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Electronic Supplementary Information

Experimental and theoretical investigation on MoS₂/MXene heterostructure as an efficient electrocatalyst for PH-universal hydrogen evolution reaction

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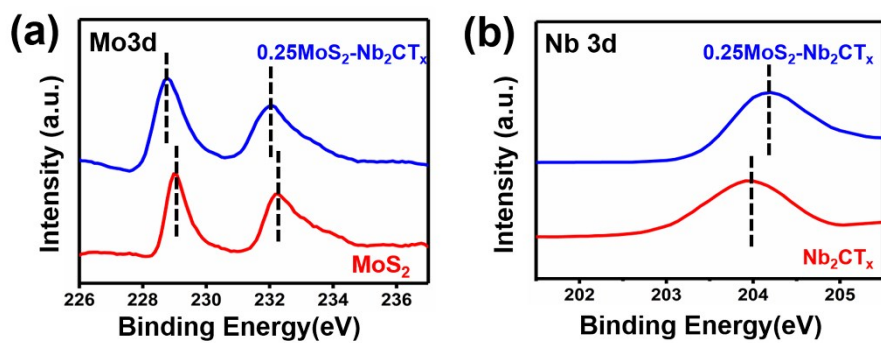


Fig. S1 (a) Mo 3d XPS spectra of 0.25 MoS₂-Nb₂CT_x heterostructure and MoS₂. (b) Nb 3d XPS spectra of 0.25 MoS₂-Nb₂CT_x heterostructure and Nb₂CT_x.

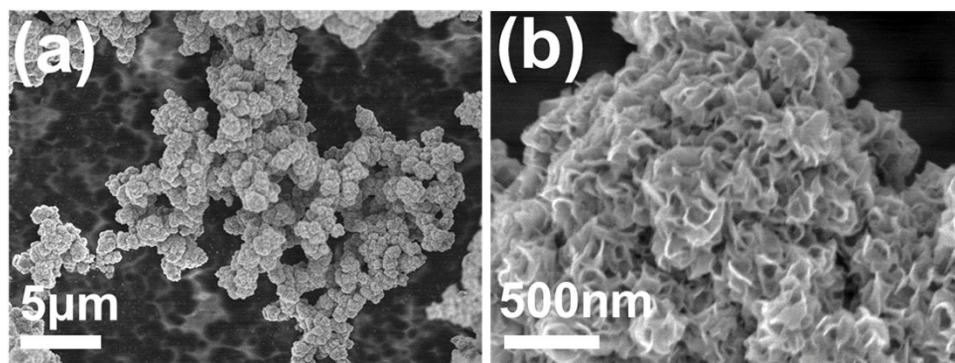


Fig. S2 SEM images of MoS₂.

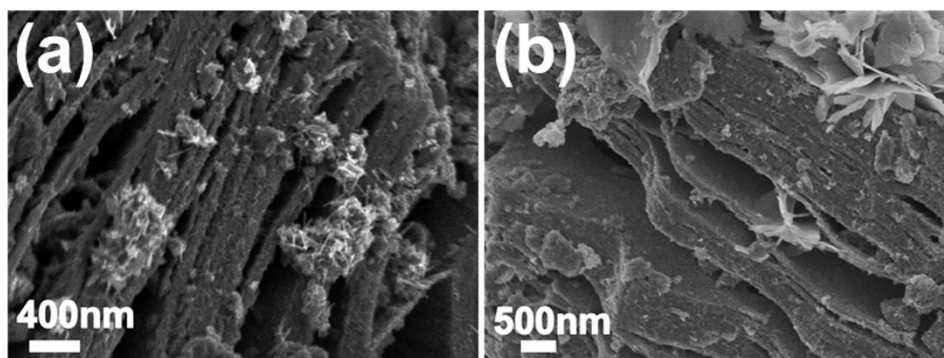


Fig. S3 SEM images of 0.15 MoS₂-Nb₂CT_x and 0.35 MoS₂-Nb₂CT_x heterostructures: (a) SEM image of 0.15 MoS₂-Nb₂CT_x heterostructure. (b) SEM image of 0.35 MoS₂-Nb₂CT_x heterostructure.

