

Electronic Supplementary Information (ESI)

Novel nitro-substituted formazan derivatives: Selective ratiometric and colorimetric chemosensors for fluoride anion sensing detectable by the naked eye

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UV-vis spectra

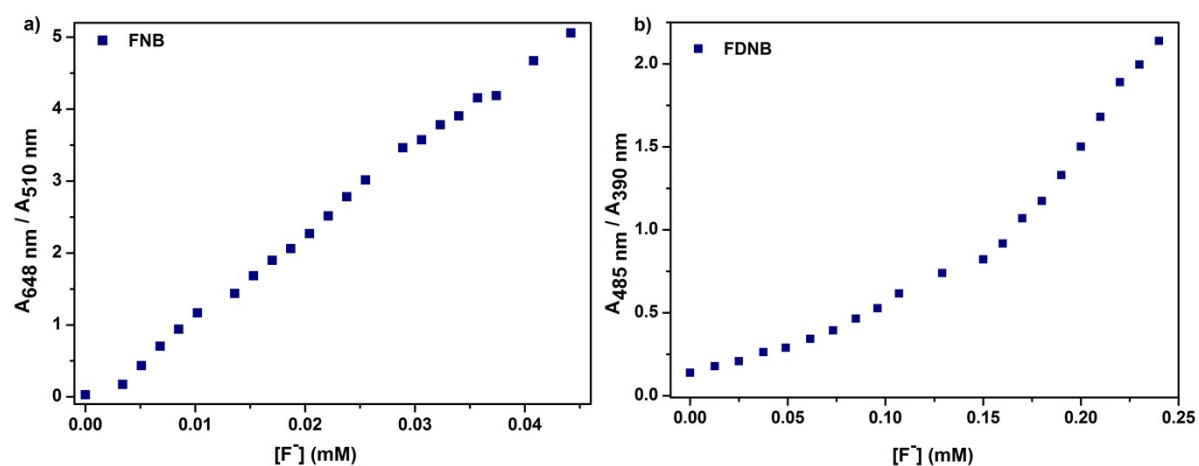


Figure S1. Ratiometric plots of **FNB** (a) and **FDNB** (b) with addition fluoride anion.

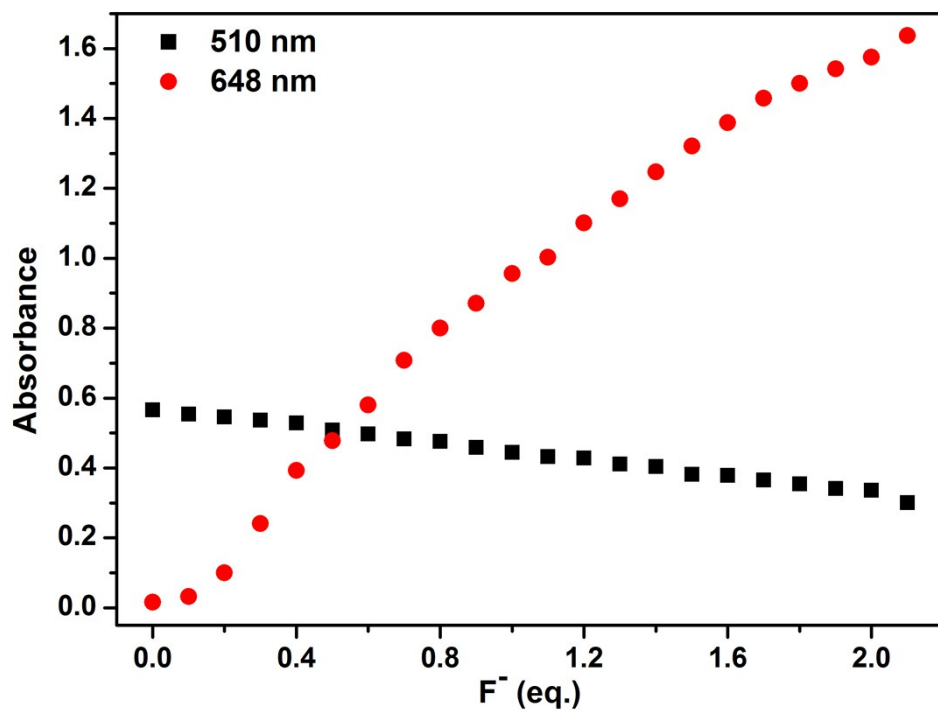


Figure S2. The titration curves of sensor **FNB** with F^- anion showing the 1:1 stoichiometry by UV-visible experiments.

