

Supporting Information

Characteristics of crystalline sputtered LaFeO₃ thin film for photoelectrochemical water splitting photocathode

Min-Kyu Son^{1,4*}, Hyunwoong Seo², Motonori Watanabe¹, Masaharu Shiratani^{3,4}, Tatsumi Ishihara¹

¹Molecular Photoconversion Devices Research Division, International Institute for Carbon-Neutral Energy Research (I²CNER), Kyushu University, 744 Motooka, Nishi-ku, Fukuoka 819-0395, Japan

²Department of Energy Engineering, Inje University, 197 Inje-ro, Gimhae-si, Gyeongsangnam-do 50834, Republic of Korea

³Faculty of Information Science and Electrical Engineering, Kyushu University, 744 Motooka, Nishi-ku, Fukuoka 819-0395, Japan

⁴Center of Plasma Nano-interface Engineering, Kyushu University, 744 Motooka, Nishi-ku, Fukuoka 819-0395, Japan

*E-mail: minkyu.son@i2cner.kyushu-u.ac.jp

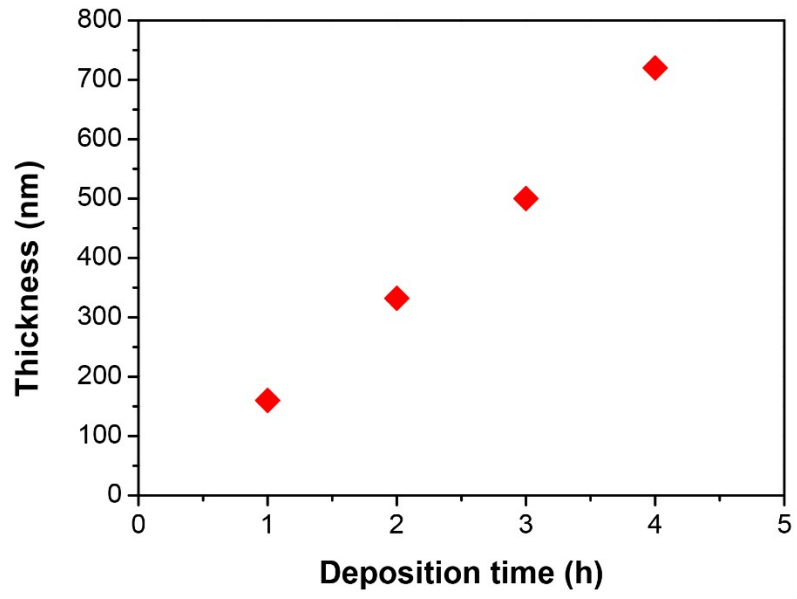


Figure S1. Thickness of sputtered LaFeO₃ film on Si wafer along with the deposition time.

