## **Supporting Information**

## Computational screening of pristine and functionalized ordered TiVC

## MXenes as highly efficient anode materials for lithium-ion batteries

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**Figure S1.** Side views for the most stable configurations of (a)  $Ti_2C$ , (b)  $Ti_2CO_2$ , (c)  $Ti_2CS_2$ , (d)  $Ti_2CF_2$ , (e)  $Ti_2C(OH)_2$ , and (f)  $V_2C$ , (g)  $V_2CO_2$ , (h)  $V_2CS_2$ , (i)  $V_2CF_2$  and (j)  $V_2C(OH)_2$  monolayers after relaxation, respectively (Referring to Figure 1 for interpretation of atomic color).



Figure S2. The calculated phonon dispersion curves of TiVC monolayer.



**Figure S3.** The pristine and functionalized TiVC MXenes after *ab initio* molecular dynamics simulations of 20 ps.



**Figure S4.** The density of states (DOS) for (a) TiVC, (b) TiVCO<sub>2</sub>, (c) TiVCS<sub>2</sub>, (d) TiVCF<sub>2</sub>, and (e) TiVC(OH)<sub>2</sub> monolayers. Note that the Fermi levels are set to zero.



**Figure S5.** The diffusion pathway of Li ion on (a) Ti and (b) V surfaces of pristine TiVC monolayer, (c) Ti and (d) V surfaces of TiVCF<sub>2</sub> monolayer, (e) Ti and (f) V surfaces of TiVCO<sub>2</sub> monolayer, and (g) Ti and (h) V surfaces of TiVCS<sub>2</sub> monolayer (Referring to Figure 1 for interpretation of atomic color).



**Figure S6.** The average adsorption energy of multi-Li ions on favorable adsorption sites of TiVC, TiVCO<sub>2</sub> and TiVCS<sub>2</sub> monolayers.

Materials		O-group	S-group	F-group	OH-group
т: ст	charges	-1.097	-0.716	-0.747	-0.753
$\Pi_2 \subset \Gamma_2$	<i>r</i> <sub>(Ti</sub> − <sub>C)</sub> , <i>r</i> <sub>(Ti</sub> − <sub>T)</sub>	2.188, 1.974	2.148, 2.381	2.097, 2.155	2.104, 2.163
TiVCT <sub>2</sub>	charges	-1.077	-0.697	-0.733	-0.744
(Ti-surface)	<b>r</b> <sub>(Ti</sub>	2.189, 1.932	2.114, 2.370	2.065, 2.134	2.072, 2.139
TiVCT <sub>2</sub>	charges	-0.990	-0.611	-0.715	-0.721
(V-surface)	$r_{(V-C)}, r_{(V-T)}$	2.152, 1.891	2.063, 2.325	2.013, 2.138	2.023, 2.130
V <sub>2</sub> CT <sub>2</sub>	charges	-0.984	-0.579	-0.674	-0.721
	$r_{(V-C)}, r_{(V-T)}$	2.057, 1.960	2.020, 2.336	2.073, 2.123	1.987, 2.111

**Table S1** The structure characteristics and the amount of obtained charges, for T-groups on  $Ti_2CT_2$ ,  $TiVCT_2$  and  $V_2CT_2$  (T = O, S, F, or OH) monolayers.

**Table S2** The adsorption energy of single Li ion on different sites of TiVC and TiVCT<sub>2</sub> (T = O, S, F, or OH) monolayers, and the bracket indicates the Li ion adsorption site after relaxation. The loss of charge amount after Li ion adsorption on surface of TiVC and TiVCT<sub>2</sub> (T = O, S, F, or OH) monolayers.

