

Electronic Supplementary Material (ESI) for New Journal of Chemistry

**Preparation of Cu-Cu₂O-CuO by solid combustion ignited by dielectric barrier
discharge and its activity towards *p*-nitrophenol reduction**

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Figure S1 Non-ignited the precursor of $\phi = 1.5$ with physically adsorbed water of 1.4%.

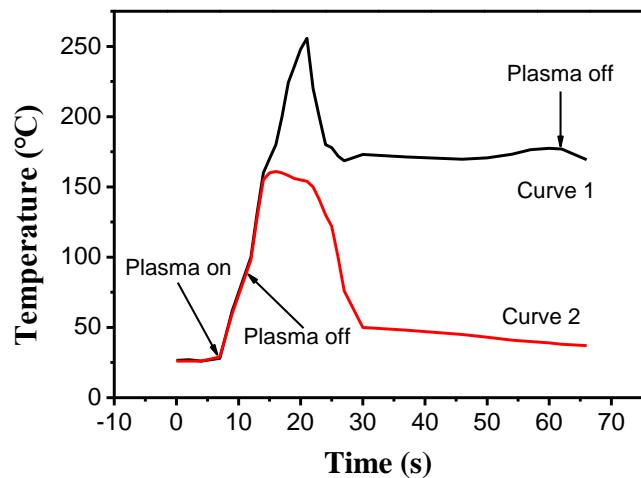
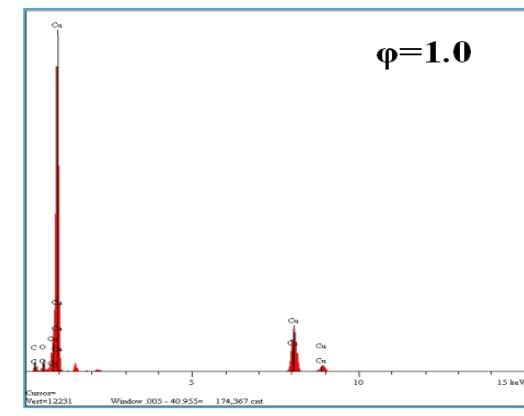
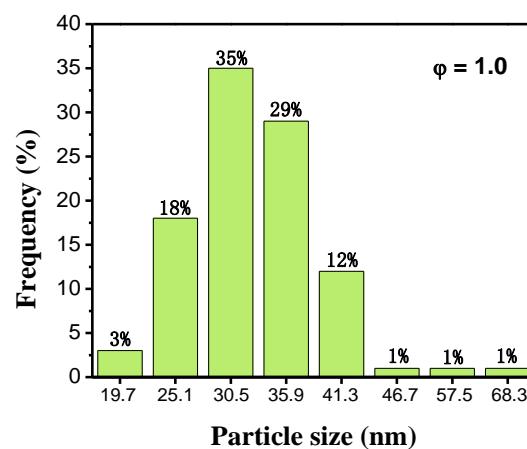
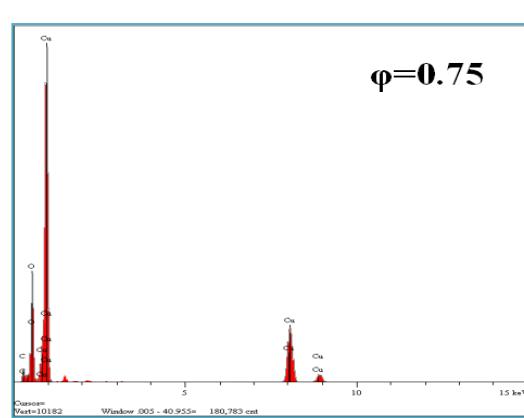
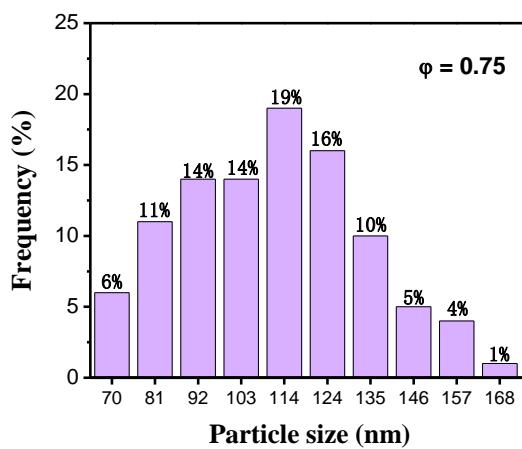
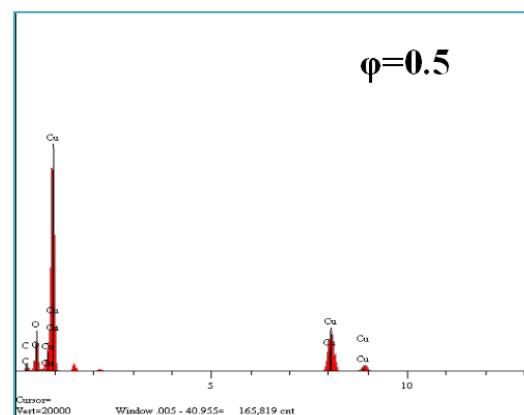
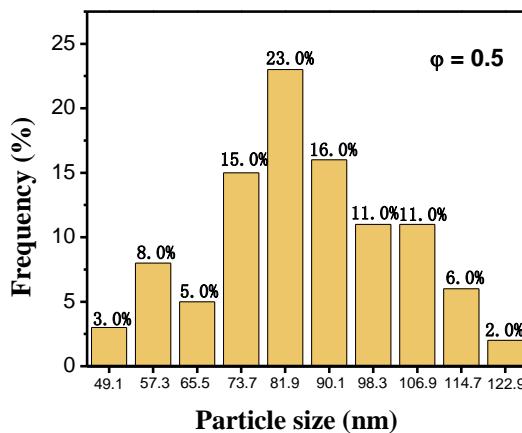


