

Supplementary Information

Ru(II)-catalyzed C–H addition and oxidative cyclization of 2-aryl quinazolinones with activated aldehydes

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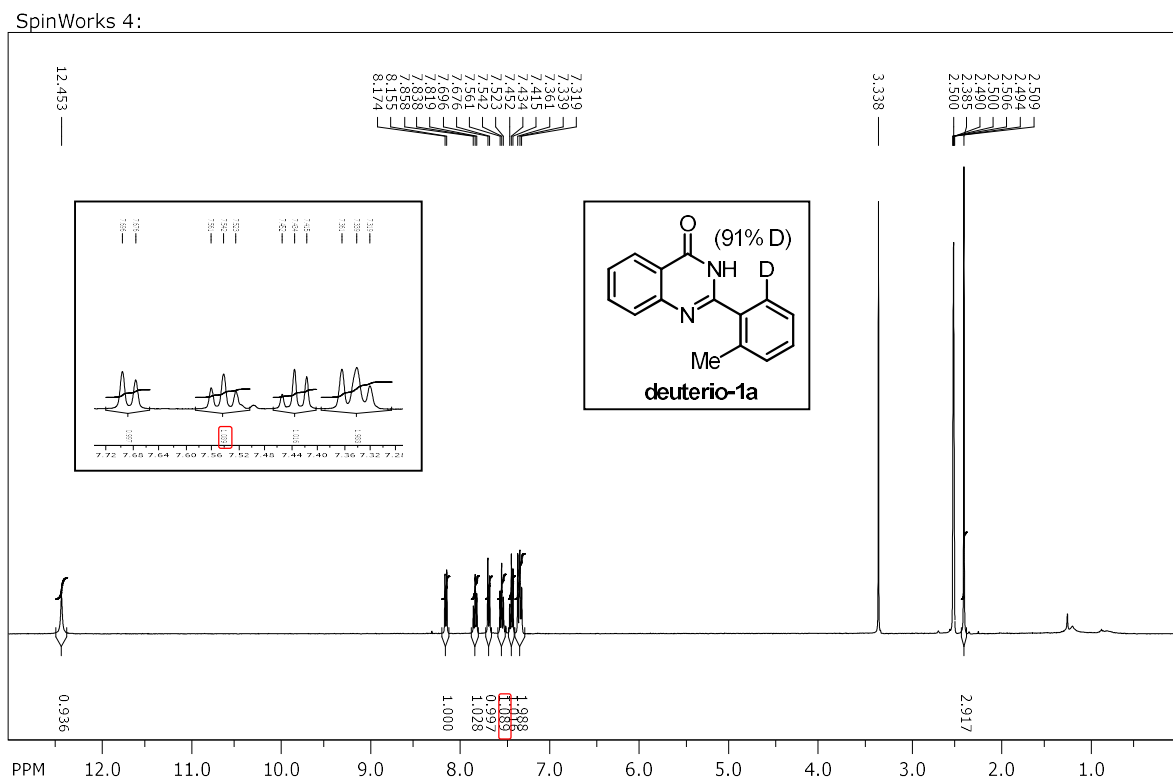
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Experimental procedure and characterization data for the synthesis of deuterio-1a

To an oven-dried sealed tube charged with **1a** (47.3 mg, 0.2 mmol, 100 mol %), [Ru(*p*-cymene)Cl₂]₂ (3.1 mg, 0.005 mmol, 2.5 mol %), NaOAc (8.2 mg, 0.1 mmol, 50 mol %) and DCE (1 mL) was added CD₃CO₂D (0.11 mL, 2.0 mmol, 10 equiv.) under air at room temperature. The reaction mixture was stirred at 80 °C for 20 h. The reaction mixture was diluted with EtOAc (3 mL) and concentrated in vacuo. The residue was purified by flash column chromatography (*n*-hexanes/EtOAc = 3:1) to afford **deuterio-1a** (37.8 mg, 80%) with 91% deuterium incorporation at *ortho*-position of aryl ring.

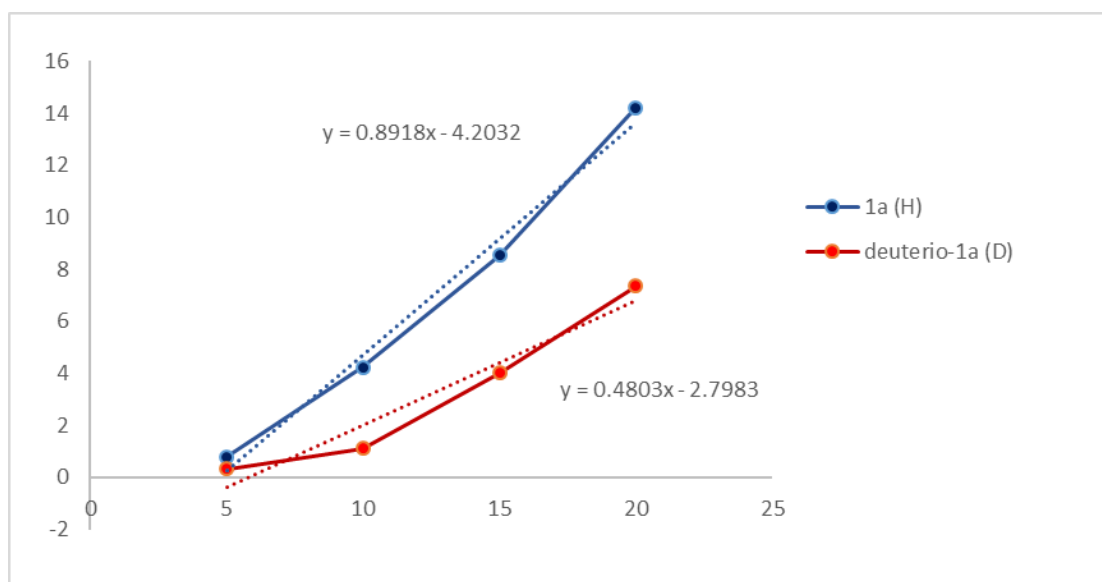
2-(2-Deuterio-6-methylphenyl)quinazolin-4(3H)-one (deuterio-1a): ¹H NMR (400 MHz, CDCl₃) δ 12.45 (s, 1H), 8.16 (d, *J* = 7.6 Hz, 1H), 7.83 (t, *J* = 8.0 Hz, 1H), 7.68 (d, *J* = 8.0 Hz, 1H), 7.54 (t, *J* = 7.6 Hz, 1.09H), 7.43 (t, *J* = 7.2 Hz, 1H), 7.36–7.31 (m, 2H), 3.38 (s, 3H).



Experimental procedure of kinetic isotope effect (KIE) experiment

To an oven-dried sealed tube charged with **1a** (47.3 mg, 0.2 mmol, 100 mol %), [Ru(*p*-cymene)Cl₂]₂ (3.1 mg, 0.005 mmol, 2.5 mol %), NaOAc (8.2 mg, 0.1 mmol, 50 mol %) and ethyl 2-oxoacetate (~50% in toluene) (**2a**) (81.7 mg, 0.4 mmol, 200 mol %) in DCE (1 mL) was added bromobenzene (78.5 mg, 0.5 mmol, 250 mol %) as an internal standard. In another reaction tube, **deuterio-1a** (47.5 mg, 0.2 mmol, 100 mol %, 91% D) was used instead of **1a**. The two reactions were allowed to stir at 80 °C. An aliquot of each reaction mixture was taken at the time of 5 min, 10 min, 15 min, and 20 min. The corresponding yield of each product **3a** was determined by GC-MS (bromobenzene as an internal standard). A kinetic isotope effect value (k_H/k_D) of 1.85 was observed.

starting material	relative yield (%) of 3a based on bromobenzene			
	5 min	10 min	15 min	20 min
1a (H)	0.78618551	4.23468618	8.54236322	14.21351523
deuterio-1a (D)	0.33251588	1.10362542	4.02012145	7.36522214



Crystal structure determination

The X-ray diffraction data for **4a** were collected on a Bruker APX-II diffractometer equipped with a monochromator in the Mo K α ($\lambda = 0.71073$ Å) incident beam. Each crystal was mounted on a glass fibre. The CCDC data were integrated and scaled using the Bruker-S SAINT software package, and the structure was solved and refined using SHELXL-2014. All hydrogen atoms were placed in the calculated positions. The crystallographic data of **4a** are listed in Table S1-Table S6. Structural information was deposited at the Cambridge Crystallographic Data Centre. The CCDC reference numbers are 2013547 for **4a**.

X-ray crystallographic data of compound 4a (CCDC 2013547)

Table S1. Crystal data and structure refinement for 4a.

Identification code	20200612_0m	
Empirical formula	C ₁₉ H ₁₆ N ₂ O ₄	
Formula weight	336.34	
Temperature	296(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	P n a 21	
Unit cell dimensions	a = 7.67620(10) Å	a = 90°.
	b = 12.3159(2) Å	b = 90°.
	c = 17.1409(3) Å	g = 90°.
Volume	1620.49(4) Å ³	
Z	4	
Density (calculated)	1.379 Mg/m ³	
Absorption coefficient	0.098 mm ⁻¹	
F(000)	704	
Crystal size	0.280 x 0.160 x 0.040 mm ³	
Theta range for data collection	2.036 to 28.326°.	
Index ranges	0 ≤ h ≤ 10, 0 ≤ k ≤ 16, 0 ≤ l ≤ 22	
Reflections collected	2082	
Independent reflections	2082 [R(int) = 0.0000]	
Completeness to theta = 25.242°	100.0 %	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	2082 / 2 / 226	
Goodness-of-fit on F ²	1.004	
Final R indices [I > 2σ(I)]	R1 = 0.0549, wR2 = 0.1498	
R indices (all data)	R1 = 0.0694, wR2 = 0.1613	
Absolute structure parameter	1.4(18)	
Extinction coefficient	n/a	
Largest diff. peak and hole	0.330 and -0.213 e.Å ⁻³	

Table S2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4a**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
O(1)	2469(4)	3983(2)	4714(2)	66(1)
O(2)	2756(3)	4108(2)	2954(1)	54(1)
O(3)	-288(4)	3073(2)	3243(2)	73(1)
O(4)	-1312(3)	4518(2)	3889(2)	71(1)
N(1)	1895(3)	5403(2)	3925(1)	43(1)
N(2)	2038(4)	7257(2)	4296(2)	49(1)
C(1)	2418(4)	4974(3)	4628(2)	49(1)
C(2)	1378(4)	4724(2)	3246(2)	44(1)
C(3)	830(4)	5600(3)	2676(2)	46(1)
C(4)	226(5)	5451(3)	1921(2)	56(1)
C(5)	-231(5)	6385(3)	1502(2)	65(1)
C(6)	-49(5)	7406(3)	1836(2)	60(1)
C(7)	586(4)	7571(3)	2583(2)	51(1)
C(8)	1013(4)	6626(2)	2999(2)	44(1)
C(9)	1697(4)	6513(2)	3795(2)	44(1)
C(10)	2666(4)	6897(3)	5018(2)	51(1)
C(11)	3071(6)	7676(3)	5585(2)	69(1)
C(12)	3675(6)	7357(4)	6312(3)	79(1)
C(13)	3908(6)	6261(4)	6480(2)	73(1)
C(14)	3507(5)	5491(3)	5939(2)	64(1)
C(15)	2877(4)	5797(3)	5202(2)	50(1)
C(16)	-164(5)	3990(2)	3466(2)	50(1)
C(17)	-2920(7)	3955(5)	4101(3)	96(2)
C(18)	-3003(11)	3798(8)	4940(4)	136(3)
C(19)	827(6)	8680(3)	2909(3)	64(1)

Table S3. Bond lengths [Å] and angles [°] for **4a**.

O(1)-C(1)	1.230(4)
O(2)-C(2)	1.395(4)
O(2)-H(2A)	0.8200
O(3)-C(16)	1.196(4)
O(4)-C(16)	1.314(4)
O(4)-C(17)	1.462(5)
N(1)-C(1)	1.376(4)
N(1)-C(9)	1.393(4)
N(1)-C(2)	1.487(4)
N(2)-C(9)	1.282(4)
N(2)-C(10)	1.400(5)
C(1)-C(15)	1.456(5)
C(2)-C(3)	1.515(4)
C(2)-C(16)	1.537(4)
C(3)-C(8)	1.386(5)
C(3)-C(4)	1.387(4)
C(4)-C(5)	1.400(5)
C(4)-H(4A)	0.9300
C(5)-C(6)	1.388(6)
C(5)-H(5A)	0.9300
C(6)-C(7)	1.385(5)
C(6)-H(6A)	0.9300
C(7)-C(8)	1.404(4)
C(7)-C(19)	1.487(5)
C(8)-C(9)	1.470(5)
C(10)-C(15)	1.400(5)
C(10)-C(11)	1.400(5)
C(11)-C(12)	1.388(6)
C(11)-H(11A)	0.9300
C(12)-C(13)	1.392(6)
C(12)-H(12A)	0.9300
C(13)-C(14)	1.362(6)
C(13)-H(13A)	0.9300
C(14)-C(15)	1.405(5)

C(14)-H(14A)	0.9300
C(17)-C(18)	1.451(7)
C(17)-H(17A)	0.9700
C(17)-H(17B)	0.9700
C(18)-H(18A)	0.9600
C(18)-H(18B)	0.9600
C(18)-H(18C)	0.9600
C(19)-H(19A)	0.9600
C(19)-H(19B)	0.9600
C(19)-H(19C)	0.9600

C(2)-O(2)-H(2A)	109.5
C(16)-O(4)-C(17)	117.9(3)
C(1)-N(1)-C(9)	123.3(3)
C(1)-N(1)-C(2)	123.2(2)
C(9)-N(1)-C(2)	113.5(2)
C(9)-N(2)-C(10)	115.8(3)
O(1)-C(1)-N(1)	119.7(3)
O(1)-C(1)-C(15)	127.0(3)
N(1)-C(1)-C(15)	113.3(3)
O(2)-C(2)-N(1)	112.6(2)
O(2)-C(2)-C(3)	111.5(2)
N(1)-C(2)-C(3)	100.3(2)
O(2)-C(2)-C(16)	110.6(2)
N(1)-C(2)-C(16)	110.1(2)
C(3)-C(2)-C(16)	111.3(2)
C(8)-C(3)-C(4)	121.8(3)
C(8)-C(3)-C(2)	111.3(3)
C(4)-C(3)-C(2)	126.9(3)
C(3)-C(4)-C(5)	116.9(3)
C(3)-C(4)-H(4A)	121.5
C(5)-C(4)-H(4A)	121.5
C(6)-C(5)-C(4)	120.5(3)
C(6)-C(5)-H(5A)	119.7
C(4)-C(5)-H(5A)	119.7
C(7)-C(6)-C(5)	123.3(3)

C(7)-C(6)-H(6A)	118.3
C(5)-C(6)-H(6A)	118.3
C(6)-C(7)-C(8)	115.5(3)
C(6)-C(7)-C(19)	121.8(3)
C(8)-C(7)-C(19)	122.8(3)
C(3)-C(8)-C(7)	121.9(3)
C(3)-C(8)-C(9)	108.7(2)
C(7)-C(8)-C(9)	129.3(3)
N(2)-C(9)-N(1)	124.9(3)
N(2)-C(9)-C(8)	128.8(3)
N(1)-C(9)-C(8)	106.2(2)
C(15)-C(10)-N(2)	123.1(3)
C(15)-C(10)-C(11)	118.8(3)
N(2)-C(10)-C(11)	118.2(3)
C(12)-C(11)-C(10)	120.2(4)
C(12)-C(11)-H(11A)	119.9
C(10)-C(11)-H(11A)	119.9
C(11)-C(12)-C(13)	120.2(3)
C(11)-C(12)-H(12A)	119.9
C(13)-C(12)-H(12A)	119.9
C(14)-C(13)-C(12)	120.3(4)
C(14)-C(13)-H(13A)	119.8
C(12)-C(13)-H(13A)	119.8
C(13)-C(14)-C(15)	120.3(4)
C(13)-C(14)-H(14A)	119.9
C(15)-C(14)-H(14A)	119.9
C(10)-C(15)-C(14)	120.1(3)
C(10)-C(15)-C(1)	119.6(3)
C(14)-C(15)-C(1)	120.3(3)
O(3)-C(16)-O(4)	126.2(3)
O(3)-C(16)-C(2)	122.6(3)
O(4)-C(16)-C(2)	111.1(3)
C(18)-C(17)-O(4)	110.3(5)
C(18)-C(17)-H(17A)	109.6
O(4)-C(17)-H(17A)	109.6
C(18)-C(17)-H(17B)	109.6

O(4)-C(17)-H(17B)	109.6
H(17A)-C(17)-H(17B)	108.1
C(17)-C(18)-H(18A)	109.5
C(17)-C(18)-H(18B)	109.5
H(18A)-C(18)-H(18B)	109.5
C(17)-C(18)-H(18C)	109.5
H(18A)-C(18)-H(18C)	109.5
H(18B)-C(18)-H(18C)	109.5
C(7)-C(19)-H(19A)	109.5
C(7)-C(19)-H(19B)	109.5
H(19A)-C(19)-H(19B)	109.5
C(7)-C(19)-H(19C)	109.5
H(19A)-C(19)-H(19C)	109.5
H(19B)-C(19)-H(19C)	109.5

Symmetry transformations used to generate equivalent atoms:

Table S4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4a**. The anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2 a^{*2}U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
O(1)	96(2)	47(1)	56(1)	6(1)	-14(1)	-5(1)
O(2)	57(1)	50(1)	55(1)	-4(1)	5(1)	2(1)
O(3)	86(2)	50(1)	82(2)	-6(1)	-1(2)	-20(1)
O(4)	65(2)	66(2)	82(2)	-11(1)	20(1)	-20(1)
N(1)	50(1)	38(1)	41(1)	0(1)	0(1)	-5(1)
N(2)	58(1)	39(1)	51(1)	-2(1)	-1(1)	0(1)
C(1)	59(2)	45(2)	42(2)	7(1)	0(1)	-2(1)
C(2)	51(2)	42(1)	40(1)	1(1)	-1(1)	-1(1)
C(3)	45(1)	49(2)	43(1)	3(1)	1(1)	-4(1)
C(4)	63(2)	59(2)	48(2)	-2(2)	-3(2)	-5(2)
C(5)	67(2)	80(2)	47(2)	10(2)	-7(2)	-3(2)
C(6)	56(2)	64(2)	61(2)	23(2)	-5(2)	-3(2)
C(7)	42(1)	49(2)	62(2)	14(1)	7(1)	0(1)
C(8)	42(1)	45(1)	47(2)	5(1)	4(1)	-3(1)
C(9)	41(1)	40(1)	49(2)	3(1)	7(1)	-2(1)
C(10)	55(2)	48(2)	50(2)	-6(1)	1(1)	-4(1)
C(11)	84(3)	60(2)	62(2)	-14(2)	-5(2)	-8(2)
C(12)	102(3)	77(3)	57(2)	-18(2)	-10(2)	-20(2)
C(13)	90(3)	78(3)	51(2)	1(2)	-13(2)	-13(2)
C(14)	77(2)	66(2)	50(2)	4(2)	-6(2)	-13(2)
C(15)	55(2)	50(2)	45(2)	-1(1)	0(1)	-7(1)
C(16)	58(2)	49(2)	43(2)	3(1)	-4(1)	-8(1)
C(17)	81(3)	117(4)	89(3)	0(3)	21(3)	-44(3)
C(18)	112(5)	205(9)	91(4)	7(5)	16(4)	-73(6)
C(19)	63(2)	48(2)	80(2)	12(2)	2(2)	2(2)

Table S5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4a**.

	x	y	z	U(eq)
H(2A)	3568	4510	2840	81
H(4A)	129	4762	1702	68
H(5A)	-659	6321	997	78
H(6A)	-370	8009	1543	72
H(11A)	2934	8410	5473	82
H(12A)	3925	7877	6689	94
H(13A)	4340	6053	6965	88
H(14A)	3652	4759	6058	77
H(17A)	-3916	4378	3930	115
H(17B)	-2963	3256	3841	115
H(18A)	-4059	3423	5071	204
H(18B)	-2985	4492	5195	204
H(18C)	-2020	3377	5107	204
H(19A)	448	9208	2534	95
H(19B)	2036	8794	3026	95
H(19C)	151	8753	3377	95

Table S6. Torsion angles [°] for **4a**.

