

Electronic Supplementary Information (ESI) for

Improved Ti species distribution and hierarchical pores in TS-1: Towards regeneration of deactivated TS-1 caused by the corrosion of alkali

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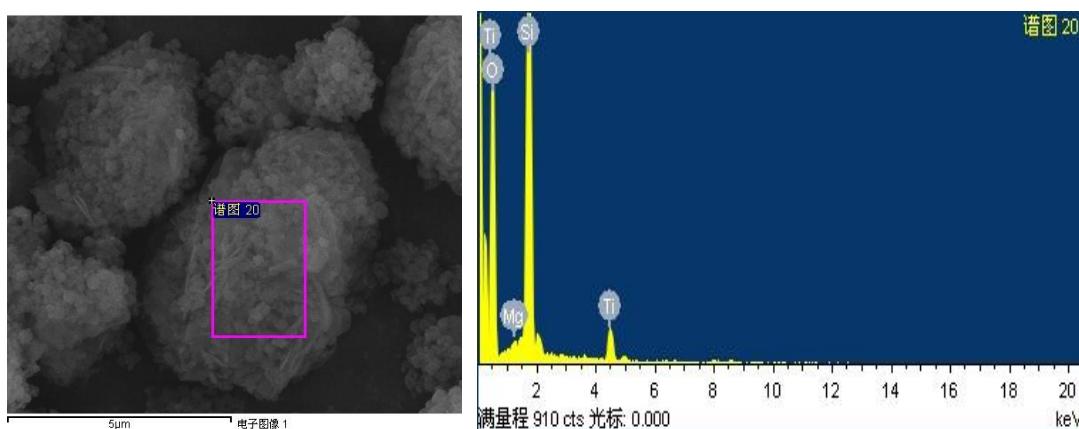


Fig. S1. EDS result of agglomerated particles doped with slice-shaped crystals in D-TS-1

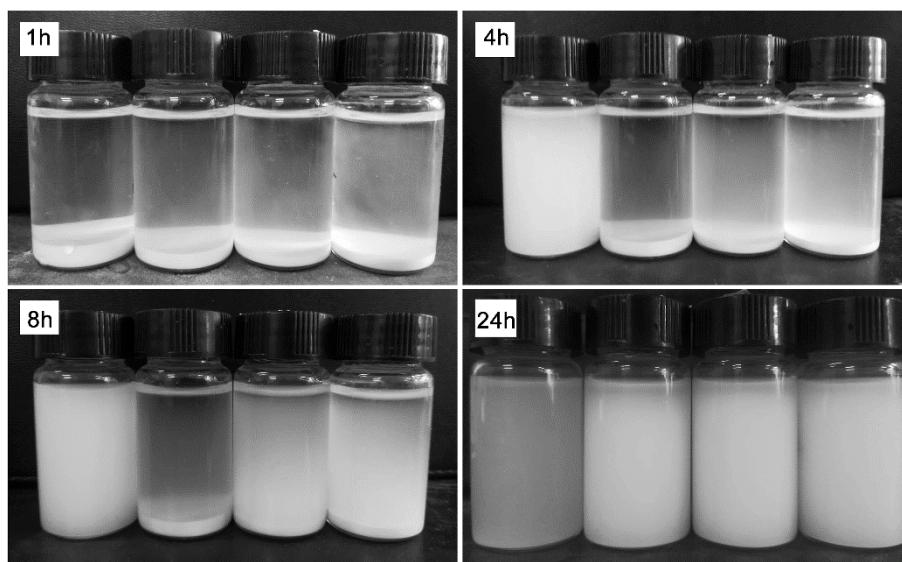


Fig. S2. Optical images of sedimentation experiment of F-TS-1, D-TS-1, R-TS-1-F and R-TS-1-S (from left to right) treated with NaOH (1 wt%) solution for different time.

