

Electronic Supplementary Information

Etch-Doping Strategy: Cobalt-Iron Bimetallic Phosphide as a Bifunction Electrocatalyst for Highly Efficient Water Splitting

Yuanyuan Wang, Shanshan Wang, Xiaogang Chen, Xuan Zhao, Shulong Chang,
Fengmei Guo, Jie Xu, Yuanyuan Shang and Yingjiu Zhang*

Key Laboratory of Material Physics, Ministry of Education, School of Physics and
Microelectronics, Zhengzhou University, Zhengzhou 450052, PR China

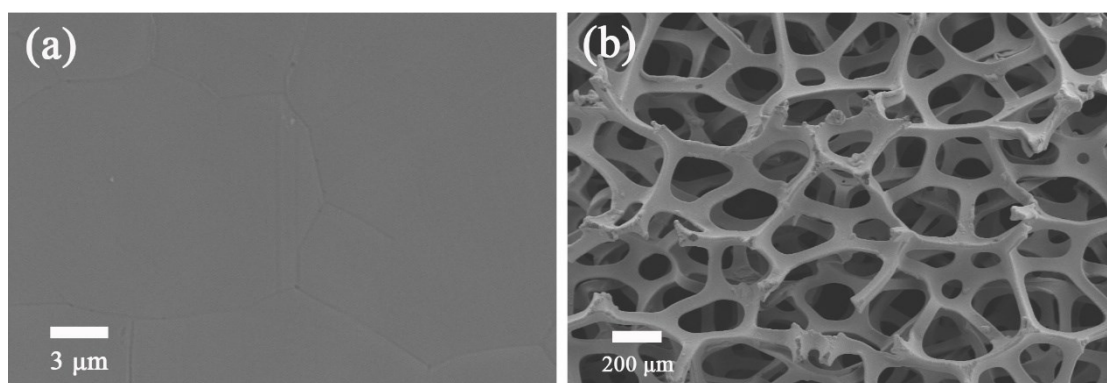


Fig. S1 (a) High and (b) low resolution SEM images of bare NF.

