

Supporting Information for

**One-step phosphating synthesis of CoP
nanosheet arrays combine with Ni₂P as
high-performance electrode for
supercapacitor**

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The synthesis of RGO electrode

The negative electrode was prepared as follows: reduced graphene oxide (RGO), acetylene black and polyvinylidene fluoride (PVDF) were dispersed by a mass ratio of 8: 1: 1 in N-methyl pyrrolidone (NMP) into the homogenous slurry. Then the slurry was coated on bare Ni foam ($1 \times 2 \text{ cm}^2$), dried in a vacuum and pressed under the pressure of 10 MPa.

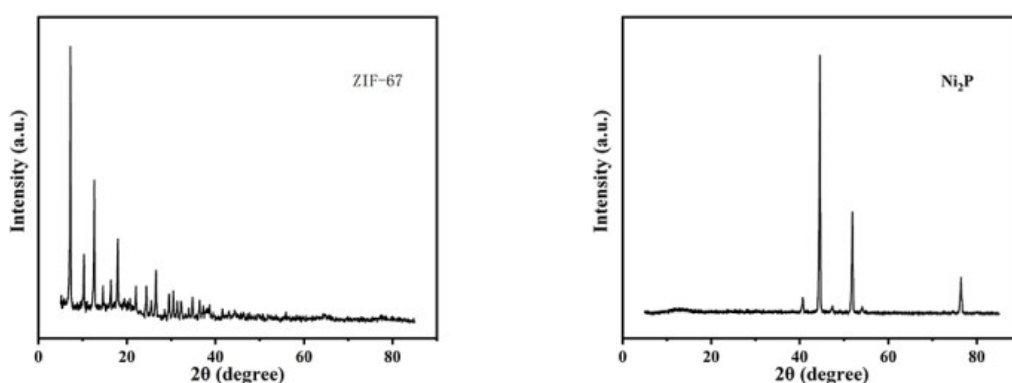


Fig. S1. XRD pattern of ZIF-67 precipitate and Ni₂P/NF.

