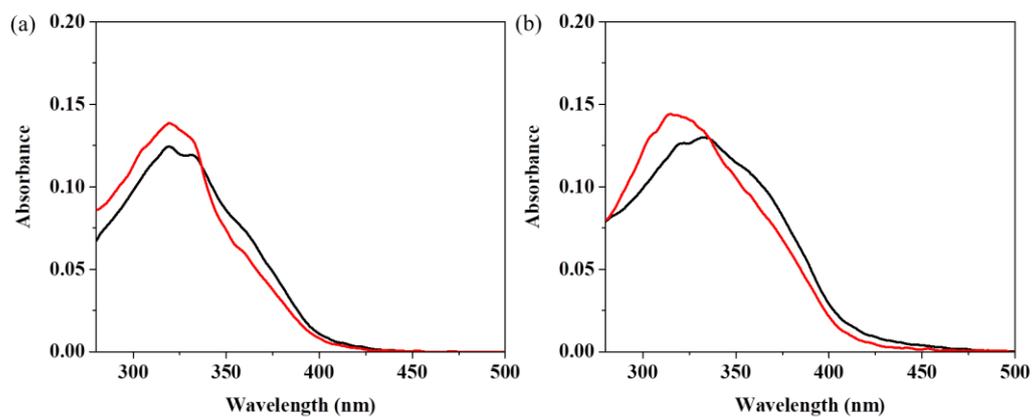


# Synthesis, Photochemical Isomerization and Photophysical Properties of Hydrazide-Hydrazone Derivatives

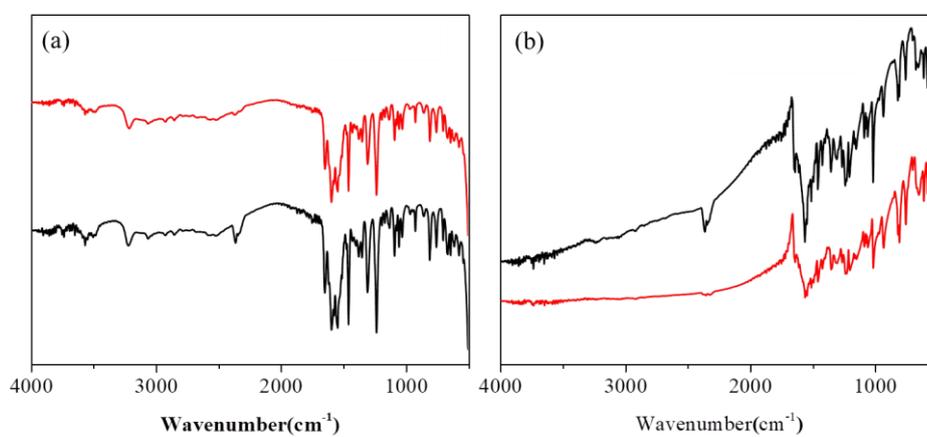
Bao-Xing Wu, Hsin-Yueh Chang, Yi-Shun Liao and Mei-Yu Yeh\*

Department of Chemistry, Chung Yuan Christian University, Republic of China, Taiwan

