

Electronic Supplementary Material (ESI) for New Journal of Chemistry.

This journal is © The Royal Society of Chemistry and the Centre National de la Recherche Scientifique 2020

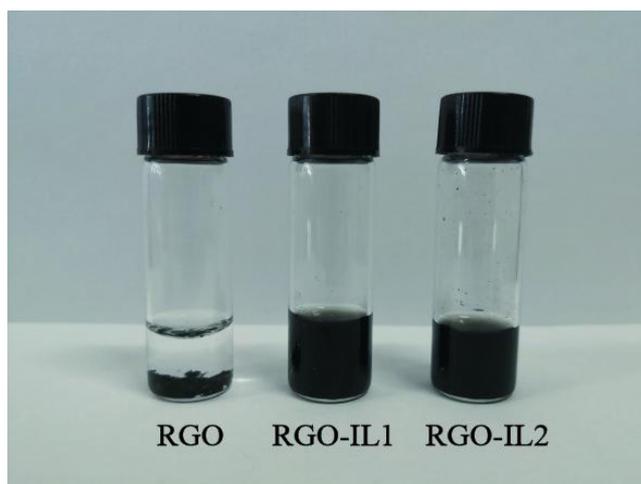
## An ionic liquid-modified reduced graphene oxide electrode material with favourable electrochemical properties

Chang Dong, Yijia Yu, Xiaoling Zhang, Liyan Huang\*, Ying Wu, Jun Li and Zhengping Liu

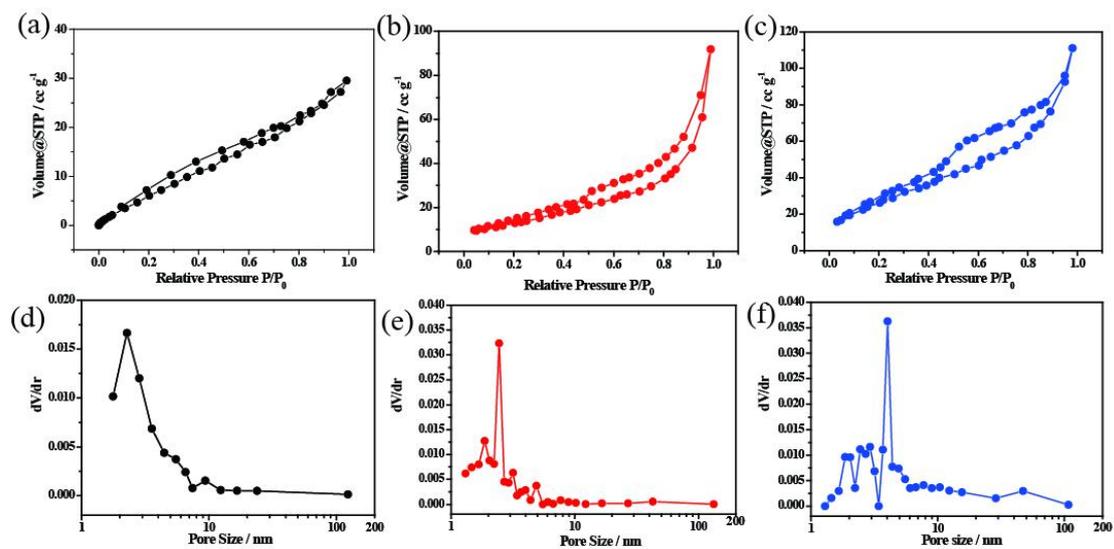
**Address:** Beijing Key Laboratory of Energy Conversion and Storage Materials, College of Chemistry, Beijing Normal University, Beijing 100875, P. R. China

**E-mail:** [hly@bnu.edu.cn](mailto:hly@bnu.edu.cn)

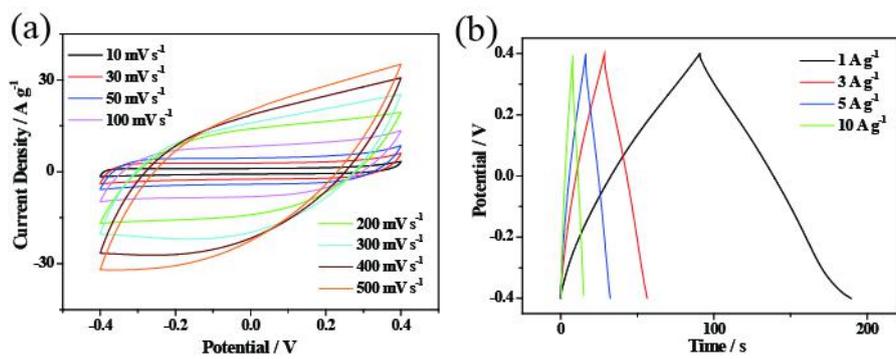
**This updated version of the Electronic Supplementary Information replaces the original version first published on-line on 31 Mar 2020. There were some minor errors in the title, Table S1 and Table S2 which have now been corrected in this updated version.**



