

Supporting information

Two Anthracene Chromophores based Metal-Organic Frameworks for Gas Absorption and Promising Nitro Aromatic Sensing

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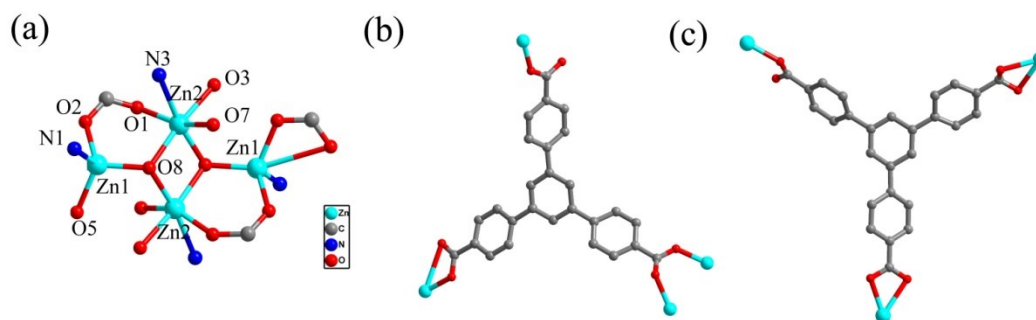


Fig. S1 (a) Tetranuclear Zn-cluster in **1**. (b) and (c) The coordination modes of btb in **1** and **2**.

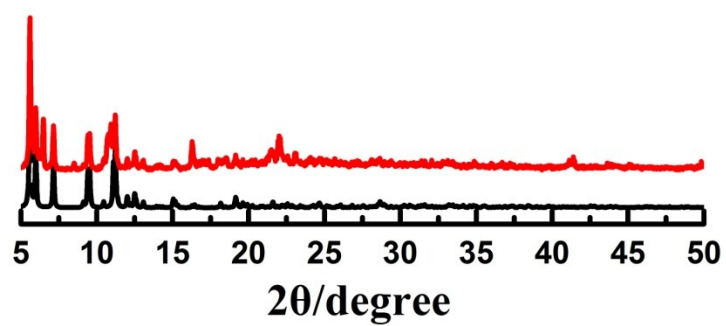


Fig. S2 PXRD patterns for **1**.

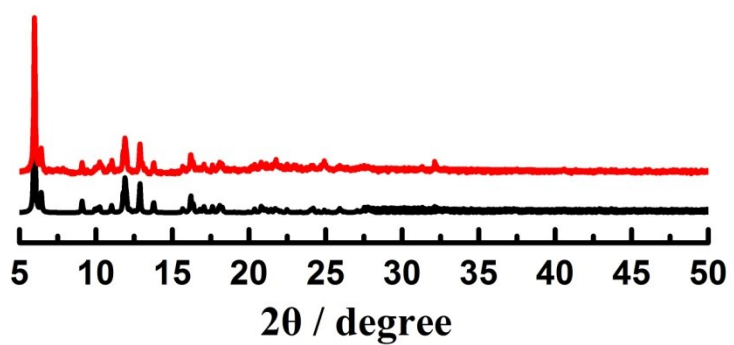


Fig. S3 PXRD patterns for **2**.

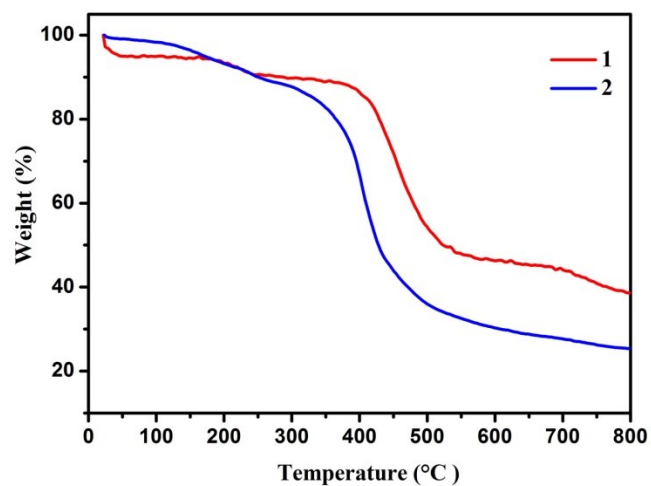


Fig. S4 TG curves for 1 and 2.

