

Synthesis of β -lactams *via* diastereoselective, intramolecular Kinugasa reactions

Oskar Popik, Barbara Grzeszczyk, Olga Staszewska-Krajewska, Bartłomiej Furman and Marek Chmielewski*

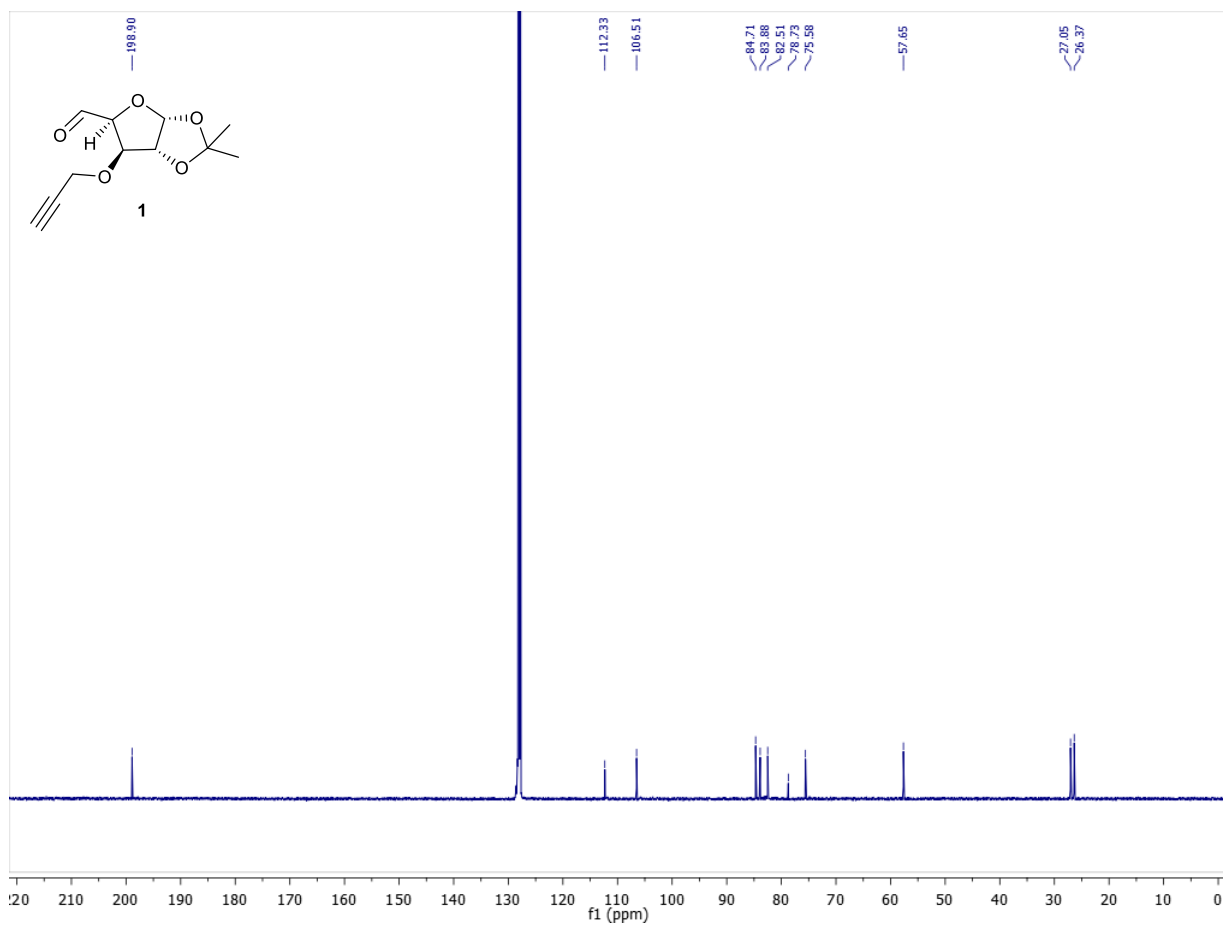
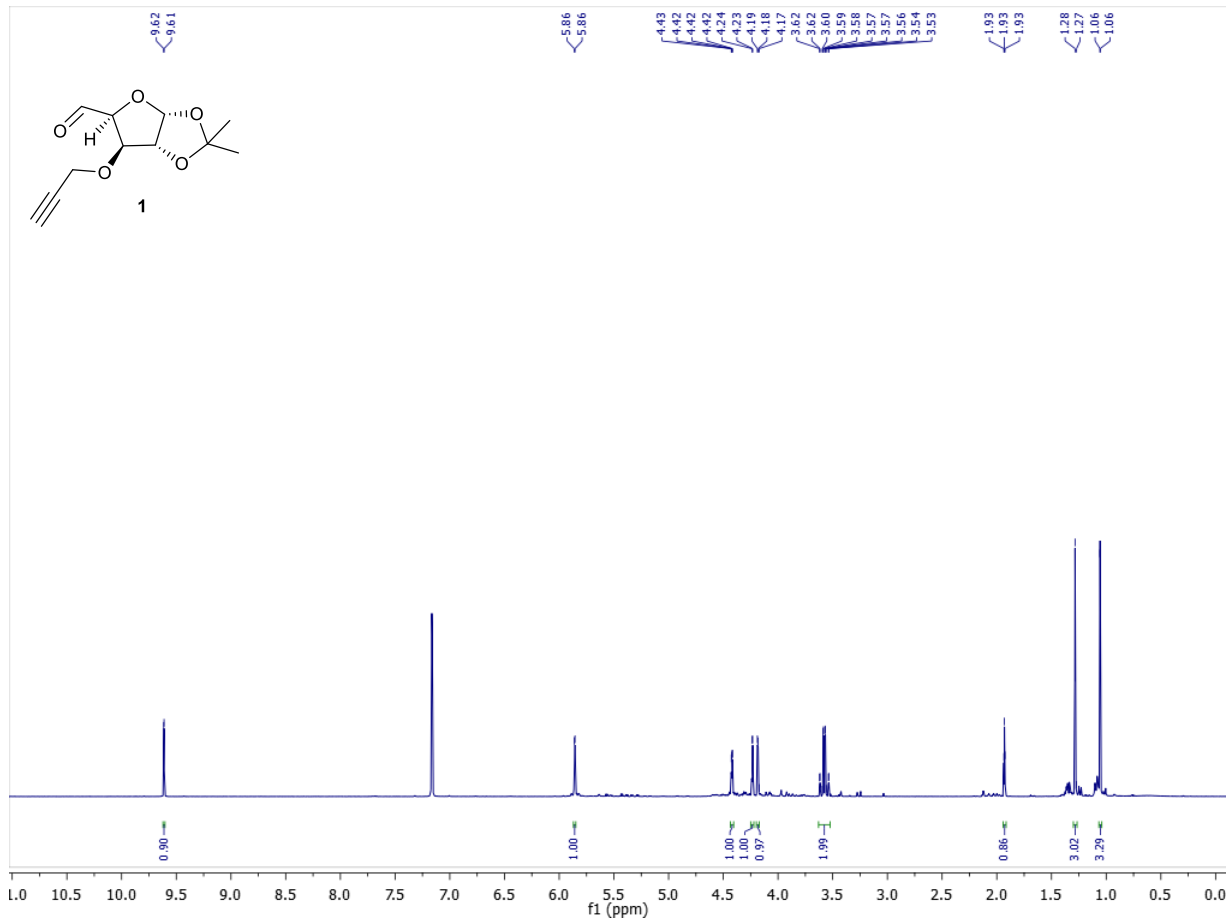
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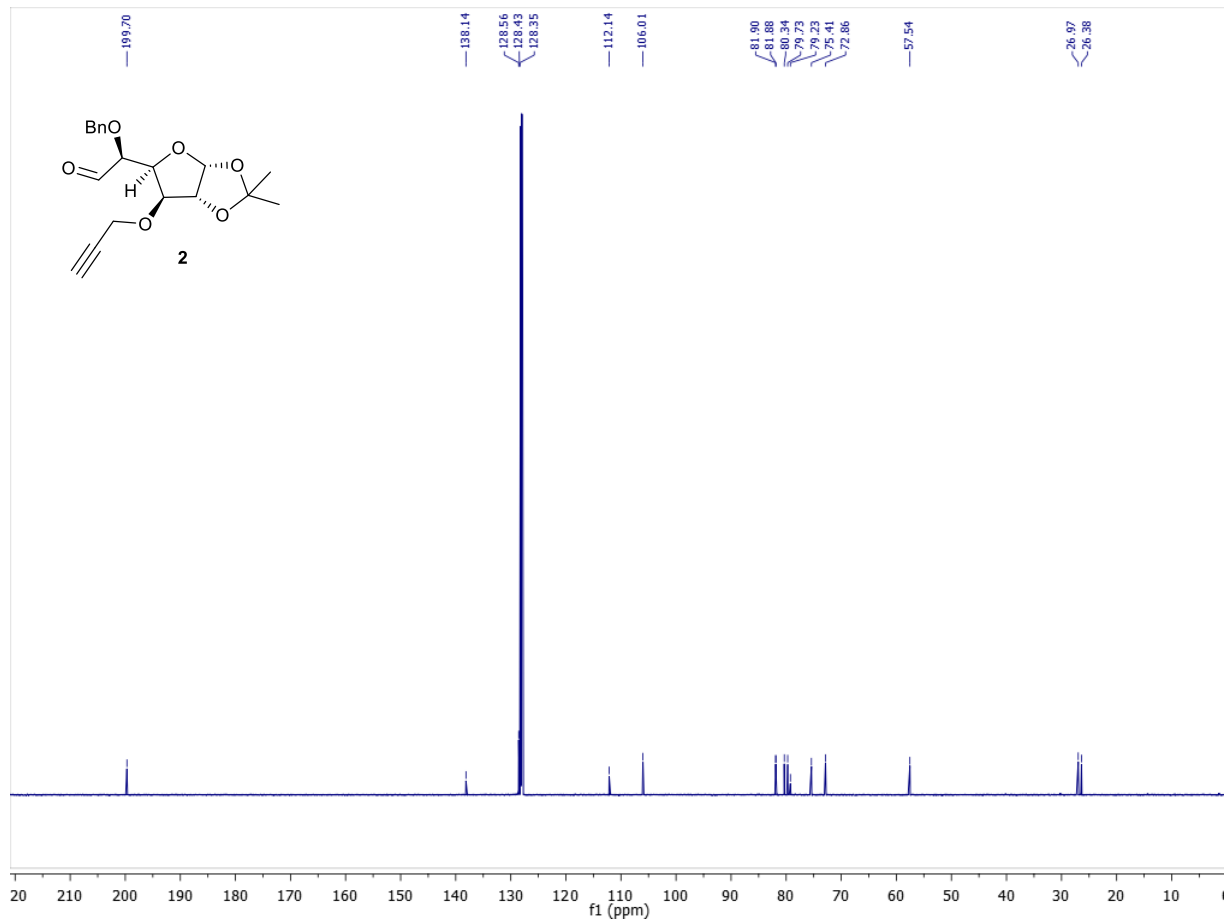
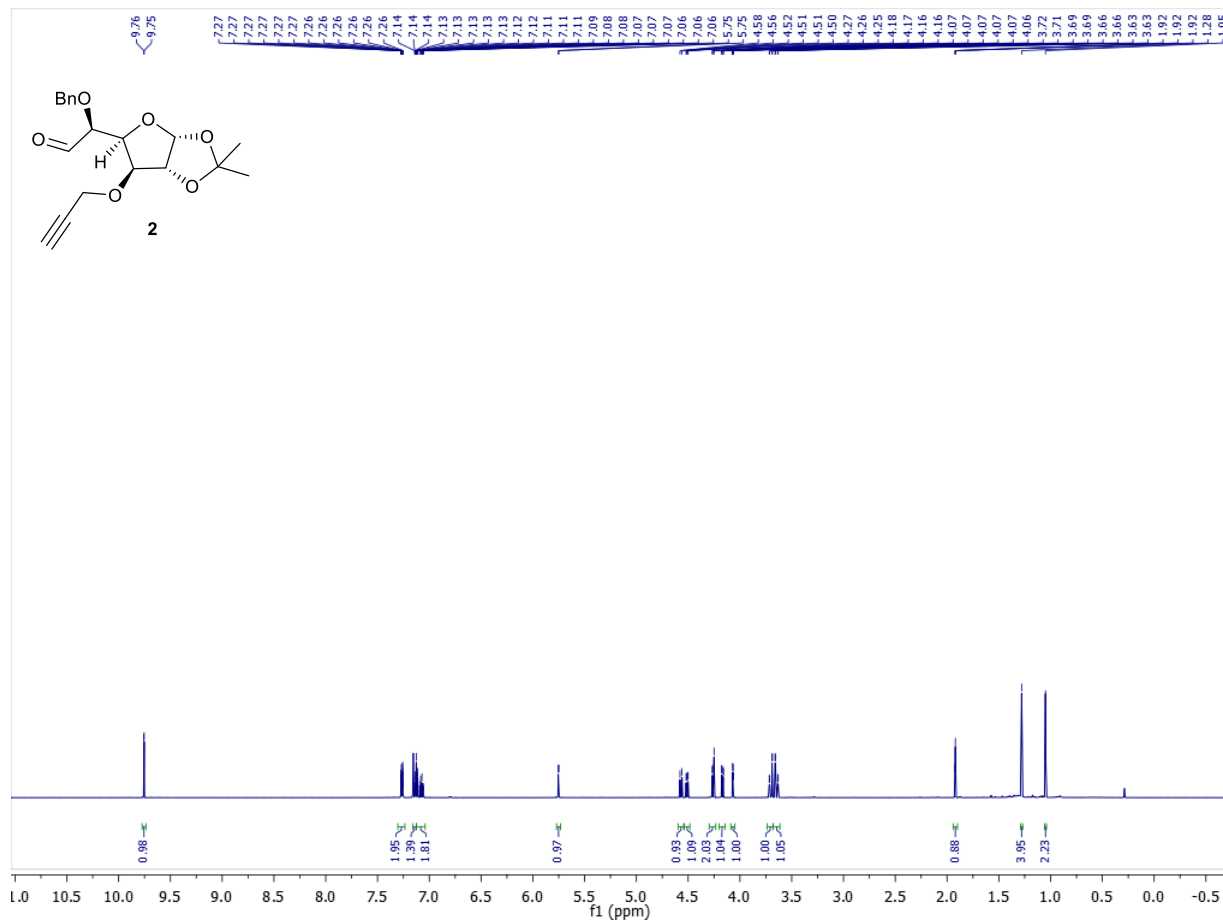
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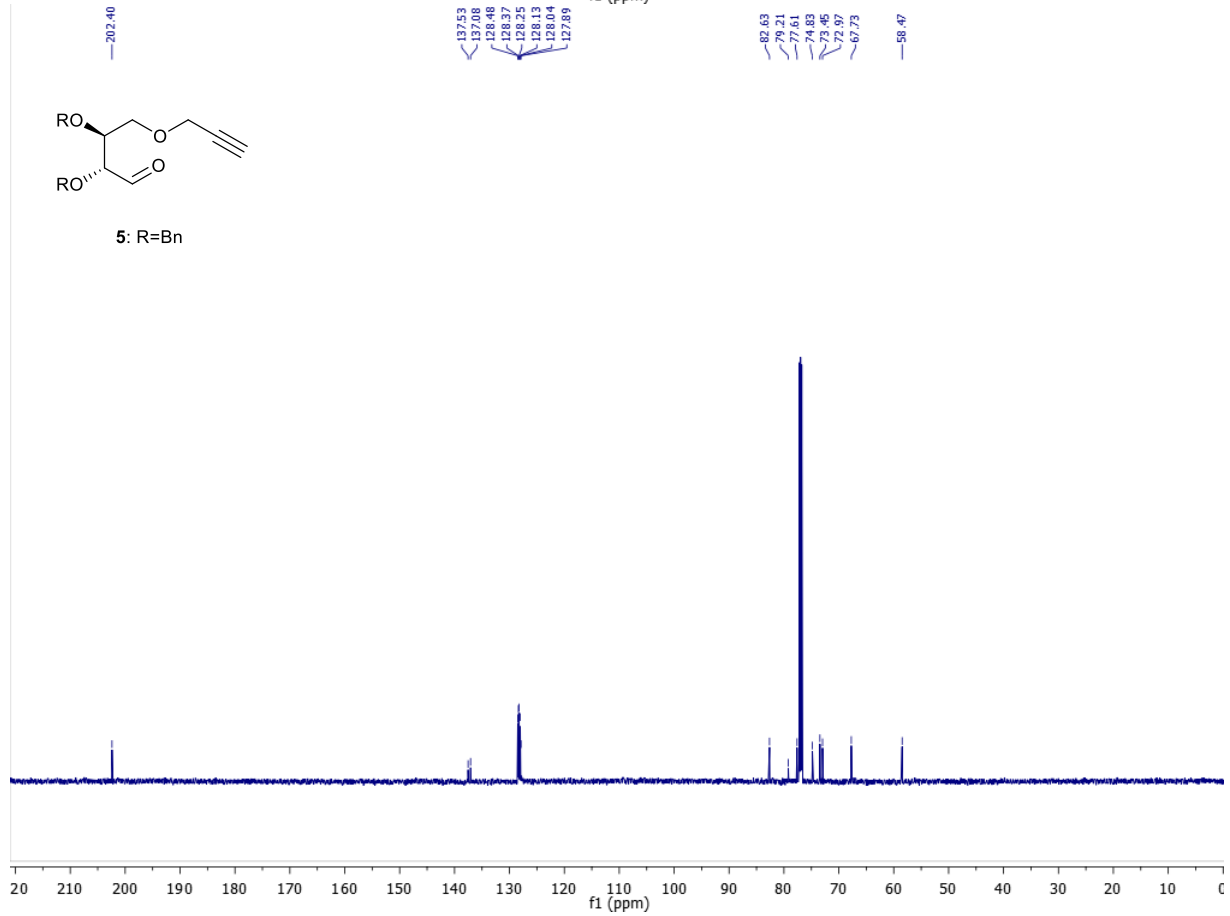
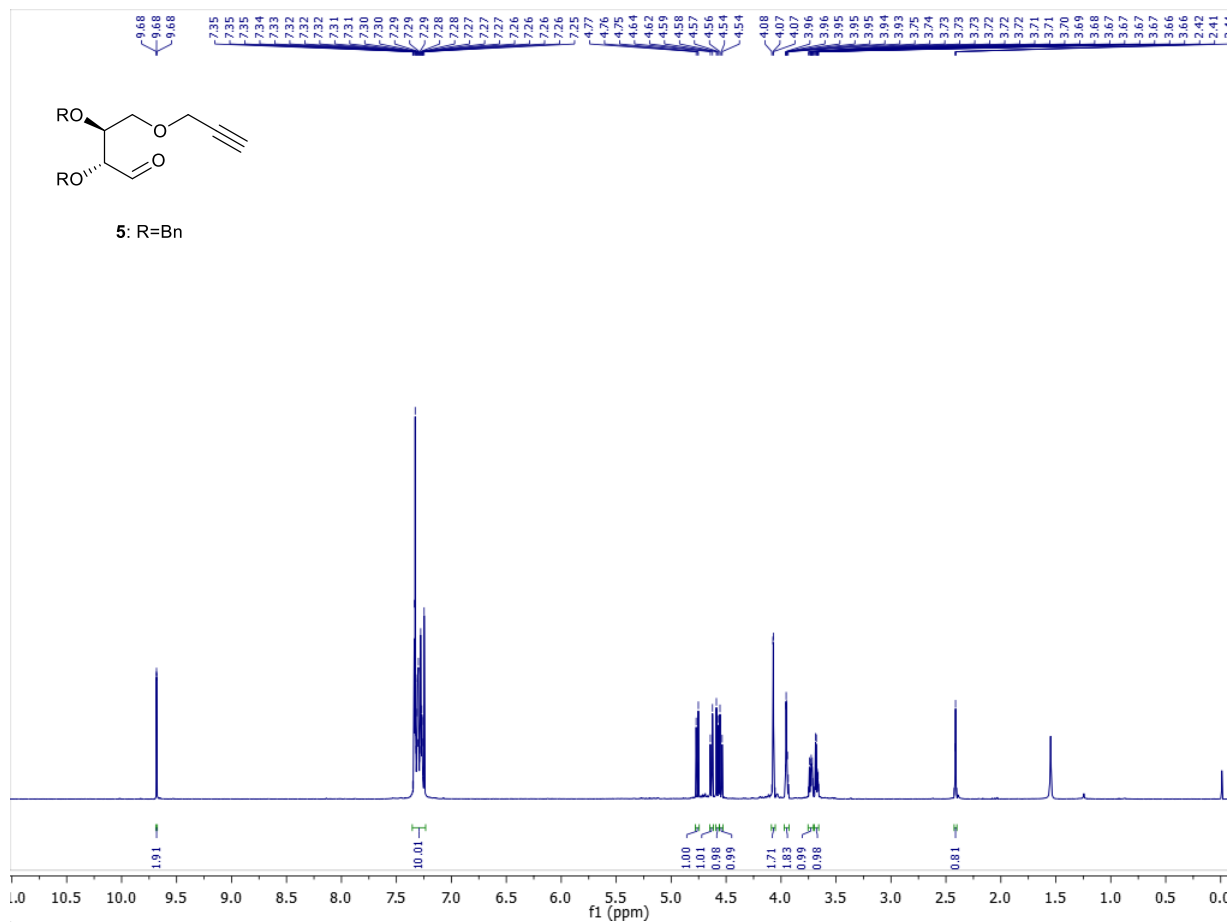
PAGE S1: Scans of ^1H and ^{13}C NMR Spectra and NOE measurements

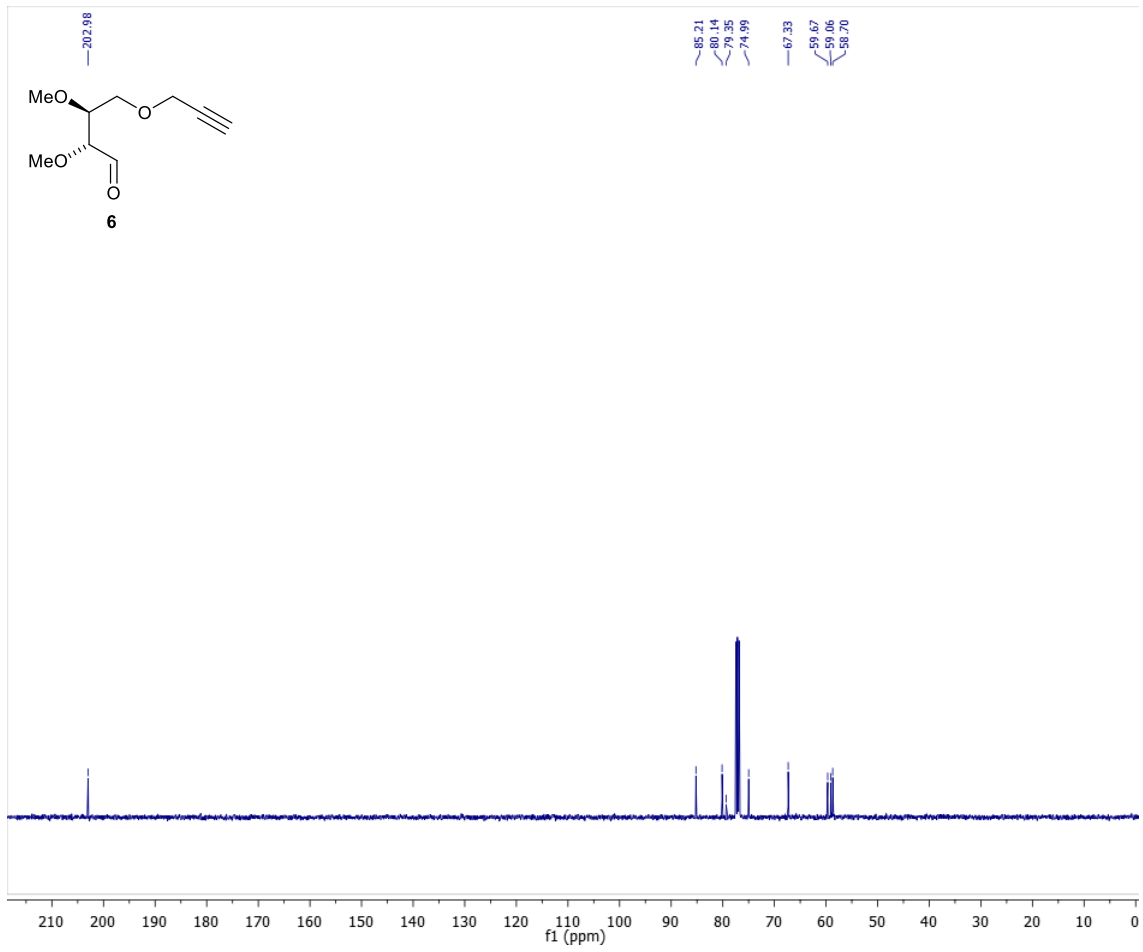
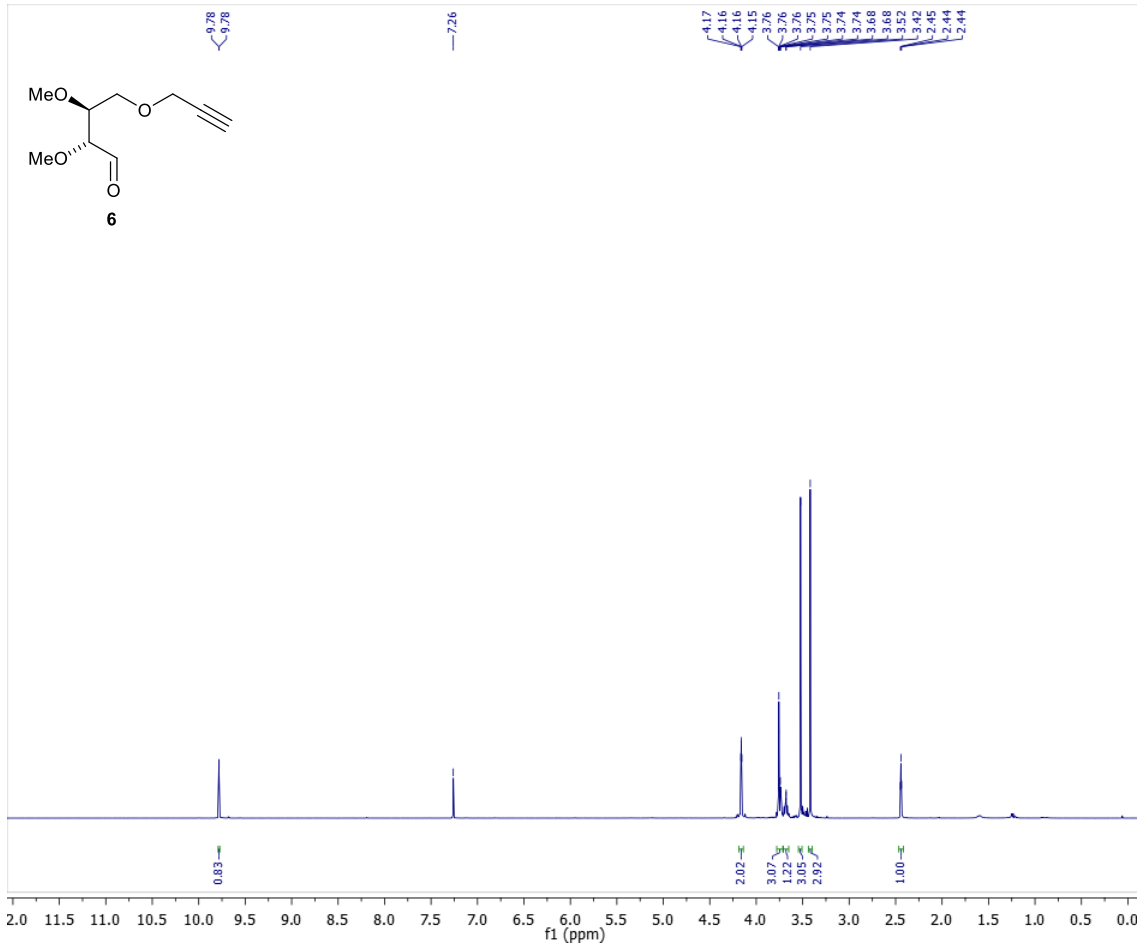
PAGE S33: Separate ECD analyses