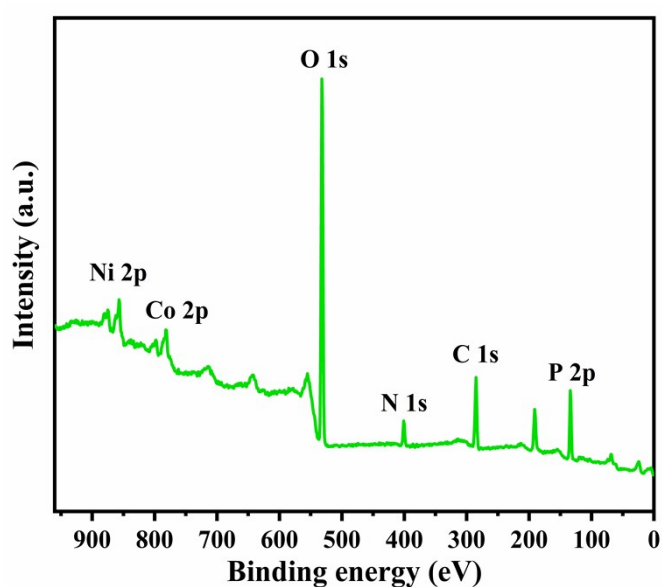


Fig. S2. wide-scan XPS spectrum of NCP/NF

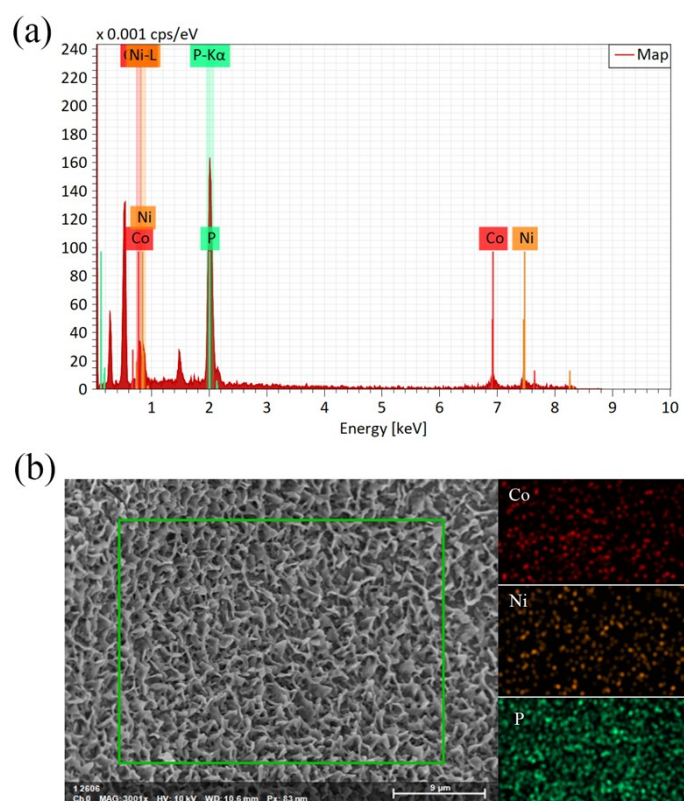


Fig. S3. EDS of NCP/NF.

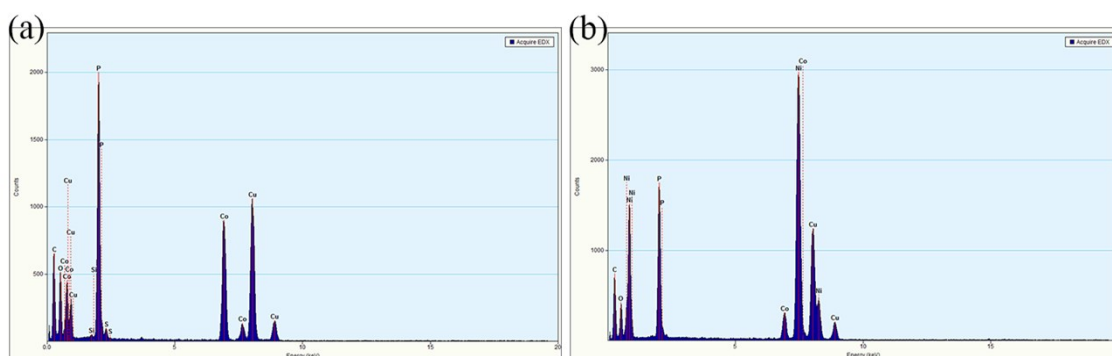


Fig. S4. EDS of (a) CoP and (b) Ni₂P.