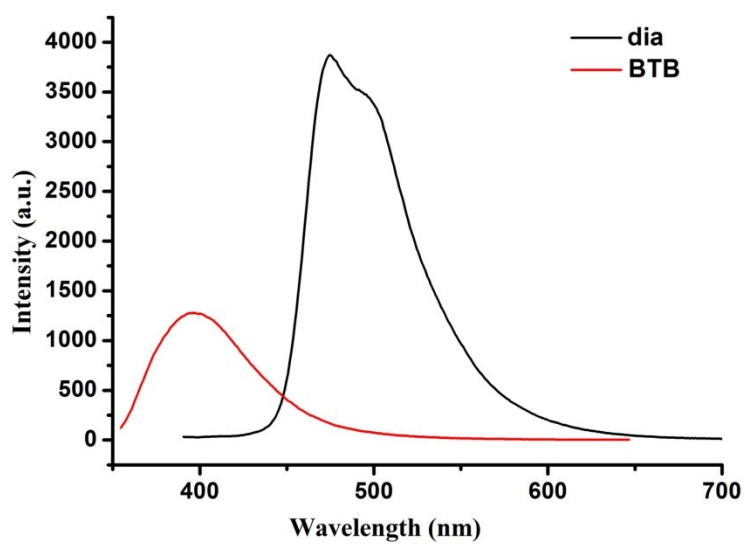


Fig. S5 Emission spectra of btb and dia ligand.

Table S1 Crystallographic data and structure refinements for **1** and **2**.

	1	2
Formula	C ₄₇ H ₃₀ N ₄ O ₈ Zn ₂	C ₇₄ H ₄₄ Cd ₂ N ₄ NaO ₁₂
Mr	909.53	1428.92
Crystal system	triclinic	monoclinic
space group	P-1	C2/c
a, Å	9.797(4)	32.143(4)
b, Å	15.252(6)	10.3824(10)
c, Å	16.948(7)	32.772(4)
α, deg	75.387(7)	90
β, deg	73.275(7)	115.790(4)
γ, deg	85.170(7)	90
V, Å ³	2346.7(16)	9847.4(18)
Z	2	4
D _x , g cm ⁻³	1.287	0.964
μ, mm ⁻¹	1.075	0.480
Unique. reflns/R _{int}	0.0577	0.0545
R ₁ [I > 2σ(I)]	0.0765	0.0721
wR ₂ (All data)	0.2153	0.2190
GOF	0.871	0.967

Table S2 Selected bond lengths [Å] and Selected angles [°] for **1**.

Zn1-N1	2.021(6)	O3#3-Zn2-O1#2	153.2(3)
Zn1-O2	1.945(4)	O3#3-Zn2-O9#2	92.47(16)
Zn1-O5#1	1.951(4)	O3#3-Zn2-N3	78.3(3)
Zn1-O9	1.934(4)	O9#2-Zn2-Zn2#2	84.6(3)
Zn2-O1#2	1.984(6)	O9-Zn2-O7	41.34(12)
Zn2-O3#3	1.969(4)	O9#2-Zn2-N3	88.19(18)
Zn2-O7	2.156(5)	N3-Zn2-Zn2#2	144.8(3)
Zn2-O9#2	2.109(4)	O2-Zn1-O5#1	120.92(18)
Zn2-O9	2.074(5)	O9-Zn1-N1	110.48(19)
Zn2-N3	2.327(8)	O9-Zn1-O5#1	109.68(18)
O2-Zn1-N1	98.6(2)	O1#2-Zn2-O7	86.4(2)
O5#1-Zn1-N1	107.4(2)	O1#2-Zn2-O9	109.5(2)
O9-Zn1-O2	108.96(17)	O3#3-Zn2-Zn2#2	96.18(15)
O1#2-Zn2-Zn2#2	106.59(19)	O3#3-Zn2-O7	89.50(19)
O1#2-Zn2-O9#2	95.37(19)	O3#3-Zn2-O9	96.8(2)
O1#2-Zn2-N3	74.9(3)	O7-Zn2-Zn2#2	130.37(15)
O9-Zn2-N3	83.53(17)	O9#2-Zn2-O7	42.19(10)
		O9-Zn2-O9#2	171.66(19)