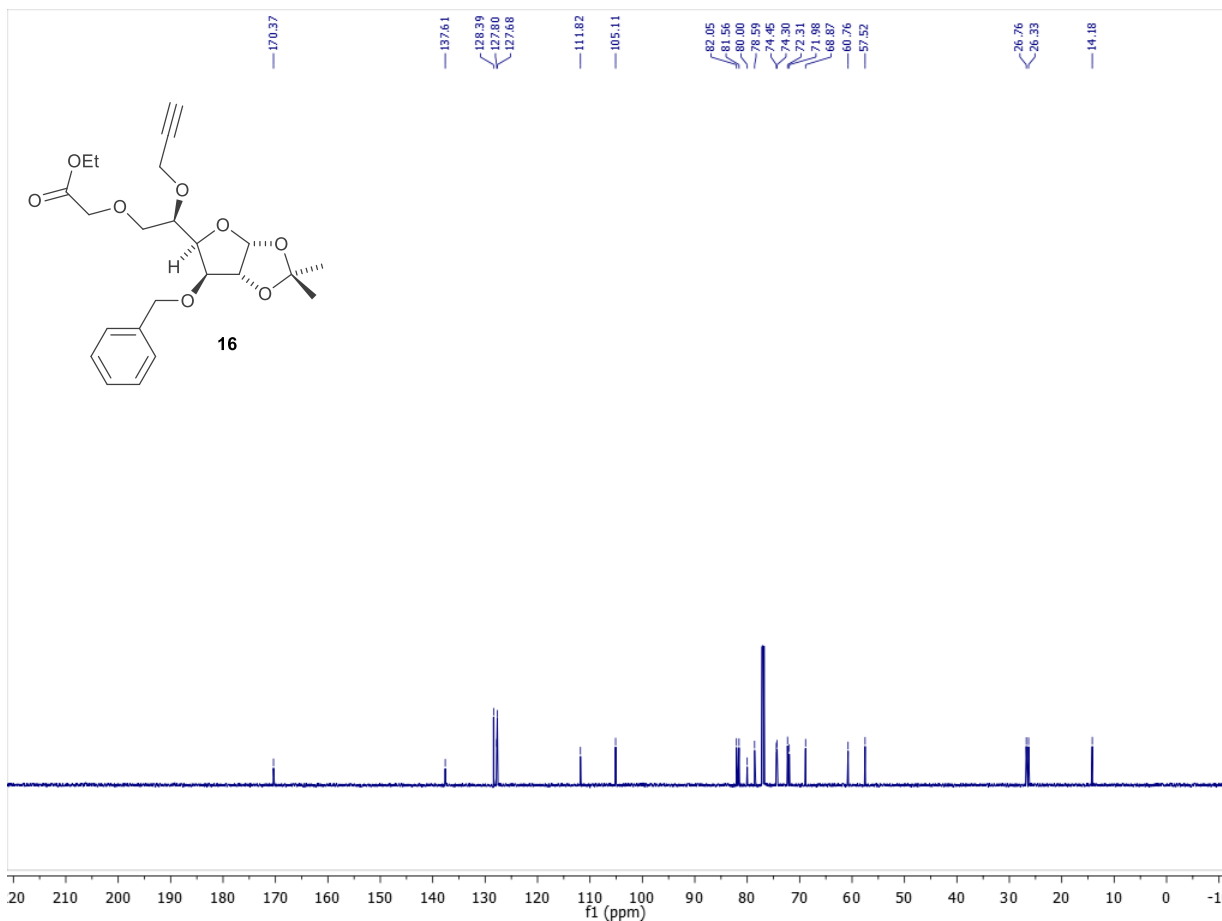
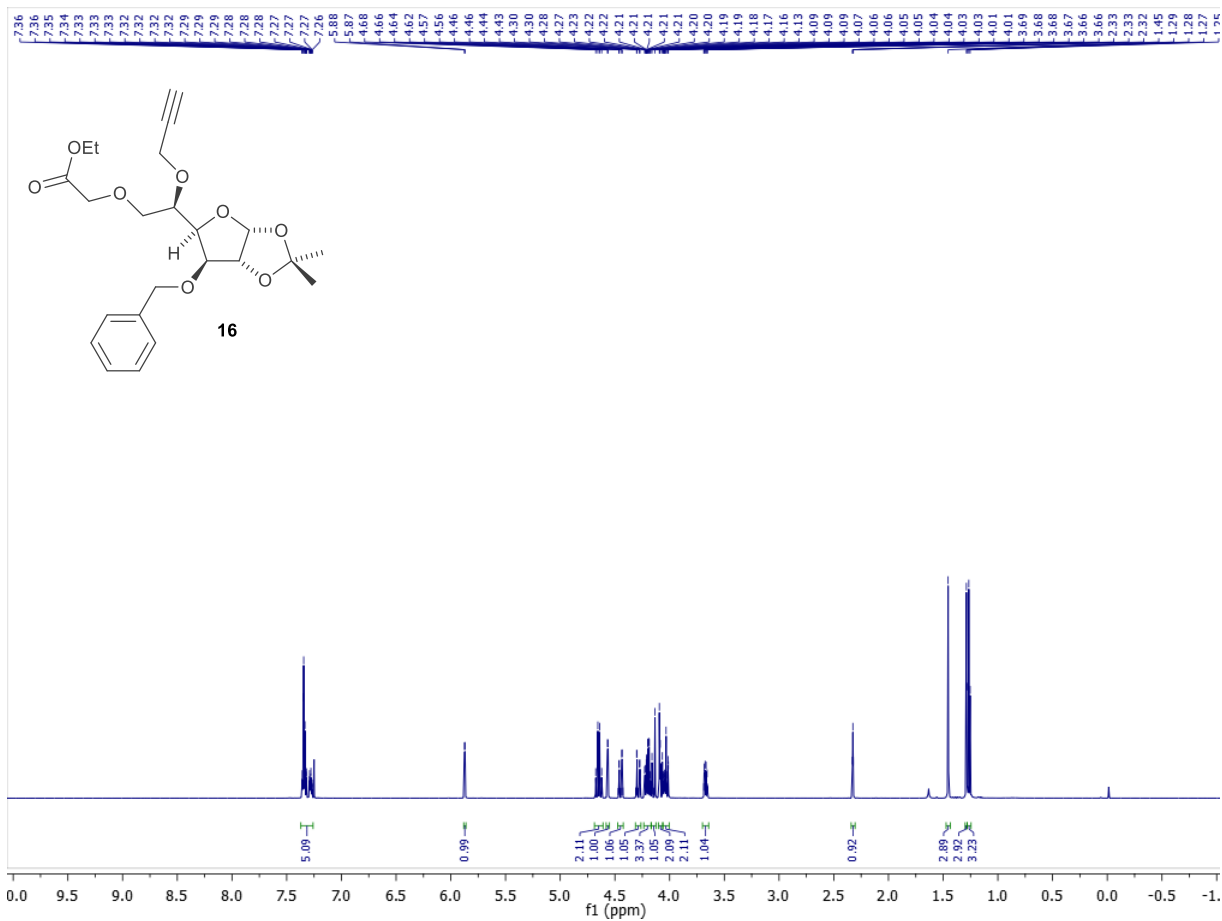
PAGE S34: X-ray crystallography of **30**

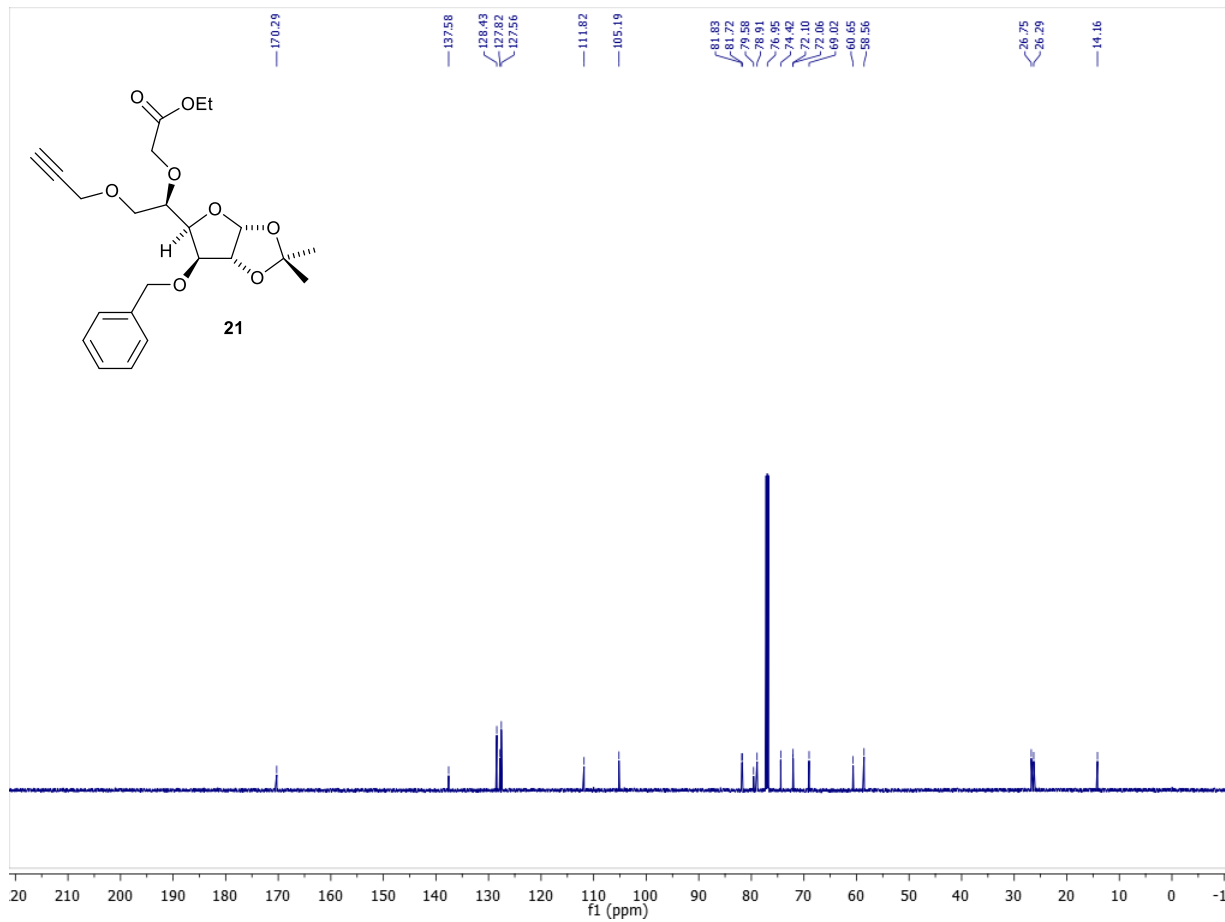
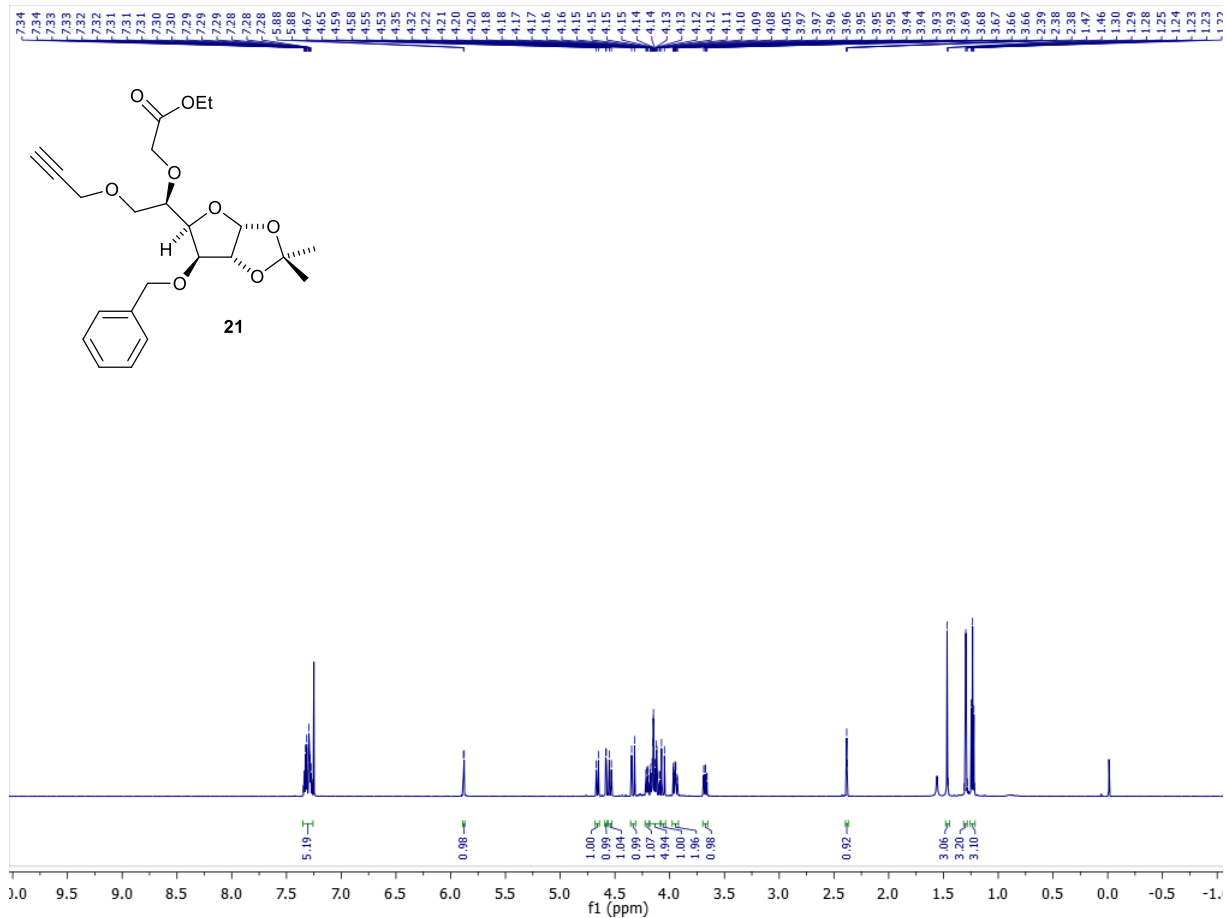


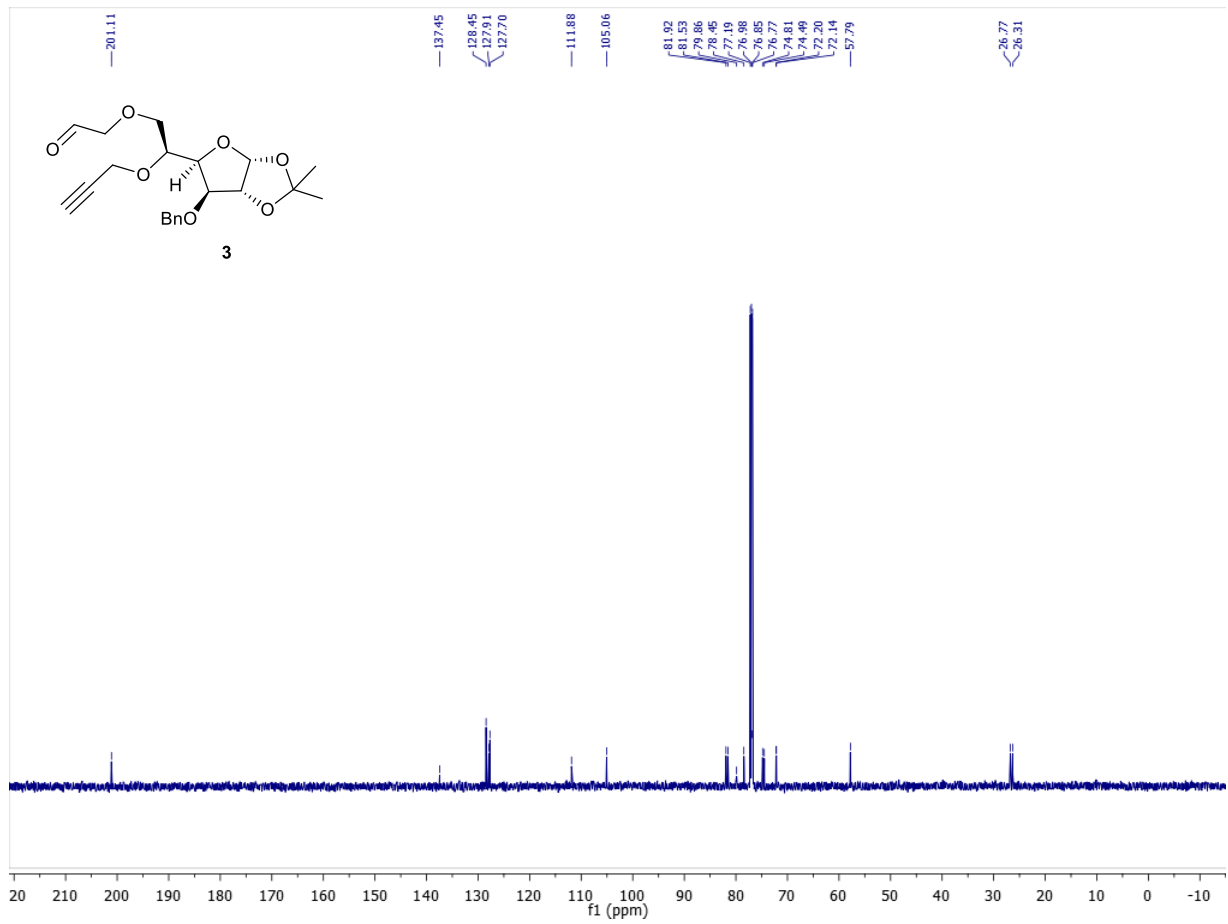
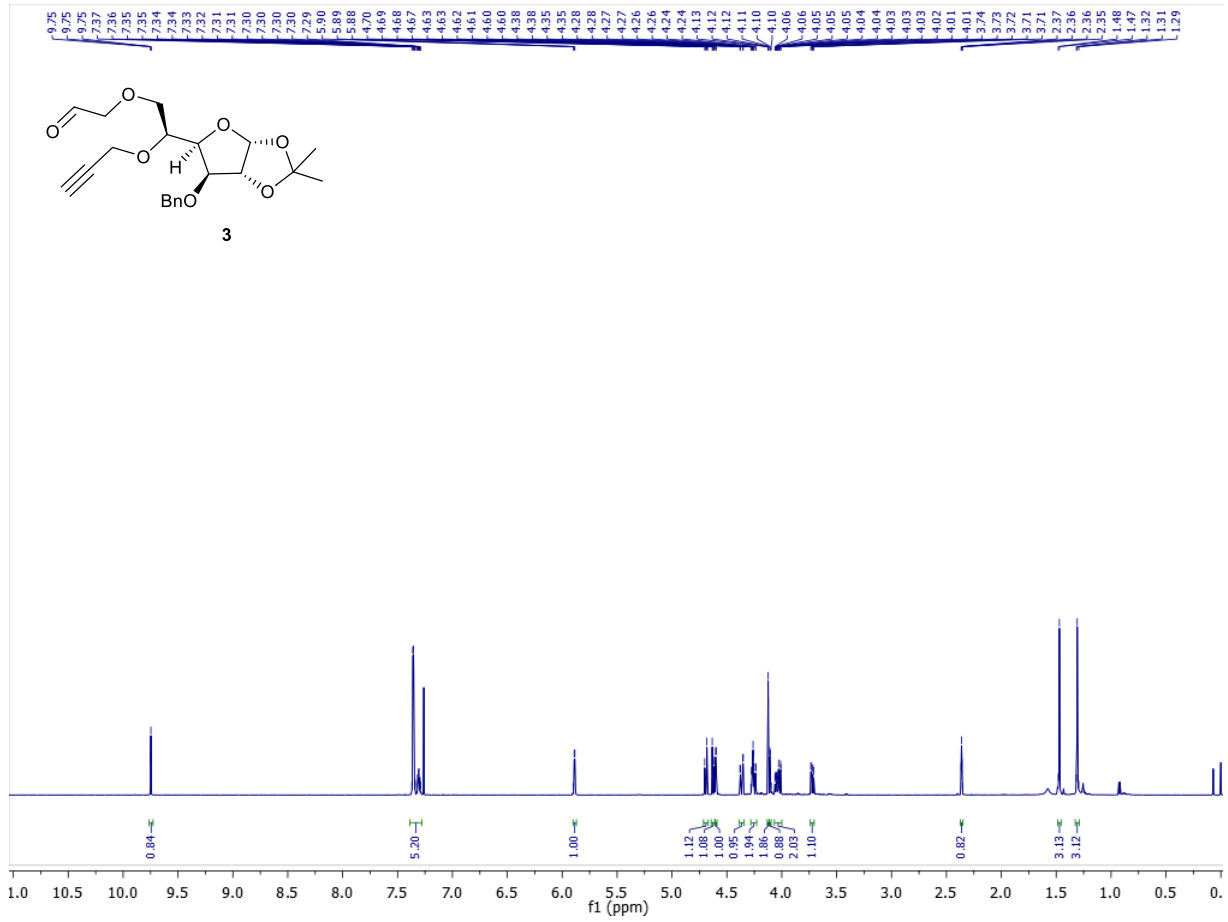