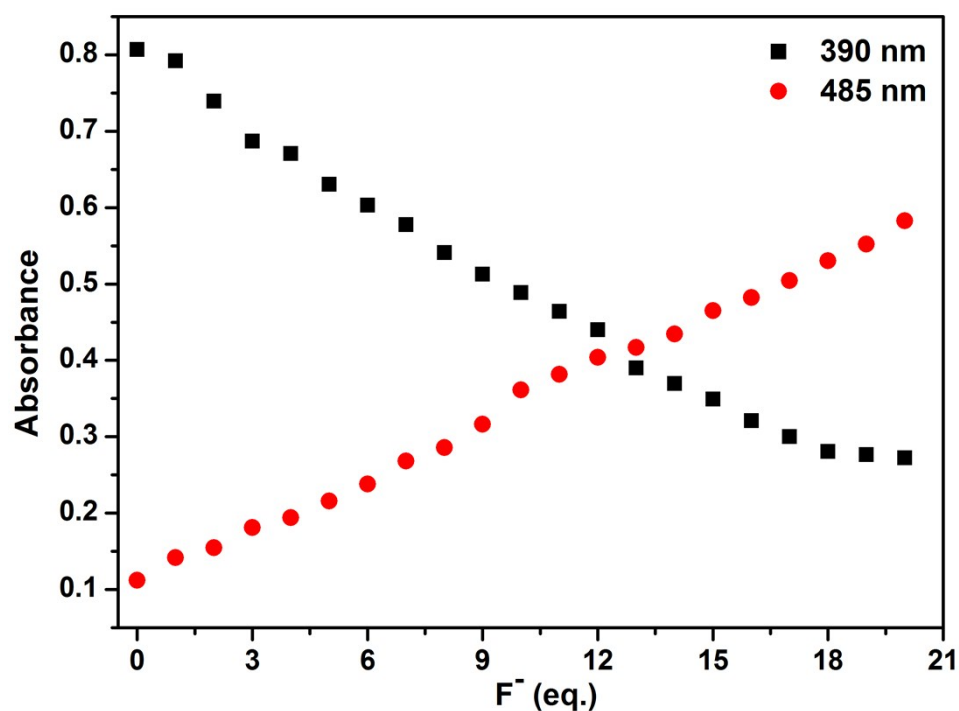


Figure S3. The titration curves of sensor **FDNB** with F^- anion showing the 1:1 stoichiometry by UV-visible experiments.