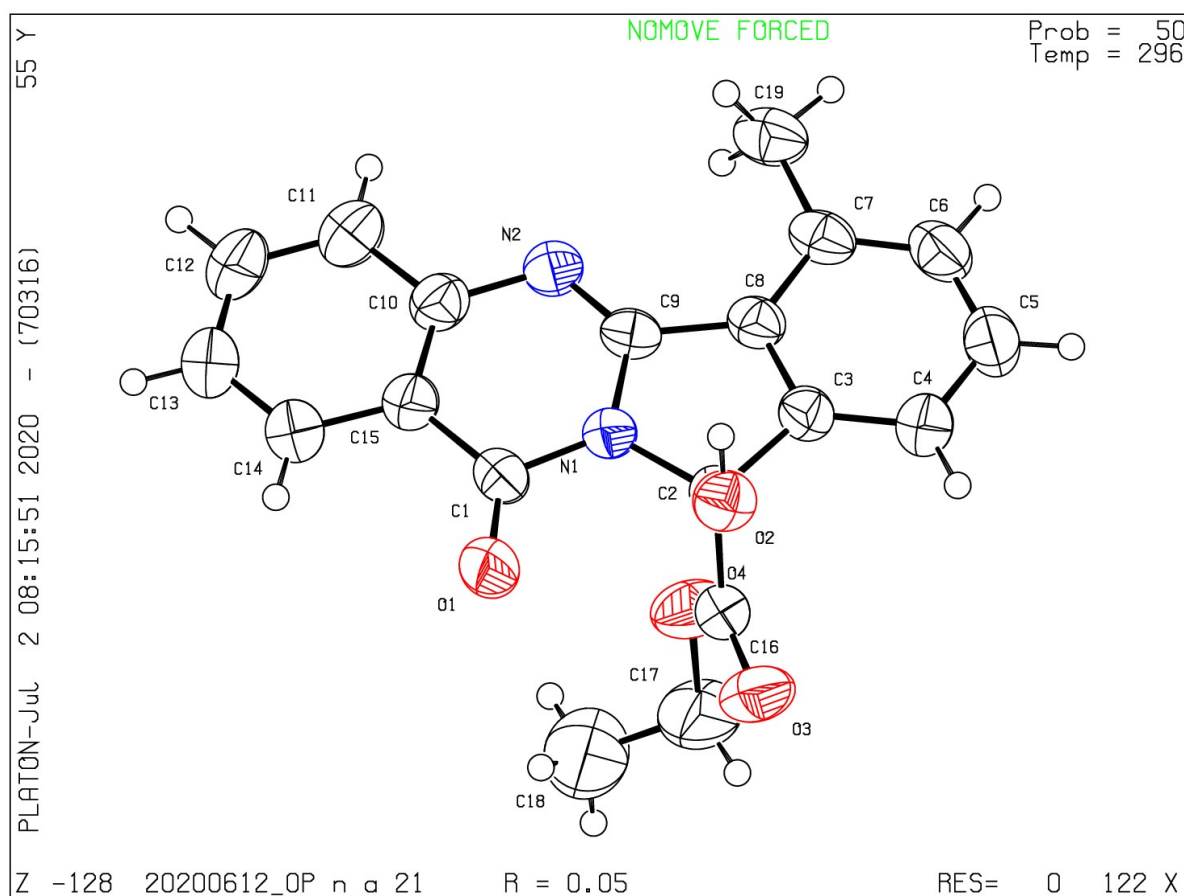
C(9)-N(1)-C(1)-O(1)	175.9(3)
C(2)-N(1)-C(1)-O(1)	-0.1(5)
C(9)-N(1)-C(1)-C(15)	-4.7(4)
C(2)-N(1)-C(1)-C(15)	179.4(3)
C(1)-N(1)-C(2)-O(2)	-65.0(4)
C(9)-N(1)-C(2)-O(2)	118.7(3)
C(1)-N(1)-C(2)-C(3)	176.3(3)
C(9)-N(1)-C(2)-C(3)	0.1(3)
C(1)-N(1)-C(2)-C(16)	58.9(4)
C(9)-N(1)-C(2)-C(16)	-117.3(3)
O(2)-C(2)-C(3)-C(8)	-119.6(3)
N(1)-C(2)-C(3)-C(8)	-0.2(3)
C(16)-C(2)-C(3)-C(8)	116.3(3)
O(2)-C(2)-C(3)-C(4)	59.9(4)
N(1)-C(2)-C(3)-C(4)	179.3(3)
C(16)-C(2)-C(3)-C(4)	-64.2(4)
C(8)-C(3)-C(4)-C(5)	-1.5(5)
C(2)-C(3)-C(4)-C(5)	179.1(3)
C(3)-C(4)-C(5)-C(6)	1.0(6)
C(4)-C(5)-C(6)-C(7)	0.4(6)
C(5)-C(6)-C(7)-C(8)	-1.3(5)
C(5)-C(6)-C(7)-C(19)	177.6(4)
C(4)-C(3)-C(8)-C(7)	0.5(5)
C(2)-C(3)-C(8)-C(7)	-179.9(3)
C(4)-C(3)-C(8)-C(9)	-179.2(3)
C(2)-C(3)-C(8)-C(9)	0.3(3)
C(6)-C(7)-C(8)-C(3)	0.8(4)
C(19)-C(7)-C(8)-C(3)	-178.0(3)
C(6)-C(7)-C(8)-C(9)	-179.5(3)
C(19)-C(7)-C(8)-C(9)	1.7(5)
C(10)-N(2)-C(9)-N(1)	0.5(4)
C(10)-N(2)-C(9)-C(8)	179.4(3)
C(1)-N(1)-C(9)-N(2)	2.9(5)
C(2)-N(1)-C(9)-N(2)	179.2(3)

C(1)-N(1)-C(9)-C(8)	-176.2(3)
C(2)-N(1)-C(9)-C(8)	0.1(3)
C(3)-C(8)-C(9)-N(2)	-179.3(3)
C(7)-C(8)-C(9)-N(2)	1.0(5)
C(3)-C(8)-C(9)-N(1)	-0.2(3)
C(7)-C(8)-C(9)-N(1)	-180.0(3)
C(9)-N(2)-C(10)-C(15)	-1.6(5)
C(9)-N(2)-C(10)-C(11)	179.6(3)
C(15)-C(10)-C(11)-C(12)	0.3(6)
N(2)-C(10)-C(11)-C(12)	179.1(4)
C(10)-C(11)-C(12)-C(13)	1.0(7)
C(11)-C(12)-C(13)-C(14)	-1.5(7)
C(12)-C(13)-C(14)-C(15)	0.9(7)
N(2)-C(10)-C(15)-C(14)	-179.7(3)
C(11)-C(10)-C(15)-C(14)	-0.9(5)
N(2)-C(10)-C(15)-C(1)	-0.5(5)
C(11)-C(10)-C(15)-C(1)	178.3(3)
C(13)-C(14)-C(15)-C(10)	0.4(6)
C(13)-C(14)-C(15)-C(1)	-178.9(4)
O(1)-C(1)-C(15)-C(10)	-177.1(3)
N(1)-C(1)-C(15)-C(10)	3.5(4)
O(1)-C(1)-C(15)-C(14)	2.1(5)
N(1)-C(1)-C(15)-C(14)	-177.3(3)
C(17)-O(4)-C(16)-O(3)	-1.3(6)
C(17)-O(4)-C(16)-C(2)	175.8(4)
O(2)-C(2)-C(16)-O(3)	-16.2(4)
N(1)-C(2)-C(16)-O(3)	-141.3(3)
C(3)-C(2)-C(16)-O(3)	108.5(4)
O(2)-C(2)-C(16)-O(4)	166.6(3)
N(1)-C(2)-C(16)-O(4)	41.5(4)
C(3)-C(2)-C(16)-O(4)	-68.7(4)
C(16)-O(4)-C(17)-C(18)	114.8(7)

Symmetry transformations used to generate equivalent atoms:

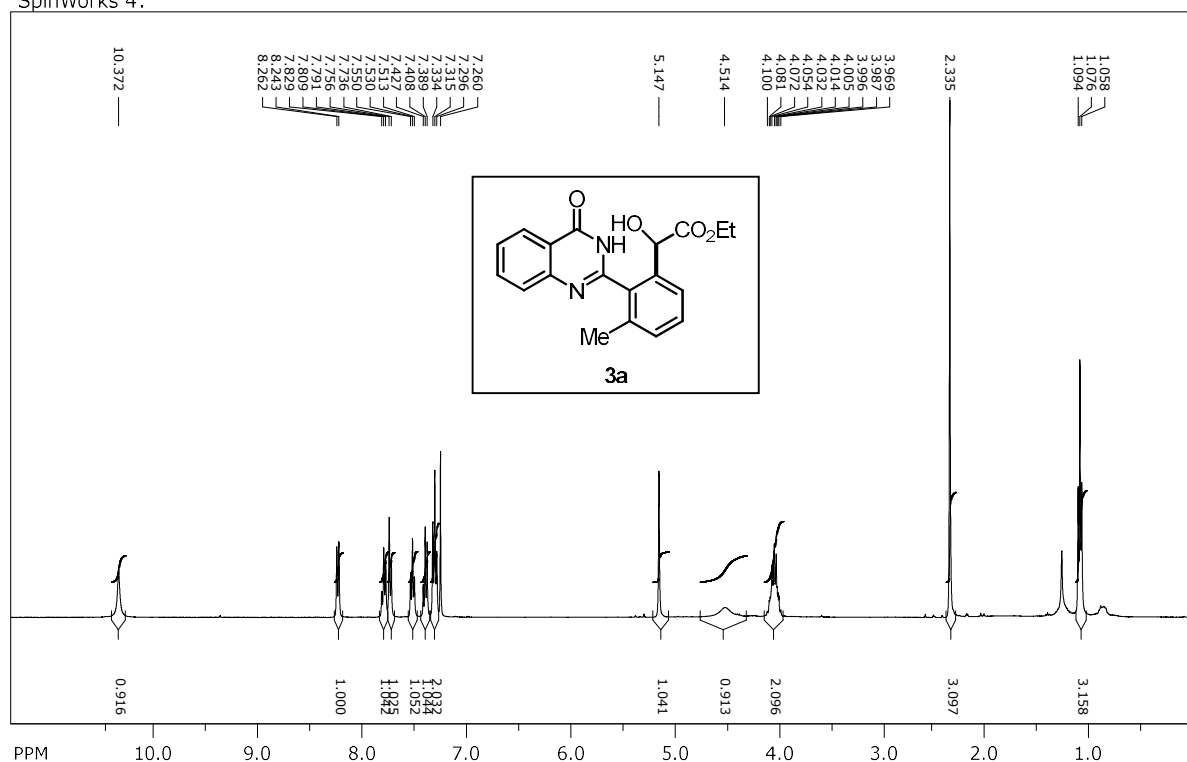
X-ray crystallographic structure of compound **4a** (CCDC 2013547)

Sample preparation (vapor diffusion): compound **4a** (5.1 mg) was dissolved with 1.5 mL of CHCl₃ in opened inner vessel, and pentane (6 mL) as an anti-solvent has been employed in closed outer vessel. After vapor diffusion for 10 days, the single crystals of compound **4a** were obtained.