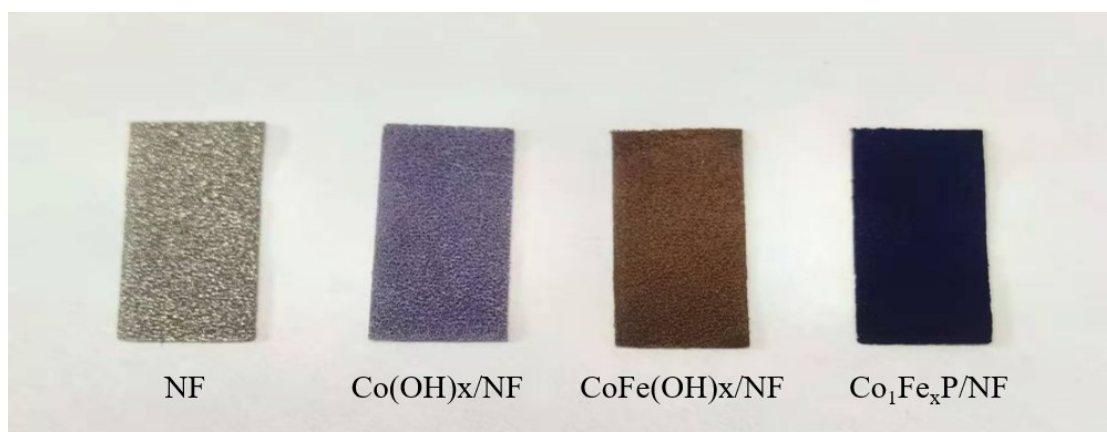


Fig. S2 Physical map of each product during the experiment

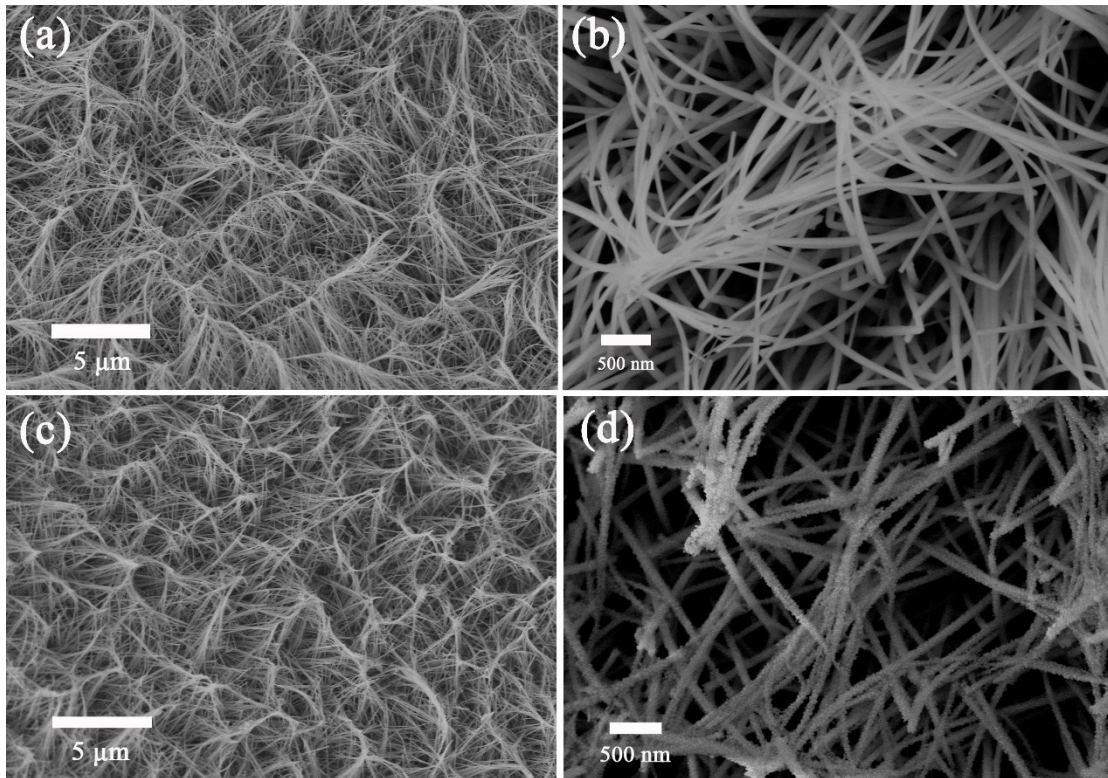


Fig. S3 (a) Low and (b) high resolution SEM images of $\text{Co(OH)}_x/\text{NF}$. (a) Low and (b) high resolution SEM images of $\text{CoFe(OH)}_x/\text{NF}$.

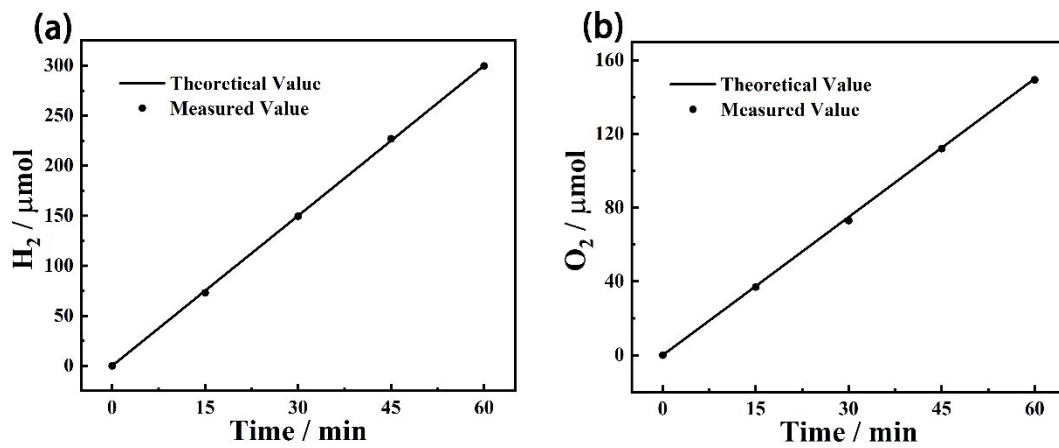


Fig. S4 (a) Electrocatalytic efficiency of H_2 production over $\text{Co}_1\text{Fe}_{0.1}\text{P}/\text{NF}$ at a current density of 20 mA cm^{-2} . (b) Electrocatalytic efficiency of O_2 production over $\text{Co}_1\text{Fe}_{0.1}\text{P}/\text{NF}$ at a current density of 20 mA cm^{-2} .