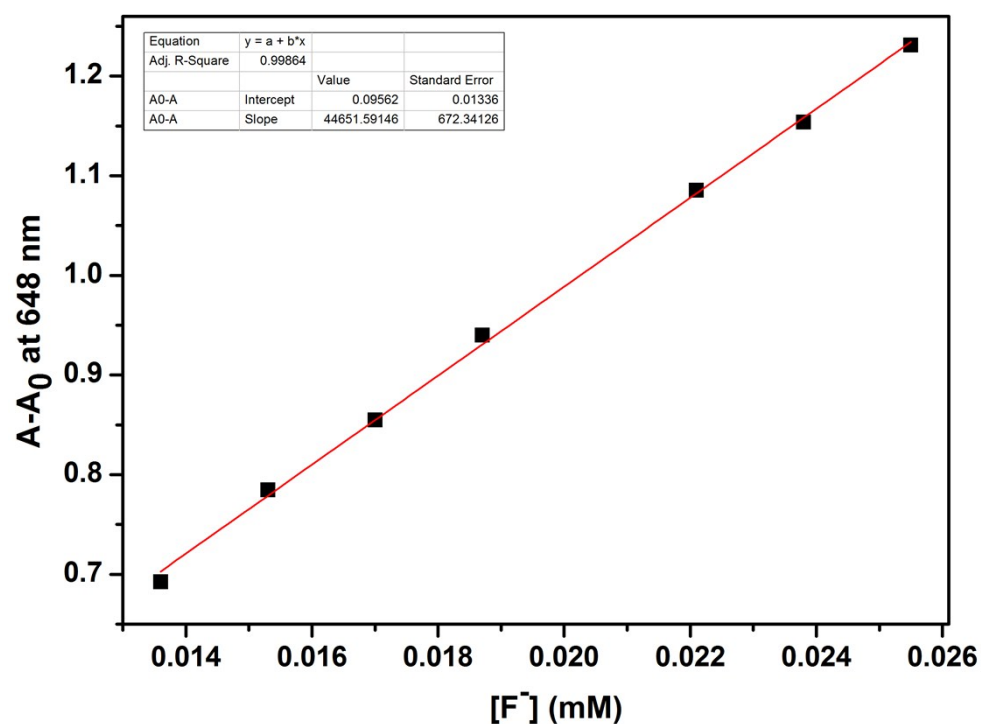


Figure S4. The absorbance change addition at 648 nm of the sensor **FNB** (17 μ M) as a function of fluoride concentration in THF. Fluoride concentration: 0.014–0.026 mM (the lower concentration part).

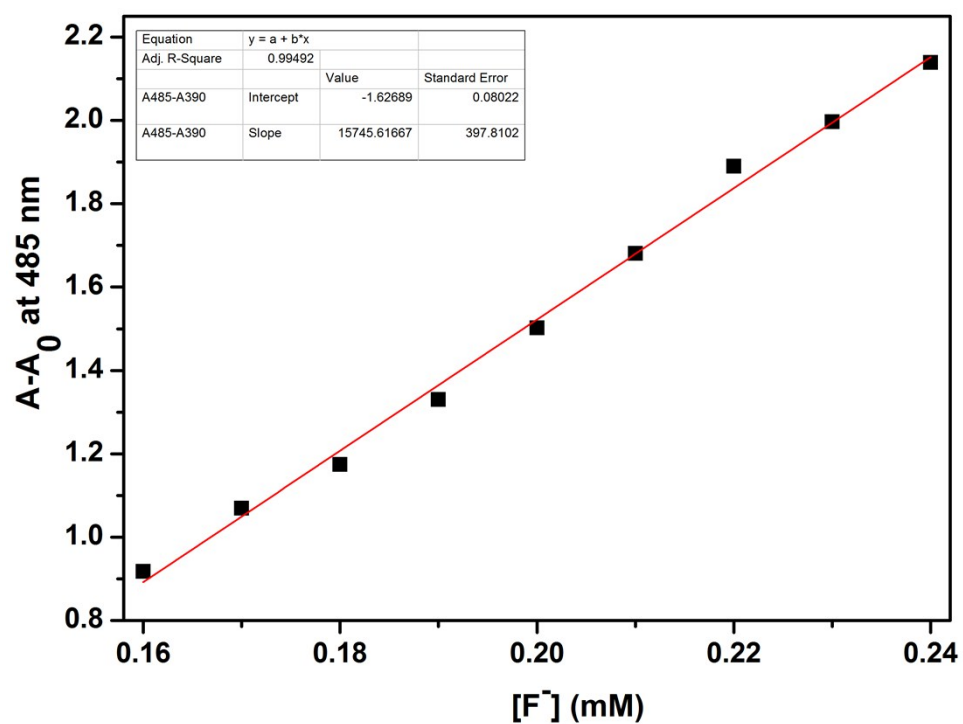


Figure S5. The absorbance change addition at 485 nm of the sensor **FDNB** (13 μM) as a function of fluoride concentration in THF. Fluoride concentration: 0.16–0.24 mM (the lower concentration part).

