

Electronic Supplementary Information

Reactivity of a carbonyl moiety in organotin(IV) compounds. Novel Pd(II) and Cu(II) complexes supported by organotin(IV) ligands

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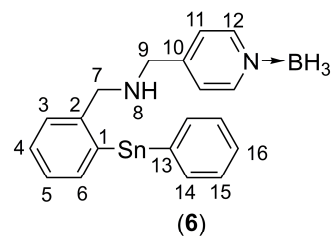
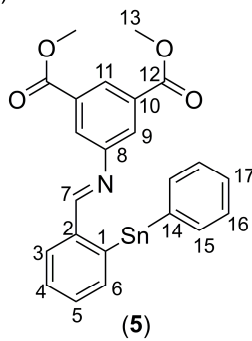
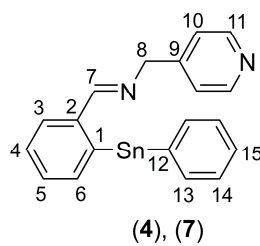
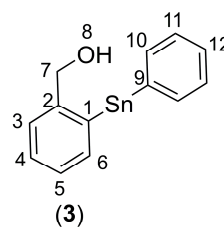
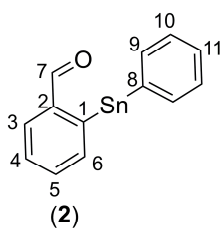
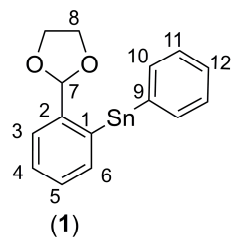
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Numbering schemes for NMR resonance assignments



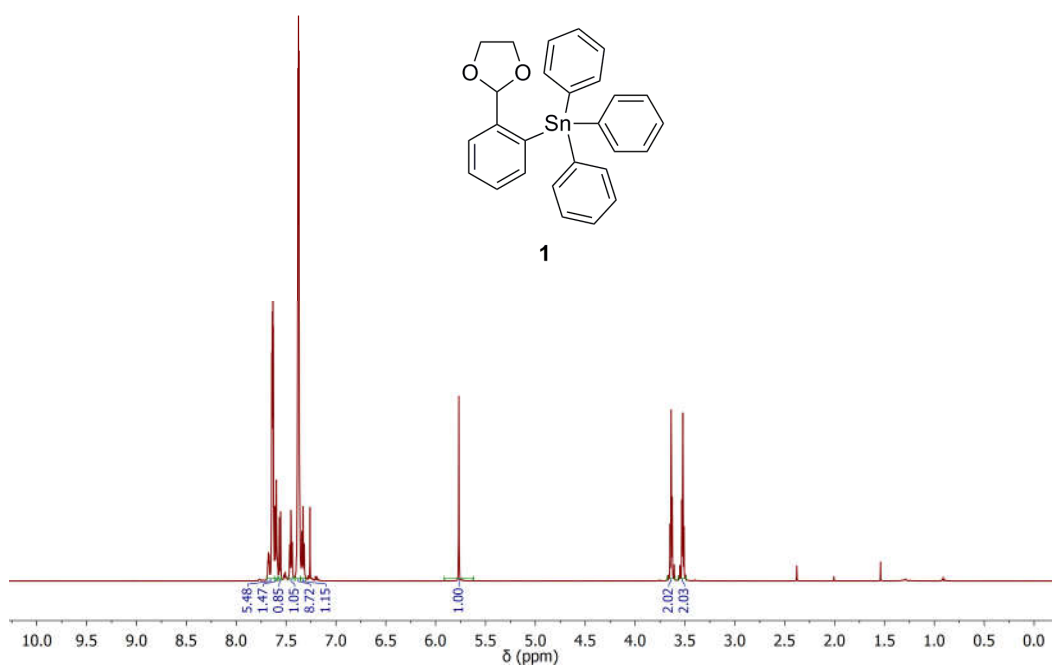


Figure S1. ¹H NMR (CDCl₃, 20 °C) spectrum of [2- $\{(\text{CH}_2\text{O})_2\text{CH}\}$ C₆H₄]₃SnPh₃ (**1**).

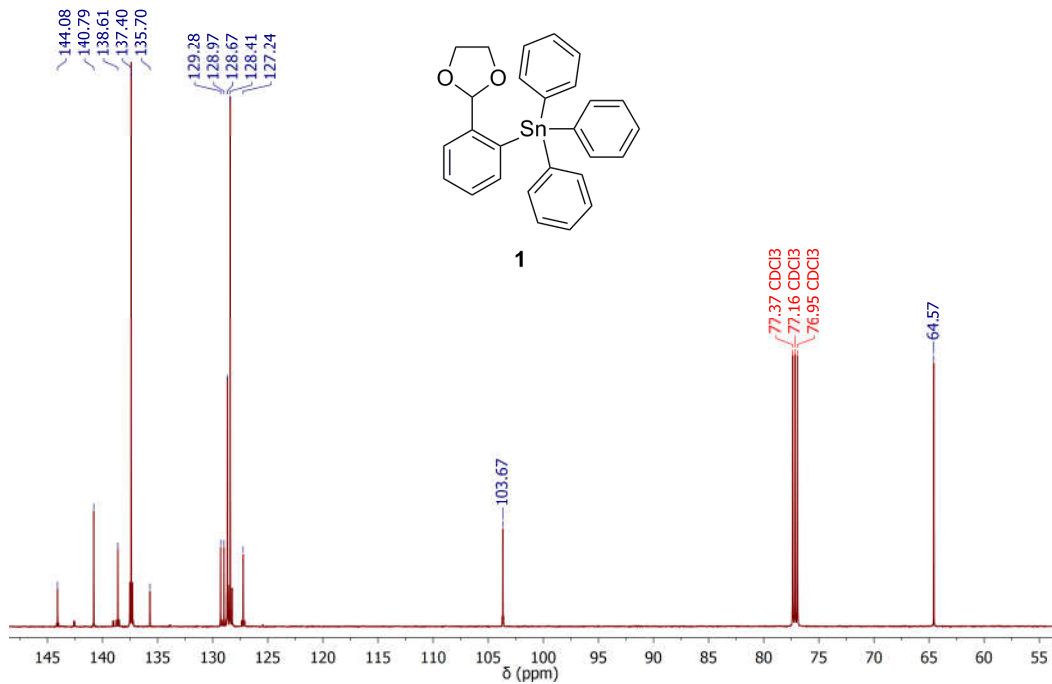


Figure S2. ¹³C{¹H} NMR (CDCl₃, 20 °C) spectrum of [2- $\{(\text{CH}_2\text{O})_2\text{CH}\}$ C₆H₄]₃SnPh₃ (**1**).

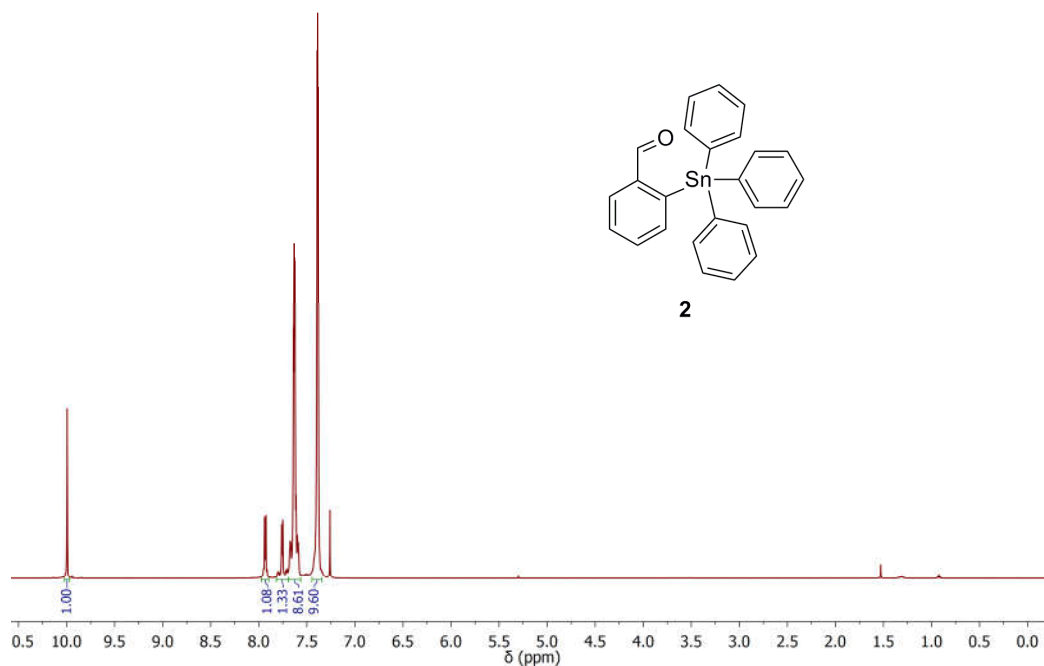


Figure S3. ¹H NMR (CDCl₃, 20 °C) spectrum of [2-(O=CH)C₆H₄]SnPh₃ (**2**).