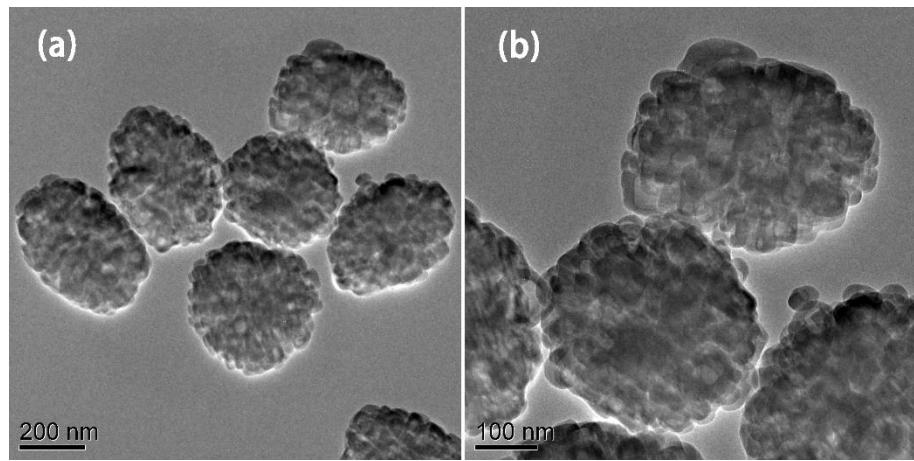


Fig. S3. TEM images of hollow TS-1(H-TS-1) (a and b)

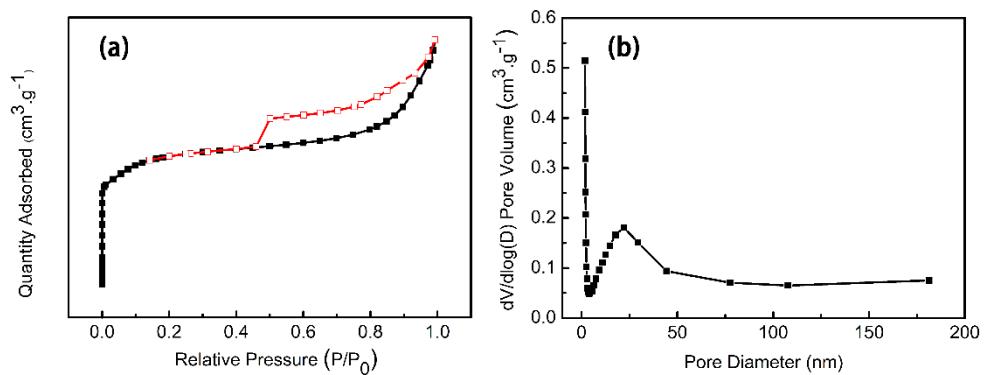


Fig. S4. N₂ adsorption–desorption isotherms (a) and pore size distribution (b) of hollow TS-1(H-TS-1).

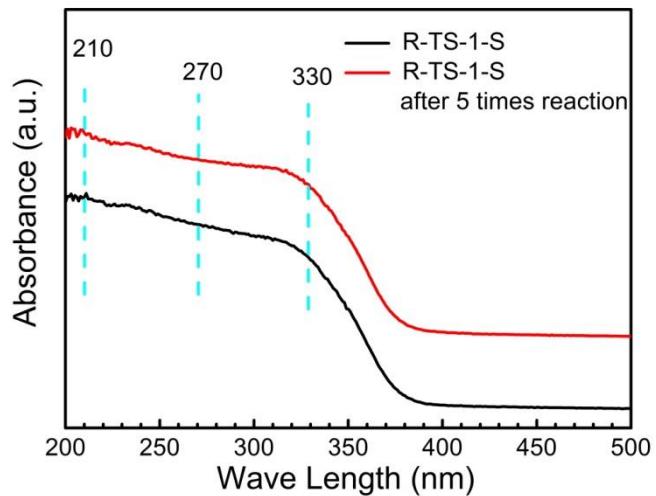


Fig. S5. Uv-vis spectra of R-TS-1-S in recycling test

Table S1. The catalytic performance for cyclohexanone ammoximation^a over a mixture of F-TS-1 and amorphous TiO₂

Samples ^b	Amount (mg)	X (%)	Y (%)	S (%)
F-TS-1	250	71.5	70.9	99.2
F-TS-1 + Amor	250+15	69.5	68.7	98.8
F-TS-1 + Amor	250+25	62.5	61.2	98.8
F-TS-1 + Amor	250+35	55.8	55.0	98.6
F-TS-1 + Amor	250+50	51.5	50.4	97.9

^aReaction conditions: t-BuOH (16.8 mL); NH₃ H₂O (70 mmol) added in eight times; H₂O₂ (70mmol) added dropwise within 70 min; cyclohexanone (60 mmol); temp., 353 K; reaction time (min), 75 min.

^b F-TS-1 and a mixture of F-TS-1 and Amor. Amor represent amorphous TiO₂.