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V-Shaped Fluorophores with a 1-Methyl-4,5-bis(arylethynyl)imidazole Skeleton Displaying Solid-state Fluorescence, Acid Responsiveness, and Remarkable Fluorescence Solvatochromism

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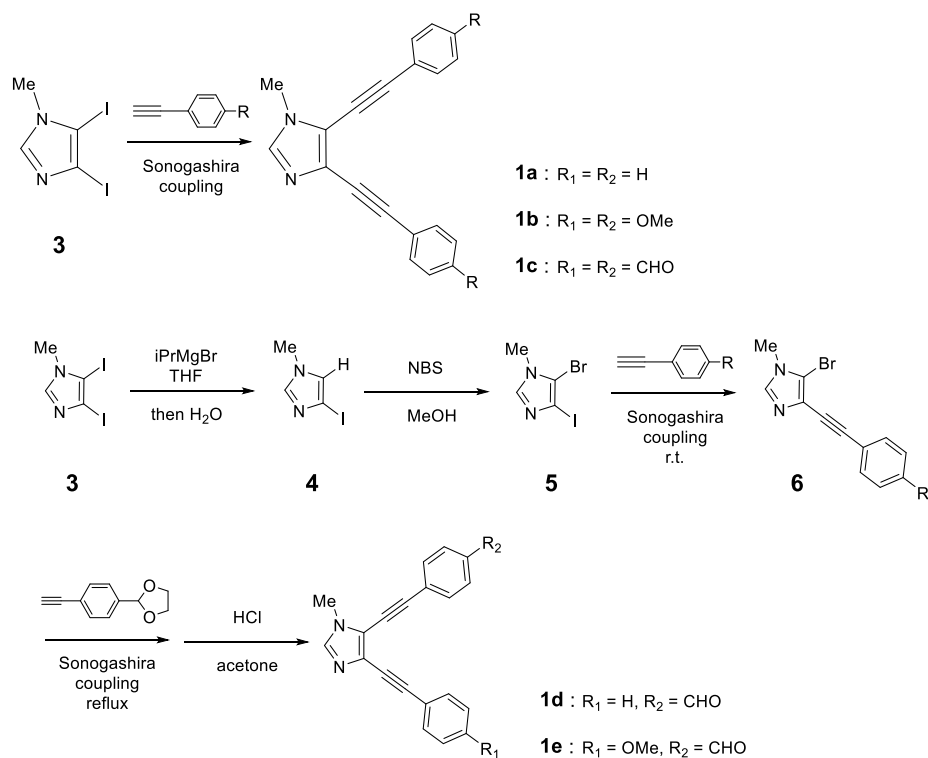
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Synthetic methods



• Preparation of 4-iodo-1-methylimidazole (**4**)

To a solution of **3** (334 mg, 1.0 mmol) in dry THF (5 mL) was added slowly isopropylmagnesium bromide (1.5 mL, 1 M in dry THF, 1.5 mmol) at 0 °C. The solution was stirred at room temperature for 1 h, then saturated ammonium chloride solution was added. The aqueous phase was extracted with CH_2Cl_2 . The combined organic phase extracts were dried (Na_2SO_4), filtered, and concentrated. The residue was purified by column chromatography over silica gel (eluent: ethyl acetate) to give **4** as a yellow oil.

Compound data

4 (184 mg, 0.88 mmol, 88%) : $^1\text{H-NMR}$ (CHCl_3 , TMS, 300 MHz) 7.34 (s, 1H), 6.98 (s, 1H), 3.69 (s, 3H) ; HRMS(ESI) calcd. for $\text{C}_4\text{H}_7\text{IN}_2^{2+}$ 209.9643, found 209.9651 [$M+2\text{H}$] $^{2+}$

- Preparation of 5-bromo-4-iodo-1-methylimidazole (**5**)

To a solution of **4** (184 mg, 0.88 mmol) in MeOH was added NBS (203 mg, 1.1 mmol). The mixture was stirred at room temperature for 15 h, then quenched with sodium sulfite solution. The aqueous phase was extracted with ethyl acetate. The combined organic phase extracts were dried (Na₂SO₄), filtered, and concentrated. The residue was purified by column chromatography over silica gel (eluent: ethyl acetate/*n*-hexane = 1 / 1) to give **5** as a white solid.

Compound data

5 (132 mg, 0.46 mmol, 52%) : ¹H-NMR (CHCl₃, TMS, 300 MHz) 7.53 (s, 1H), 3.68 (s, 3H) ; ¹³C-NMR (CHCl₃, TMS, 75 MHz) 139.75, 111.27, 86.33, 34.33 ; IR 3094, 1488, 1237, 945, 658 ; HRMS(ESI) calcd. for C₄H₅BrIN₂⁺ 286.8675, found 286.8681 [*M*+H]⁺ ; Melting point 108-109 °C

- Preparation of 5-bromo-4-(phenylethynyl)-1-methylimidazole (**6d**)

To a mixture of **5** (574 mg, 2.0 mmol), PdCl₂(PPh₃)₂ (35 mg, 0.050 mmol) and CuI (19 mg, 0.10 mmol) in THF (4 mL) and Et₃N (1.5 mL) was added dropwise ethynylbenzene (347 μL, 3.0 mmol) under nitrogen atmosphere. The reaction mixture was stirred at room temperature for 16 h. THF and Et₃N were then evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (eluent: ethyl acetate / *n*-hexane = 2 / 1) to give **6d** as a brown solid.

Compound data

6d (384 mg, 1.5 mmol, 70%) : ¹H-NMR (CHCl₃, TMS, 300 MHz) 7.57-7.55 (m, 3H), 7.36-7.32 (m, 3H), 3.65 (s, 3H) ; ¹³C-NMR (CHCl₃, TMS, 100 MHz) 139.72, 138.30, 131.48, 128.30, 128.27, 124.87, 122.86, 109.04, 91.85, 81.64, 33.42 ; IR (KBr) 3399, 3107, 3056, 2947, 2217, 1492, 757, 690 ; HRMS(ESI) calcd. for C₁₂H₁₀BrN₂⁺ 261.0022, found 260.9999 [*M*+H]⁺ ; Melting point 106-107 °C

- Preparation of 5-bromo-4-(4-methoxyphenylethynyl)-1-methylimidazole (**6e**)

Using the procedure described above for the synthesis of **6d**. A mixture of **5** (287 mg, 1.0 mmol), PdCl₂(PPh₃)₂ (35 mg, 0.050 mmol), CuI (19 mg, 0.10 mmol), 1-ethynyl-4-methoxybenzene (198 mg, 1.5 mmol) in DMF (2 mL) and Et₃N (1.5 mL) was stirred at room temperature for 16 h under nitrogen atmosphere. Purification by column chromatography over silica gel (eluent: ethyl acetate / *n*-hexane = 2 / 1) gave **6e** as a pale yellow solid.

