

Mild generation of surface oxygen vacancy on CeO₂ for improved CO₂ photoreduction activity

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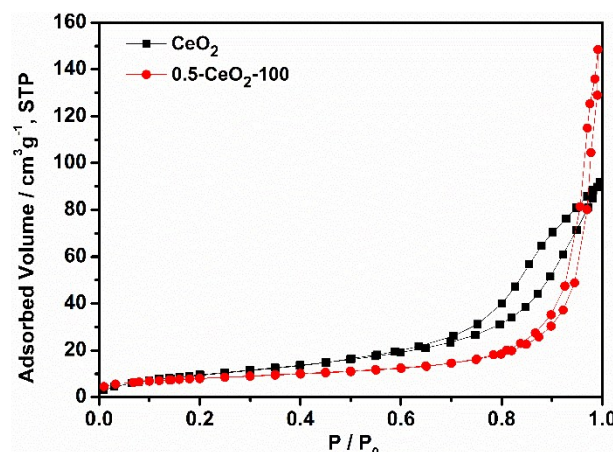


Fig. S1 The N₂ adsorption/desorption curves of CeO₂ and 0.5-CeO₂-100.

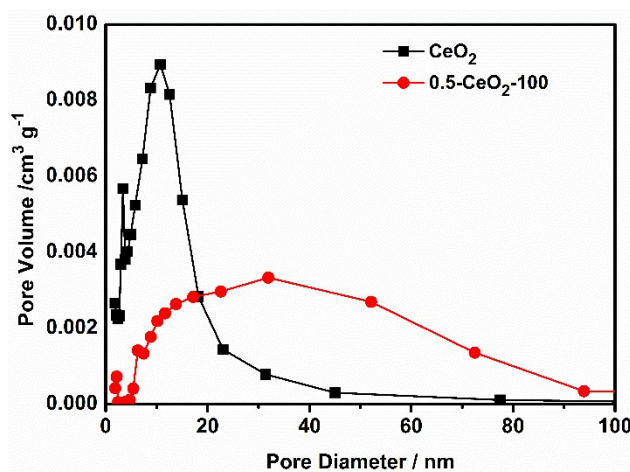


Fig. S2 The pore size distribution curves of CeO₂ and 0.5-CeO₂-100.

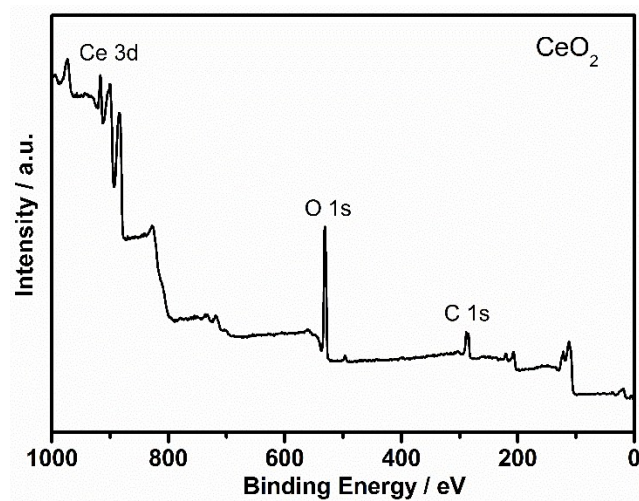


Fig. S3 The XPS survey spectrum of CeO_2 .

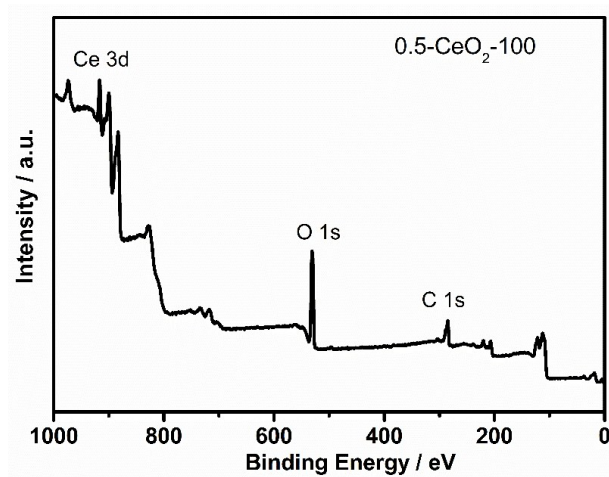


Fig. S4 The XPS survey spectrum of 0.5- CeO_2 -100.