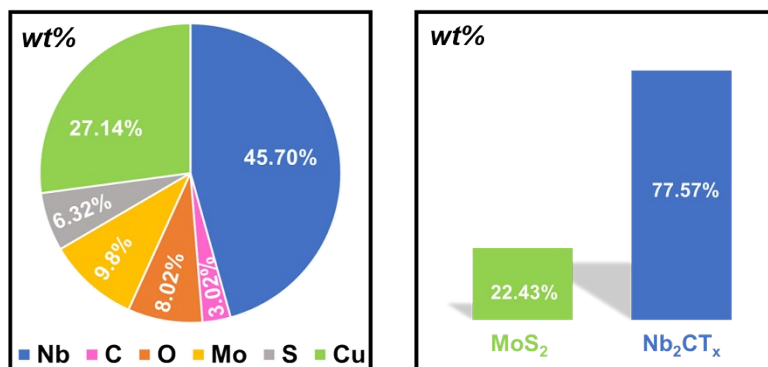


Fig. S4 Mass ratio diagram of elements and chemical composition in 0.25 MoS₂-Nb₂CT_x sample. Because the proportion of Al element is only 0.12%, it is not exhibited in the figure.

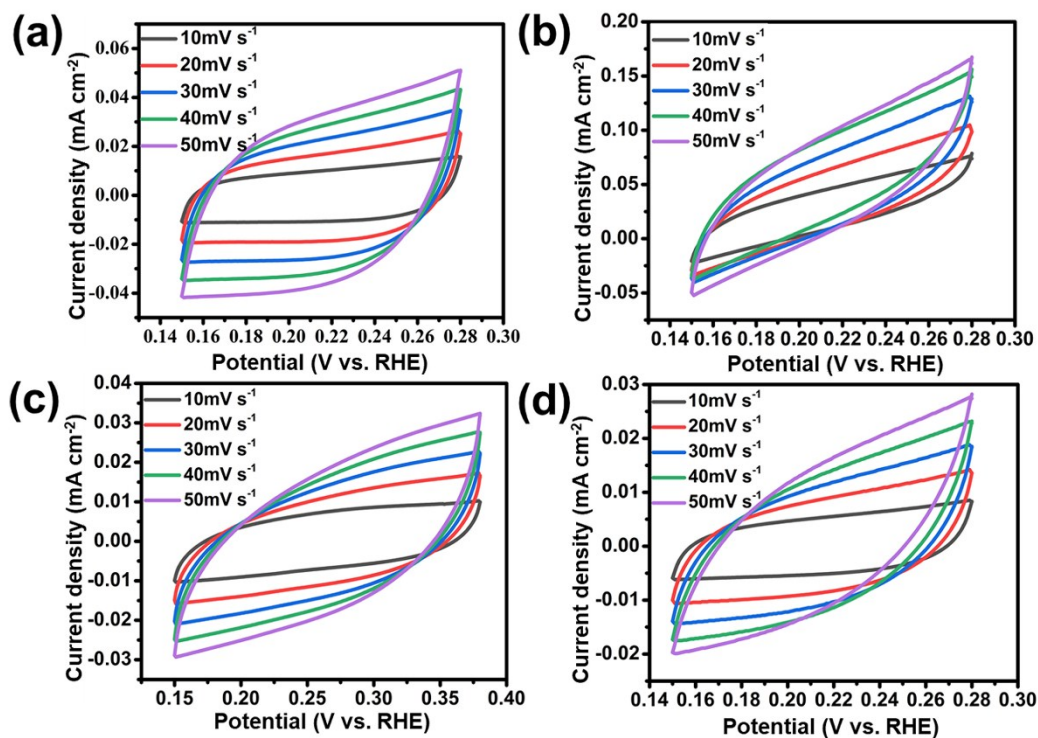


Fig. S5 CV Cyclic voltammetry curves of MoS₂, Nb₂CT_x, 0.15 MoS₂-Nb₂CT_x, and 0.35 MoS₂-Nb₂CT_x: (a) CV Cyclic voltammetry curve of MoS₂, (b) CV Cyclic voltammetry curve of Nb₂CT_x, (c) CV Cyclic voltammetry curve of 0.15 MoS₂-Nb₂CT_x, (d) CV Cyclic voltammetry curve of 0.35 MoS₂-Nb₂CT_x.

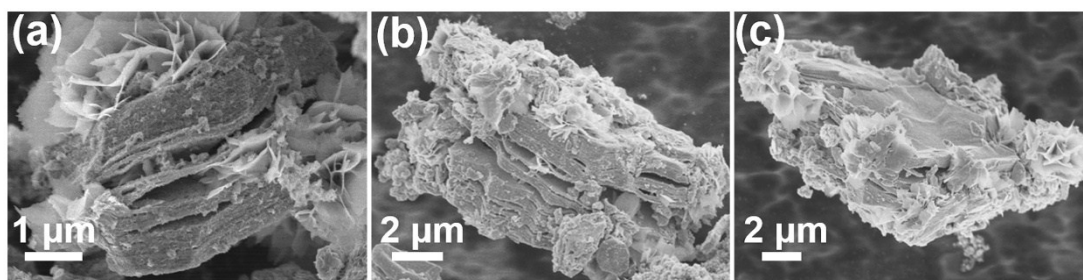


Fig. S6 SEM images of 0.25 MoS₂-Nb₂CT_x before and after the cycles.

(a) SEM image of 0.25 MoS₂-Nb₂CT_x before the cycles. (b) SEM image of the composite after the potential sweeps for 3000 cycles in acidic medium. (c) SEM image of the composite after the potential sweeps for 3000 cycles in alkaline medium.

Table S1. Comparison of HER performances for MoS₂-Nb₂CT_x with related MoS₂-based catalysts.