Compound data

6e (106 mg, 0.36 mmol, 36%) : $^1\text{H-NMR}$ (CHCl_3 , TMS, 300 MHz) 7.53 (s, 1H), 7.49 (d, J = 9.0 Hz, 2H), 6.86 (d, J = 9.0 Hz, 2H), 3.82 (s, 3H), 3.64 (s, 3H) ; $^{13}\text{C-NMR}$ (CHCl_3 , TMS, 100 MHz) 159.78, 133.17, 132.23, 132.13, 128.66, 128.54, 115.21, 114.05, 92.02, 55.41 ; IR (KBr) 3133, 3007, 2947, 2840, 2536, 2210, 1604, 1502, 1285, 1170, 1027, 835, 690 ; HRMS(ESI) calcd. for $\text{C}_{13}\text{H}_{12}\text{BrN}_2\text{O}^+$ 291.0128, found 291.0120 $[\text{M}+\text{H}]^+$; Melting point 113-115 °C

• Preparation of 4,5-bis(phenylethynyl)-1-methylimidazole (**1a**)

To a mixture of **3** (167 mg, 0.50 mmol), $\text{PdCl}_2(\text{PPh}_3)_2$ (17.5 mg, 0.025 mmol) and CuI (9.5 mg, 0.050 mmol) in DMF (2 mL) and Et_3N (750 μL) was dropwise added ethynylbenzene (165 μL , 1.5 mmol) under nitrogen atmosphere. The reaction mixture was stirred at room temperature for 16 h. DMF and Et_3N were then evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (eluent: ethyl acetate / *n*-hexane = 1 / 1) to give brown solid. The white plate crystal of **1a** was prepared by recrystallization from CH_2Cl_2 / *n*-hexane.

Compound data

1a (88.4 mg, 0.31 mmol, 61%) : $^1\text{H-NMR}$ (CHCl_3 , TMS, 300 MHz) 7.58-7.55 (m, 4H), 7.46 (s, 1H), 7.40-7.32 (m, 6H), 3.75 (s, 3H) ; IR (KBr) 3400, 3109, 2948, 2194, 2211, 1490, 756, 688, 631 ; HRMS(ESI) calcd. for $\text{C}_{20}\text{H}_{15}\text{N}_2^+$ 283.1230, found 283.1254 $[\text{M}+\text{H}]^+$; elemental analysis calcd. for C : 85.08, H : 5.00, N : 9.92, found C : 85.26, H : 5.02, N : 9.83 ; Melting point 110-111 °C

• Preparation of 4,5-bis(4-methoxyphenylethynyl)-1-methylimidazole (**1b**)

Using the procedure described above for the synthesis of **1a**. A mixture of **3** (167 mg, 0.50 mmol), $\text{PdCl}_2(\text{PPh}_3)_2$ (17.5 mg, 0.025 mmol), CuI (9.5 mg, 0.050 mmol), 1-ethynyl-4-methoxybenzene (198 mg, 1.5 mmol) in DMF (2 mL) and Et_3N (750 μL) was stirred at room temperature for 18 h under nitrogen atmosphere. Purification by column chromatography over silica gel (eluent: ethyl acetate / *n*-hexane = 4 / 1) gave brown solid. The white needle crystal of **1b** was prepared by recrystallization from CH_2Cl_2 / *n*-hexane.

Compound data

1b (136 mg, 0.40 mmol, 80%) : $^1\text{H-NMR}$ (CHCl_3 , TMS, 300 MHz) 7.52-7.48 (m, 4H), 7.43 (s, 1H), 6.91-6.85 (m, 4H), 3.85 (s, 3H), 3.82 (s, 3H), 3.72 (s, 3H) ; IR (KBr) 2941, 2840,

2202, 1604, 1497, 1248, 835 ; HRMS(ESI) calcd. for $C_{22}H_{19}N_2O_2^+$ 343.1441, found 343.1414 $[M+H]^+$; elemental analysis calcd. for C : 77.17, H : 5.30, N : 8.18, found C : 77.17, H : 5.28, N : 8.17 ; Melting point 102-104 °C

- Preparation of 4,5-bis(4-formylphenylethynyl)-1-methylimidazole (**1c**)

Using the procedure described above for the synthesis of **1a**. A mixture of **3** (60 mg, 0.18 mmol), $PdCl_2(PPh_3)_2$ (6.3 mg, 0.0090 mmol), CuI (3.4 mg, 0.018 mmol), 4-ethynylbenzaldehyde (70 mg, 0.54 mmol) in THF (1 mL) and Et_3N (300 μ L) was stirred at room temperature for 16 h under nitrogen atmosphere. Purification by column chromatography over silica gel (eluent: ethyl acetate / *n*-hexane = 2 / 1) gave **1c** as a yellow solid.

Compound data

1c (25 mg, 0.075 mmol, 42%) : 1H -NMR ($CHCl_3$, TMS, 300 MHz) 10.05 (s, 1H), 10.03 (s, 1H), 7.93-7.86 (m, 4H), 7.73-7.70 (m, 4H), 7.53 (s, 1H), 3.79 (s, 3H) ; ^{13}C -NMR ($CHCl_3$, TMS, 100 MHz) 191.54, 191.34, 139.10, 136.01, 135.58, 132.23, 132.13, 131.95, 129.83, 129.70, 129.57, 129.28, 128.69, 128.56, 128.25, 120.24, 99.39, 92.17, 86.57, 80.19, 32.89 ; IR (KBr) 3124, 2848, 2746, 2205, 1691, 1599, 1206, 1160, 806, 788 ; HRMS(ESI) calcd. for $C_{22}H_{15}N_2O_2^+$ 339.1128, found 339.1112 $[M+H]^+$; elemental analysis calcd. for C : 78.09, H : 4.17, N : 8.28, found C : 77.74, H : 4.14, N : 8.15 ; Melting point 203-206 °C

- Preparation of 4-phenylethynyl-5-(4-formylphenylethynyl)-1-methylimidazole (**1d**)

To a mixture of **6d** (157 mg, 0.60 mmol), $PdCl_2(PPh_3)_2$ (21 mg, 0.030 mmol) and CuI (11 mg, 0.060 mmol) in DMF (2.4 mL) and Et_3N (900 μ L) was added 2-(4-ethynylphenyl)-1,3-dioxolane (157 mg, 0.90 mmol) under nitrogen atmosphere. The reaction mixture was stirred at 90 °C for 16 h. After the reaction, DMF and Et_3N were then evaporated under reduced pressure. The residue was purified by column chromatography over silica gel (eluent: toluene / acetone = 4 / 1) to give yellow solid. Further purification was performed by column chromatography over silica gel (eluent: CH_2Cl_2). The obtained solid was dissolved in acetone and hydrochloric acid (1*N*) was added. The mixture was stirred at room temperature for 30 minutes. The aqueous phase was extracted with CH_2Cl_2 . The combined organic phase extracts were dried (Na_2SO_4), filtered, and concentrated. The residue was purified by column chromatography over silica gel (eluent: ethyl acetate / *n*-hexane = 2 / 1) to give **1d** as yellow solid. The yellow plate crystal of **1d** was prepared by recrystallization from CH_2Cl_2 / *n*-hexane.