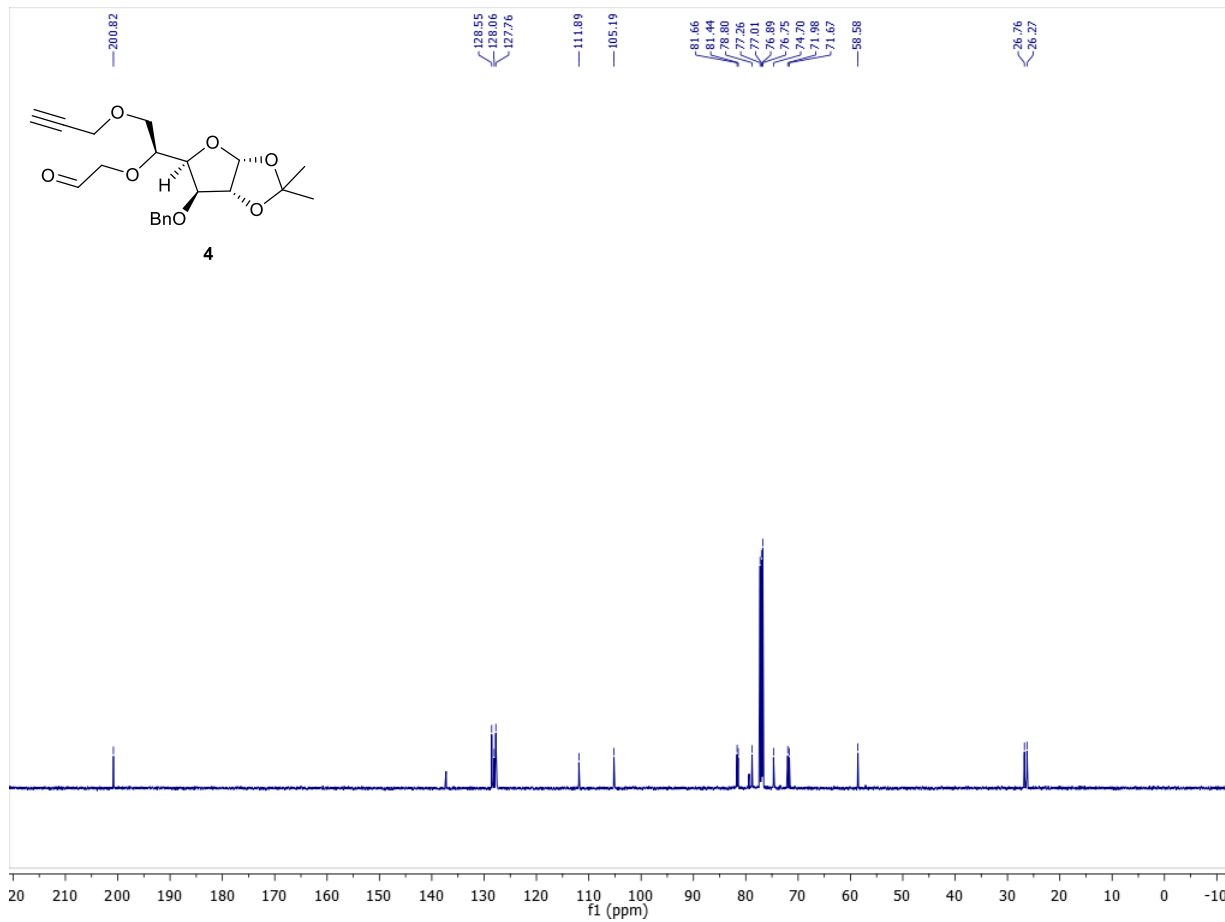
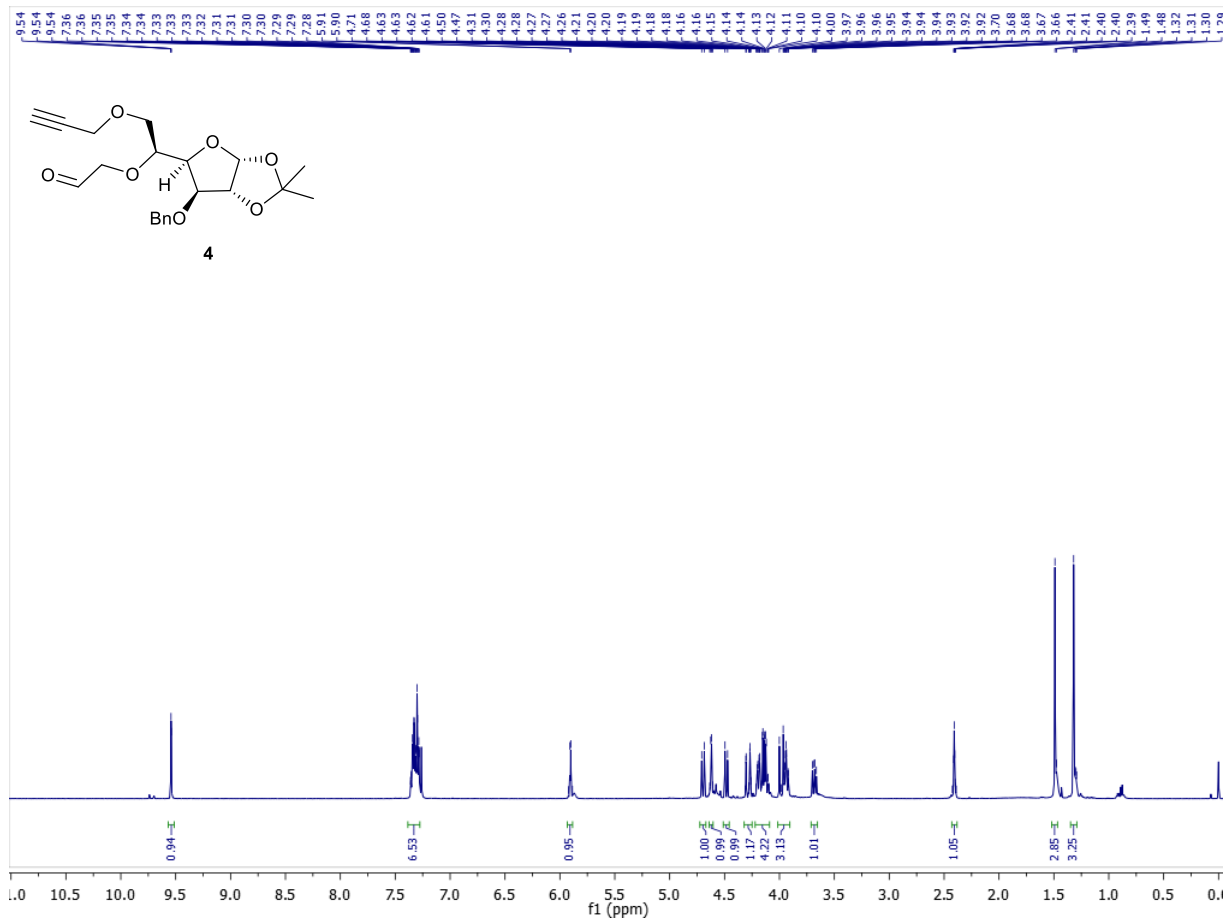


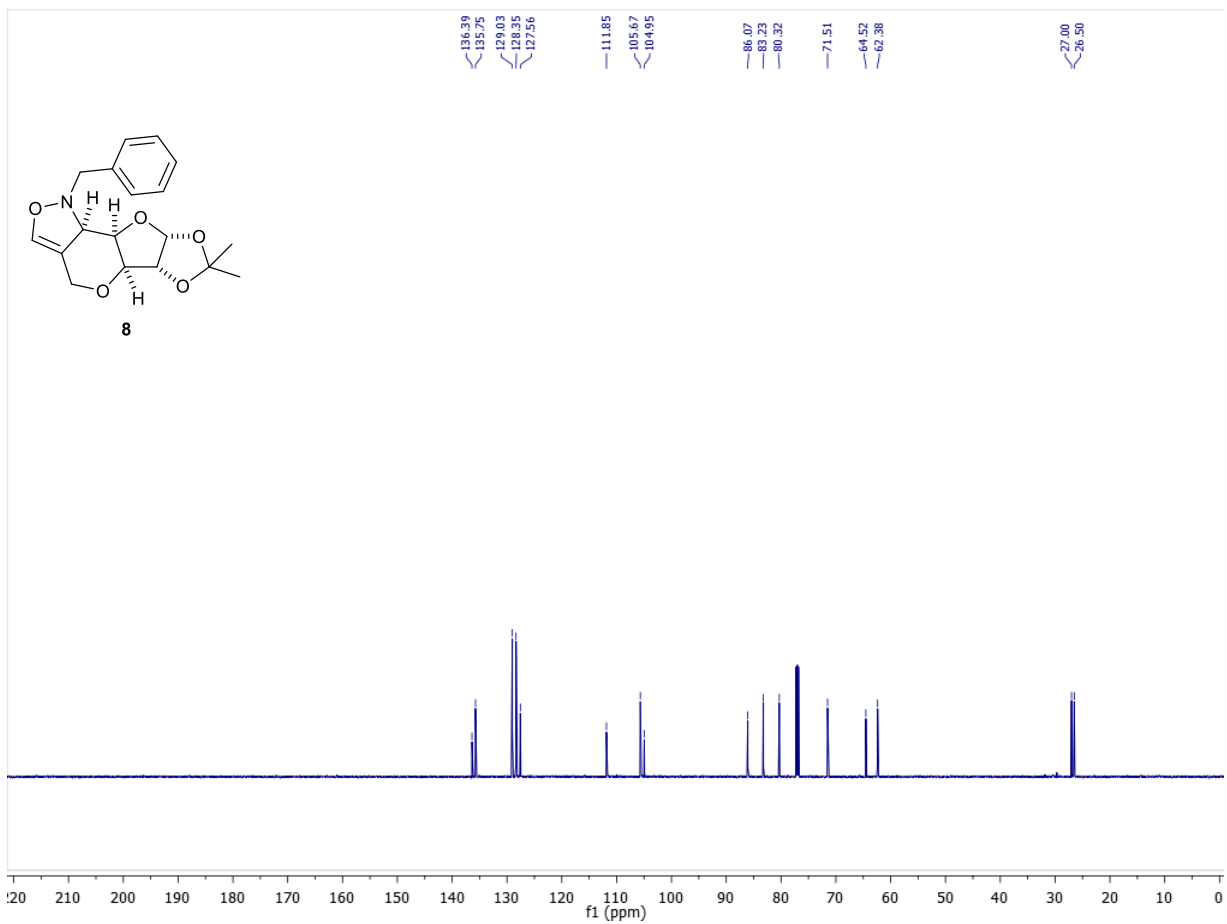
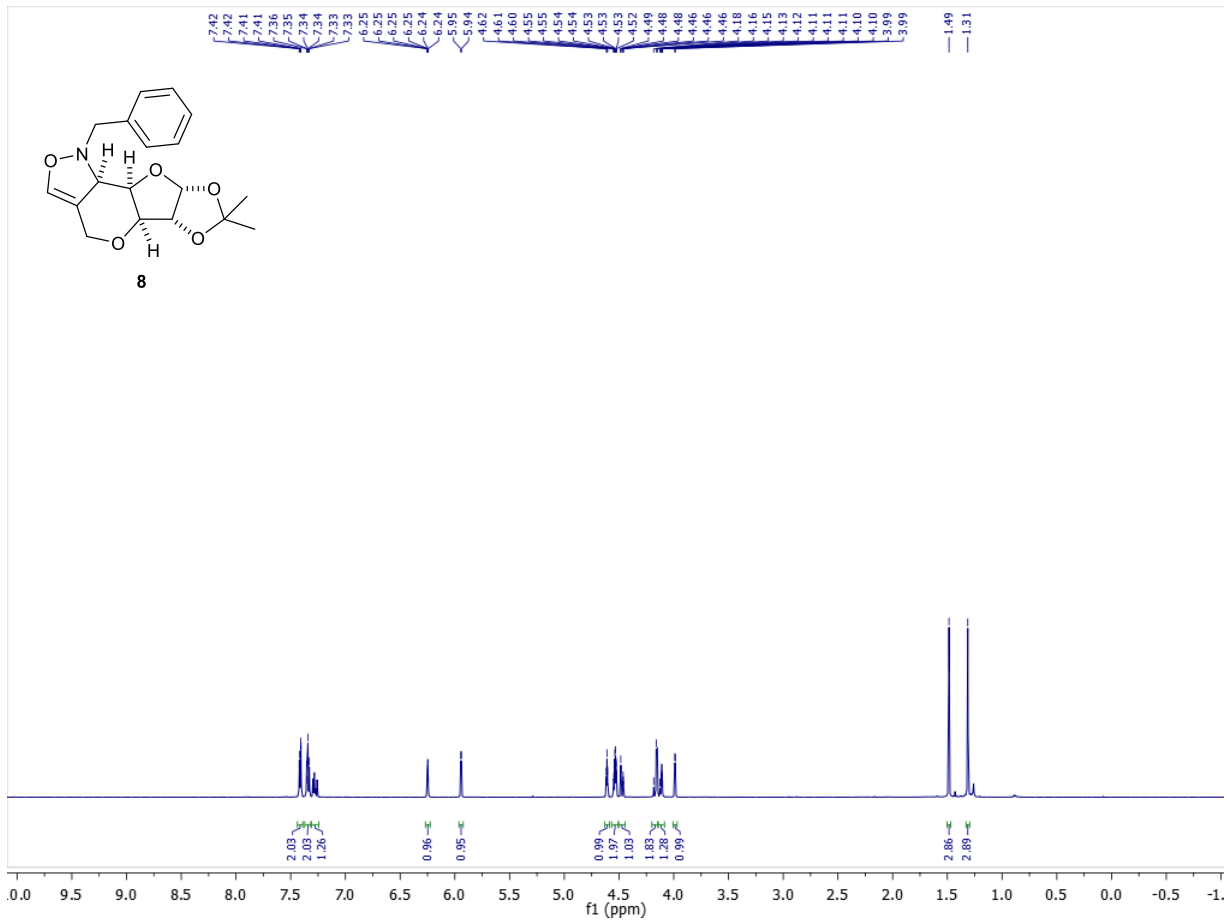


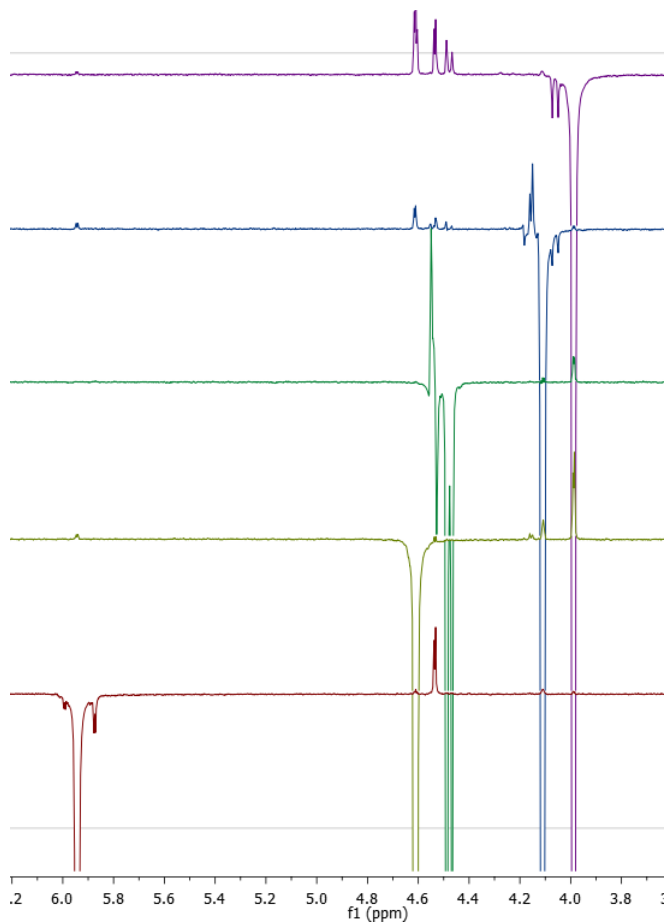




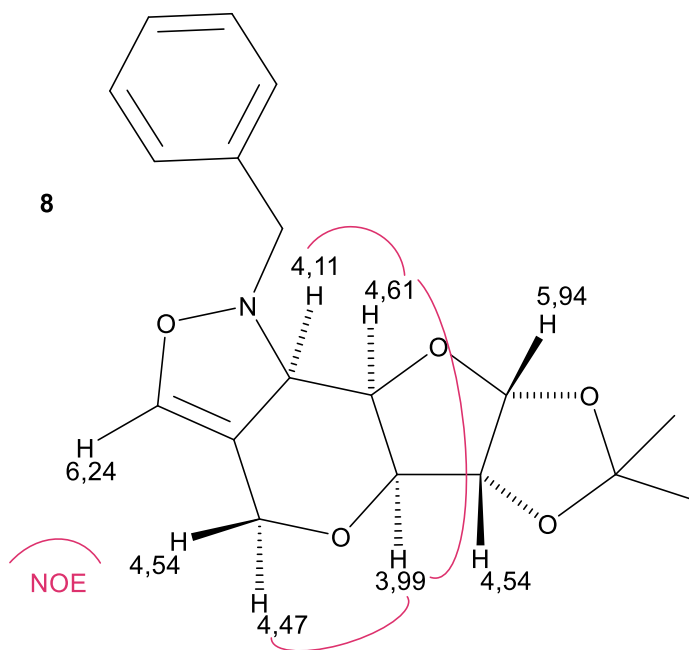


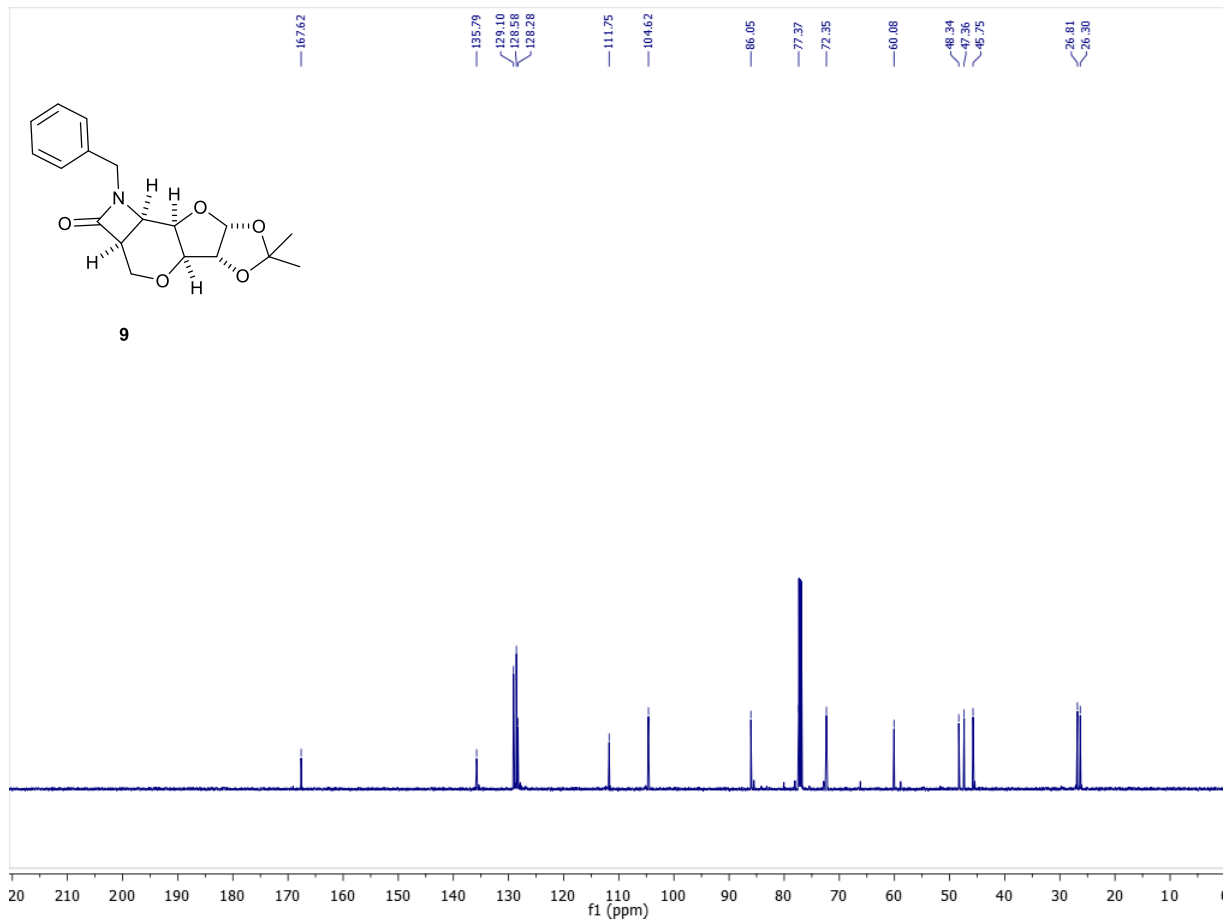
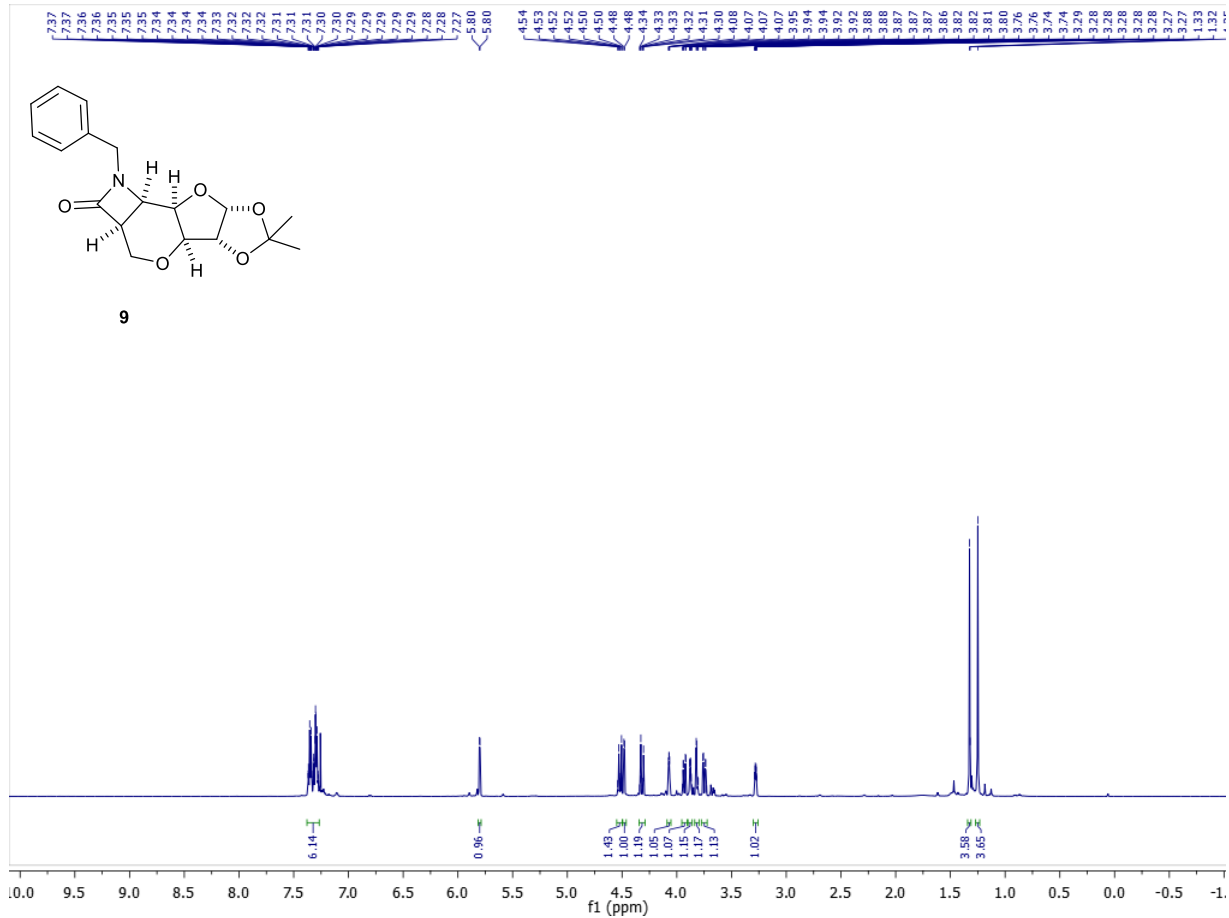


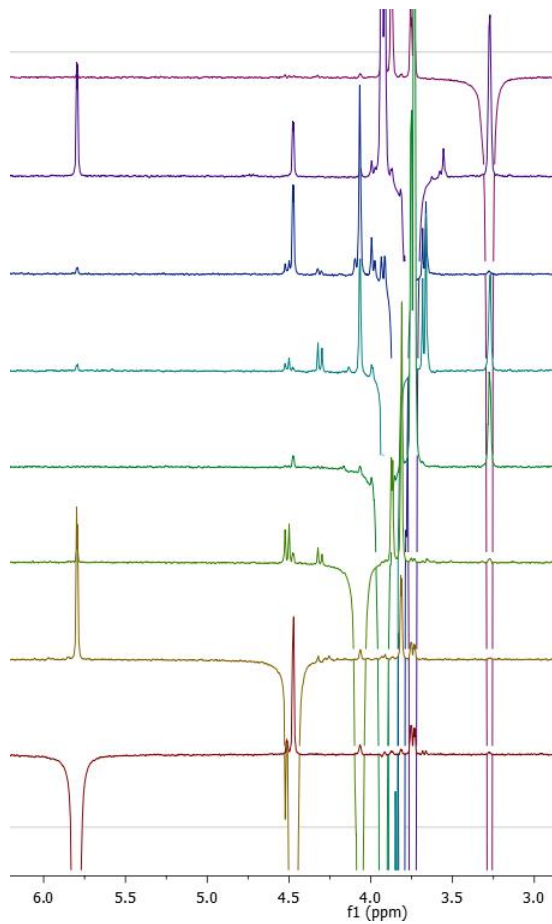




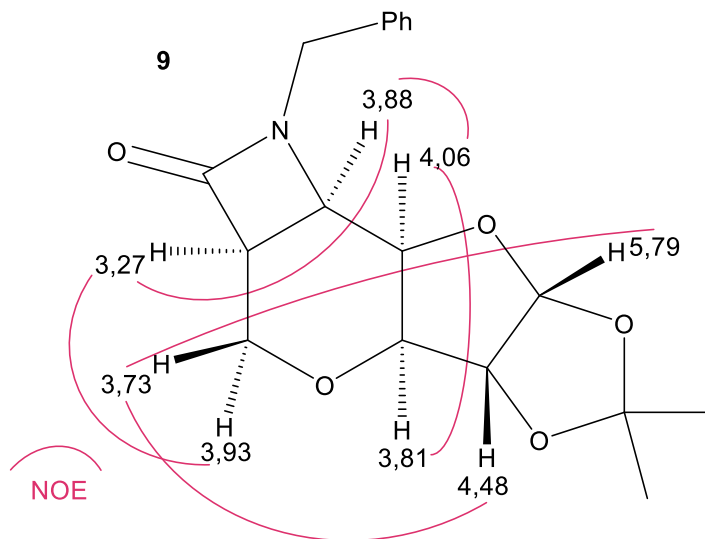
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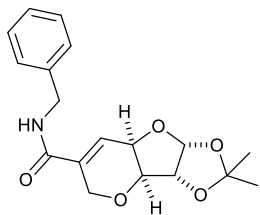