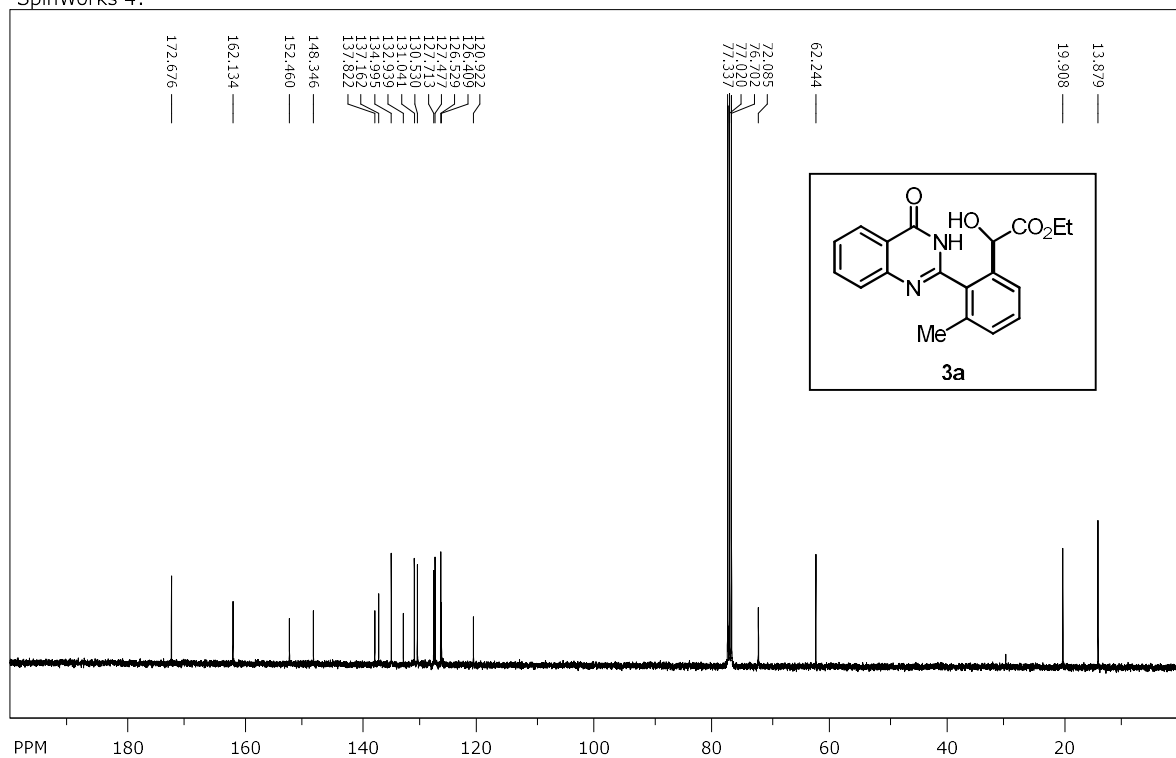


¹H NMR and ¹³C NMR spectra of all products

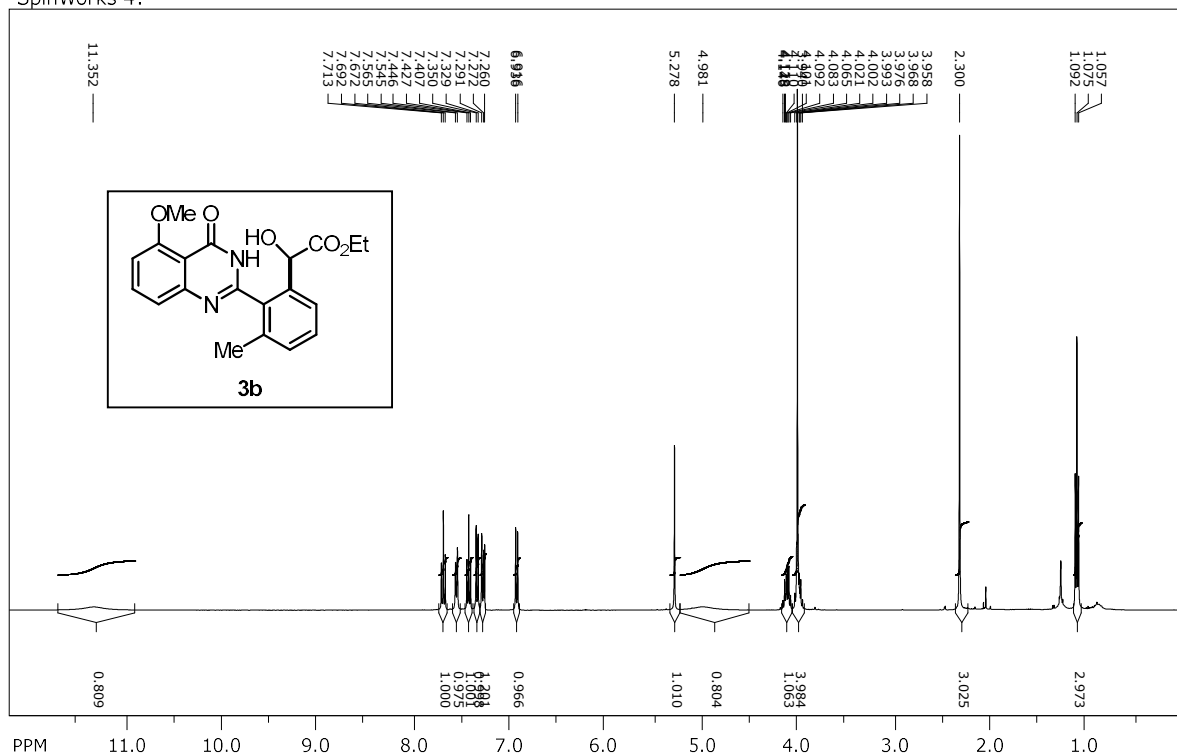
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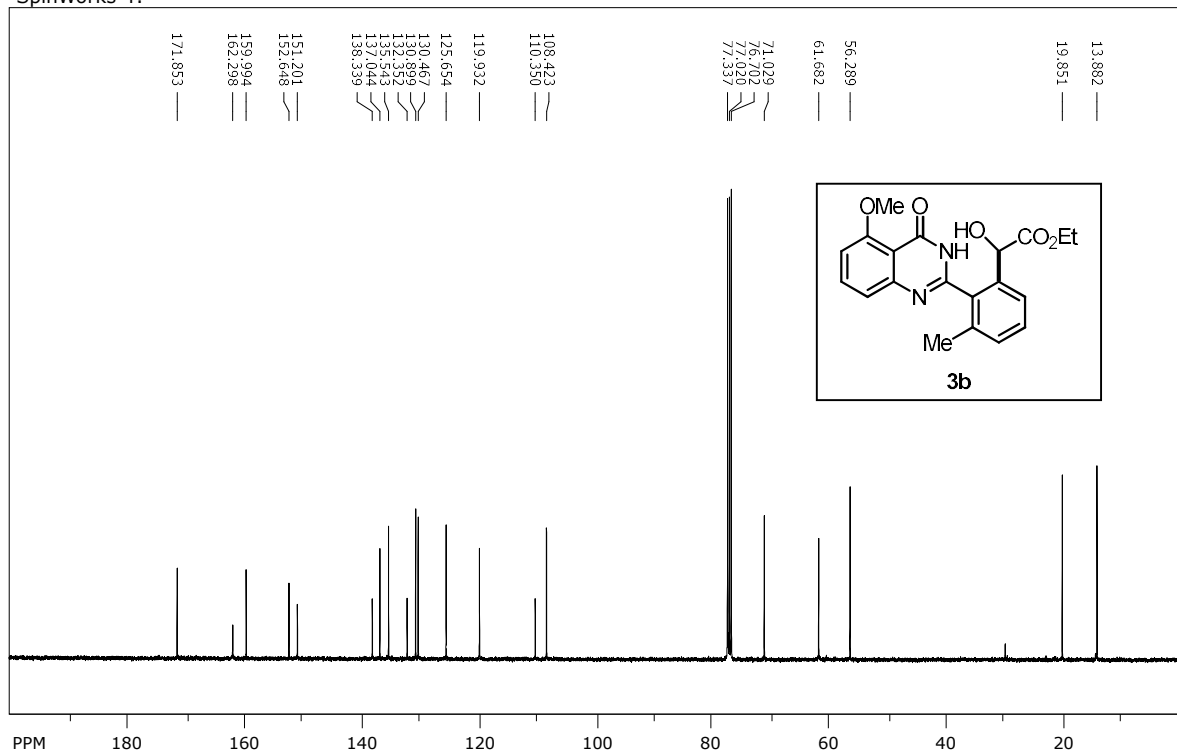
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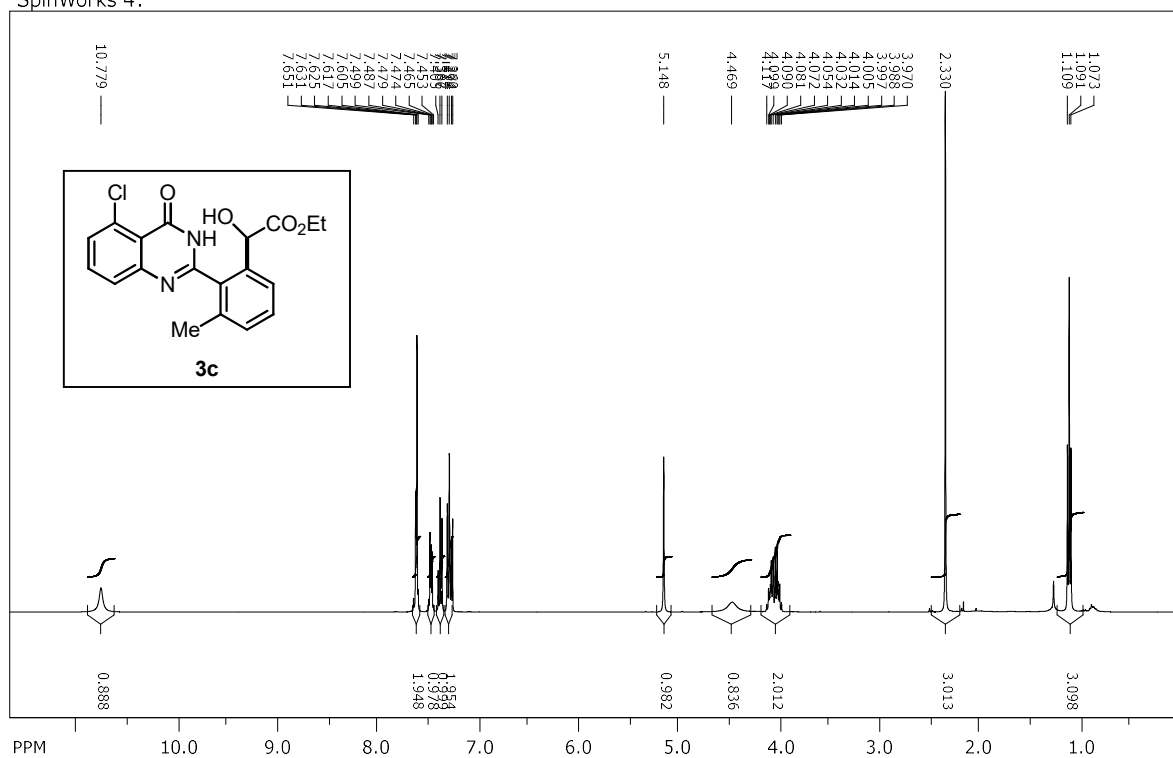
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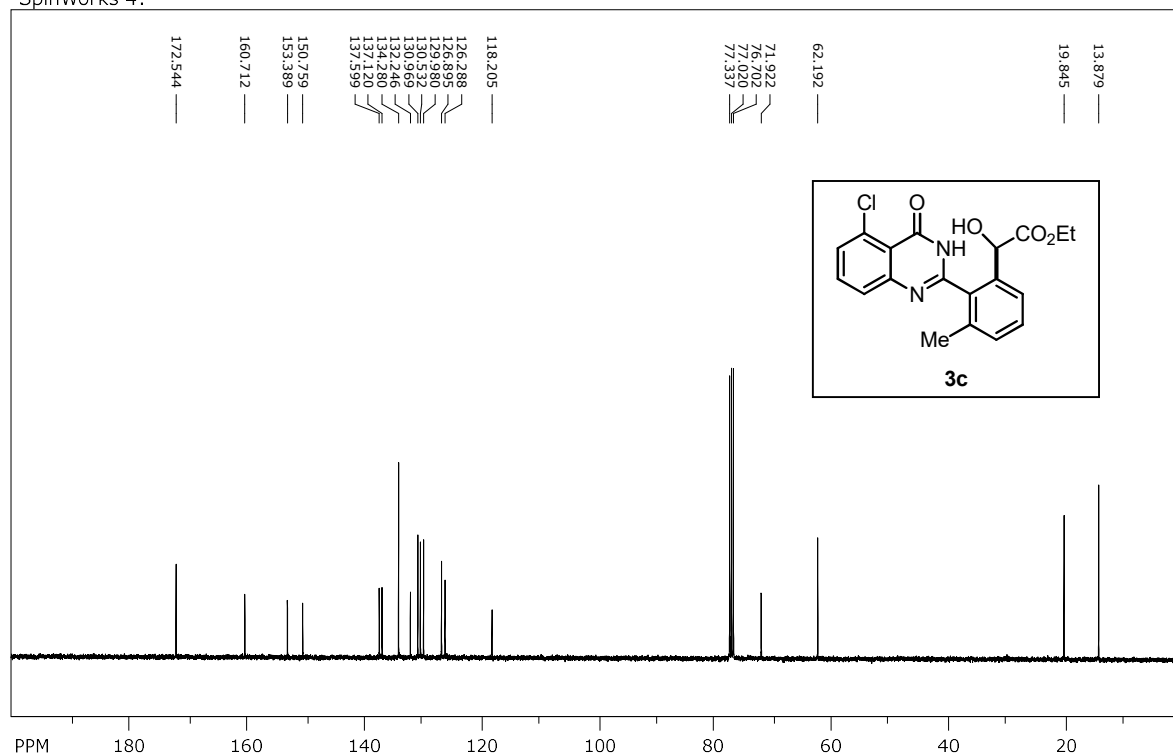
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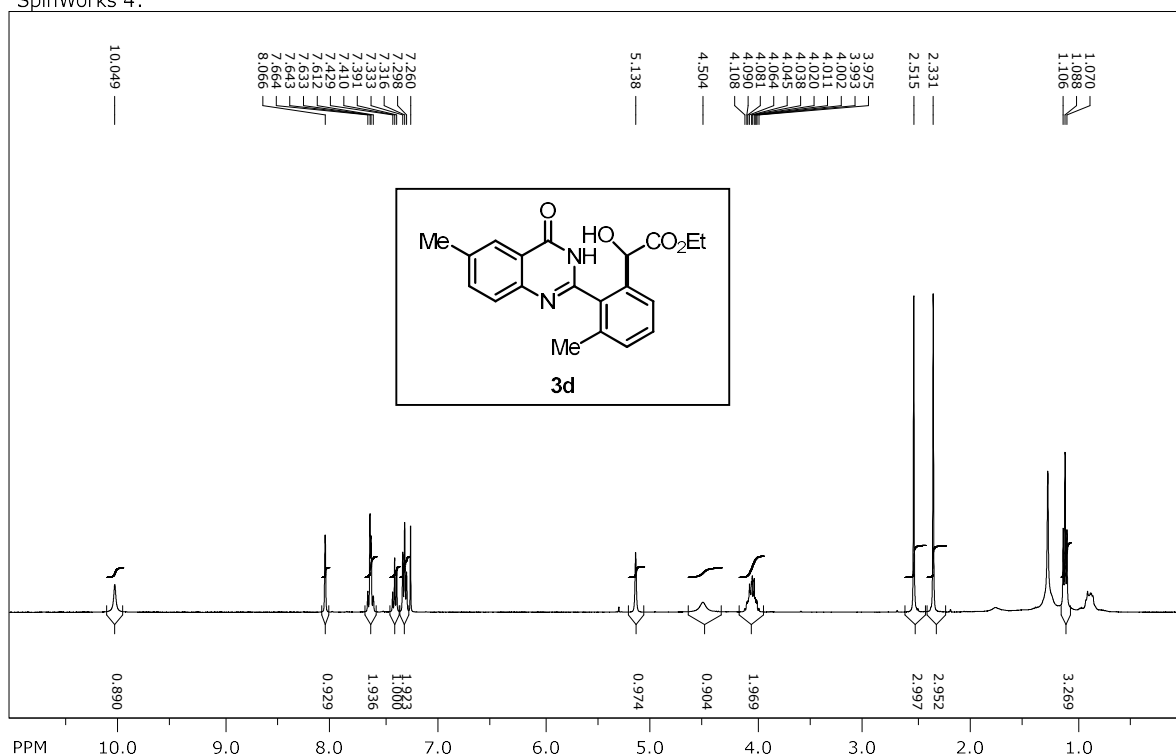
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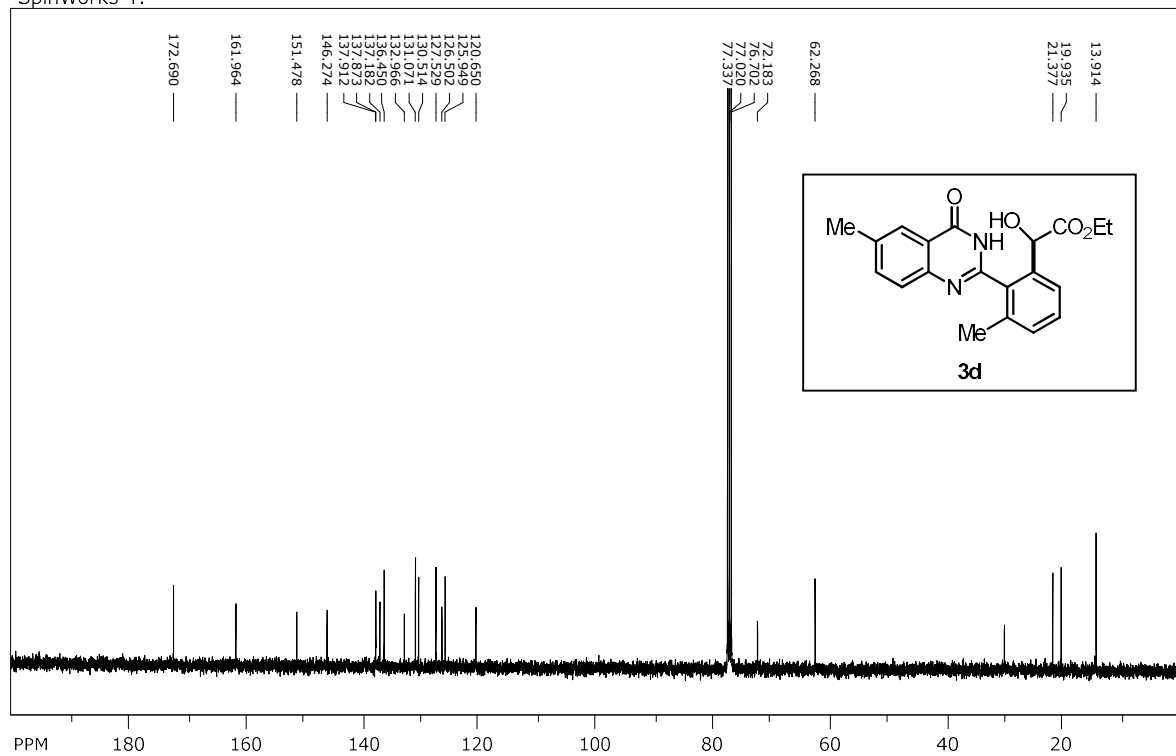
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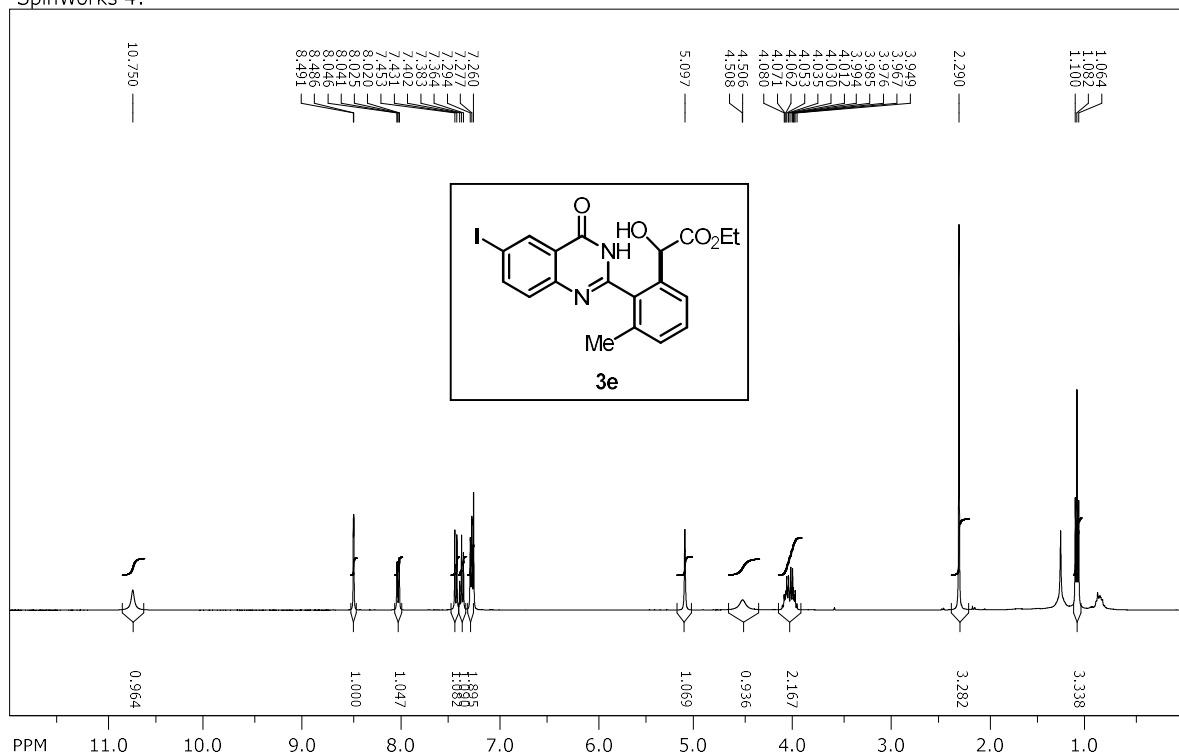
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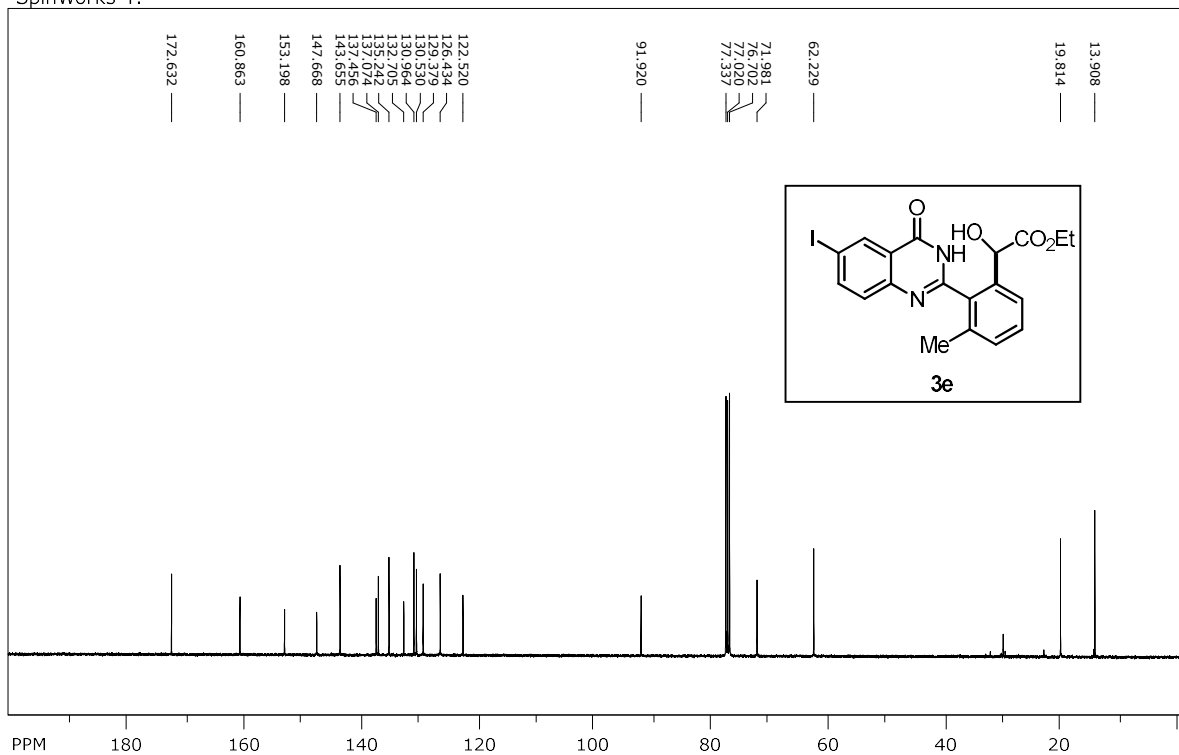
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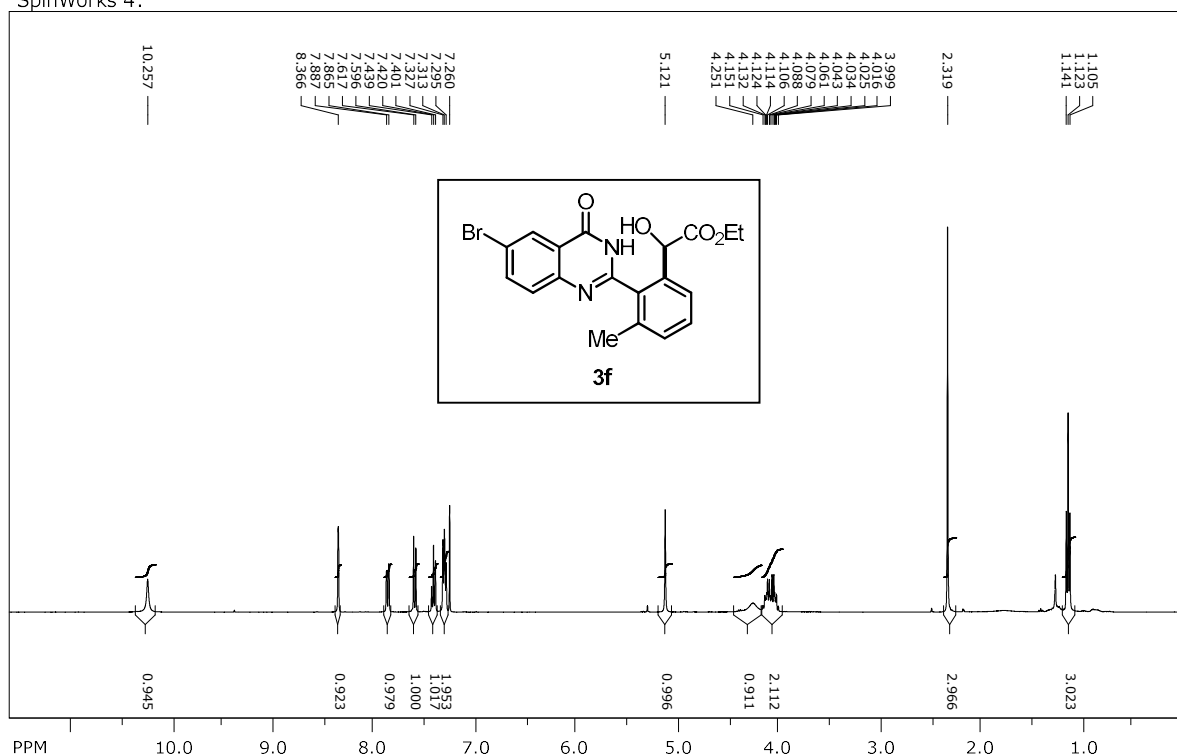
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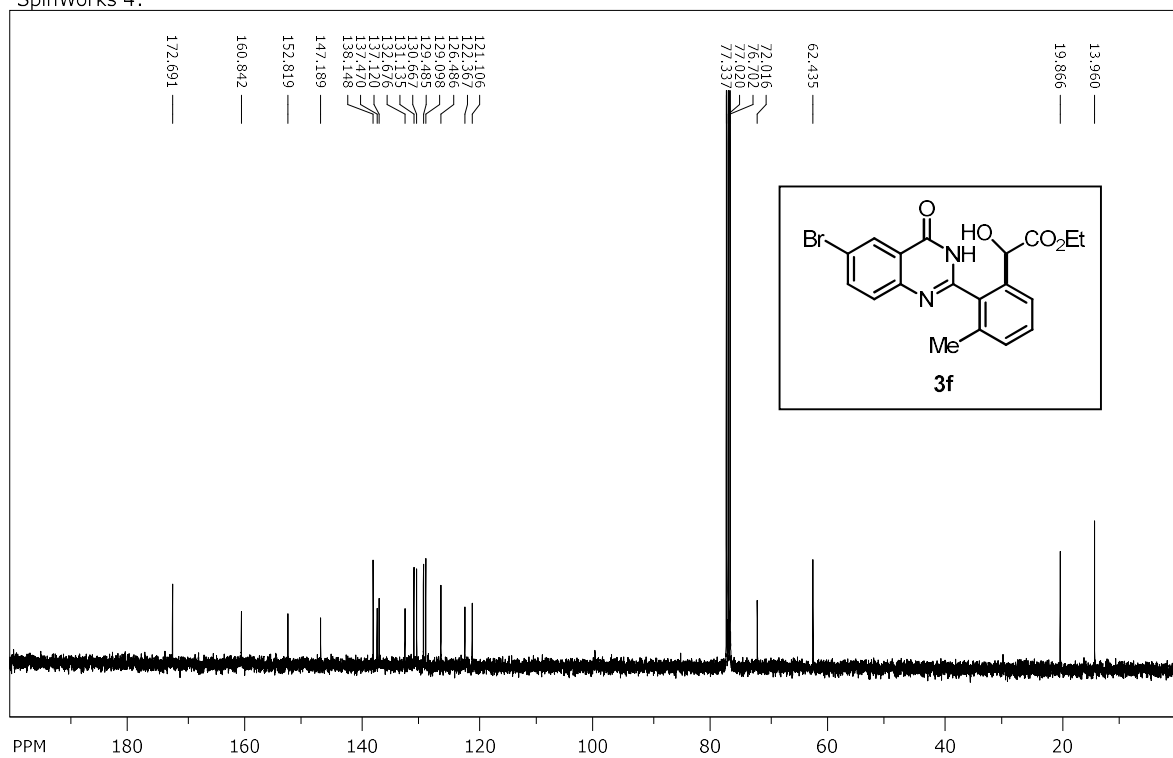
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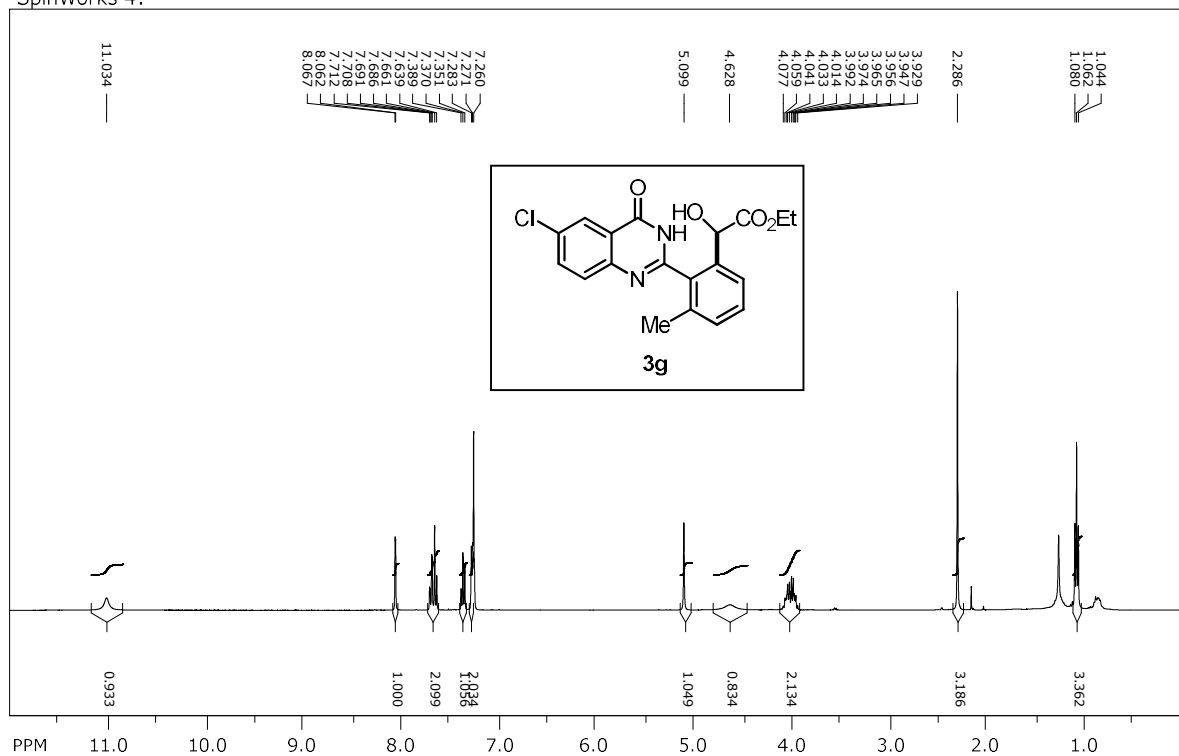