Compound data

1d (38 mg, 0.12 mmol, 20%) : $^1\text{H-NMR}$ (CHCl_3 , TMS, 300 MHz) 10.04 (s, 1H), 7.90 (d, J = 8.7 Hz, 2H), 7.70 (d, J = 8.7 Hz, 2H), 7.60-7.57 (m, 2H), 7.50 (s, 1H), 7.37-7.35 (m, 3H), 3.78 (s, 3H) ; IR (KBr) 3093, 2824, 2728, 2206, 2189, 1698, 1600, 1205, 825, 755 ; HRMS(ESI) calcd. for $\text{C}_{21}\text{H}_{15}\text{N}_2\text{O}^+$ 311.1179, found 311.1166 $[\text{M}+\text{H}]^+$; elemental analysis calcd. for C : 81.27, H : 4.55, N : 9.03, found C : 81.21, H : 4.51, N : 8.84 ; Melting point 162-163 °C

• Preparation of 4-(4-methoxyphenylethynyl)-5-(4-formylphenylethynyl)-1-methylimidazole (**1e**)

Using the procedure described above for the synthesis of **1d**. A mixture of **6e** (85 mg, 0.29 mmol), $\text{PdCl}_2(\text{PPh}_3)_2$ (10 mg, 0.015 mmol), CuI (5.5 mg, 0.029 mmol), 2-(4-ethynylphenyl)-1,3-dioxolane (76 mg, 0.15 mmol) in DMF (2 mL) and Et_3N (750 μL) was stirred at 90 °C for 18 h under nitrogen atmosphere. Purification by column chromatography over silica gel (eluent: ethyl acetate / *n*-hexane = 4 / 1) gave brown oil. After reaction using hydrochloric acid, purification by column chromatography over silica gel (eluent: ethyl acetate / *n*-hexane = 2 / 1) gave **1e** as a yellow solid. The yellow plate crystal of **1e** was prepared by recrystallization from CH_2Cl_2 / *n*-hexane.

Compound data

1e (62 mg, 0.18 mmol, 63%) : $^1\text{H-NMR}$ (CHCl_3 , TMS, 300 MHz) 10.04 (s, 1H), 7.90 (d, J = 8.1 Hz, 2H), 7.69 (d, J = 8.1 Hz, 2H), 7.52 (d, J = 8.7), 7.49 (s, 1H), 6.89 (d, J = 8.7 Hz, 2H), 3.84 (s, 3H), 3.77 (s, 3H); IR (KBr) 3121, 2839, 2215, 2196, 1693, 1601, 1248, 1207, 1164, 1031, 794 ; HRMS(ESI) calcd. for $\text{C}_{22}\text{H}_{17}\text{N}_2\text{O}_2^+$ 341.1285, found 341.1255 $[\text{M}+\text{H}]^+$; elemental analysis calcd. for C : 77.63, H : 4.74, N : 8.23, found C : 77.60, H : 4.70, N : 8.10 ; Melting point 146-148 °C

General procedure of **1**-TFA salt

To a solution of **1** in dichloromethane was added dropwise excess amount of TFA. Slow evaporation gave solid of **1**-TFA after 1 day. A single crystal was prepared by recrystallization from CH_2Cl_2 / *n*-hexane.

Compound data

1a-TFA : $^1\text{H-NMR}$ (CHCl_3 , TMS, 400 MHz) 8.14 (s, 1H), 7.60-7.57 (m, 4H), 7.44-7.41 (m, 3H), 7.40-7.35 (m, 3H), 3.85 (s, 3H) ; $^{13}\text{C-NMR}$ (CHCl_3 , TMS, 100 MHz) 161.96, 161.59, 137.02, 131.97, 131.87, 129.97, 129.47, 128.77, 128.56, 123.31, 121.68, 121.14, 120.67,

101.60, 96.96, 74.25, 33.98 ; IR (KBr) 3113, 3047, 2359, 2218, 1683, 1193, 1132, 753, 687 ; HRMS(ESI) calcd. for $C_{20}H_{15}N_2^+$ 283.1230, found 283.1227 [*M*] ; Melting point 105-107 °C

Compound data

1d-TFA : 1H -NMR ($CHCl_3$, TMS, 400 MHz) 10.1 (s, 1H), 8.13 (s, 1H), 7.93 (d, *J* = 8.4 Hz, 2H), 7.73 (d, *J* = 8.4 Hz, 2H), 7.40-7.37 (m, 2H), 7.61-7.58 (m, 2H), 3.87 (s, 3H) ; ^{13}C -NMR ($CHCl_3$, TMS, 100 MHz) 191.24, 160.66, 136.41, 133.61, 132.62, 132.29, 132.27, 132.20, 131.36, 129.81, 128.87, 128.76, 127.26, 114.28, 113.60, 97.48, 55.47 ; IR (KBr) 3422, 3129, 2362, 1702, 1604, 1193, 1127 ; HRMS(ESI) calcd. for $C_{21}H_{15}N_2O^+$ 311.1179, found 311.1181 [*M*] ; Melting point 155-157 °C

Compound data

1e-TFA : 1H -NMR ($CHCl_3$, TMS, 400 MHz) 10.1(s, 1H), 8.28 (s, 1H), 7.93 (d, *J* = 7.8 Hz, 2H), 7.72 (d, *J* = 7.8 Hz, 2H), 7.54 (d, *J* = 8.2 Hz, 2H), 6.91 (d, *J* = 8.2 Hz, 2H), 3.87 (s, 3H) ; ^{13}C -NMR ($CHCl_3$, TMS, 100 MHz) 191.23, 160.72, 136.46, 133.66, 132.66, 132.27, 132.20, 129.82, 128.89, 128.77, 127.19, 114.29, 113.45, 97.83, 55.47, 53.53 ; IR (KBr) 3127, 2212, 1694, 1602, 1514, 1255, 1207, 1167 ; HRMS(ESI) calcd. for $C_{22}H_{17}N_2O_2^+$ 341.1285, found 341.1283 [*M*] ; Melting point 133-135 °C

