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Supporting Information

Selective functionalization of 6-amino-6-methyl-1,4perhydrodiazepine for the synthesis of a library of polydentate chelators

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1. HPLC methods and chromatograms

L1

Semi-preparative HPLC gradient conditions:

Solvent A: H₂O TFA 0.1%; Solvent B: MeOH TFA 0.1%; Flow: 20 ml/min; $t_R = 3.6$ min

Time (min)	Solvent A (%)	Solvent B (%)
0	60	40
2,00	60	40
7,00	39	61
8,00	0	100
11,00	0	100

Analytical HPLC gradient conditions:

Solvent A: H₂O TFA 0.1%; Solvent B: MeOH TFA 0.1%; Flow: 1 ml/min; $t_R = 14.2 \text{ min}$

Time (min)	Solvent A (%)	Solvent B (%)
0	70	30
2,00	70	30
16,00	0	100
19,00	0	100



Figure S1. ESI⁺ MS (*bottom*) and UV (254 nm, *middle*) HPLC chromatograms of ligand L1.

Solvent A: H₂O TFA 0.1%; Solvent B: MeOH TFA 0.1%; Flow: 1 ml/min; $t_R = 10.1$ min

Time (min)	Solvent A (%)	Solvent B (%)
0	90	10
2,00	90	10
16,00	0	100
19,00	0	100



Figure S2. ESI⁺ MS (*bottom*) and UV (254 nm, *top*) HPLC chromatograms of ligand L2.

Solvent A: H₂O TFA 0.1%; Solvent B: MeOH TFA 0.1%; Flow: 1 ml/min; $t_R = 12.3$ min

Time (min)	Solvent A (%)	Solvent B (%)
0	90	10
2,00	90	10
16,00	0	100
19,00	0	100



Figure S3. ESI⁺ MS (*bottom*) and UV (254 nm, *middle*) HPLC chromatograms of ligand L3.

Analytical HPLC gradient conditions:

Solvent A: H₂O TFA 0.1%; Solvent B: MeOH TFA 0.1%; Flow: 1 ml/min; $t_R = 2.0$ min

Time (min)	Solvent A (%)	Solvent B (%)
0	99	1
2,00	99	1
10,00	0	100
13,00	0	100



Figure S4. ESI⁺ MS (*bottom*) and UV (254 nm, *top*) HPLC chromatograms of ligand L4.

Semi-preparative HPLC gradient conditions:

Solvent A: H₂O TFA 0.1%; Solvent B: MeOH TFA 0.1%; Flow: 20 ml/min; $t_R = 6.0$ min

Time (min)	Solvent A (%)	Solvent B (%)
0	95	5
2,00	95	5
10,00	47	53
11,00	0	100
15,00	0	100

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Solvent A: H<sub>2</sub>O TFA 0.1%; Solvent B: MeOH TFA 0.1%; Flow: 1 ml/min; t_R = 10.6 min
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Time (min)	Solvent A (%)	Solvent B (%)
0	95	5
2,00	95	5
16,00	0	100
19,00	0	100



Figure S5. ESI⁺ MS (*bottom*) and UV (254 nm, *middle*) HPLC chromatograms of ligand L5.

Semi-preparative HPLC gradient conditions:

Solvent A: H₂O TFA 0.1%; Solvent B: MeOH TFA 0.1%; Flow: 20 ml/min; $t_R = 6.1$ min

Time (min)	Solvent A (%)	Solvent B (%)
0	95	5
2,00	95	5
9,00	53	47
10,00	0	100
13,00	0	100

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Solvent A: H<sub>2</sub>O TFA 0.1%; Solvent B: MeOH TFA 0.1%; Flow: 1 ml/min; t_R = 12.2 min
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Time (min)	Solvent A (%)	Solvent B (%)
0	99	0
2,00	99	0
16,00	0	100
19,00	0	100



Figure S6. ESI⁺ MS (*bottom*) HPLC chromatogram of ligand L6.

2. NMR spectra



Figure S7. ¹H NMR spectrum in CDCl₃ of intermediate 1 (resonances from residual solvents are also observable).



Figure S8. ¹³C NMR spectrum in CDCl₃ of intermediate 1.



Figure S9. ¹H NMR spectrum in D₂O of ligand L1.



Figure S10. ¹³C NMR spectrum in D₂O of ligand L1 (TFA as counterion is also observable).



Figure S11. ¹H NMR spectrum in CDCl₃ of intermediate 2.



Figure S12. ¹³C NMR spectrum in CDCl₃ of intermediate 2.



Figure S13. ¹H NMR spectrum in D₂O of ligand L2.



Figure S14. ¹³C NMR spectrum in D₂O of ligand L2 (TFA as counterion is also slightly observable).



Figure S15. ¹H NMR spectrum in CD₃OD of ligand L3.



Figure S16. ¹³C NMR spectrum in CD₃OD of ligand L3.



Figure S17. ¹H NMR spectrum in CDCl₃ of intermediate 4.





Figure S19. ¹H NMR spectrum in CD₃OD of ligand L4 (residual Et₂O from precipitation is also observable).



Figure S20. ¹³C NMR spectrum in CD₃OD of ligand L4 (residual Et₂O from precipitation is also observable).



Figure S21. ¹H NMR spectrum in CDCl₃ of intermediate 5.



Figure S22. ¹³C NMR spectrum in CDCl₃ of intermediate 5.



Figure S23. ¹H NMR spectrum in CD₃CN of ligand L5.



Figure S24. ¹³C NMR spectrum in CD₃CN of ligand L5.



Figure S25. ¹H NMR spectrum in CDCl₃ of intermediate 7.





Figure S27. ¹H NMR spectrum in CDCl₃ of intermediate 8.





Figure S29. ¹H NMR spectrum in CDCl₃ of intermediate 9.



Figure S30. ¹³C NMR spectrum in CDCl₃ of intermediate 9.



Figure S31. ¹H NMR spectrum in D₂O of ligand L6.



Figure S32. ¹³C NMR spectrum in D₂O of ligand L6.



Figure S33. ¹H NMR spectrum in CDCl₃ of intermediate 10.