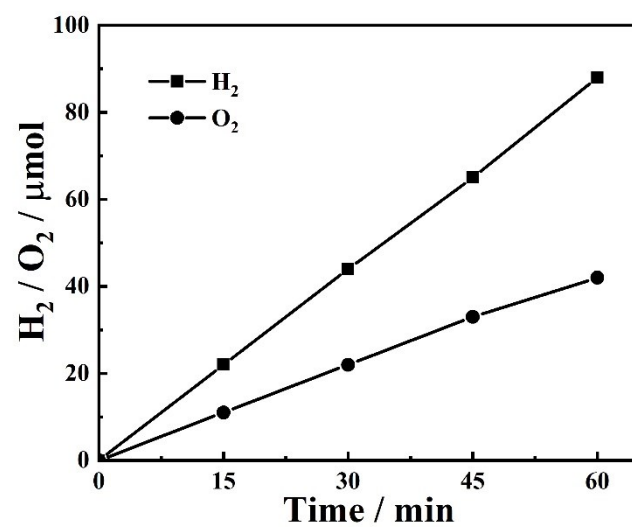


Fig. S5 H₂ and O₂ production over Co₁Fe_{0.1}P/NF at a current density of 20 mA cm⁻².

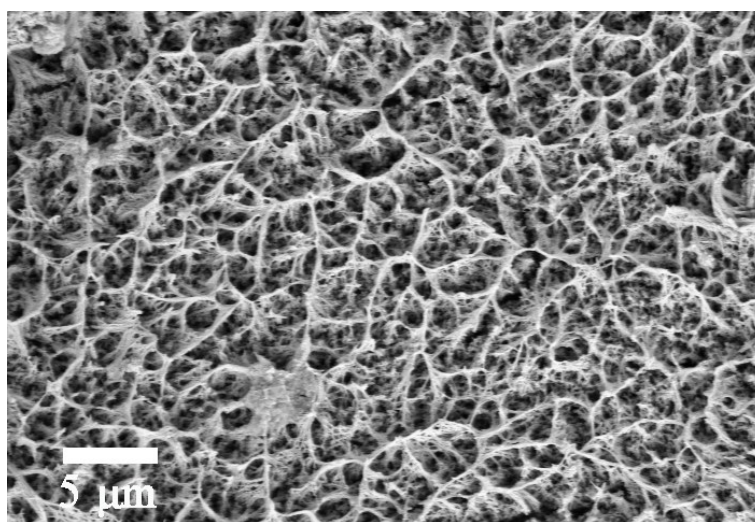


Fig. S6 SEM images of Co₁Fe_{0.1}P/NF catalyst after 24 h OER test.

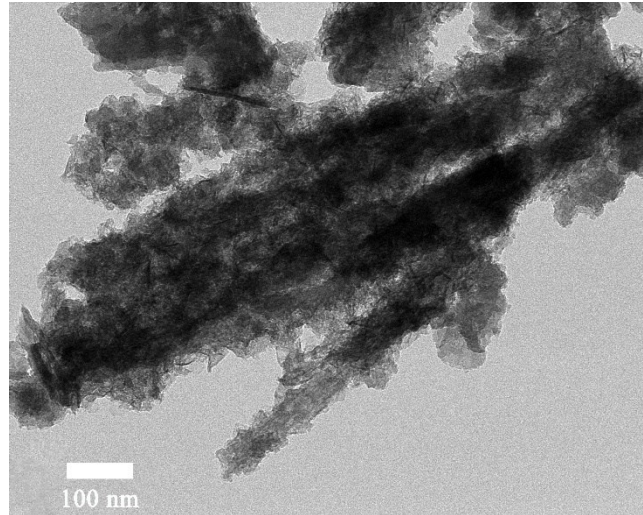


Fig. S7 TEM images of $\text{Co}_1\text{Fe}_{0.1}\text{P/NF}$ catalyst after 24 h OER test.