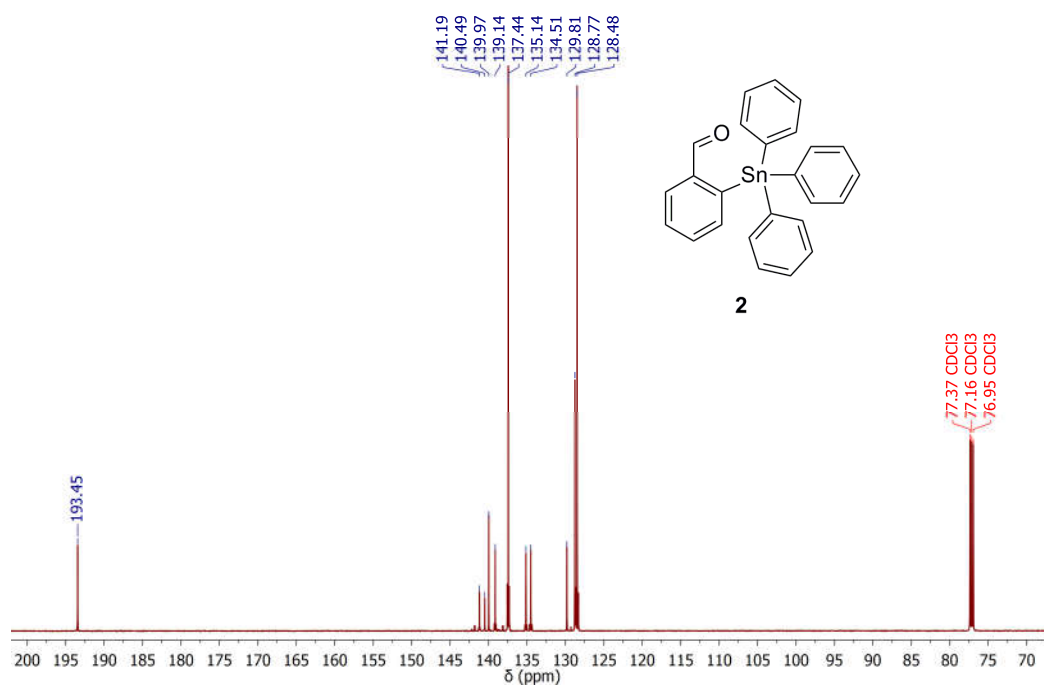


Figure S4. ¹³C {¹H} NMR (CDCl₃, 20 °C) spectrum of [2-(O=CH)C₆H₄]SnPh₃ (**2**).

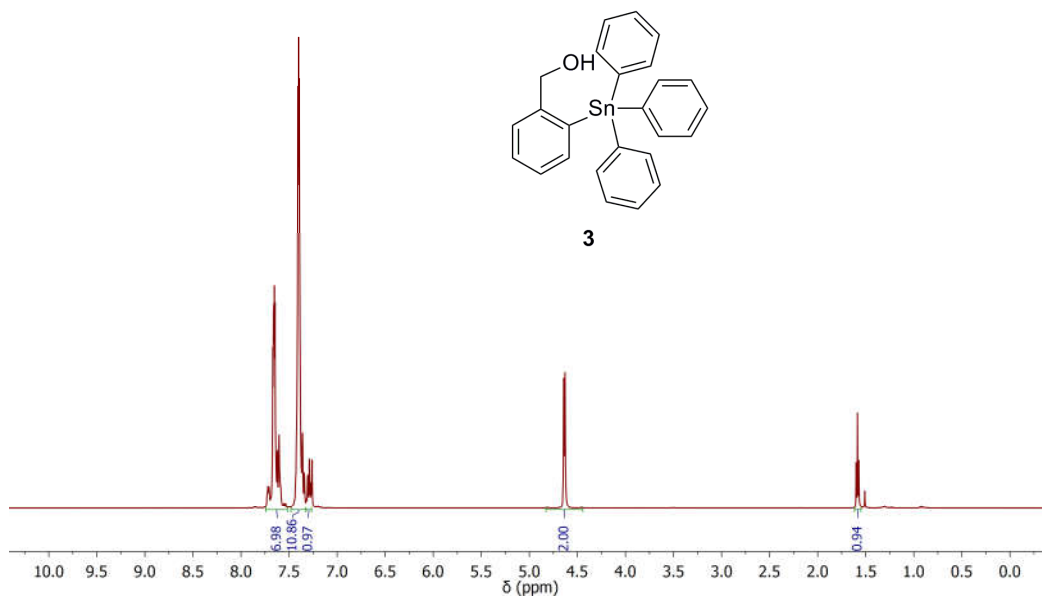


Figure S5. ¹H NMR (CDCl₃, 20 °C) spectrum of [2-(HOCH₂)C₆H₄]SnPh₃ (**3**).

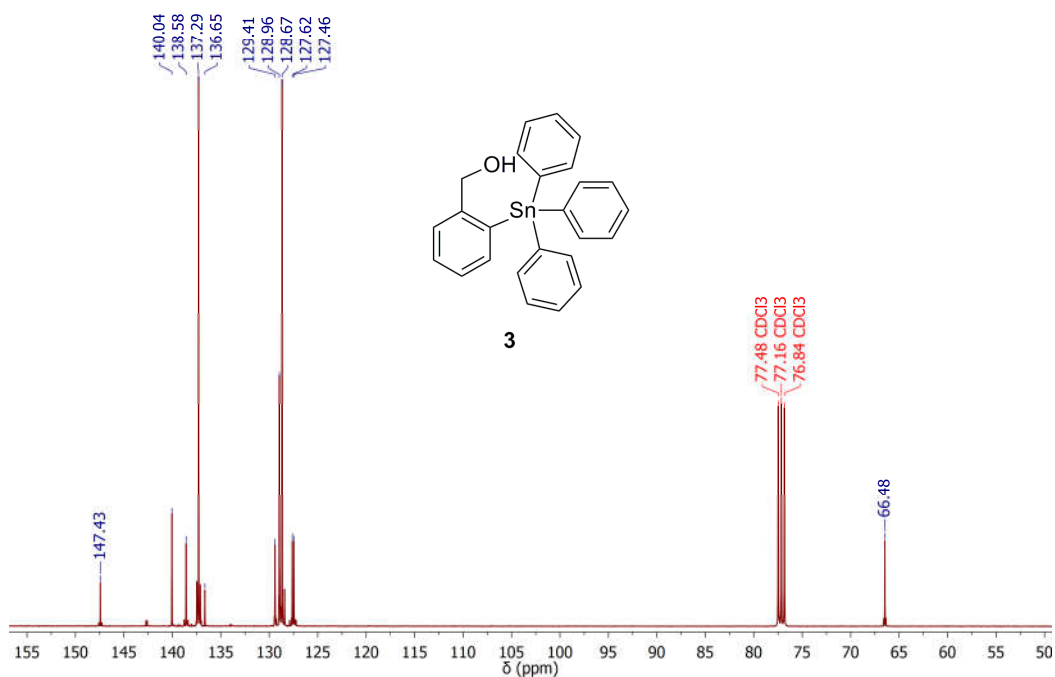


Figure S6. ¹³C {¹H} NMR (CDCl₃, 20 °C) spectrum of [2-(HOCH₂)C₆H₄]SnPh₃ (**3**).

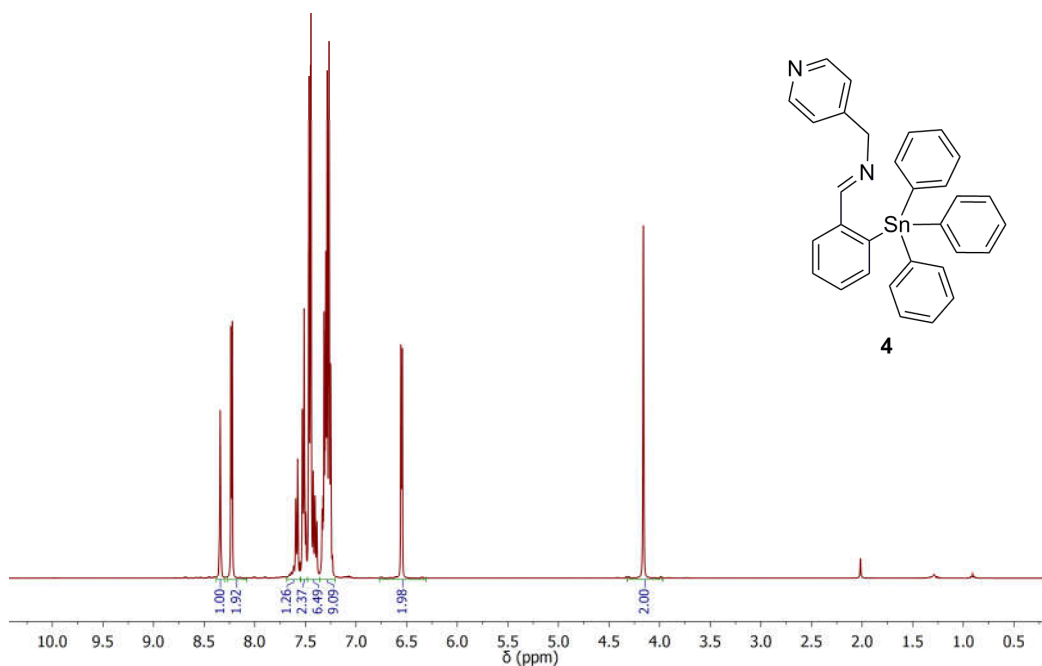


Figure S7. ¹H NMR (CDCl₃, 20 °C) spectrum of [2-(4'-PyCH₂N=CH)C₆H₄]SnPh₃ (**4**).

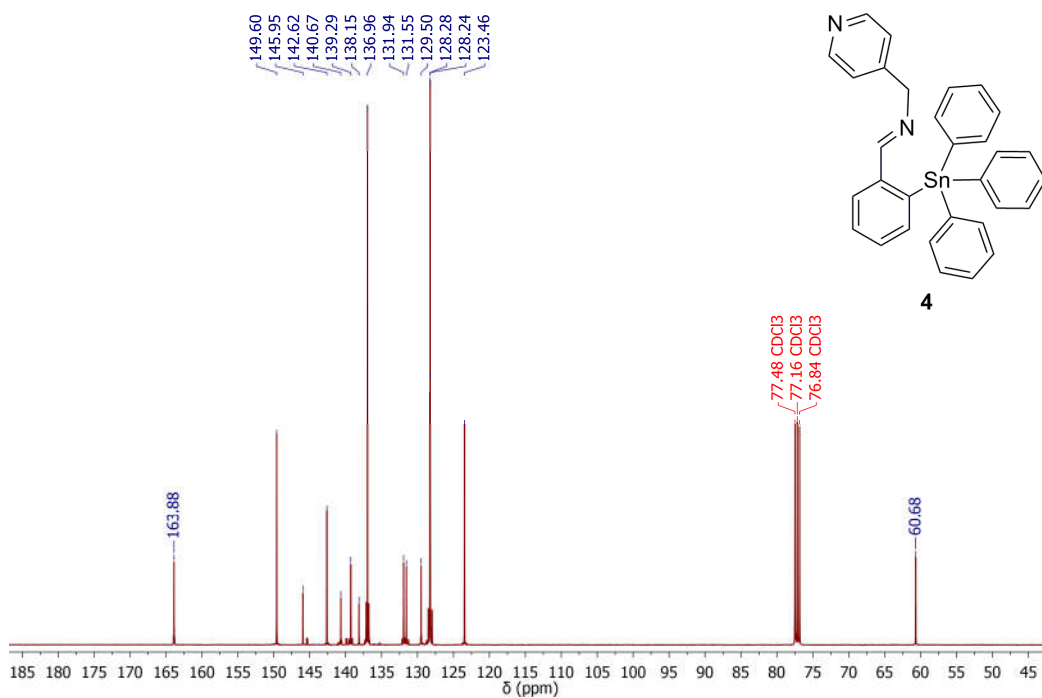


Figure S8. ¹³C{¹H} NMR (CDCl₃, 20 °C) spectrum of [2-(4'-PyCH₂N=CH)C₆H₄]SnPh₃ (**4**).

