AgNO₃-Catalyzed Decarboxylative Cross-coupling Reaction: An Approach to Coenzyme Q

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Experimental Procedures

All reactions were monitored by TLC (SiO₂, petrol ether/EtOAc 5:1), Melting points were measured on Melting Point M-565 (BUCHI). NMR and mass spectra were recorded on a Bruker Avanc III-HD 400 NMR and a TripleTOF Mass spectrometers, respectively. All reagents: e.g. Potassium Persulfate ($(K_2S_2O_8)$, Ammonium persulphate ($(NH_4)_2S_2O_8$) were purchased from Adamas, P. R. China, and used without further purification.

Table 1. Control experiments

	CoQ ₀	$\frac{\text{catalyst, oxidant, air}}{\text{HOOC}-(\text{CH}_2)_2 \text{CH}_3}$		(CH ₂) ₂ CH ₃
Entry	Oxidant	Catalyst	Solvent	Yield (%)
1	$K_2S_2O_8$	Ag ₂ CO ₃	CH ₃ CN	10
2	$K_2S_2O_8$	Ag_2O	CH ₃ CN	8
3	$K_2S_2O_8$	AgOAc	CH ₃ CN	14
4	$(NH_4)_2S_2O_8$	AgNO ₃	CH ₃ CN	25
5	H_2O_2	AgNO ₃	CH ₃ CN	0
6	TBHP	AgNO ₃	CH ₃ CN	trace
7	TBPB	AgNO ₃	CH ₃ CN	12
8	DTBP	AgNO ₃	CH ₃ CN	trace
9	TBPB	Bi(OTf) ₃	CH ₃ CN	0
10	TBPB	$Cu(OAc)_2$	CH ₃ CN	trace
11	TBPB	$Pd(OAc)_2$	CH ₃ CN	trace

Reaction Conditions: **CoQ**₀ (0.02 mol), butanoic acid (1.2 equiv), Catalyst 40%, Oxidant 2 equiv (DTBP: *di*-tertbutyl peroxide. TBHP=*tert*-butyl hydroperoxide, TBPB=*tert*-butyl peroxybenzoate). 80 °C, 4h

Synthesis of CoQ compounds

To a solution of Coenzyme Q_0 (3.64 g, 0.02 mol) and carboxylic acids (0.03mol) in acetonitrile 80 mL was added AgNO₃ (1.35 g, 8 mmol). The mixture was heated to 80 °C and a solution of K₂S₂O₈ (10.81 g, 0.04 mol) in distilled water 80 mL was added dropwise over 2 h, then the reaction mixture was stirred for another 2 h, with TLC monitoring unitil the starting material was consumed. The resulting mixture was cooled and extracted with CH₂Cl₂. The organic layer was washed with water, then dried over anhydrous Na₂SO₄ and evaporated under reduced pressure. The residue was purified by column chromatograph on silica gel (PE/EtOAc= 5:1) to give CoQ compounds.

Compound 3, red oil;

¹H NMR (400 MHz, CDCl₃) δ 3.99 (s, 3H, OCH₃), 3.98 (s, 3H, OCH₃), 2.45 (t, 2H, *J* = 8.0 Hz, CH₂), 2.02 (s, 3H, CH₃), 1.39-1.49 (m, 2H, CH₂), 0.96 (t, *J* = 8.0 Hz, 3H, CH₃).

¹³C NMR (101 MHz, CDCl₃) δ 184.7 (C=O), 184.2 (C=O), 144.3, 144.2, 142.8, 138.9, 61.2(OCH₃), 28.3, 22.1, 14.3, 11.9 (CH₃).

Compound 4, red oil;

¹H NMR (400 MHz, CDCl₃) δ4.00 (s, 3H, OCH₃), 3.99 (s, 3H, OCH₃), 2.45 (t, 2H, *J* = 8.0 Hz, CH₂), 2.01 (s, 3H, CH₃), 1.15-1.45 (m, 14H, CH₂), 0.88 (t, *J* = 8.0 Hz, 3H, CH₃).

¹³C NMR (101 MHz, CDCl₃) δ 184.7 (C=O), 184.2 (C=O), 144.3, 144.2, 143.1, 138.6, 61.2 (OCH₃), 31.9, 29.8, 29.5, 29.4, 29.3, 28.7, 26.4, 22.7, 14.1, 11.9 (CH₃).

Compound **5**, red oil^[1];

¹H NMR (400MHz, CDCl₃) δ 3.91 (s, 6H, OCH₃), 3.55 (t, *J* = 8.0Hz, 2H, CH₂OH), 2.52 (s, 1H, OH), 2.36 (t, *J* = 8.0Hz, 2H), 2.00 (s, 3H, CH₃), 1.48(t, *J* = 8.0Hz, 2H), 1.10-1.45 (m, 14H).

¹³CNMR (100MHz, (CD₃)₂CO) δ 184.5 (C=O), 183.9 (C=O), 144.0, 142.8(2C), 138.5, 62.6(OCH₃), 60.9(OCH₃), 32.5, 29.6, 29.2, 29.1, 29.0, 28.5, 26.2, 25.5, 11.7 (CH₃).

Idebenone (compound 6), red solid, m.p. 53-55 °C (Lit. ^[2] 52-54 °C).

¹H NMR (400 MHz, CDCl₃) δ 4.00 (s, 3H, OCH₃),3.99 (s,3H, OCH₃), 3.62-3.66 (m, 2H, CH₂), 2.45 (t, 2H, *J* = 8.0 Hz, CH₂), 2.01 (s, 3H, CH₃), 1.61 (s,1H,OH), 1.59-1.52 (m, 2H), 1.42-1.22 (m, 14H); ¹³CNMR (100 MHz, CDCl₃) δ 184.7 (C=O), 184.2 (C=O), 144.3, 144.2, 143.1, 138.7, 63.1(OCH₃), 61.2(OCH₃), 32.8, 29.8, 29.5, 29.4, 29.3, 28.7, 26.4, 25.7, 11.9 (CH₃).

References

- [1] J. Wang, S. Li, T. Yang and J. Yang, European Journal of Medicinal Chemistry, 2014, 86, 710-713
- [2] J. Wang, S. Li, T. Yang and J. Yang, Tetrahedron, 2014, 70, 9029-9032.





¹H NMR spectrum of compound **4**



¹³C NMR spectrum of compound **4**



¹H NMR spectrum of compound **5**



¹³C NMR spectrum of compound **5**