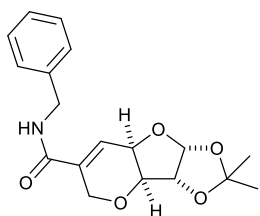
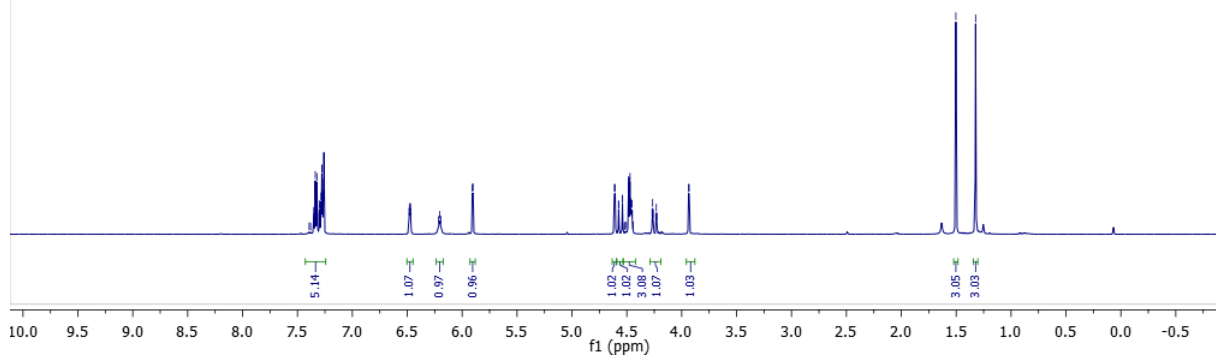
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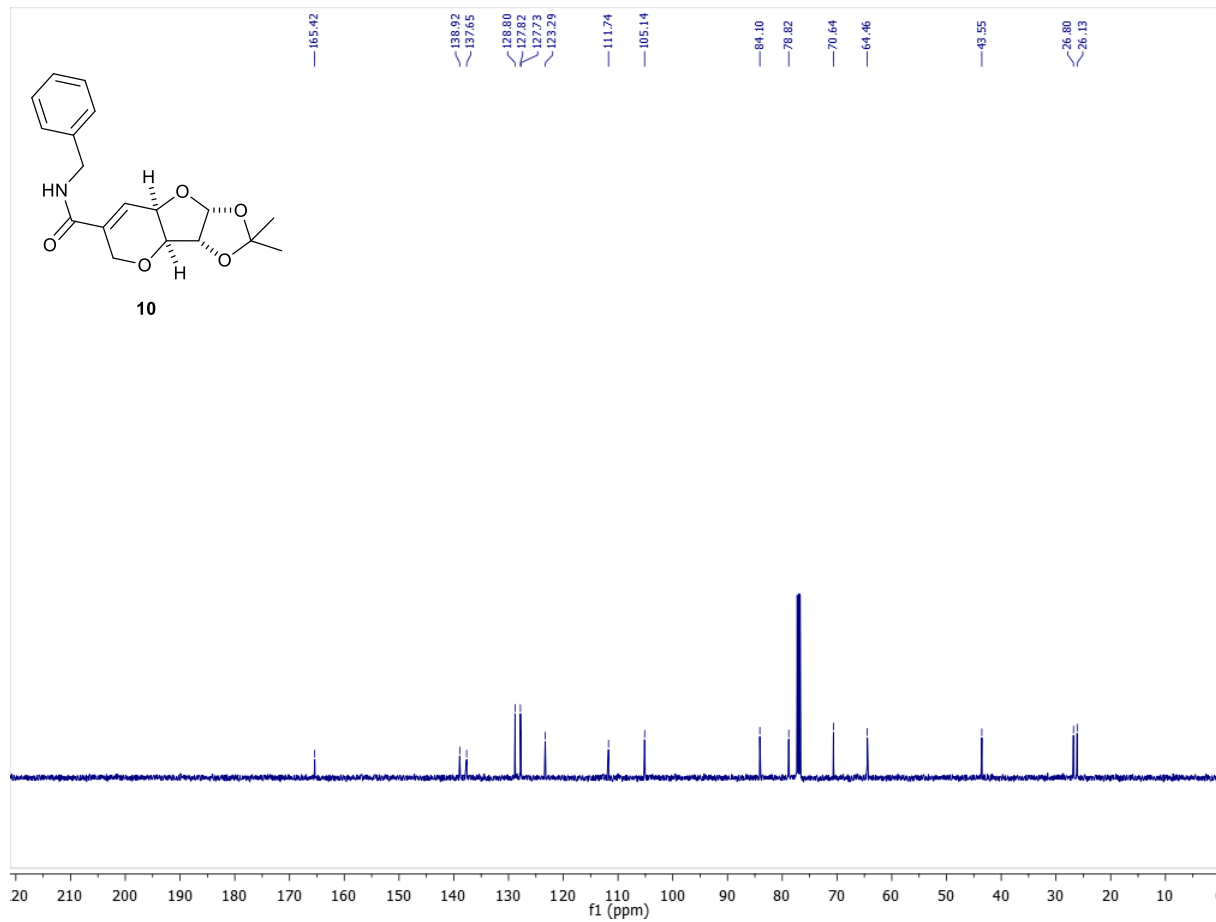
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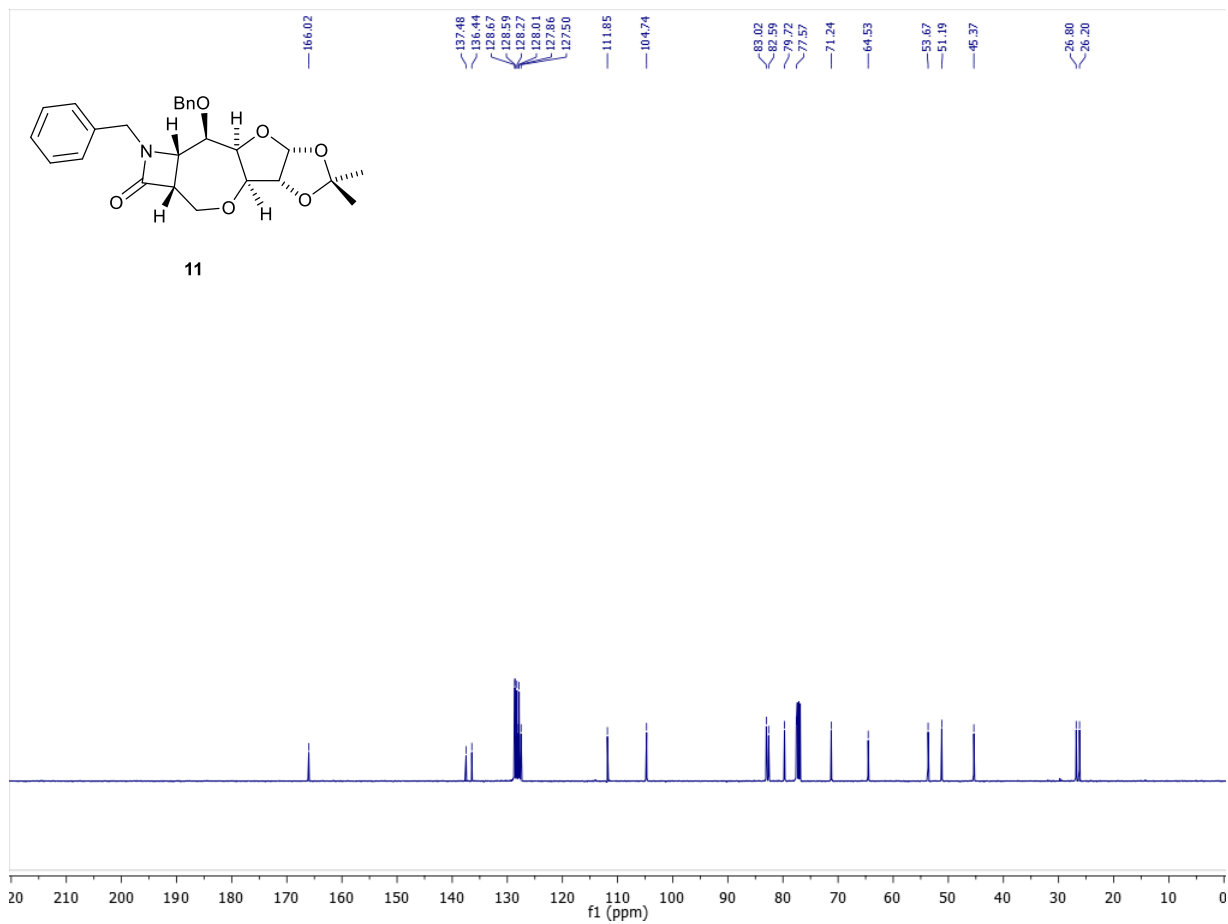
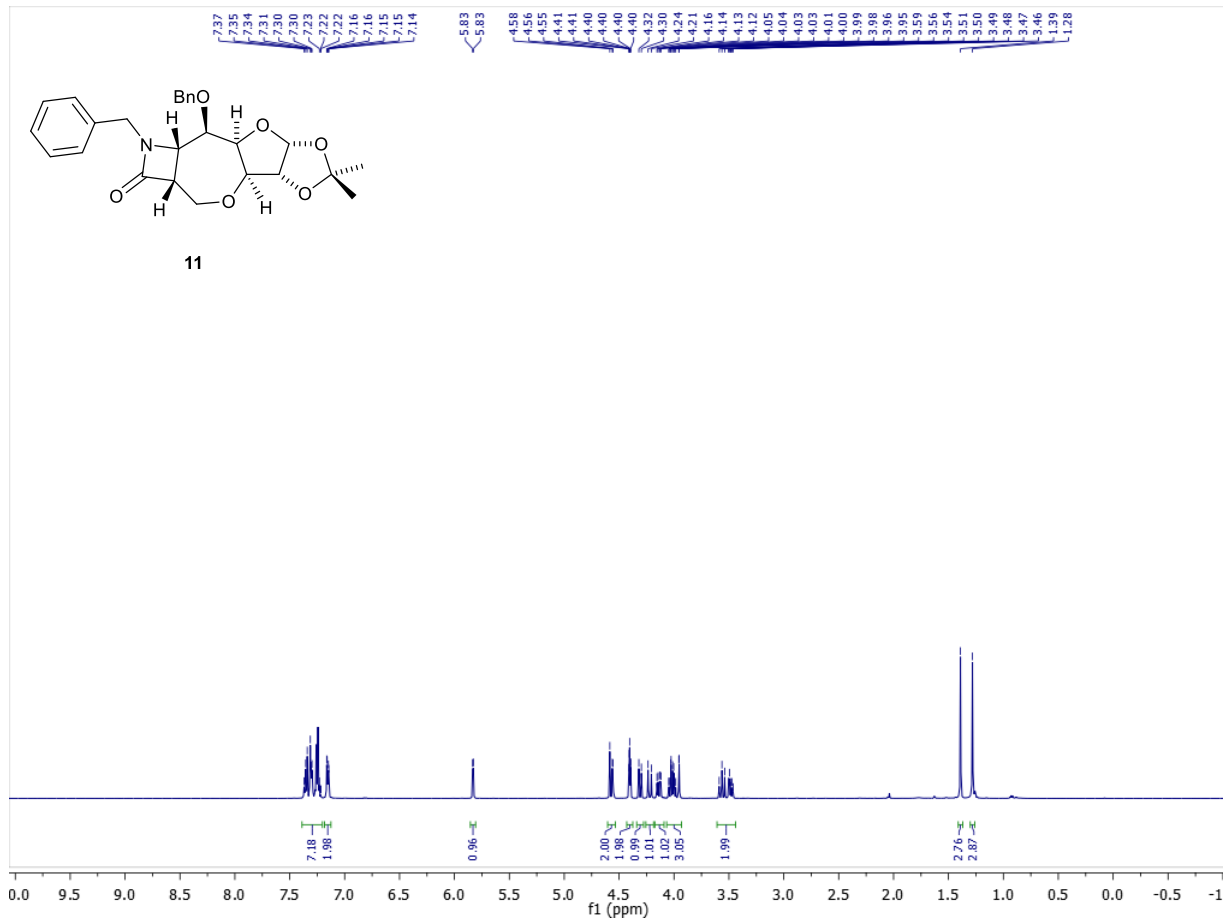


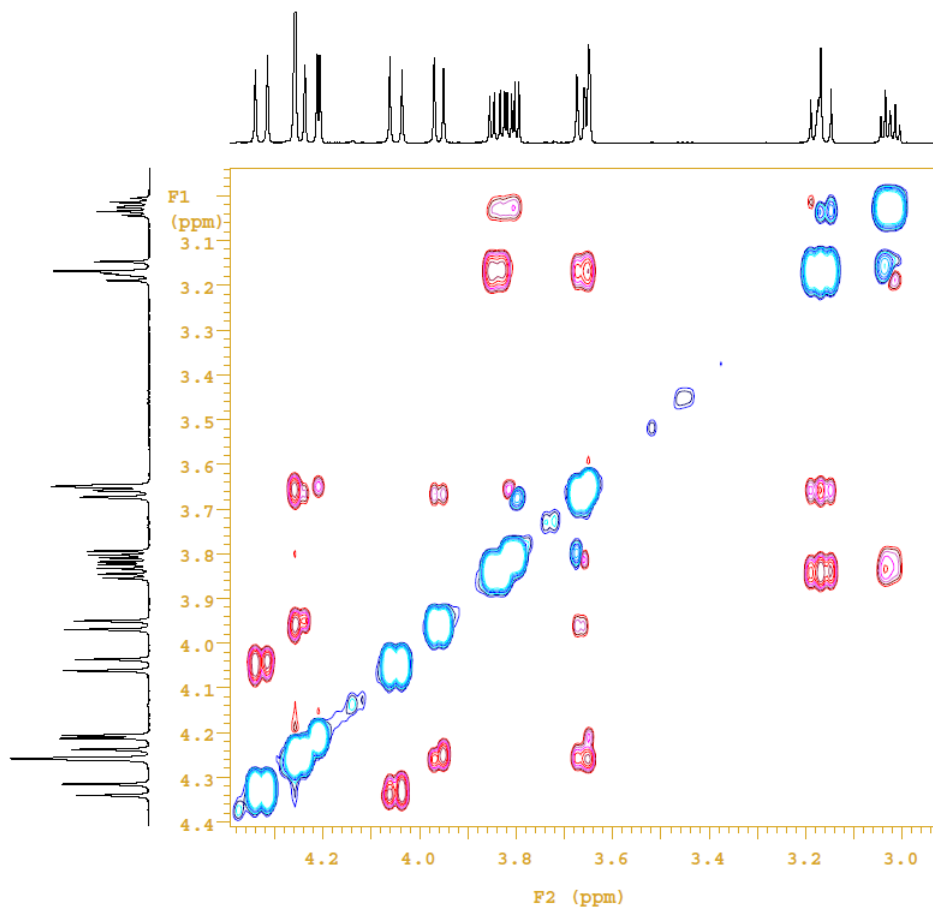
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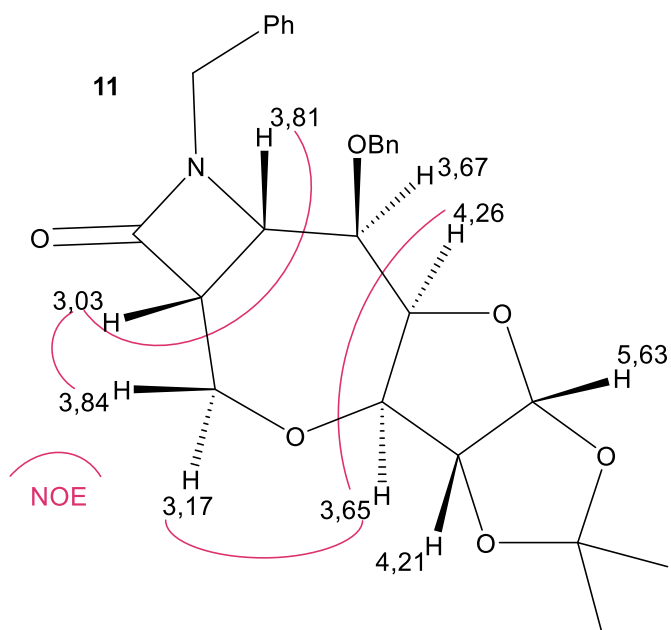
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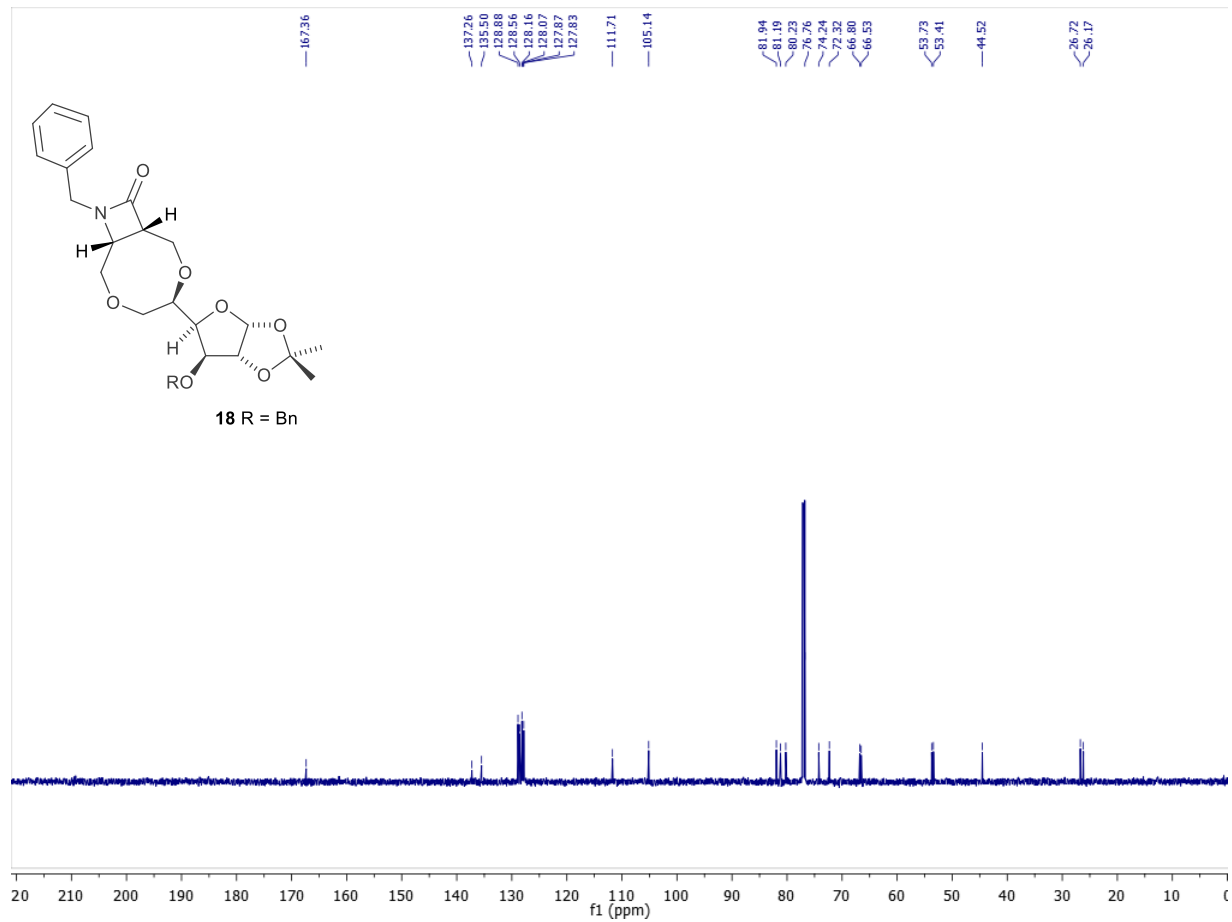
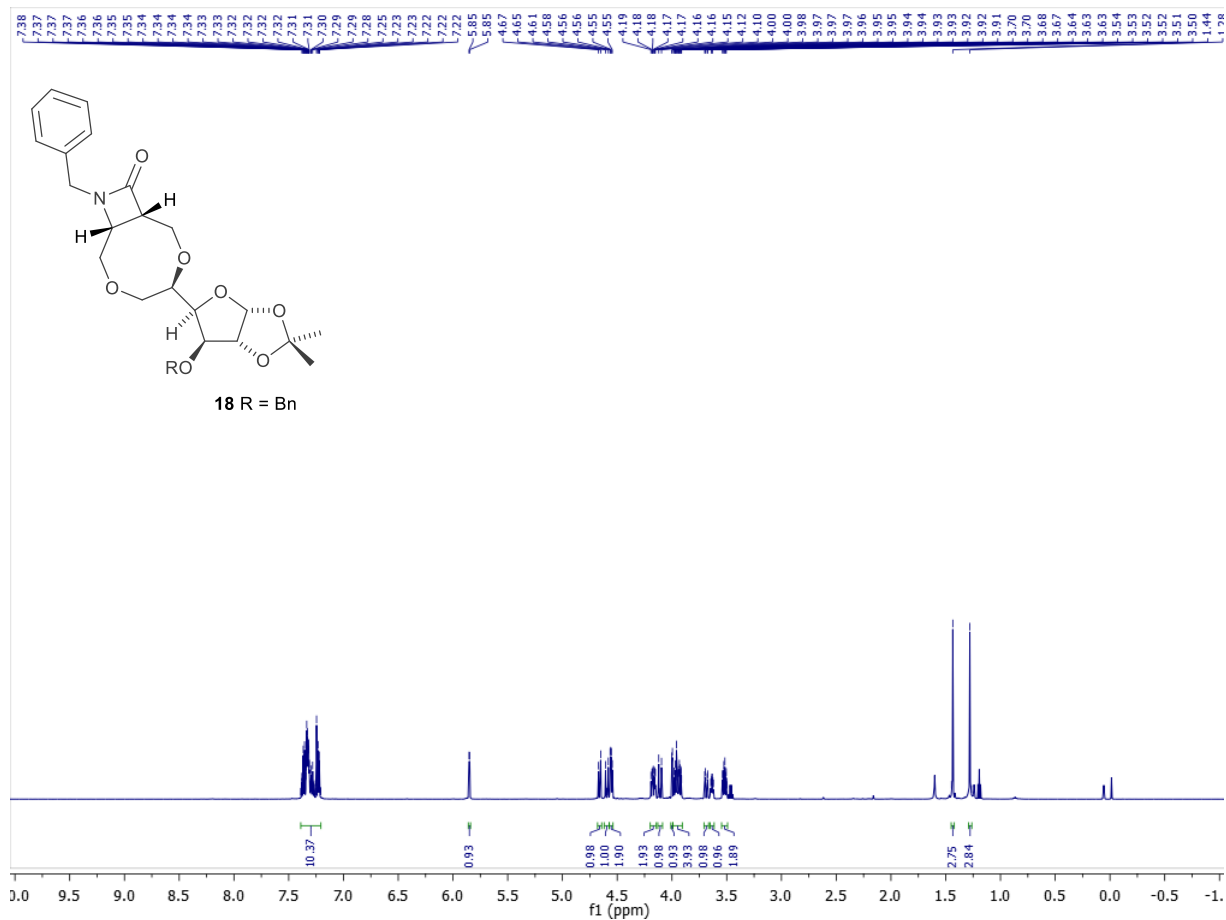


