

Electronic Supplementary Material

**A Facile Synthesis of Mesoporous Graphitic Carbon Nitride Supported Palladium Nanoparticles as
Highly Effective and Reusable Catalysts for the Stille Coupling Reactions under Mild Conditions**

Erbay Kalay,^{[a],[b]} Sultan Cetin,^[a] Safacan Kolemen,^{[a],[c],*} Önder Metin^{[a],[c],*}

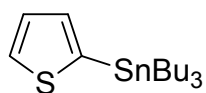
^[a]Department of chemistry, College of Sciences, Koç University, 34450 Istanbul, TURKEY

^[b]Kars Vocational School, Kafkas University, 36100 Kars, TURKEY

^[c]Surface Science and Technology Center (KUYTAM), Koç University, 34450 Istanbul, TURKEY

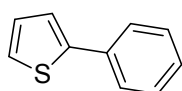
e-mail: ometin@ku.edu.tr, skolemen@ku.edu.tr

Tributyl(thiophen-2-yl)stannane ¹



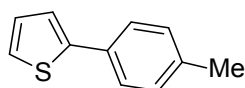
¹H NMR (500 MHz, Chloroform-*d*) δ 7.69 – 7.66 (m, 1H), 7.30 – 7.28 (m, 1H), 7.22 (dd, $J = 3.3, 0.8$ Hz, 1H), 1.65 – 1.56 (m, 6H), 1.37 (sext, $J = 7.3$ Hz, 6H), 1.18 – 1.10 (m, 6H), 0.93 (t, $J = 7.3$ Hz, 9H).

2-phenylthiophene ²



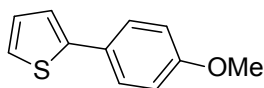
Light beige crystals; ¹H NMR (500 MHz, Chloroform-*d*) δ 7.62 (d, $J = 8.3, 1.2$ Hz, 2H), 7.38 (t, $J = 7.7$ Hz, 2H), 7.32-7.26 (m, 3H), 7.08 (dd, $J = 5.1, 3.6$ Hz, 1H),

2-(*p*-tolyl)thiophene ³



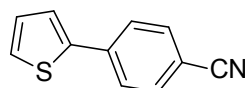
White solid; ¹H NMR (500 MHz, Chloroform-*d*) δ 7.53 (d, $J = 8.2$ Hz, 2H), 7.29-7.26 (m, 2H), 7.21 (d, $J = 7.9$ Hz, 2H), 7.08 (dd, $J = 5.1, 3.6$ Hz, 1H), 2.39 (s, 3H).

2-(4-methoxyphenyl)thiophene ⁴



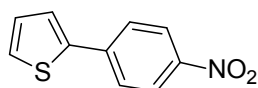
Colorless solid; ¹H NMR (500 MHz, Chloroform-*d*) δ 7.55 (d, $J = 8.9$ Hz, 2H), 7.23-7.21 (m, 2H), 7.06 (dd, $J = 5.1, 3.6$ Hz, 1H), 6.94-6.92 (m, 2H), 3.85 (s, 3H).

4-(thiophen-2-yl)benzonitrile ⁵



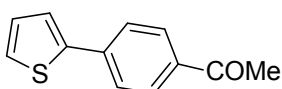
White solid; ¹H NMR (500 MHz, Chloroform-*d*) δ 7.71-7.65 (m, 4H), 7.43 (dd, $J = 3.7, 1.0$ Hz, 1H), 7.41 (dd, $J = 5.1, 1.0$ Hz, 1H), 7.14 (dd, $J = 5.1, 3.7$ Hz, 1H).

2-(4-nitrophenyl)thiophene ⁶



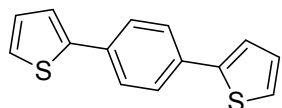
Yellow solid; ¹H NMR (500 MHz, Chloroform-*d*) δ 8.24 (d, *J* = 8.9 Hz, 2H), 7.75 (d, *J* = 8.9 Hz, 2H), 7.48 (dd, *J* = 3.7, 1.1 Hz, 1H), 7.44 (dd, *J* = 5.1, 1.1 Hz, 1H), 7.15 (dd, *J* = 5.1, 3.7 Hz, 1H).

1-(4-(thiophen-2-yl)phenyl)ethanone ⁶



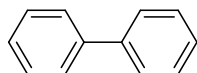
White solid; ¹H NMR (500 MHz, Chloroform-*d*) δ 7.98 (d, *J* = 8.6 Hz, 2H), 7.70 (d, *J* = 8.6 Hz, 2H), 7.44 (dd, *J* = 3.7, 1.1 Hz, 1H), 7.38 (dd, *J* = 5.1, 1.1 Hz, 1H), 7.13 (dd, *J* = 5.1, 3.7 Hz, 1H), 2.62 (s, 3H).

1,4-di(thiophen-2-yl)benzene ⁷



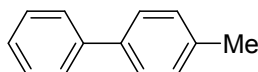
Shiny yellow spangles; ¹H NMR (500 MHz, Chloroform-*d*) δ 7.64 (s, 4H), 7.35 (dd, *J* = 3.6, 1.0 Hz, 2H), 7.30 (dd, *J* = 5.1, 1.0 Hz, 2H), 7.11 (dd, *J* = 5.1, 3.6 Hz, 2H).

1,1'-biphenyl ⁴



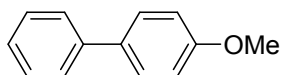
White solid; ¹H NMR (500 MHz, Chloroform-*d*) δ 7.65-7.64 (m, 4H), 7.50-7.47 (m, Hz, 4H), 7.38 (m, 2H).

4-methyl-1,1'-biphenyl ⁸



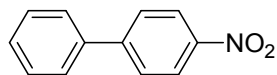
White solid; ¹H NMR (500 MHz, Chloroform-*d*) δ 7.62 (td, *J* = 8.7, 8.1, 1.2 Hz, 2 H), 7.53-7.50 (m, 2 H), 7.47-7.42 (m, 2 H), 7.35-7.32 (m, 1 H), 7.28 (d, *J* = 8.5 Hz, 2 H), 2.43 (s, 3H).

4-methoxy-1,1'-biphenyl ⁹



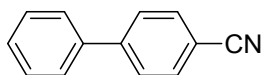
White solid; ^1H NMR (500 MHz, Chloroform-*d*) δ 7.58-7.54 (m, 4H), 7.45-7.41 (m, 2H), 7.34-7.30 (m, 1H), 7.00 (d, J = 8.8 Hz, 2H), 3.87 (s, 3H).

4-nitro-1,1'-biphenyl ¹⁰



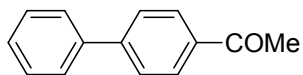
Yellow solid; ^1H NMR (500 MHz, Chloroform-*d*) δ 8.31 (d, J = 8.8 Hz, 2H), 7.74 (d, J = 8.8 Hz, 2H), 7.66-7.62 (m, 2H), 7.53-7.50 (m, 2H), 7.48-7.45 (m, 1H).