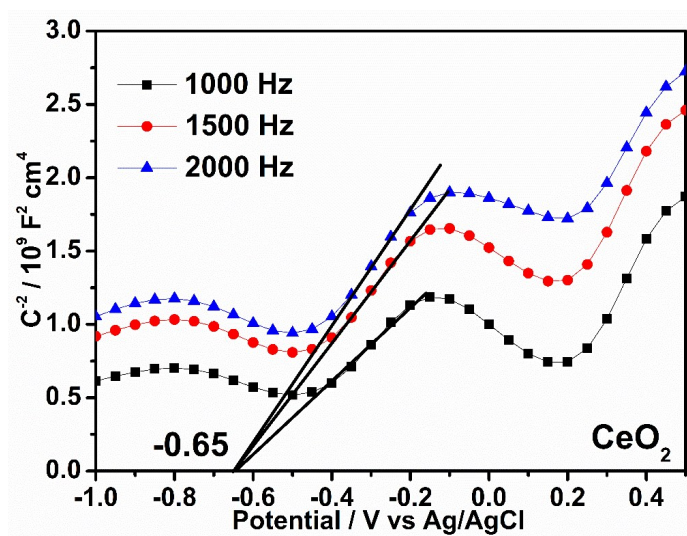


Fig. S5 The Mott-Schottky plots of CeO_2 at different frequencies.

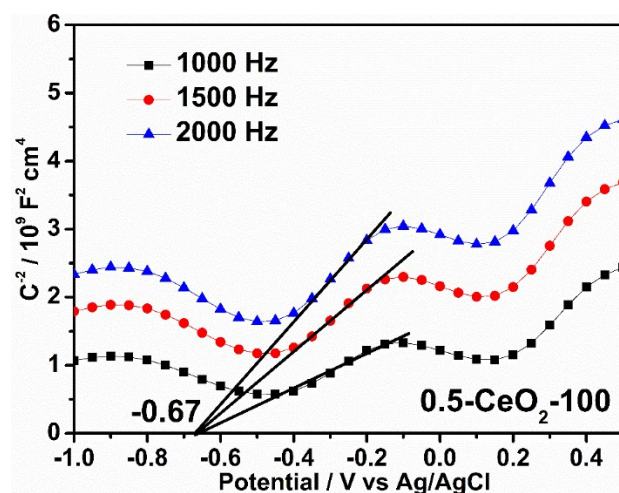


Fig. S6 The Mott-Schottky plots of 0.5-CeO₂-100 at different frequencies.

Table S1 The full width at half-maximum (FWHM) and oxygen vacancy concentration of CeO₂ and 0.5-CeO₂-100.

Samples	FWHM (cm ⁻¹)	oxygen vacancy concentration (× 10 ²⁰ cm ⁻³)
CeO ₂	50.1	0.02
0.5-CeO ₂ -100	27.3	1.06

Table S2 CO₂ photoreduction activity of CeO₂ previously reported and in this work.

Strategy	Product	The yield of product	Reference
Oxygen vacancy introduced CeO ₂	CO	1.4 μmol/g (5 h)	This work
Oxygen vacancy introduced CeO ₂	CH ₃ OH	5.8 μmol/g (10 h)	1
Cu ₂ O-decorated CeO ₂	CO	1.2 μmol/g (8 h)	2
Cr-doped CeO ₂	CO	16.2 μmol/g (8 h)	3
	CH ₄	10.1 μmol/g (8 h)	
Cu and oxygen vacancy introduced CeO ₂	CO	12.55 μmol/g (5 h)	4
Zr-doped CeO ₂	CO	8 μmol/g (8 h)	5
Pt-embedded CeO ₂	CO	6.0 mmol/min/g	6
	CH ₃ OH	32 μmol/g (4 h)	
CeO ₂ /Bi ₂ MoO ₆ heterojunction	C ₂ H ₅ OH	25 μmol/g (4 h)	7

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