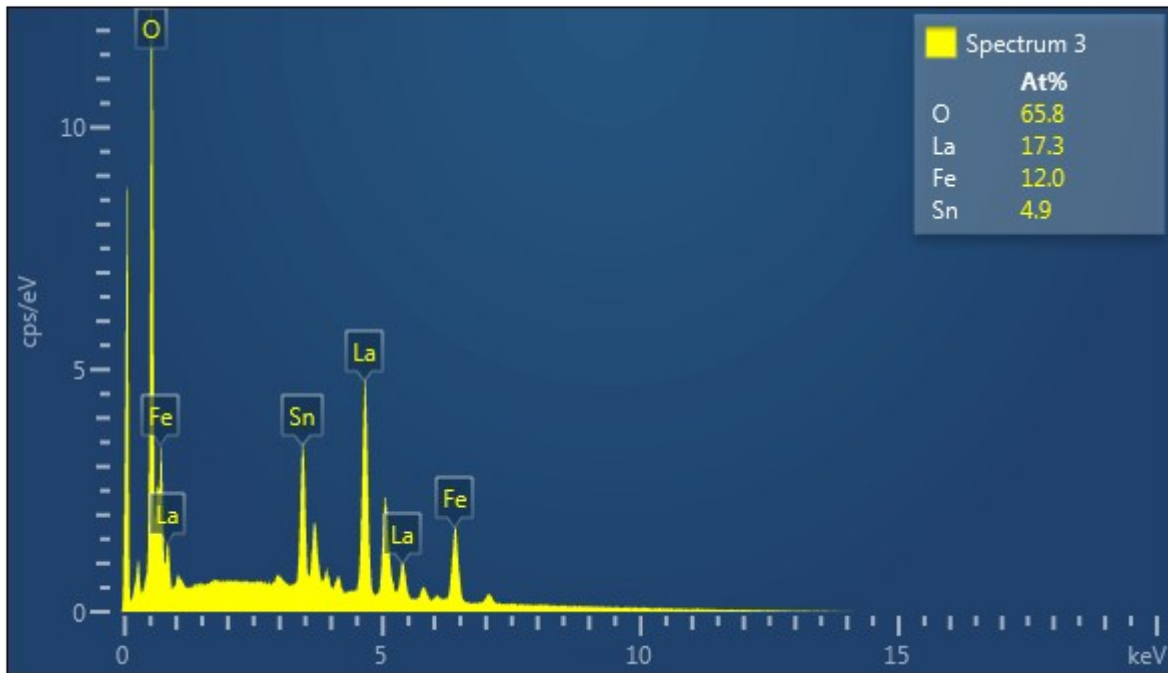


Figure S2. Compositional analysis of the crystalline sputtered LaFeO_3 film with post annealing process in air at 550 °C from EDX characterization.

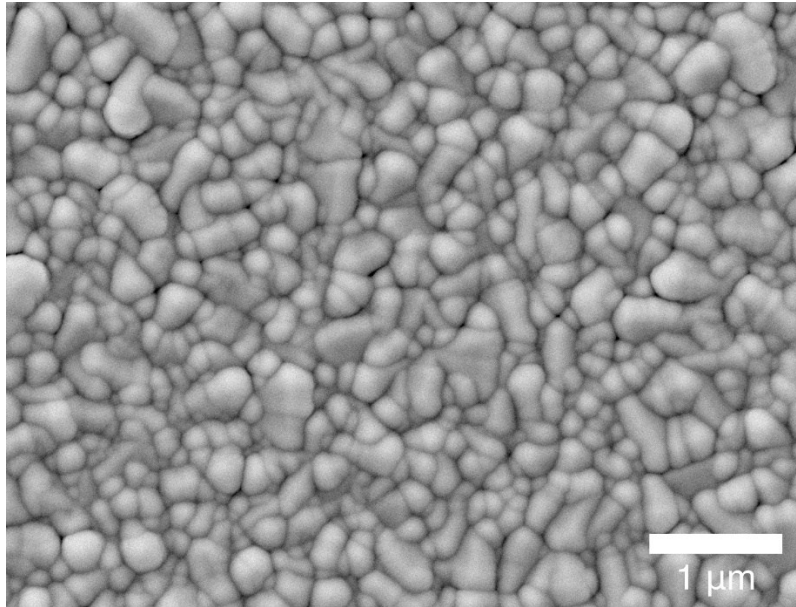


Figure S3. Top-view SEM image of the pristine sputtered LaFeO₃ film before post annealing process.