Figure S2 The change of bulk temperature with discharge time in the process of DBD plasma induced combustion.



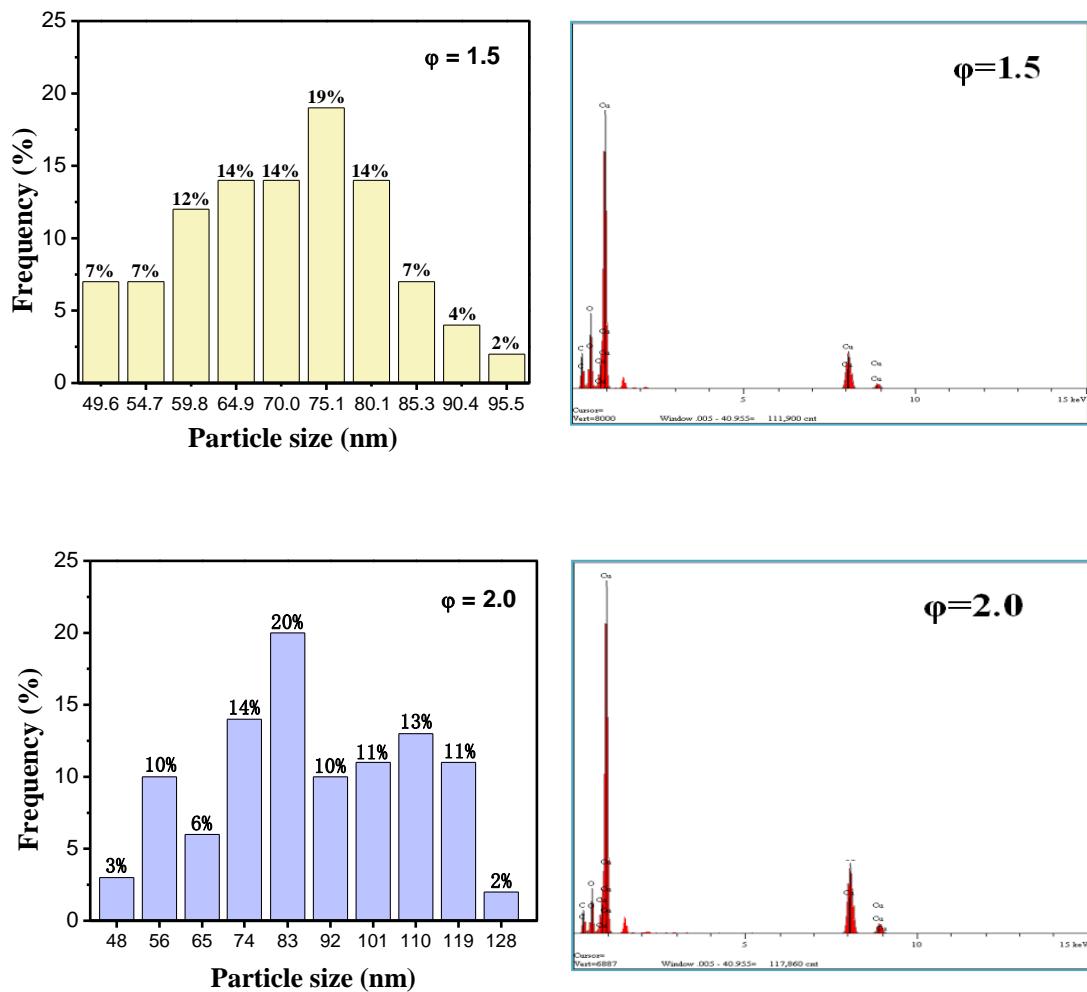


Figure S3 Size distribution histogram and EDX of $\phi = 0.5-2.0$ samples and the changes of copper and oxygen surface concentration with ϕ .