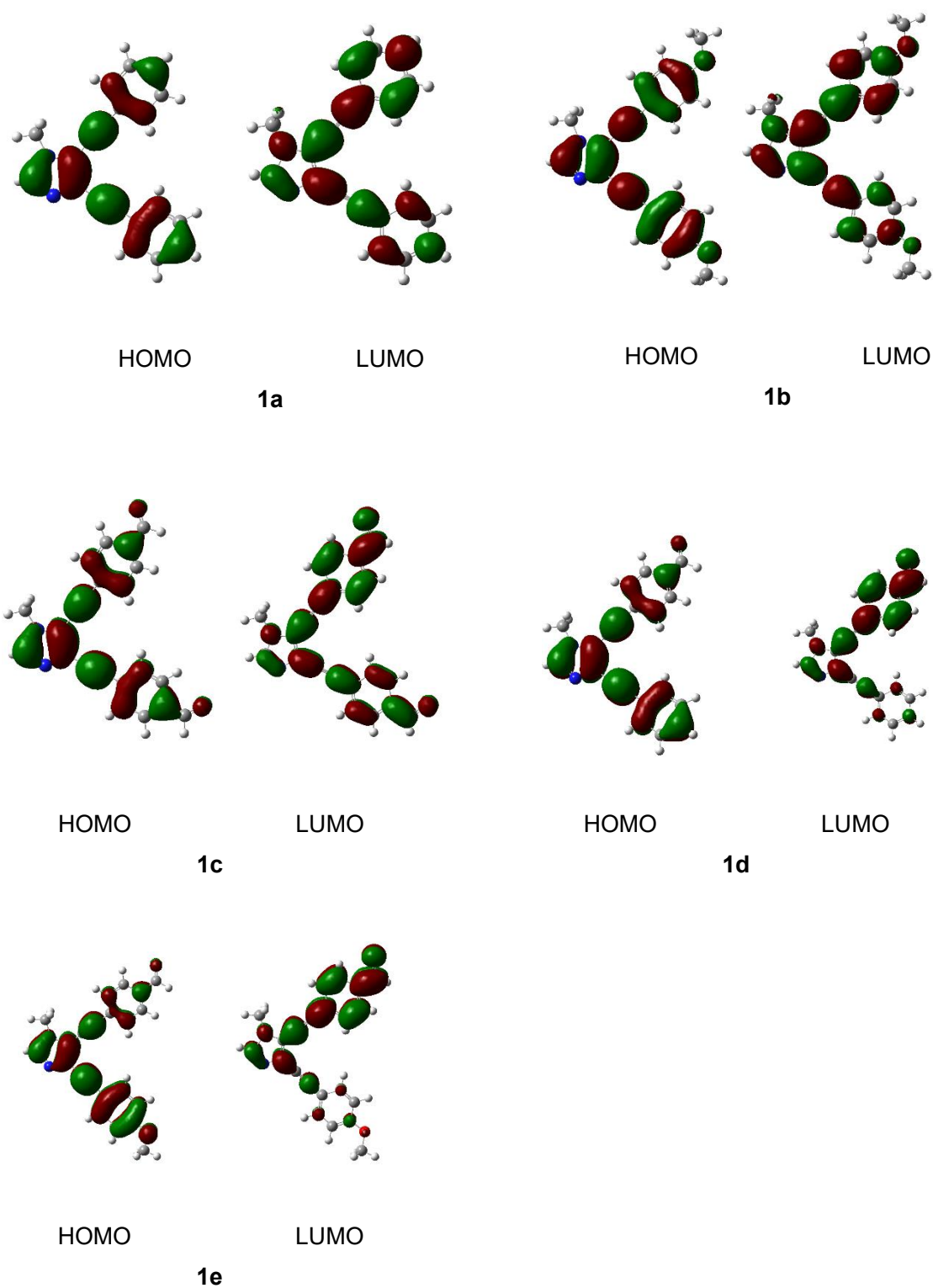


Figure S1. Electron distribution of frontier orbitals of **1a-e**.