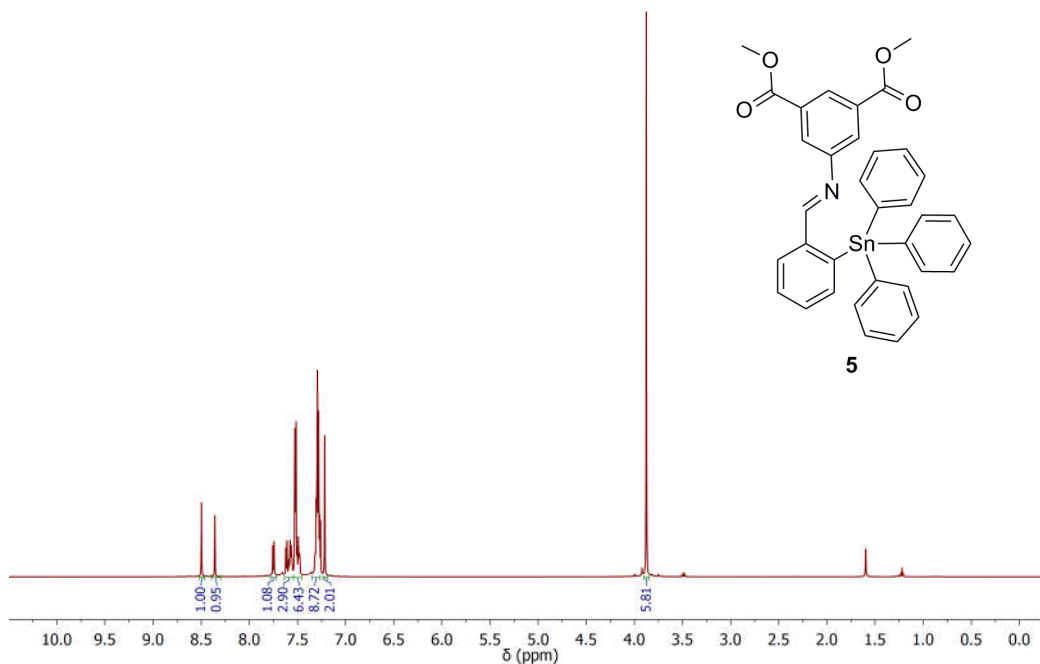


Figure S9. ¹H NMR (CDCl₃, 20 °C) spectrum of [2-(3',5'-(MeOOC)₂-C₆H₃N=CH)C₆H₄]SnPh₃ (**5**).

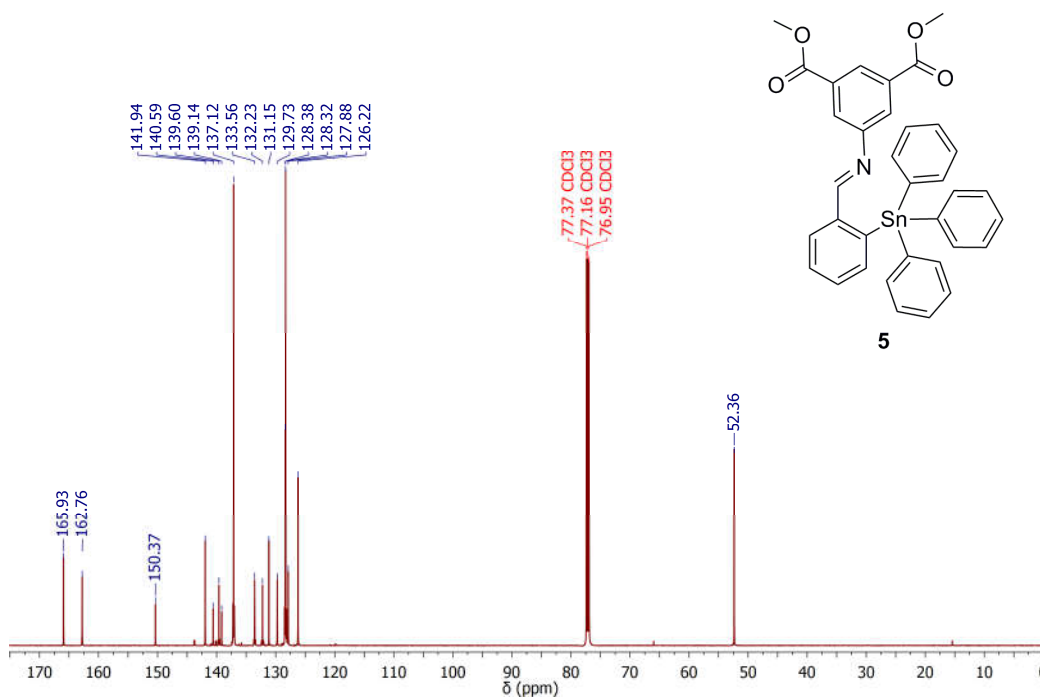


Figure S10. ¹³C {¹H} NMR (CDCl₃, 20 °C) spectrum of [2-(3',5'-(MeOOC)₂-C₆H₃N=CH)C₆H₄]SnPh₃ (**5**).

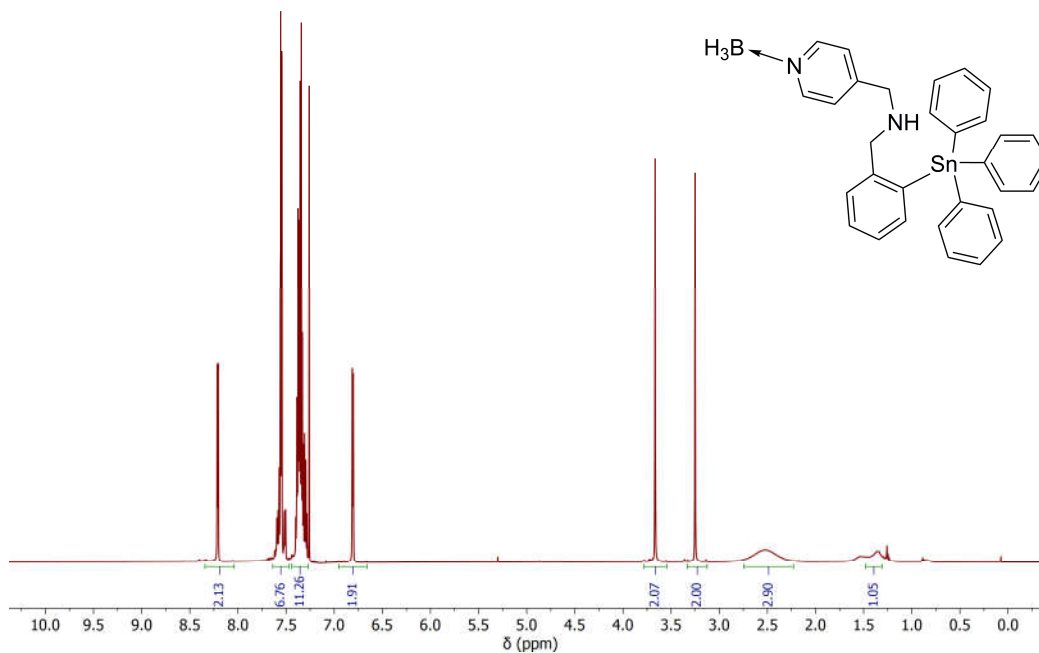


Figure S11. ¹H NMR (CDCl₃, 20 °C) spectrum of [2-(4'-PyCH₂NH-CH₂)C₆H₄]SnPh₃·BH₃ (**6**).

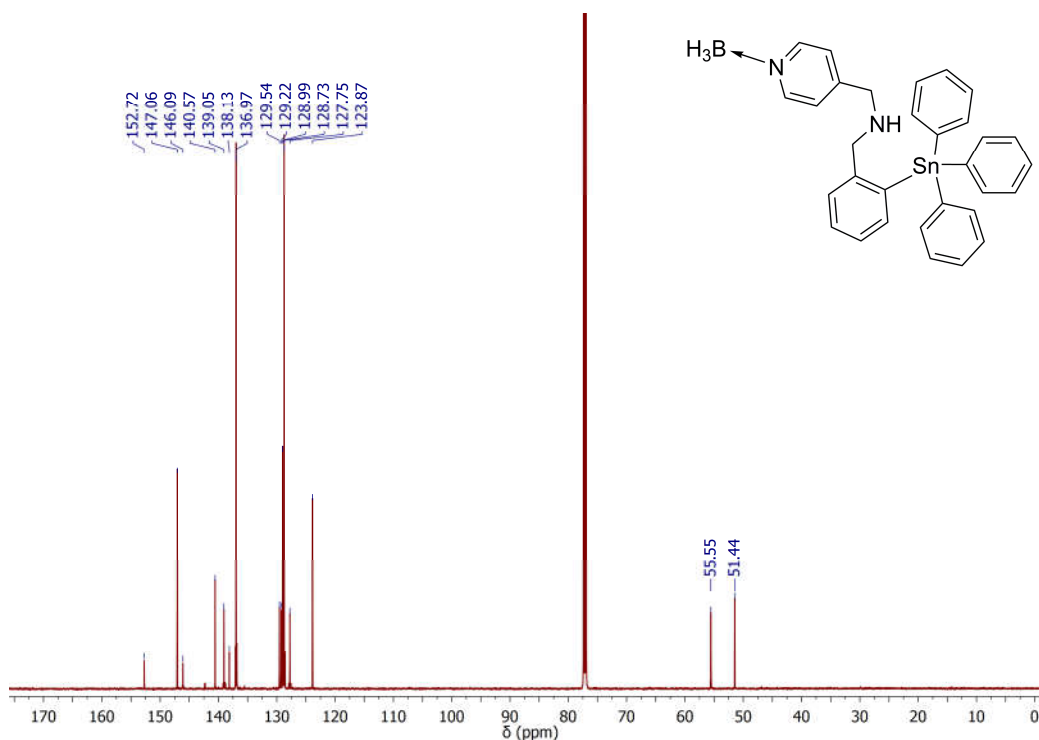


Figure S12. ¹³C {¹H} NMR (CDCl₃, 20 °C) spectrum of [2-(4'-PyCH₂NH-CH₂)C₆H₄]SnPh₃·BH₃ (**6**).

