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## Supporting Information

## Efficient Heterogeneous Catalysis by Pendant Metalloporphyrin-functionalized Polythiophenes for Electrochemical Carbon Dioxide Reduction

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Figure S1 <sup>1</sup>H-NMR spectrum of compound 3 in CDCl<sub>3</sub>.



## Figure S2 <sup>13</sup>C-NMR spectrum of compound 3 in CDCl<sub>3</sub>.

S3





Figure S4 Absorption spectrum of compound 3 in toluene.



**Figure S5** Emission spectrum of compound **3** in toluene ( $\lambda_{ex} = 421$  nm).



Figure S6 <sup>1</sup>H-NMR spectrum of compound 1 in CDCl<sub>3</sub>.

8/ppm





δ/ppm

S8



Figure S8 HR-ESI mass spectrum of compound 1.



Figure S9 Absorption spectrum of compound 1 in toluene.



**Figure S10** Calibration curve of **1** in toluene ( $\lambda_{max} = 422 \text{ nm}$ ).



**Figure S11** Emission spectrum of compound 1 in toluene ( $\lambda_{ex} = 422$  nm).



# Figure S12 <sup>1</sup>H-NMR spectrum of compound Zn-1 in CDCl<sub>3</sub>.

S13





8/ppm



Figure S14 HR-ESI mass spectrum of compound Zn-1.



Figure S15 Absorption spectrum of compound Zn-1 in toluene.







Figure S17 Emission spectrum of compound Zn-1 in toluene ( $\lambda_{ex}$  = 430 nm).

## Mass Spectrum List Report

### Analysis Info

Analysis Name Method Sample Name

30

1042.1838

6633

3.1

15.5

0.1781

OSCUSW20180121001\_1.d Tune\_wide\_POS\_Natee20130403.m CoPTE CoPTE

Acquisition Date 1/24/2018 8:57:29 AM Operator Administrator Instrument micrOTOF 72

**Acquisition Parameter** Set Corrector Fill 50 V Source Type ESI Ion Polarity Positive Set Pulsar Pull Set Pulsar Push 337 V 337 V Capillary Exit Scan Range n/a 200.0 V Scan Begin 50 m/z Hexapole RF 600.0 V Set Reflector 1300 V Scan End 3000 m/z Skimmer 1 70.0 V Set Flight Tube 9000 V 25.0 V Hexapole 1 Set Detector TOF 2295 V Intens. x10<sup>5</sup> +MS, 1.0min #(61) 982.1758 2.0 1.5 1.0 686.1688 0.5 1443.2302 1236.1481 1700.2289 200 400 600 800 1000 1200 1400 1600 1800 m/z m/z 1% S/N FWHM Res. 489.1700 9485 4.4 20.2 0.1047 4674 2 491.5948 14385 6.7 31.2 0.0938 5242 3 492.0976 10258 4.8 21.9 0.1129 4358 4 520.2106 24208 11.2 52.3 0.1134 4589 5 521,2136 10615 4.9 22.2 0.1164 4478 6 563.2511 56666 26.3 122.7 0.1237 4555 564.2519 24066 11.2 51.2 0.1152 4900 8 565.2597 6740 3.1 13.3 0.1404 4025 9 663.4547 5672 2.6 10.5 0.1718 3862 10 686.1688 70332 32.7 151.2 0.1406 4881 11 687.1710 35465 16.5 75.3 0.1395 4927 12 688,1699 9273 4.3 18.3 0.1415 4863 13 726.2017 21720 10.1 45.3 0.1423 5104 14 727.2016 12894 6.0 26.1 0.1414 5143 15 758.2081 8825 17.4 0.1555 4.1 4877 16 759.2033 7172 3.3 13.8 0.1629 4661 НŃ 17 784.2102 27355 12.7 58.7 0.1528 5133 18 785.2115 16031 7.4 33.6 0.1516 5180 19 982.1758 215290 100.0 525.4 0.1920 5116 20 983,1775 151832 70.5 370.3 0.1904 5164 21 984.1755 78033 36.2 1897 0.1985 4959 22 985.1763 31717 14.7 76.2 0.2021 4874 23 986.1733 11139 5.2 25.8 0.1857 5310 24 999.1735 13152 31.0 6.1 0.1972 5068 25 26 27 1000.1797 9084 4.2 21.0 0.1956 5114 1014.1702 1015.1706 8947 4.2 20.9 0.1892 5360 5688 2.6 12.7 0.1902 5336 28 1040.1811 13922 6.5 34.0 0.2074 5016 29 1041.1856 10243 24.6 4.8 0.1961 5310

Figure S18 HR-ESI mass spectrum of compound Co-1.

5852



Figure S19 Absorption spectrum of compound Co-1 in toluene.



**Figure S20** Calibration curve of **Co-1** in toluene ( $\lambda_{max} = 432$  nm).



**Figure S21** Emission spectrum of compound **Co-1** in toluene ( $\lambda_{ex} = 432$  nm).

**Table S1** TON, TOF and %FE for CO formation during a 6-h CPE experiment of ECR of

Time (h)	<b>TON</b> ( $\times 10^3$ )	$TOF / s^{-1}$	%FE
0.5	1.4	0.8	35.8
1.0	1.2	0.7	36.9
1.5	1.2	0.7	36.9
2.0	1.2	0.6	35.2
2.5	1.1	0.6	35.4
3.0	1.1	0.6	35.5
3.5	1.1	0.6	35.7
4.0	1.0	0.6	35.5
4.5	1.1	0.6	36.1
5.0	1.1	0.6	35.3
5.5	1.1	0.7	35.2
6.0	1.2	0.6	36.0
Average	$1.4 \times 10^{4 a}$	0.6	35.8

CO<sub>2</sub> under catalysis of **poly(Co-1)**/carbon paper

<sup>a</sup> A sum value



Figure S22 Cyclic voltammograms of the pre- and post-electrolysis **poly(Zn-1)**/carbon paper samples



Figure S23 Cyclic voltammograms of the pre- and post-electrolysis poly(Co-1)/carbon paper samples