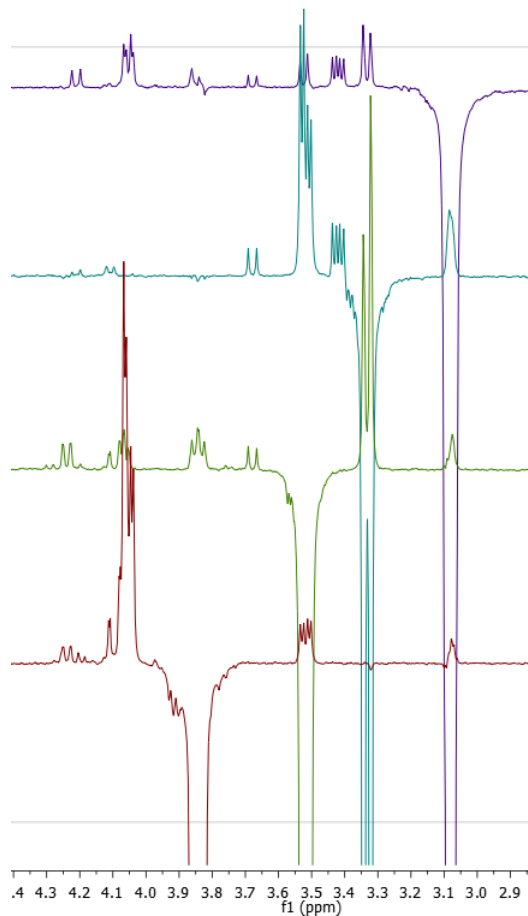




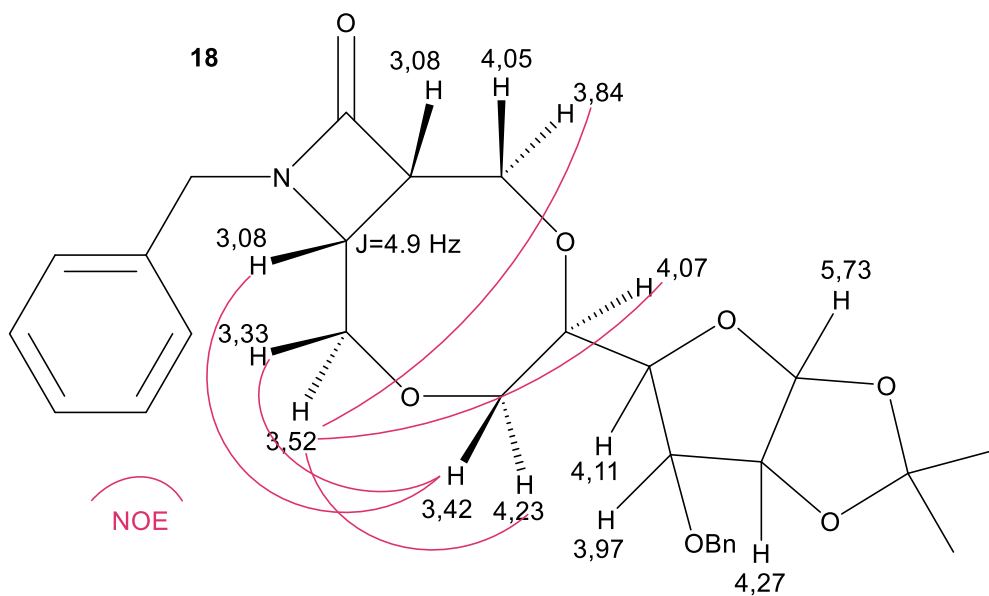
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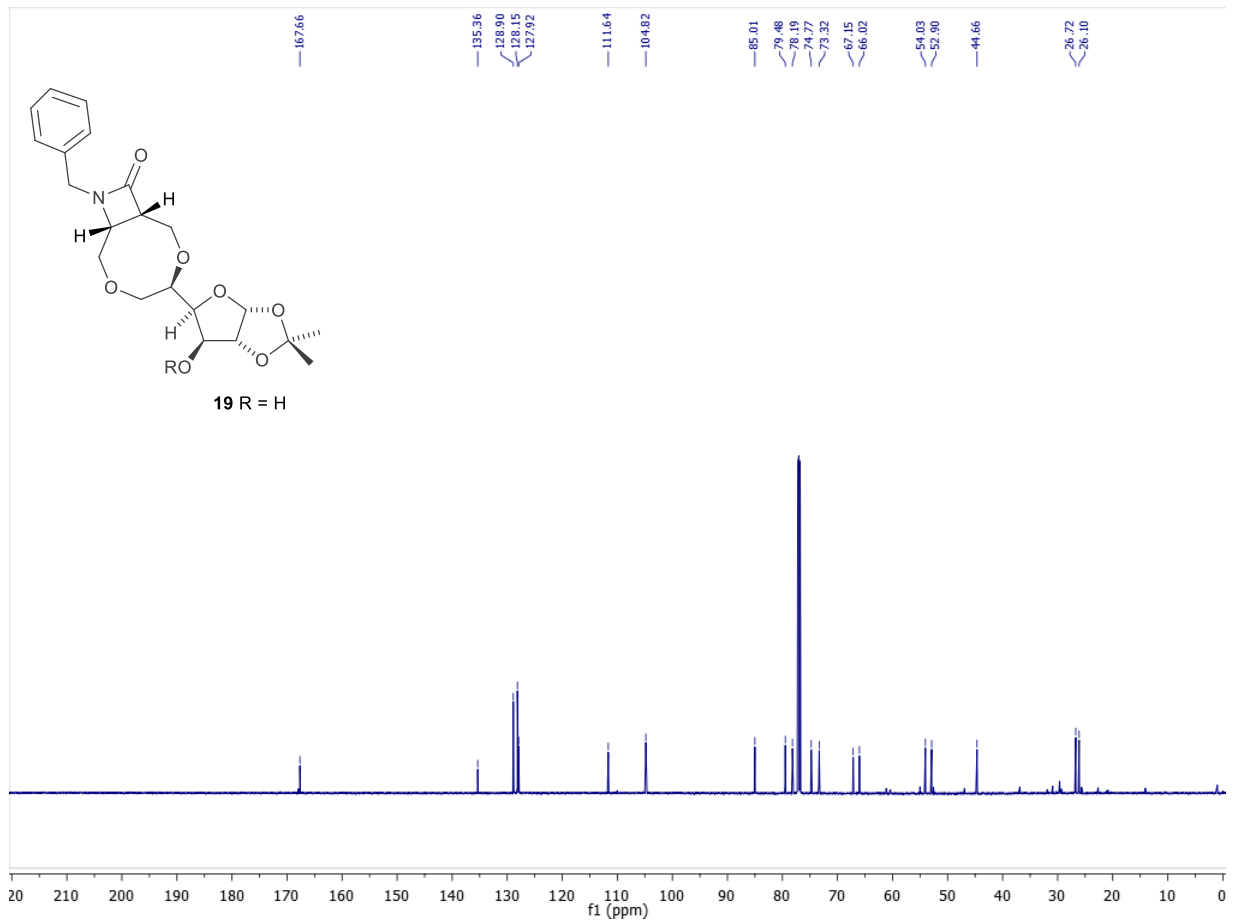
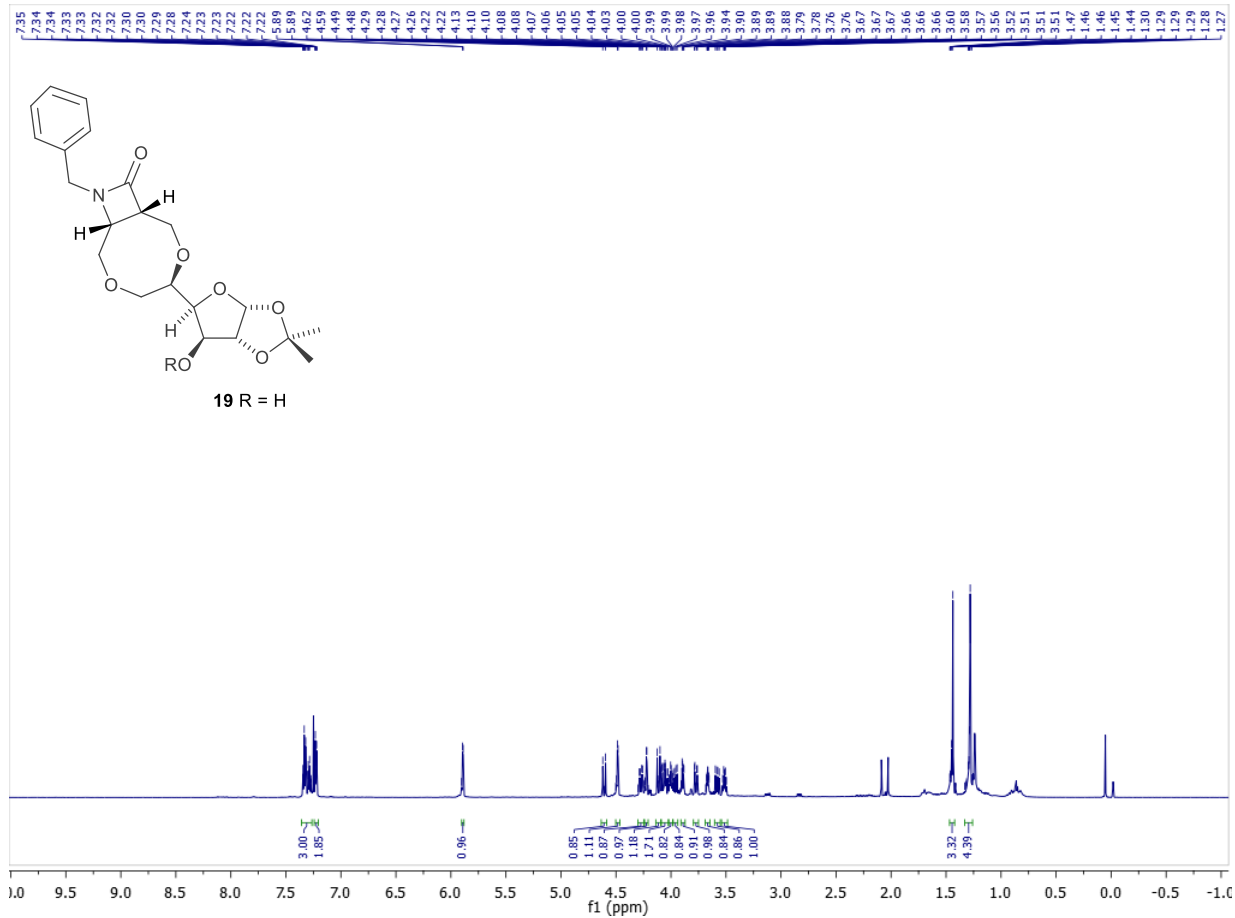


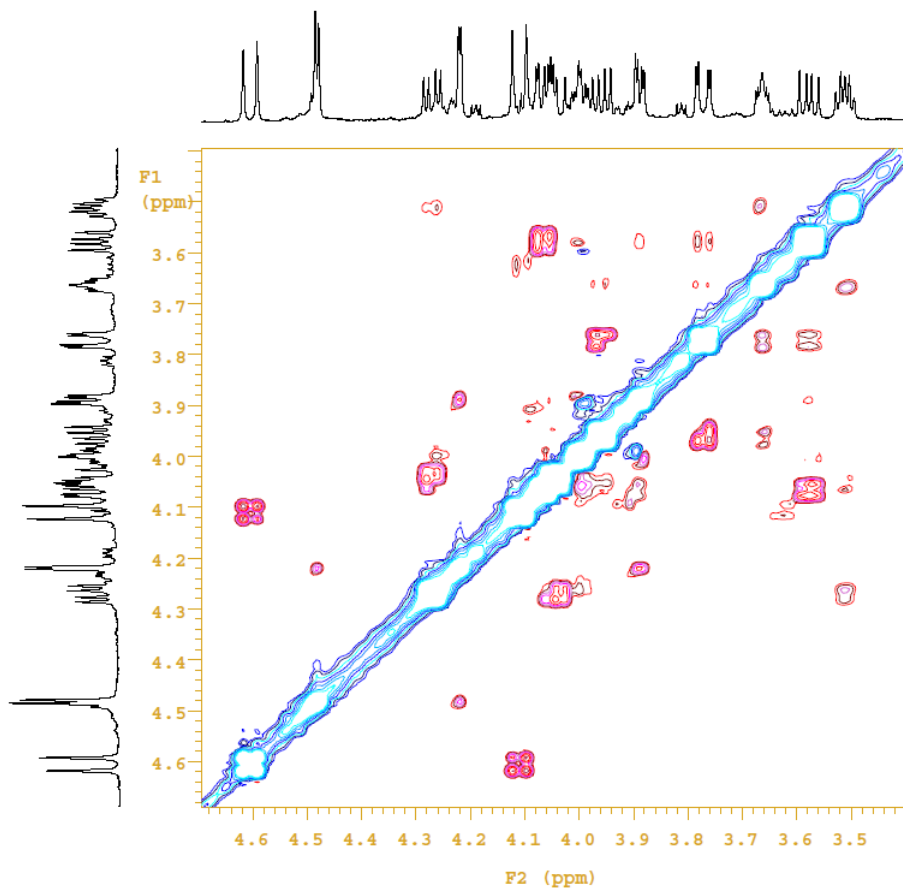




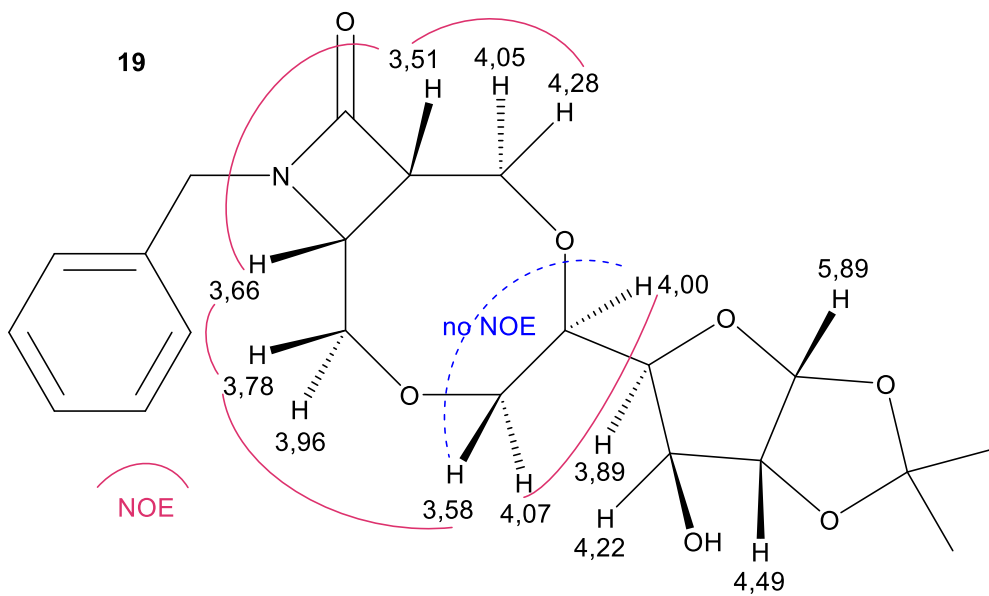
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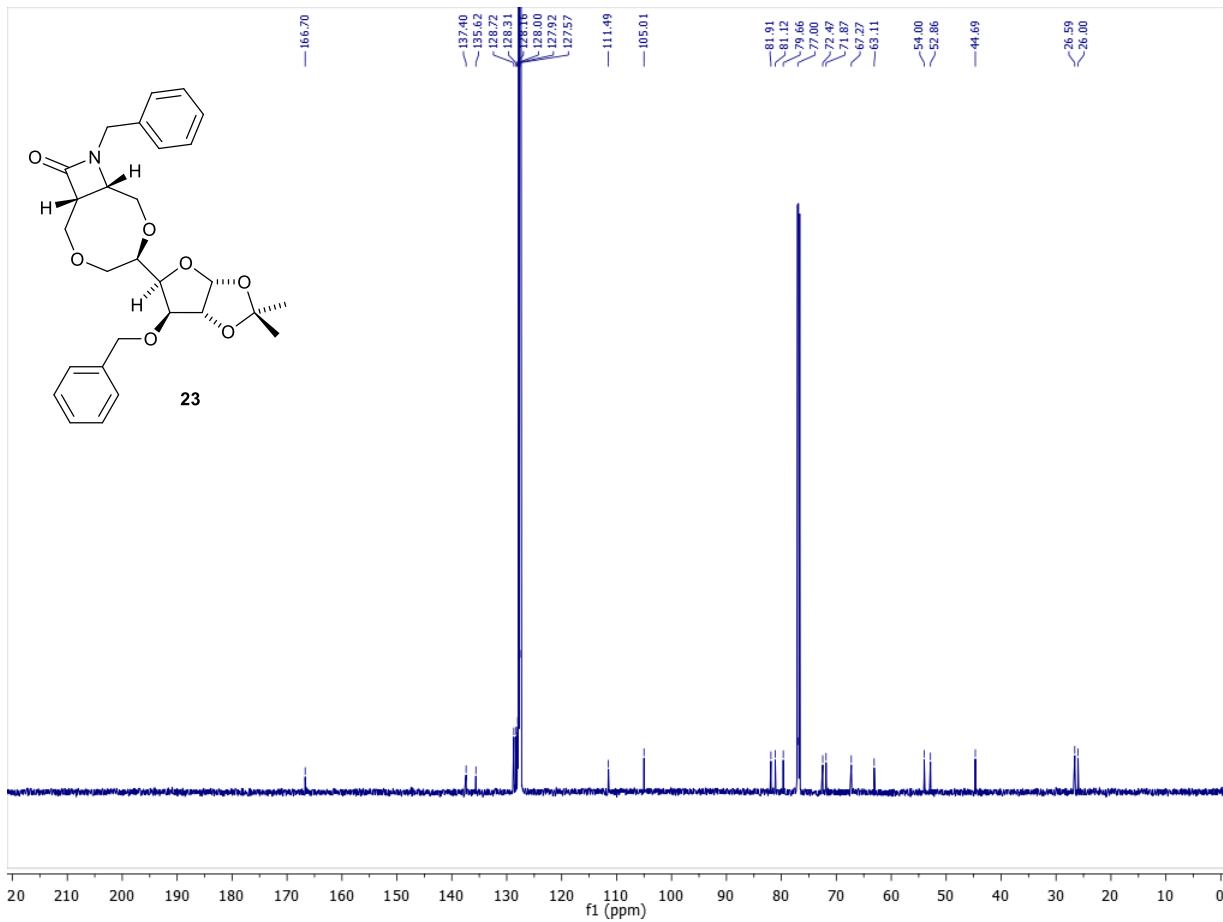
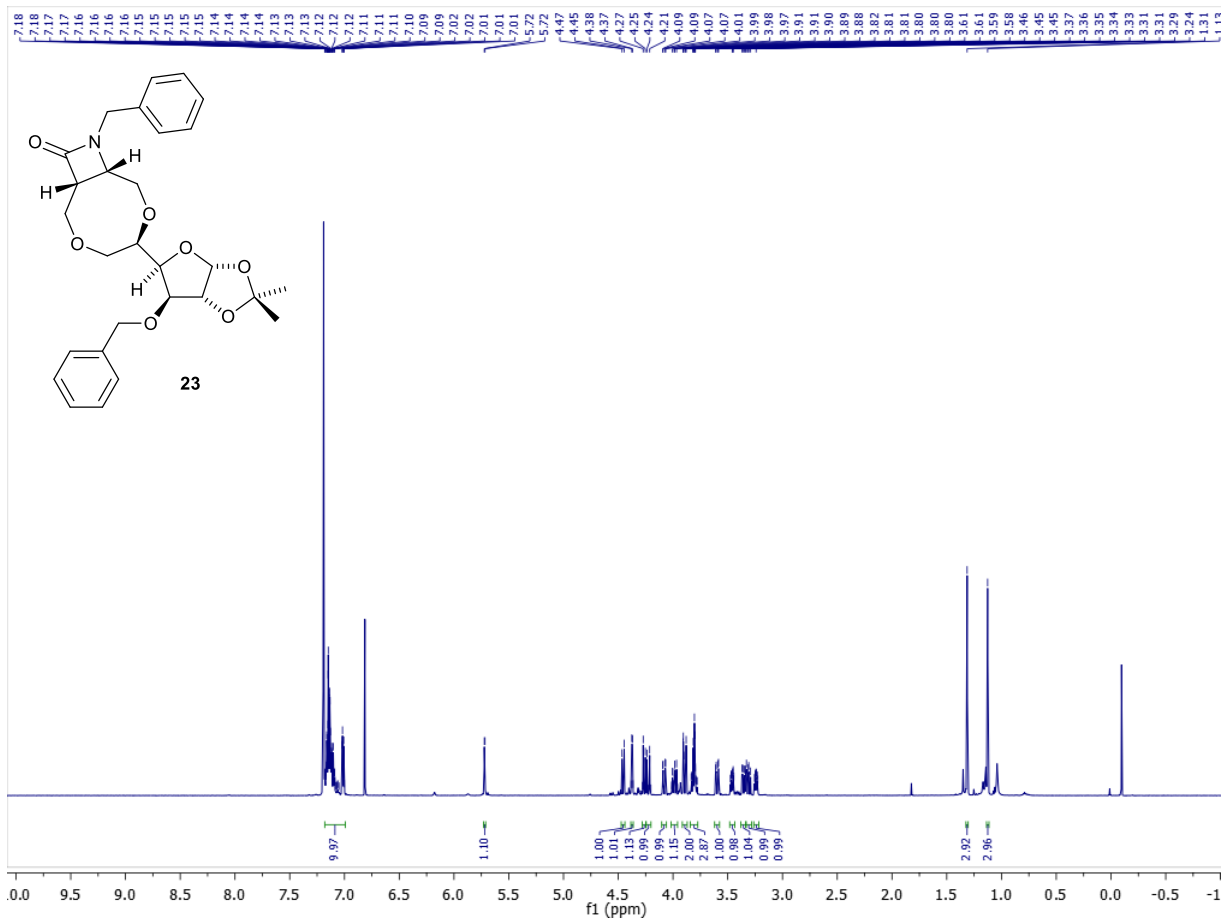


