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Synthesis, structural characterization and *in vitro* antiproliferative effects on carcinoma cells of novel organotin(IV) compounds with nicotinate and isonicotinate moieties.

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ELECTRONIC SUPPLEMENTARY INFORMATION

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Figure S5. ¹H NMR spectra (CDCl₃, 400.1 MHz, 21 °C) of compound 6.







aromatic region.



	3	4	5	7
Empirical formula	C ₁₇ H ₁₉ NO ₄ Sn	C ₁₇ H ₁₉ NO ₄ Sn	$C_{15}H_{15}NO_3Sn$	$C_{17}H_{19}NO_4Sn$
Formula weight	420.02	420.02	375.97	420.02
Temperature (K)	297(2)	297(2)	100(2)	297(2)
Crystal system	triclinic	orthorhombic	orthorhombic	triclinic
Space group	<i>P</i> -1	Pbca	P212121	<i>P</i> -1
a (Å)	9.142(2)	11.6366(11)	7.2969(6)	8.942(2)
<i>b</i> (Å)	9.455(2)	11.8098(11)	10.3968(8)	10.955(2)
<i>c</i> (Å)	11.142(3)	25.255(2)	19.3079(12)	11.156(3)
α (°)	104.291(4)	90	90	118.128(4)
β (°)	107.472(4)	90	90	98.558(4)
γ (°)	91.291(4)	90	90	105.645(4)
Volume (Å ³)	885.3(3)	3470.7(6)	1464.78(19)	877.8(3)
Ζ	2	8	4	2
D_{calc} (g cm ⁻³)	1.576	1.608	1.705	1.589
Absorption coefficient (mm ⁻¹)	1.461	1.491	1.705	1.474
<i>F</i> (000)	420	1680	744	420
Crystal size (mm)	0.18x0.19x0.25	0.29x0.28x0.15	0.06x0.08x0.11	0.2x0.26x0.28
θ range for data collection (°)	1.988 to 25.00	2.38 to 25.089	2.23 to 28.31	2.16 to 25.1
Reflections collected	8642	31704	19773	8608
Independent reflections	3118	3084	3644	3126
	$[R_{int}=0.034]$	$[R_{int}=0.0625]$	$[R_{int} = 0.0259]$	$[R_{int}=0.0452]$
Absorption correction	Multi-scan ¹	Multi-scan ¹	Multi-scan ¹	Multi-scan ¹
Data / restraints / parameters	3118/0/210	3084/0/210	3644/0/184	3126/0/211
Goodness-of-fit on F^2	1.051	1.081	1.072	1.001
Final <i>R</i> indices $[I \ge 2\sigma(I)]$	$R_1 = 0.0327$	$R_1 = 0.0389$	$R_1 = 0.0136$	$R_1 = 0.0393$
	$wR_2 = 0.0778$	$wR_2 = 0.0940$	$wR_2 = 0.0336$	$wR_2 = 0.0920$
R indices (all data)	$R_1 = 0.0367$	$R_1 = 0.0553$	$R_1 = 0.0141$	$R_1 = 0.0494$
	$wR_2 = 0.0808$	$wR_2 = 0.1068$	$wR_2 = 0.0339$	$wR_2 = 0.0991$
Largest difference peak and	0.77 and -0.28	1.20 and -0.32	0.44 and -0.27	0.81 and -0.32
hole (e Å ⁻³)				
CCDC No.	2004064	2004065	2004066	2004067

Table S1. X-ray crystal data and structure refinement for compounds 3, 4, 5 and 7.

¹G. M. Sheldrick, *SADABS, Program for area detector adsorption correction*, Institute for Inorganic Chemistry, University of Göttingen, Germany, 1996.

CCDC 2004064, 2004065, 2004066 and 2004067 for **3**, **4**, **5** and **7**, respectively, contain the supplementary crystallographic data for this contribution. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

$[2-{(CH_2O)_2CH}C_6H_4]Me_2Sn[O(O)CC_5H_4N-4] (3)$

- the crystal contains a 1:1 mixture of $pR_{O(1)}-R_{O(1)}-S_{C(7)}$ - and $pS_{O(1)}-S_{O(1)}-R_{C(7)}-3$ isomers



Figure S10. Molecular structure of $pR_{O(1)}-R_{O(1)}-S_{C(7)}-3$ isomer (*left*) and $pS_{O(1)}-S_{O(1)}-R_{C(7)}-3$ isomer (*right*) in the crystal of **3** (only methyne hydrogens are shown).