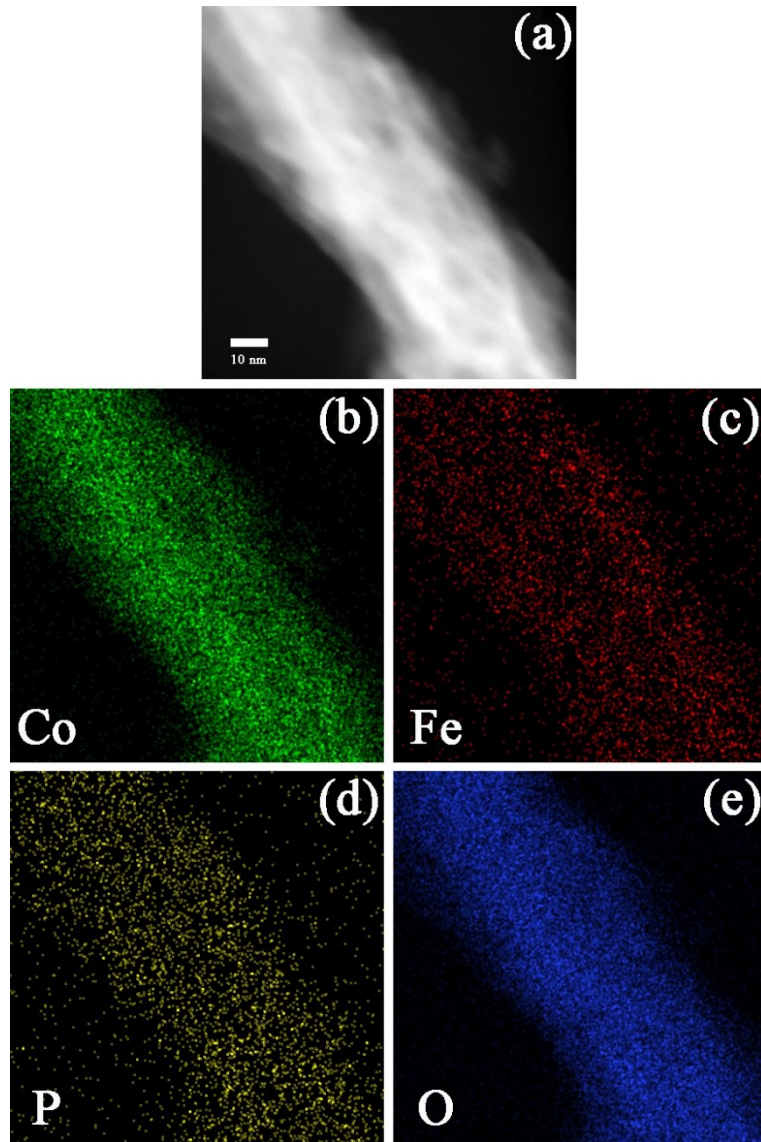


Fig. S8 Mapping images of $\text{Co}_1\text{Fe}_{0.1}\text{P/NF}$ catalyst after 24 h OER test.