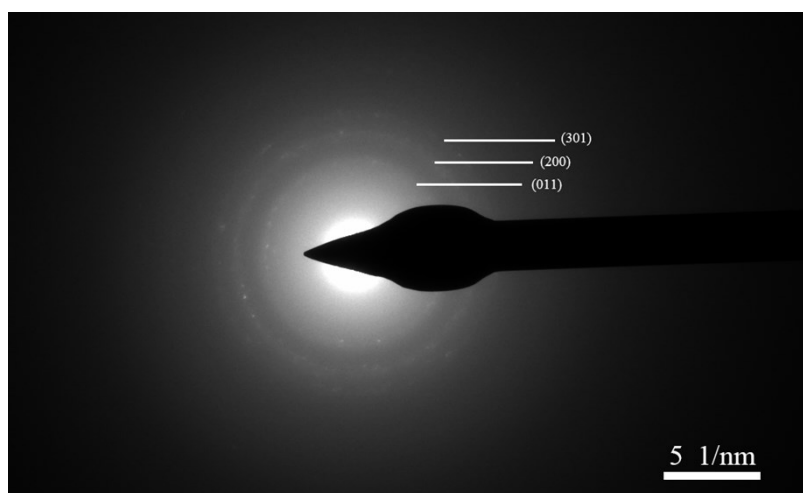


Fig. S5 The selected area electron diffraction (SAED) of CoP

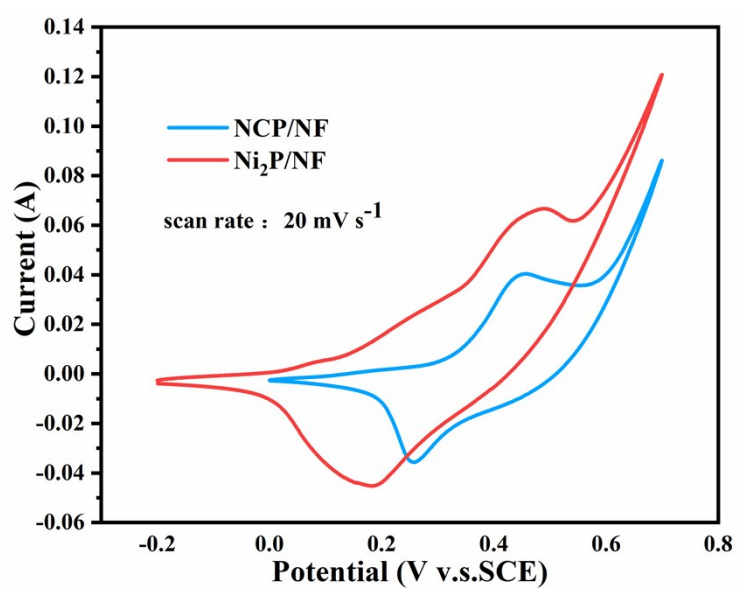


Fig. S6 CV curves of NCP/NF and $\text{Ni}_2\text{P/NF}$ at 20 mV s^{-1} .

Electrode	Synthetic method	Capacitance/capacity	Reference
Ni ₂ P-NF	Hydrothermal and CVD	2031 mF cm ⁻² at 1mA cm ⁻²	48
CoP	mechanical alloying method	447.5 F g ⁻¹ at 1 A g ⁻¹	51
CoP Microcubes	low-temperature phosphorization	560 F g ⁻¹ at 1 A g ⁻¹	52
urchin-like Ni ₂ P/CoP	CVD	1038 F g ⁻¹ at 1 A g ⁻¹	56
Co ₂ P nanoflowers	Chemical method	416 F g ⁻¹ at 1 A g ⁻¹	57
Hollow Co ₂ P nanoflowers	Thermal decomposition	412.7 F g ⁻¹ at 1 A g ⁻¹	58
This work	CVD	1.43 C cm ⁻² (3575 mF cm ⁻²)	

Table 1. Materials for transition metal phosphides electrodes prepared by different methods and their electrochemical performance as a super capacitor.

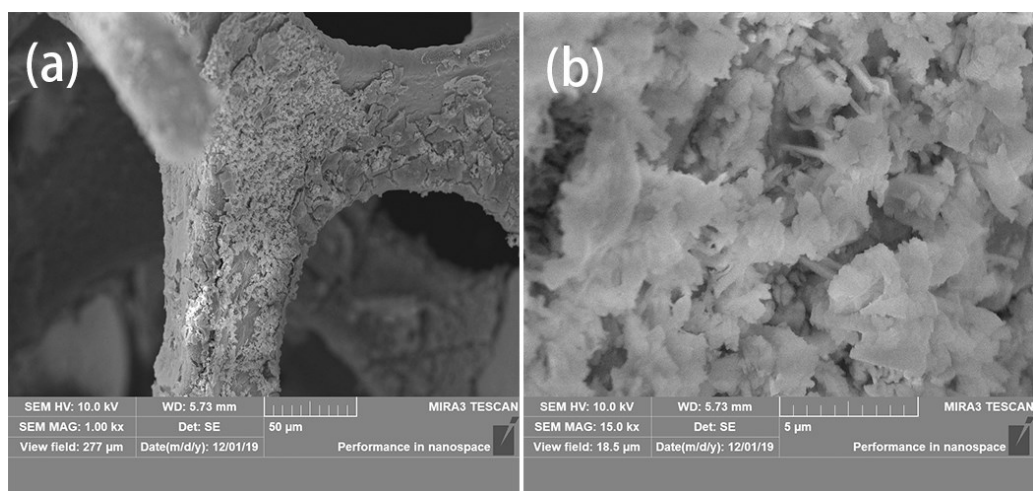


Fig. S7. SEM image of NCP/NF after 5000 cycles at different magnification.