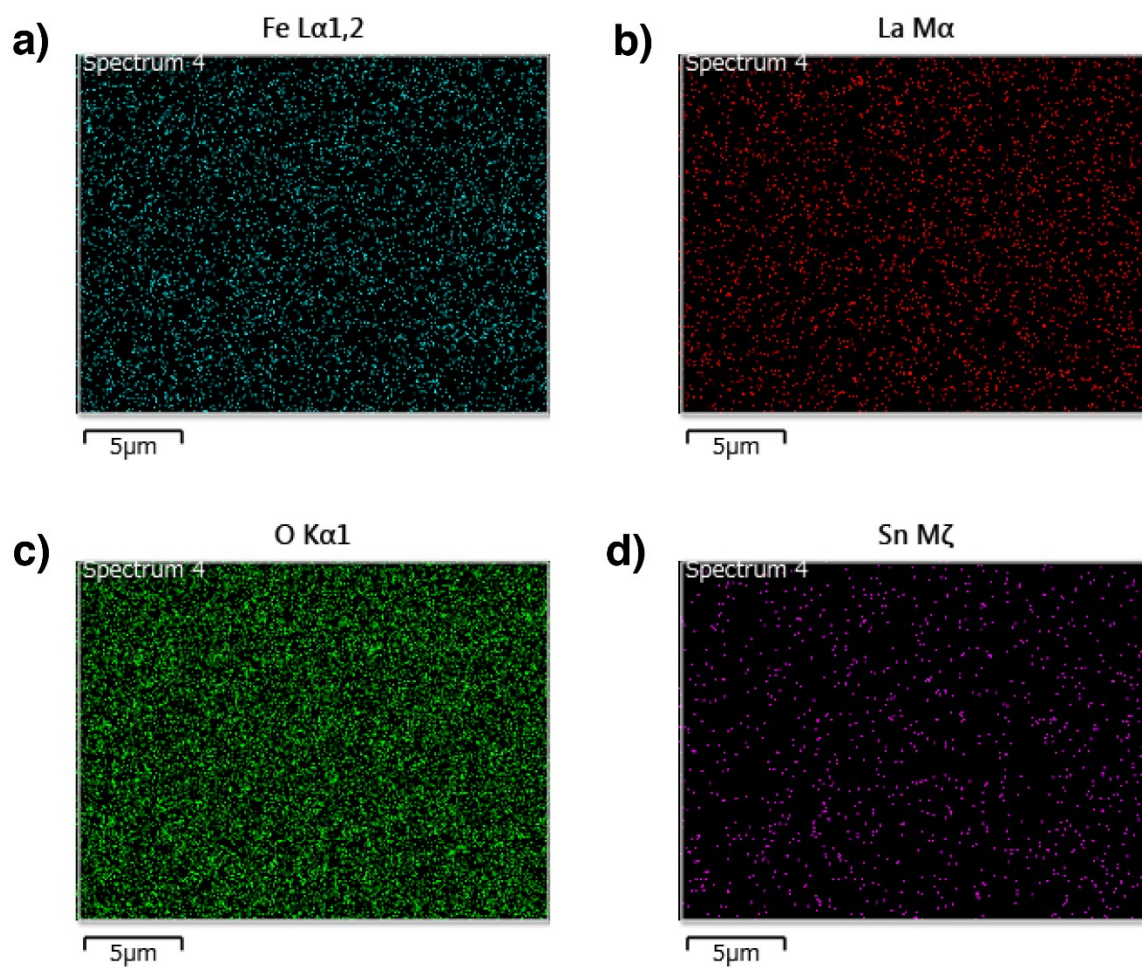


Figure S4. Element mapping of the crystalline sputtered LaFeO_3 film with post annealing process in air at 550 °C from EDX characterization: a) Fe, b) La, c) O and d) Sn, respectively.

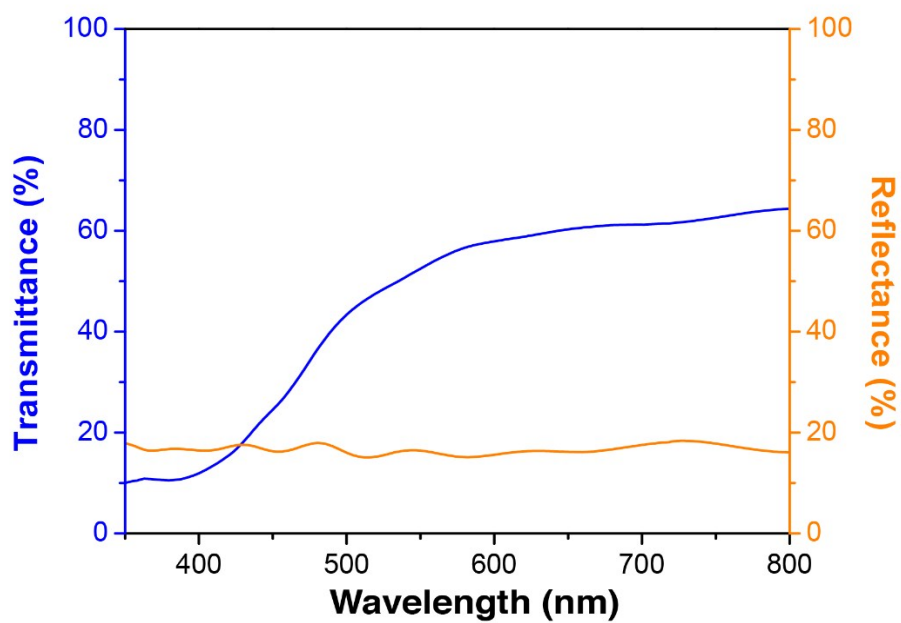


Figure S5. Transmittance and reflectance of the crystalline sputtered LaFeO₃ film on the FTO substrate.