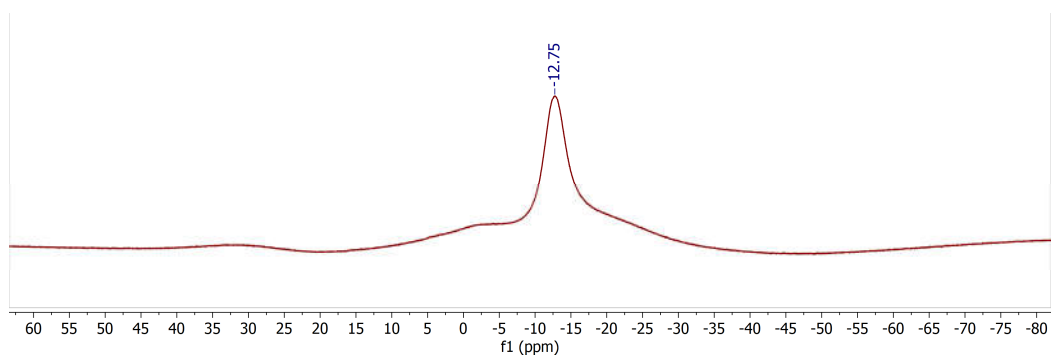


Figure S13. $^{11}\text{B}\{^1\text{H}\}$ NMR (CDCl_3 , 20 °C) spectrum of [2-(4'-PyCH₂NH-CH₂)C₆H₄]SnPh₃·BH₃ (**6**).

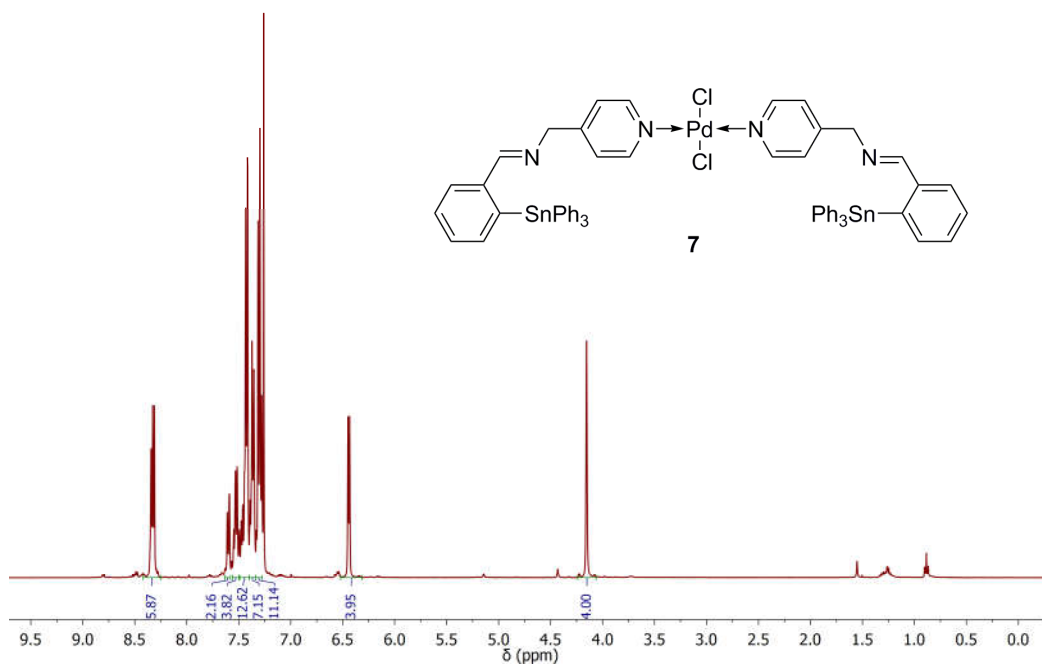


Figure S14. ^1H NMR (CDCl_3 , 20 °C) spectrum of $[\{2-(4'\text{-PyCH}_2\text{N=CH})\text{C}_6\text{H}_4\text{SnPh}_3\}_2\text{PdCl}_2]$ (**7**).

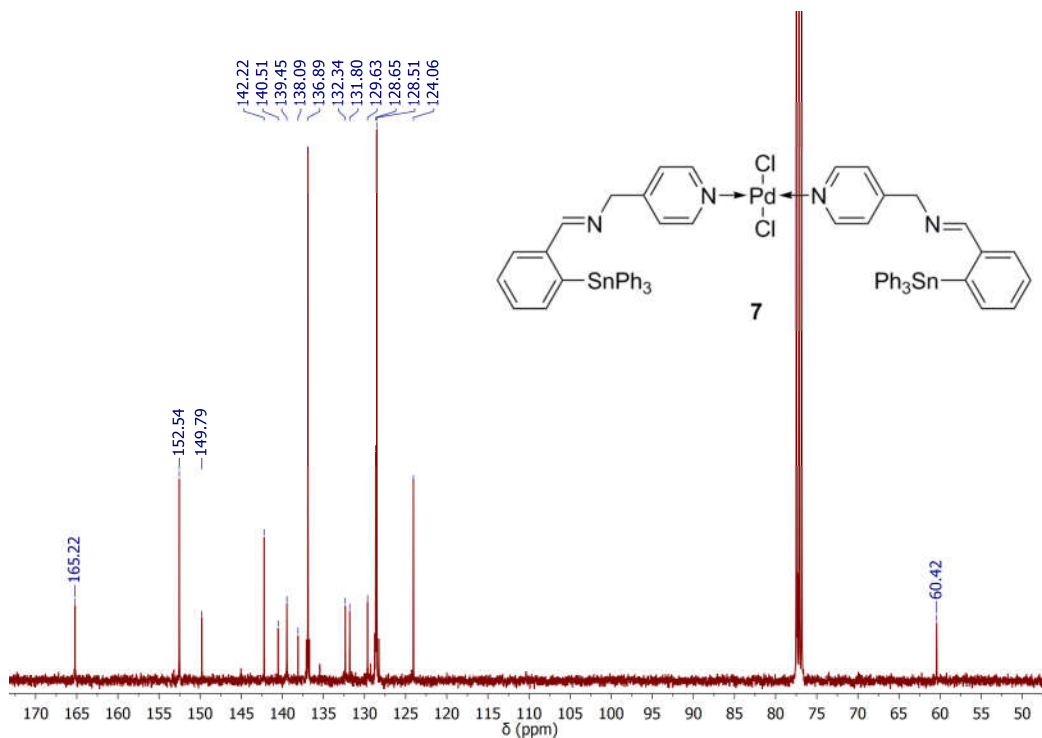


Figure S15. ^{13}C NMR (CDCl_3 , 20 °C) spectrum of $[\{2-(4'\text{-PyCH}_2\text{N=CH})\text{C}_6\text{H}_4\text{SnPh}_3\}_2\text{PdCl}_2]$ (**7**).

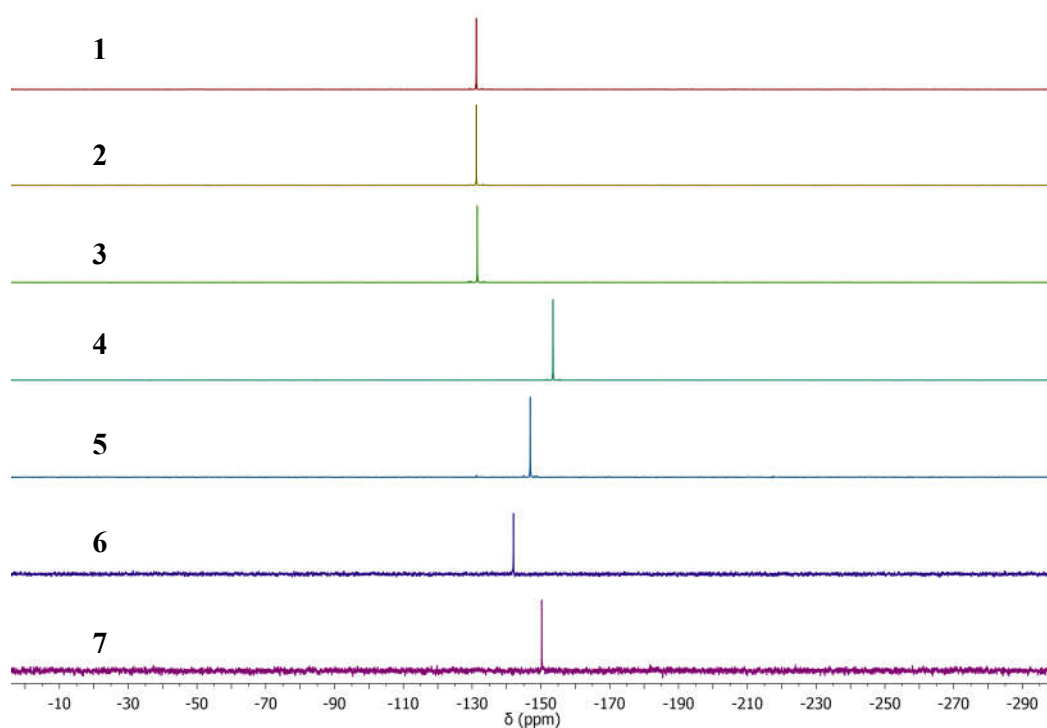


Figure S16. ^{119}Sn NMR (CDCl_3 , 20 $^\circ\text{C}$) stacked spectra of compounds 1-7.