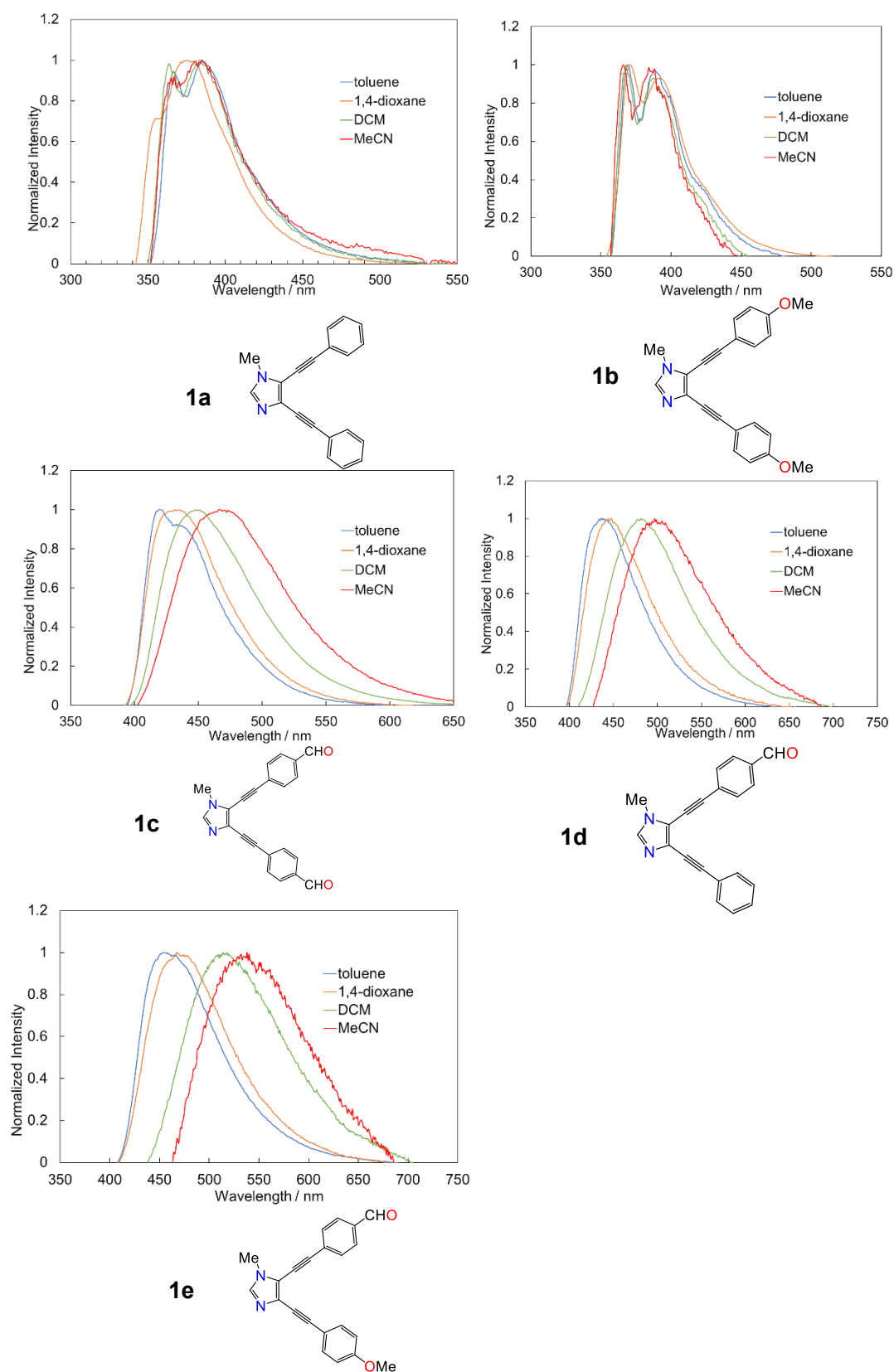


Figure S2. Fluorescence spectra of **1a-e** in various solvents

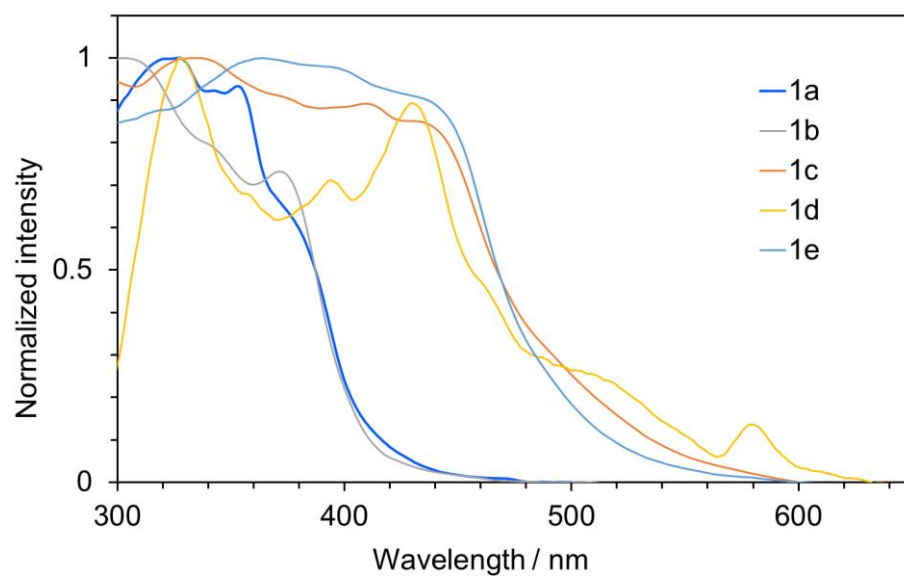


Figure S3. Diffuse reflection spectra of **1a-e** in the solid-state.

Table S1. Optimized atom coordinates of **1a**.

Total energy: -727.7954746 A.U.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-5.827955	-1.021816	-0.000004
2	6	0	-5.501515	-2.377774	0.000082
3	6	0	-4.162028	-2.765784	0.000140
4	6	0	-3.154303	-1.809203	0.000111
5	6	0	-3.471782	-0.438250	0.000024
6	6	0	-4.827098	-0.058163	-0.000034
7	6	0	-2.446952	0.546292	-0.000006
8	6	0	-1.584001	1.395179	-0.000026
9	6	0	2.329702	-2.465997	-0.000281
10	6	0	-0.579907	2.383694	-0.000038
11	6	0	0.805847	2.176371	-0.000072
12	6	0	1.572078	1.004932	-0.000103
13	6	0	2.982255	-1.218503	-0.000036
14	6	0	2.230053	-0.013767	-0.000117
15	6	0	3.065609	-3.644143	-0.000257
16	6	0	4.459731	-3.605259	0.000008
17	6	0	5.116447	-2.375023	0.000250
18	6	0	4.389614	-1.191087	0.000228
19	7	0	1.358251	3.449569	-0.000071
20	6	0	0.314426	4.326992	-0.000038
21	7	0	-0.854483	3.734240	-0.000017
22	1	0	-6.868593	-0.713428	-0.000050
23	1	0	-6.286150	-3.127412	0.000105
24	1	0	-3.901601	-3.819516	0.000207
25	1	0	-5.078820	0.996453	-0.000101
26	1	0	2.548657	-4.598357	-0.000448
27	1	0	5.030763	-4.527850	0.000025
28	1	0	6.201107	-2.337966	0.000456
29	1	0	4.901553	-0.235022	0.000417
30	1	0	0.472564	5.396456	-0.000030
31	6	0	2.774833	3.769497	0.000143
32	1	0	3.261082	3.360176	-0.887223
33	1	0	3.260553	3.361684	0.888504
34	1	0	2.888505	4.853029	-0.000756
35	1	0	-2.112244	-2.109259	0.000156
36	1	0	1.245931	-2.494782	-0.000491

Table S2. Optimized atom coordinates of **1b**.

Total energy: -1109.2513894 A.U.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.000000	0.000000
2	6	0	0.000000	0.000000	1.398405
3	6	0	1.220974	0.000000	2.087057
4	6	0	2.416140	0.000000	1.392668
5	6	0	2.437038	0.000000	-0.017689
6	6	0	1.206807	0.000000	-0.692860
7	6	0	3.662831	0.000000	-0.734330
8	6	0	4.701330	0.000009	-1.357019
9	6	0	7.588664	0.000388	3.320469
10	6	0	5.910104	0.000084	-2.081255
11	6	0	7.208035	0.000157	-1.554187
12	6	0	7.678603	0.000177	-0.235415
13	6	0	8.518736	0.000369	2.258872
14	6	0	8.077614	0.000211	0.910147
15	6	0	8.011929	0.000479	4.635579
16	6	0	9.380222	0.000553	4.942468
17	6	0	10.316792	0.000535	3.903825
18	6	0	9.884409	0.000444	2.581215
19	7	0	8.044666	0.000213	-2.662348
20	6	0	7.236331	0.000172	-3.760783
21	7	0	5.961199	0.000013	-3.459627
22	1	0	-0.926887	0.000000	-0.559612
23	1	0	1.201056	0.000000	3.171209
24	1	0	1.200333	0.000000	-1.777107
25	1	0	7.300441	0.000494	5.453682
26	1	0	11.379795	0.000591	4.109189
27	1	0	10.618221	0.000430	1.782646
28	1	0	7.641974	0.000201	-4.762941
29	6	0	9.496166	0.000496	-2.639618
30	1	0	9.872838	-0.886734	-2.127190
31	1	0	9.872580	0.888637	-2.128564
32	1	0	9.862006	-0.000256	-3.665905
33	1	0	3.355165	0.000000	1.934831
34	1	0	6.528565	0.000331	3.093765
35	6	0	11.057200	0.000717	6.647138
36	1	0	11.062032	0.000778	7.735373
37	1	0	11.574872	-0.893310	6.283408
38	1	0	11.574787	0.894756	6.283314
39	6	0	-2.386923	0.000000	1.539804
40	1	0	-3.121122	0.000000	2.343299
41	1	0	-2.527080	0.893757	0.922299
42	1	0	-2.527081	-0.893755	0.922297
43	8	0	9.689424	0.000639	6.267153
44	8	0	-1.118281	0.000000	2.175627

Table S3. Optimized atom coordinates of **1c**.