[1,1'-biphenyl]-4-carbonitrile ¹⁰

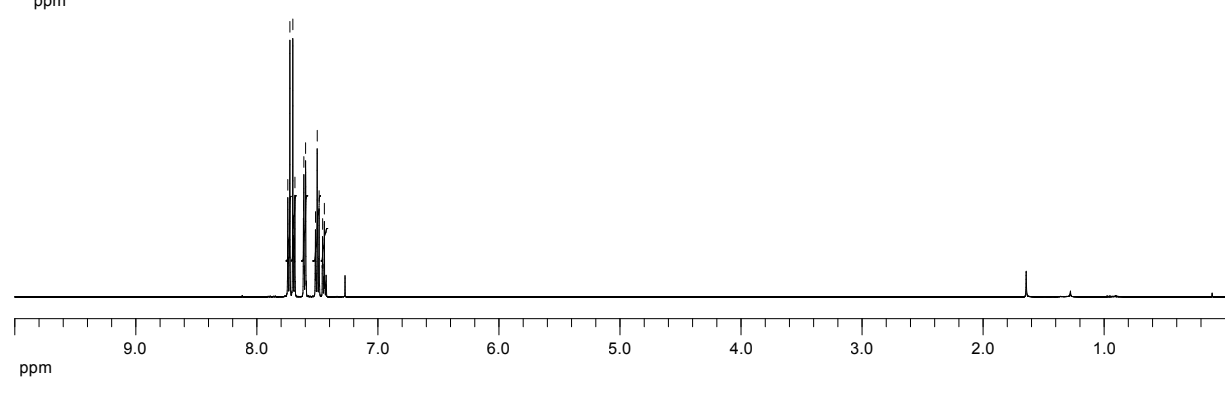
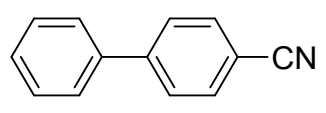
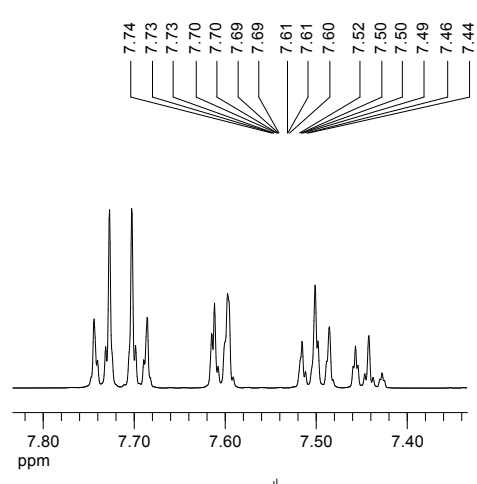
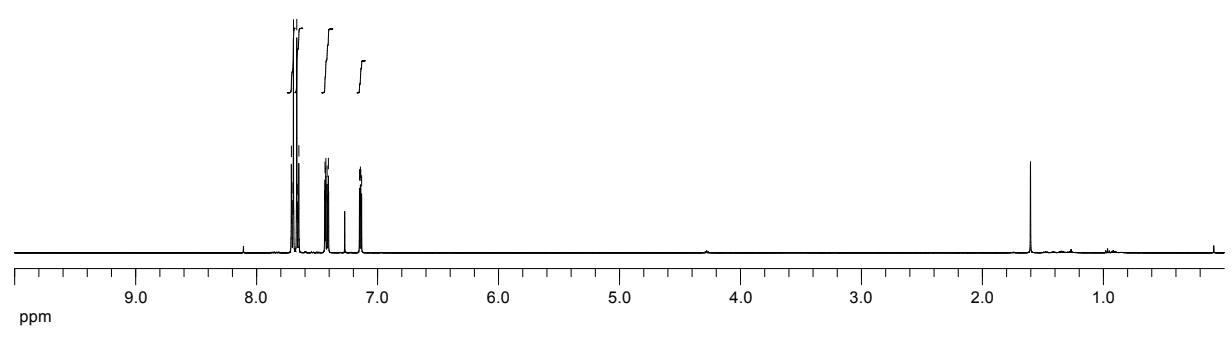
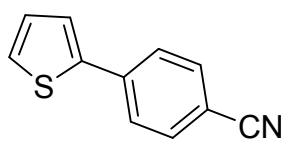
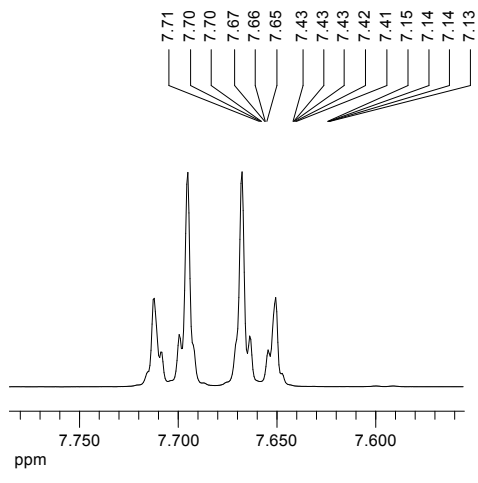