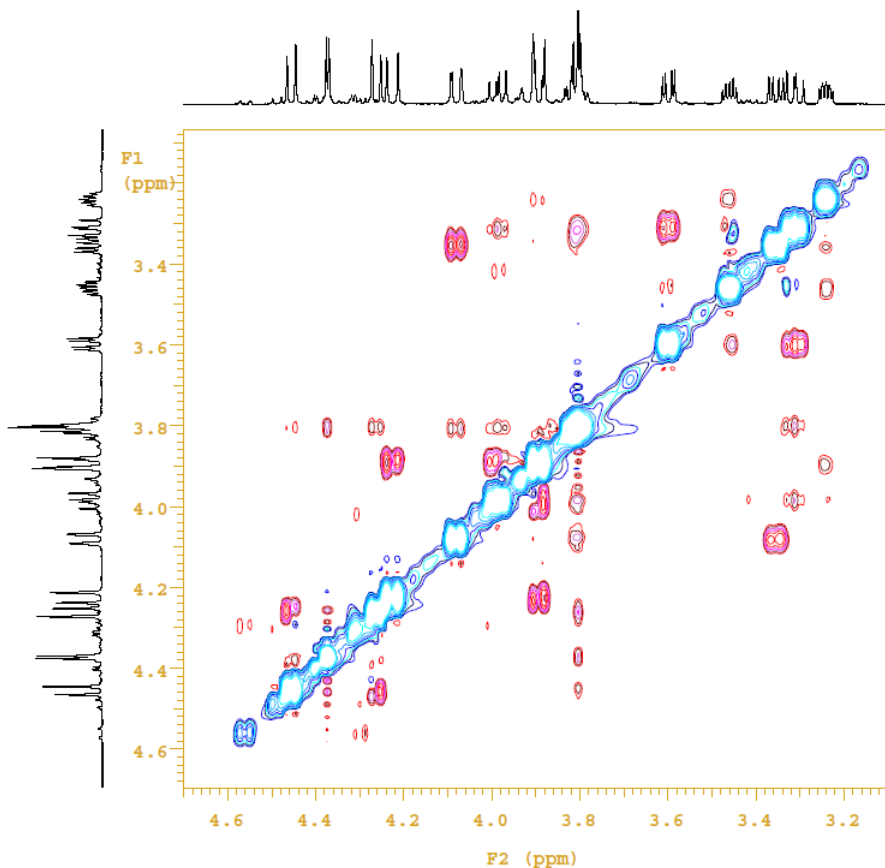




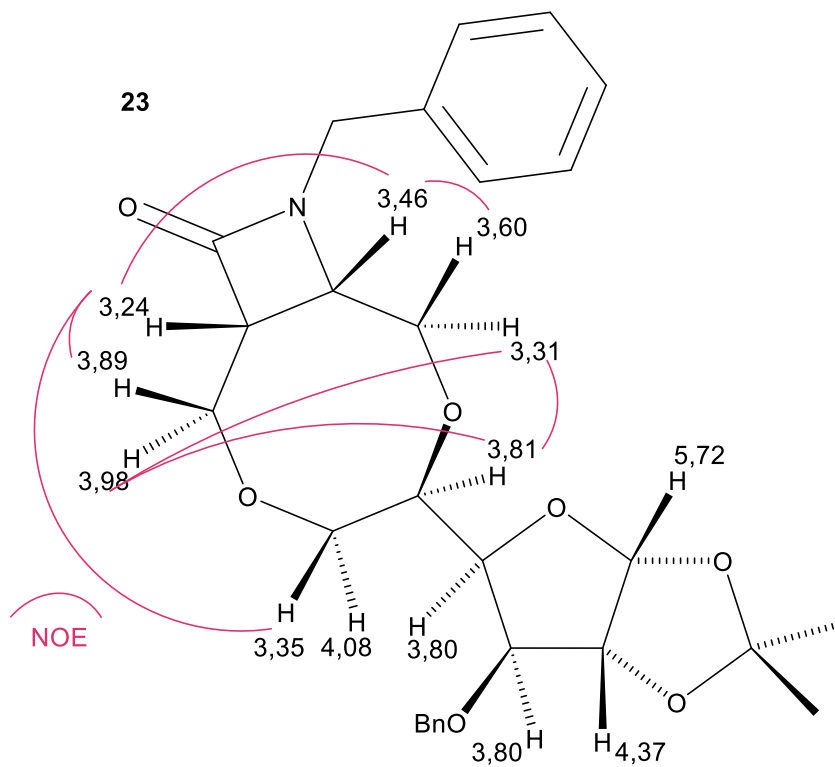
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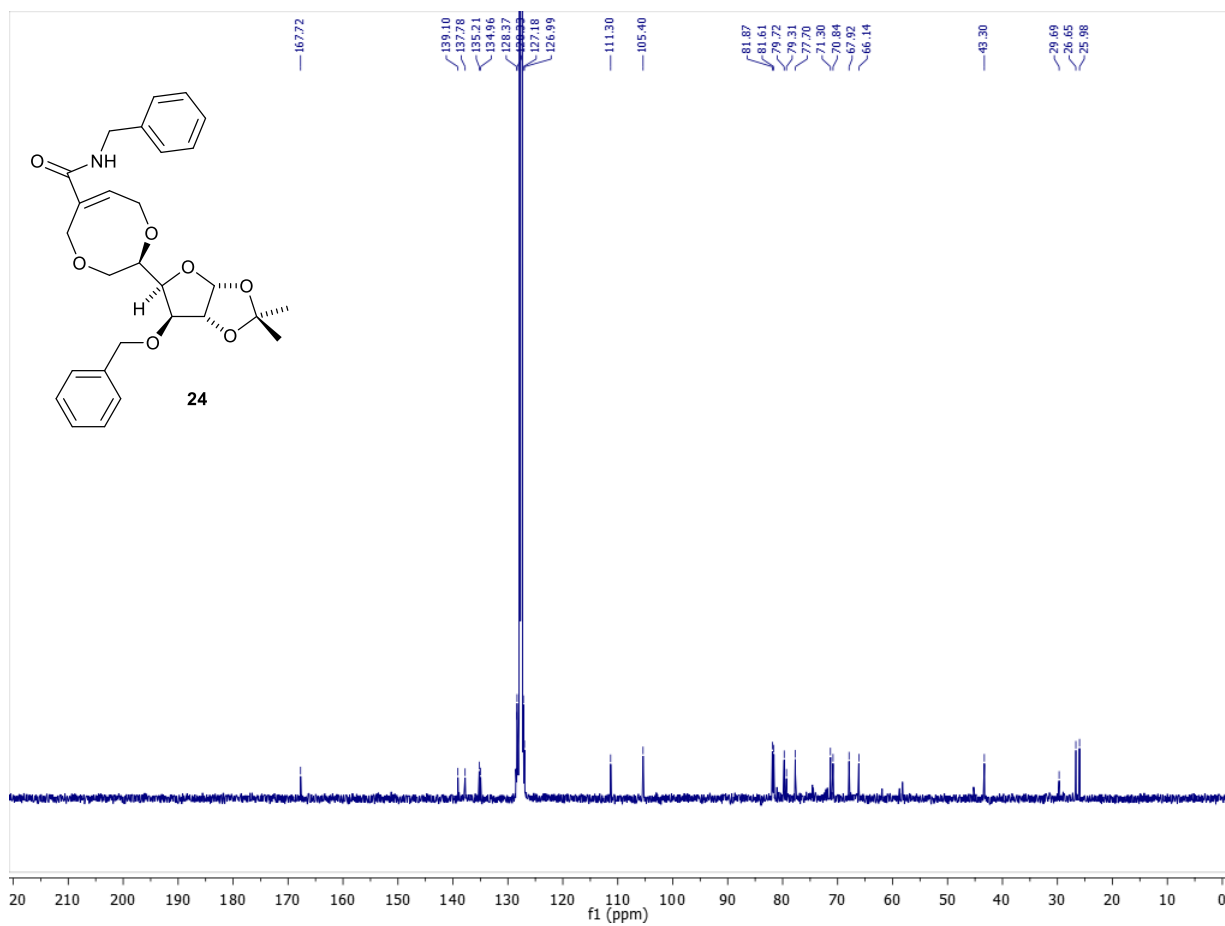
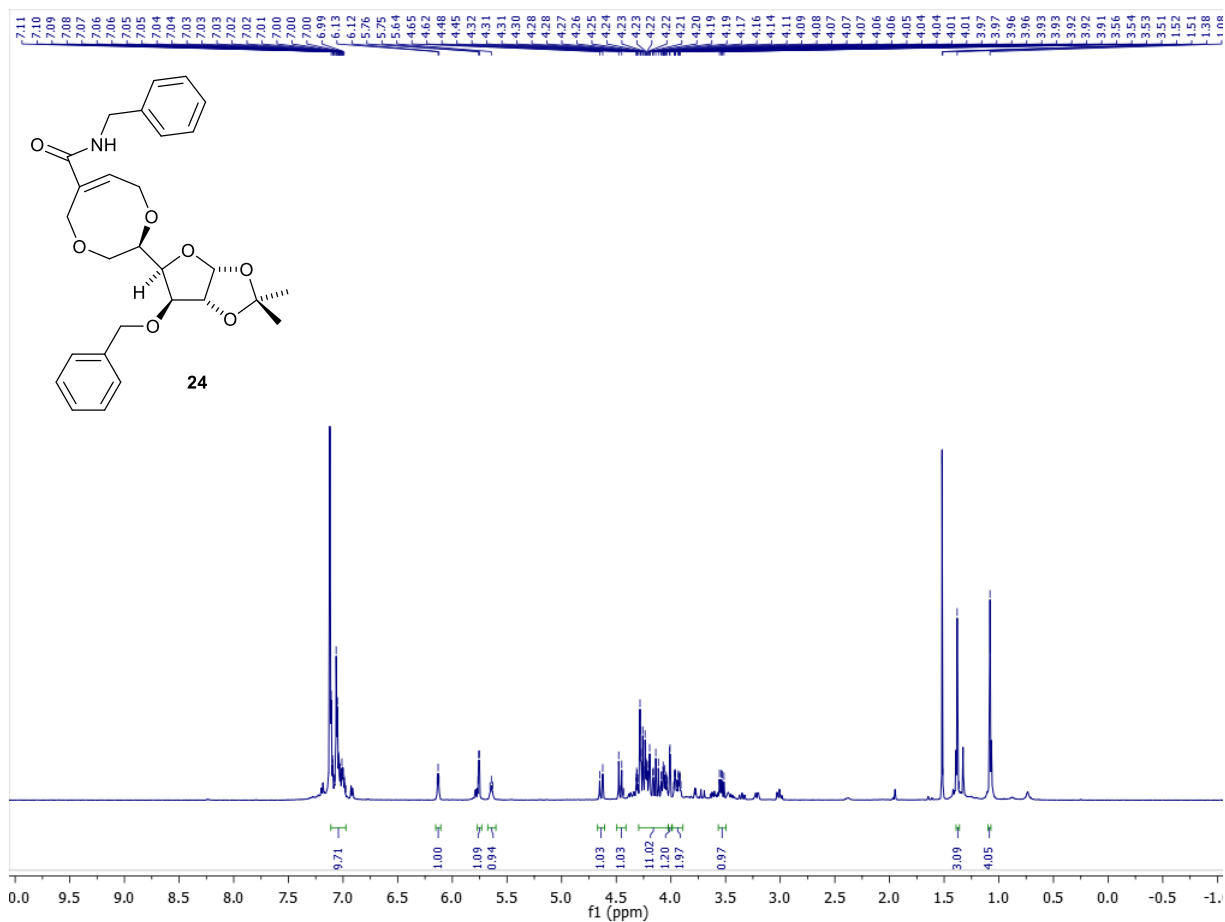


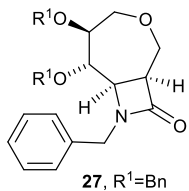
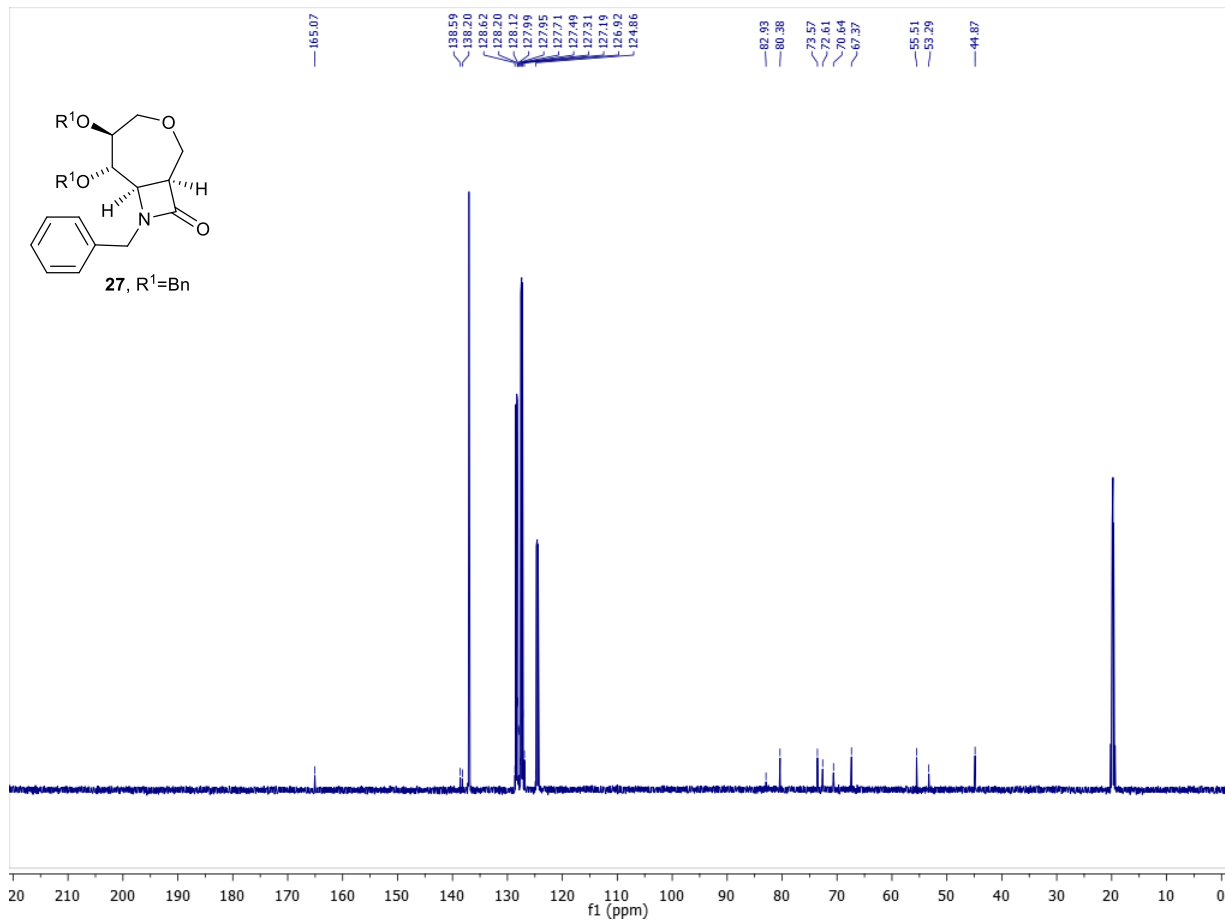
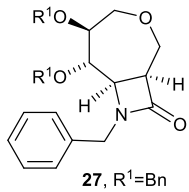
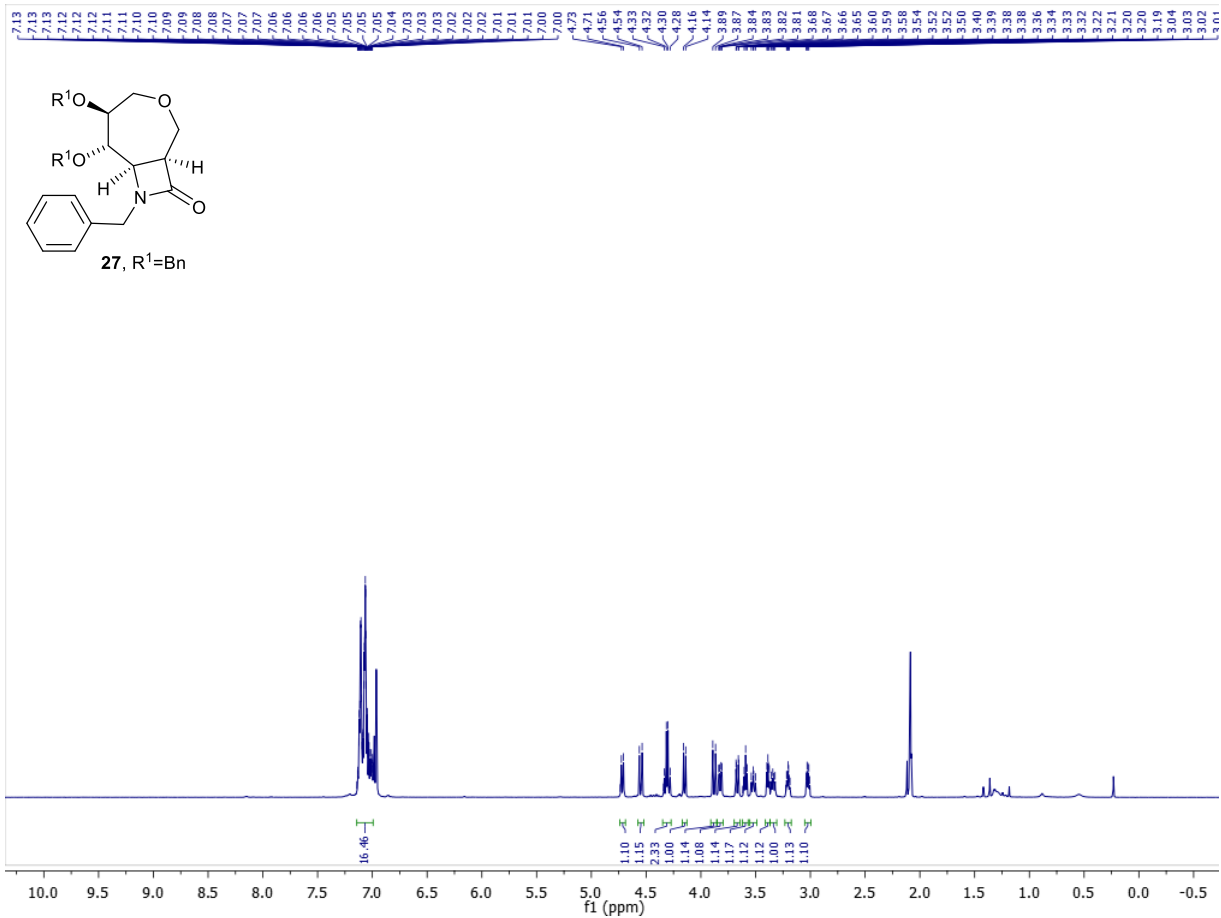


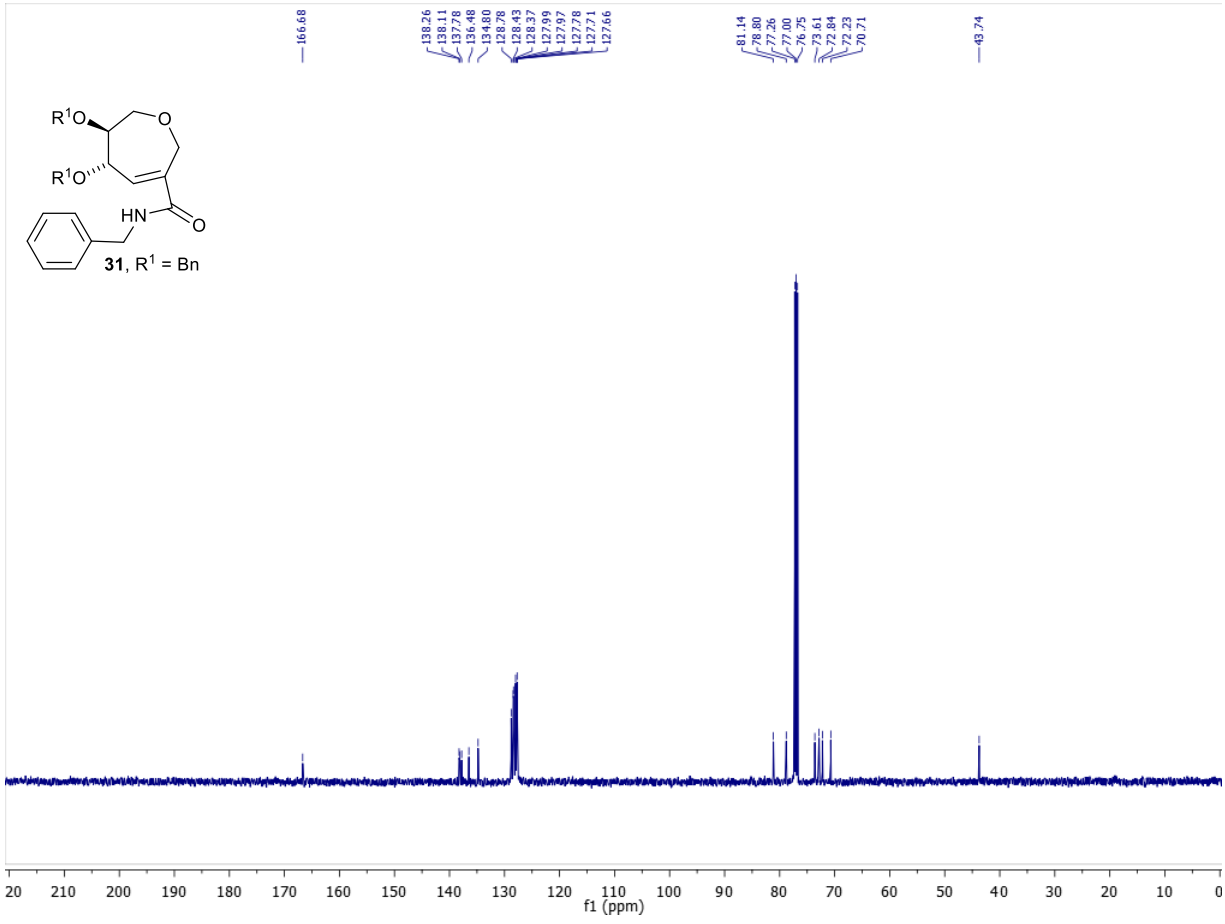
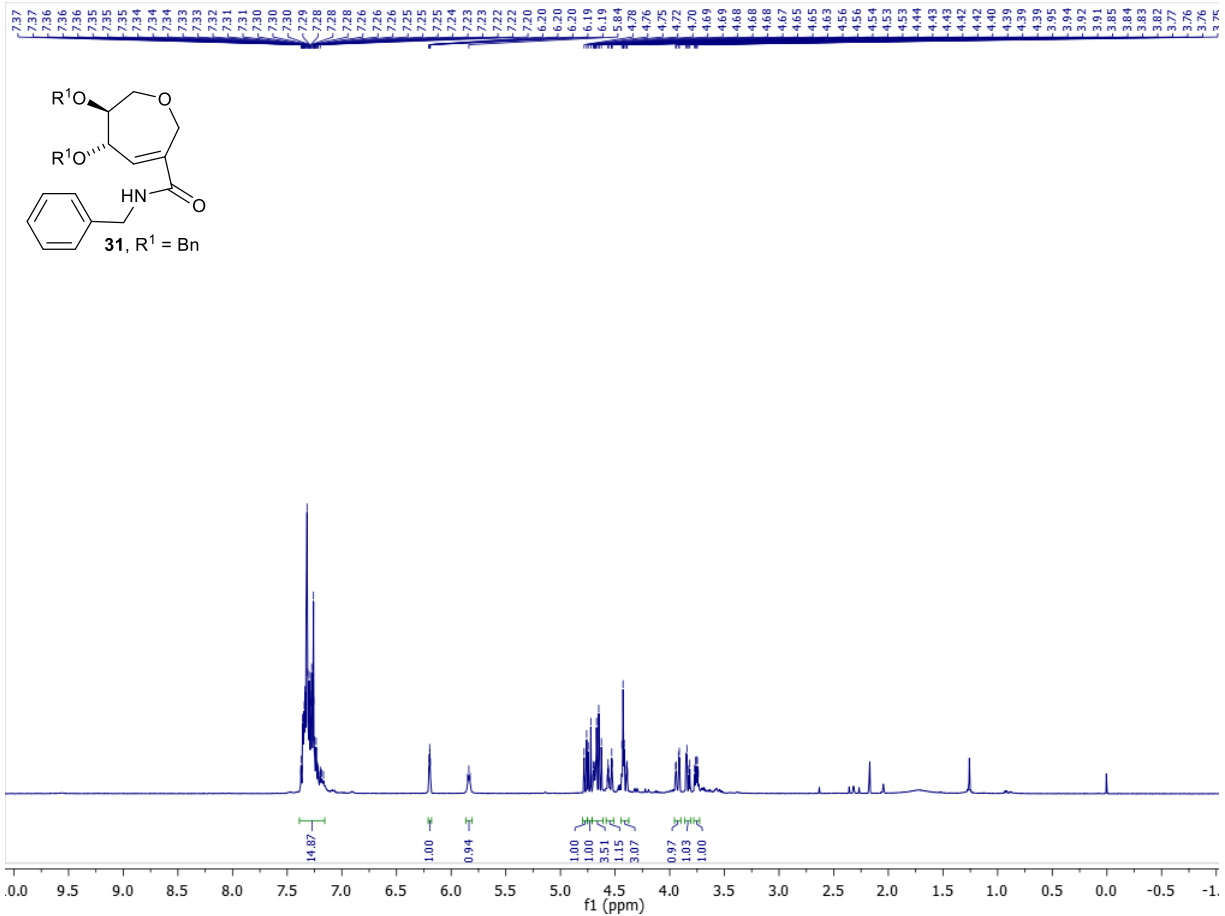


