

Supporting Information

Superlattice films of semiconducting oxide and rare-earth hydroxide nanosheets for tunable and efficient photoluminescent energy transfer

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Table S1. Elemental analysis results (mass%) of the as-prepared samples

sample	Estimated composition	Elemental analysis	
		Calculated	Measured
LGdH:Eu	(Gd _{0.96} Eu _{0.04}) ₂ (OH) _{5.51} (C ₁₂ H ₂₅ SO ₄) _{0.49} ·1.9H ₂ O	Gd 36.1 Eu 1.8	Gd 36.1 Eu 1.7
LGdH:Tb	(Gd _{0.98} Tb _{0.02}) ₂ (OH) _{5.46} (C ₁₂ H ₂₅ SO ₄) _{0.54} ·2.0H ₂ O	Gd 43.3 Tb 1.0	Gd 43.3 Tb 1.0

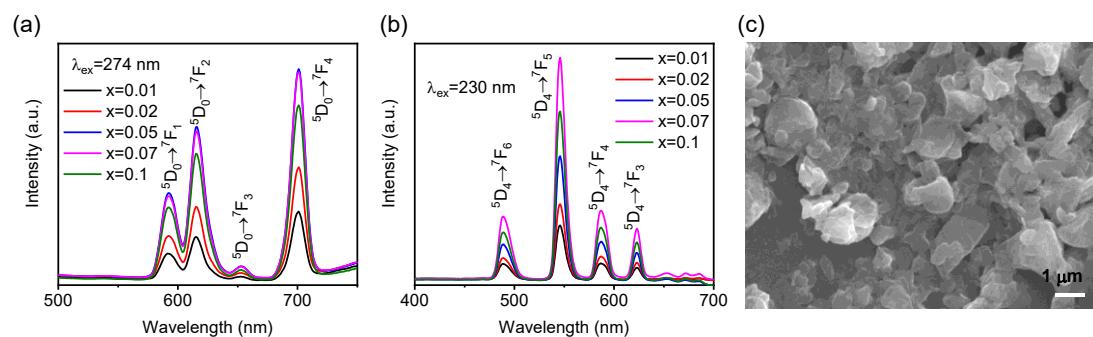


Figure S1. Photoluminescence emission spectra for (a) LGd_{1-x}H:Eu_x and (b) LGd_{1-x}H:Tb_x with different doped amount of Eu³⁺ and Tb³⁺; (c) SEM images of LGd_{0.95}H:Tb_{0.05}.

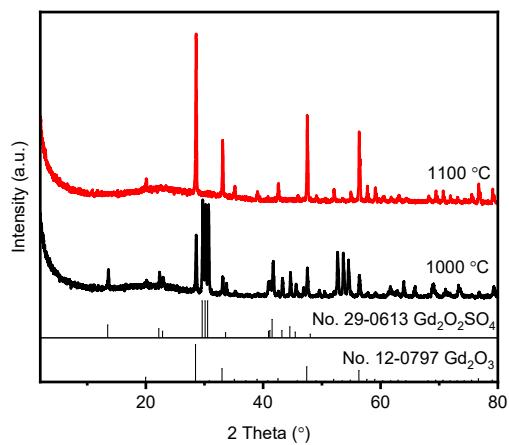


Figure S2. XRD patterns of LGdH:Eu powder sample after calcined at 1000 °C and 1100 °C for 7 h.

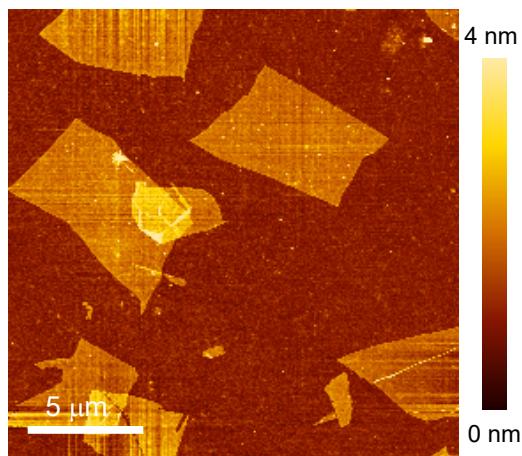


Figure S3. AFM image of Ti_{0.87}O₂^{0.52-} nanosheets.

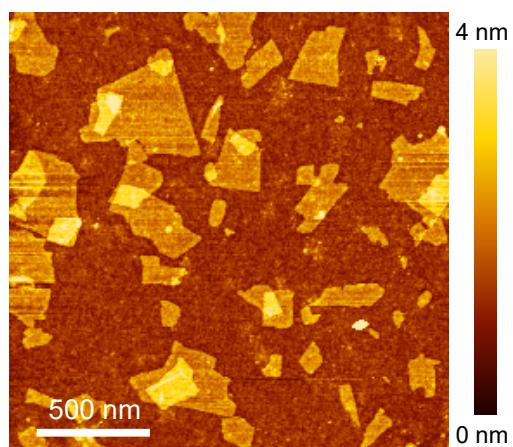


Figure S4. AFM image of TaO₃⁻ nanosheets.