[2-{(CH2O)2CH}C6H4]Me2Sn[O(O)CC5H4N-3] (4)

- the crystal contains a 1:1 mixture of $pR_{O(1)}-S_{O(1)}-R_{C(7)}$ - and $pS_{O(1)}-R_{O(1)}-S_{C(7)}$ -4 isomers



Figure S11. Molecular structure of $pR_{O(1)}-S_{O(1)}-R_{C(7)}-4$ isomer (*left*) and $pS_{O(1)}-R_{O(1)}-S_{C(7)}-4$ isomer (*right*) in the crystal of 4 (only methyne hydrogens are shown).



Figure S12. C–H···O and C–H··· π interactions in the crystal of **3** [symmetry equivalent atoms (–*x*, *1*–*y*, *1*–*z*), (*1*–*x*, *1*–*y*, *2*–*z*) and (*x*, –*1*+*y*, *z*) are given by "a", "b" and "c", respectively].

Intermolecular C–H…O interactions		
C(10)–H(10B)····O(2a)	2.56 Å	$\Sigma r_{\rm cov}({\rm O},{\rm H}) = 0.96 {\rm ~\AA}$
		$\Sigma r_{vdW}(O,H) = 2.60 \text{ Å}$

Intermolecular C–H··· π interactions		
$C(13)-H(13)\cdots Ph_{centroid}{C(1b)-C(6b)}$	2.80 Å	$\gamma = 5.0^{\circ}$
$C(5c)-H(5c)-Py_{centroid} \{N(1),C(13)-C(17)\}$	3.02	$\gamma = 9.4^{\circ}$



Figure S13. Dimer association in the crystal of **3** [symmetry equivalent atoms (-x, 1-y, 1-z), (1-x, 1-y, 2-z) and (x, -1+y, z) are given by "a", "b" and "c", respectively].



Figure S14. View along a axis in the packing of 3 showing a 2D layer (only hydrogen atoms involved in the intermolecular contacts are shown).



Figure S15. View along c axis in the packing of 3 showing a 2D layer (only hydrogen atoms involved in the intermolecular contacts are shown).



Figure S16. View along *b* axis in the packing of **3** showing parallel 2D layers (only hydrogen atoms involved in the intermolecular contacts are shown).



Figure S17. C–H··· π interactions in the crystal of 4 [symmetry equivalent atom (0.5–*x*, 0.5+*y*, *z*) is given by "a"].



Figure S18. Zig-zag chain in the crystal of 4 [symmetry equivalent atom (0.5-x, 0.5+y, z) is given by "a"].



Figure S19. Parallel zig-zag chains in the crystal of 4. View along *a*, *b* and *c* axis.



Figure S20. C–H···O intra- and intermolecular interactions in the crystal of **5** [symmetry equivalent atoms (0.5-x, 1-y, -0.5+z), (1-x, 0.5+y, 0.5-z), (x, 1+y, z), (0.5-x, 1-y, 0.5+z), (x, -1+y, z) and (1-x, -0.5+y, 0.5-z) are given by "a", "b", "c", "d", "e" and "f", respectively].

Intramolecular C-H···O inter	actions			
C(10)–H(10)···O(2)	2.48 Å			
C(6)–H(6)····O(2)	2.49 Å			
Intermolecular C–H…O interactions				
$C(3)-H(3)\cdots O(3d)$	2.48 Å	$\Sigma r_{\rm cov}({\rm O,H}) = 0.96$ Å		
C(12)–H(12)····O(3f)	2.49 Å	$\Sigma r_{vdW}(O,H) = 2.60 \text{ Å}$		
C(11)–H(11)····O(1e)	2.54 Å			



Figure S21. View along *a* axis of the crystal packing in 5.



Figure S22. View along b and c axis of the 3D supramolecular architecture in the crystal of 5.



Figure S23. Polymer chain formed *via* C–H…N interactions in the crystal of 7 [symmetry equivalent atoms (1+x, y, -1+z), (2+x, y, -2+z), (-1+x, y, 1+z) and (-2+x, y, 2+z) are given by "a", "b", "c" and "d", respectively].



Figure S24. Parallel chains in the crystal packing of 7.



Figure S25. View along *a* and *b* axis in the crystal packing of **7**.



Figure S26. View along *c* axis in the crystal packing of 7



Figure S27. IR spectrum of compound 3.



Figure S28. ATR-IR spectrum of compound 4.



Figure S29. ATR-IR spectrum of compound 6.



Figure S30. IR spectrum of compound 7.



Figure S31. HRMS spectrum of compound 3.



Figure S32. HRMS spectrum of compound 4.



Figure S33. HRMS spectrum of compound 6.



Figure S34. HRMS spectrum of compound 7.