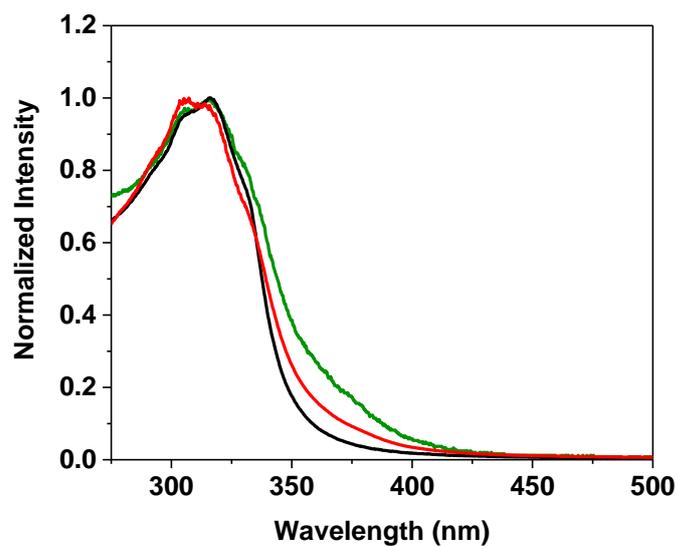
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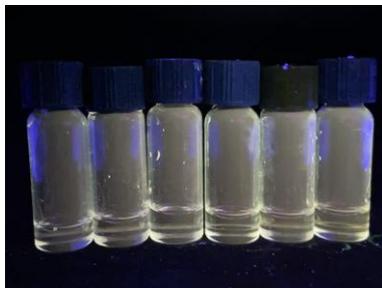
**Fig. S1** UV-Vis absorption spectra of (a) **3b** and (b) **3c** before irradiation (black) and after irradiation (red) with 365 nm light (Molecule concentration: 50  $\mu$ M).



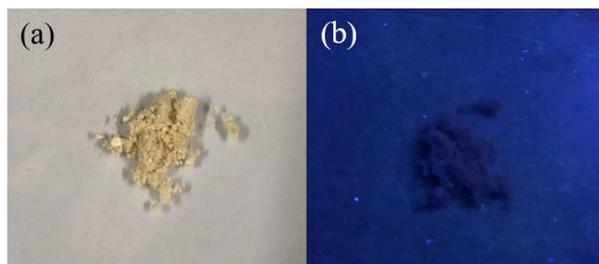
**Fig. S2** FT-IR spectra of (a) **3b** and (b) **3c** before irradiation (black) and after irradiation (red) with 365 nm light.



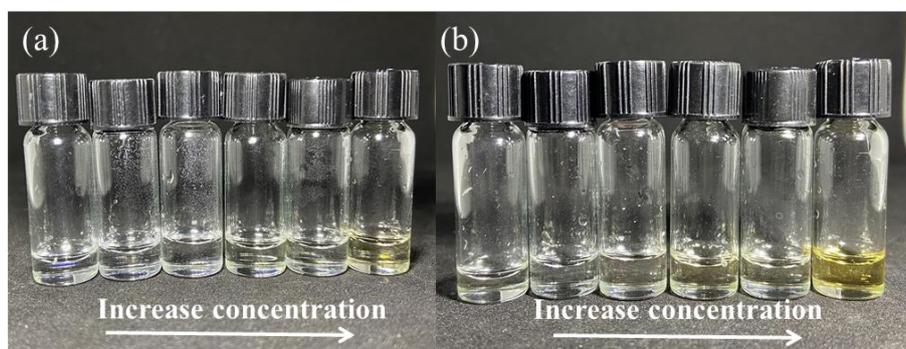
**Fig. S3** Concentration dependence of UV-Vis absorption spectra of **3a** after irradiation with 365 nm light in DMSO (black for 50  $\mu\text{M}$ , red for 500  $\mu\text{M}$  and green for 5,000  $\mu\text{M}$ ).