Total energy: -1106.858041 A.U.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.000000	0.000000
2	6	0	0.000000	0.000000	1.399379
3	6	0	1.221219	0.000000	2.088277
4	6	0	2.415819	0.000000	1.392504
5	6	0	2.422242	0.000000	-0.019504
6	6	0	1.192926	0.000000	-0.706776
7	6	0	3.644907	0.000000	-0.737151
8	6	0	4.681769	-0.000001	-1.362658
9	6	0	7.587074	-0.000885	3.295375
10	6	0	5.884657	-0.000002	-2.093489
11	6	0	7.186753	-0.000021	-1.571933
12	6	0	7.657990	-0.000042	-0.255810
13	6	0	8.510931	-0.000571	2.232149
14	6	0	8.058201	-0.000110	0.889489
15	6	0	8.035522	-0.001327	4.607307
16	6	0	9.405286	-0.001471	4.892624
17	6	0	10.328875	-0.001162	3.837757
18	6	0	9.892416	-0.000720	2.526135
19	7	0	8.016514	-0.000014	-2.683054
20	6	0	7.202372	0.000076	-3.776487
21	7	0	5.927939	0.000126	-3.469267
22	1	0	-0.944619	0.000000	-0.537096
23	1	0	1.204169	0.000000	3.172556
24	1	0	1.193159	0.000000	-1.790569
25	1	0	7.316817	-0.001566	5.422174
26	1	0	11.386863	-0.001279	4.075455
27	1	0	10.607036	-0.000482	1.710806
28	1	0	7.601903	0.000100	-4.780908
29	6	0	9.470239	0.000084	-2.670024
30	1	0	9.848456	-0.887952	-2.161047
31	1	0	9.848389	0.888698	-2.161995
32	1	0	9.828282	-0.000460	-3.698632
33	1	0	3.361227	0.000000	1.923029
34	1	0	6.525982	-0.000765	3.074999
35	6	0	9.874060	-0.001944	6.295564
36	8	0	11.035049	-0.002097	6.635378
37	1	0	9.063646	-0.002161	7.056744
38	6	0	-1.277574	0.000000	2.142994
39	8	0	-1.377154	0.000000	3.348986
40	1	0	-2.187179	0.000000	1.503093

Table S4. Optimized atom coordinates of **1d**.

Total energy: -993.5028143 A.U.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.000000	0.000000
2	6	0	0.000000	0.000000	1.394688
3	6	0	1.210918	0.000000	2.086374
4	6	0	2.414686	0.000000	1.392649
5	6	0	2.426690	0.000000	-0.014375
6	6	0	1.198729	0.000000	-0.702254
7	6	0	3.653356	0.000000	-0.731906
8	6	0	4.689602	0.000001	-1.357547
9	6	0	7.596183	-0.000433	3.297222
10	6	0	5.893641	0.000005	-2.088675
11	6	0	7.196134	-0.000097	-1.568390
12	6	0	7.668524	-0.000224	-0.253346
13	6	0	8.521118	-0.000398	2.234331
14	6	0	8.069756	-0.000305	0.891960
15	6	0	8.042981	-0.000404	4.609383
16	6	0	9.412445	-0.000293	4.897372
17	6	0	10.337126	-0.000210	3.843144
18	6	0	9.902370	-0.000238	2.531174
19	7	0	8.026776	-0.000045	-2.681315
20	6	0	7.212289	0.000080	-3.772897
21	7	0	5.937165	0.000113	-3.464482
22	1	0	-0.939426	0.000000	-0.543403
23	1	0	1.216641	0.000000	3.171709
24	1	0	1.201372	0.000000	-1.786445
25	1	0	7.323155	-0.000470	5.423427
26	1	0	11.394958	-0.000124	4.081888
27	1	0	10.618141	-0.000174	1.716783
28	1	0	7.610607	0.000142	-4.777897
29	6	0	9.479733	0.000113	-2.668584
30	1	0	9.858821	-0.887680	-2.159395
31	1	0	9.858710	0.888741	-2.160752
32	1	0	9.838070	-0.000664	-3.697291
33	1	0	3.357402	0.000000	1.928554
34	1	0	6.535402	-0.000524	3.075123
35	6	0	9.877915	-0.000263	6.300136
36	8	0	11.038050	-0.000249	6.644630
37	1	0	9.064826	-0.000406	7.058989
38	1	0	-0.938440	0.000000	1.939492

Table S5. Optimized atom coordinates of **1e**.

Total energy: -1108.0549967 A.U.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.000000	0.000000	0.000000
2	6	0	0.000000	0.000000	1.398824
3	6	0	1.221031	0.000000	2.088262
4	6	0	2.415970	0.000000	1.394295
5	6	0	2.436236	0.000000	-0.016067
6	6	0	1.206497	0.000000	-0.692406
7	6	0	3.661195	0.000000	-0.733026
8	6	0	4.698246	0.000004	-1.358220
9	6	0	7.599168	-0.000238	3.298055
10	6	0	5.902667	0.000022	-2.088021
11	6	0	7.205254	-0.000017	-1.567213
12	6	0	7.677061	-0.000082	-0.252405
13	6	0	8.525889	-0.000137	2.236400
14	6	0	8.077014	-0.000130	0.893634
15	6	0	8.043536	-0.000276	4.610981
16	6	0	9.412564	-0.000215	4.901894
17	6	0	10.339024	-0.000116	3.849054
18	6	0	9.906734	-0.000077	2.536318
19	7	0	8.036999	0.000023	-2.680529
20	6	0	7.222864	0.000082	-3.771270
21	7	0	5.947385	0.000016	-3.463836
22	1	0	-0.926892	0.000000	-0.559390
23	1	0	1.200295	0.000000	3.172301
24	1	0	1.200511	0.000000	-1.776598
25	1	0	7.322106	-0.000354	5.423608
26	1	0	11.396395	-0.000070	4.089740
27	1	0	10.624026	0.000000	1.723280
28	1	0	7.621232	0.000122	-4.776298
29	6	0	9.489673	0.000179	-2.667093
30	1	0	9.868832	-0.887527	-2.157794
31	1	0	9.868711	0.888619	-2.158969
32	1	0	9.848436	-0.000488	-3.695628
33	1	0	3.354927	0.000000	1.936552
34	1	0	6.538837	-0.000286	3.073930
35	6	0	-2.387243	0.000000	1.540547
36	1	0	-3.119625	0.000000	2.345445
37	1	0	-2.527587	0.893969	0.923806
38	1	0	-2.527587	-0.893967	0.923802
39	8	0	-1.116829	0.000000	2.175004
40	6	0	9.875196	-0.000257	6.305062
41	8	0	11.034450	-0.000282	6.652780
42	1	0	9.060621	-0.000409	7.062326

Table S6. Crystal data of **1a-e**.