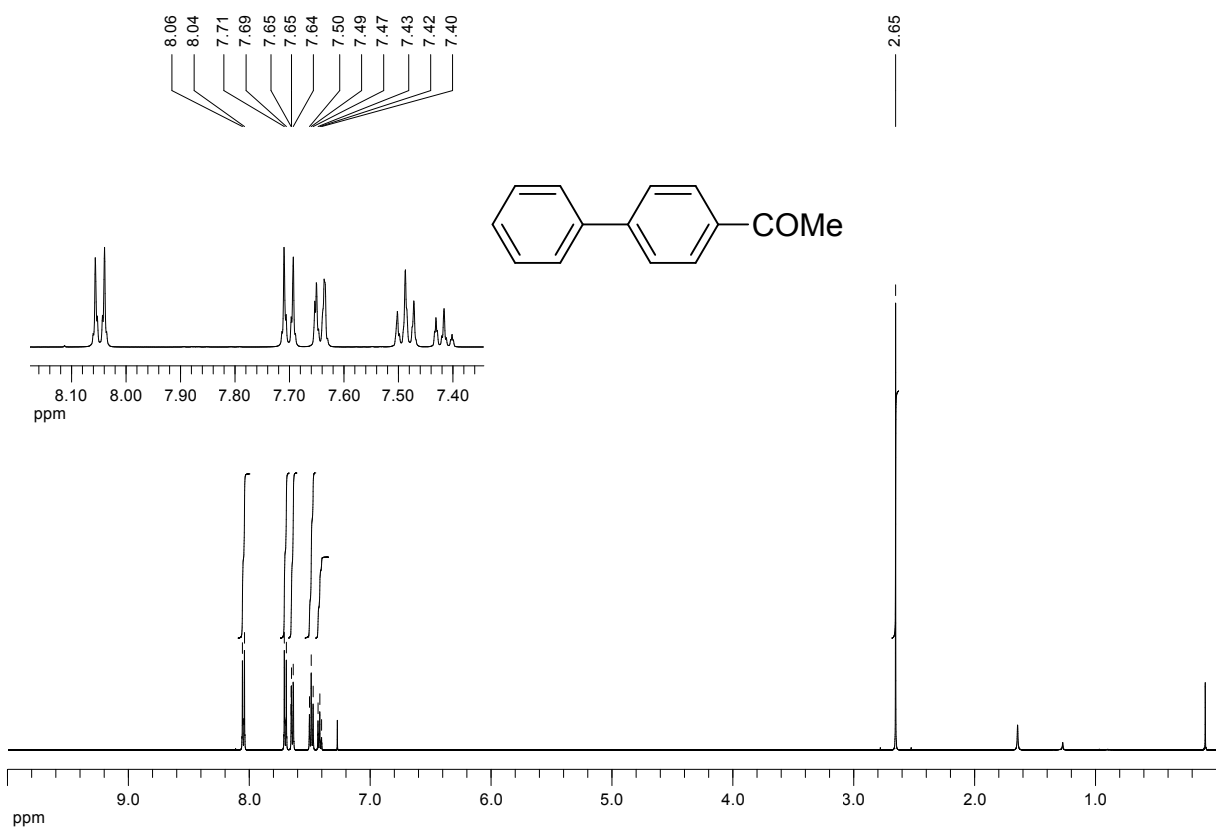
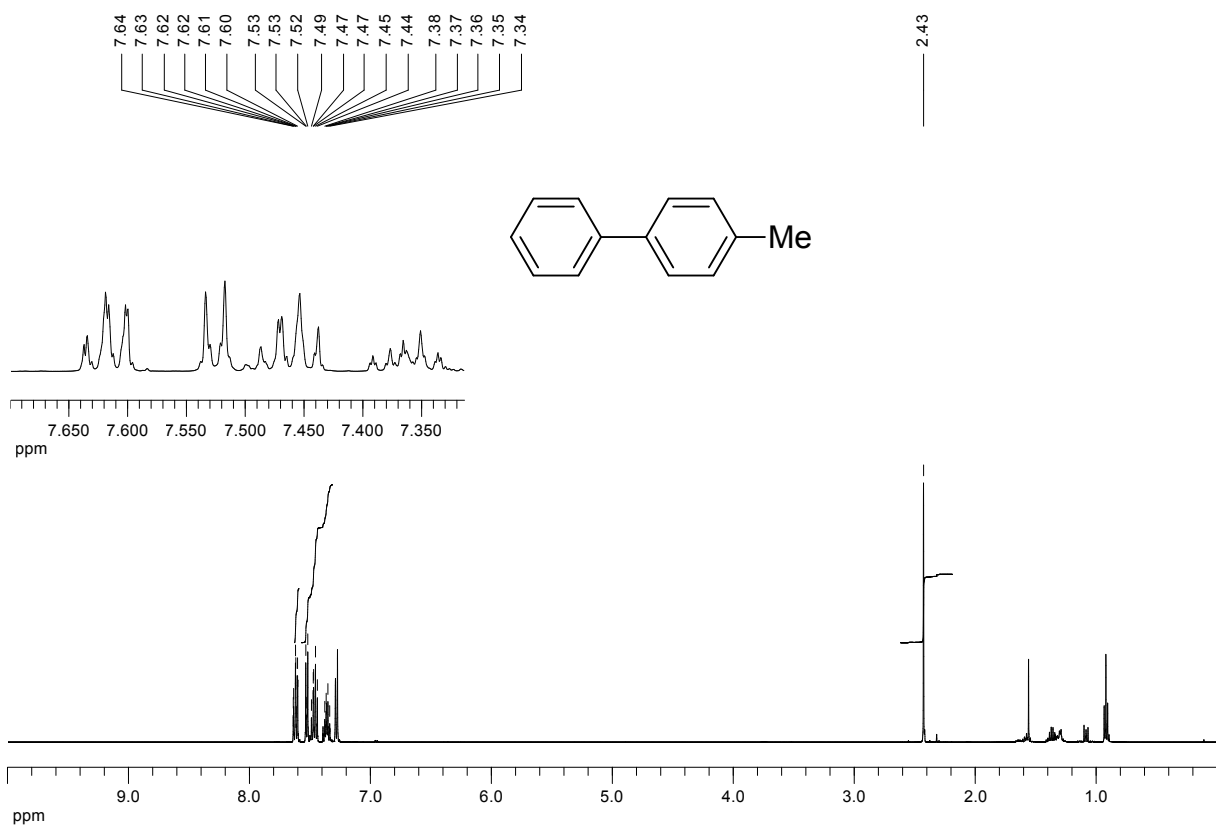
White solid; ^1H NMR (500 MHz, Chloroform-*d*) δ 7.74-7.68 (m, 4H), 7.61-7.60 (m, 2H), 7.50 (dd, J = 8.2, 6.7 Hz, 2H), 7.46-7.42 (m, 1H).

