$  / Sr-Fe oxide-1 (e).  $2\text{MSO}_4^{2-}$  &  $1\text{MCl}$  / Sr-Fe oxide-1 and (f).  $2\text{MSO}_4^{2-}$  &  $2\text{MCl}$  / Sr-Fe oxide-1.



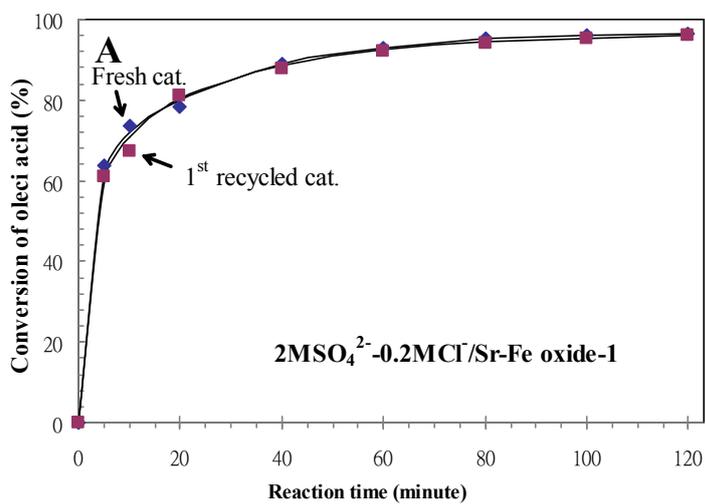
S4. XRD patterns of  $2\text{MSO}_4^{2-} / \text{Sr-Fe oxide-1}$ ,  $2\text{MSO}_4^{2-} \& 0.2\text{MCl} / \text{Sr-Fe oxide-1}$  and  $2\text{MSO}_4^{2-} \& 2\text{MCl} / \text{Sr-Fe oxide-1}$ .

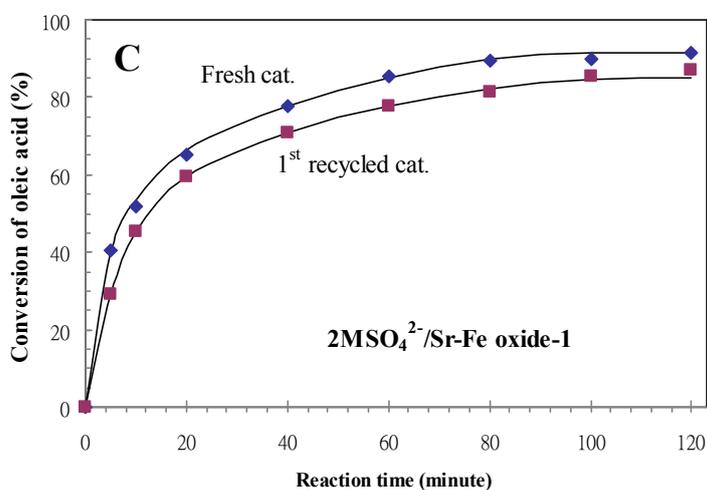
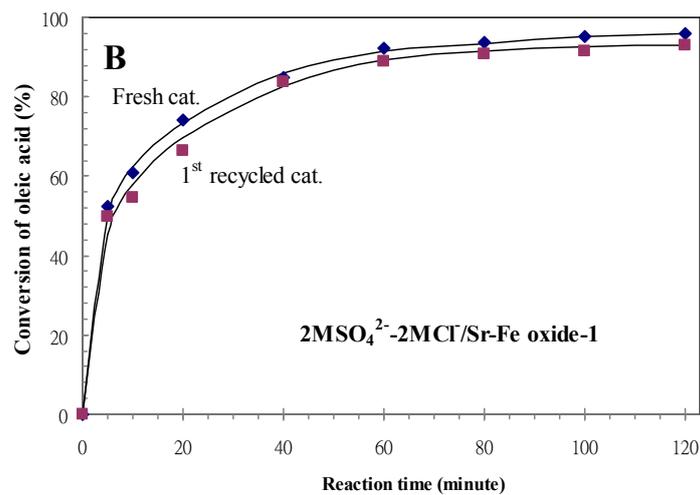


S5. Conversion versus reaction time for oleic acid esterification catalyzed by (a).  $2\text{MSO}_4^{2-} \& 2\text{MCl} / \text{Sr-Fe oxide-1}$ , (b).  $2\text{MCl} - 2\text{MSO}_4^{2-} / \text{Sr-Fe oxide-1}$ , (c).  $2\text{MSO}_4^{2-} / \text{Sr-Fe oxide-1}$ , (d).  $2\text{MSO}_4^{2-} - 2\text{MCl} / \text{Sr-Fe oxide-1}$  and (e).  $2\text{MCl} / \text{Sr-Fe oxide-1}$  at 100°C.