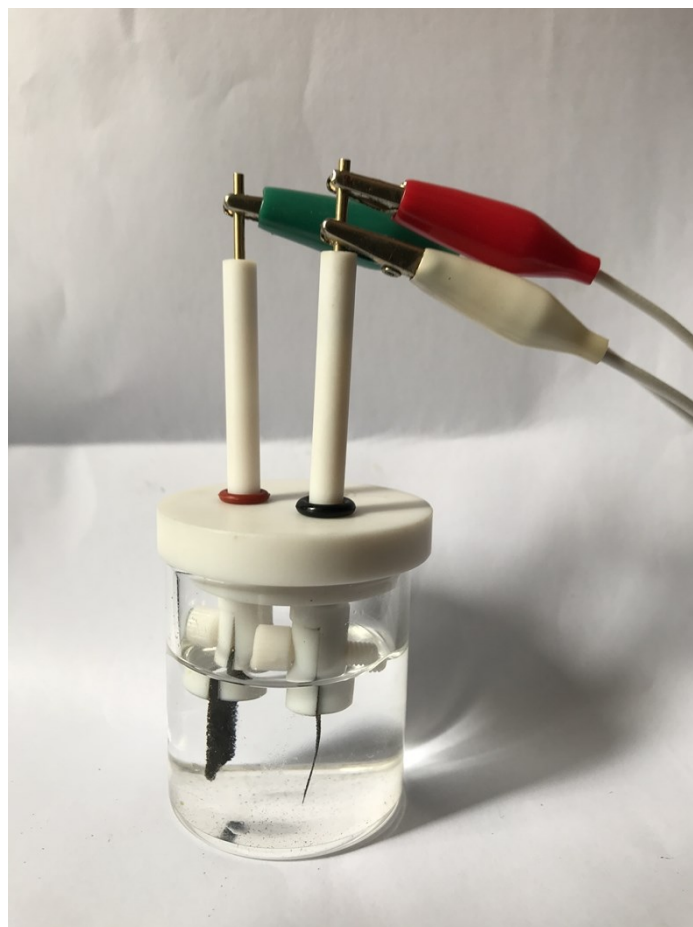


Fig. S8 The digital photograph of ASC.

