

Supporting Information for

Study on Voltage Drop of Vanadium Nitride/Carbon Composites Derived from Pectin/VCl₃ Membrane as Supercapacitors Anode Material

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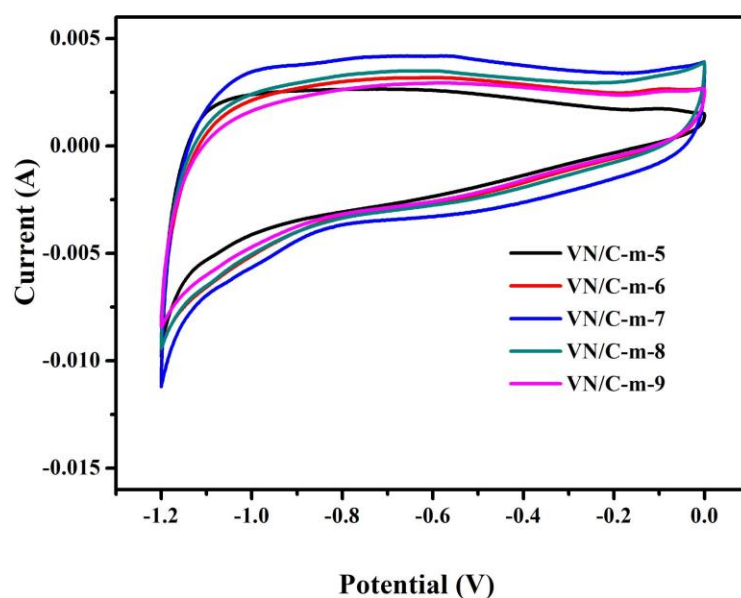
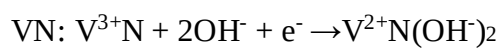
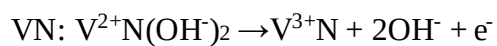


Figure S1 cyclic voltammograms (CVs) curves at the rates of 5 mV/s for VN/C-m (VN/C-m-5, VN/C-m-6, VN/C-m-7, VN/C-m-8, and VN/C-m-9)

The active material here is a composite which consists of VN and active carbon. Therefore, the CV and GCD curves could show the contribution of two energy storage mechanisms. The possible reactions of VN should be the following ^[S1]:



discharge process (0V to -1.2V):



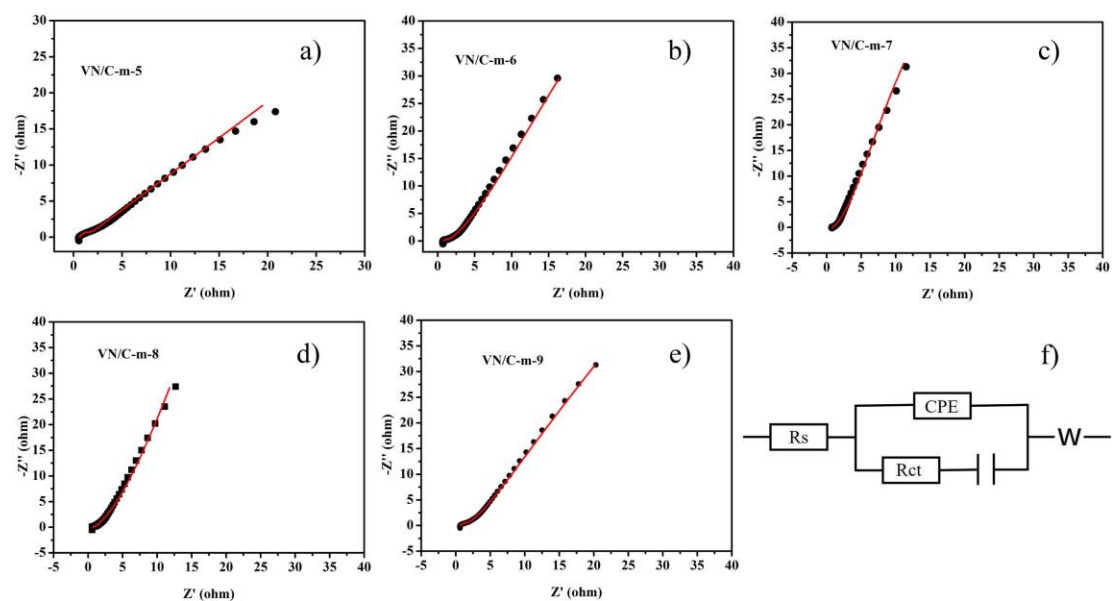


Figure S2 Fitting of Nyquist plots: (a) VN/C-m-5, (b) VN/C-m-6, (c) VN/C-m-7, (d) VN/C-m-8, (e) VN/C-m-9; and (f) circuit diagram

