

Electronic Supplementary Information for the manuscript

Unveiling the origin of room-temperature ferromagnetism in monolayer VSe₂: the role of extrinsic effects

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S1. Vibrational experiments on chemisorption on monolayer VS₂

To elucidate the eventual formation of surface vanadium-oxide phases, we have carried out a vibrational investigation by high-resolution electron energy loss spectroscopy (HREELS) on monolayer VSe₂ on graphite, which was prepared using the exact experimental procedures reported in the Experimental methods of the paper by Bonilla et al. ¹.

The use of HREELS provides unambiguous fingerprint regarding surface chemical bonds and adsorption at solid surfaces. Specifically, the inspection of the HREELS spectrum is a particularly powerful tool to identify vanadium-oxide phases and, correspondingly, the oxidation status of V atoms. As a matter of fact, VO₂ is characterized by surface phonons at 55 and 95 meV ²; V₂O₃ by an intense feature at 68 meV ³; and V₂O₅ by several modes at 45, 74, 105, and 126 meV ⁴. However, the vibrational spectrum in Fig. R1 does not exhibit any feature even after exposure to 10⁵ L of O₂. This unambiguously excludes the formation of surface vanadium-oxide phases, that, if existing, do not cover more than 1-2% of the surface area (corresponding to the minimal sensitivity of HREELS technique).

Notably, the vibrational spectrum remains featureless even after exposure to 10⁵ L of H₂, H₂O and CO. The absence of adsorbate-related vibrational modes proves the chemical inertness of monolayer VSe₂ to most common gases, thus experimentally validating our theoretical model concerning chemical reactivity.

We also recorded the vibrational spectrum for the surface kept one year in air. The spectrum is dominated by C-H bending and stretching vibrations at 171 and 365 meV ⁵, respectively. This could

arise from hydrocarbon contamination from ambient air⁶ adsorbed at Se vacancies of the VSe₂ monolayer.

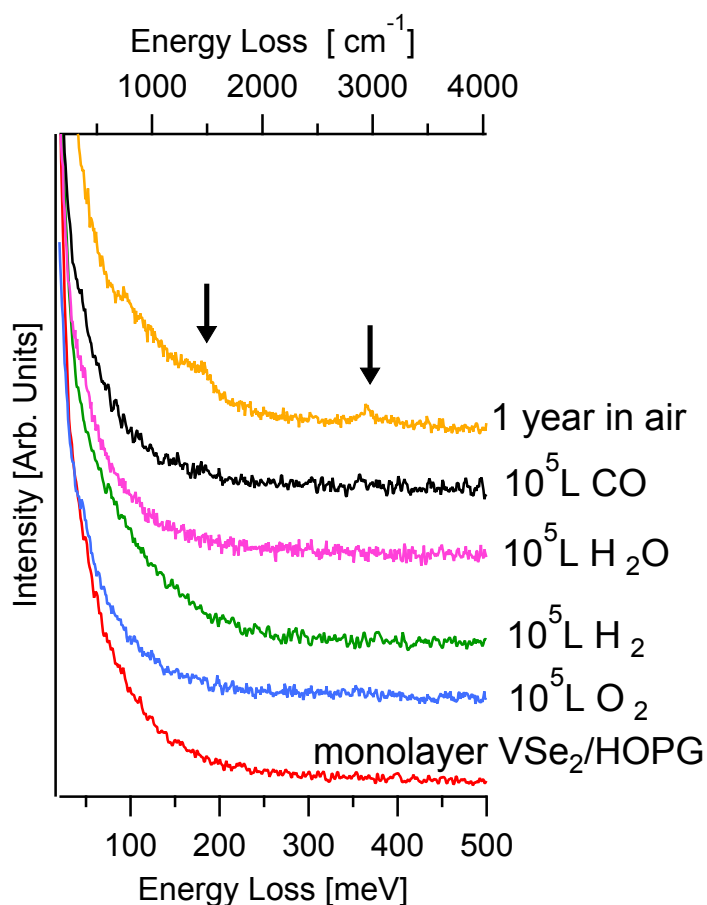


Figure S1: Vibrational spectra for monolayer VSe₂ grown on highly oriented pyrolytic graphite (HOPG) and the same surface modified by the exposure of O₂, H₂, H₂O and CO. The vibrational spectrum for the same surface kept one year in air is also reported.

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