

## Electronic Supplementary Information for-

### Charge transport in phenazine-fused triphenylene discotic mesogens doped with CdS nanowires<sup>†</sup>

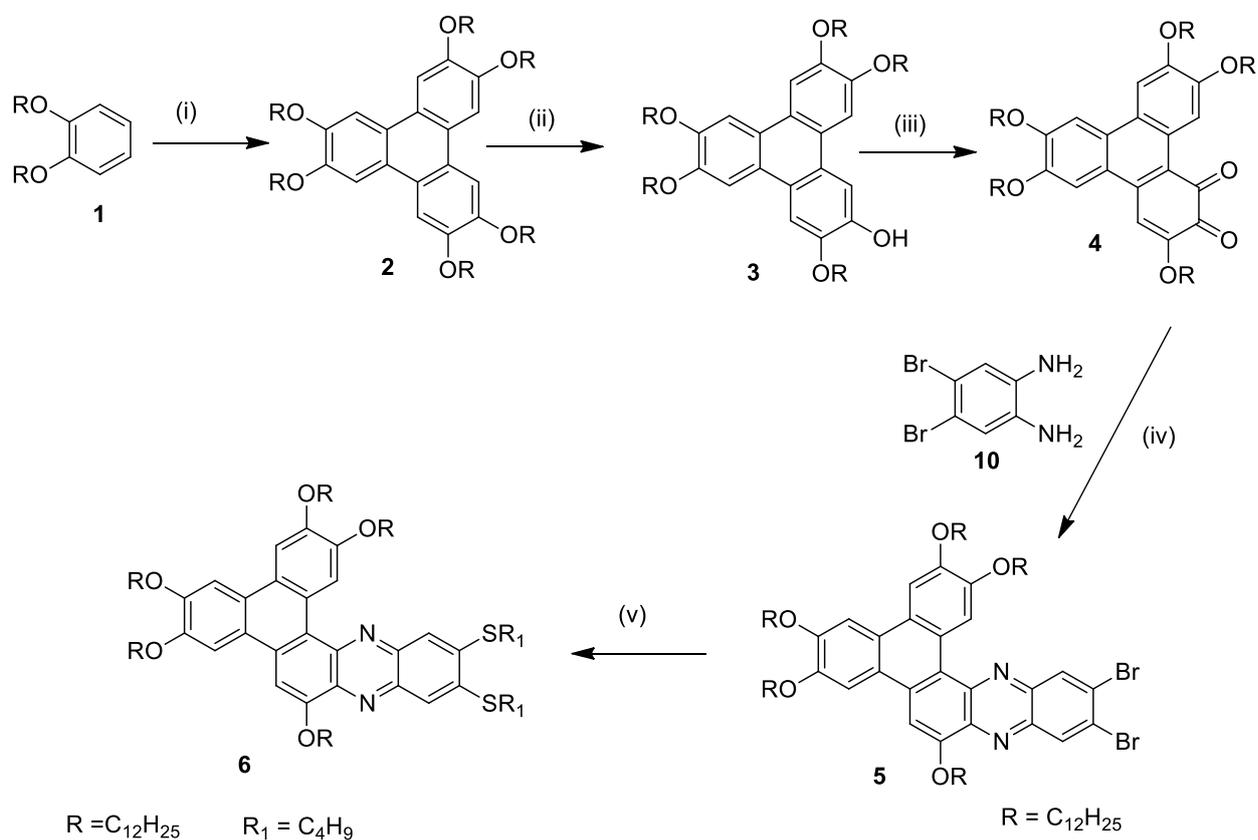
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### Synthesis of phenazine-fused triphenylene discotic liquid crystal

The synthesis of extended triphenylene fused mesogenic derivative as shown in **Scheme 1**.



**Scheme 1.** (i) FeCl<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, r.t, 30 min; (ii) catechol boron bromide, CH<sub>2</sub>Cl<sub>2</sub>, r.t, 24 h; (iii) CAN, CH<sub>3</sub>CN, r.t, 30 min; (iv) CH<sub>3</sub>COOH: toluene (7:3), reflux, 6 h; (v) alkanethiol, Cs<sub>2</sub>CO<sub>3</sub>, DMAC, reflux, 24 h; (vi) *p*-TosCl, pyridine, r.t, 24 h; (vii) Br<sub>2</sub>, NaOAc, acetic acid, 110 °C, 3 h; (viii) Con. H<sub>2</sub>SO<sub>4</sub>, 110 °C, 15 min.