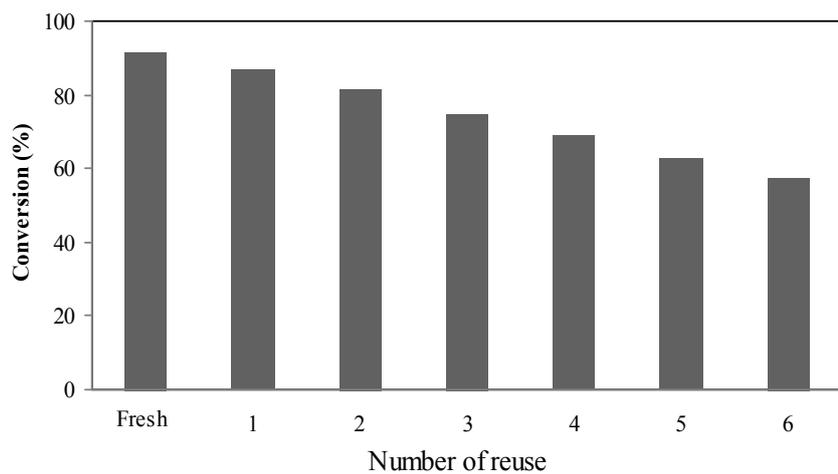


S6. Conversion versus reaction time for oleic acid esterification catalyzed by ◆  $2\text{MSO}_4^{2-}$  &  $2\text{MCl}^-$  / Sr-Fe oxide-1, ▲  $2\text{MSO}_4^{2-}$  &  $1\text{MCl}^-$  / Sr-Fe oxide-1, ◆  $2\text{MSO}_4^{2-}$  &  $0.5\text{MCl}^-$  / Sr-Fe oxide-1, ■  $2\text{MSO}_4^{2-}$  &  $0.2\text{MCl}^-$  / Sr-Fe oxide-1 and ●  $2\text{MSO}_4^{2-}$  &  $0.1\text{MCl}^-$  / Sr-Fe oxide-1 at  $100^\circ\text{C}$ .