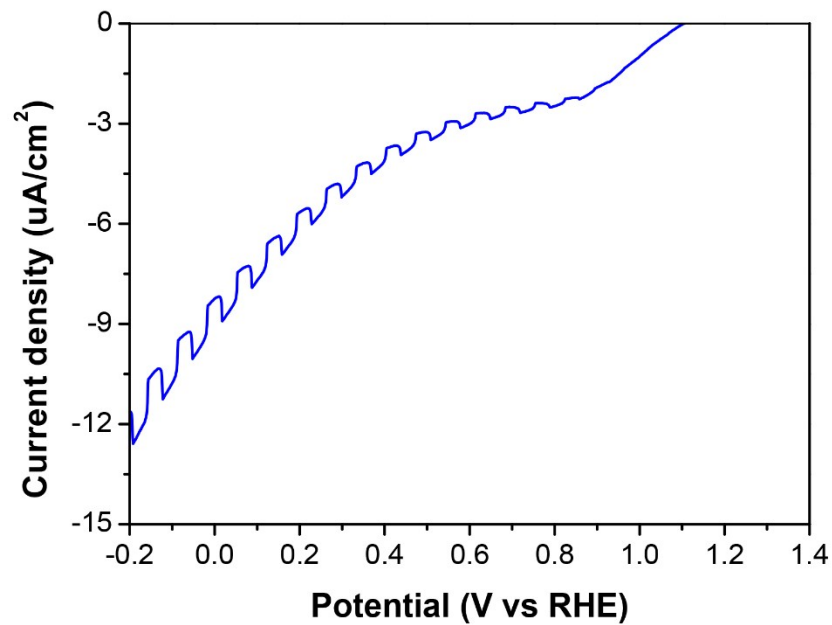


Figure S6. Current density-potential (J-E) characteristics of amorphous sputtered LaFeO₃ photocathode in strong alkaline solution (pH 13.6) under chopped one sun illumination.

Table S1. Photoelectrochemical performances of LaFeO₃ photocathode in other literatures and our work.

| Fabrication method | Photocathode composition | Electrolyte | Current density ($\mu\text{A cm}^{-2}$) | Onset potential (V vs RHE) | Stability | Ref. |
|----------------------|--|--|---|----------------------------|---------------------------------|----------|
| Electrodeposition | LaFeO ₃ | 0.1 M NaOH saturated with O ₂ , pH 13 (0.1 M NaOH, pH 13) | -95 at 0.75 V vs RHE (-1.5 at 0.5 V vs RHE) | 1.41 (1.27) | 1 hr (1 hr) | 23 |
| Electrodeposition | K-doped LaFeO ₃ | 0.1 M NaOH saturated with O ₂ , pH 13 (0.1 M NaOH, pH 13) | -268 at 0.6 V vs RHE (-11 at 0.6 V vs RHE) | 1.2 (1.2) | 15 hr (-) | 24 |
| Spray pyrolysis | LaFeO ₃ | 0.1 M NaOH, pH 13 | -160 at 0.26 V vs RHE | 1.2 | 20 hr (20% decrease) | 25 |
| Spray pyrolysis | Ag incorporated LaFeO ₃ | 0.1 M NaOH, pH 13 | -74 at 0.6 V vs RHE | 1.2 | 24 hr (5 % decrease) | 26 |
| Spray pyrolysis | Ni incorporated LaFeO ₃ | 0.1 M NaOH, pH 13 | -66 at 0.6 V vs RHE | 1.2 | 20 hr | 27 |
| Sol-gel spin coating | LaFeO ₃ with Au buffer layer | 0.1 M Na ₂ SO ₄ , pH 7 | -19 at 0.6 V vs RHE | 1.0 | 50 min | 28 |
| Sol-gel spin coating | Ni-B catalysed LaFeO ₃ | 0.1 M Na ₂ SO ₄ , pH 7 | -23 at 0.6 V vs RHE | 1.0 | 3.8 hr (82% decrease) | 29 |
| Spray pyrolysis | Dye-sensitized LaFeO ₃ (NiP catalysed dye sensitized LaFeO ₃) | 1.0 M KOH saturated with O ₂ , pH 13.6 (0.5 M Na ₂ SO ₄ , pH 3) | -150 at 0.75 V vs RHE (-20 at 0.6 V vs RHE) | 1.3 (1.1) | 400 sec (400 sec, 20% decrease) | 31 |
| RF sputtering | LaFeO ₃ | 0.1 M NaOH, pH 13.6 | -25 at 0V vs RHE | 1.3 | 30 min | Our work |