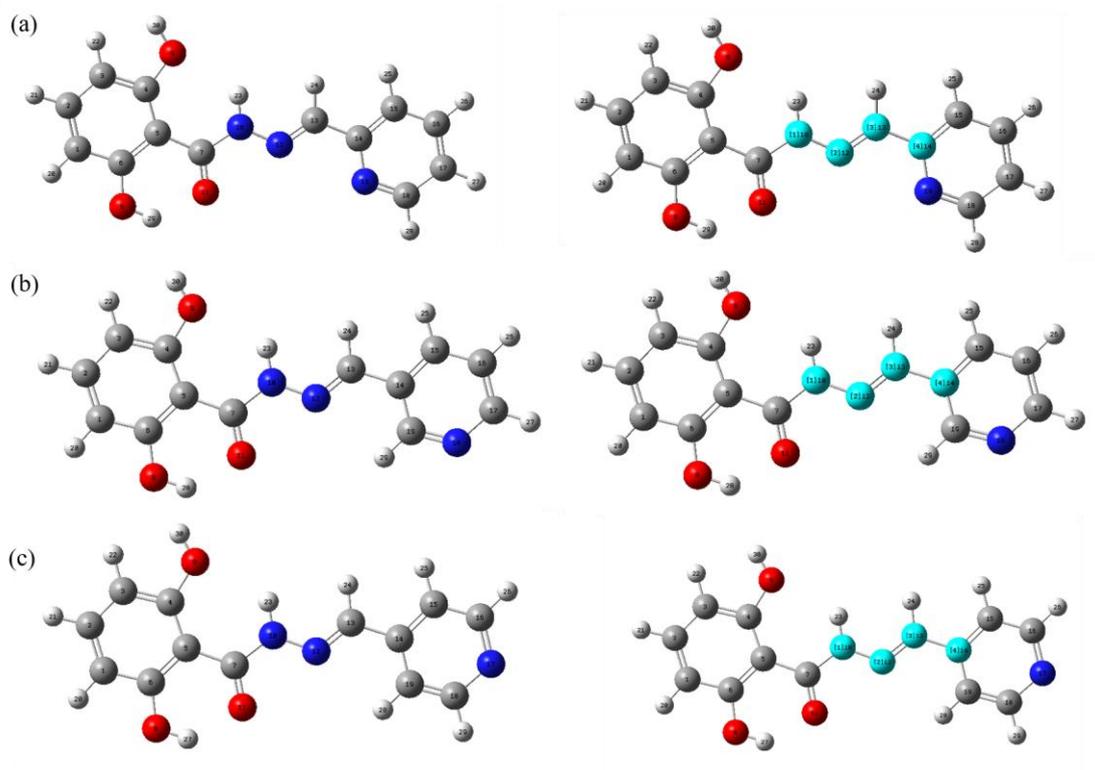
**Fig. S4** Optical images of **3a** before irradiation with 365 nm light taken under UV illumination (Concentrations from left to right: 50, 100, 500, 1000, 5000  $\mu\text{M}$ ).



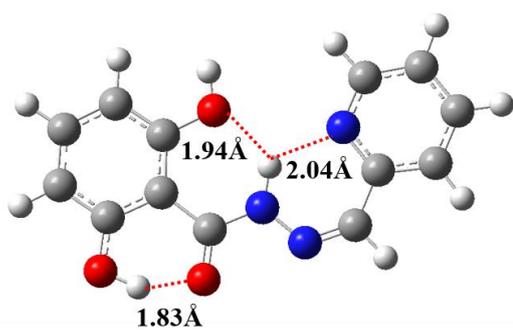
**Fig. S5** Photographs of solid powders of **3a** taken under (a) ambient room lighting and (b) UV illumination.



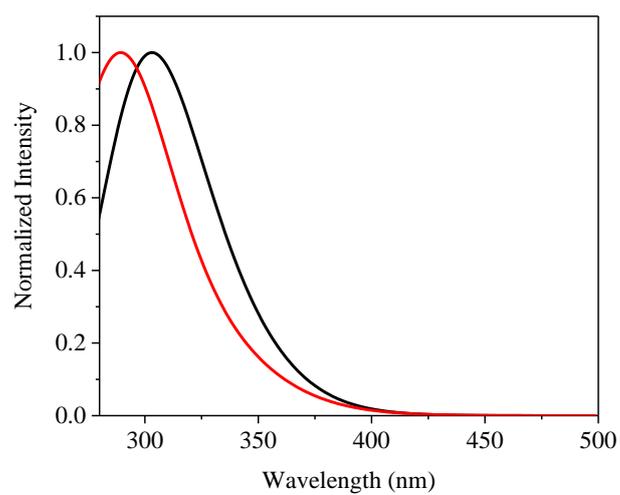
**Fig. S6** Optical images of **3a** (a) before irradiation and (b) after irradiation with 365 nm light taken under ambient room lighting (Concentrations from left to right: 50, 100, 500, 1000, 5000  $\mu\text{M}$ ).