### General procedure for the synthesis of final compound

The monohydroxytriphenylene **3** was synthesised by following reported procedure [1]. Oxidation of the 2-hydroxy-3,6,7,10,11-pentakis(alkyloxy) triphenylene **3** with ceric ammonium nitrate (CAN) gives 3,6,7,10,11-pentakis(alkyloxy)triphenylene-1,2-diones **4** [2]. The intermediate 4,5-dibromobenzene-1,2-diamine **10** compound was synthesized following a reported procedure [3] and it was condensed with triphenylene-1,2-diquinone **4** in presence of glacial acetic acid in toluene (7:3) under reflux condition to afford intermediate 9,10-dibromo-2,3,6,14,15-pentaalkoxyphenanthro [9, 10-a]phenazine compounds (**5**). Further, reaction of 1-butanethiol with intermediate mesogenic compound **5** in presence of cesium carbonate under reflux condition produces the desired extended phenazine based triphenylene mesogenic derivative (**6**). All the intermediates and final compounds were purified by column chromatography, followed by recrystallization with appropriate solvent and well characterized using spectral and elemental analysis.

### Reference

- [1] S. Kumar, M. Manickam, *Synthesis*. 1998, 1119–1122.
- [2] S. Kumar, M. Manickam, S. K. Varshney, D. S. S. Rao, S. K. Prasad, *J. Mater. Chem.* 2000, 10, 2483–2489
- [3] J. Shao, J. Chang, C. Chi, *Org. Biomol. Chem.* 2012, 10, 7045–7052.

Figure S1. 3D crystal structure of CdS. It has been obtained using the refined parameters in VESTA software with ball-and-stick and polyhedron models of hexagonal phase.

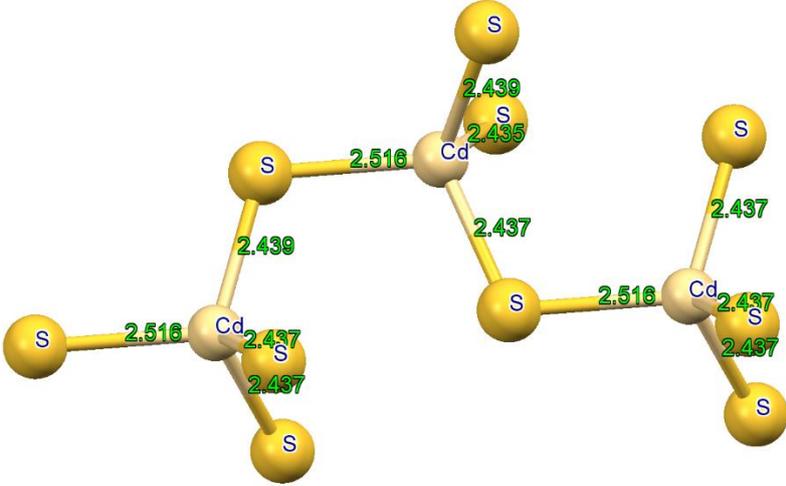


Figure S2. Average roughness profile of the PFT/CdS composite.

