

## Supplementary Information

# Catalytic oxidation mechanisms of carbon monoxide over single- and double-vacancy Mn-embedded graphene

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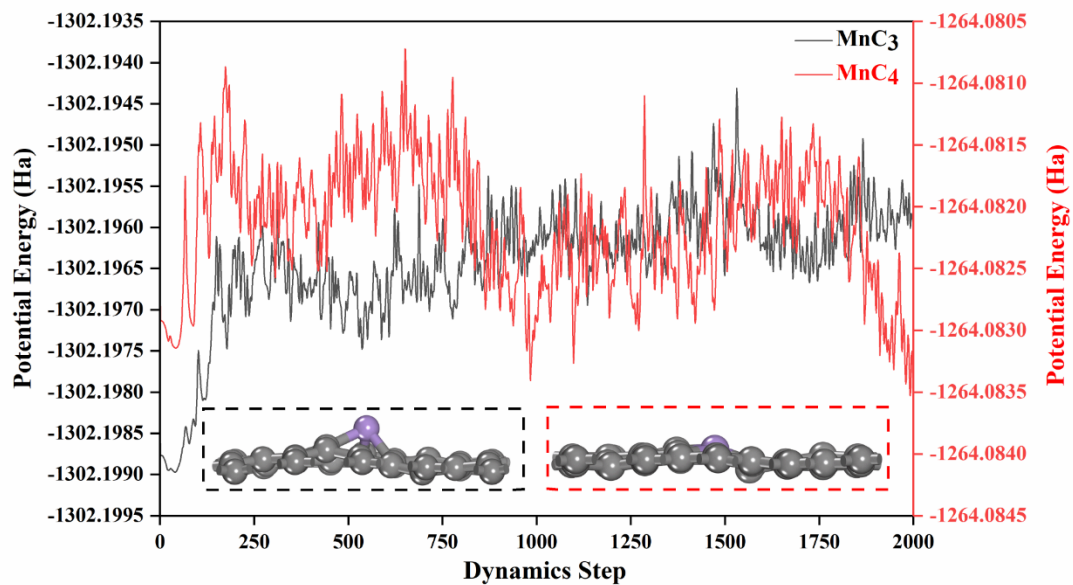


Fig. S1. The final structure (MnC<sub>3</sub> and MnC<sub>4</sub>) from dynamics simulation at 1000 K and potential energy change.

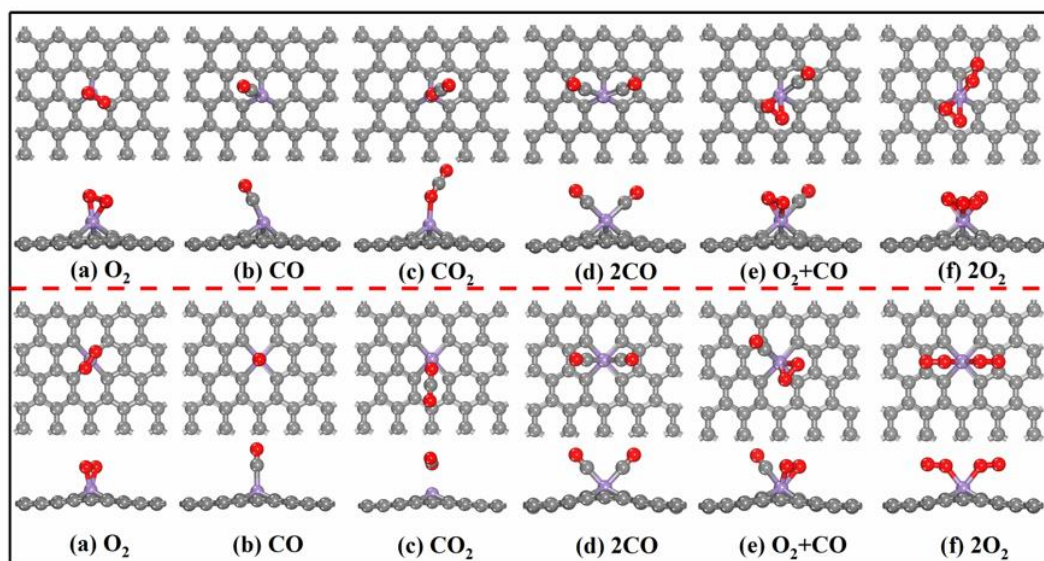


Fig. S2. The most stable configuration of O<sub>2</sub>, CO, CO<sub>2</sub> and bimolecular co-adsorption on MnC<sub>3</sub> and MnC<sub>4</sub>.

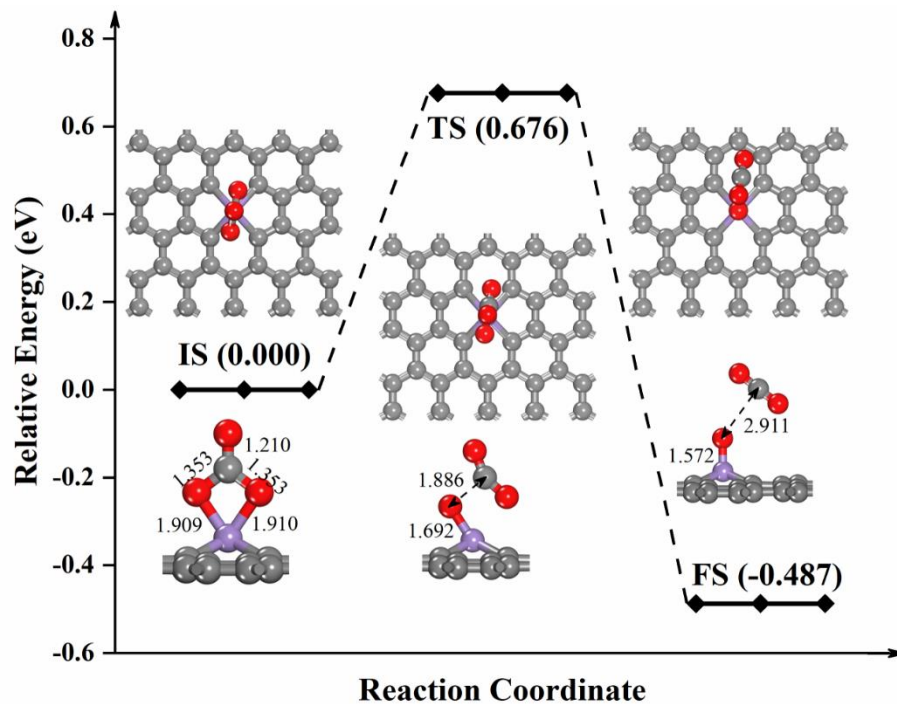


Fig. S3. OCOO is cracked into O and CO<sub>2</sub>.

Table S1 The energy barrier of RDS (reaction mechanism) and second CO<sub>2</sub> generation of MnC<sub>3</sub> and MnC<sub>4</sub>, \* represents the substrate at room temperature (298.15 K), -- represents the CO oxidation process has not been studied (or the reaction is not RDS and the effect of temperature has not been studied), the unit is eV.

Models	ER	LH	O+CO→CO <sub>2</sub>	TER
MnC <sub>3</sub>	0.832	0.829	--	--
*MnC <sub>3</sub>	0.835	0.761	--	--
MnC <sub>4</sub>	--	0.568	0.851	1.011
*MnC <sub>4</sub>	--	0.538	0.767	1.083