	1a	1b	1c	1d	1e
Formula	C ₂₀ H ₁₄ N ₂	C ₂₂ H ₁₈ N ₂ O ₂	C ₂₂ H ₁₄ N ₂ O ₂	C ₂₁ H ₁₄ N ₂ O	C ₂₂ H ₁₆ N ₂ O ₂
F. W.	282.33	342.38	338.35	310.34	340.37
T/ K	300	300	300	302	302
Crystal system	Triclinic	Monoclinic	Monoclinic	Monoclinic	Triclinic
Space group	P1	P2 ₁ /n	P2 ₁ /c	P2 ₁ /c	P-1
a/ Å	9.6262(10)	10.4483(9)	9.5584(3)	10.8306(3)	9.3320(4)
b/ Å	12.6518(13)	7.9356(7)	22.8679(6)	16.0766(5)	9.8300(4)
c/ Å	13.3053(14)	22.7149(18)	8.4503(2)	11.0441(3)	11.3679(5)
$\alpha(^{\circ})$	87.690(4)	90	90	90	97.151(2)
$\beta(^{\circ})$	81.189(4)	95.017(3)	106.7870(10)	119.2370(10)	114.1460(10)
$\gamma(^{\circ})$	80.098(4)	90	90	90	102.008(2)
V/ Å ³	1577.3(3)	1876.2(3)	1768.36(8)	1678.01(8)	904.73(7)
Z	4	4	4	4	2
$\rho_{\text{calcd}} / \text{g cm}^{-3}$	1.189	1.212	1.271	1.228	1.249
R ₁ (I > 2 σ (I))	0.0591	0.0982	0.0425	0.0433	0.0479
wR ₂	0.175	0.295	0.1656	0.1382	0.1572
GOF	0.790	1.438	1.454	0.779	1.723

Table S7. Result of TD-DFT calculation of the contact c dimer in the crystal of **1b**

Excited State 1: Singlet-A 3.6328 eV 341.29 nm f=0.5180 <S**2>=0.000
179 -> 181 0.67253
180 -> 181 -0.18648

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -2217.78381399

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.6929 eV 335.73 nm f=0.0627
<S**2>=0.000
179 -> 181 0.17014
180 -> 181 0.64087
180 -> 182 0.22275

Excited State 3: Singlet-A 3.7260 eV 332.75 nm f=0.5760
<S**2>=0.000
180 -> 181 -0.21137
180 -> 182 0.65811

Excited State 4: Singlet-A 3.7616 eV 329.61 nm f=0.0134
<S**2>=0.000
179 -> 182 0.69801

Excited State 5: Singlet-A 4.2276 eV 293.27 nm f=0.1417
<S**2>=0.000
177 -> 181 0.18587
179 -> 183 0.62430
179 -> 184 0.15727
180 -> 183 -0.14173

Table S8. Result of TD-DFT calculation of the contact c dimer in the crystal of **1c**

Excited State 1: Singlet-A 2.9220 eV 424.31 nm f=0.0283 <S**2>=0.000
176 -> 177 0.67944
176 -> 179 -0.18590

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -2213.08123339

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.1270 eV 396.50 nm f=0.0057
<S**2>=0.000
175 -> 177 0.51348
176 -> 178 -0.48164

Excited State 3: Singlet-A 3.1455 eV 394.17 nm f=1.0712
<S**2>=0.000
175 -> 177 0.47857
176 -> 178 0.51235

Excited State 4: Singlet-A 3.3728 eV 367.61 nm f=0.0380
<S**2>=0.000
175 -> 178 0.68034
176 -> 179 -0.17023

Excited State 5: Singlet-A 3.3764 eV 367.21 nm f=0.0015
<S**2>=0.000
175 -> 178 0.17576
176 -> 177 0.18449
176 -> 179 0.64752

Table S9. Result of TD-DFT calculation of the contact c trimer in the crystal of **1b**

Excitation energies and oscillator strengths:

Excited State	1:	Singlet-A	3.6303 eV	341.52 nm	f=0.6215
<S**2>=0.000					
	268 -> 271	0.67627			
	269 -> 271	-0.13258			
	270 -> 271	0.10799			
This state for optimization and/or second-order correction.					
Total Energy, E(TD-HF/TD-KS) = -3326.74468713					
Copying the excited state density for this state as the 1-particle RhoCl density.					
Excited State	2:	Singlet-A	3.6762 eV	337.26 nm	f=0.2828
<S**2>=0.000					
	268 -> 271	0.11677			
	269 -> 271	0.32152			
	269 -> 272	0.38377			
	270 -> 271	-0.30002			
	270 -> 272	-0.35045			
Excited State	3:	Singlet-A	3.7012 eV	334.99 nm	f=0.2285
<S**2>=0.000					
	268 -> 271	0.12332			
	269 -> 271	0.37958			
	269 -> 272	-0.32986			
	270 -> 271	-0.35052			
	270 -> 272	0.30784			
Excited State	4:	Singlet-A	3.7187 eV	333.40 nm	f=0.5389
<S**2>=0.000					
	269 -> 272	-0.23690			
	269 -> 273	0.31150			
	270 -> 272	-0.32197			
	270 -> 273	0.47809			
Excited State	5:	Singlet-A	3.7444 eV	331.12 nm	f=0.0065
<S**2>=0.000					
	268 -> 272	0.66033			
	268 -> 273	0.20764			

Table S10. Result of TD-DFT calculation of the contact c trimer in the crystal of **1c**

Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 2.9284 eV 423.38 nm f=0.0322
<S**2>=0.000
264 -> 265 -0.32293
264 -> 266 0.59991
264 -> 269 -0.16841

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -3319.67579720

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.0120 eV 411.64 nm f=0.0343
<S**2>=0.000
263 -> 265 0.61262
263 -> 268 -0.16300
264 -> 265 -0.25463
264 -> 266 -0.15310

Excited State 3: Singlet-A 3.0521 eV 406.23 nm f=0.0387
<S**2>=0.000
263 -> 265 0.29176
264 -> 265 0.57206
264 -> 266 0.27272

Excited State 4: Singlet-A 3.1315 eV 395.93 nm f=0.5169
<S**2>=0.000
262 -> 265 0.55333
263 -> 266 -0.23676
264 -> 267 0.36137

Excited State 5: Singlet-A 3.1408 eV 394.75 nm f=0.0062
<S**2>=0.000
262 -> 265 -0.41594
263 -> 266 -0.14166
264 -> 267 0.54660

