

**Electronic Supplement information**

**UV-light-assisted green preparation of Bi/BiOBr/RGO composites with oxygen vacancies toward enhanced photocatalytic removal of organic dye**

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## 1. Experimental Section

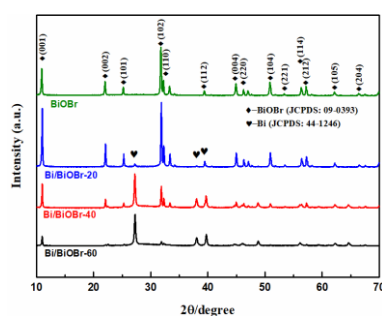
### Preparation of the photocatalyst

Synthesis of BiOBr: 2 mmol of  $\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$  was dissolved in 45 mL of distilled water, and magnetically stirred at 10 min., and also stirred at 10 min. Then, another 45 mL of distilled water containing 2 mmol KBr was added drop wise into the above solution, and the mixed solutions was further stirred for 30 min. The reaction solution was transferred into a 100 mL Teflon-lined autoclave and maintained at  $160^\circ\text{C}$  for 12 h. After cooling to room temperature, the resulted precipitates were collected and washed by deionized water and ethanol, respectively, then finally dried at  $60^\circ\text{C}$  for 12 h. And the sample was named as BiOBr.

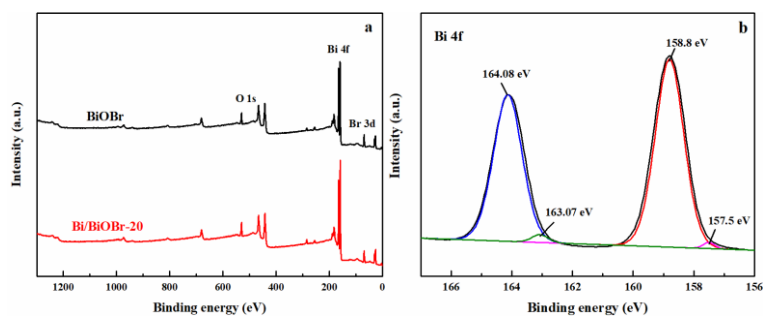
Synthesis of Bi/BiOBr: 0.1 g of BiOBr was added into 50 mL of ethanol and the mixture solution was exposed to UV-light for 20, 40 and 60 min while stirring, respectively. Afterwards, the precipitate was collected and washed successively by deionized water and ethanol, finally dried at  $60^\circ\text{C}$  for 12 h. The resulting samples were marked as Bi/BiOBr-20, Bi/BiOBr-40 and Bi/BiOBr-60.

## 2. Results and discussion

### 2.1. Characterization of Bi/BiOBr composites



**Fig. S1** XRD patterns of BiOBr, Bi/BiOBr-20, Bi/BiOBr-40 and Bi/BiOBr-60 composites.



**Fig. S2** XPS spectra of BiOBr and Bi/BiOBr-20 photocatalyst survey pattern (a); Bi 4f (b).

**Table. S1** The chemical contents (%) of BiOBr/RGO and Bi/BiOBr/RGO.

Name	BiOBr/RGO	Bi/BiOBr/RGO-40
O 1s	23.09	19.53
C 1s	20.74	14.66
Bi 4f	20.30	22.73
Br 3d	35.86	43.08