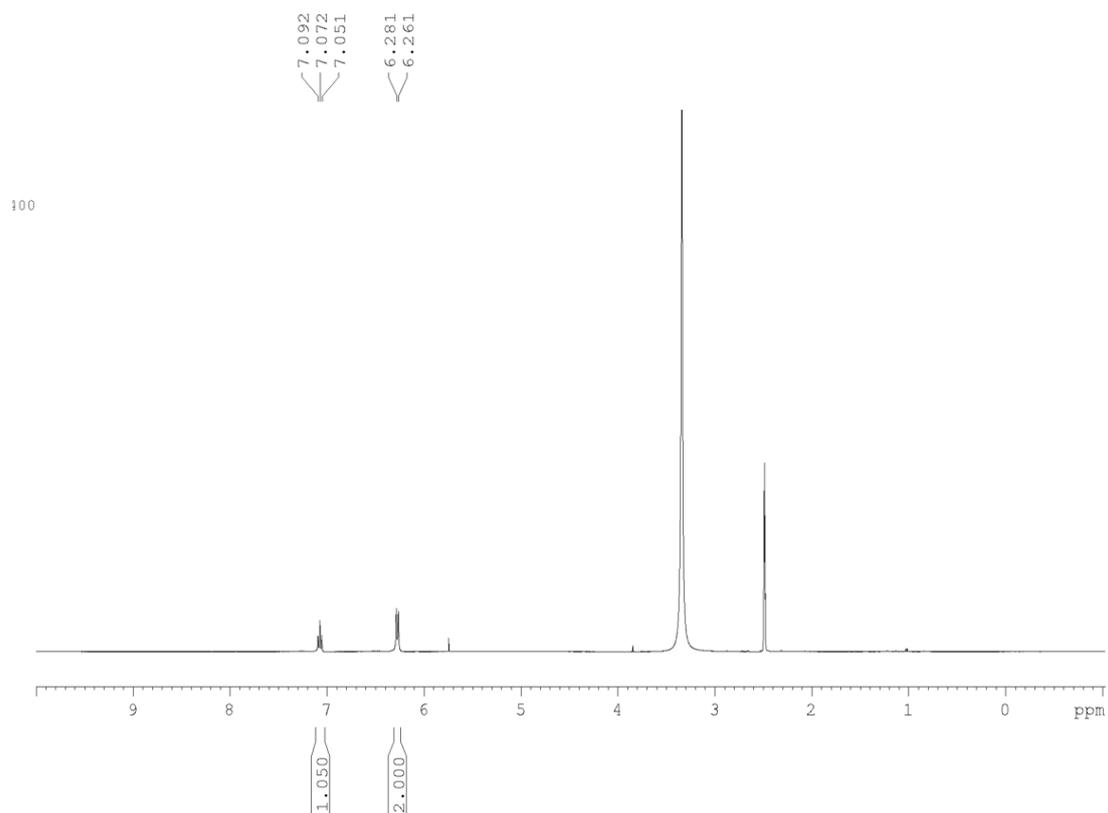
**Fig. S7** The calculated isomerization of (a) **3a**, (b) **3b** and (c) **3c** along a reaction coordinate corresponding to the N<sub>10</sub>-N<sub>12</sub>=C<sub>13</sub>-C<sub>14</sub> dihedral angle.