Table S1. X-ray crystal data and structure refinement for compounds **1**, **2**, **4** and **5**

	1·C₇H₈	2	4	5
Empirical formula	C ₆₁ H ₅₆ O ₄ Sn	C ₂₅ H ₂₀ OSn	C ₃₁ H ₂₆ N ₂ Sn	C ₃₅ H ₂₉ NO ₄ Sn
Formula weight	1090.43	455.10	545.23	646.28
Temperature (K)	150(2)	100(2)	297(2)	150(2)
Crystal system	Monoclinic	Orthorhombic	Triclinic	Triclinic
Space group	<i>P</i> 2 ₁ / <i>n</i>	<i>Pna</i> 2 ₁	<i>P</i> -1	<i>P</i> -1
<i>a</i> (Å)	9.844(2)	16.9881(6)	9.898(16)	10.0636(1)
<i>b</i> (Å)	18.861(4)	15.2425(5)	11.551(2)	11.6223(2)
<i>c</i> (Å)	14.068(3)	7.7375(2)	12.613(3)	14.212(2)
α (°)	90	90	70.529(4)	96.462(2)
β (°)	108.167(4)	90	80.243(4)	97.444(2)
γ (°)	90	90	67.538(3)	113.734(2)
Volume (Å ³)	2481.8(10)	2003.56(11)	1255.0(5)	1483.8(4)
<i>Z</i>	2	4	2	2
<i>D</i> _{calc} (g cm ⁻³)	1.459	1.509	1.443	1.447
Absorption coefficient (mm ⁻¹)	1.055	1.286	1.039	0.900
<i>F</i> (000)	1108	912	552	656
Crystal size (mm)	0.29x0.25x0.22	0.08x0.08x0.11	0.29x0.28x0.26	0.23x0.26x0.33
θ range for data collection (°)	1.867 to 25.099	2.40 to 28.28	1.714 to 24.999	1.469 to 24.999
Reflections collected	24041	28944	11831	14095
Independent reflections	4423	4967	4367	5182
	[<i>R</i> _{int} = 0.0481]	[<i>R</i> _{int} = 0.0333]	[<i>R</i> _{int} = 0.0361]	[<i>R</i> _{int} = 0.0255]
Absorption correction	Multi-Scan ¹	Multi-Scan ¹	Multi-Scan ¹	Multi-Scan ¹
Data / restraints / parameters	4423 / 123 / 335	4967 / 1 / 244	4367 / 0 / 307	5182 / 0 / 372
Goodness-of-fit on <i>F</i> ²	1.095	1.085	1.026	1.088
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> _I = 0.0385 <i>wR</i> ₂ = 0.0984	<i>R</i> _I = 0.0327 <i>wR</i> ₂ = 0.0857	<i>R</i> _I = 0.0324 <i>wR</i> ₂ = 0.0657	<i>R</i> _I = 0.0252 <i>wR</i> ₂ = 0.0614
<i>R</i> indices (all data)	<i>R</i> _I = 0.0457 <i>wR</i> ₂ = 0.1031	<i>R</i> _I = 0.0370 <i>wR</i> ₂ = 0.0885	<i>R</i> _I = 0.0402 <i>wR</i> ₂ = 0.0690	<i>R</i> _I = 0.0271 <i>wR</i> ₂ = 0.0623
Largest difference peak and hole (e Å ⁻³)	1.421 and -0.477	1.494 and -0.942	0.48 and -0.301	0.55 and -0.21
CCDC No.	2003707	2003708	2003705	2003706

G. M. Sheldrick, *SADABS, Program for area detector adsorption correction*, Institute for Inorganic Chemistry, University of Göttingen, Germany, 1996.

Table S2. X-ray crystal data and structure refinement for compounds **6-8**.

	6	7	8
Empirical formula	C ₃₁ H ₃₁ BN ₂ Sn	C ₆₂ H ₅₂ Cl ₂ N ₄ PdSn ₂	C ₇₂ H ₅₄ CuF ₁₂ N ₄ O ₄ Sn
Formula weight	561.08	1267.75	1568.11
Temperature (K)	297(2)	100(2)	100(2)
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	Cc	<i>P</i> 2 ₁ / <i>n</i>	<i>C</i> 2/ <i>c</i>
<i>a</i> (Å)	14.6278(17)	8.2544(7)	27.3118(9)
<i>b</i> (Å)	13.3165(15)	17.3898(14)	12.8432(5)
<i>c</i> (Å)	14.9311(17)	18.3621(16)	20.0980(7)
α (°)	90	90	90
β (°)	111.563(2)	95.750(3)	111.2470(1)
γ (°)	90	90	90
Volume (Å ³)	2704.9(5)	2622.5(4)	6570.6(4)
<i>Z</i>	4	2	4
<i>D</i> _{calc} (g cm ⁻³)	1.378	1.605	1.585
Absorption coefficient (mm ⁻¹)	0.966	1.431	1.162
<i>F</i> (000)	1144	1264	3132
Crystal size (mm)	0.20x0.25x0.29	0.05x0.05x0.07	0.06x0.09x0.14
θ range for data collection (°)	2.140 to 25.099	2.23 to 29.48	2.18 to 28.30
Reflections collected	12809	75729	111572
Independent reflections	4780	7288	8170
	[<i>R</i> _{int} = 0.0477]	[<i>R</i> _{int} = 0.0651]	[<i>R</i> _{int} = 0.0366]
Absorption correction	Multi-Scan ¹	Multi-Scan ¹	Multi-Scan ¹
Data / restraints / parameters	4780 / 9 / 332	7288 / 2510 / 651	8170 / 360 / 487
Goodness-of-fit on <i>F</i> ²	1.036	1.006	1.030
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> _I = 0.0389 <i>wR</i> ₂ = 0.0904	<i>R</i> _I = 0.0477 <i>wR</i> ₂ = 0.1071	<i>R</i> _I = 0.0298 <i>wR</i> ₂ = 0.0702
<i>R</i> indices (all data)	<i>R</i> _I = 0.0416 <i>wR</i> ₂ = 0.0919	<i>R</i> _I = 0.0951 <i>wR</i> ₂ = 0.1335	<i>R</i> _I = 0.0346 <i>wR</i> ₂ = 0.0735
Largest difference peak and hole (e Å ⁻³)	1.068 and -0.317	0.566 and -0.758	2.241 and -1.327
CCDC No.	2003704	2003709	2003710

¹G. M. Sheldrick, *SADABS, Program for area detector adsorption correction*, Institute for Inorganic Chemistry, University of Göttingen, Germany, 1996.

[2- $\{(\text{CH}_2\text{O})_2\text{CH}\}\text{C}_6\text{H}_4\text{]\text{SnPh}_3$ (1**)**

- the crystal contains a 1:1 mixture of $p\text{S}_{\text{O}(1)}\text{-S}_{\text{C}(7)}\text{-1}$ and $p\text{R}_{\text{O}(1)}\text{-R}_{\text{C}(7)}\text{-1}$

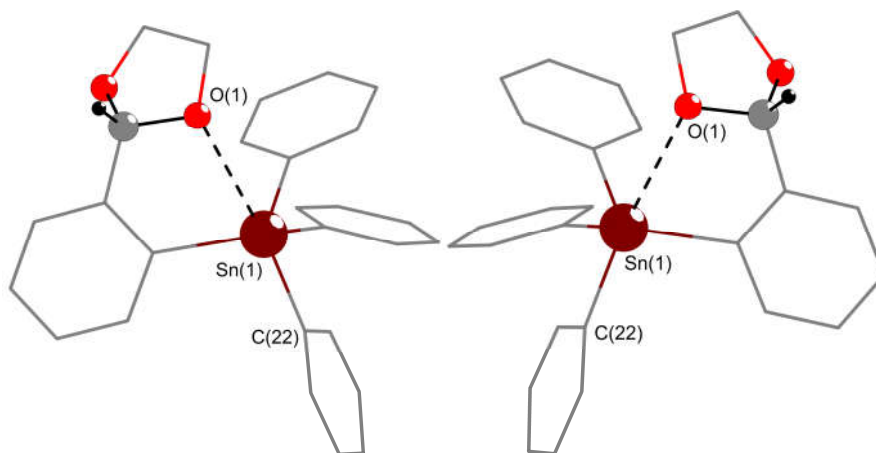


Figure S17. Molecular structure of $p\text{S}_{\text{O}(1)}\text{-S}_{\text{C}(7)}\text{-1}$ isomer (*left*) and $p\text{R}_{\text{O}(1)}\text{-R}_{\text{C}(7)}\text{-1}$ isomer (*right*) in the crystal of **1** (only methyne hydrogen is shown).

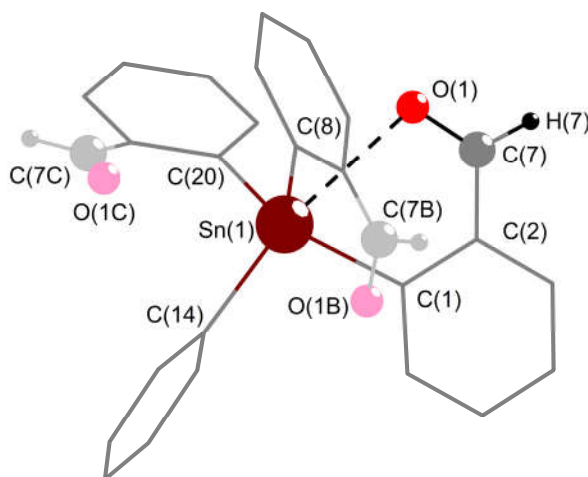


Figure S18. Molecular structure of **3** showing the carbonyl fragment located on three different phenyl rings with a site having preponderant occupancy [0.69 C(7)] over the others [0.18 C(7B) and 0.13 C(7C)] (only carbonyl hydrogen atoms are shown).