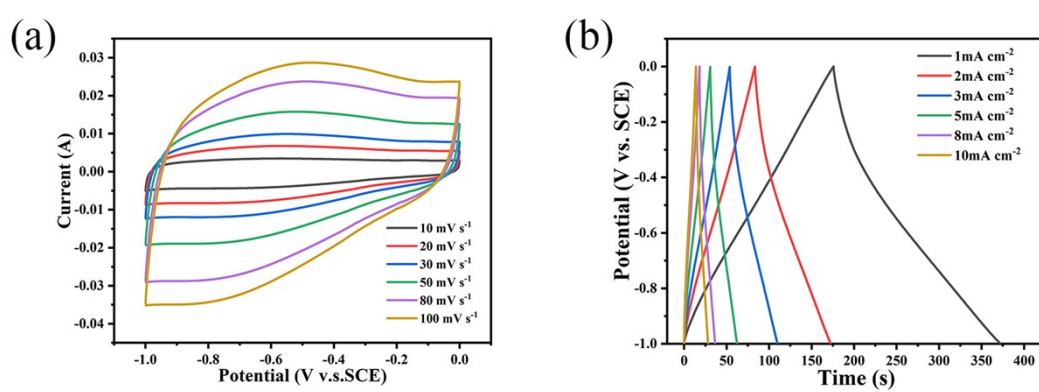


Fig. S9. (a) CV and (b) GCD curves of the RGO electrode

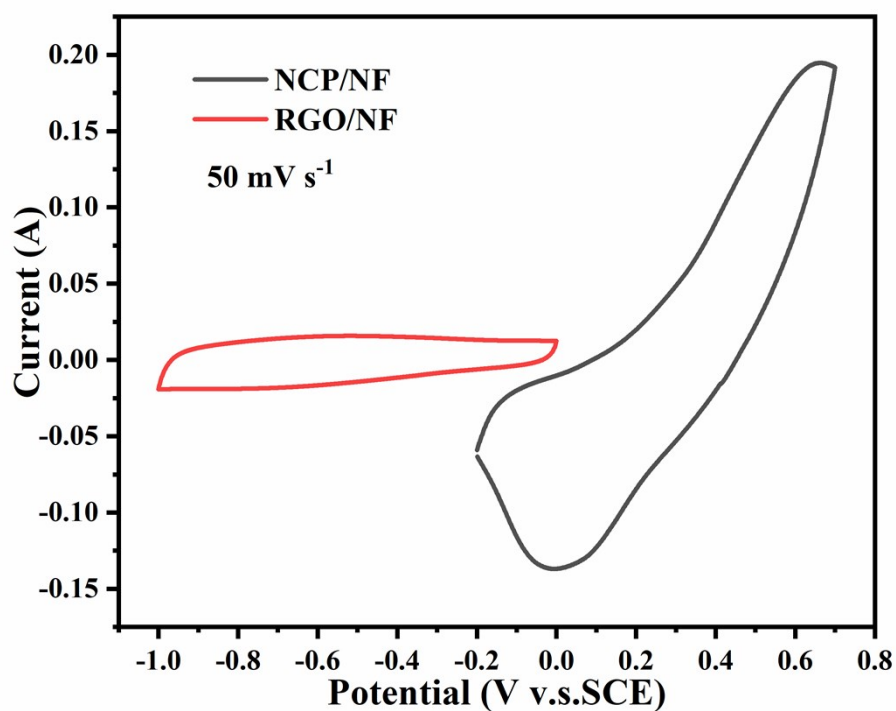


Fig. S10. CV curves of the anode and cathode

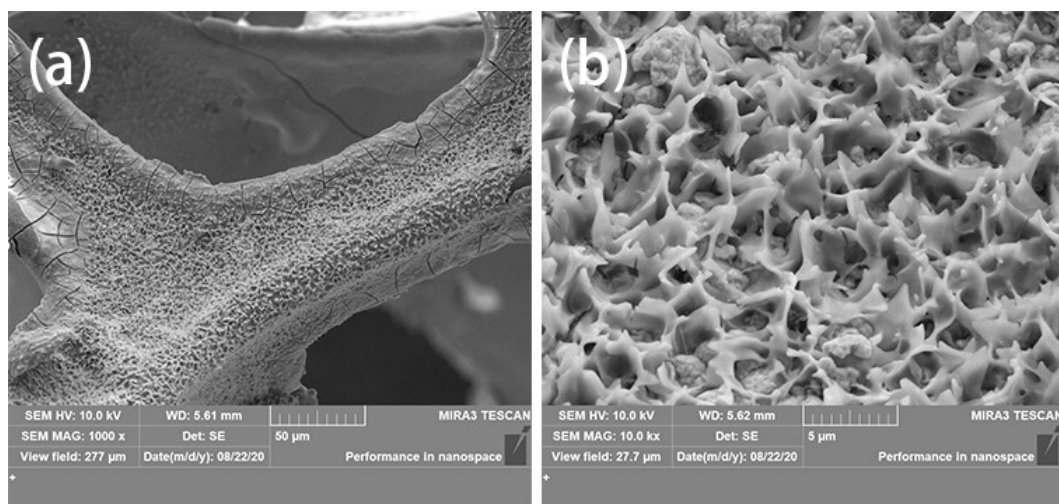


Fig. S11. SEM image of ASC after 5000 cycles.

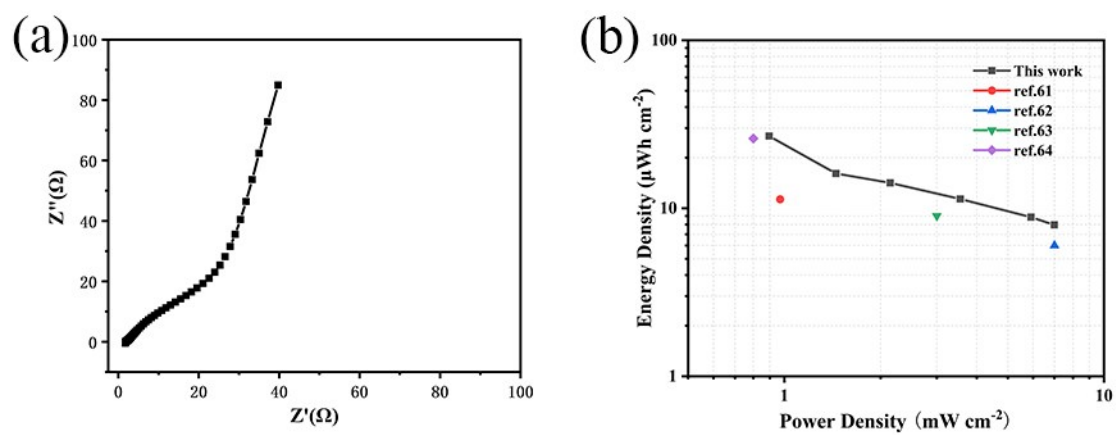


Fig. S12. The (a) Nyquist plots and (b) Ragone plot of ACS.