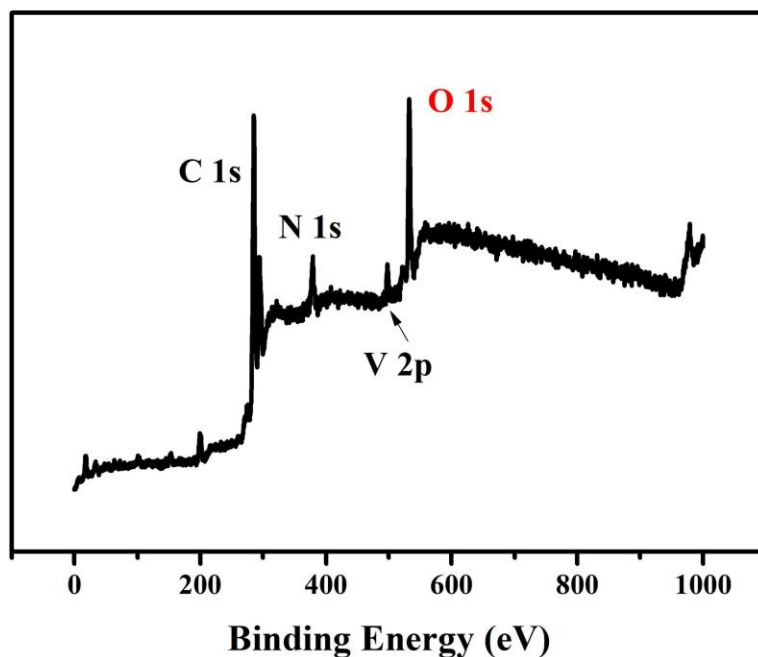


Figure S3 XPS full spectrum for VN/C-m-7.

From our analysis results and previous works ^[S2-S6], we can see that due to the good water retention of pectin, it eventually leads to an extremely increase in the oxygen content of the composite. And from the XPS results, this kind of oxygen is derived from chemical adsorption ^[S7, S8]. Much increased oxygen functional groups could be grafted into carbon surface due to the catalytic oxygen adsorption, which greatly enhances the surface polarity and thereby the wettability toward aqueous electrolyte

^[S9].

Table S1 Fitting value of Nyquist plots

sample	R_s	R_{ct}	W
VN/C-m-5	0.5715	0.5824	0.1406
VN/C-m-6	0.7773	0.5237	0.1941
VN/C-m-7	0.7565	0.1931	0.3869
VN/C-m-8	0.6567	0.2196	0.2502
VN/C-m-9	0.6613	0.6196	0.1579

Table S2 Comparisons of cycle performance with other reported articles.

Materials	Electrolyte	Stability, cycle number, test current	Ref.
VN/C composite	6 M KOH	89.3 %, 2000, 1 A g ⁻¹	Our work
VN-400 °C	KOH, PH≈14	62.5 %, 1000, 50 mV s ⁻¹	S10
VN-400 °C	1 M KOH	45 %, 350, 50 mV s ⁻¹	S11
VN	1 M KOH	20 %, 200, 10 mV s ⁻¹	S12
V ₂ O ₃ -180 min (VN)	2 M KOH	56 %, 1000, 1A g ⁻¹	S13
sputtered VN thin films	1 M KOH	78 %, 2000, 25 mV s ⁻¹	S14
VN/PCNPs	2 M KOH	66 %, 1000, 1A g ⁻¹	S15
VN/CNT	0.5 M Na ₂ SO ₄	85 %, 2000, 200 mV s ⁻¹	S16

Table S3 Energy dispersive spectrometer (EDS) analysis results for VN/C-m-7

Element	wt %	Atom %
C	60.35	68.43
N	3.72	3.62
O	31.42	26.75
V	4.5	1.2
Total	100	100

References:

- S1. Djire, A.; Pande, P.; Deb, A.; Siegel, J. B.; Ajenifujah, O. T.; He, L.; Sleightholme, A. E.; Rasmussen, P. G.; Thompson, L. T., Unveiling the pseudocapacitive charge storage mechanisms of nanostructured vanadium nitrides using in-situ analyses. *Nano Energy* **2019**, *60*, 72-81.
- S2. Liu, Y.; Liu, L.; Tan, Y.; Niu, L.; Kong, L.; Kang, L.; Ran, F., Carbon nanosphere@vanadium nitride electrode materials derived from metal-organic nanospheres self-assembled by NH_4VO_3 , chitosan, and amphiphilic block copolymer. *Electrochimica Acta* **2018**, *262*, 66-73.
- S3. Liu, Y.; Liu, L.; Tan, Y.; Kong, L.; Kang, L.; Ran, F., Well-dispersed vanadium nitride on porous carbon networks derived from block copolymer of PAN-b-PDMC-b-PAN absorbed with ammonium metavanadate for energy storage application. *The Journal of Physical Chemistry C* **2017**, *122* (1), 143-149.
- S4. Yang, Y.; Zhao, L.; Shen, K.; Liu, Y.; Zhao, X.; Wu, Y.; Wang, Y.; Ran, F., Ultra-small vanadium nitride quantum dots embedded in porous carbon as high performance electrode materials for capacitive energy storage. *Journal of Power Sources* **2016**, *333*, 61-71.
- S5. Ran, F.; Wu, Y.; Jiang, M.; Tan, Y.; Liu, Y.; Kong, L.; Kang, L.; Chen, S., Nanocomposites based on hierarchical porous carbon fiber@vanadium nitride nanoparticles as supercapacitor electrodes. *Dalton Trans* **2018**, *47* (12), 4128-4138.
- S6. Wu, Y.; Yang, Y.; Zhao, X.; Tan, Y.; Liu, Y.; Wang, Z.; Ran, F., A novel hierarchical porous 3D structured vanadium nitride/carbon membranes for high-performance supercapacitor negative electrodes. *Nano-micro letters* **2018**, *10* (4), 63.
- S7. Jing, L.; Xu, Z.; Sun, X.; Shang, J.; Cai, W., The surface properties and photocatalytic activities of ZnO ultrafine particles. *Applied Surface Science* **2001**, *180* (3-4), 308-314.
- S8. Yang, S.; Zhu, W.; Jiang, Z.; Chen, Z.; Wang, J., The surface properties and the activities in catalytic wet air oxidation over $\text{CeO}_2\text{-TiO}_2$ catalysts. *Applied Surface Science* **2006**, *252* (24), 8499-8505.
- S9. Sun, F.; Qu, Z.; Gao, J.; Wu, H. B.; Liu, F.; Han, R.; Han, R.; Wang, L.; Pei, T.; Lu, Y., In situ doping boron atoms into porous carbon nanoparticles with increased

- oxygen graft enhances both affinity and durability toward electrolyte for greatly improved supercapacitive performance. *Advanced Functional Materials*, **2018**, 28 (41), 1804190.
- S10. Choi, D.; Blomgren, G. E.; Kumta, P. N., Fast and reversible surface redox reaction in nanocrystalline vanadium nitride supercapacitors. *Advanced Materials* **2006**, 18 (9), 1178-1182.
- S11. Choi, D.; Kumta, P. N., Chemically synthesized nanostructured VN for pseudocapacitor application. *Electrochemical and Solid-State Letters* **2005**, 8 (8), A418-A422.
- S12. Hanumantha, P. J.; Datta, M. K.; Kadakia, K.; Okoli, C.; Patel, P.; Kumta, P. N., Vanadium nitride supercapacitors: Effect of processing parameters on electrochemical charge storage behavior. *Electrochimica Acta* **2016**, 207, 37-47.
- S13. Hou, Z.; Guo, K.; Li, H.; Zhai, T., Facile synthesis and electrochemical properties of nanoflake VN for supercapacitors. *CrystEngComm* **2016**, 18 (17), 3040-3047.
- S14. Achour, A.; Lucio-Porto, R.; Solaymani, S.; Islam, M.; Ahmad, I.; Brousse, T., Reactive sputtering of vanadium nitride thin films as pseudo-capacitor electrodes for high areal capacitance and cyclic stability. *Journal of Materials Science: Materials in Electronics* **2018**, 29 (15), 13125-13131.
- S15. Yang, Y.; Shen, K.; Liu, Y.; Tan, Y.; Zhao, X.; Wu, J.; Niu, X.; Ran, F., Novel hybrid nanoparticles of vanadium nitride/porous carbon as an anode material for symmetrical supercapacitor. *Nano-micro letters* **2017**, 9 (1), 6.
- S16. Ouldhamadouche, N.; Achour, A.; Lucio-Porto, R.; Islam, M.; Solaymani, S.; Arman, A.; Ahmadpourian, A.; Achour, H.; Le Brizoual, L.; Djouadi, M. A.; Brousse, T., Electrodes based on nano-tree-like vanadium nitride and carbon nanotubes for micro-supercapacitors. *Journal of Materials Science & Technology* **2018**, 34 (6), 976-982.