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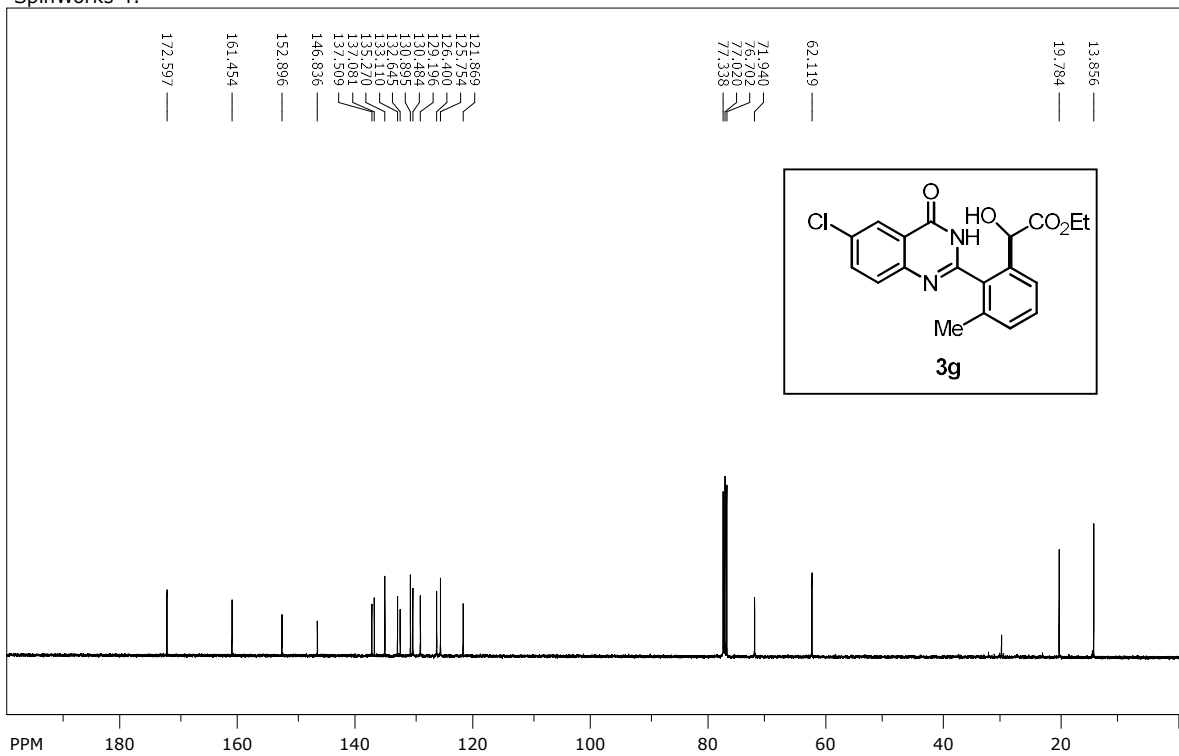
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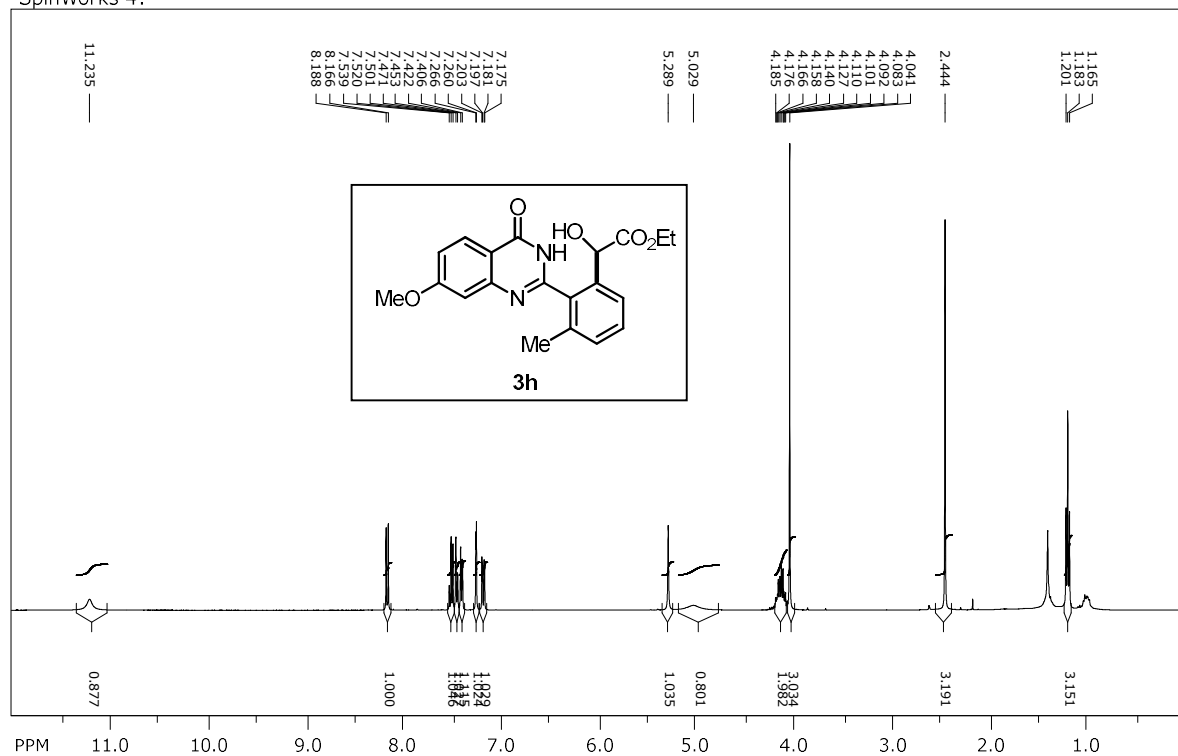
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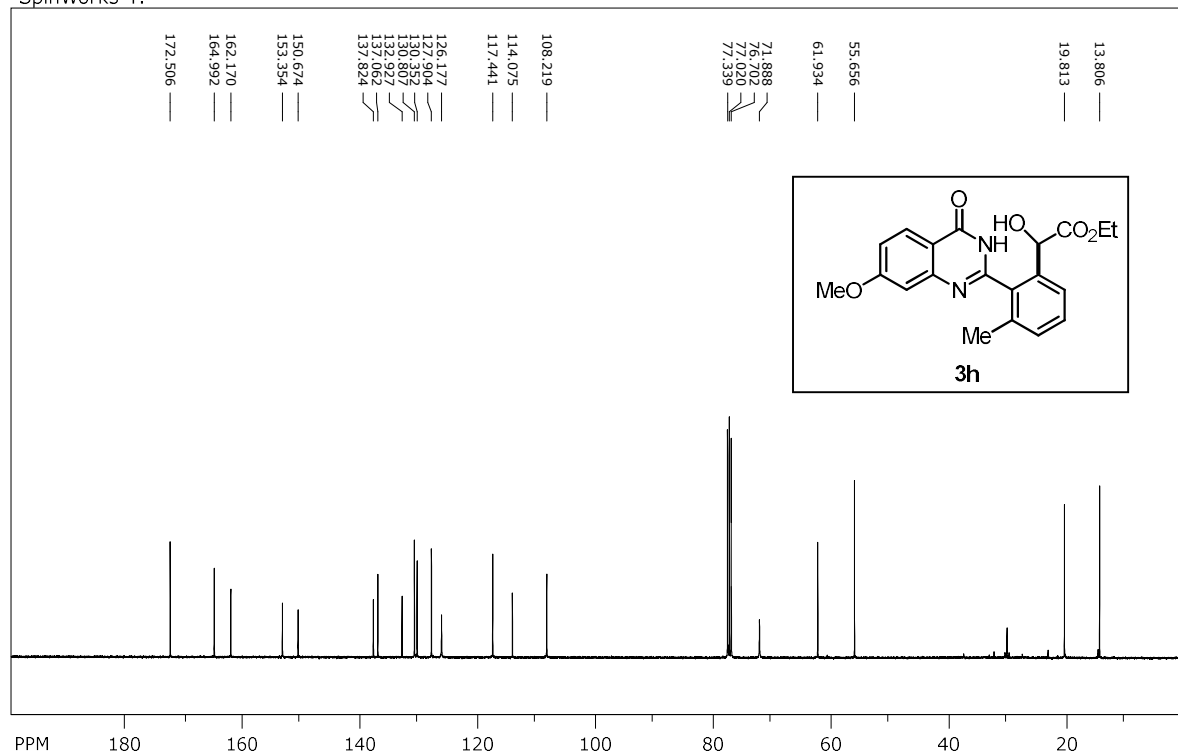
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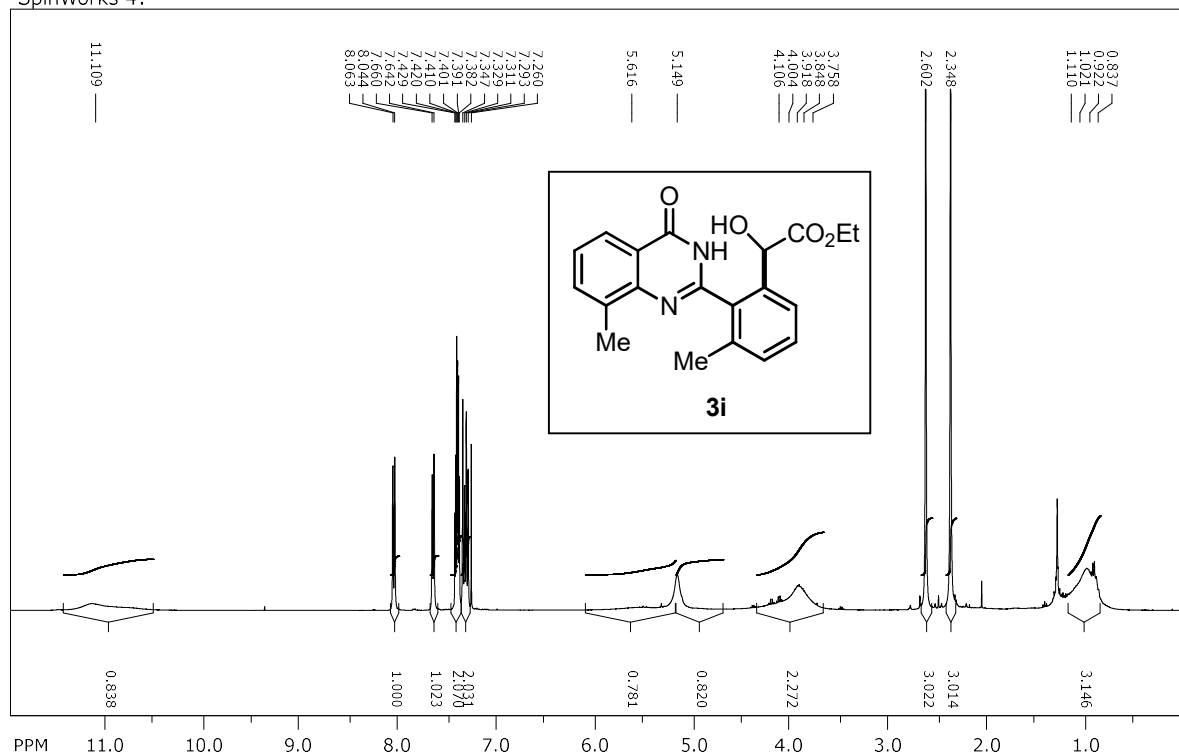
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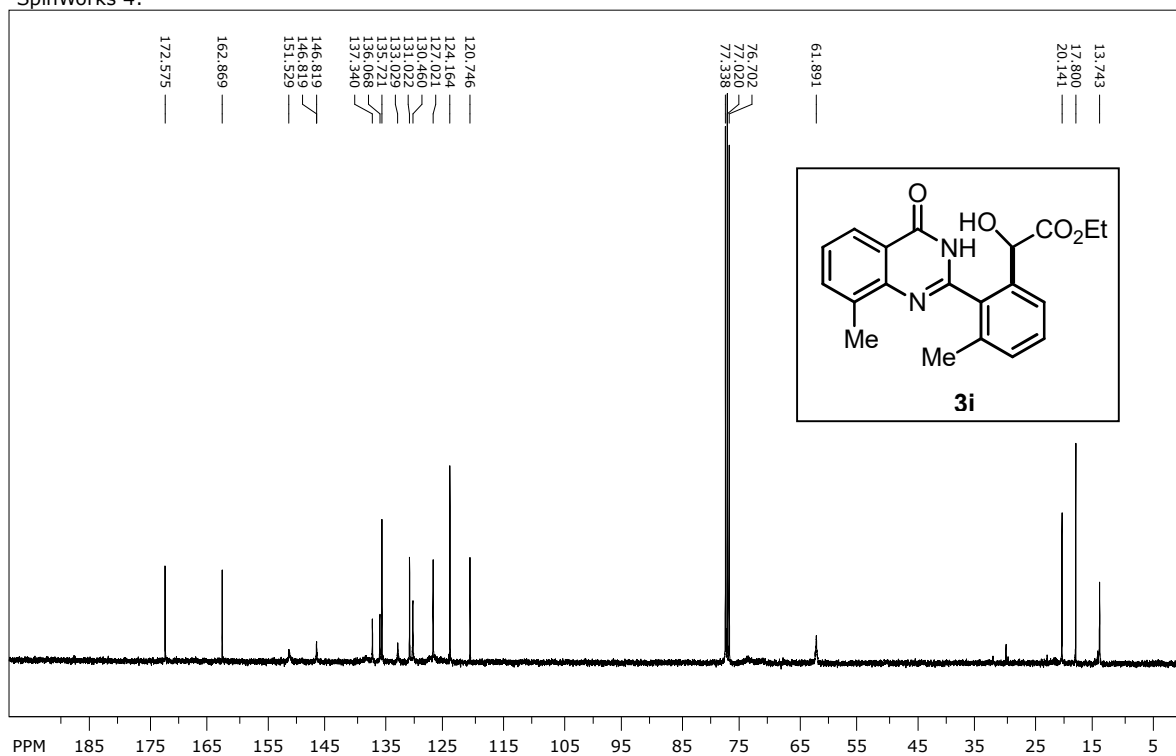
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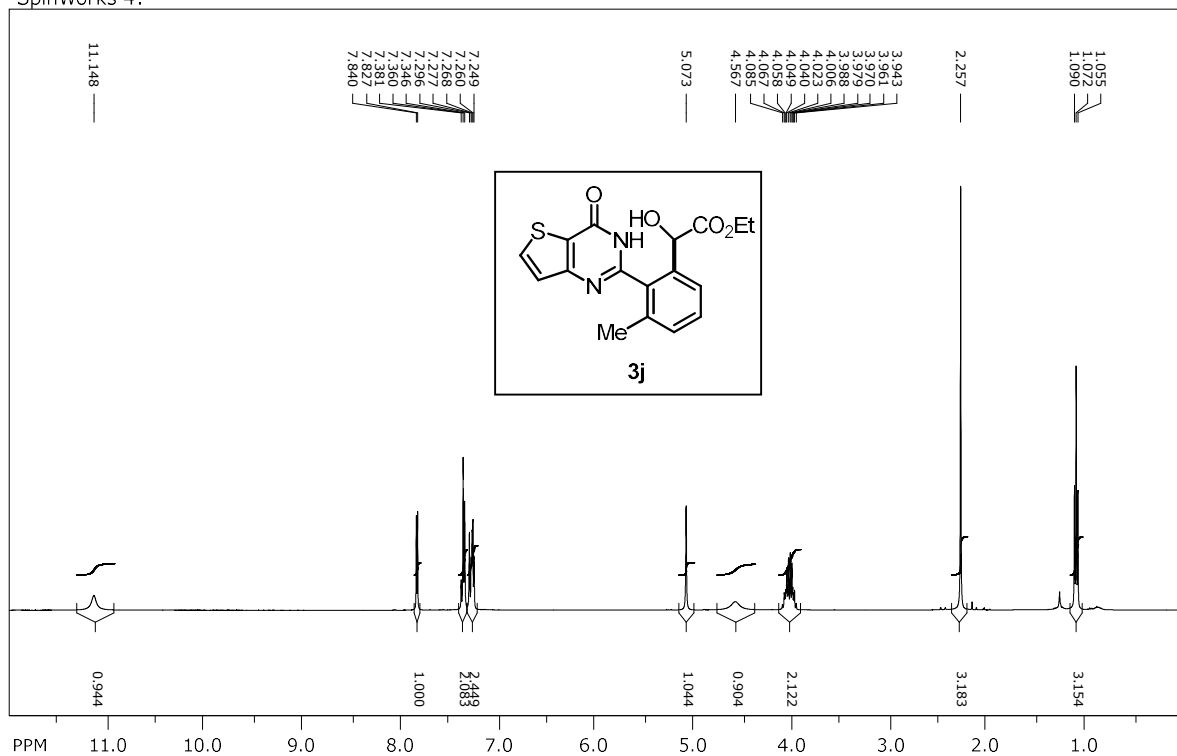
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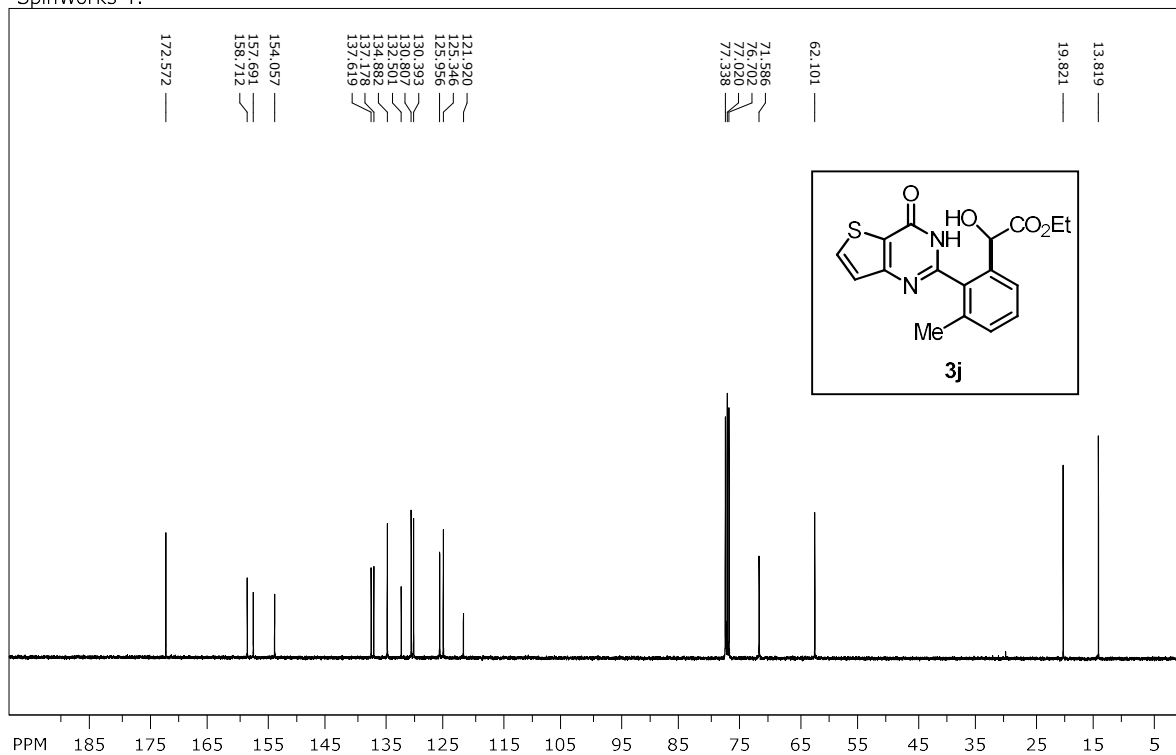
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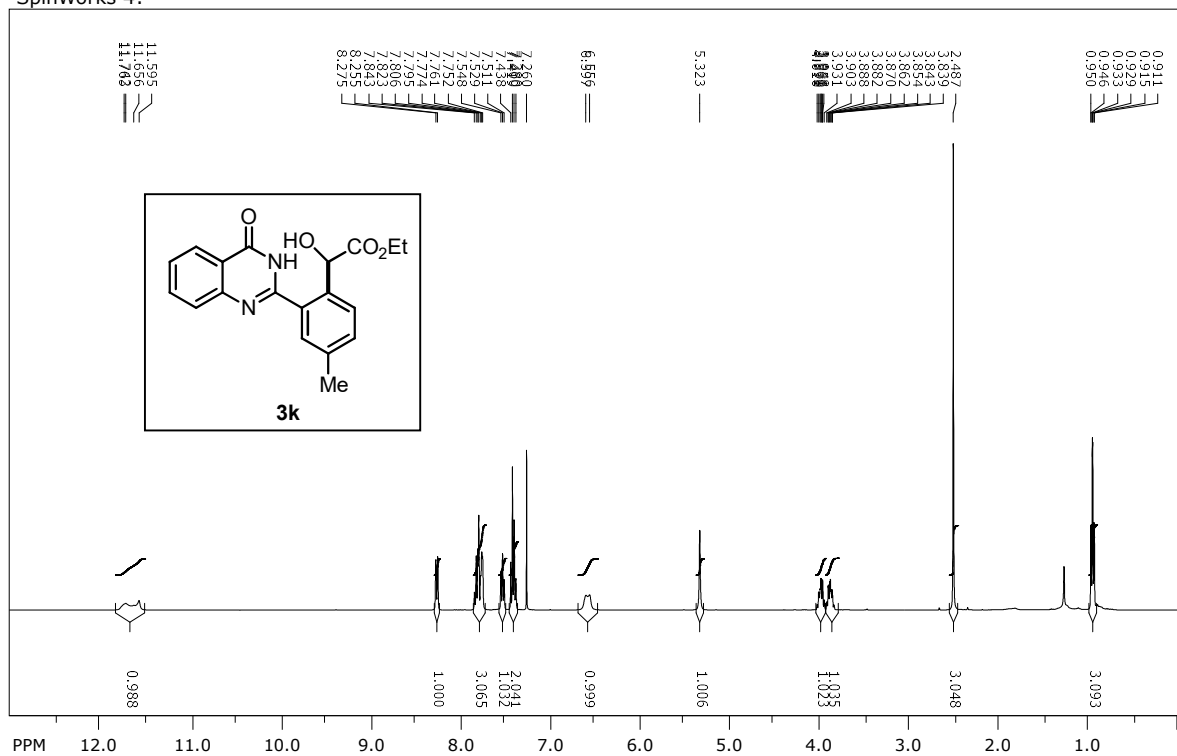
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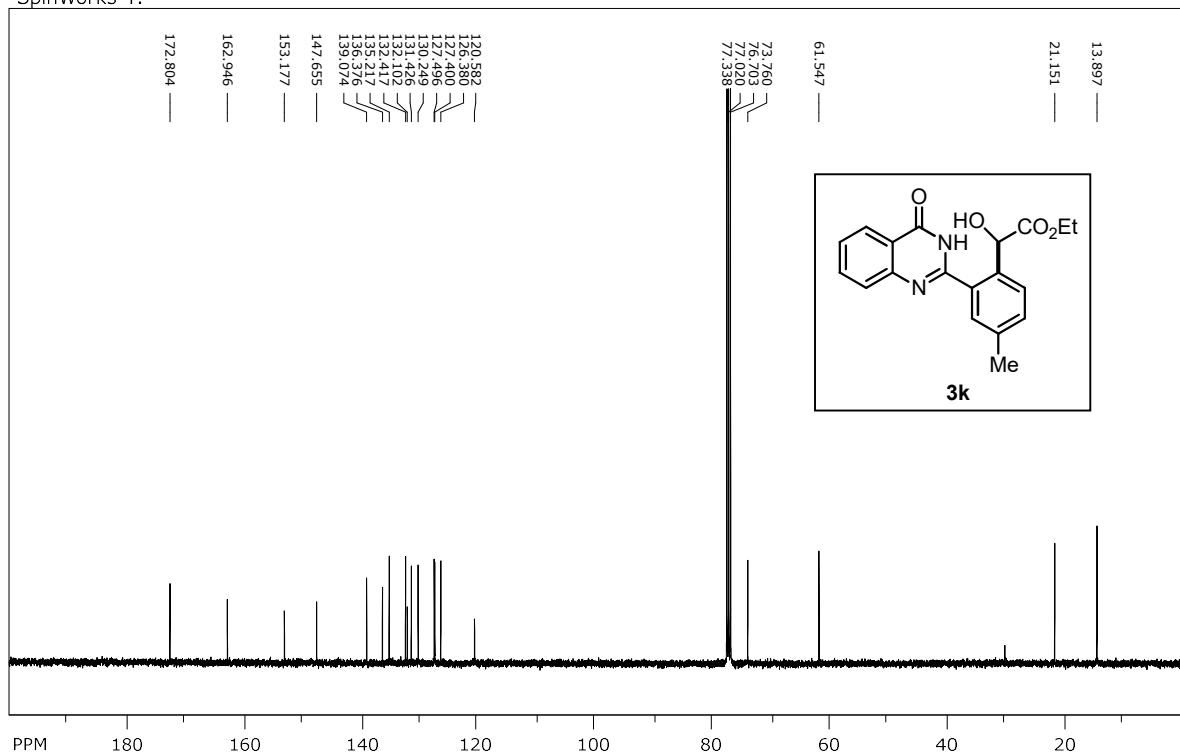
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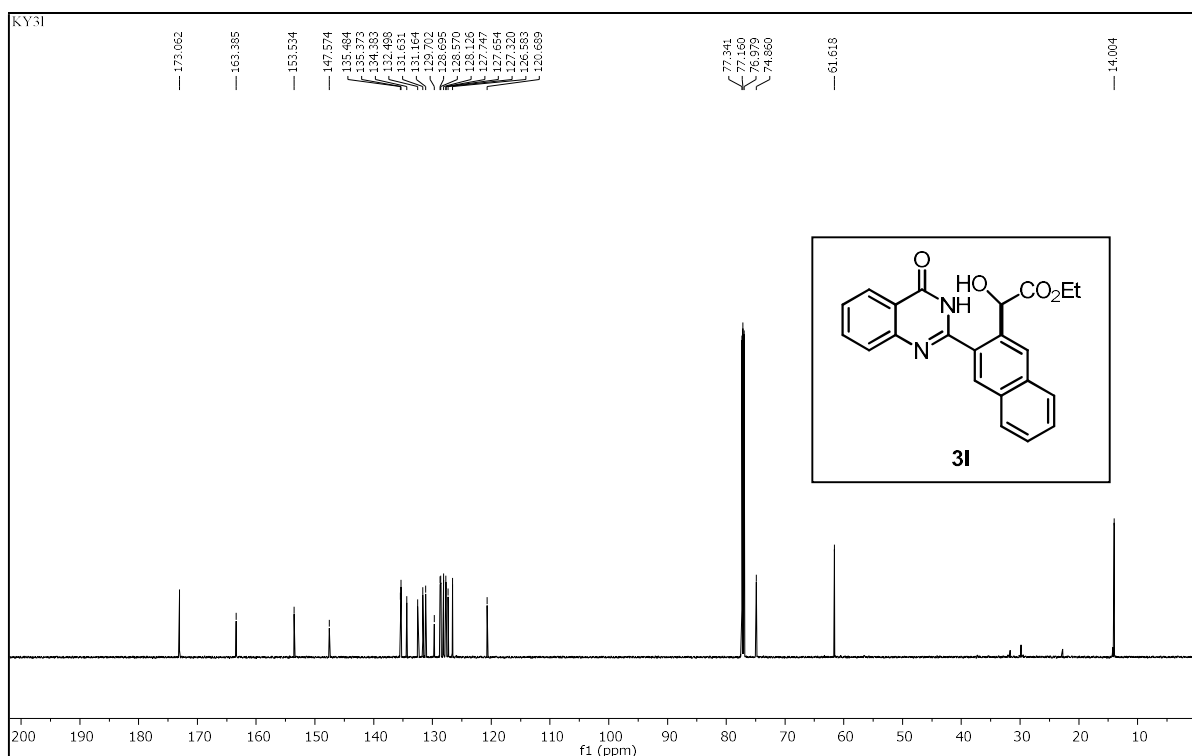
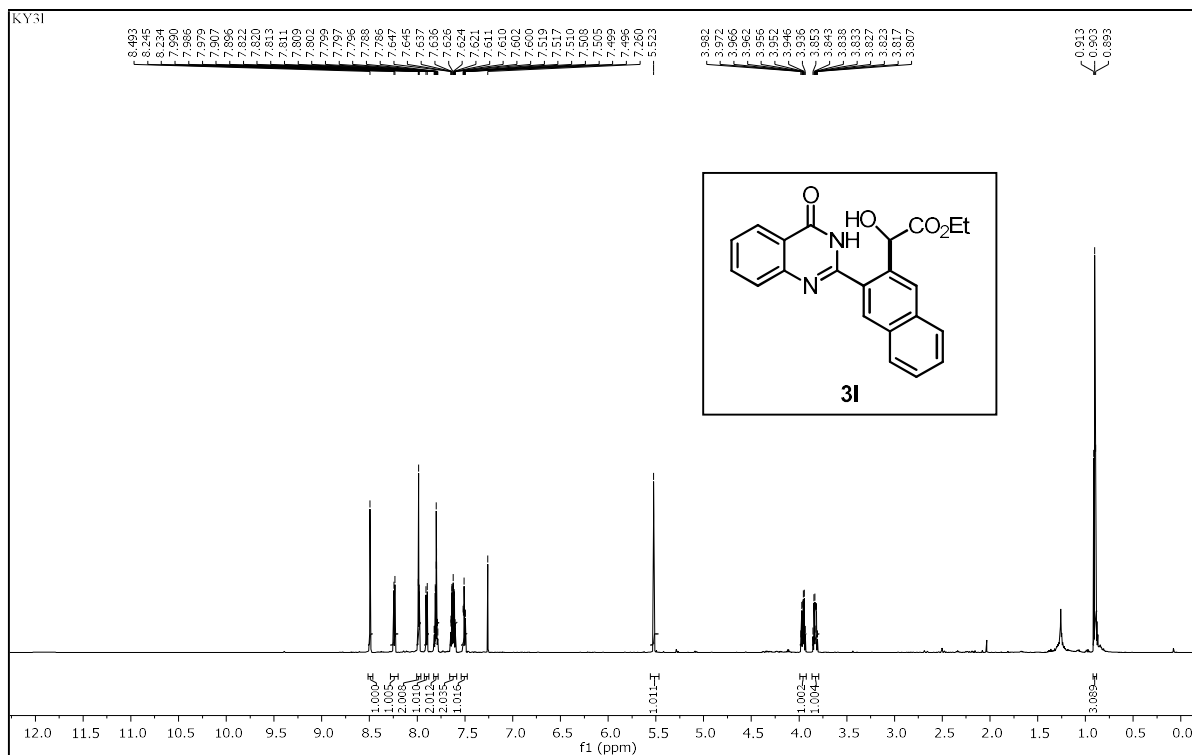