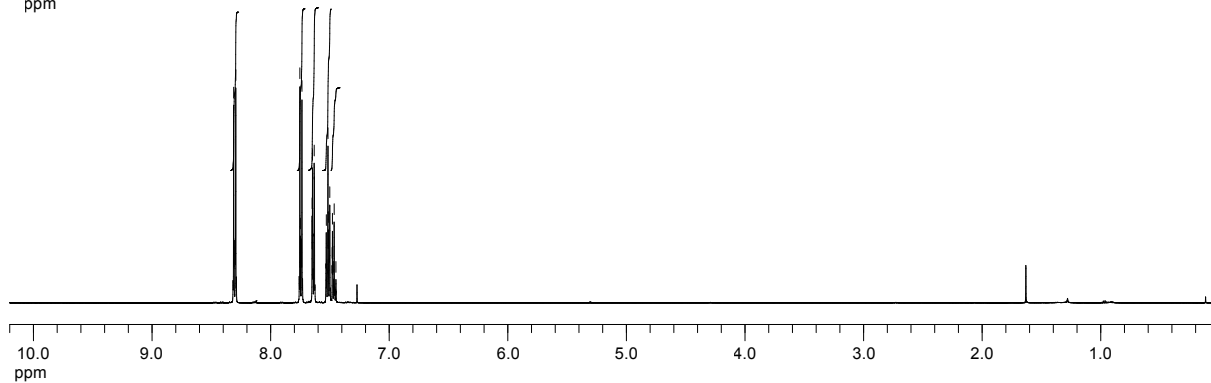
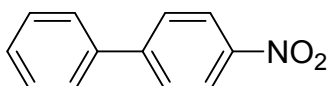
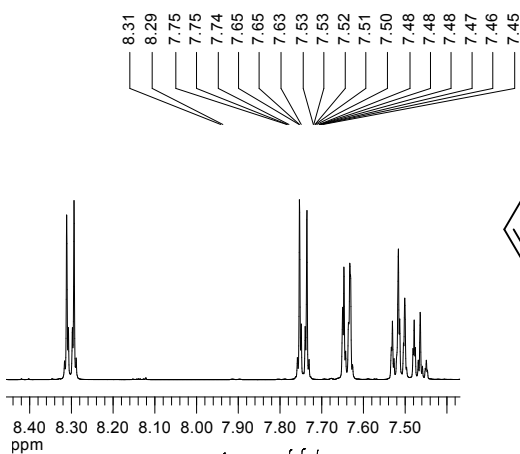
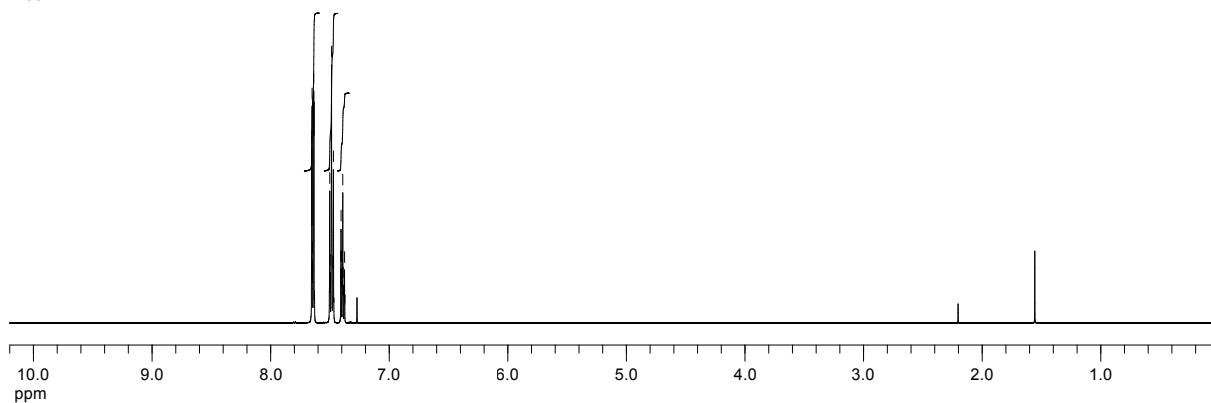
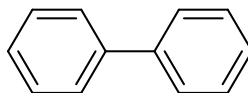
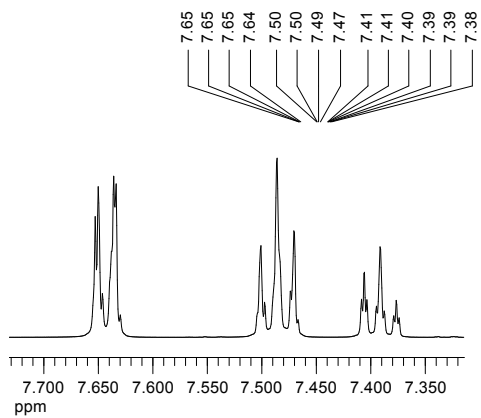
1-([1,1'-biphenyl]-4-yl)ethanone ⁹

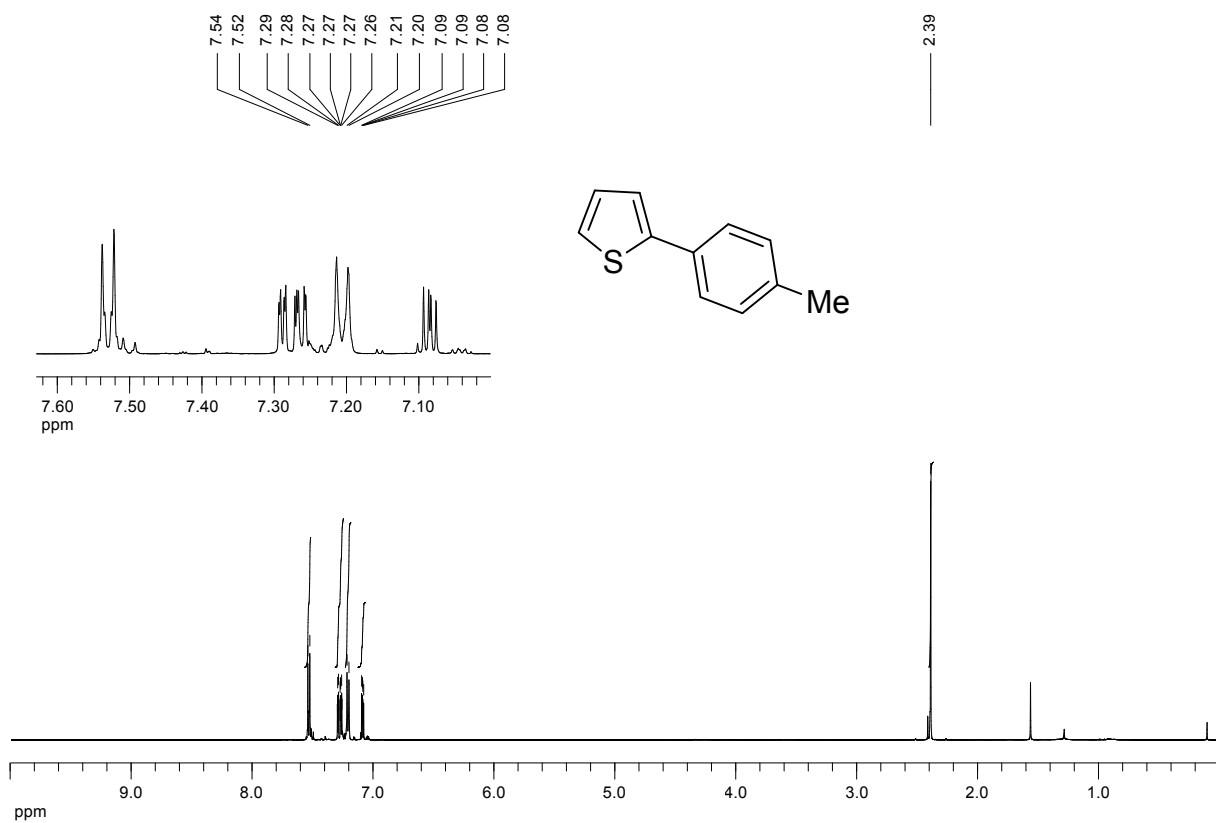
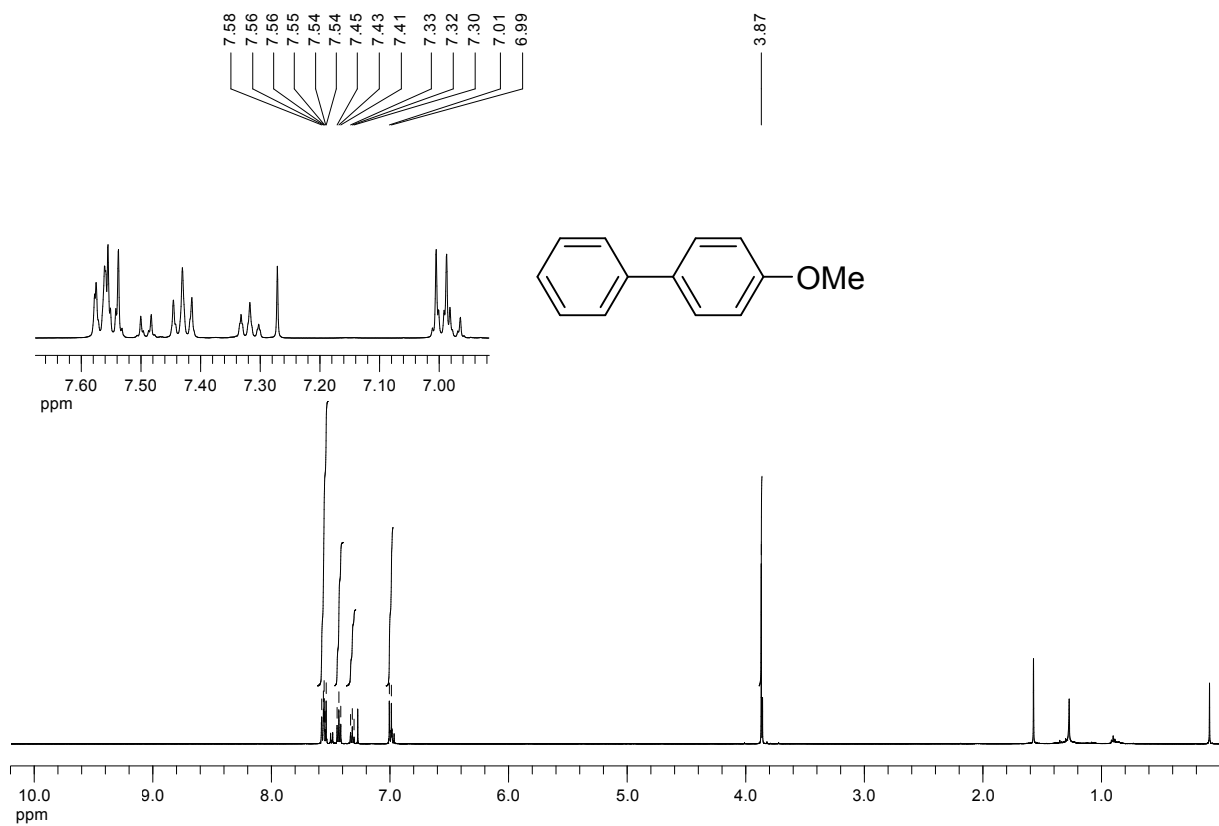