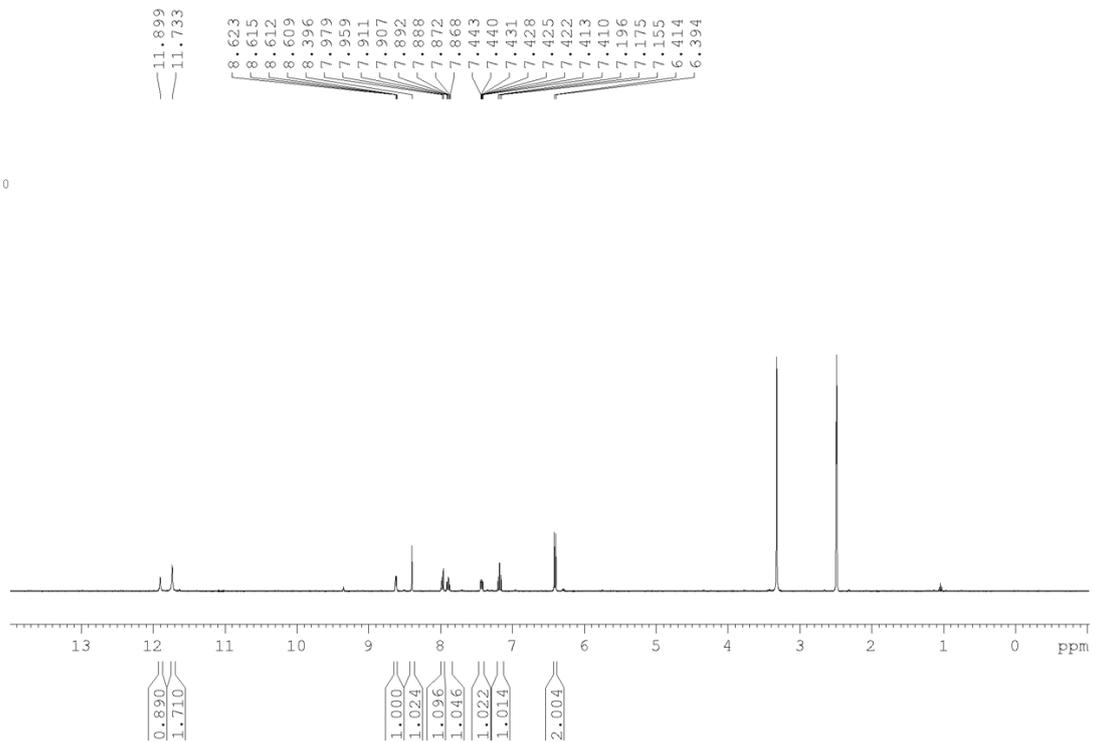
**Fig. S8** The optimized structure of **3a-Z**.



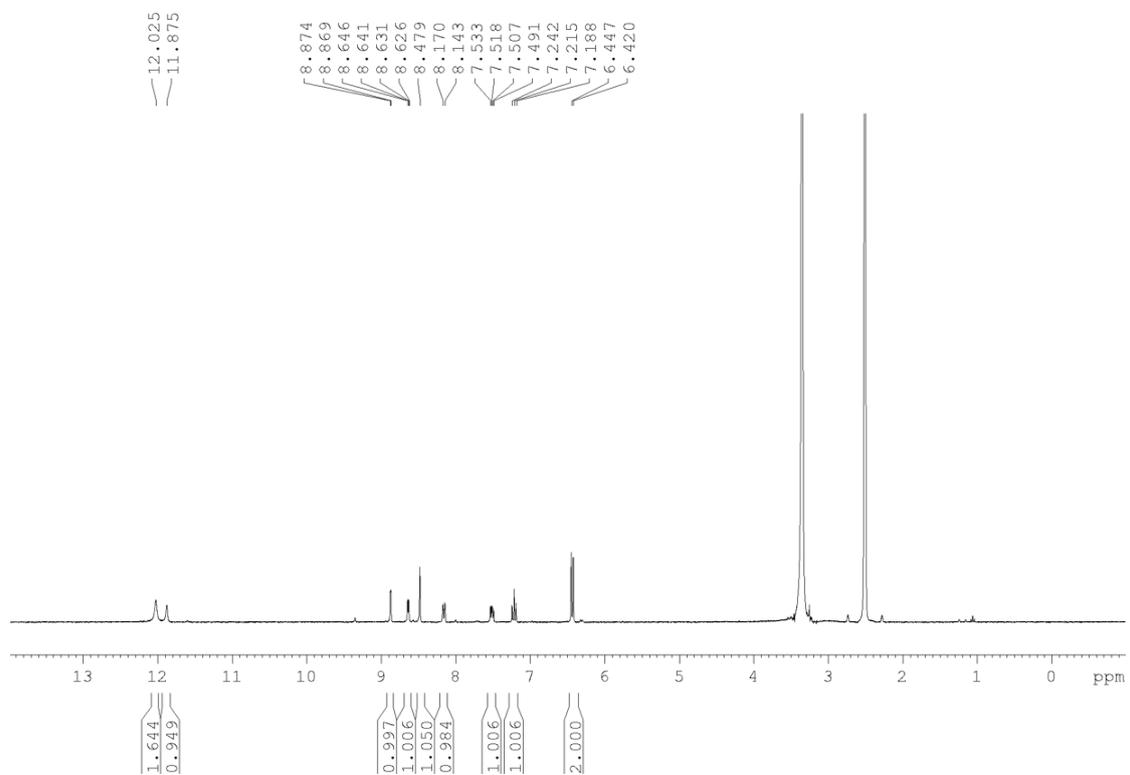
**Fig. S9** The calculated absorption spectra of **3a-E** (black) and **3a-Z** (red).