	Adsorption energy (eV)			Transfer charge (e <sup>-</sup> )		
Adsorption Site	C-site	H-site	T-site	C-site	H-site	<b>T-site</b>
Ti <sub>2</sub> C	-0.706	-0.721	-0.700(C-H)	0.835	0.835	0.835
TiVC (Ti-Surface)	-0.749	-0.735	-0.749(C)	0.837	0.838	0.837
TiVC (V-Surface)	-0.839	-0.845	-0.845(H)	0.836	0.835	0.836
V <sub>2</sub> C	-0.911	-0.891	-0.911(C)	0.837	0.839	0.838
Ti <sub>2</sub> CO <sub>2</sub>	-1.974	-1.974(C)	-1.658	0.882	0.883	0.904
TiVCO <sub>2</sub> (Ti-Surface)	-2.450	-2.340(T)	-2.340	0.885	0.898	0.896
TiVCO <sub>2</sub> (V-Surface)	-2.643	-1.826	-2.446	0.885	0.915	0.897
V <sub>2</sub> CO <sub>2</sub>	-2.856	-2.974(C)	-2.685	0.887	0.886	0.898
Ti <sub>2</sub> CS <sub>2</sub>	-1.704	-1.786(T)	-1.786	0.882	0.874	0.875
TiVCS <sub>2</sub> (Ti-Surface)	-1.763	-1.865(T)	-1.865	0.881	0.872	0.872
TiVCS <sub>2</sub> (V-Surface)	-1.218	-1.340(T)	-1.340	0.887	0.874	0.874
V <sub>2</sub> CS <sub>2</sub>	-0.960	-0.960(C)	-1.085	0.888	0.888	0.873
Ti <sub>2</sub> CF <sub>2</sub>	-0.778	-0.778(C)	-0.625	0.896	0.896	0.905
TiVCF <sub>2</sub> (Ti-Surface)	-0.881	-0.769(T)	-0.769	0.898	0.905	0.904
TiVCF <sub>2</sub> (V-Surface)	-1.508	-1.390(T)	-1.390	0.894	0.902	0.902
V <sub>2</sub> CF <sub>2</sub>	-1.101(T)	-1.187	-1.094	0.901	0.897	0.901
Ti <sub>2</sub> C(OH) <sub>2</sub>	0.768	0.803	0.765	-0.101	0.072	-0.088
TiVC(OH) <sub>2</sub> (Ti-	0.720	0.750	0.717	0 102	0 1 1 5	0.006
Surface)	0.720	0.730	0./1/	-0.102	0.113	-0.090
TiVC(OH) <sub>2</sub> (V-Surface)	0.794	0.826	0.789	-0.013	0.131	0.010
V <sub>2</sub> C(OH) <sub>2</sub>	0.670	0.670	0.633	0.247	0.212	0.094

**Table S3** The classical diffusion constant (k), with and without Wigner ZPE-tunneling corrected diffusion constant (kwig/tunn) and percentage of tunneling at three different temperatures (T=100, 200 and 300 K) for Li ion on pristine and functionalized  $Ti_2C$ , TiVC and V<sub>2</sub>C monolayers.

Pristino	т	K (s-1)	K (s-1)	QMT
I I Isune	1	<b>N</b> (S <sup>-</sup> )	Kwig/tunn (S <sup>-</sup> )	(%)
	100	$2.007 \times 10^{11}$	$2.032 \times 10^{11}$	1.230
Ti <sub>2</sub> C	200	$5.078 \times 10^{11}$	5.146×10 <sup>11</sup>	1.321
	300	6.920×10 <sup>11</sup>	6.976×10 <sup>11</sup>	0.803
TiVC (Ti	100	3.754×10 <sup>11</sup>	$4.031 \times 10^{11}$	6.872
five (11-	200	$8.962 \times 10^{11}$	9.190×10 <sup>11</sup>	2.481
Surface)	300	$1.198 \times 10^{12}$	$1.213 \times 10^{12}$	1.237
	100	$6.062 \times 10^{11}$	6.646×10 <sup>11</sup>	8.787
TiVC (V-Surface)	200	$1.366 \times 10^{12}$	$1.386 \times 10^{12}$	1.443
	300	$1.791 \times 10^{12}$	$1.799 \times 10^{12}$	0.445
	100	$2.803 \times 10^{11}$	$3.787 \times 10^{11}$	25.984
V <sub>2</sub> C	200	8.945×10 <sup>11</sup>	9.990×10 <sup>11</sup>	10.460
	300	$1.317 \times 10^{12}$	$1.393 \times 10^{12}$	5.456

