Supporting Information

Synergistic Effects of Heteroatom decorated MXene catalyst for CO Reduction Reaction

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Figure S1. Adsorption free energy comparison between *CO and *OC on TM-HACs.



Figure S2. Top and side view of pristine and *CHO on TM-HACs, where Ag, Cd and Au atoms are drawn in blue, green and yellow.



Figure S3. (a) The top view of Mo-HAC, where the *S* represents the area enclosed by six Oxygen atoms near heteroatom active center. They are 22.82, 23.66, 23.72 and 23.84 Å² for pristine, *CO, *COH and *CHO Mo-HAC, respectively.



Figure S4. The scaling relationships between $\Delta G_{ads}(*CO)$ and $\Delta G_{ads}(*COH)$ or $\Delta G_{ads}(*CHO)$, respectively.



Figure S5. Reaction free energy comparison between $\Delta G(*COH \rightarrow *C)$ and $\Delta G(*COH \rightarrow *CHOH)$.



Figure S6. Free energy diagram for CORR with all possible reaction intermediates on Ir, Ru, Os, and Rh-HAC, respectively. The energetically favorite route was drawn in blue, and yellow for the undesirable products.



Figure S7. Free energy diagram for CORR under solvation condition with all possible reaction intermediates on Ir, Mo, Ru, W, Os, and Rh-HAC, respectively. The energetically favorite route was drawn in blue, and yellow for the undesirable products.



Figure S8. The calculated projected DOSs of pristine Mo₂TiC₂O₂ layer, Mo-HAC, and W-HAC.



Figure S9. Comparison adsorption free energies of H_2O , H, and CO on the selected candidates.



Figure S10. DFT calculated dissolution potentials (U_{diss}) of these selected candidates, comparing with that of the synthesized Pt-HAC.





Figure S11. Fluctuation of temperature and distance between anchor TM and the middle Ti-layer (*d*) versus time for TM-HACs, respectively. The snapshots at the beginning and the end of the trajectory are also shown.

Adsorbed species	$E_{\rm ZPE}({ m eV})$	E _{TS} (eV)
СО	0.14	0.62
H_2	0.27	0.41
H ₂ O	0.60	0.59
CH ₄	1.19	0.57
CH ₃ OH	1.37	0.74
*C0	0.21	0.12
*СОН	0.50	0.12
*СНО	0.44	0.13
*C	0.10	0.02
*СНОН	0.80	0.10
*CH	0.35	0.05
*CH ₂	0.62	0.06
*CH ₃	0.94	0.08
*CH4	1.20	0.27
*CH ₂ OH	1.08	0.17
*CH ₃ OH	1.36	0.21

 Table S1 Calculated ZPE and TS energies for different species, where the label *

 denotes the status of adsorption.