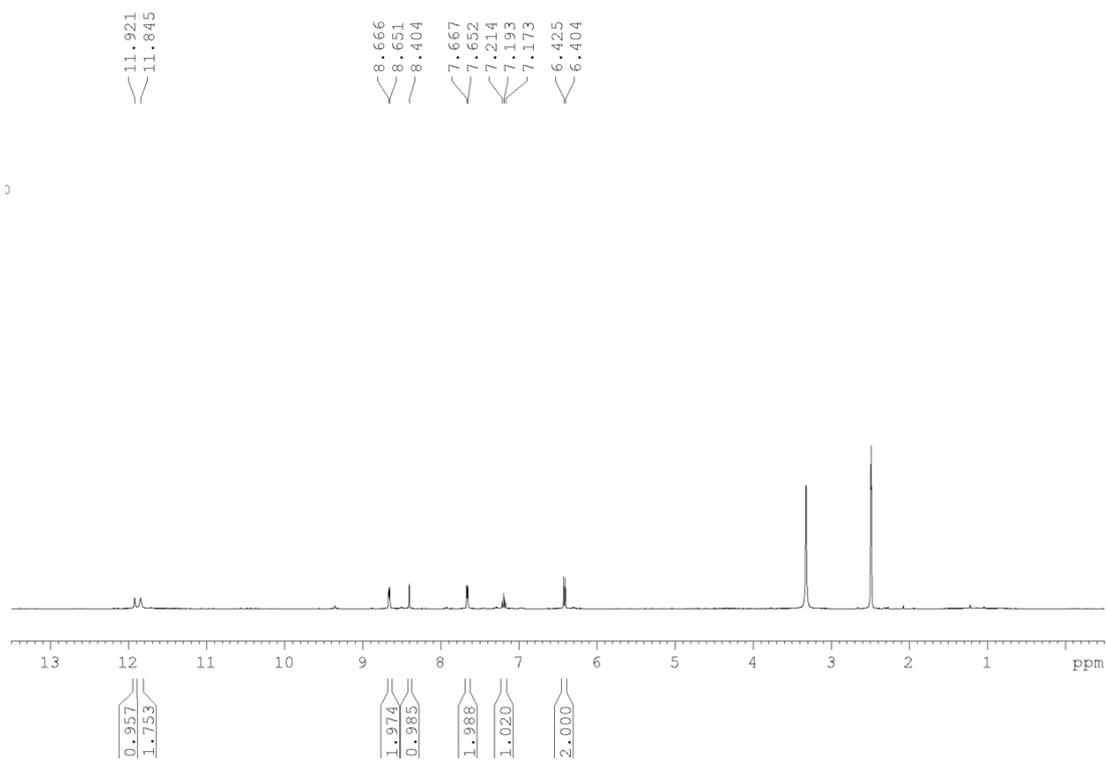
**Fig. S10**  $^1\text{H}$  NMR spectrum for **2** in  $\text{DMSO-}d_6$ .