F-group	Т	<i>k</i> in s <sup>-1</sup>	K <sub>wig/tunn</sub> in s <sup>-1</sup>	Percentage of QMT
	100	0.762	1.039	26.603
Ti <sub>2</sub> CF <sub>2</sub>	200	1.277×10 <sup>6</sup>	$1.341 \times 10^{6}$	4.773
	300	$1.517 \times 10^{8}$	$1.541 \times 10^{8}$	1.557
TiVCE. (Ti	100	$1.823 \times 10^{2}$	$2.600 \times 10^{2}$	29.885
$\frac{11VC\Gamma_2(11-}{Surface})$	200	$3.781 \times 10^{7}$	$4.076 \times 10^{7}$	7.237
Surface)	300	2.239×10 <sup>9</sup>	$2.310 \times 10^{9}$	3.074
	100	5.481×10	7.539×10	27.298
TiVCF <sub>2</sub> (V-Surface)	200	$2.153 \times 10^{7}$	$2.265 \times 10^{7}$	4.945
	300	$1.577 \times 10^{9}$	$1.604 \times 10^{9}$	1.683
	100	$2.385 \times 10^{4}$	$6.500 \times 10^{3}$	/
V <sub>2</sub> CF <sub>2</sub>	200	$2.285 \times 10^{8}$	$1.421 \times 10^{8}$	/
	300	4.854×10 <sup>9</sup>	3.829×10 <sup>9</sup>	/

O-group	Т	<i>k</i> in s <sup>-1</sup>	K <sub>wig/tunn</sub> in s <sup>-1</sup>	Percentage of QMT
	100	1.794×10 <sup>-5</sup>	1.902×10-3	99.057
Ti <sub>2</sub> CO <sub>2</sub>	200	2.942×10 <sup>3</sup>	$1.172 \times 10^{5}$	97.490
	300	1.610×10 <sup>6</sup>	5.139×10 <sup>7</sup>	96.867
	100	2.431×10	2.665×10	8.780
$\frac{11000}{2}(11-$	200	$1.137 \times 10^{7}$	$1.129 \times 10^{7}$	/
Surface)	300	8.821×10 <sup>8</sup>	8.739×10 <sup>8</sup>	/
	100	2.479	0.662	/
TiVCO <sub>2</sub> (V-Surface)	200	$1.094 \times 10^{6}$	$6.880 \times 10^{5}$	/
	300	8.326×10 <sup>7</sup>	$6.617 \times 10^{7}$	/
	100	4.037×10	9.386×10	56.989
V <sub>2</sub> CO <sub>2</sub>	200	$1.680 \times 10^{7}$	$2.143 \times 10^{7}$	21.605
	300	1.255×10 <sup>9</sup>	$1.403 \times 10^{9}$	10.549

S-group	Т	<i>k</i> in s <sup>-1</sup>	K <sub>wig/tunn</sub> in s <sup>-1</sup>	Percentage of QMT
	100	$1.823 \times 10^{3}$	$2.659 \times 10^{3}$	31.440
Ti <sub>2</sub> CS <sub>2</sub>	200	$1.185 \times 10^{8}$	$1.293 \times 10^{8}$	8.353
	300	4.766×10 <sup>9</sup>	$4.946 \times 10^{9}$	3.639
TWCS (T)	100	$1.412 \times 10^{3}$	$1.893 \times 10^{3}$	25.409
Surface)	200	9.730×10 <sup>7</sup>	$1.039 \times 10^{8}$	6.352
Surface)	300	3.989×10 <sup>9</sup>	$4.101 \times 10^{9}$	2.731
	100	$2.564 \times 10^{3}$	$3.041 \times 10^{3}$	15.686
TiVCS <sub>2</sub> (V-Surface)	200	$1.247 \times 10^{8}$	$1.278 \times 10^{8}$	2.426
	300	4.553×10 <sup>9</sup>	$4.587 \times 10^{9}$	0.741
	100	$1.323 \times 10^{4}$	$1.821 \times 10^{4}$	27.348
V <sub>2</sub> CS <sub>2</sub>	200	$2.857 \times 10^{8}$	$3.099 \times 10^{8}$	7.809
	300	7.955×10 <sup>9</sup>	8.250×10 <sup>9</sup>	3.576