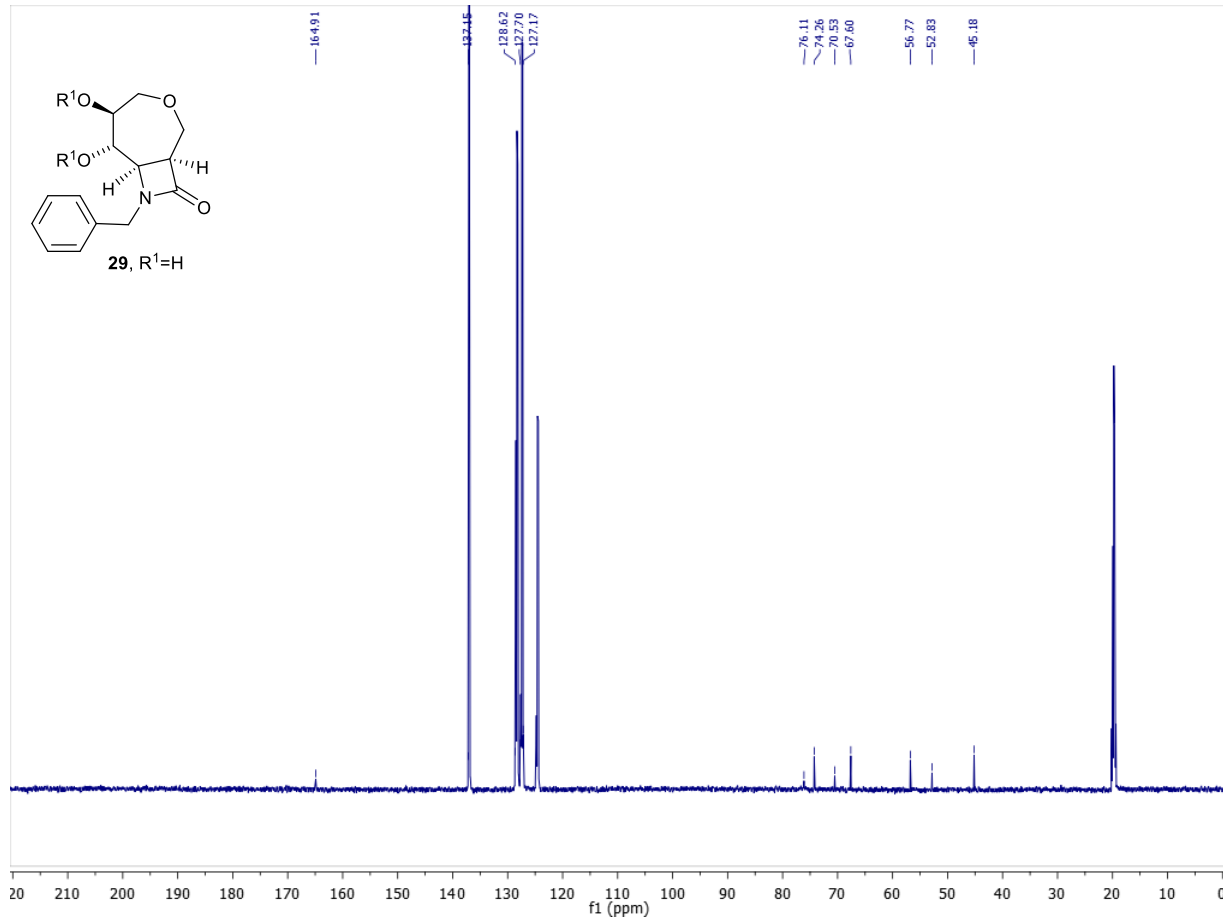
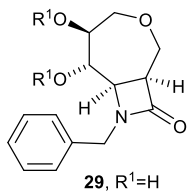
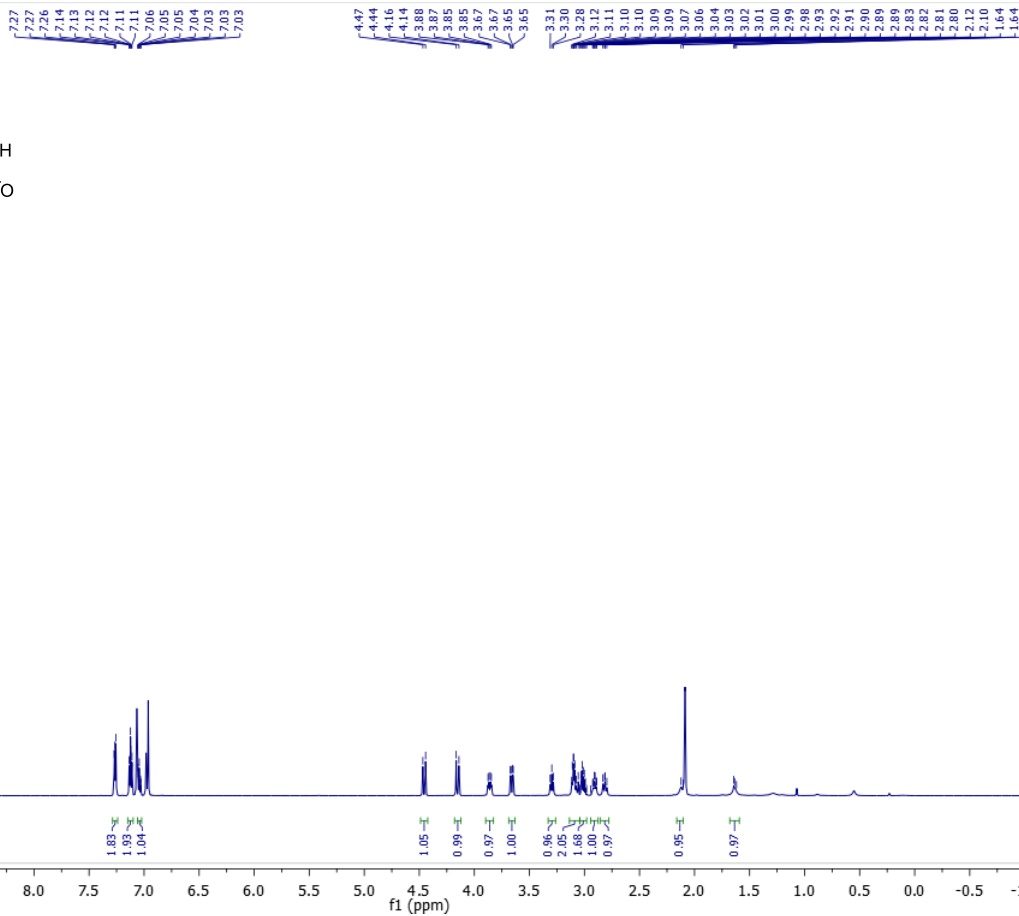
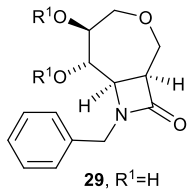
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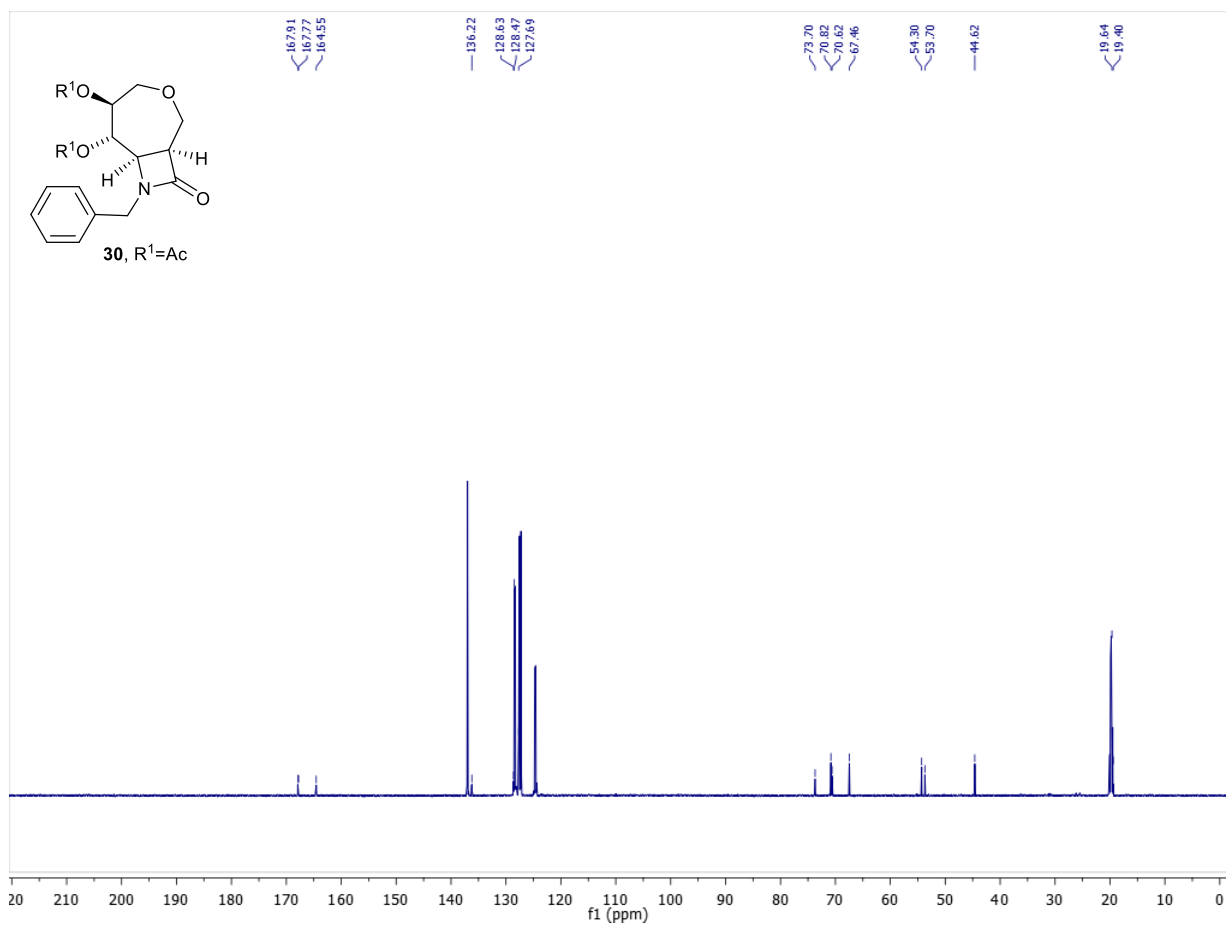
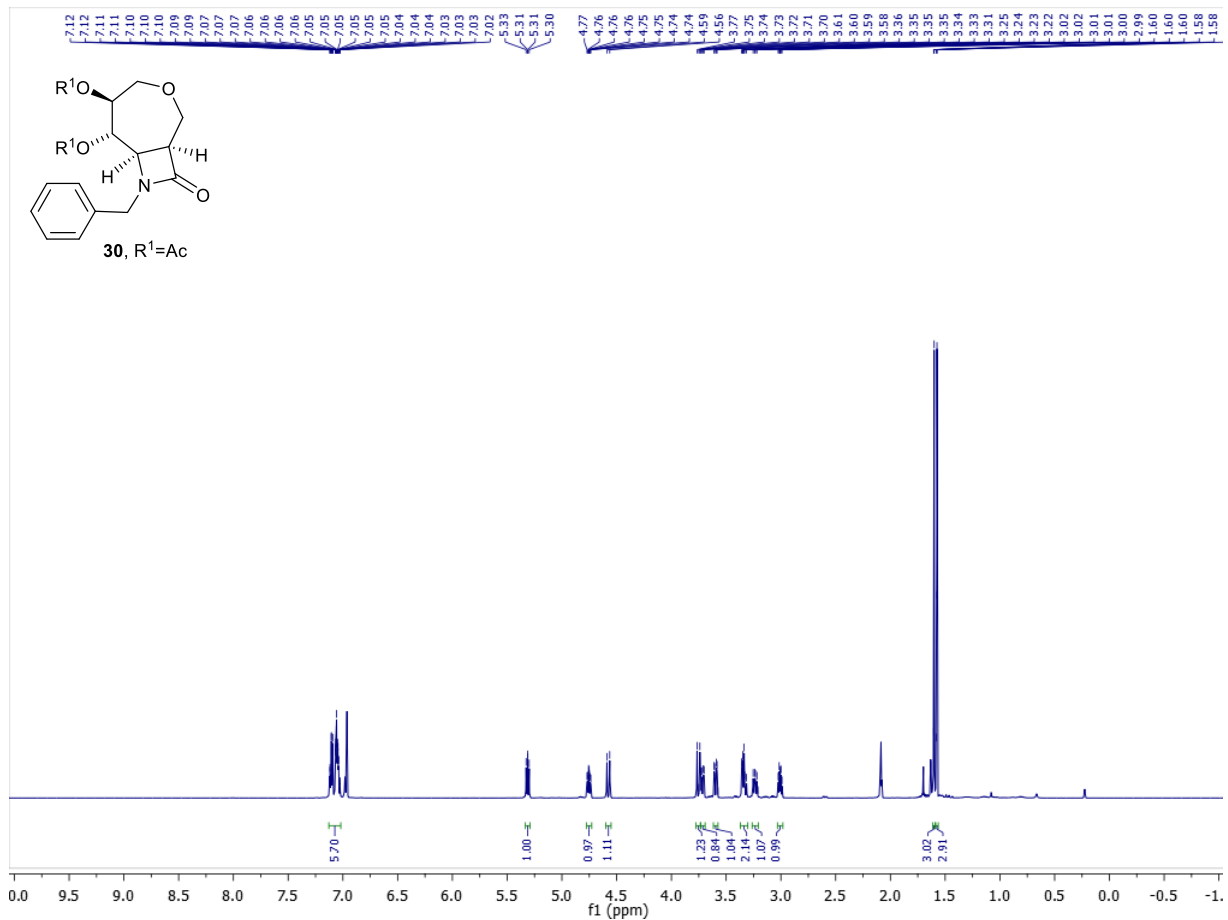


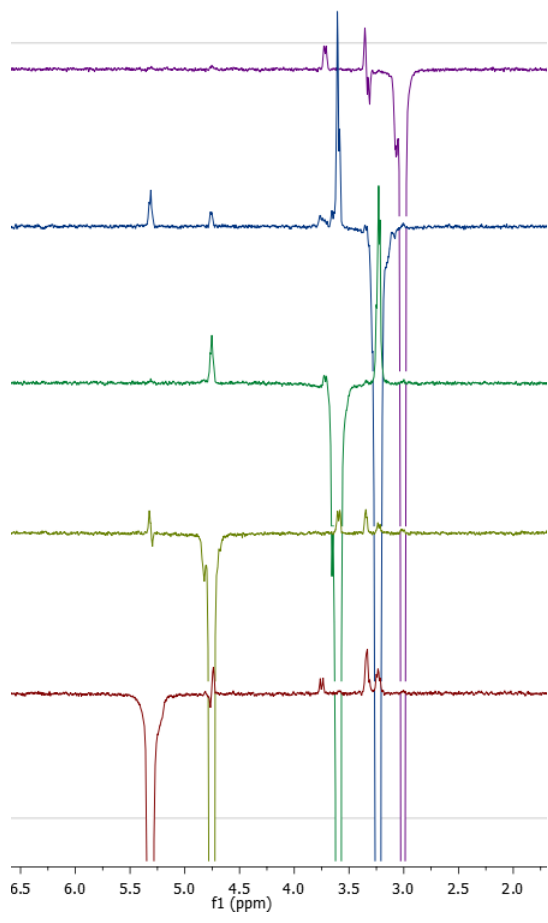




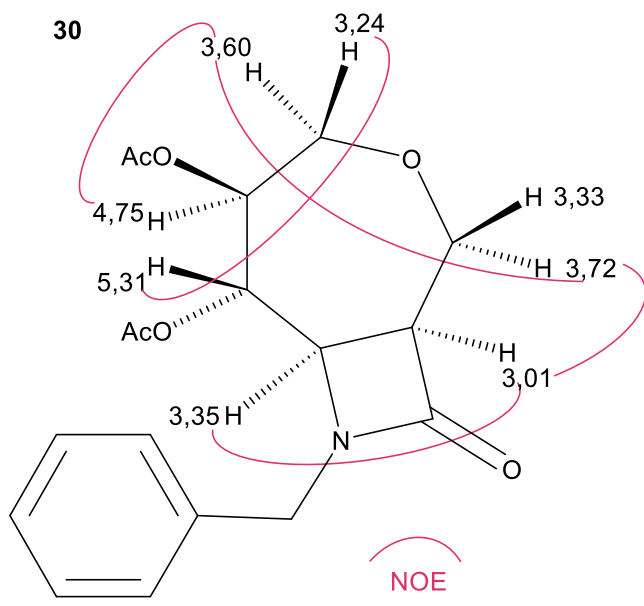


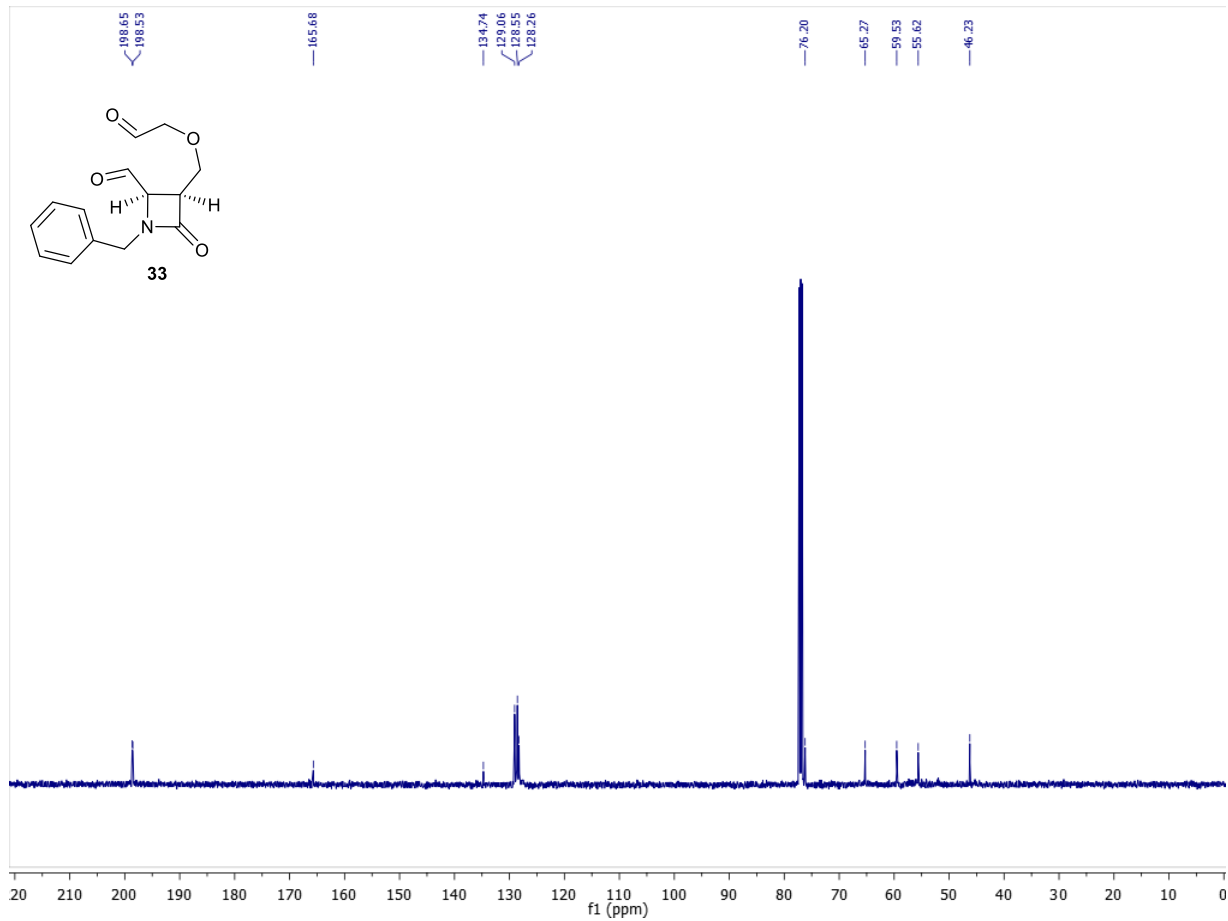
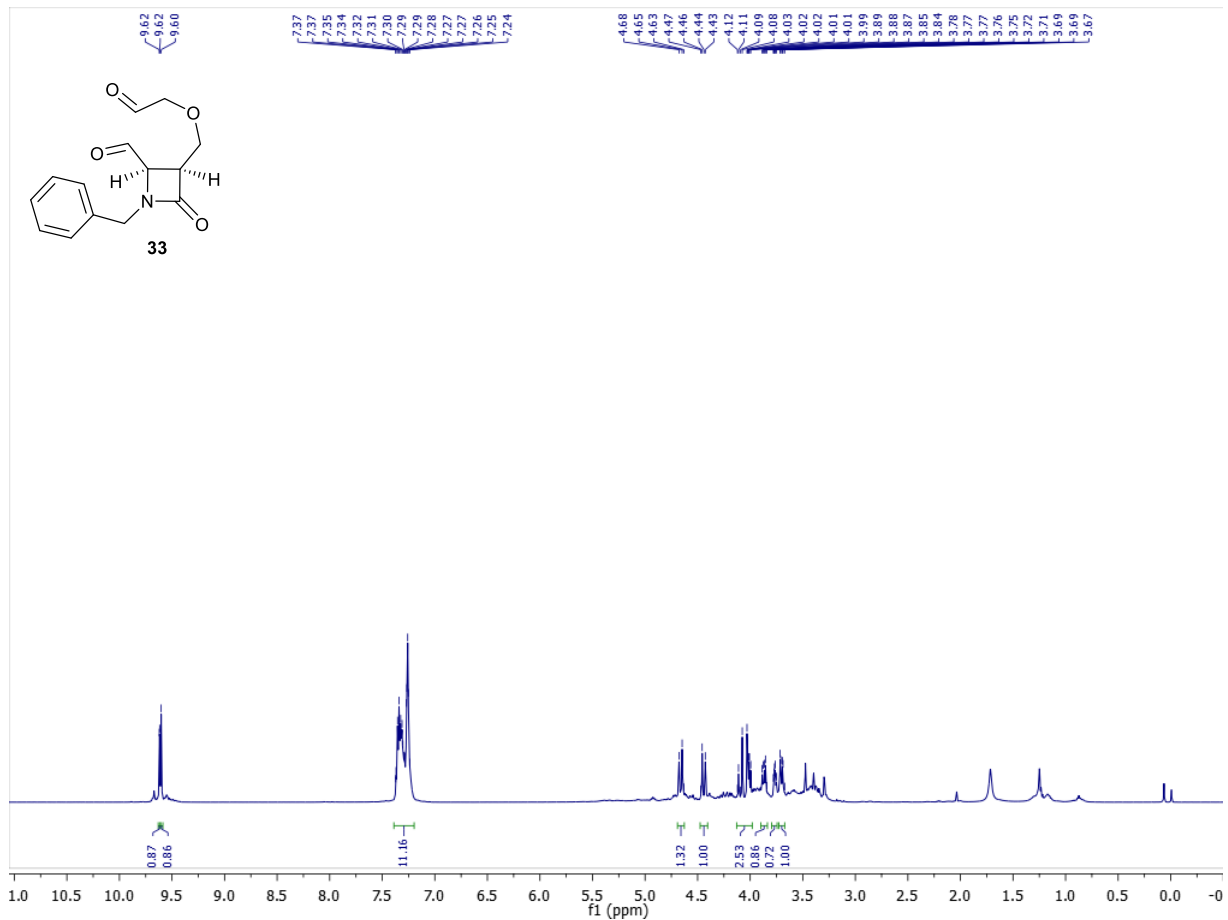


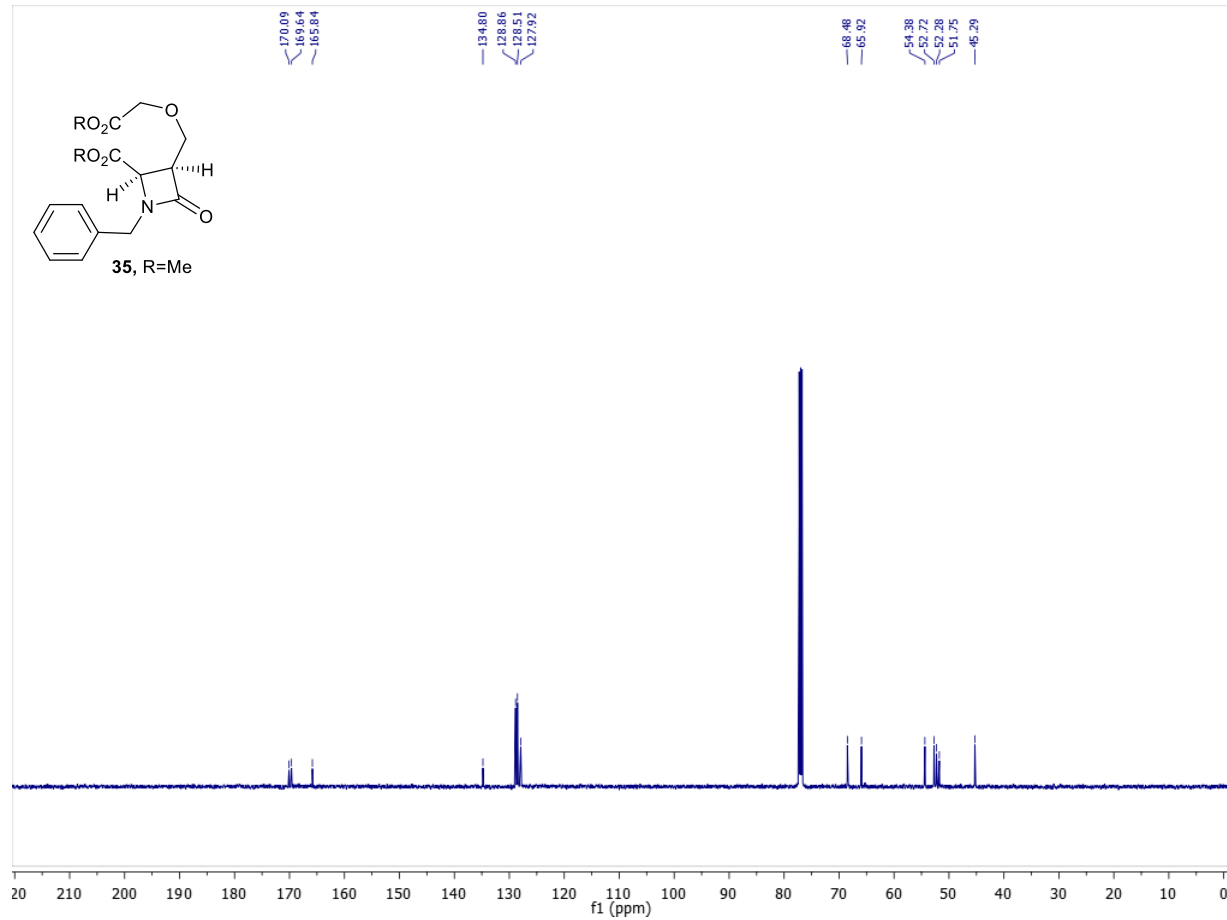
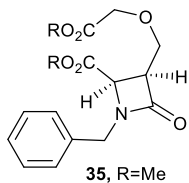
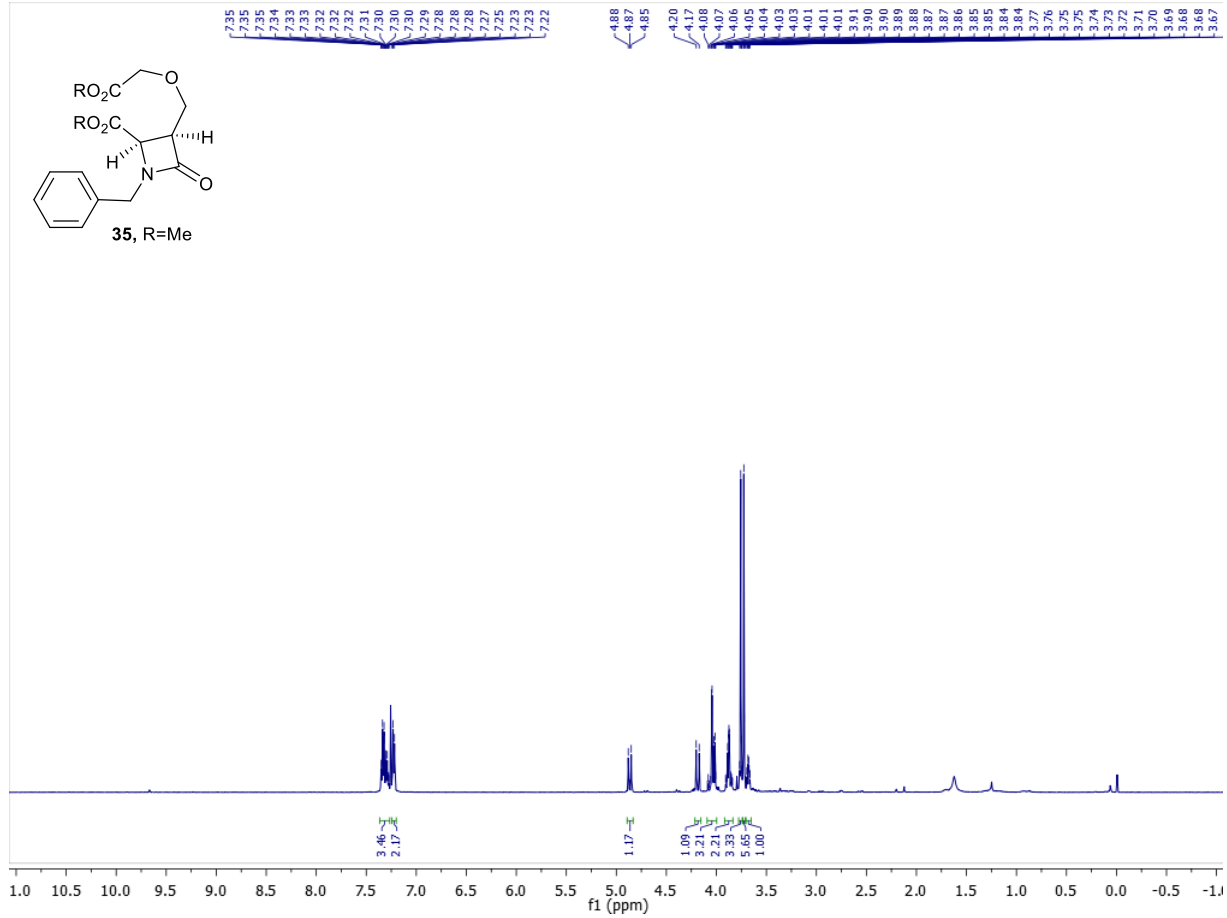
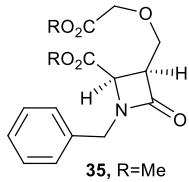


