## Functionalized MXenes as Effective Polyselenides Immobilizer for Lithium-Selenium Batteries: A Density Functional Theory (DFT) Study

Rahul Jayan, Md Mahbubul Islam\*

Department of Mechanical Engineering, Wayne State University, Detroit, MI 48202, USA \*Corresponding author: <u>mahbub.islam@wayne.edu</u>

Table S1. Structural data of the bare and functionalized Mxenes

AM	Lattice Parameters (A <sup>o</sup> )		Bond length (A <sup>o</sup> )	
	a = b	Ti - C	Ti - Ti	Ti – S/O/F/Cl
Ti <sub>3</sub> C <sub>2</sub>	12.370	2.05	2.92	-
$Ti_3C_2S_2$	12.493	2.16	3.03	2.40
$Ti_3C_2O_2$	12.108	2.18	3.10	1.97
$Ti_3C_2F_2$	12.215	2.06	2.94	2.16
$Ti_3C_2Cl_2$	12.108	2.05	2.93	2.47

Table S2. The binding energies (in eV) of higher order polyselenides with the electrolyte solvents

Higher order Li <sub>2</sub> Se <sub>n</sub> polyselenides	DME	DOL
Li <sub>2</sub> Se <sub>4</sub>	0.87	0.82
Li <sub>2</sub> Se <sub>6</sub>	0.92	0.79
Li <sub>2</sub> Se <sub>8</sub>	0.90	0.83



Fig S1. Partial density of states (PDOS) of Ti<sub>3</sub>C<sub>2</sub>. The fermi level is indicated with a vertical line



Figure S2. Top and side views of the structures of Li<sub>2</sub>Se<sub>n</sub> adsorbed monolayer graphene



Figure S3. The side and top views of the most stable structures of  $Li_2Se_n$  adsorbed on  $Ti_3C_2$ 



Figure S4. The side and top views of the optimized structures of  $Ti_3C_2X_2$  (X = S, O, F, Cl)



Figure S5. The top views of the most stable structures of  $Li_2Se_n$  adsorbed on  $Ti_3C_2S_2$  and  $Ti_3C_2O_2$ .



Figure S6. The side and top views of the optimized structures of Li<sub>2</sub>Se<sub>n</sub> adsorbed Ti<sub>3</sub>C<sub>2</sub>F<sub>2</sub>



Figure S7. The side and top views of the most favorable configurations of  $Li_2Se_n$  adsorbed  $Ti_3C_2Cl_2$ 



Figure S8. The most stable geometric configurations of ( $Li_2Se_n \ge 4$ ) bound DOL/ DME solvents



Figure S9. Charge density difference of  $Li_2Se_n$  (n = 1,4,8) on graphene. The iso-surface level is set at 0.0020 e Å<sup>-3</sup>. The green and red colors denote charge accumulation and depletion, respectively.



Figure S10. Charge density difference of  $Li_2Se_n$  (n = 1,4,8) on  $Ti_3C_2F_2$  and  $Ti_3C_2Cl_2$ . The iso-surface level is set at 0.0020 e Å<sup>-3</sup>. The green and red colors denote charge accumulation and depletion, respectively.



Figure S11. Total DOS of  $Li_2Se_n$  (n = 1,4,8) adsorbed  $Ti_3C_2S_2$ .



Figure S12. Total DOS of  $Li_2Se_n$  (n = 1,4,8) adsorbed  $Ti_3C_2O_2$ 



Figure S13. Total DOS of  $Li_2Se_n$  (n = 1,4,8) adsorbed  $Ti_3C_2F_2$ 



Figure S14. Total DOS of  $Li_2Se_n$  (n = 1,4,8) adsorbed  $Ti_3C_2Cl_2$ 



 $Figure \ S15. \ Total \ DOS \ of \ Ti_3C_2-S/O \ and \ Ti_3C_2-S/O-Li_2Se \ calculated \ using \ PBE-GGA \ and \ GGA+U \ method$