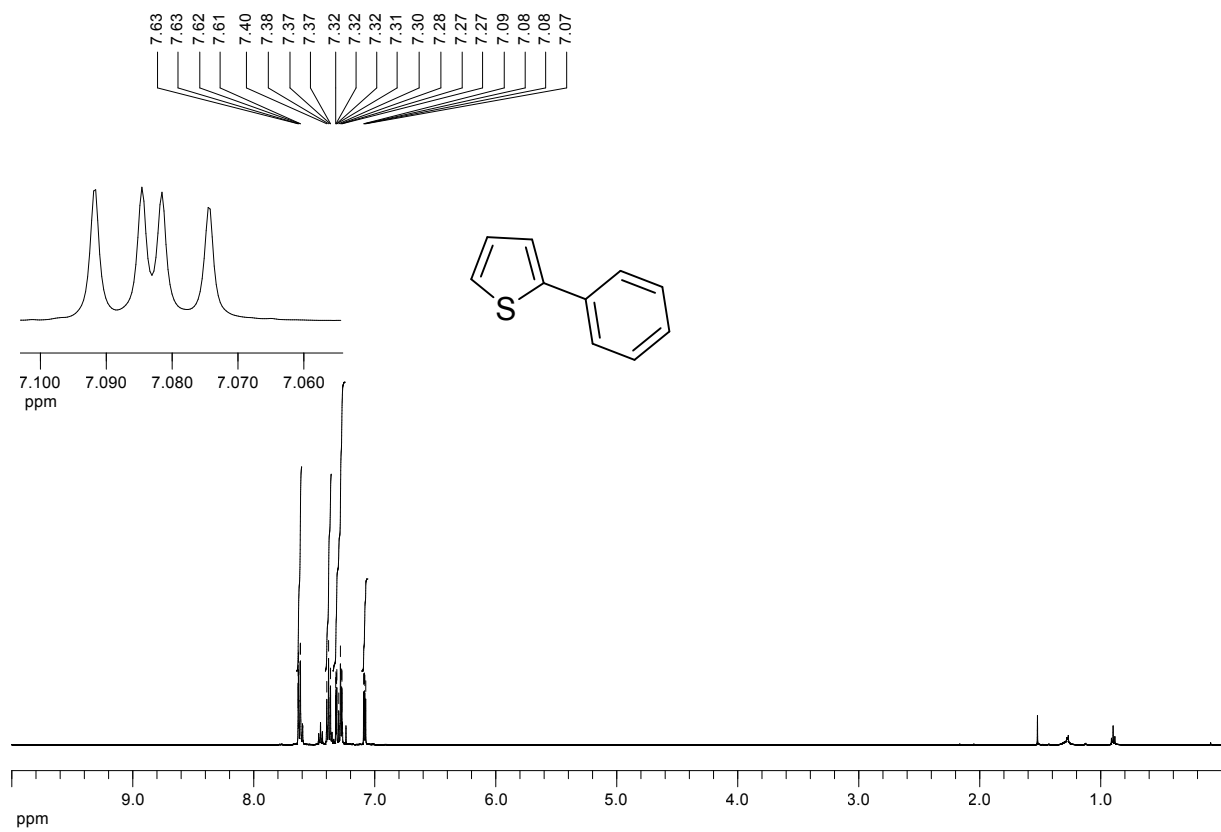
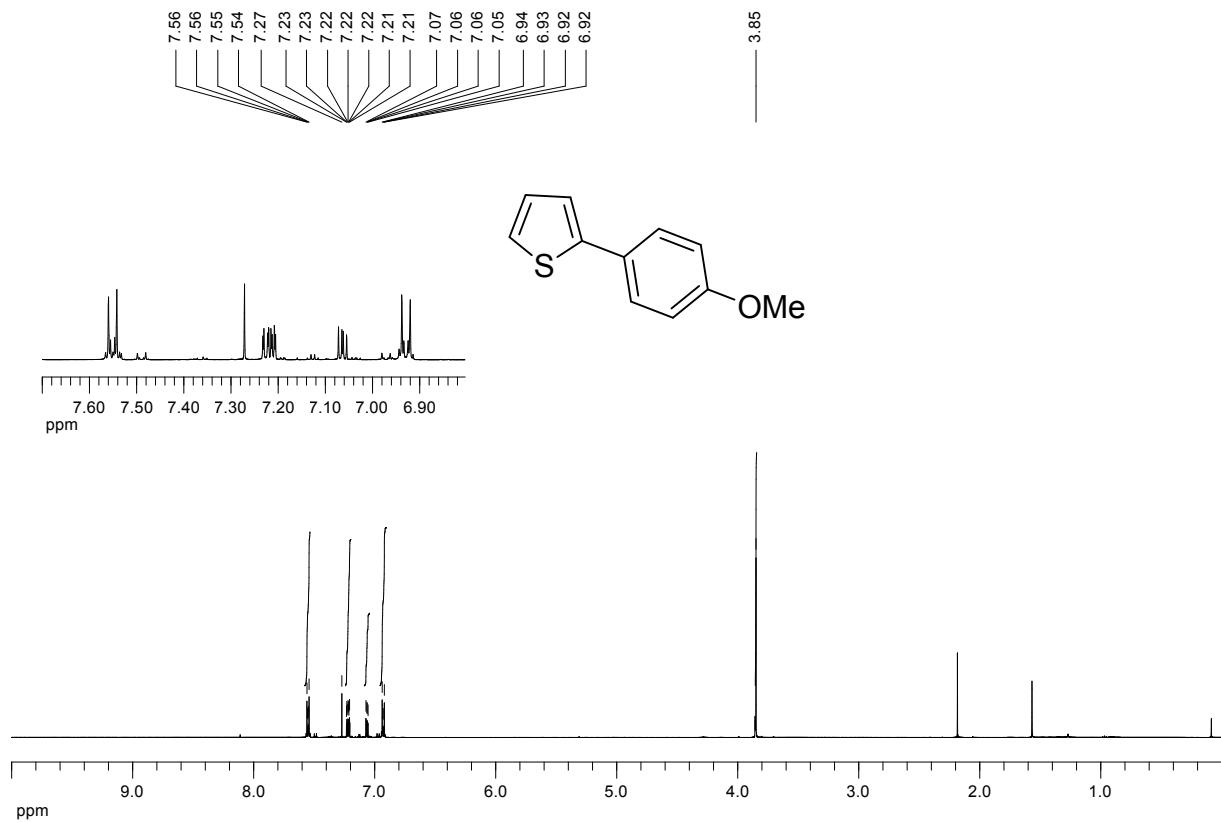
White solid; ^1H NMR (500 MHz, Chloroform-*d*) δ 8.05 (d, J = 8.4 Hz, 2H), 7.70 (d, J = 8.4 Hz, 2H), 7.66-7.62 (m, J = 8.4 Hz, 2H), 7.51-7.46 (m, 2H), 7.44-7.39 (m, 1H), 2.65 (s, 3H).