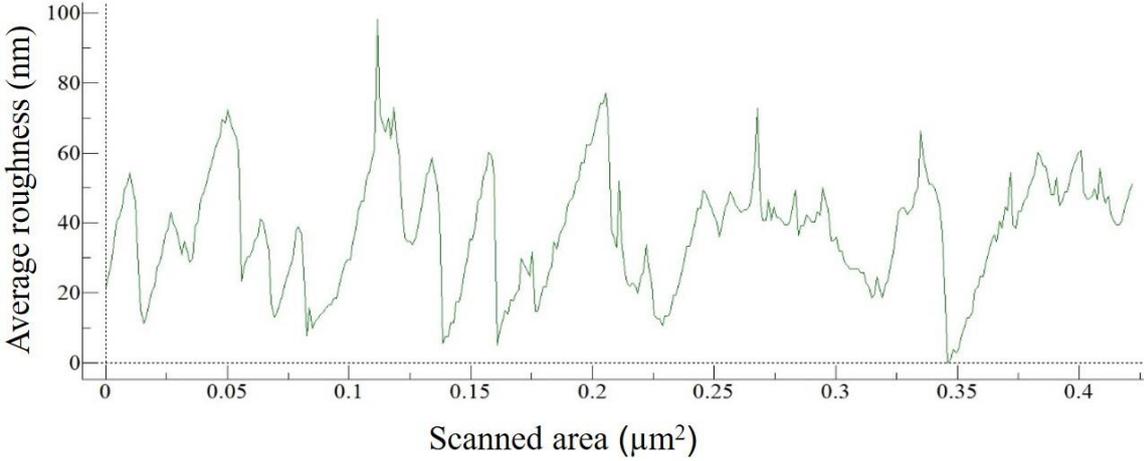


Figure S3. Thickness profilometric image of the ITO-PFTDLC/CdS-Au heterojunction.

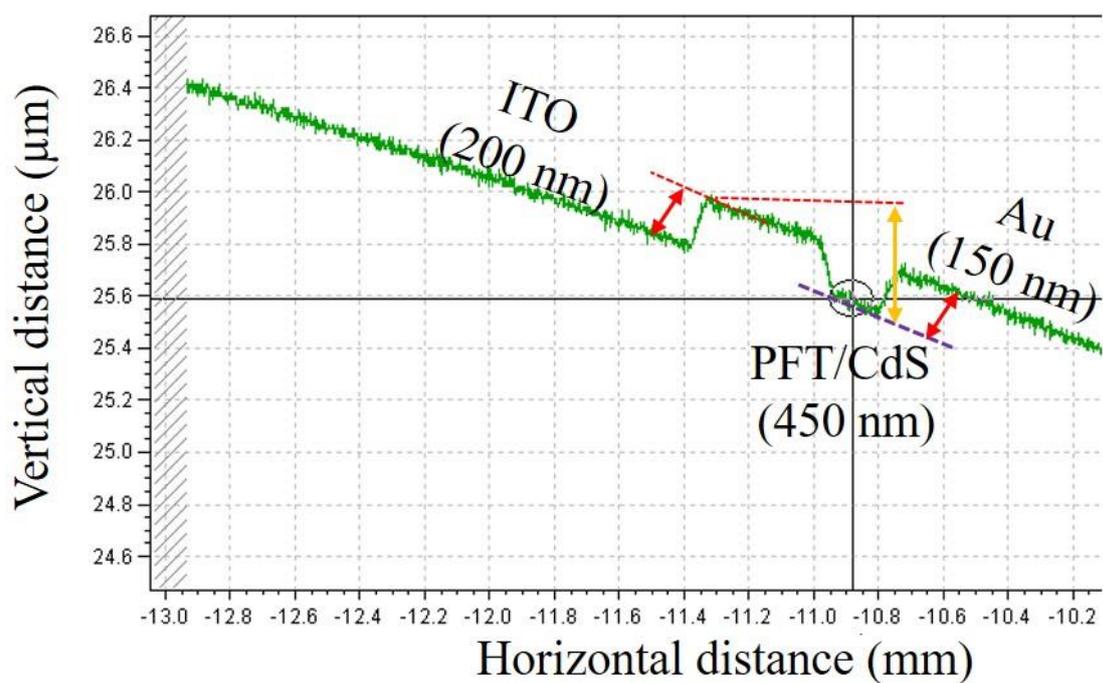


Figure S4. High resolution TEM image of CdS NWs.