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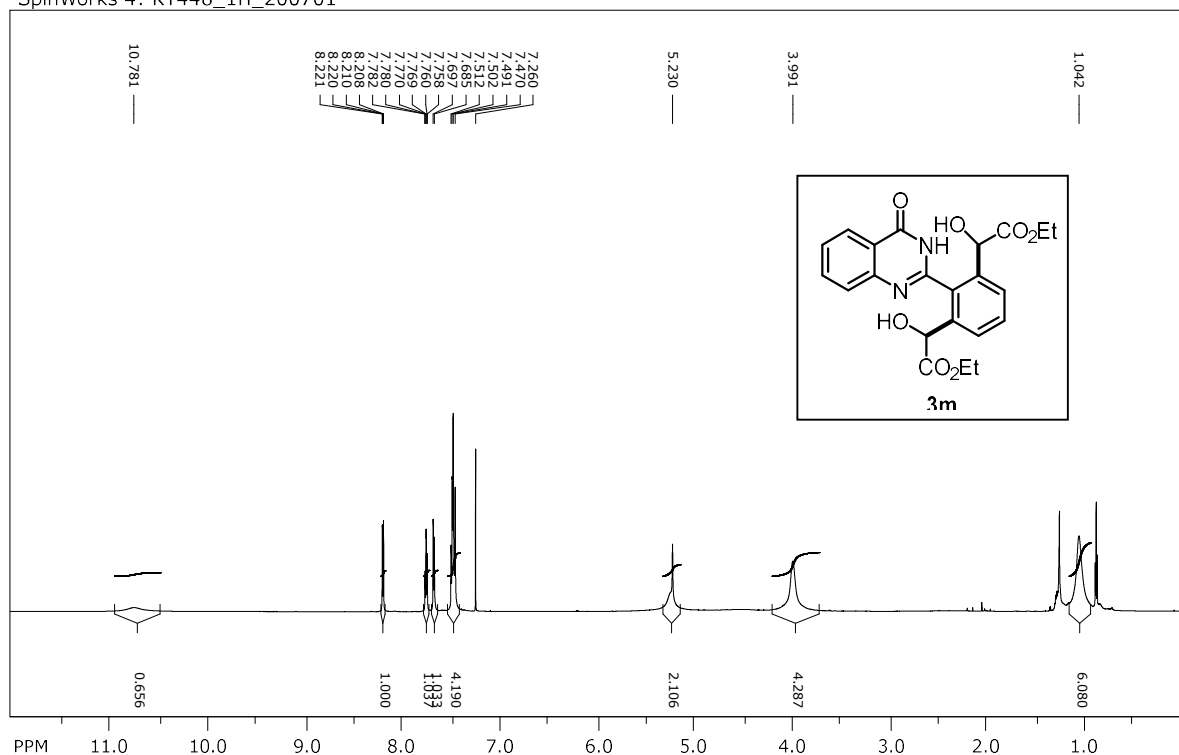