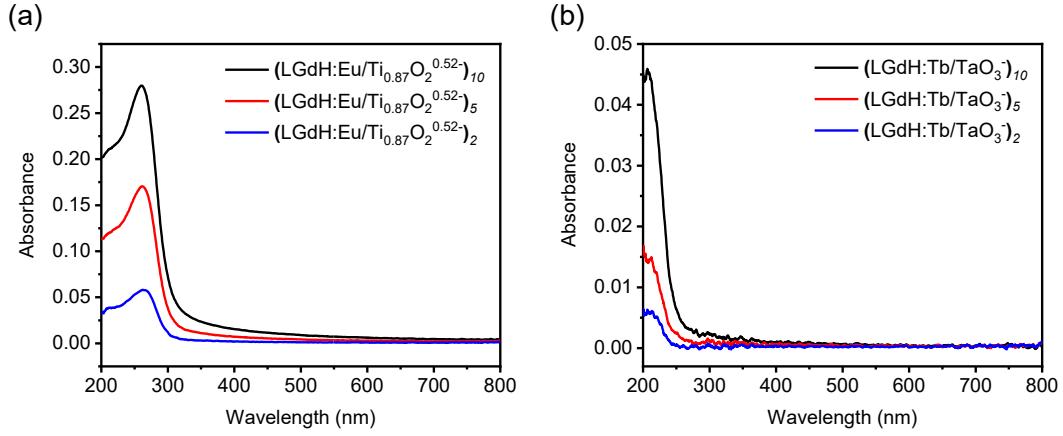


Figure S5. Typical UV-vis absorption spectra of (a) (LGdH:Eu/Ti_{0.87}O₂^{0.52-})_n and (b) (LGdH:Tb/TaO₃)_n films ($n = 2, 5, 10$).

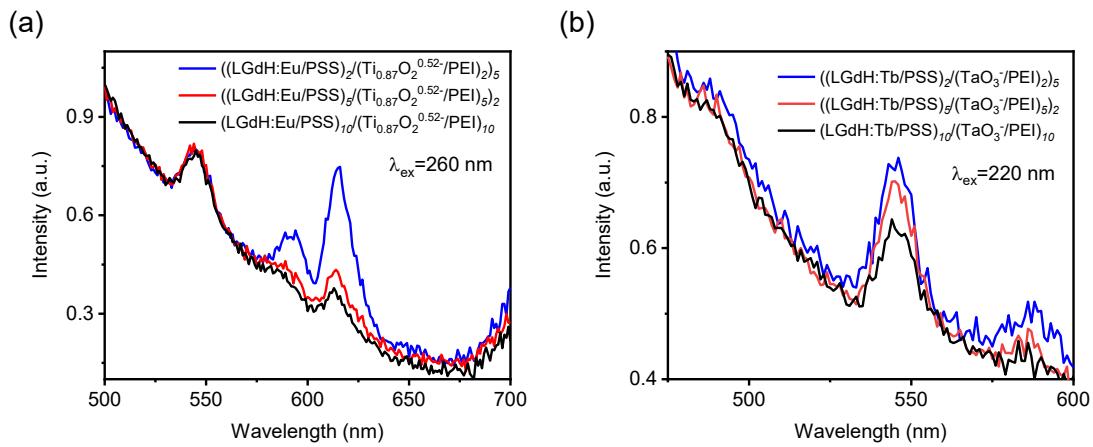


Figure S6. Normalized emission spectra of (a) ((LGdH:Eu/PSS)_m/(Ti_{0.87}O₂^{0.52-}/PEI)_m)_n and (b) ((LGdH:Tb/PSS)_m/(TaO₃/PEI)_m)_n series with a total of 20 layers ($m = 2, 5$ and 10 , while $n = 5, 2$ and 1 , respectively). PSS or PEI was used as charge-balancing counterions for assembling rare-earth hydroxide nanosheets and oxide nanosheets, respectively.

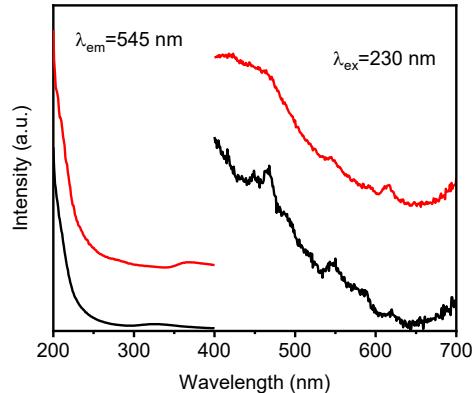


Figure S7. Photoluminescence excitation and emission spectra of (LGdH:Tb/PSS)₁₀ (black trace) and (LGdH:Tb/Ti_{0.87}O₂^{0.52-})₁₀ (red trace) films.

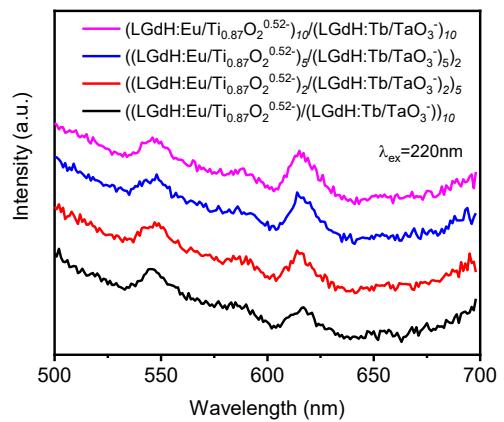


Figure S8. Photoluminescence emission spectra of $((\text{LGdH:Eu/Ti}_{0.87}\text{O}_2^{0.52-})_m/(\text{LGdH:Tb/TaO}_3^-)_n$ films ($m = 1, 2, 5$ and 10 , while $n = 10, 5, 2$ and 1 , respectively) under excitation of 220 nm .