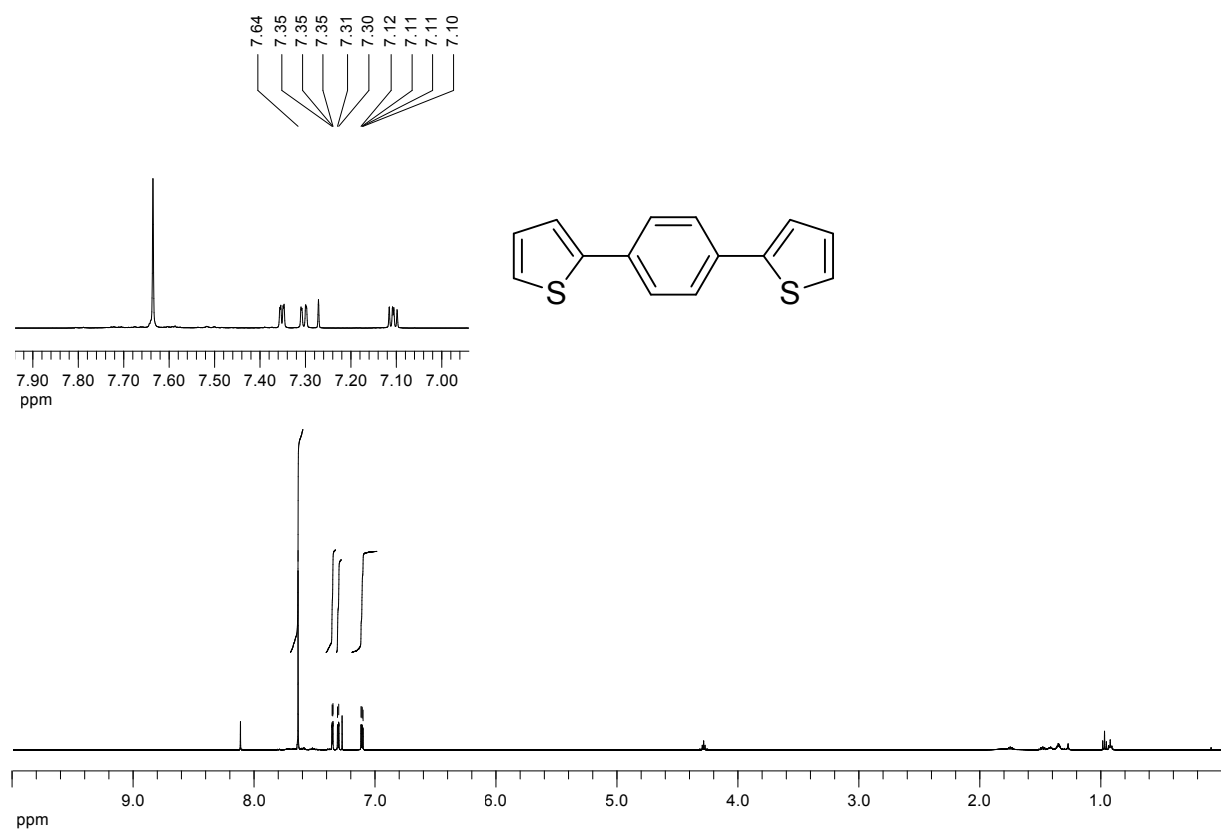
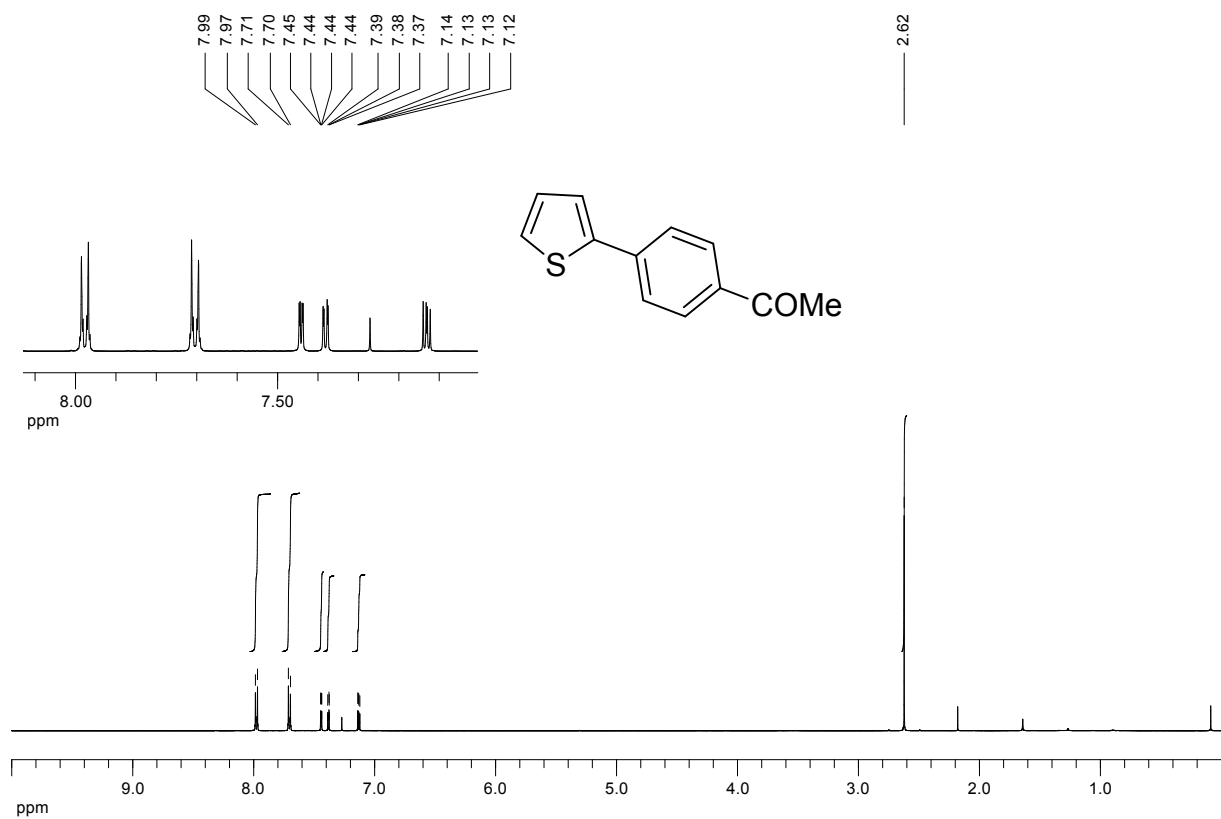