NMR spectra

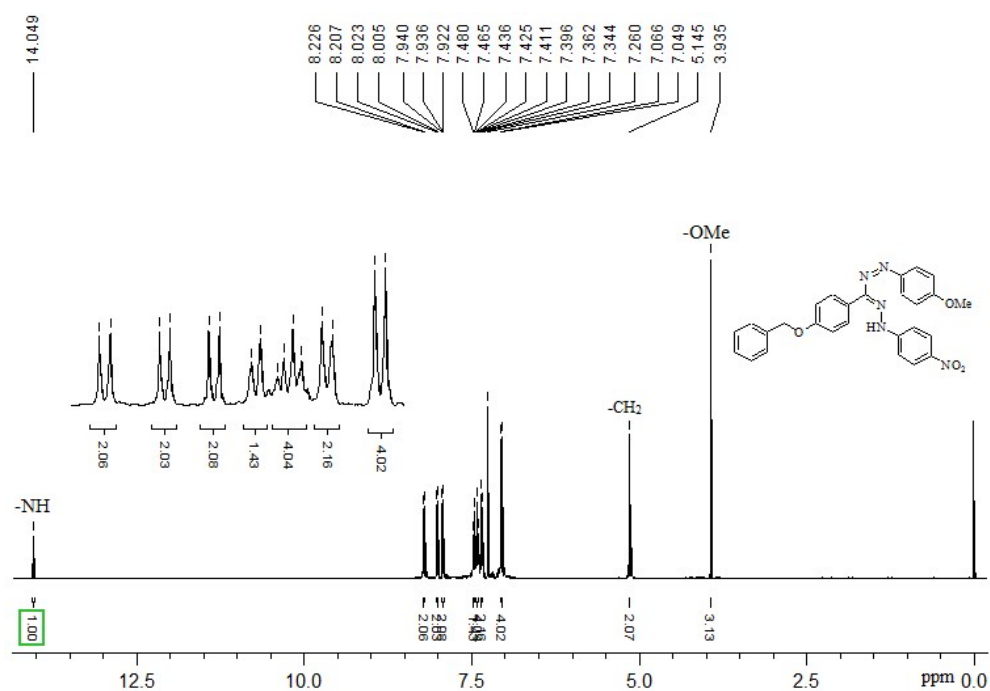


Figure S6. ¹H NMR (500 MHz, CDCl₃) spectrum of compound **FNB**.

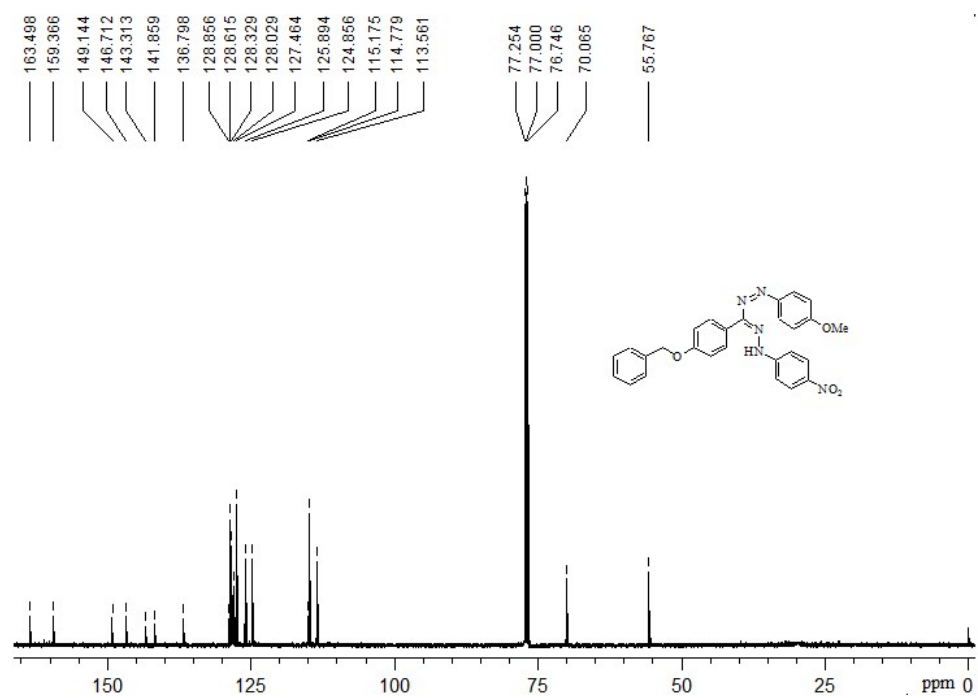


Figure S7. ¹³C NMR (126 MHz, CDCl₃) spectrum of compound **FNB**.