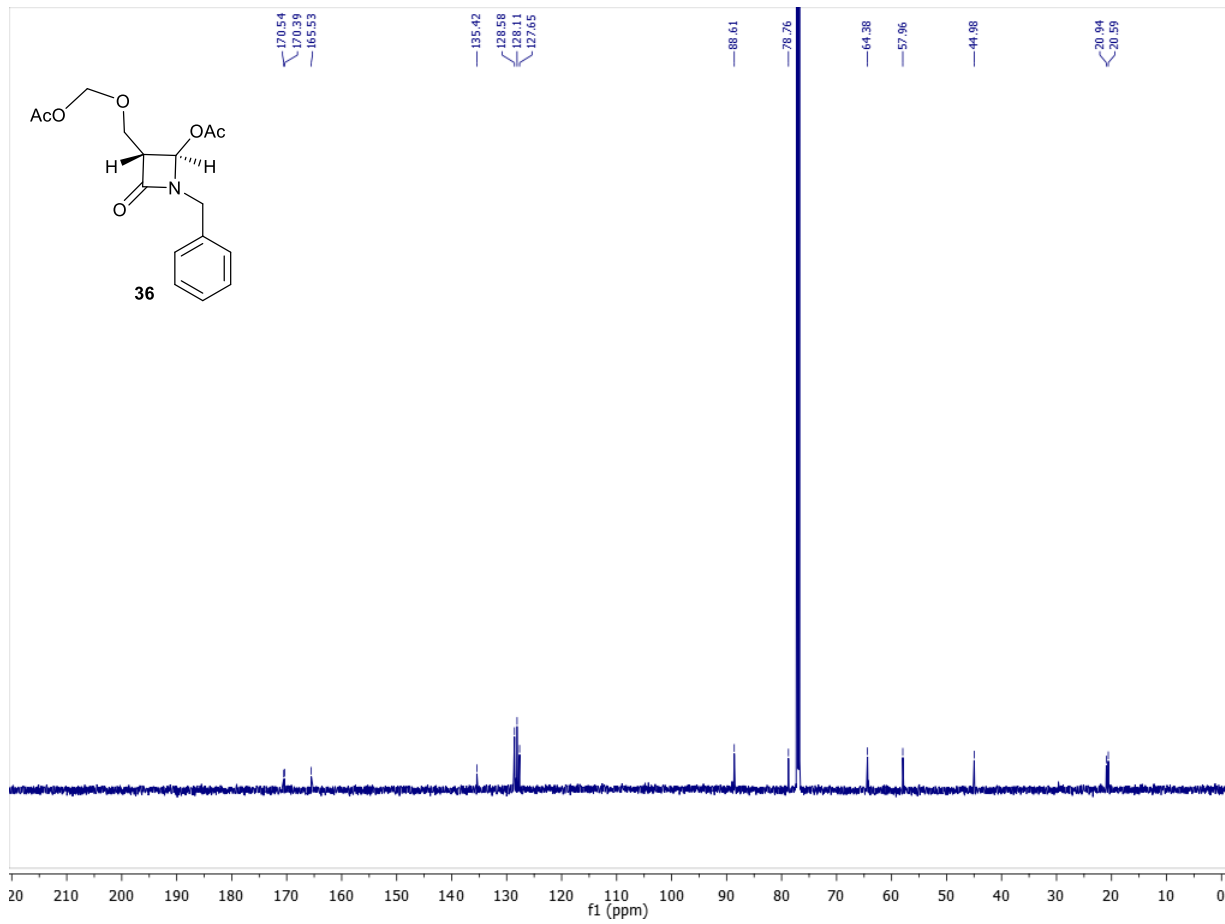
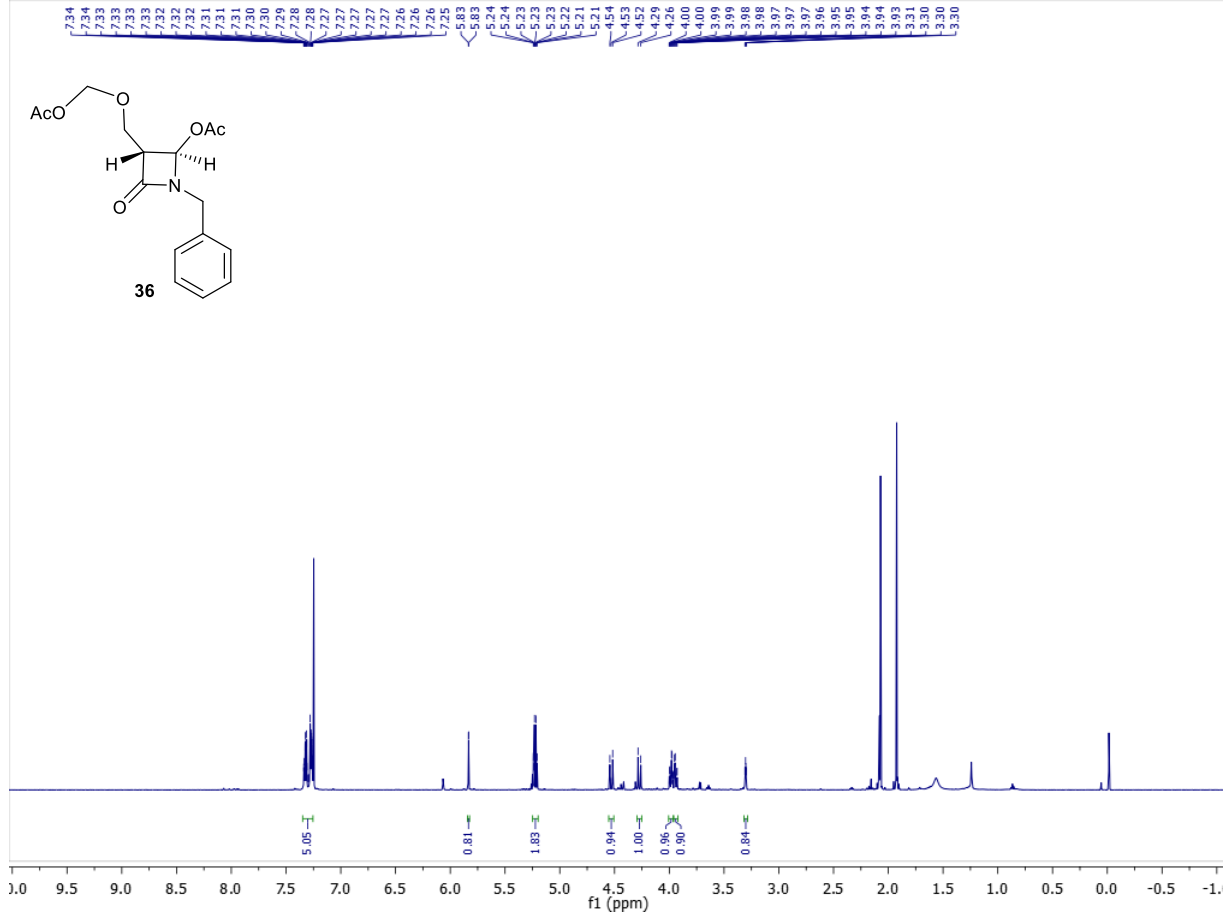


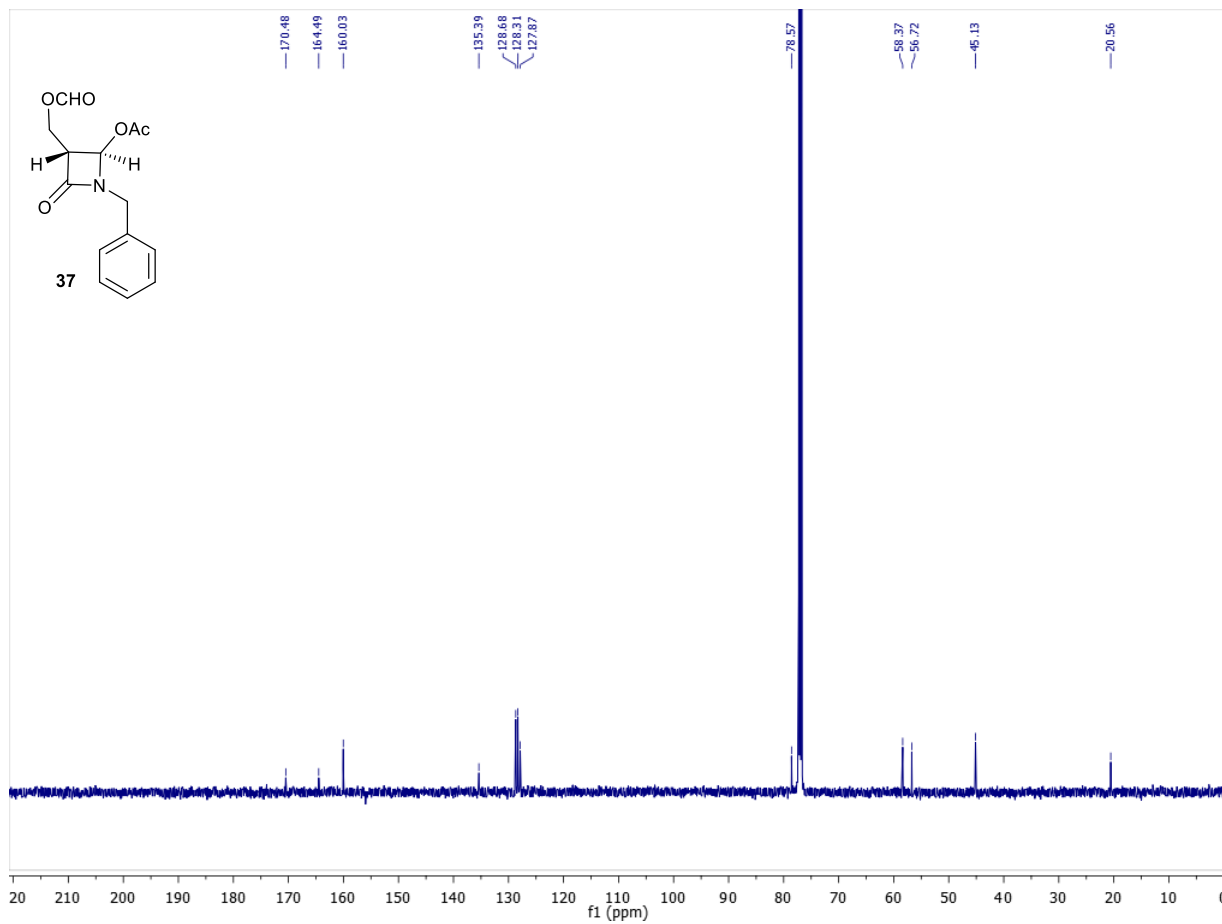
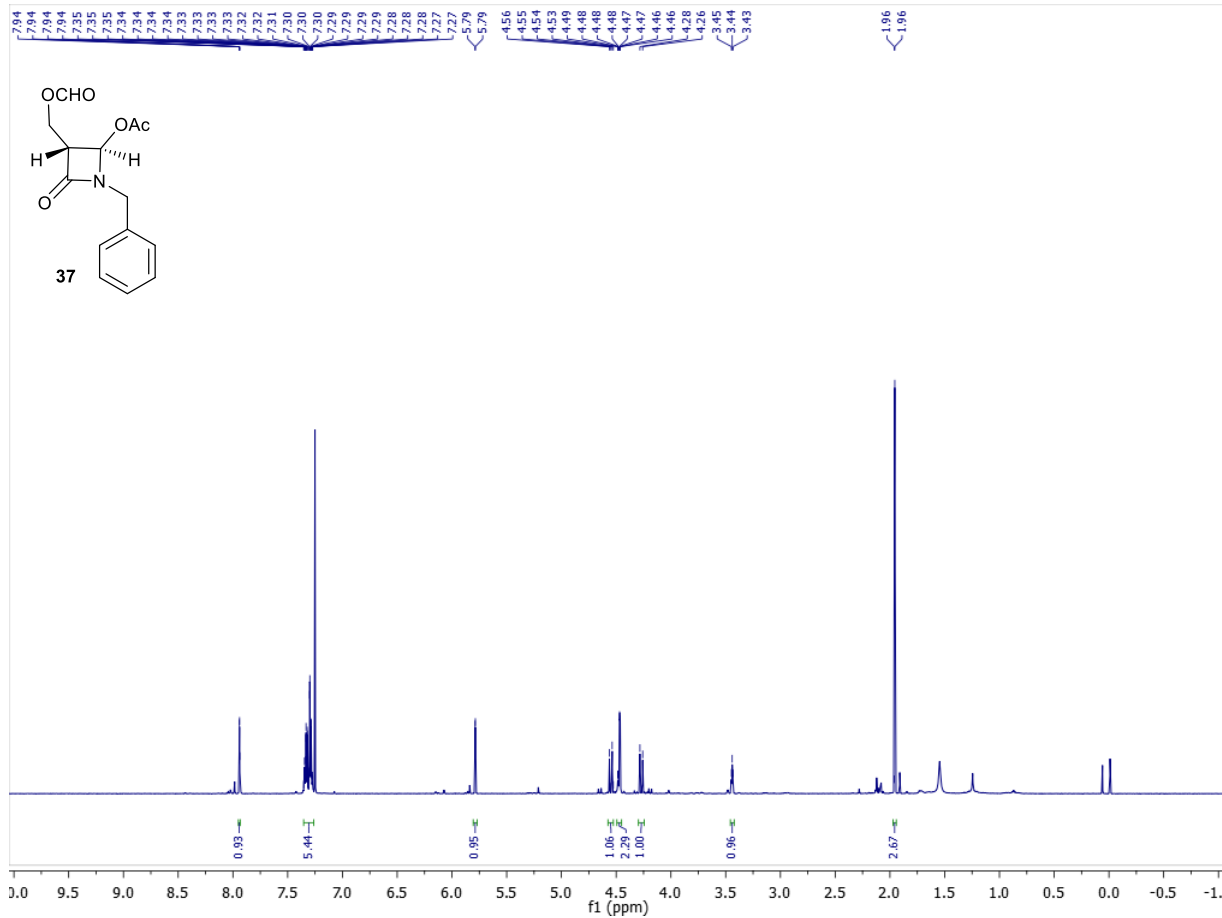
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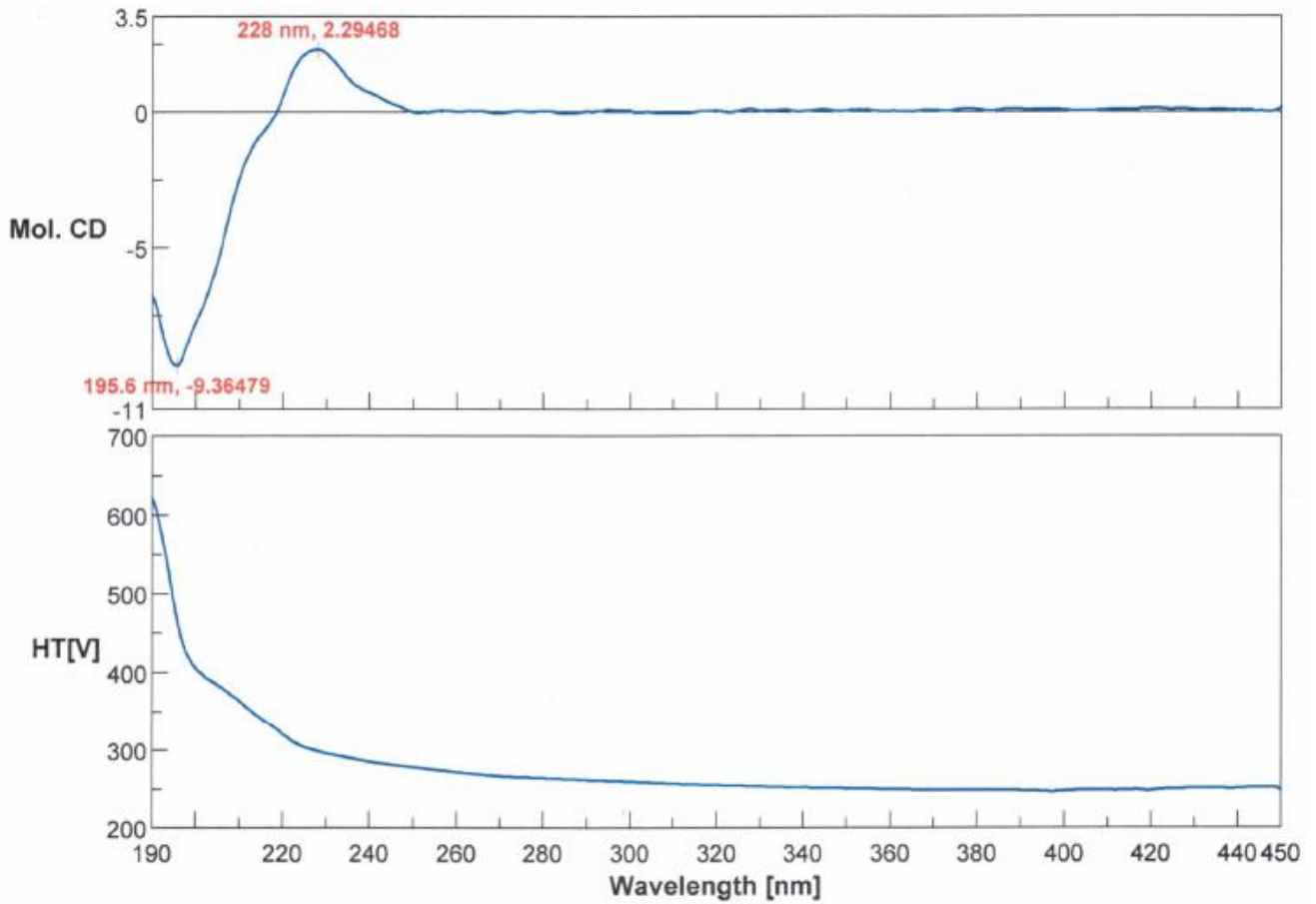






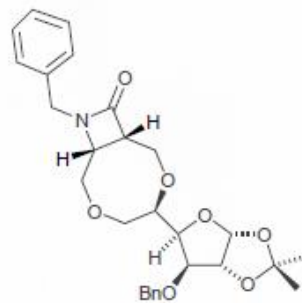


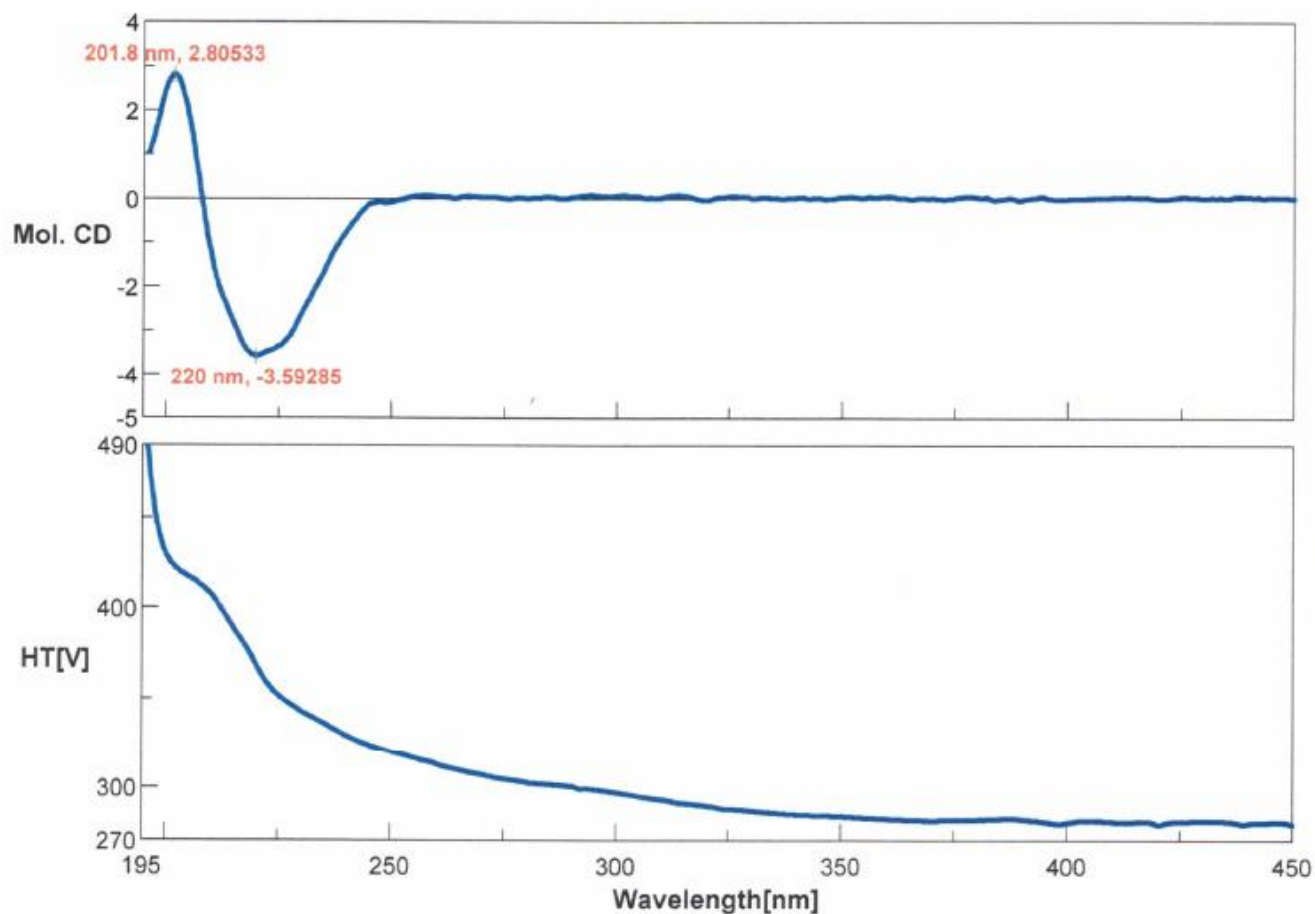




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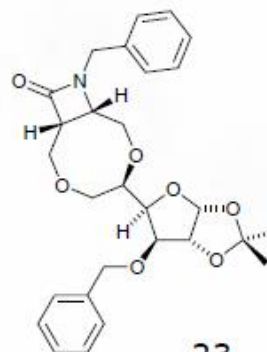
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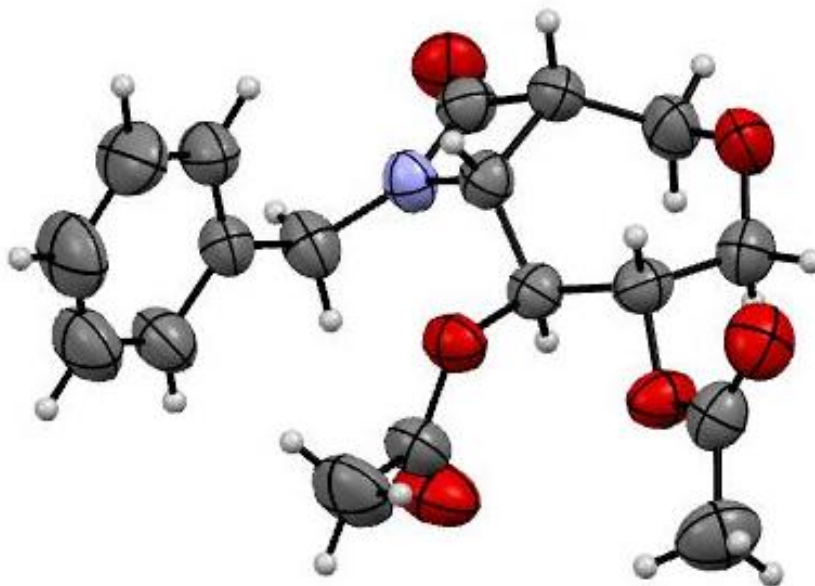


Figure S1. ORTEP plot of **30**, represented by thermal ellipsoids shown at the 50% probability level.

Colorless crystals suitable for X-ray structural analysis were obtained by slow evaporation of heptane–ethyl acetate solution of **30**. Crystal data were obtained on a Bruker X8 APEX II diffractometer with CCD detector employing graphite monochromated Cu-K α radiation ($\lambda=1.54178$ Å) at 296(2) K and operating in the ϕ - ω scan mode. The structure was solved by direct methods SHELXS-2014^[1] and refined with full-matrix least-squares calculations on F^2 using SHELX-2014. ^[1] All non-hydrogen atoms were refined anisotropically. The hydrogen atom positions were geometrically idealized and allowed to ride on their parent atoms. Crystallographic data for **30** have been deposited at the Cambridge Crystallographic Data Centre (deposition no. **CCDC 1975296**). Copies of these data can be obtained free of charge *via* www.ccdc.cam.ac.uk/conts/retrieving.html or from the Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge CB21EZ, UK. [fax: (+44) 1223-336-033; or email: deposit@ccdc.cam.ac.uk].

[1] G. M. Sheldrick, SHELXL-2014. Program for the Refinement of Crystal Structures from Diffraction Data, University of Göttingen, Germany, 2014.

Table S1. Summary of the crystal parameters and refinement metrics for **30**.

Chemical formula	C ₁₈ H ₂₁ NO ₆	θ range [°]	3.53 to 68.42
Formula weight	347.36 g/mol	Index ranges	-6<= <i>h</i> <=5, -20<= <i>k</i> <=19, -22<= <i>l</i> <=20
CCDC number	1975296	Reflections collected	14421
Crystal appearance	colorless needle	Reflections unique	3302 [R(int) = 0.0711]
Crystal size [mm]	0.070 x 0.103 x 0.558	Completeness	100%
Crystal system	Orthorhombic	Absorption correction	numerical
Space group	P2 ₁ 2 ₁ 2 ₁	Max and min transmission	0.9460 and 0.6630
<i>a</i> [Å]	5.7158(8)	Solution method	direct methods
<i>b</i> [Å]	16.615(2)	Refinement method	Full-matrix least-squares on F ²
<i>c</i> [Å]	19.036(2)	Data/restraints/parameters	3302 / 0 / 258
α [°]	90	Goodness-of-fit on F²	1.017
β [°]	90	Final R indices [I>2σ(I)]	2566 data; R ₁ = 0.0427, wR ₂ = 0.1027
γ [°]	90	R indices (all data)	R ₁ = 0.0583, wR ₂ = 0.1132
<i>V</i> [Å³]	1807.8(4)	Absolute structure parameter	0.3(4)
<i>Z</i>	4	Largest diff peak and hole [eÅ⁻³]	0.119 and -0.133
T [K]	296(2)	Extinction coefficient	0.0022(5)
<i>D</i>_{calc.} [g/cm³]	1.276		
μ [mm⁻¹]	0.803		
F(000)	736		