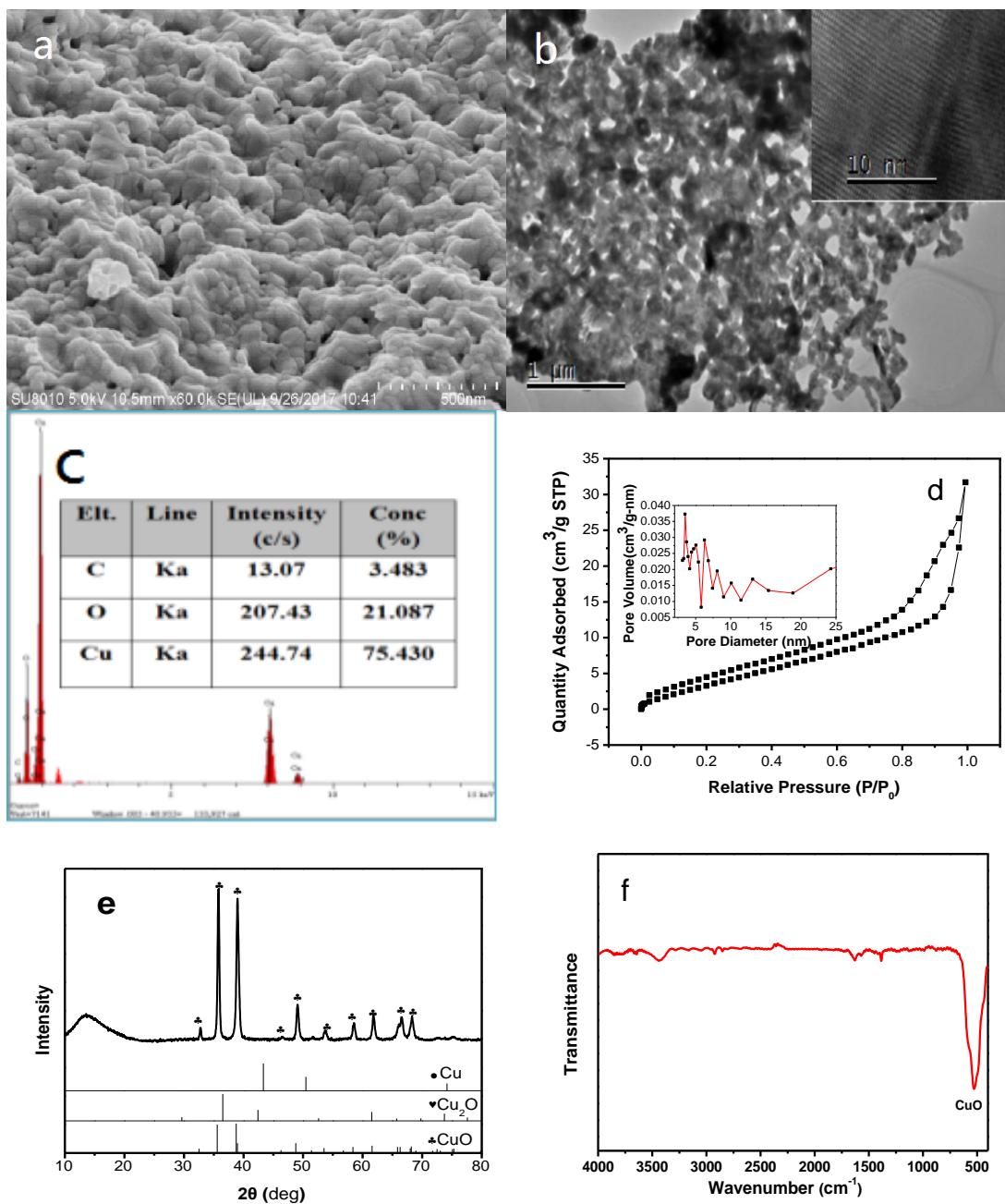


Figure S4 Characterization results of the sample prepared by chemical method (a) SEM (b) TEM (c) EDS (d) N_2 adsorption and desorption curves (e) XRD and (f) FTIR images of the chemical prepared sample.