[2-(4-PyCH₂NH-CH₂)C₆H₄][SnPh₃·BH₃ (6**)**

- the crystal contains a 1:1 mixture of *pS_{N(1)}-S_{N(1)}*-**6** and *pR_{N(1)}-R_{N(1)}*-**6**

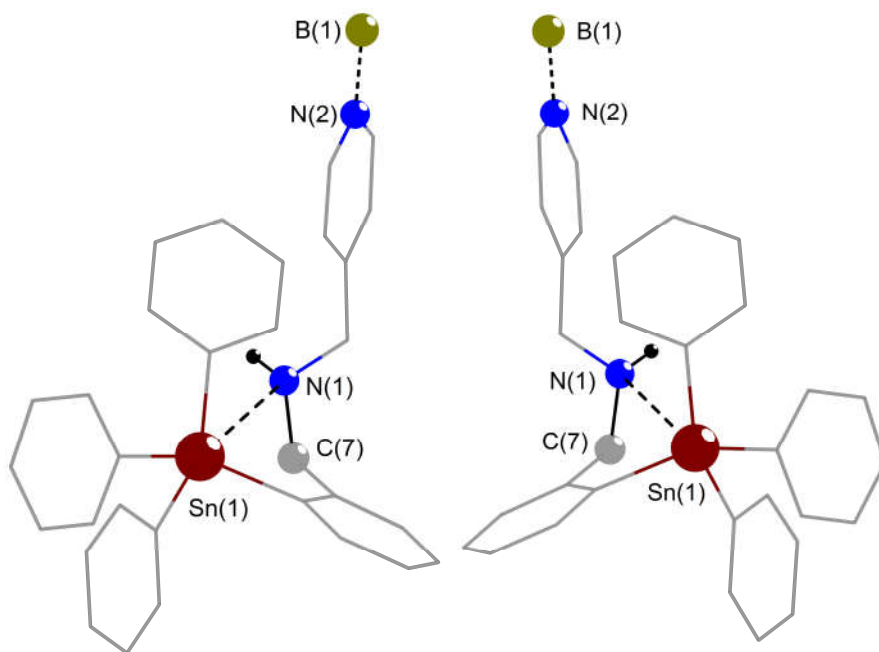


Figure S19. Molecular structure of *pS_{N(1)}-S_{N(1)}*-**6** isomer (*left*) and *pR_{N(1)}-R_{N(1)}*-**6** isomer (*right*) in the crystal of **6** (only amine hydrogen is shown).

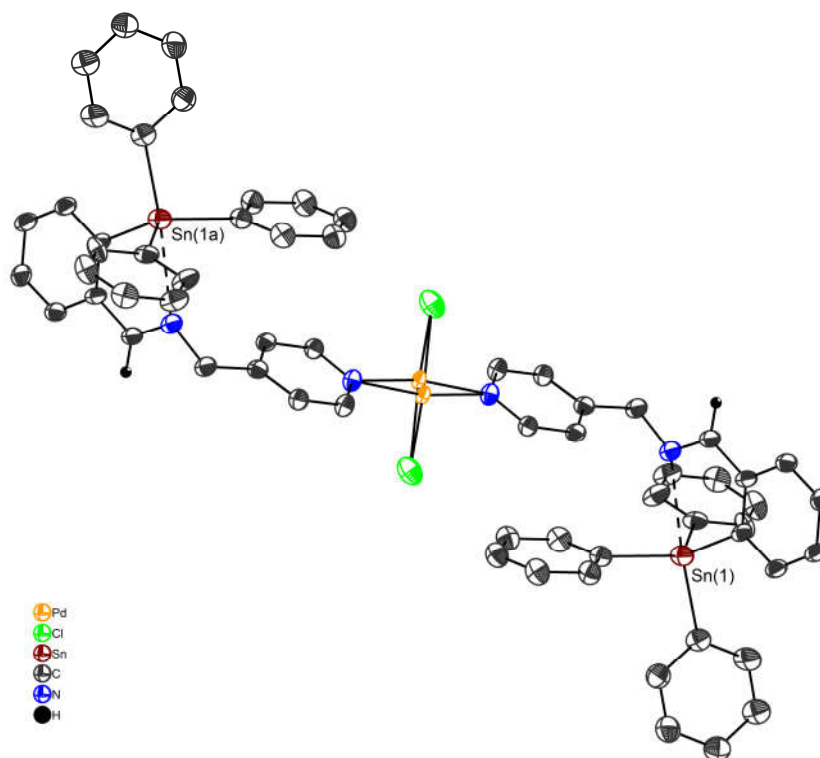


Figure S20. ORTEP representation at 30% probability of **7**, displaying only the major occupancy site of the molecule (68%) (only imine hydrogen atoms are shown). Selected interatomic distances (Å) and bond angles (°): Sn(1)–C distances (Å): 2.192(10), 2.194(11), 2.157(7), 2.154(8), Sn(1)···N distance 2.737(12) (Å), axial angle N–Sn(1)–C = 174.4(4)°.

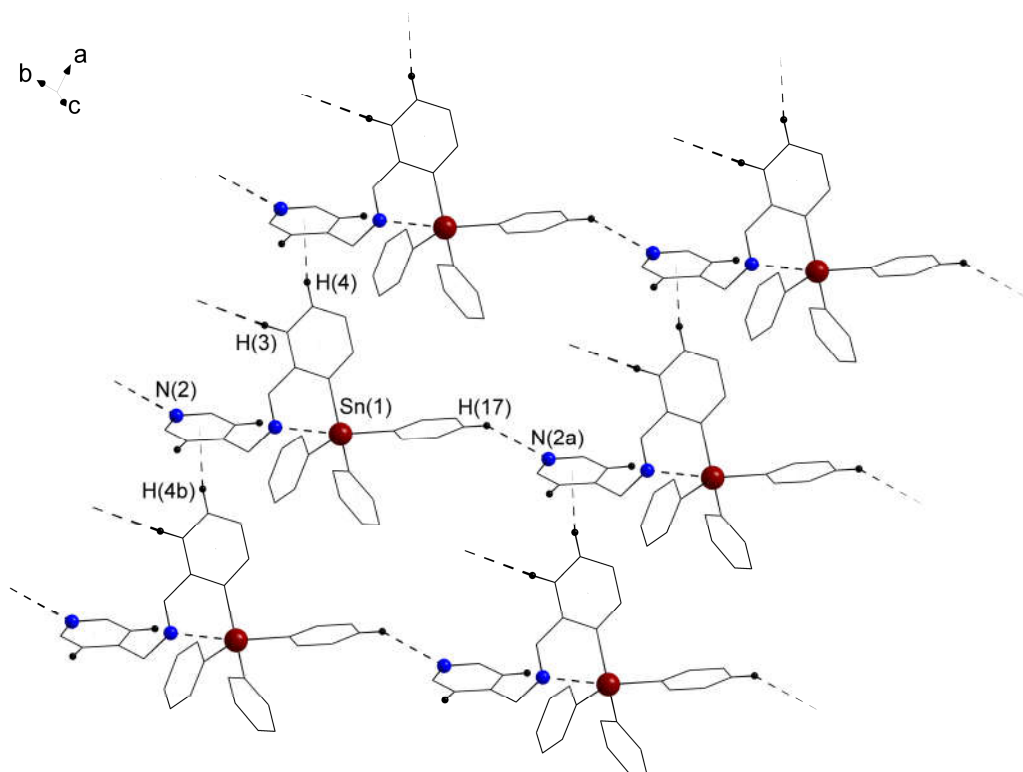


Figure S21. Packing of the molecules in the crystal of **4**, showing the interconnected chains forming a layer (only hydrogen atoms involved in intermolecular contacts are shown). [Symmetry equivalents atom ($l+x, -l+y, l+z$), ($-l+x, y, z$) are given by “a”, “b”, respectively].

- intermolecular distance	H(17)···N(2a)	2.66 Å	$\sum r_{\text{vdW}}(\text{N,H})$ 2.75 Å $\gamma = 12.2^\circ$
	C(4b)–H(4b)···Ph _{centroid} {Py}	2.89 Å	

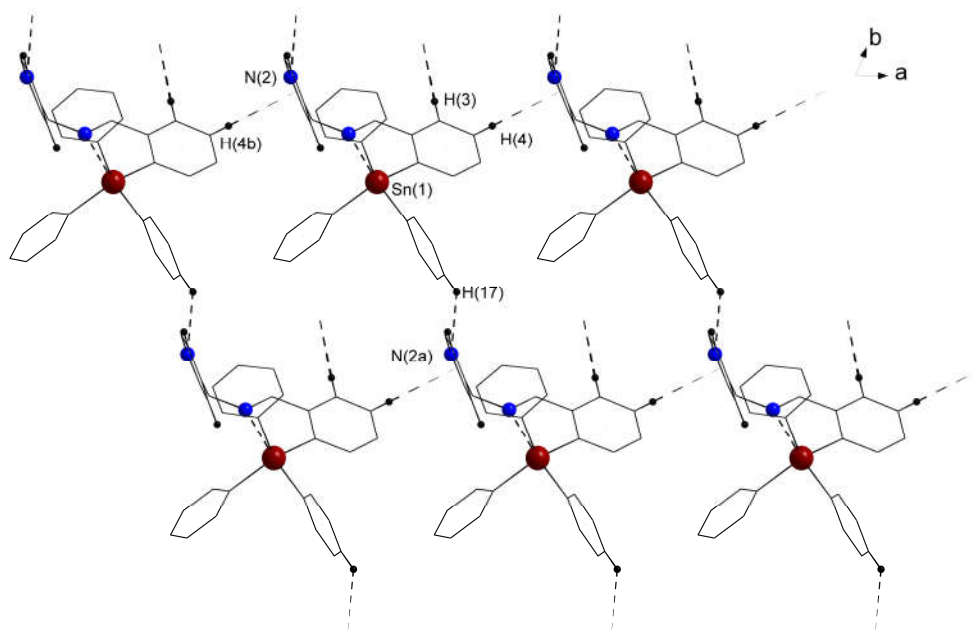


Figure S22. Viewing along *c* axis of the packing of molecules in the crystal of **4**, showing the interconnected chains forming a layer (only hydrogen atoms involved in intermolecular contacts are shown). [Symmetry equivalents atom ($I+x, -I+y, I+z$), ($-I+x, y, z$) are given by “a”, “b”, respectively].

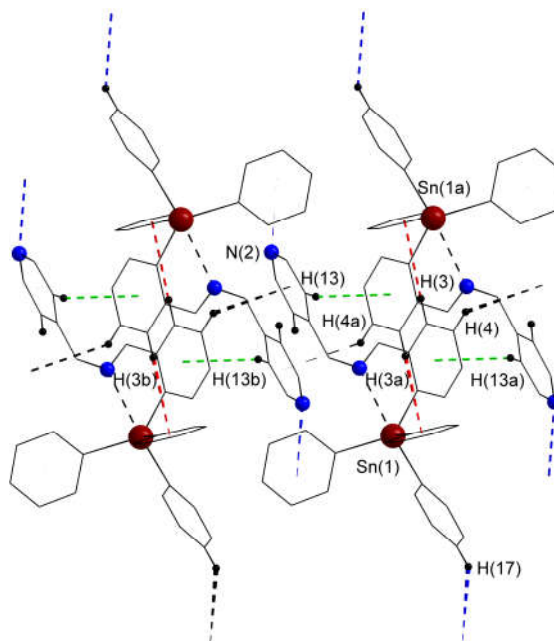


Figure S23. Packing of the molecules in the crystal **4**, showing the interconnected layers forming 3D supramolecular architecture (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms ($I-x, I-y, -z$), ($-x, I-y, -z$) are given by “a” and “b” respectively].