| Catalyst | Overpotential (mV) | Tafel slope (mV dec ⁻¹) | R _{ct} | Electrolyte | Reference |
|---|--------------------|-------------------------------------|-----------------|--------------------------------------|-----------|
| MoS ₂ -Nb ₂ CT _x | 127 | 56.2 | 63.1 | 0.5 M H ₂ SO ₄ | This work |
| MoS ₂ -Ti ₃ C ₂ T _x | 152 | 70 | - | 0.5 M H ₂ SO ₄ | 1 |
| MoS ₂ -GF | 157 | 93 | 47.3 | 0.5 M H ₂ SO ₄ | 2 |
| Ni@NC/MoS ₂ -P | 325 | 118.2 | 28.6 | 0.5 M H ₂ SO ₄ | 3 |
| Cu-MoS ₂ /rGo | 244 | 127 | - | 0.5 M H ₂ SO ₄ | 4 |
| CoP/CN @ MoS ₂ | 144 | 69 | - | 0.5 M H ₂ SO ₄ | 5 |
| MoS ₂ -Nb ₂ CT _x | 141 | 93.4 | 231.4 | 1 M KOH | This work |
| (NiFe) _x -MoS ₂ | 285 | 94 | - | 1 M KOH | 6 |
| MoS ₂ -Co(OH) ₂ | 179 | 62 | - | 1 M KOH | 7 |
| MoS ₂ @3DC | 252 | 102.8 | 316.3 | 1 M KOH | 8 |
| Fe-MoS ₂ | 163 | 181 | - | 1 M KOH | 9 |
| Co-MoS ₂ | 215 | 153 | 175 | 1 M KOH | 10 |
| Ni-P/MoS ₂ | 155 | 108 | 2758 | 1 M KOH | 11 |

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