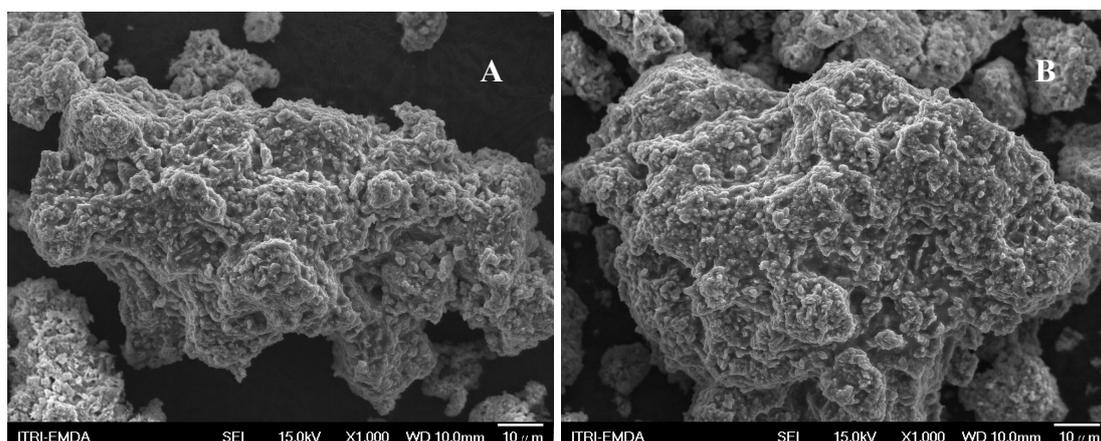




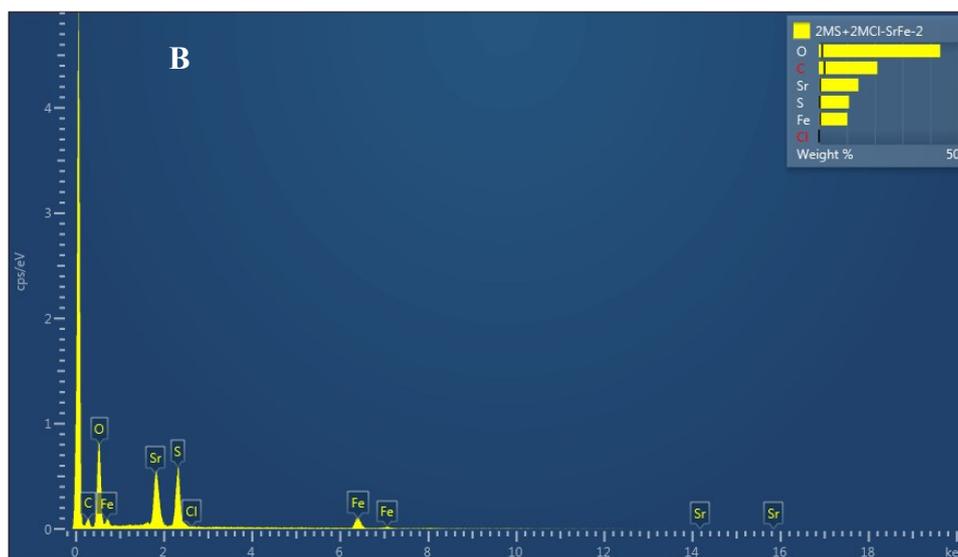
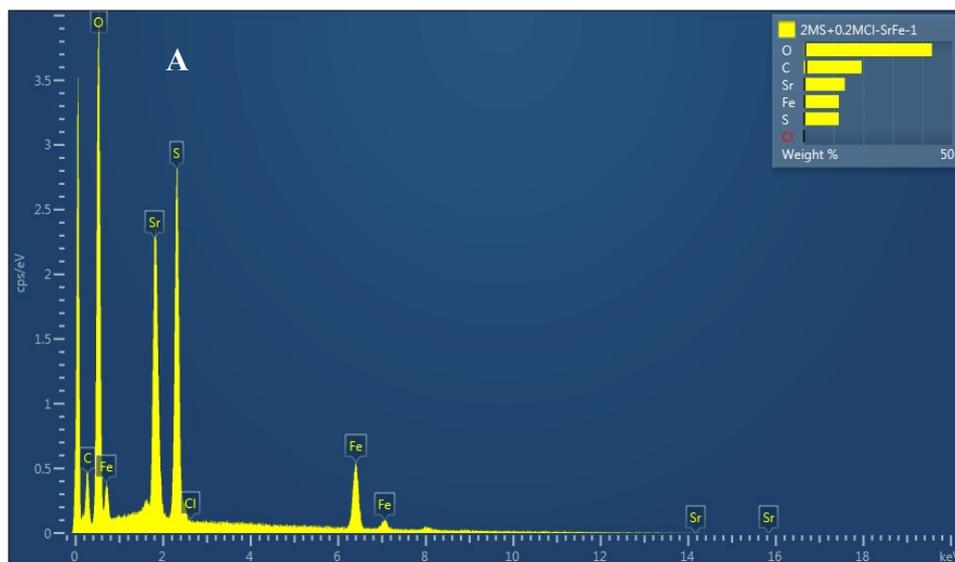
S7. Catalytic conversion of oleic acid over fresh and 1<sup>st</sup> recycled (A).  $2\text{MSO}_4^{2-}$  &  $0.2\text{MCl/Sr-Fe oxide-1}$ , (B).  $2\text{MSO}_4^{2-}$  &  $2\text{MCl/Sr-Fe oxide-1}$  and (C).  $2\text{MSO}_4^{2-}/\text{Sr-Fe oxide-1}$  as a function of reaction time. (Reaction temperature =  $100^\circ\text{C}$ ).



S8. Stability studies of  $\text{SO}_4^{2-}/\text{Sr-Fe oxide-1}$  in catalyzing the esterification of oleic acid with methanol.



S9. SEM images of (A).  $2\text{MSO}_4^{2-}\&0.2\text{MCl}^-/\text{Sr-Fe oxide-1}$  and (B).  $2\text{MSO}_4^{2-}\&2\text{MCl}^-/\text{Sr-Fe oxide-1}$



S10. surface-elemental mapping results of (A).  $2\text{MSO}_4^{2-}$ -& $0.2\text{MCl}^-/\text{Sr-Fe oxide-1}$  and (B).  $2\text{MSO}_4^{2-}$ -& $2\text{MCl}^-/\text{Sr-Fe oxide-1}$