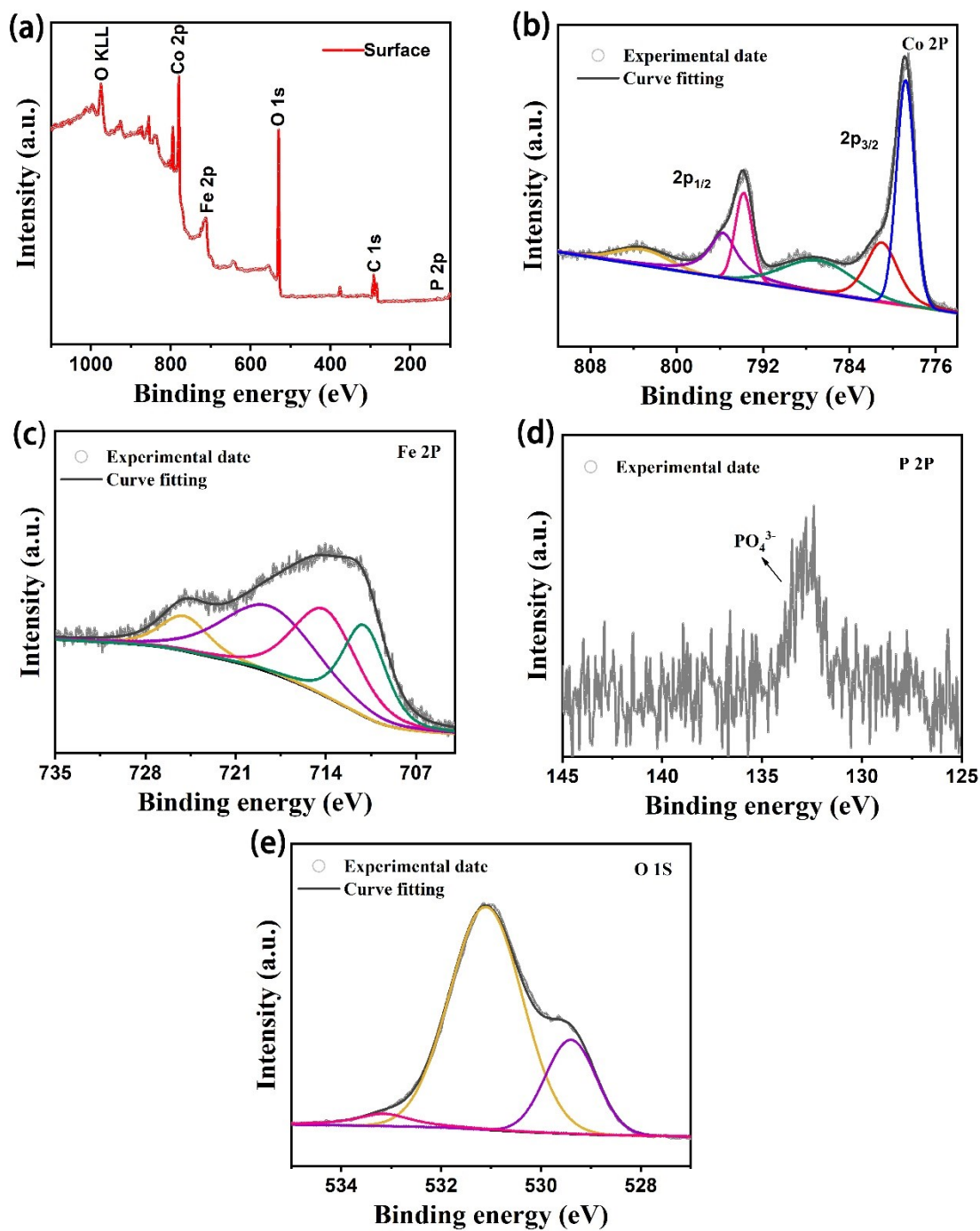


Fig. S9 (a) XPS survey spectrum of Co₁Fe_{0.1}P/NF nanowires after 24 h OER test; The XPS spectra of (b) Co 2p, (c) Fe 2p, and (d) P 2p in the Co₁Fe_{0.1}P/NF.

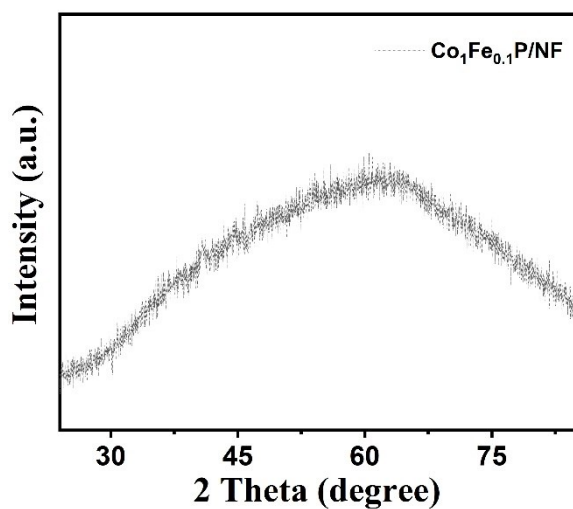


Fig. S10 XRD patterns of $\text{Co}_1\text{Fe}_{0.1}\text{P/NF}$ after 24 h OER test.

Table S1 Related parameters of nickel foam.

	thickness	Specific surface area	Porosity
NF	1.0 mm	$320 \text{ g/m}^2 \pm 20$	95%-98%

Table S2 Load capacity of different batches of the same sample.

	Quality (mg)	average value (mg)
NF	95.1/89.7/90.7	91.83
$\text{Co}_1\text{Fe}_{0.1}\text{P/NF}$	105.2/100.0/99.0	100.14
Sample load	10.1/10.3/8.3	9.57

Table S3 Inductively coupled plasma of 1M KOH after OER test.

Metal ions	P
Content (mg/L)	0.70

Table S4 Summary of TMP-based electrocatalysts for HER, OER and overall water splitting in 1 M KOH.

Catalyst	Electrolyte	Overpotential @J ₁₀ /mV for HER	Overpotential @ J ₁₀ /mV for OER	Overpotential @J ₁₀ /V for overall water splitting	Reference
CoFeP@Ru	1M KOH	94	270	1.60	[1]
Ni ₂ P/Cu ₃ P	1M KOH	88	262	1.60	[2]
A-NiSe ₂ P	1M KOH	111	272	1.62	[3]
P-Fe ₃ N@NC	1M KOH	102	270	1.61	[4]
MnCo ₂ O ₄ @Ni ₂ P	1M KOH	57	240	1.63	[5]
Ni ₂ P@rGO	1M KOH	142	260	1.61	[6]
CoP@3D Ti ₃ C ₂ -MXene	1M KOH	168	280	1.58	[7]
Co _{0.6} Fe _{0.4} P NPs	1M KOH	133	298	1.57	[8]
CoS _{0.46} P _{0.54}	1M KOH	101	302	1.62	[9]
Co ₁ Fe _{0.1} P/NF	1M KOH	73	257	1.60	This work