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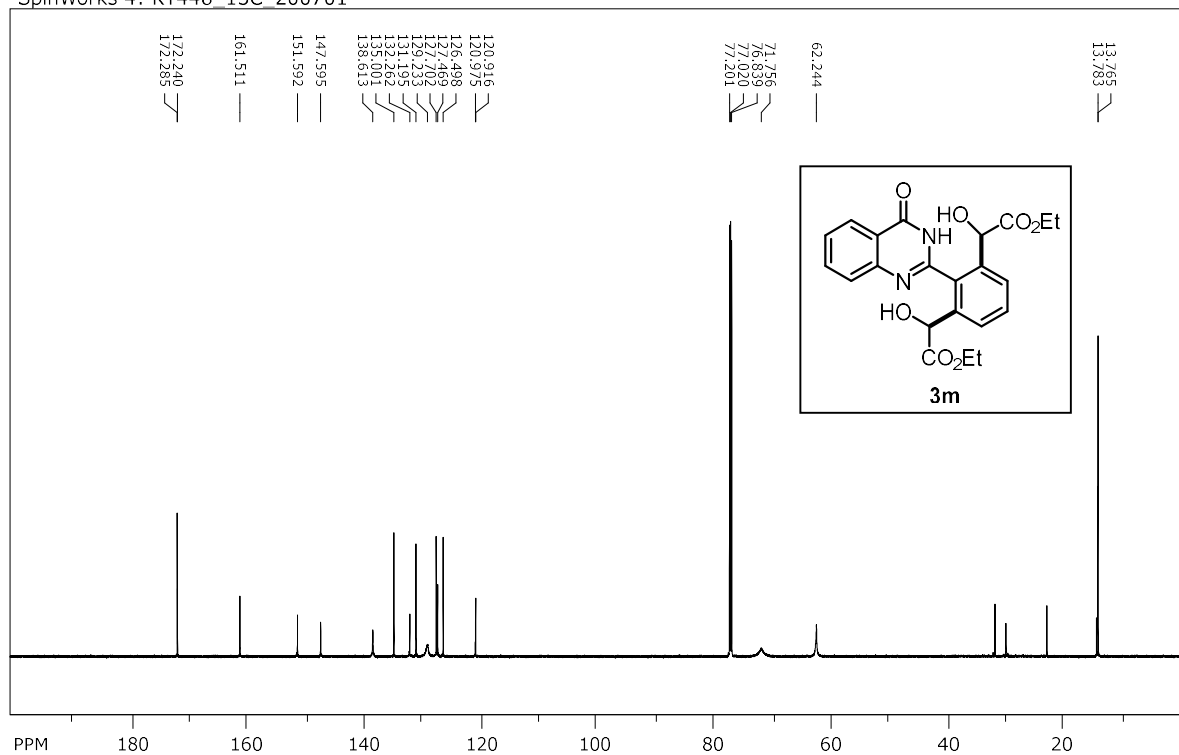


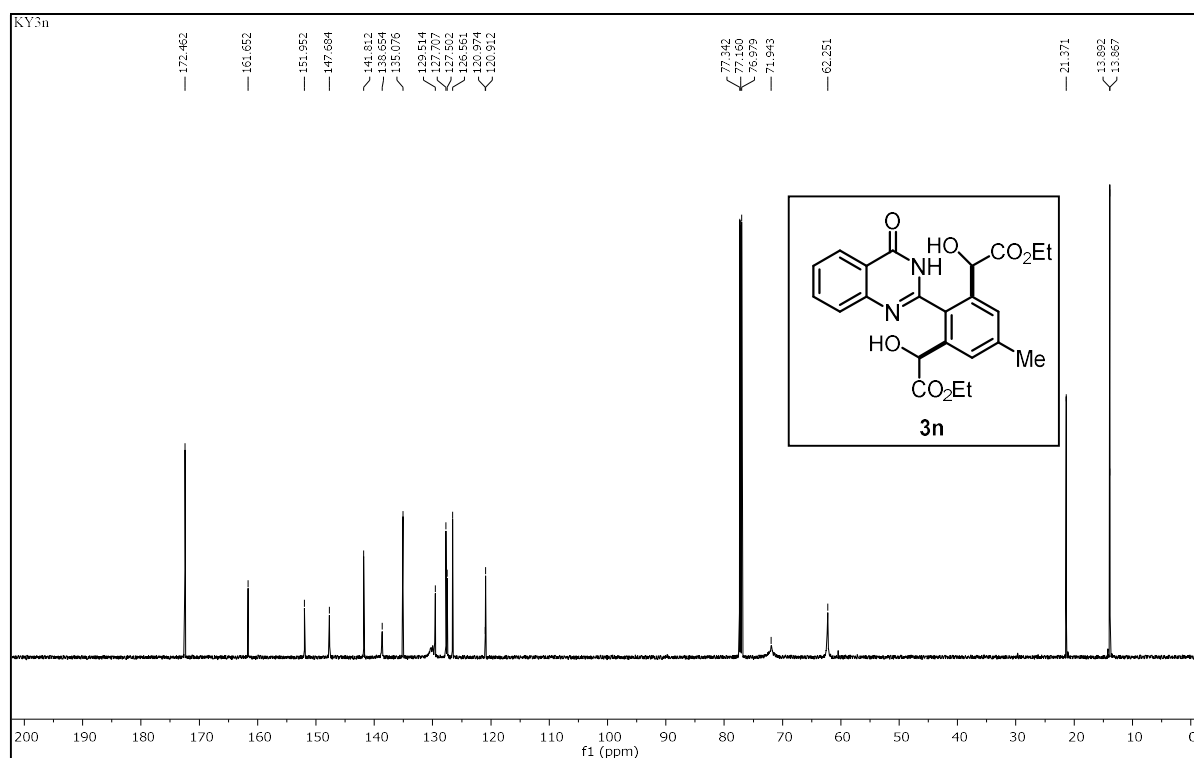
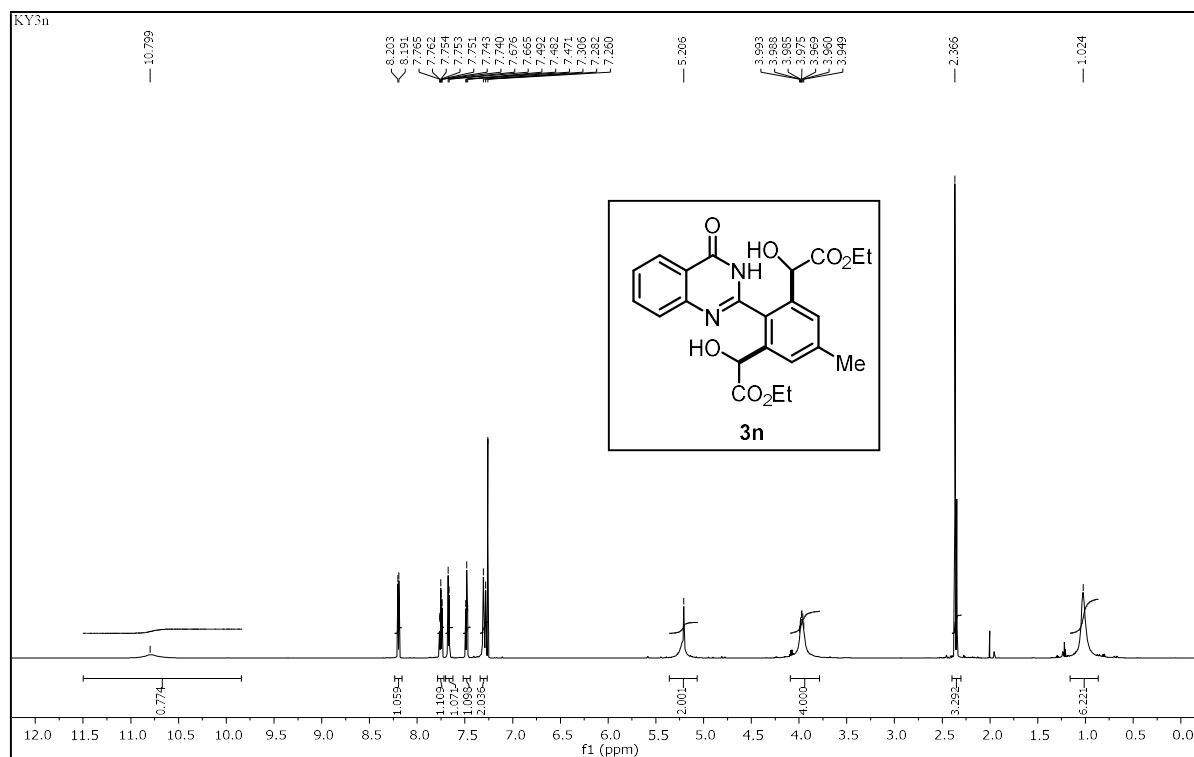