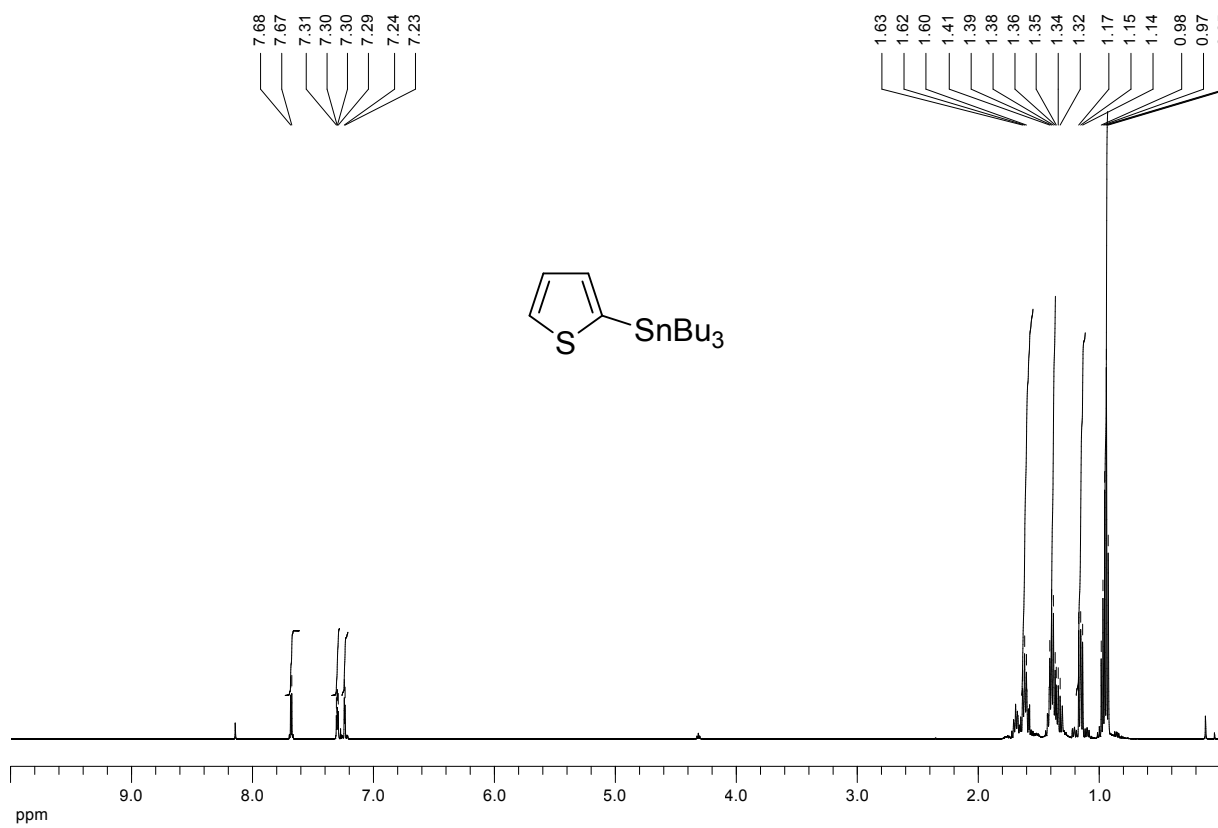
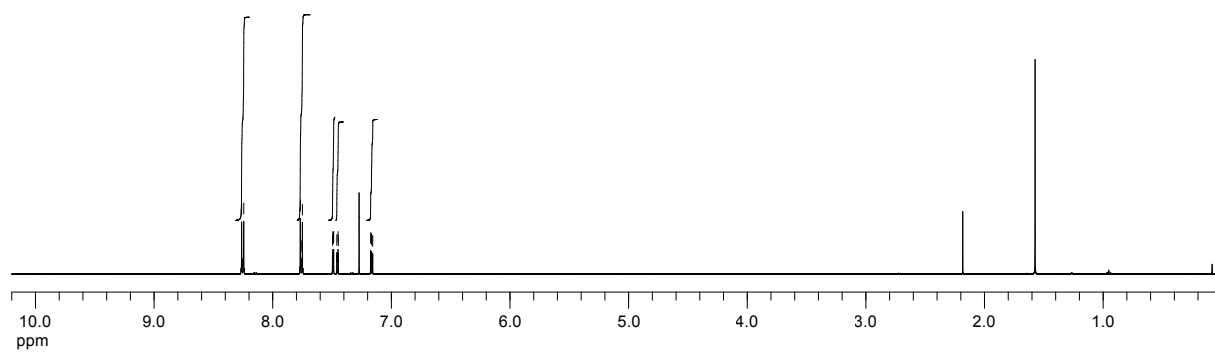
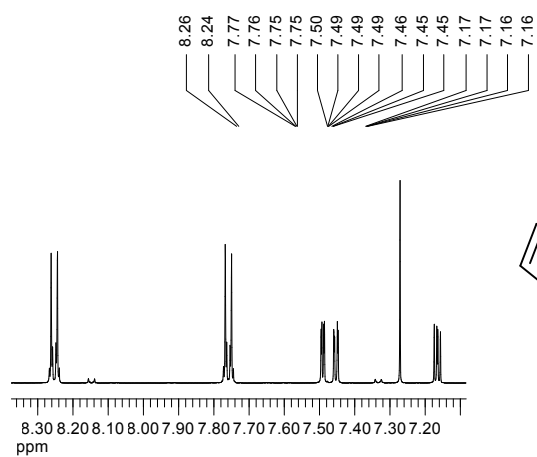


Table S1. Heterogeneous Pd-based catalysts that have been tested in Stille cross-coupling reactions.