**Fig. S11**  $^1\text{H}$  NMR spectrum for **3a** in  $\text{DMSO-}d_6$ .

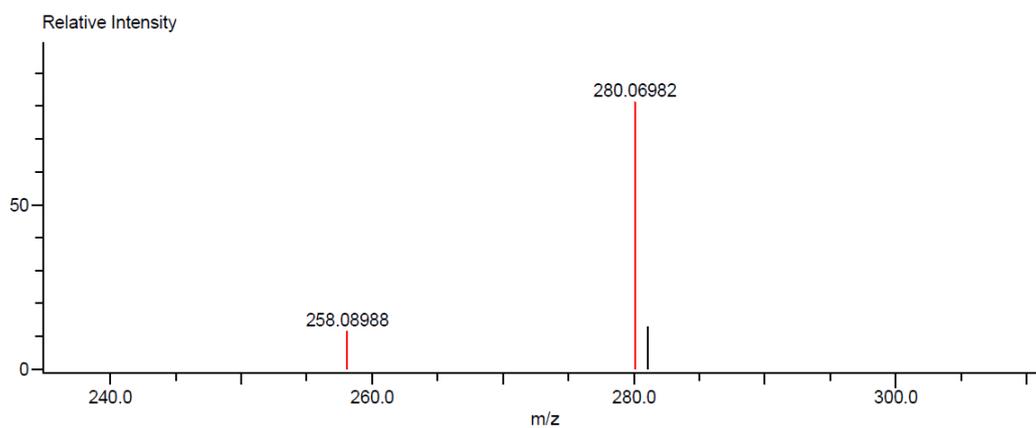


**Fig. S12**  $^1\text{H}$  NMR spectrum for **3b** in  $\text{DMSO-}d_6$ .



**Fig. S13**  $^1\text{H}$  NMR spectrum for **3c** in  $\text{DMSO-}d_6$ .

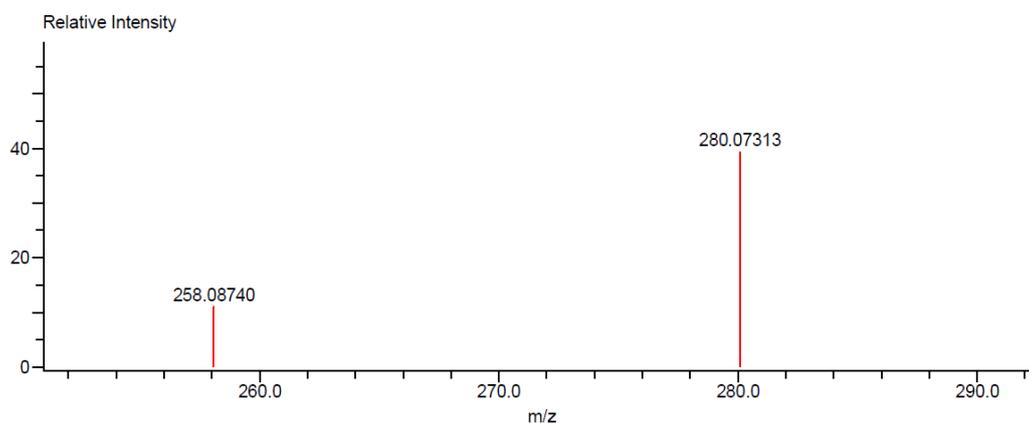
Charge number:1 Tolerance:300.00[ppm], 250.00 .. 250.... Unsaturation Number:-200.0 .. 300.0 (...)  
 Element: $^{12}\text{C}$ :13 .. 13,  $^1\text{H}$ :0 .. 12,  $^{14}\text{N}$ :3 .. 3,  $^{23}\text{Na}$ :0 .. 1,  $^{16}\text{O}$ :3 .. 3



Mass	Intensity	Calc. Mass	Mass Difference [mDa]	Mass Difference [ppm]	Possible Formula
258.08988	1032.87	258.08787	2.01	7.81	$^{12}\text{C}_{13}^{1}\text{H}_{12}^{14}\text{N}_3^{16}\text{O}_3$
280.06982	7432.50	280.06981	0.01	0.03	$^{12}\text{C}_{13}^{1}\text{H}_{11}^{14}\text{N}_3^{23}\text{Na}^{16}\text{O}_3$