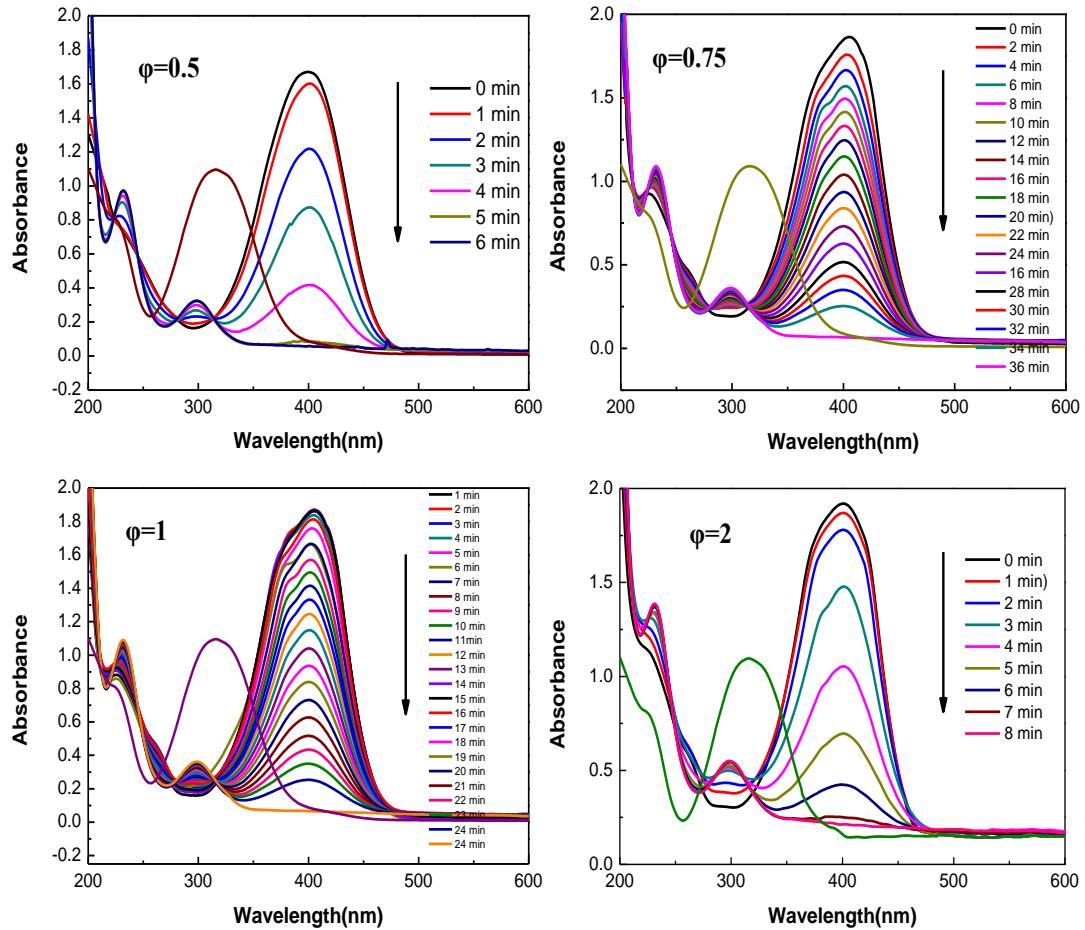


Figure S5 Comparison of the catalytic activities of the Cu-Cu₂O-CuO samples with different φ for the reduction of PNP by NaBH₄.

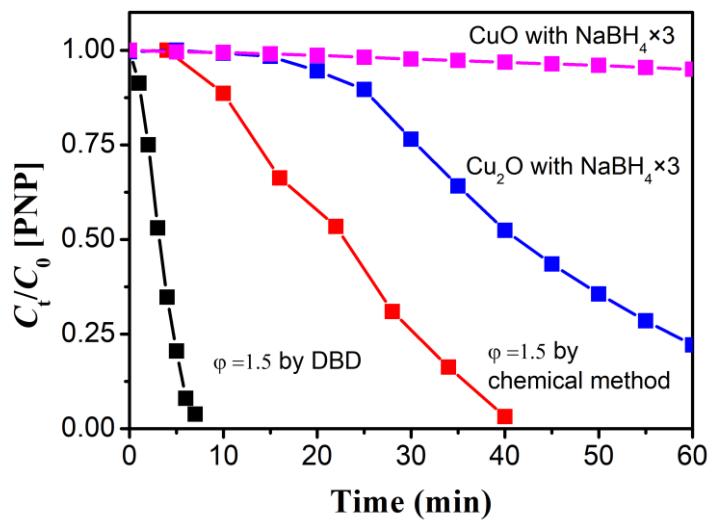


Figure S6 Comparison of the catalytic activities of the DBD and chemical method