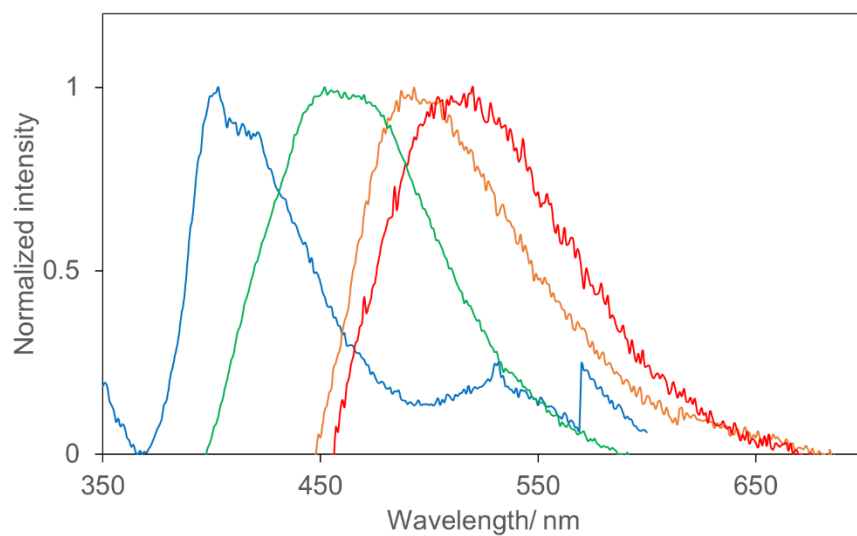


Figure S4. Solid-state fluorescence spectra of **1a**, **b**, **d**, **e** after exposure to HCl vapor. (Blue: **1a** ($\Phi = 0.01$), Green: **1b** ($\Phi = 0.03$), Orange: **1d** ($\Phi < 0.01$), and Red: **1e** ($\Phi < 0.01$)) Compound **1c** showed no fluorescence.

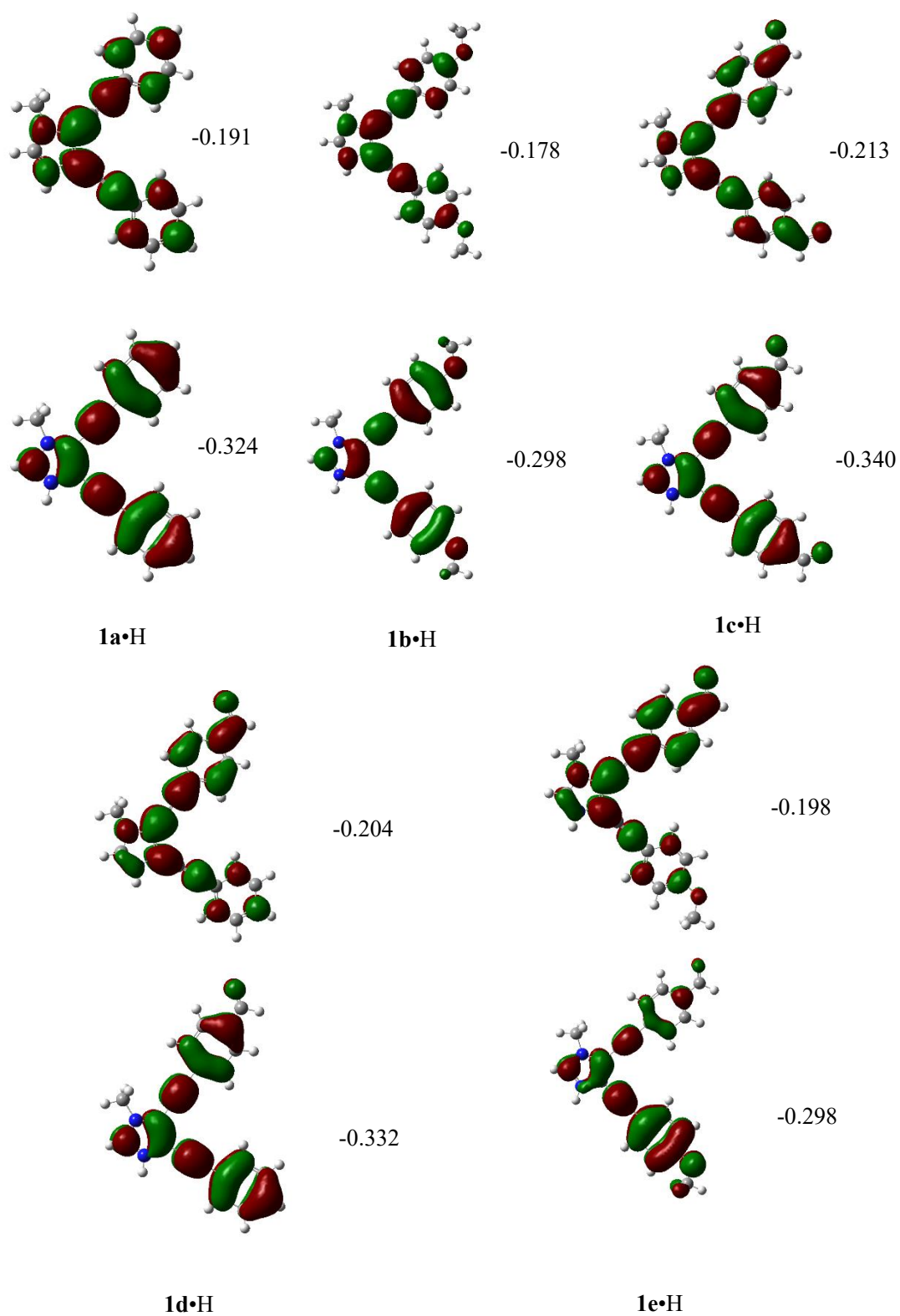


Figure S5. Electron distribution of **1a-e•H** cation of HOMO and LUMO.

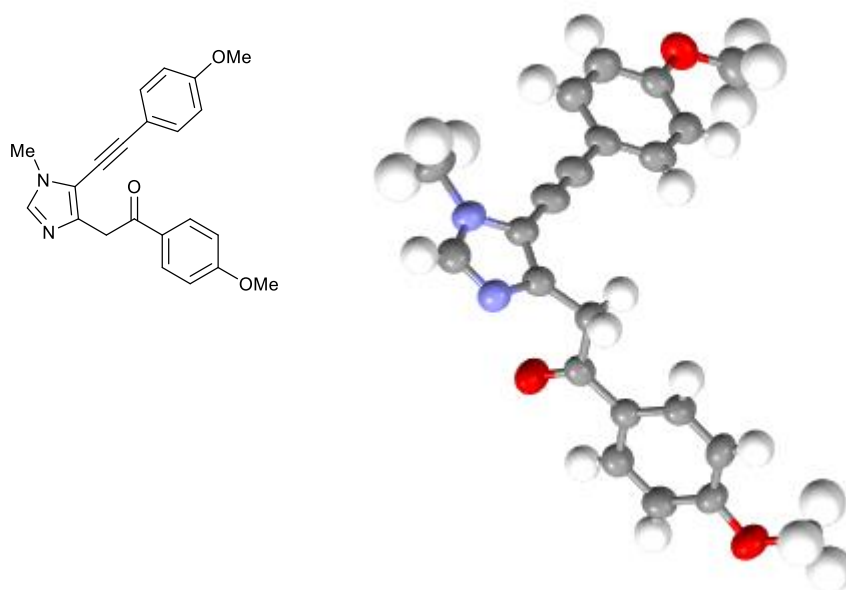


Figure S6. estimated structure of decomposition compound from **1b**+TFA.