$C(3)-H(3) \cdots Ph_{\text{centroid}}\{C(26)-C(31)\}$	2.85 Å	$\gamma = 6.4^\circ$
$C(13b)-H(13b) \cdots Ph_{\text{centroid}}\{C(1)-C(6)\}$	2.95 Å	$\gamma = 18.2^\circ$

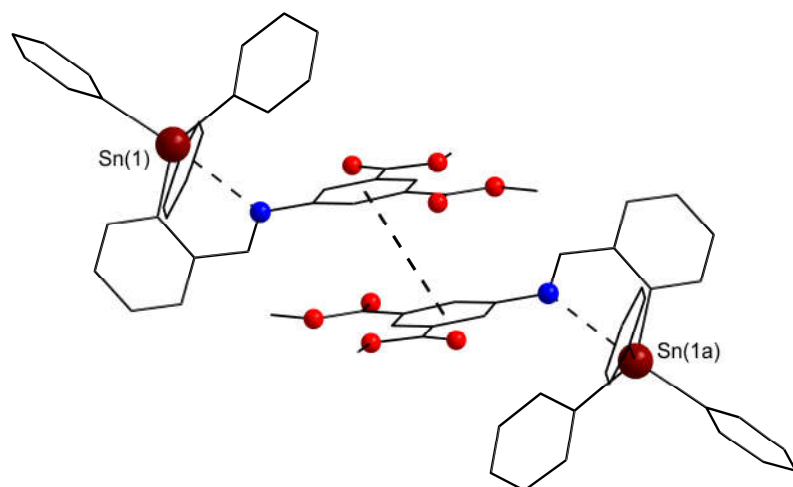


Figure S24. Offset $\pi\cdots\pi$ stacking in the crystal of **5** (hydrogen atoms are omitted for clarity) [symmetry equivalent ($2-x$, $1-y$, $2-z$) is corresponding to “a”].

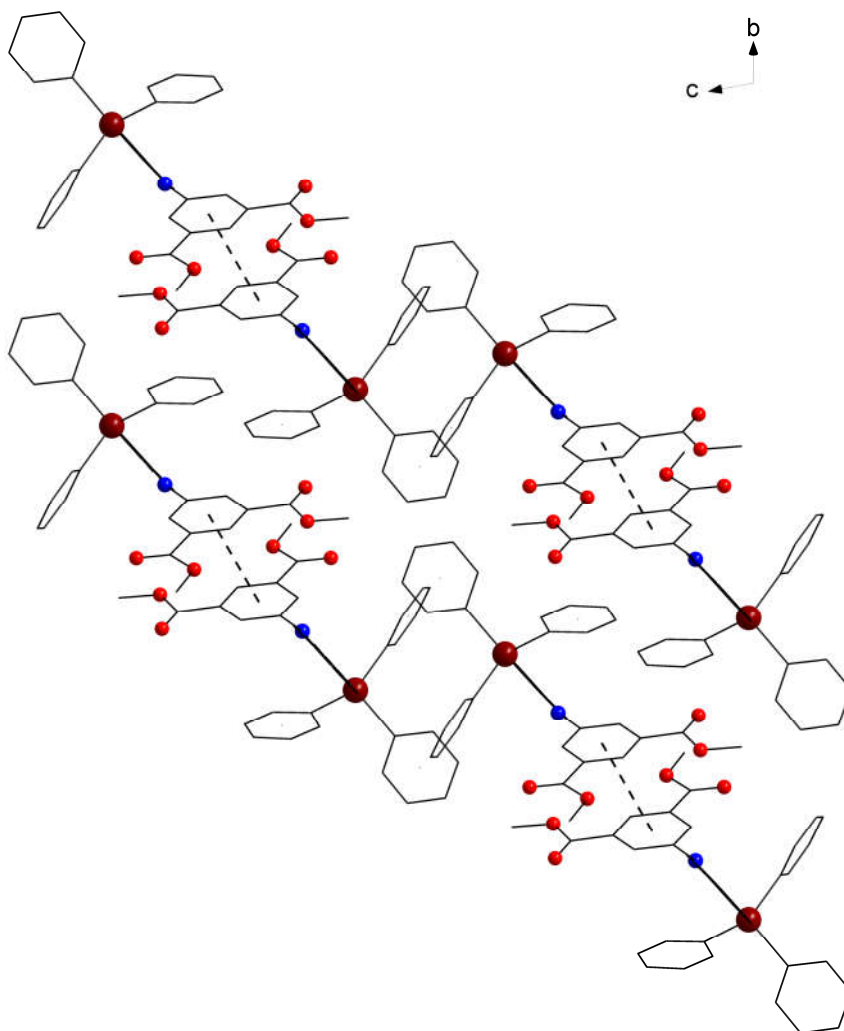


Figure S25. View along a axis of the crystal packing of **5** (hydrogen atoms are omitted for clarity).

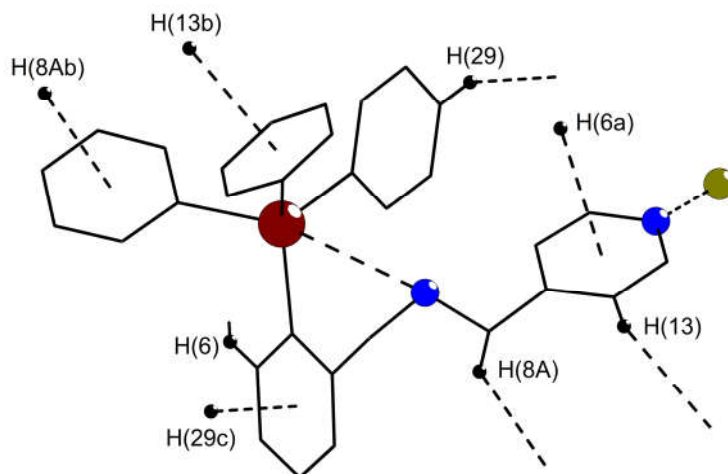


Figure S26. CH $\cdots\pi$ interactions in the crystal of **6** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms ($0.5+x$, $0.5-y$, $0.5+z$), ($-0.5+x$, $0.5+y$, z) and ($-0.5+x$, $-0.5+y$, z) are given by “a”, “b” and “c”, respectively].

Intermolecular CH $\cdots\pi$ interactions

C(6)–H(6) \cdots Ph _{centroid} {Py}	3.08 Å	$\gamma = 9.0^\circ$
C(8)–H(8A) \cdots Ph _{centroid} {C(14)–C(19)}	3.06 Å	$\gamma = 18.9^\circ$
C(13)–H(13) \cdots Ph _{centroid} {C(20)–C(25)}	2.79 Å	$\gamma = 5.3^\circ$
C(29)–H(29) \cdots Ph _{centroid} {C(1)–C(6)}	2.83 Å	$\gamma = 18.3^\circ$

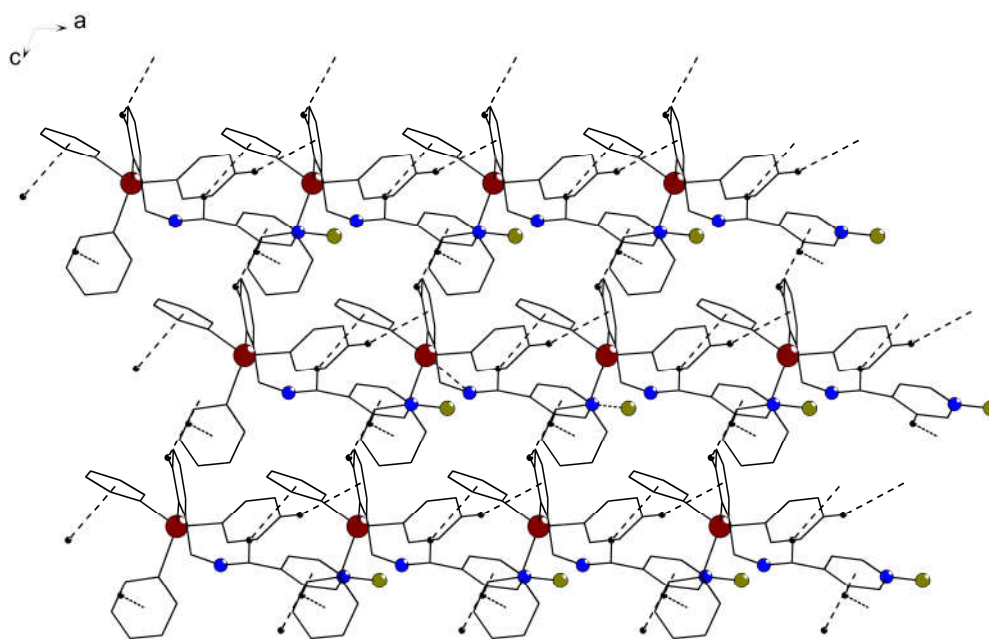


Figure S27. View along b axis of the 3D supramolecular architecture in the crystal packing of **6** (only hydrogen atoms involved in intermolecular contacts are shown).