**Figure S1.** Photograph of the water dispersions of RGO, RGO-IL1 and RGO-IL2.



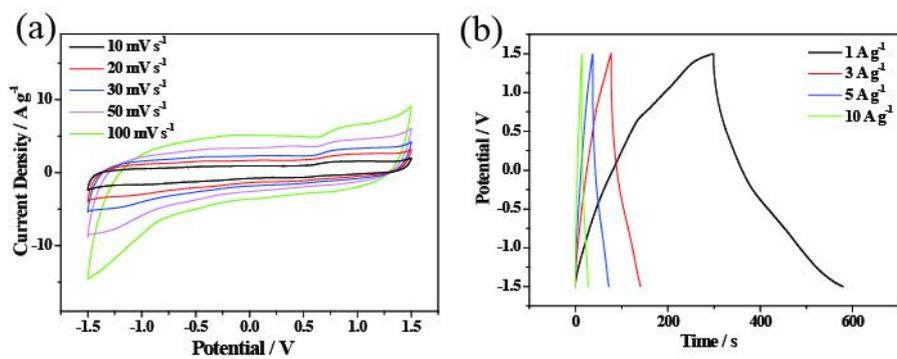
**Figure S2.** Nitrogen adsorption isotherms of (a) RGO, (b) RGO-IL1 and (c) RGO-IL2. BJH desorption pore size distribution curves of (d) RGO, (e) RGO-IL1 and (f) RGO-IL2.



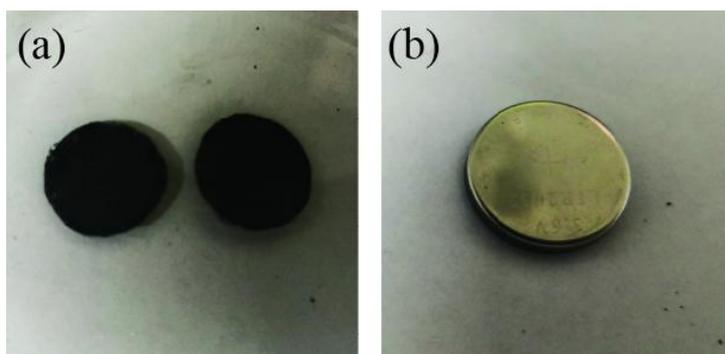
**Figure S3.** (a) CV curves of RGO at different scanning rates in the potential range from  $-0.4$  to  $0.4$  V in  $1.0$  M  $\text{H}_2\text{SO}_4$ . (b) GCD curves of RGO at current density from  $1$  to  $10$   $\text{A g}^{-1}$  in  $1.0$  M  $\text{H}_2\text{SO}_4$ .

**Table S1.** Comparison of RGO-IL1 and RGO-IL2 with other graphene electrodes in terms of specific capacitance and measurement conditions.

Electrodes	Electrolyte	Current density ( $A g^{-1}$ )	Specific capacitance ( $F g^{-1}$ )	Ref.
OHCF	1M TEABF <sub>4</sub> /AN	1	188	44
CNDHS	1M TEABF <sub>4</sub> /AN	1	183	45
The crumpled graphene balls	5M KOH	0.1	181	46
Cl-RGO	1M H <sub>2</sub> SO <sub>4</sub>	1	178	47
Ionothermal-RGO	6M KOH	0.2	155	48
MCNM	1M TEABF <sub>4</sub> /AN	1	149	49
CNT/graphene	PVA-H <sub>3</sub> PO <sub>4</sub>	1	138	50
RGO	1M H <sub>2</sub> SO <sub>4</sub>	1	114	This work
RGO-IL1	1M H <sub>2</sub> SO <sub>4</sub>	1	173	This work
RGO-IL2	1M H <sub>2</sub> SO <sub>4</sub>	1	193	This work



**Figure S4.** (a) CV curves of RGO at different scanning rates in the potential range from  $-1.5$  to  $1.5$  V in [Bmim]PF<sub>6</sub>/AN (m:m = 1:1). (b) GCD curves of RGO at current density from 1 to  $10 \text{ A g}^{-1}$  in [Bmim]PF<sub>6</sub>/AN (m:m = 1:1).



**Figure S5.** Photographs of (a) the completed RGO-IL electrode slices and (b) the assembled coin-type supercapacitor based on RGO-IL.

**Table S2.** Comparison of RGO-IL1 and RGO-IL2 with other ionic liquid modified RGO supercapacitor in terms of potential window and energy density.

Electrodes	Electrolyte	Potential window (V)	Energy density ( Wh kg <sup>-1</sup> )	Ref.
PIL:RG-O	1M EMIM-NTf <sub>2</sub>	3.5	6.50	30
ILG	1M [Bmim]PF <sub>6</sub> /AN	2.2	25.30	32
PIL:RGO	[MPPy][TFSI]	2.0	40.00	34
rGO-Im-IL	PVA-NaCl	2.0	36.67	54
IL-RGO	1M EMIMBF <sub>4</sub>	4.0	49.00	61
FGO-178/POAP	0.1M HClO <sub>4</sub>	1.1	48.33	62
RGO-IL1	1M [Bmim]PF <sub>6</sub> /AN	3.0	32.02	This work
RGO-IL2	1M [Bmim]PF <sub>6</sub> /AN	3.0	50.19	This work