**Fig. S14** HRMS spectrum for **3a**.

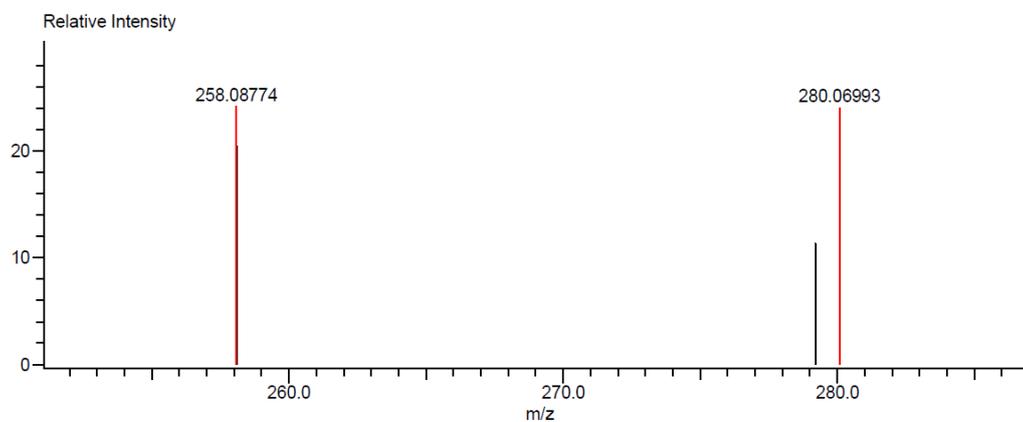
Charge number:1 Tolerance:300.00[ppm], 250.00 .. 250.... Unsaturation Number:-200.0 .. 300.0 (...)  
 Element:<sup>12</sup>C:13 .. 13, <sup>1</sup>H:0 .. 12, <sup>14</sup>N:3 .. 3, <sup>23</sup>Na:0 .. 1, <sup>16</sup>O:3 .. 3



Mass	Intensity	Calc. Mass	Mass Difference [mDa]	Mass Difference [ppm]	Possible Formula
258.08740	137.51	258.08787	-0.46	-1.80	<sup>12</sup> C <sub>13</sub> <sup>1</sup> H <sub>12</sub> <sup>14</sup> N <sub>3</sub> <sup>16</sup> O <sub>3</sub>
280.07313	491.43	280.06981	3.32	11.85	<sup>12</sup> C <sub>13</sub> <sup>1</sup> H <sub>11</sub> <sup>14</sup> N <sub>3</sub> <sup>23</sup> Na <sub>1</sub> <sup>16</sup> O <sub>3</sub>

**Fig. S15** HRMS spectrum for **3b**.

Charge number:1 Tolerance:300.00[ppm], 250.00 .. 250.... Unsaturation Number:-200.0 .. 300.0 (...)  
 Element:<sup>12</sup>C:13 .. 13, <sup>1</sup>H:0 .. 12, <sup>14</sup>N:3 .. 3, <sup>23</sup>Na:0 .. 1, <sup>16</sup>O:3 .. 3



Mass	Intensity	Calc. Mass	Mass Difference [mDa]	Mass Difference [ppm]	Possible Formula
258.08774	476.08	258.08787	-0.13	-0.50	<sup>12</sup> C <sub>13</sub> <sup>1</sup> H <sub>12</sub> <sup>14</sup> N <sub>3</sub> <sup>16</sup> O <sub>3</sub>
280.06993	473.36	280.06981	0.12	0.44	<sup>12</sup> C <sub>13</sub> <sup>1</sup> H <sub>11</sub> <sup>14</sup> N <sub>3</sub> <sup>23</sup> Na <sub>1</sub> <sup>16</sup> O <sub>3</sub>

**Fig. S16** HRMS spectrum for **3c**.