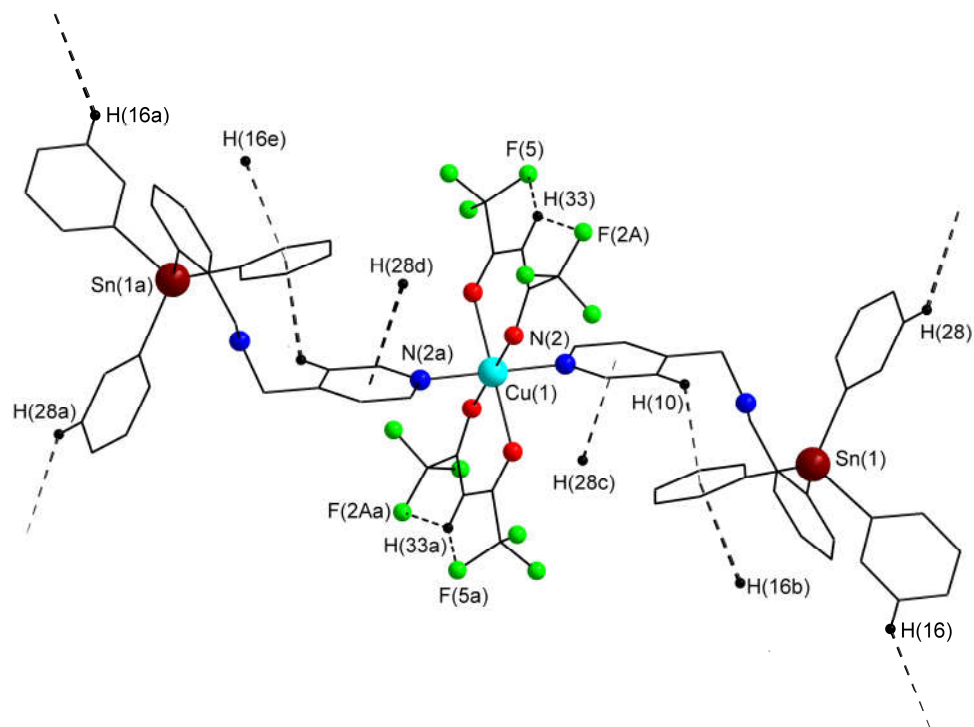


Figure S28. CH $\cdots\pi$ interactions in the crystal of **8** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms ($1-x, 1-y, 1-z$), ($1.5-x, 0.5-y, 2-z$), ($1.5-x, -0.5+y, 1.5-z$), ($-0.5+x, 1.5-y, -0.5+z$) and ($-0.5+x, 0.5+y, -1+z$) are given by “a”, “b”, “c”, “d” and “e”, respectively; only the major occupancy site of the disordered CF₃ fragment is depicted].

Intramolecular H \cdots F interactions

F(5)–(H33)	2.50 Å	$\sum r_{\text{cov}}(\text{F}, \text{H}) = 0.88 \text{ Å}$
F(2A)–(H33)	2.29 Å	$\sum r_{\text{vdW}}(\text{F}, \text{H}) = 2.55 \text{ Å}$

Intramolecular CH $\cdots\pi$ interactions

C(10)–H(10) \cdots Ph _{centroid} {C(20)–C(25)}	2.77 Å	$\gamma = 10.4^\circ$
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Intermolecular CH $\cdots\pi$ interactions

C(16)–H(16) \cdots Ph _{centroid} {C(20)–C(25)}	2.88 Å	$\gamma = 5.3^\circ$
C(28)–H(28) \cdots Ph _{centroid} {Py}	2.96 Å	$\gamma = 14.1^\circ$

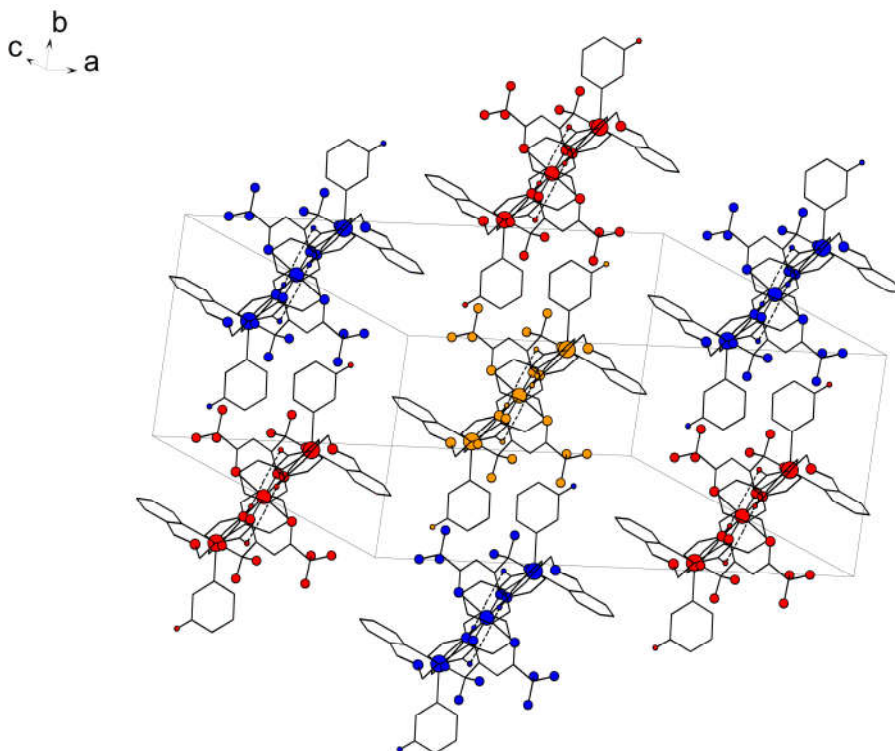


Figure S29. View of the parallel chain-like polymers in the crystal of **8** generated by the $\text{CH}\cdots\pi$ interaction $\text{C}(16)\text{--H}(16)\cdots\text{Ph}_{\text{centroid}}\{\text{C}(20)\text{--C}(25)\}$ (only hydrogen atoms involved in intermolecular contacts are shown).

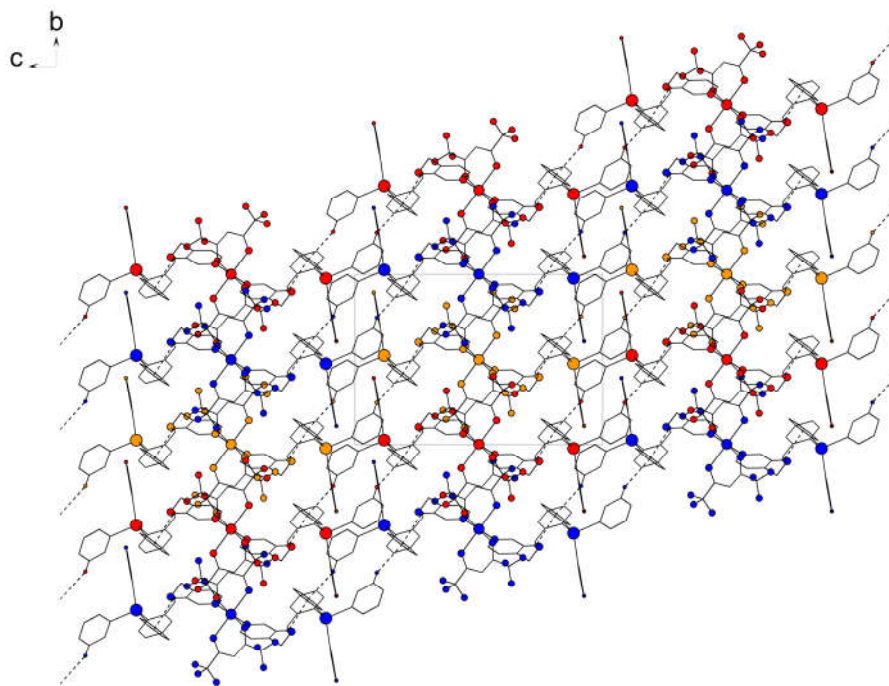


Figure S30. View along a axis of the parallel chain-like polymers in the crystal of **8** generated by the $\text{CH}\cdots\pi$ interaction $\text{C}(16)\text{--H}(16)\cdots\text{Ph}_{\text{centroid}}\{\text{C}(20)\text{--C}(25)\}$ (only hydrogen atoms involved in intermolecular contacts are shown).

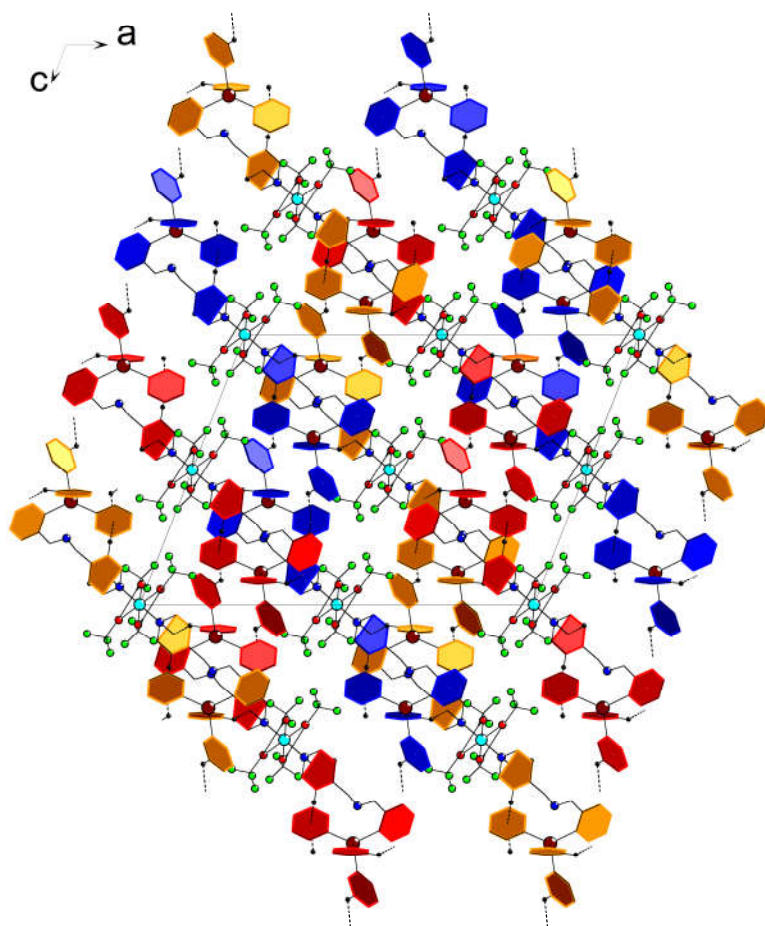


Figure S31. View along b axis of the 3D supramolecular architecture in the crystal of **8**. Interconnected chain-like polymers are highlighted in different colours (only hydrogen atoms involved in intermolecular contacts are shown).