SpinWorks 4: KY448_1H_200701

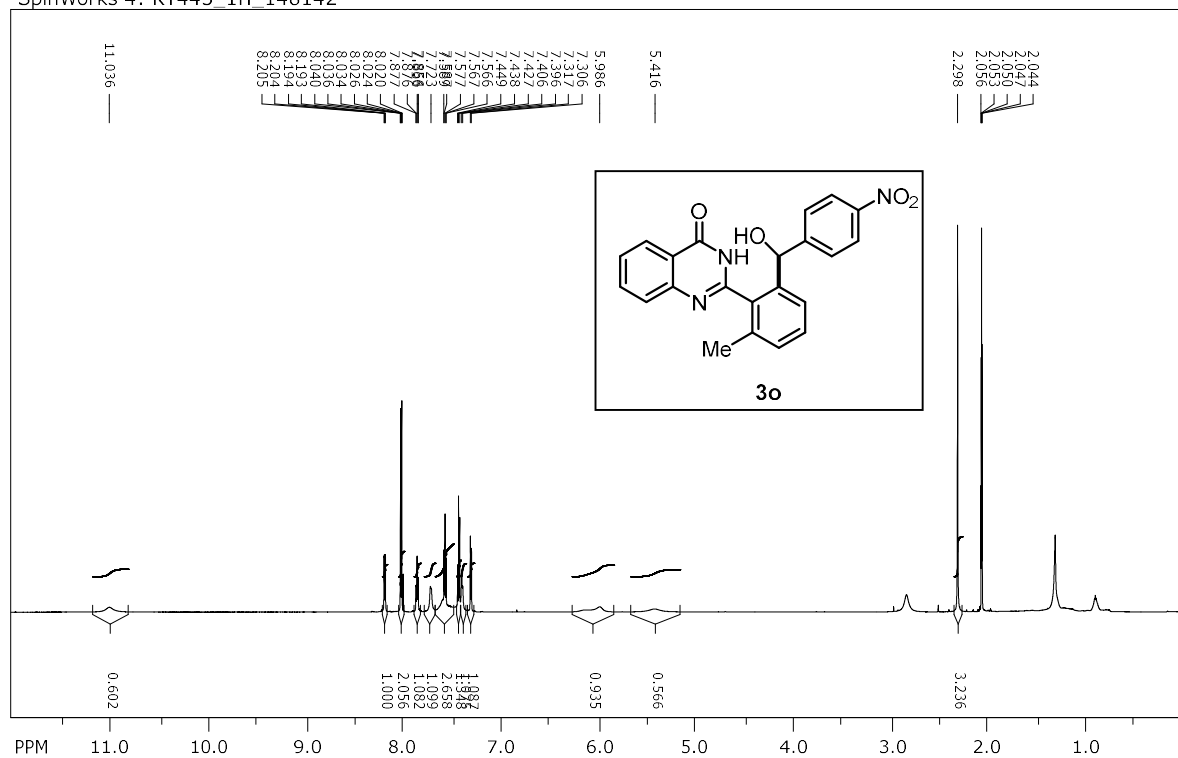


SpinWorks 4: KY448_13C_200701

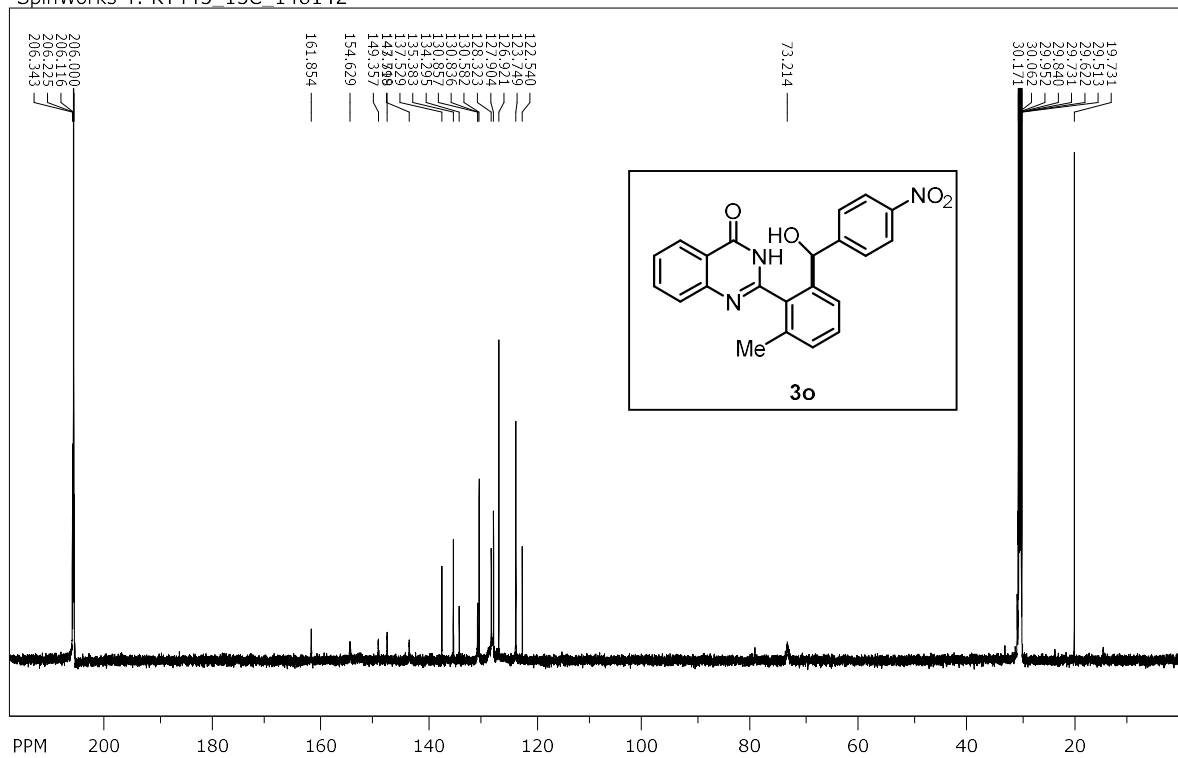




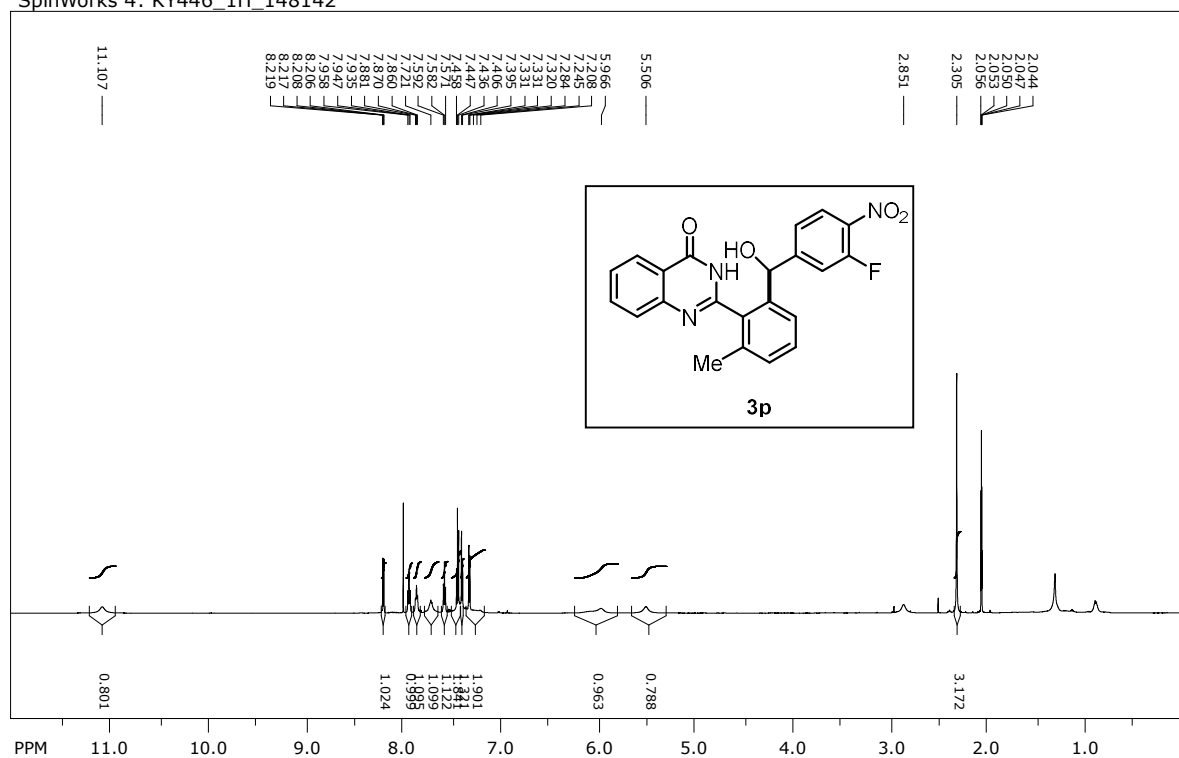
SpinWorks 4: KY443_1H_148142



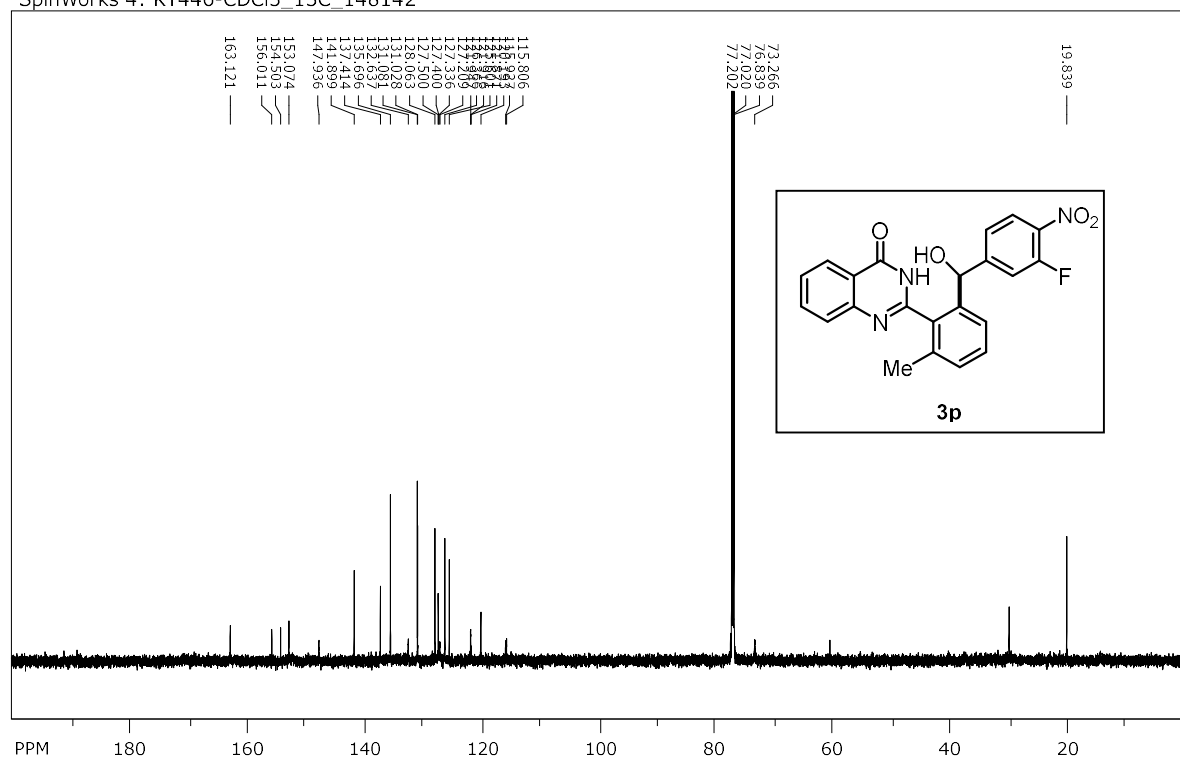
SpinWorks 4: KY443_13C_148142



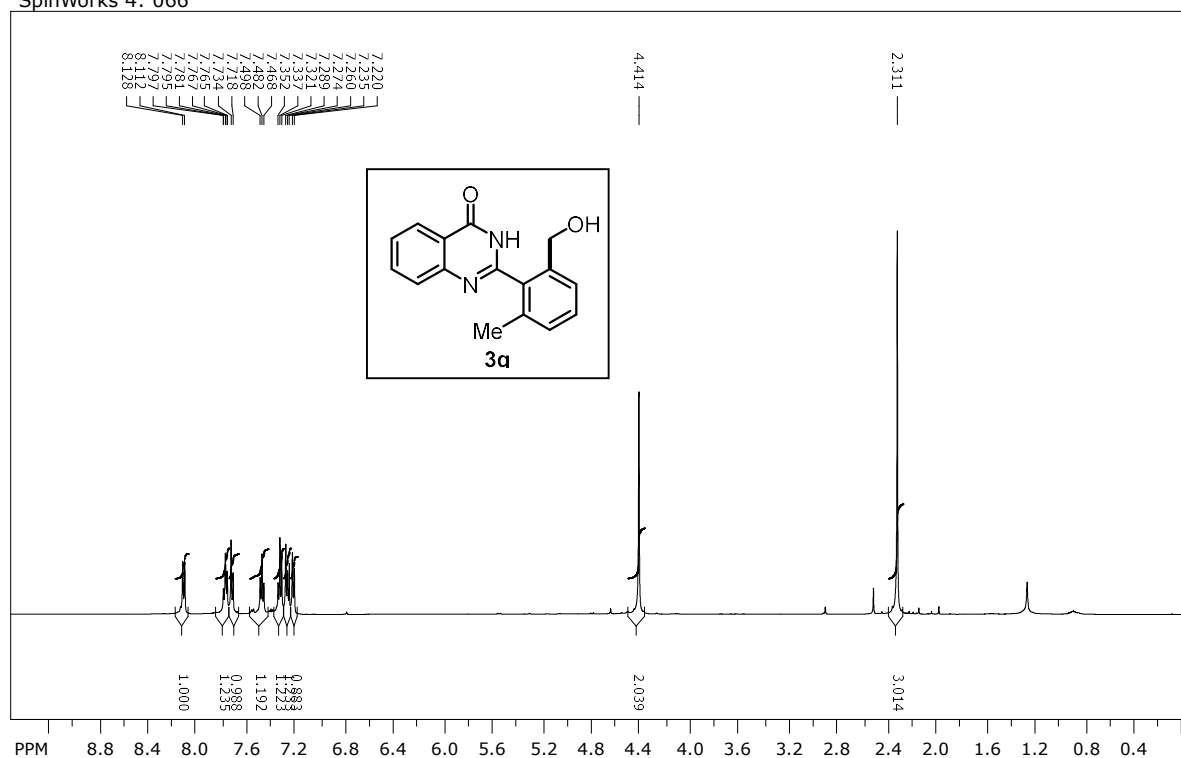
SpinWorks 4: KY446_1H_148142



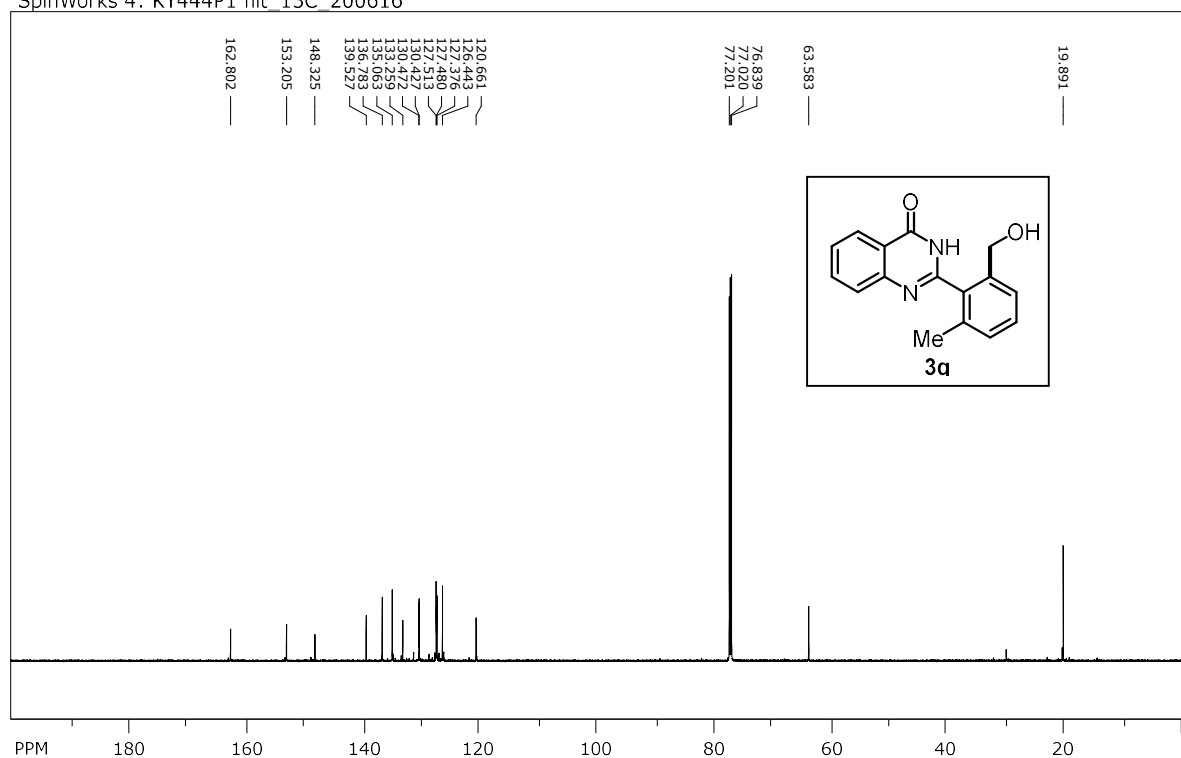
SpinWorks 4: KY446-CDCl3_13C_148142



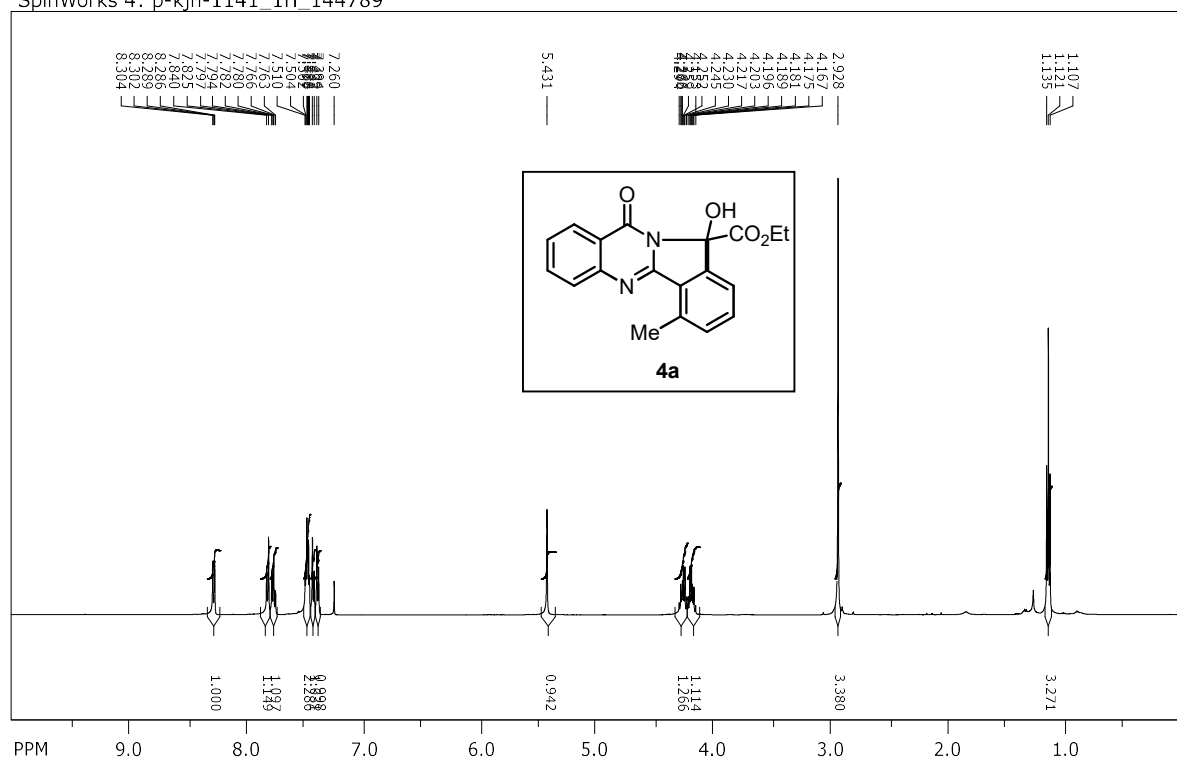
SpinWorks 4: 066



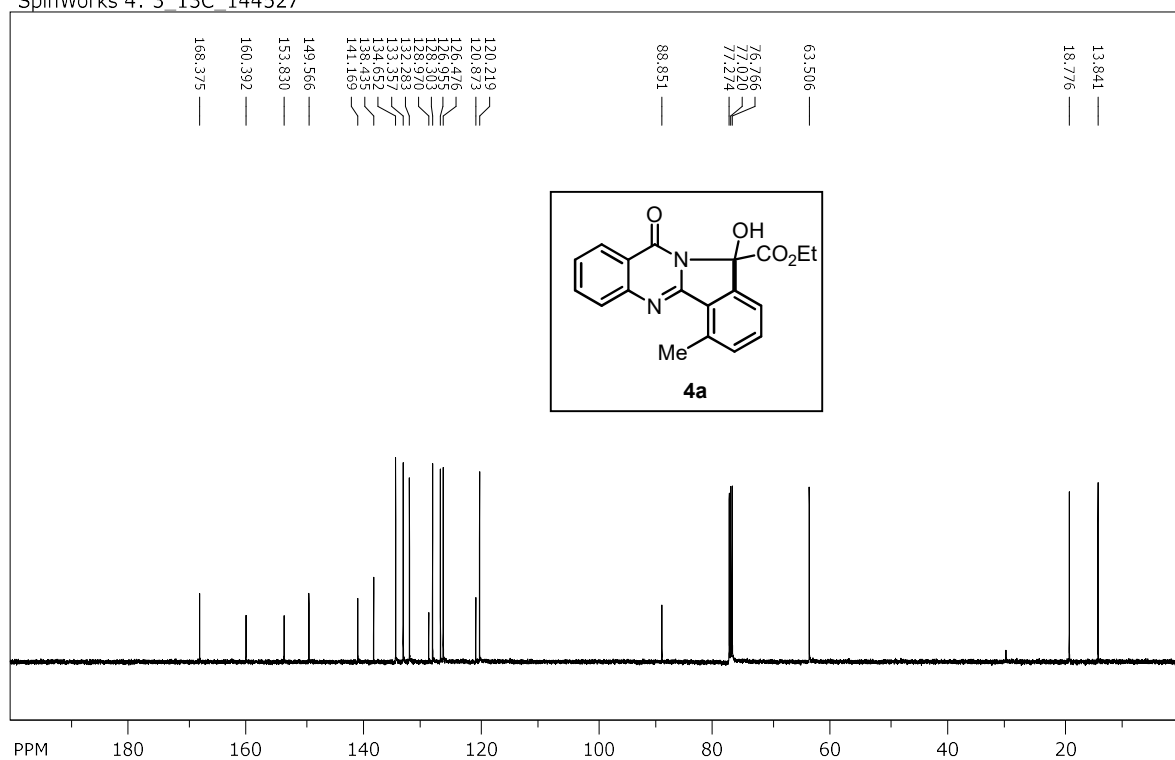
SpinWorks 4: KY444P1 filt 13C 200616



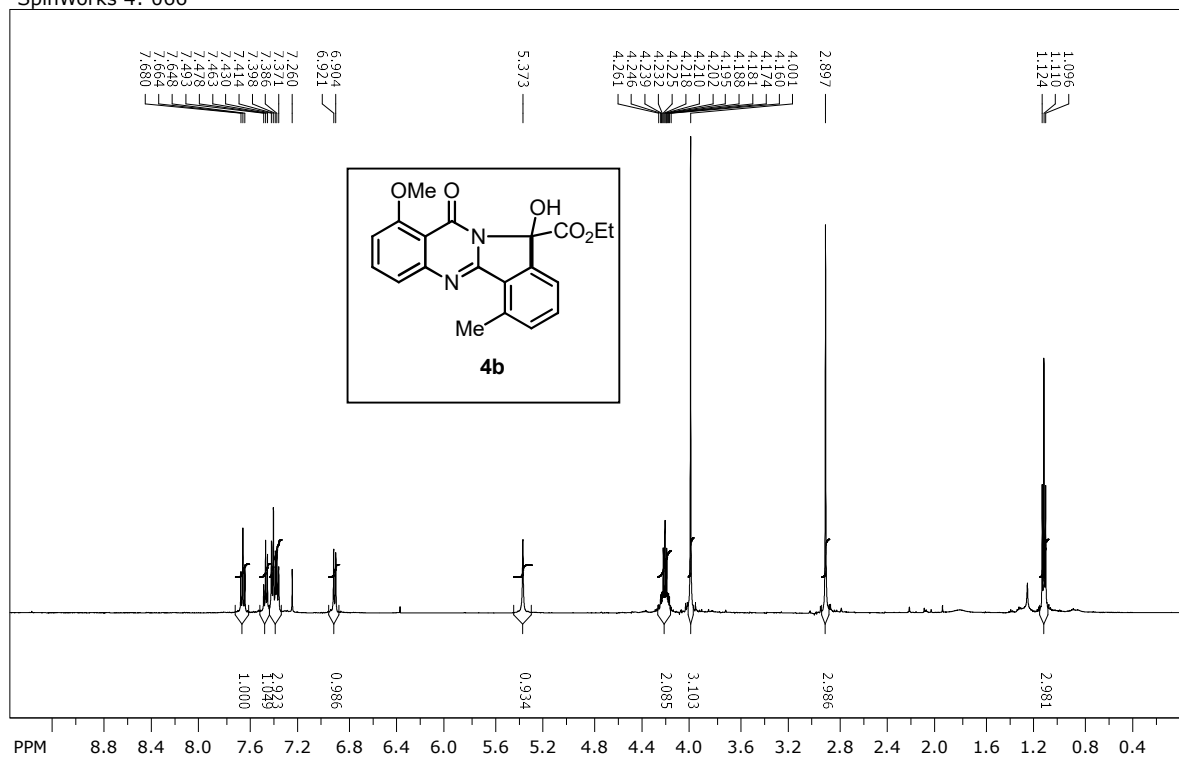
SpinWorks 4: p-kjh-1141_1H_144789



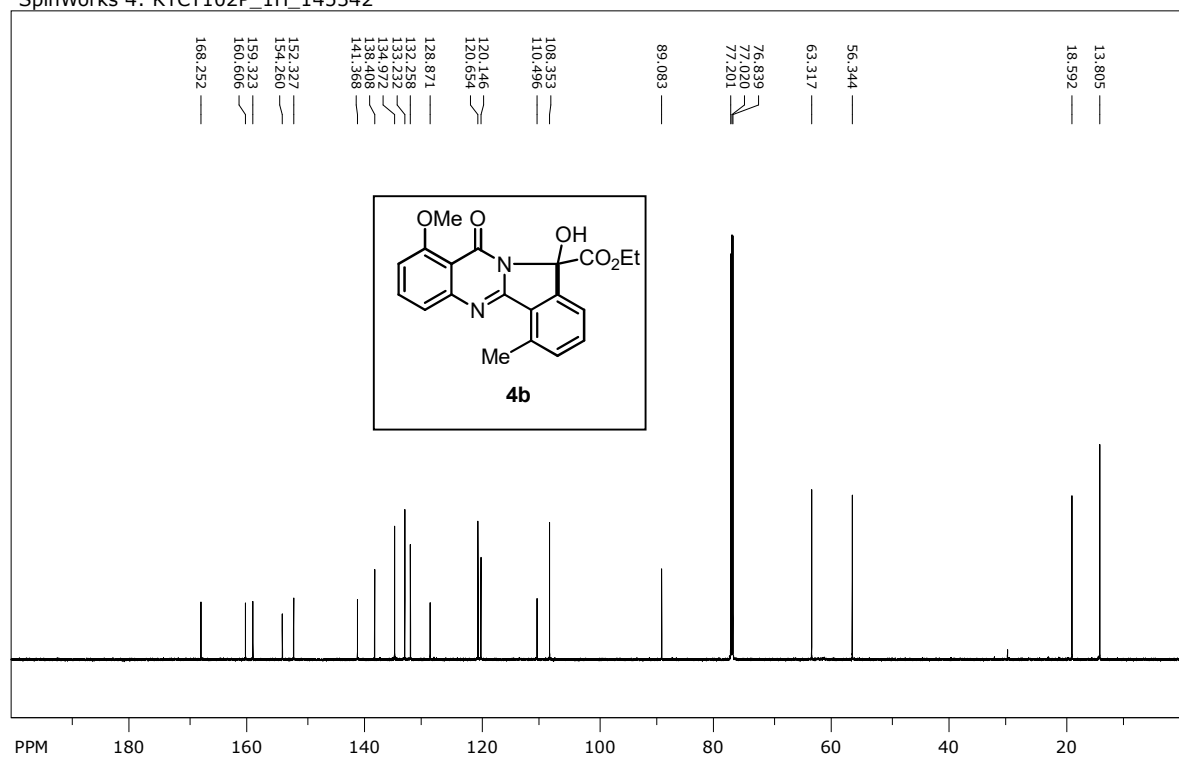
SpinWorks 4: 3_13C_144527



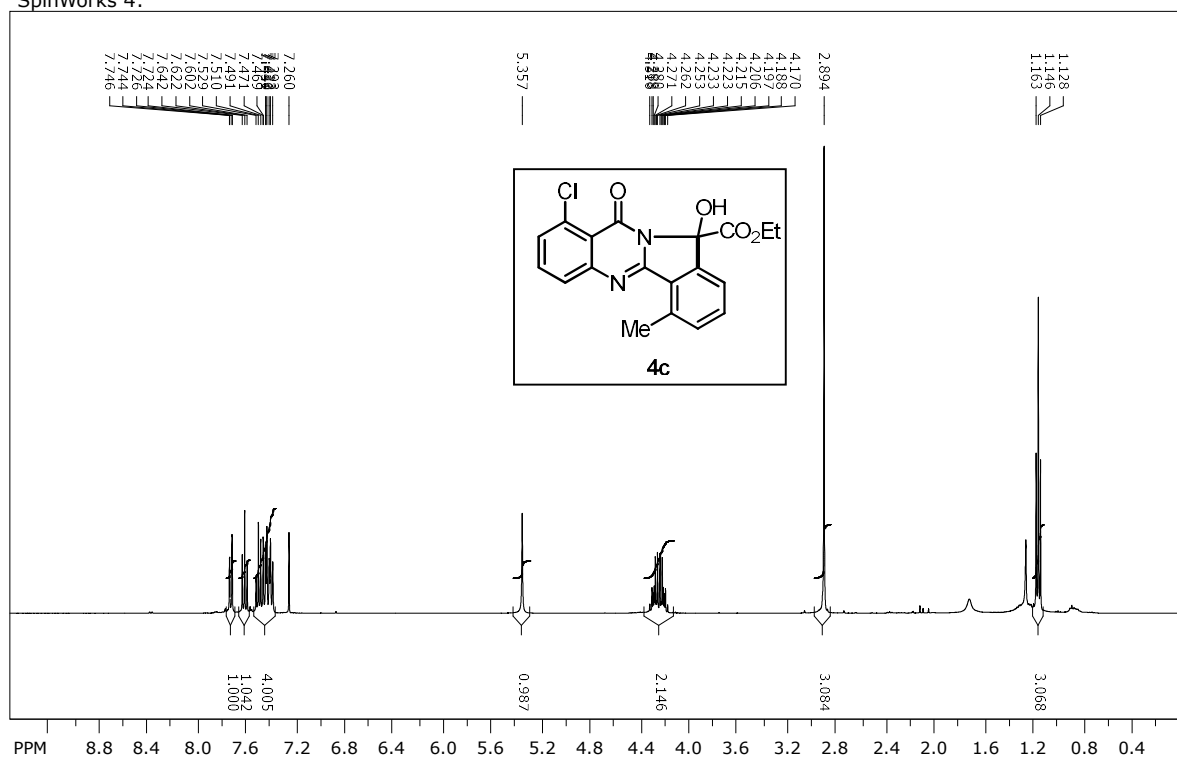
SpinWorks 4: 066



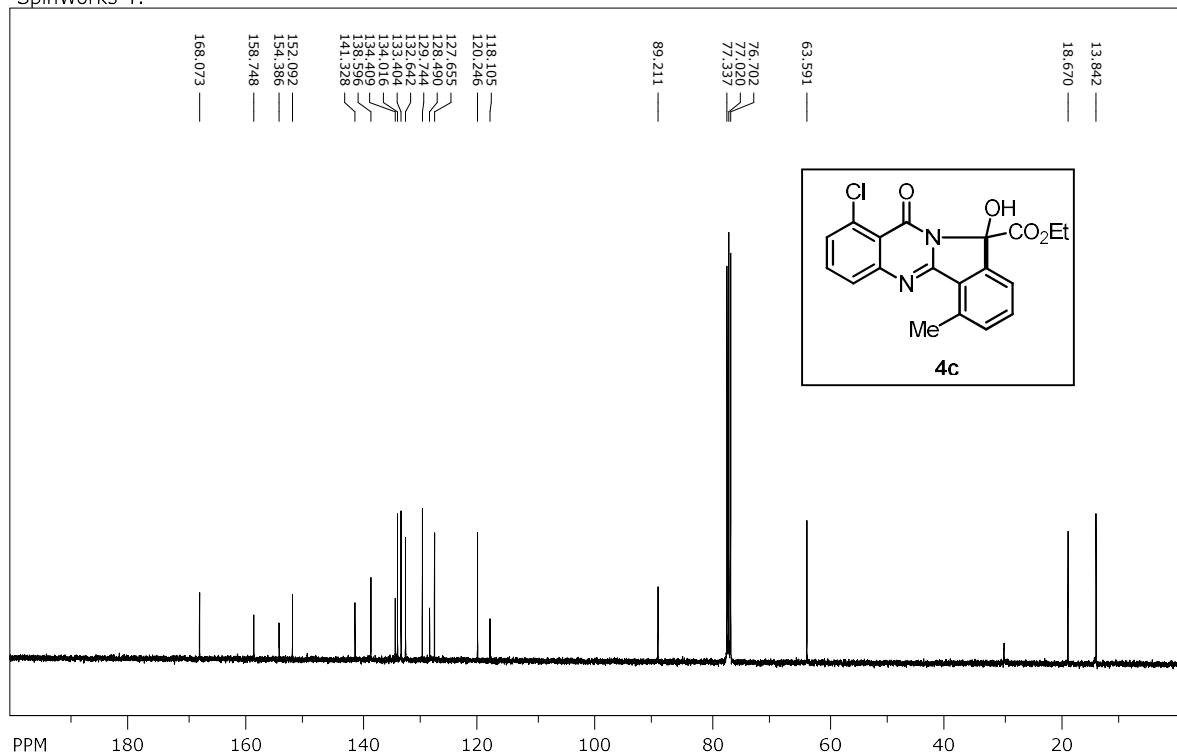
SpinWorks 4: KYCT102P_1H_145342



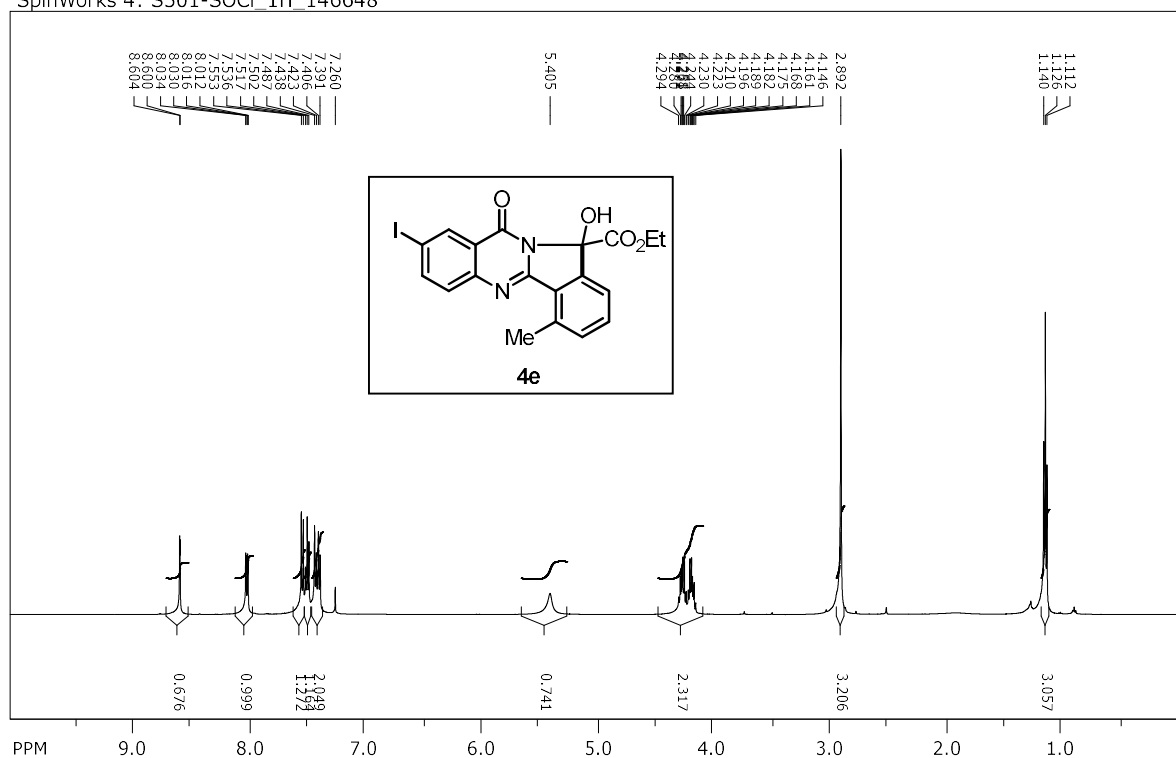
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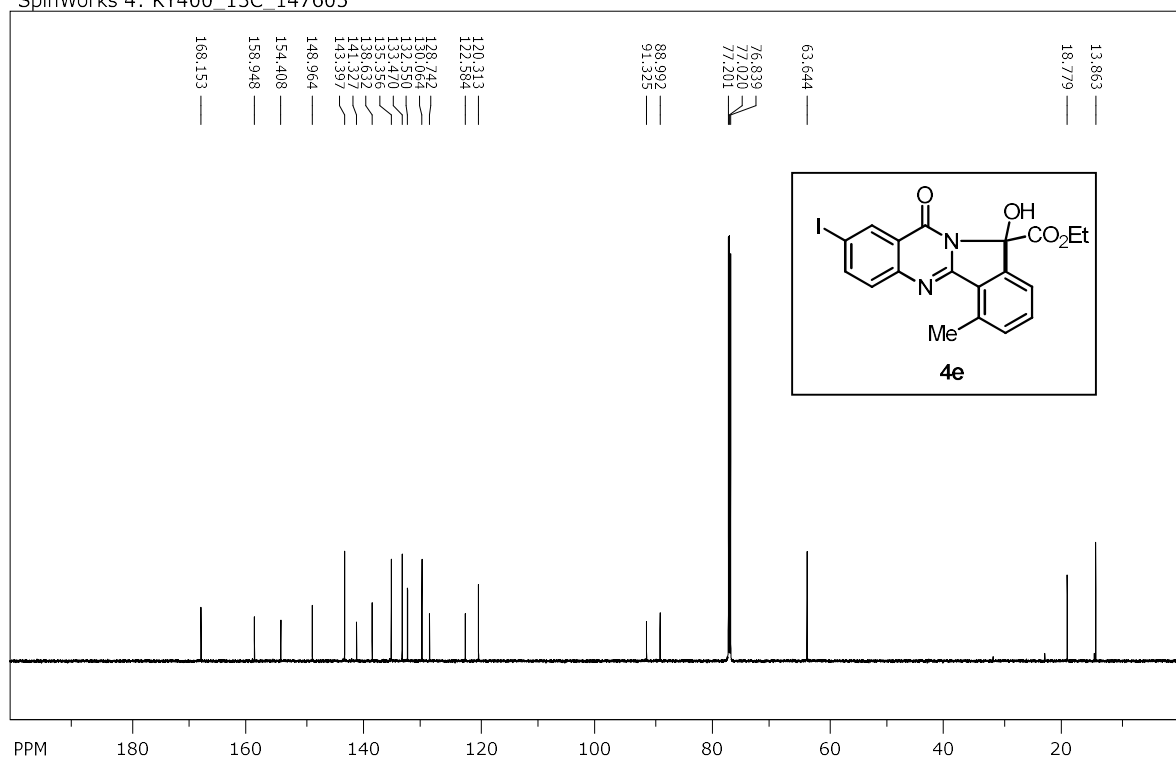
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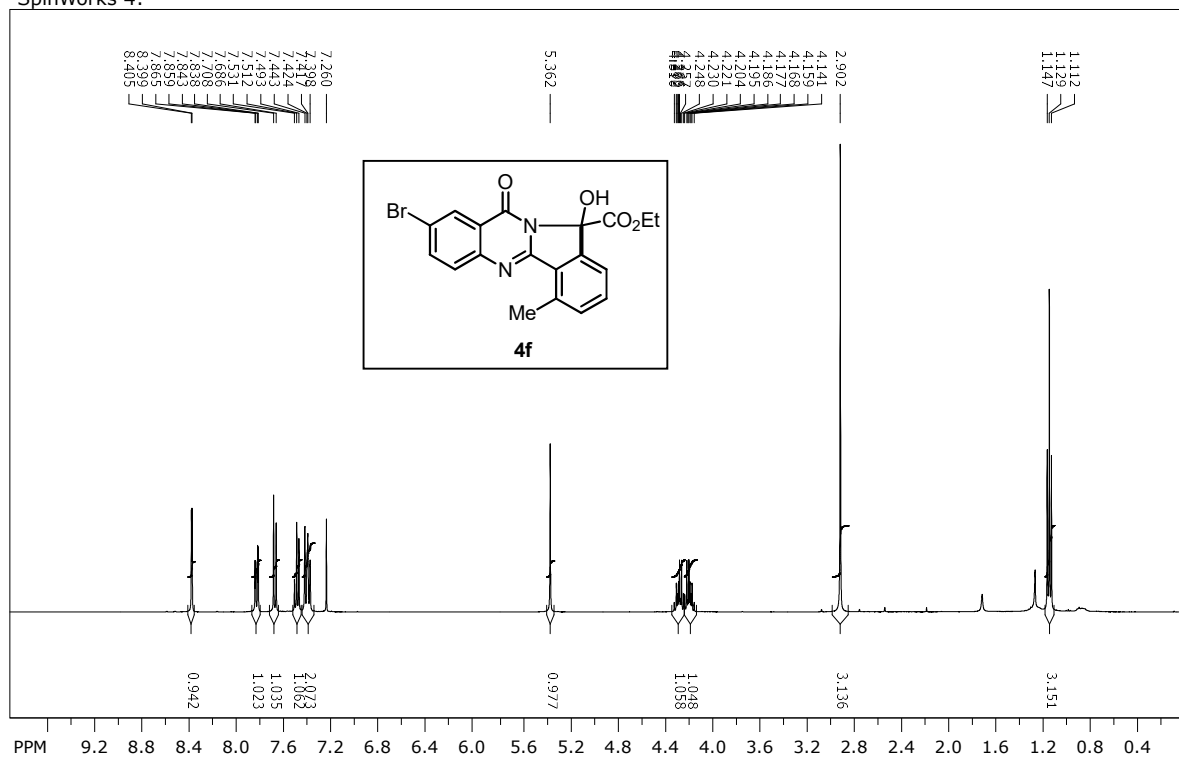
SpinWorks 4: S501-SOCl_1H_146648