No	Catalyst	Mol (%)	Base	Solvent	Temp (°C)	Time (h)	Yield (%)			Recycles	Ref.
							PhI	PhBr	PhCl		
1	Pd _{np-2}	1.0	K ₂ CO ₃	DMF	110	7-16	-	80-94	-	6	10
2	NAP-Mg-Pd	1-3	NaOAc	NMP	100	2-18	90-95	88	70-92	4	11
3	Pd-EGCG-CF.	1-2.5	-	DMF	60-80	5-24	84-88	-	-	5	12
4	Fe ₃ O ₄ /SiO ₂ -PAP-Pd	0.75	Na ₂ CO ₃	PEG	80	0.5-4	40-96	89-98	51	5	13
5	MCM-41-2P-Pd	0.5-2	-	DMF	25-100	4-24	4-23	-	-	10	14
6	PdNPs@H ₂ P-CMP	0.5	-	H ₂ O	80	5	-	-	88-96	10	15
7	PS-dpp-Pd(0)	1.0	-	DMF	65-120	7-14	83-88	-	71-75	5	16
8	Pd/CaCO ₃	0.5-5	K ₂ CO ₃	EtOH/ H ₂ O	80	24	75-93	66-98	60	3	17
9	Fe ₃ O ₄ @PTA-Pd	5	DMAP	PEG	100	0.5-2.5	78-97	93-98	79	2	18
10	PdNPs/C@Fe ₃ O ₄	0.73-1.5	K ₂ CO ₃	DMF	100	12-24	82-99	45-99	68-82	5	8
11	Pd(0)-SMT-MCM-41	1.2	K ₂ CO ₃	PEG	80	20-165min	95-97	94-97	95-96	-	19
12	SBA-15-EDTA-Pd	0.87	-	DMSO	120	4-20	80-94	85-90	76	-	20
13	SBA-Pd	0.157	Na ₂ CO ₃	PEG	80	10-360min	85-96	68-95	71	6	21
14	Boehmite@tryptophan-Pd	0.075	Na ₂ CO ₃	EtOH	70	50-210min	87-96	77-94	78	5	22
15	Pd(0)-guanidine@MCM41		KOH	DMSO	80	20-185min	65-97	65-93	-	10	23
16	Fe ₃ O ₄ @MCM-41@Pd-SPATB	0.94	K ₂ CO ₃	PEG	80	40-140min	88-94	84-93	-	5	24

References

- 1 T. Akiyama, T. Taniguchi, N. Saito, R. Doi, T. Honma, Y. Tamenori, Y. Ohki, N. Takahashi, H. Fujioka, Y. Sato and M. Arisawa, *Green Chem.*, 2017, **19**, 3357–3369.
- 2 C. Istanbuluoglu, S. Göker, G. Hizalan, S. O. Hacıoglu, Y. A. Udum, E. D. Yildiz, A. Cirpan and L. Toppare, *New J. Chem.*, 2015, **39**, 6623–6630.
- 3 P. S. Gribanov, Y. D. Golenko, M. A. Topchiy, L. I. Minaeva, A. F. Asachenko and M. S. Nechaev, *Chem. Eur. J.*, 2018, **2018**, 120–125.
- 4 M. E. Budén, J. F. Guastavino and R. A. Rossi, *Org. Lett.*, 2013, **15**, 1174–1177.
- 5 D. P. Hari, P. Schroll and B. König, *J. Am. Chem. Soc.*, 2012, **134**, 2958–2961.
- 6 Y. Yabe, T. Maegawa, Y. Monguchi and H. Sajiki, *Tetrahedron*, 2010, **66**, 8654–8660.
- 7 D. Saha, R. Sen, T. Maity and S. Koner, *Langmuir*, 2013, **29**, 3140–3151.
- 8 B. S. Kumar, R. Anbarasan, A. J. Amali and K. Pitchumani, *Tetrahedron Lett.*, 2017, **58**, 3276–3282.
- 9 J. H. Li, B. X. Tang, L. M. Tao, Y. X. Xie, Y. Liang and M. B. Zhang, *J. Org. Chem.*, 2006, **71**, 7488–7490.
- 10 J. F. Cívicos, D. A. Alonso and C. Nájera, *Adv. Synth. Catal.*, 2011, **353**, 1683–1687.
- 11 M. L. Kantam, S. Roy, M. Roy, B. Sreedhar and B. M. Choudary, *Adv. Synth. Catal.*, 2005, **347**, 2002–2008.
- 12 B. Mohammadi-Aghdam, S. Bahari and R. Molaei, *J. Chem. Sci.*, 2013, **125**, 813–817.
- 13 A. Ghorbani-Choghamarani and M. Norouzi, *Appl. Organomet. Chem.*, 2016, **30**, 140–147.
- 14 W. Hao, J. Sha, S. Sheng and M. Cai, *J. Mol. Catal. A: Chem.*, 2009, **298**, 94–98.
- 15 N. Huang, Y. Xu and D. Jiang, *Sci. Rep.*
- 16 S. Bahari, B. Mohammadi-Aghdam, R. Molaei and Z. Gharibi, *Can. J. Chem.*, 2012, **90**, 784–789.
- 17 A. v. Coelho, A. L. F. de Souza, P. G. de Lima, J. L. Wardell and O. A. C. Antunes, *Tetrahedron Lett.*, 2007, **48**, 7671–7674.

- 18 A. Ghorbani-Choghamarani and M. Norouzi, *New J. Chem.*, 2016, **40**, 6299–6307.
- 19 N. Noori, M. Nikoorazm and A. Ghorbani-Choghamarani, *J. Porous Mater.*, 2016, **23**, 1467–1481.
- 20 J. Rathod, P. Sharma, P. Pandey, A. P. Singh and P. Kumar, *J. Porous Mater.*, 2017, **24**, 837–846.
- 21 A. Ghorbani-Choghamarani, A. A. Derakhshan, M. Hajjami and L. Rajabi, *Catal. Lett.*, 2017, **147**, 110–127.
- 22 A. Ghorbani-Choghamarani, M. Mohammadi, R. H. E. Hudson and T. Tamoradi, *Appl. Organomet. Chem.*, 2019, **33**, 1–11.
- 23 H. Filian, A. Ghorbani-Choghamarani and E. Tahanpesar, *J. Porous Mater.*, 2019, **26**, 1091–1101.
- 24 M. Nikoorazm, F. Ghorbani, A. Ghorbani-Choghamarani and Z. Erfani, *Appl. Organomet. Chem.*, 2018, **32**, 1–13.