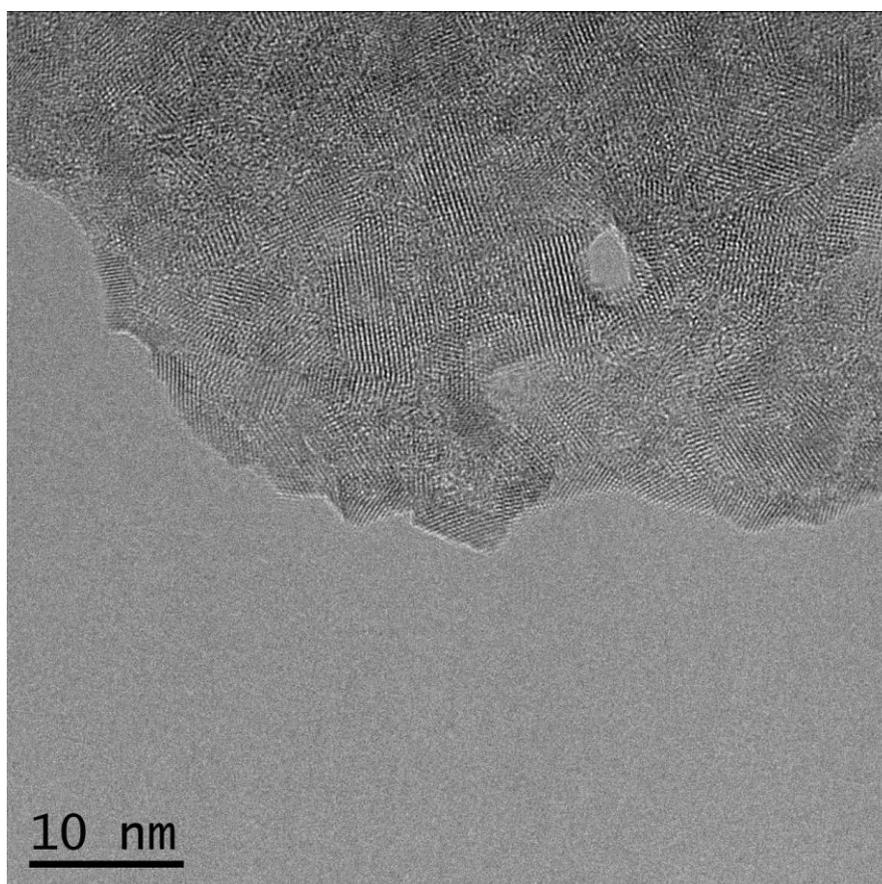


Table S1. X-ray data of the PFT DLC and PFT/CdS composite.

Compound	2 $\theta$ ( $^{\circ}$ )	d-spacing observed (calculated) $\text{\AA}$	Phase/Lattice constant	Col <sub>h</sub> parameter (r)	Miller indices	Alkyl- chain length ( $\text{\AA}$ )	Core-Core separation ( $\text{\AA}$ )	Intercolumnar distance ( $\text{\AA}$ )
<b>PFT DLC</b>	3.59	24.58 (24.57)	Col <sub>h</sub> a = 28.38	1	100	4.50	3.55	28.38
	6.22	14.19 (14.18)		$1/\sqrt{3}$	110			
	7.19	12.28 (12.28)		1/2	200			
<b>PFT/CdS composite</b>	3.67	24.06 (24.05)	Col <sub>h</sub> a = 27.78	1	100	4.47	3.50	27.78
	6.35	13.94 (13.89)		$1/\sqrt{3}$	110			
	7.29	12.10 (12.02)		1/2	200			

Table S2. Details of transit time for the PFT DLC and PFT/CdS composite at 90 V.

Temperature ( $^{\circ}\text{C}$ )	Transit time (PFT DLC) (sec)	Transit time (PFT/CdS composite) (sec)
70	0.000130175	1.40E-04
80	6.68805E-05	1.10E-04
90	4.06029E-05	8.44E-05
100	3.48631E-05	5.07E-05
110	2.90968E-05	4.26E-05
120	2.14839E-05	2.80E-05

Methods to calculate the hole mobility:

**(I) Direct calculation method:**

In this method, first we obtain a best transient photocurrent curve by examining it at various voltages; and we obtain the transit time ( $\tau$ ); then, the hole mobility ( $\mu_h$ ) is calculated using the formula;  $\mu_h = d^2 / \tau.V$ , where  $\tau$  is the transit time obtained by photocurrent curves,  $V$  is the applied voltage and  $d$  is the thickness of ITO cell.

**(II) Calculation via slope of transit time versus 1/V:**

In this method, we first plot a curve between transit time and inverse of voltages (here we take several voltages) i.e.  $\tau = \left( \frac{d^2}{\mu} \right) \left[ \frac{1}{V} \right]$ . In this relation the value of slope is equivalent to  $d^2/\mu$  having intercept value equals to zero. Then mobility is calculated using the formula;  $\mu = d^2 / slope$ .