prepared samples of $\varphi=1.5$ and commercial CuO and Cu_2O for the reduction of PNP by NaBH_4 .

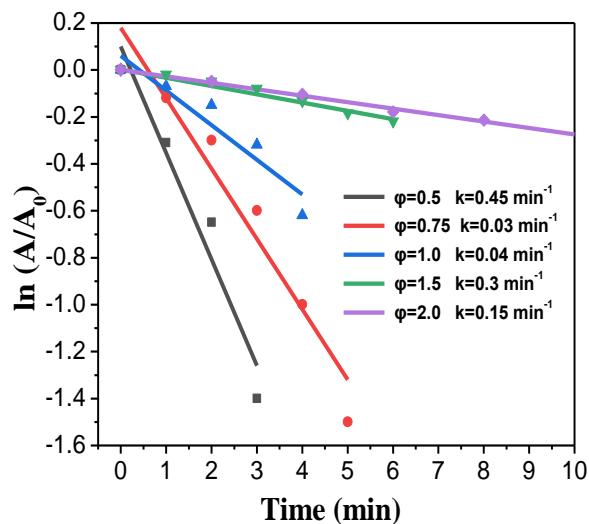


Figure S7 Apparent rate constants for the reduction of PNP by NaBH_4 using the $\text{Cu}-\text{Cu}_2\text{O}-\text{CuO}$ with $\varphi = 0.5, 0.75, 1$ and 2 .