Symmetry transformations used to generate equivalent atoms: 1 -1+X,+Y,1+Z; 2 -X,-Y,-1-Z; 3 1-X,-1-Y,-1-Z.

Table S3 Selected bond lengths [Å] and Selected angles [°] for **2**.

Cd1-O1#1	2.487(4)	O3-Cd1-O6#2	93.8(3)
Cd1-O2#	2.277(4)	O3-Cd1-O8#2	147.9(3)
Cd1-N1	2.250(4)	O3-Cd1-N1	107.00(18)
Cd1-O3	2.167(5)	O3-Cd1-O6#2	93.8(3)
Cd1-O5#2	2.400(10)	O3-Cd1-O8#2	147.9(3)
Cd1-O6#2	2.358(8)	O3-Cd1-N1	107.00(18)
Cd1-O7#2	2.507(8)	O3-Cd1-O6#2	93.8(3)
Cd1-O8#2	2.384(11)	O3-Cd1-O8#2	147.9(3)
O1#1-Cd1-O7#2	139.4(3)	O2#1-Cd1-O1#1	54.02(13)
O2#1 Cd1-O5#2	108.6(3)	O2#1-Cd1-O6#2	113.4(3)
O2#1 Cd1-O7#2	86.3(3)	O2#1 Cd1 O8#2	88.8(3)
N1-Cd1-1#1	81.38(15)	N1-Cd1-O2#1	131.14(17)
N1-Cd1-O5#2	79.4(3)	N1-Cd1-O6#2	107.9(3)
N1-Cd1-O7#2	129.6(3)	N1-Cd1-O8#2	95.2(3)
O3-Cd1-O1#1	95.47(19)	O3-Cd1-O2#1	95.47(19)
O3-Cd1-N1	107.00(18)	O3-Cd1-O5#2	94.9(2)
		O3-Cd1-O7#2	105.3(3)

Symmetry transformations used to generate equivalent atoms: 1 +X,2-Y,1/2+Z; 2 -1/2+X,-1/2+Y,+Z.

Table S4 The comparison of K_{sv} for **1**, **2** and other reported MOFs.

MOF	K_{sv}	Ref
1	$2.84 \times 10^5 \text{ M}^{-1}$	In this work
2	$2.50 \times 10^5 \text{ M}^{-1}$	In this work
[Tb(L)(OH)]·x(solv)	$7.73 \times 10^{-2} \text{ ppm}^{-1}$	1
Zr-NDC/Tz and Zr-NDC/CN)	$1.8 \times 10^4 \text{ M}^{-1}$	2
[Cd(NDC) _{0.5} (PCA)].xG	$3.5 \times 10^4 \text{ M}^{-1}$	3
[(CH ₃) ₂ NH ₂] ₃ [Zn ₄ Na(BP TC) ₃].4CH ₃ OH·2DMF	$3.2 \times 10^4 \text{ M}^{-1}$	4
[Zn(NDC)(H ₂ O)] _n	$6 \times 10^4 \text{ M}^{-1}$	5
[Cd(NDC)(H ₂ O)] _n	$2.385 \times 10^4 \text{ M}^{-1}$	5
{[Tb(L) _{1.5} (H ₂ O)]·3H ₂ O} _n	$7.47 \times 10^4 \text{ M}^{-1}$	6

Reference

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