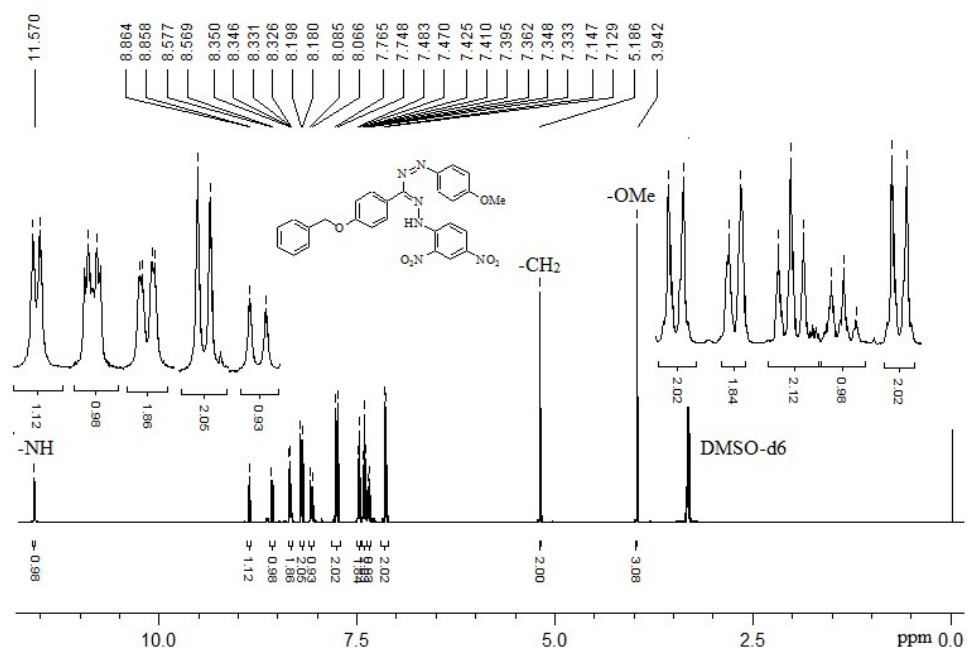


Figure S8. ¹H NMR (500 MHz, DMSO-*d*₆) spectrum of compound **FDNB**.

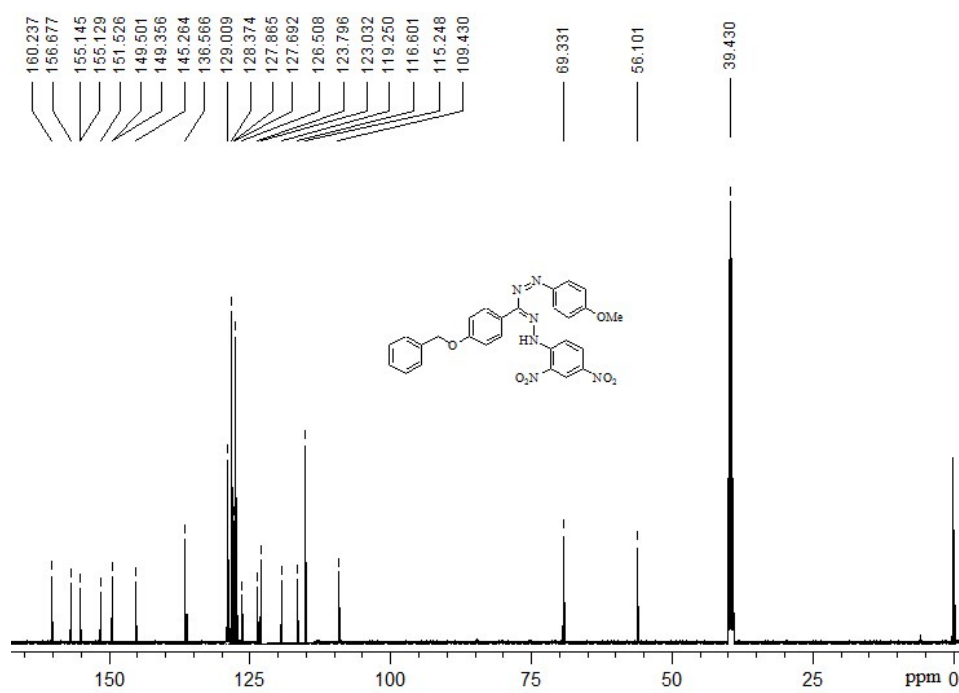


Figure S9. ¹³C NMR (126 MHz, DMSO-*d*₆) spectrum of compound **FDNB**.