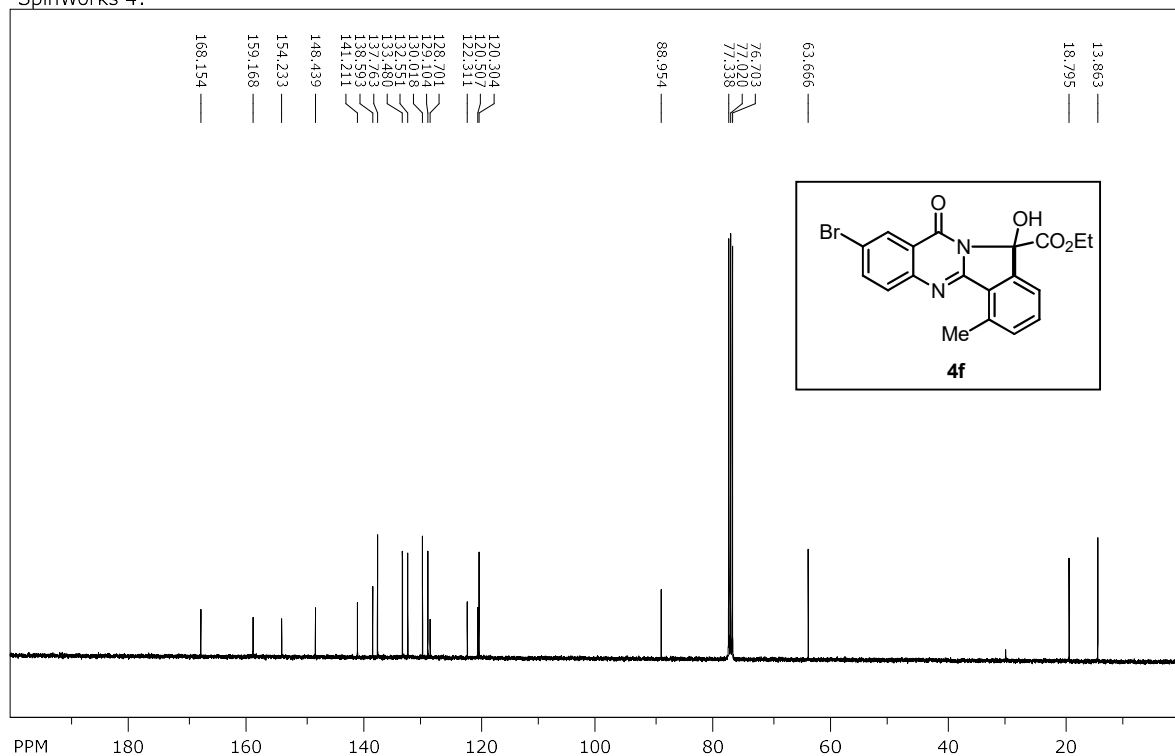
SpinWorks 4: KY400_13C_147605



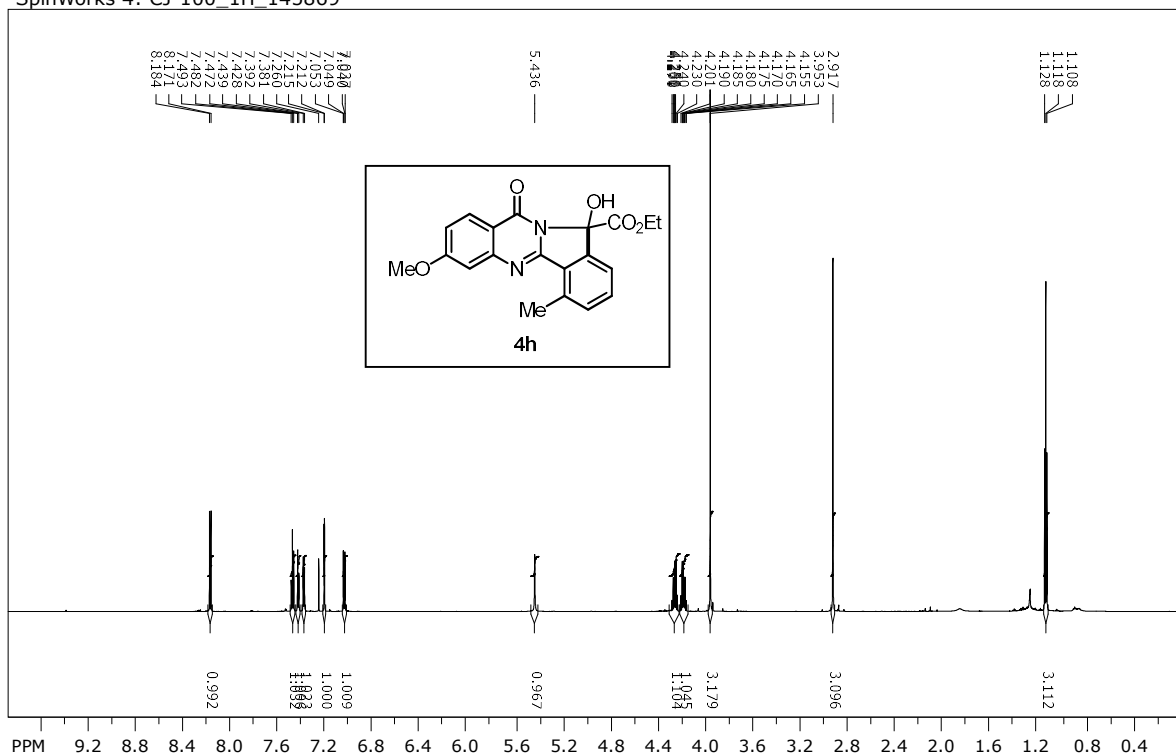
SpinWorks 4:



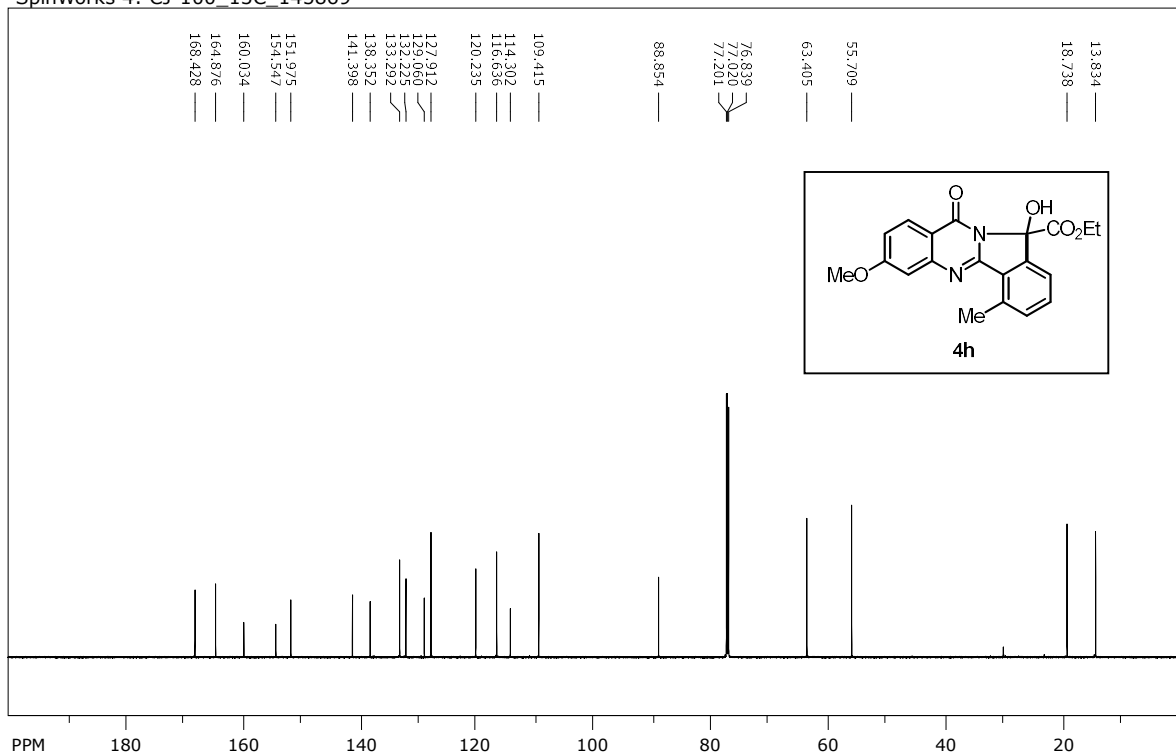
SpinWorks 4:



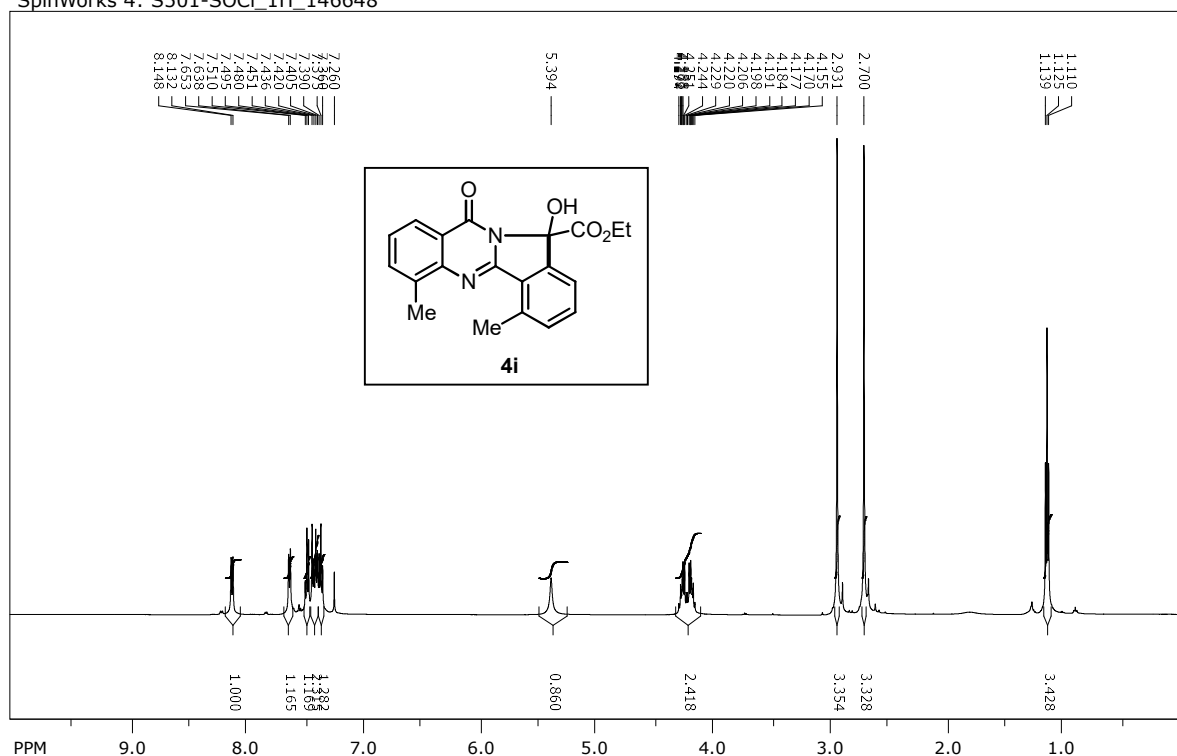
SpinWorks 4: CJ-100_1H_145869



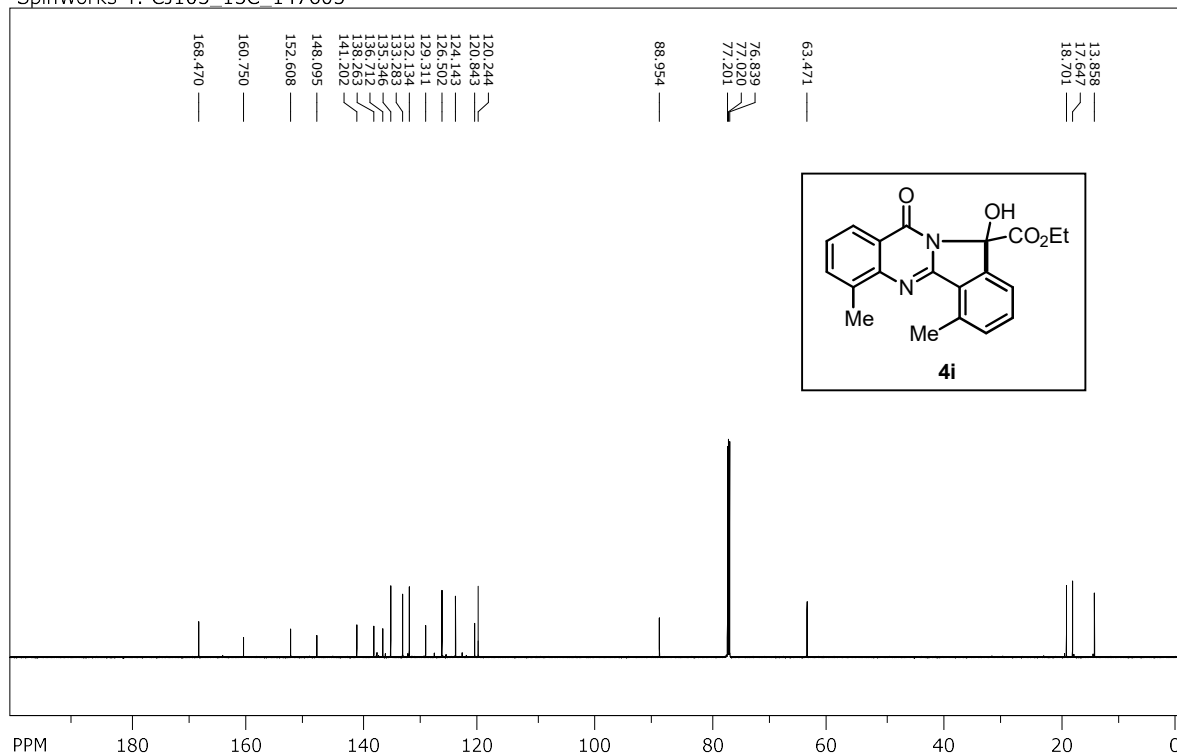
SpinWorks 4: CJ-100_13C_145869



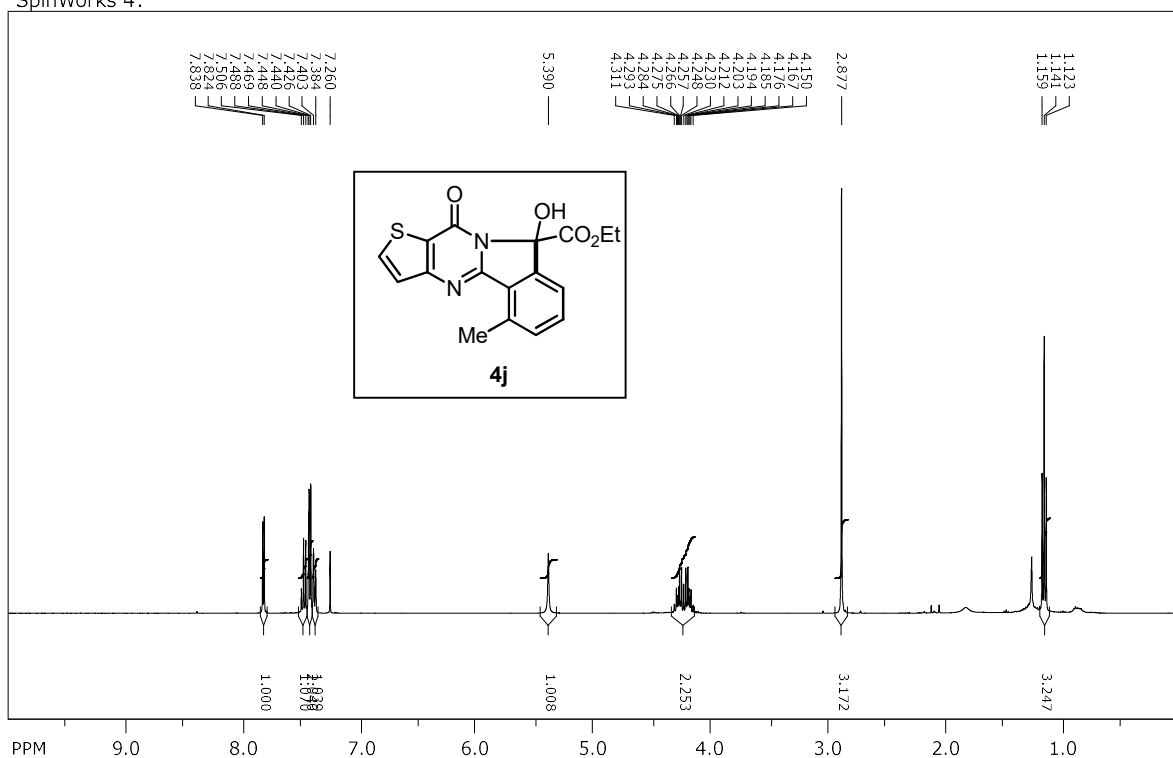
SpinWorks 4: S501-SOCl_1H_146648



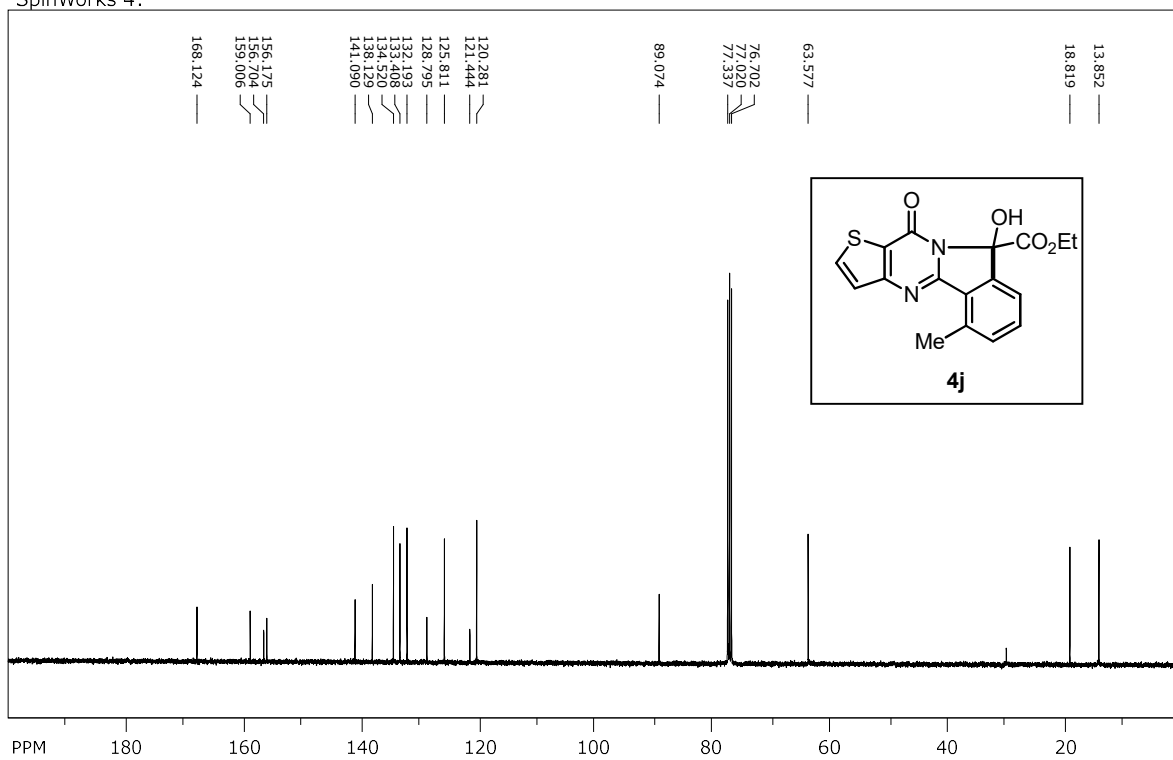
SpinWorks 4: CJ103_13C_147605



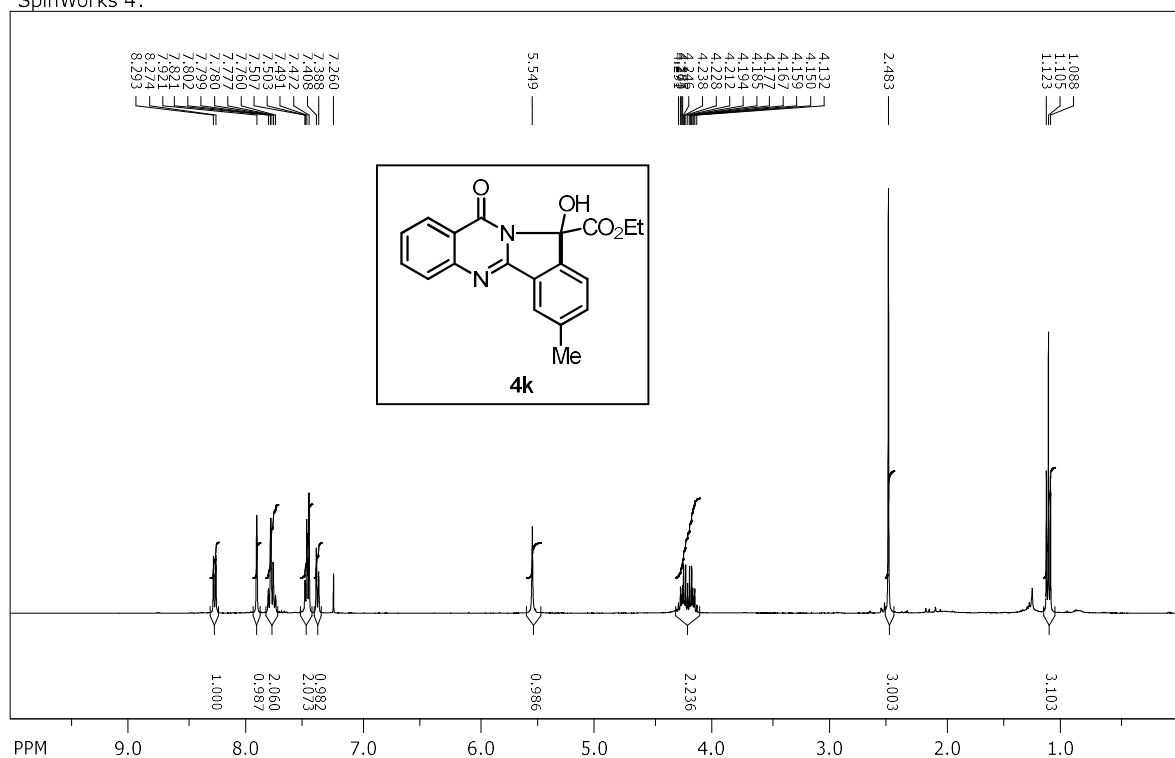
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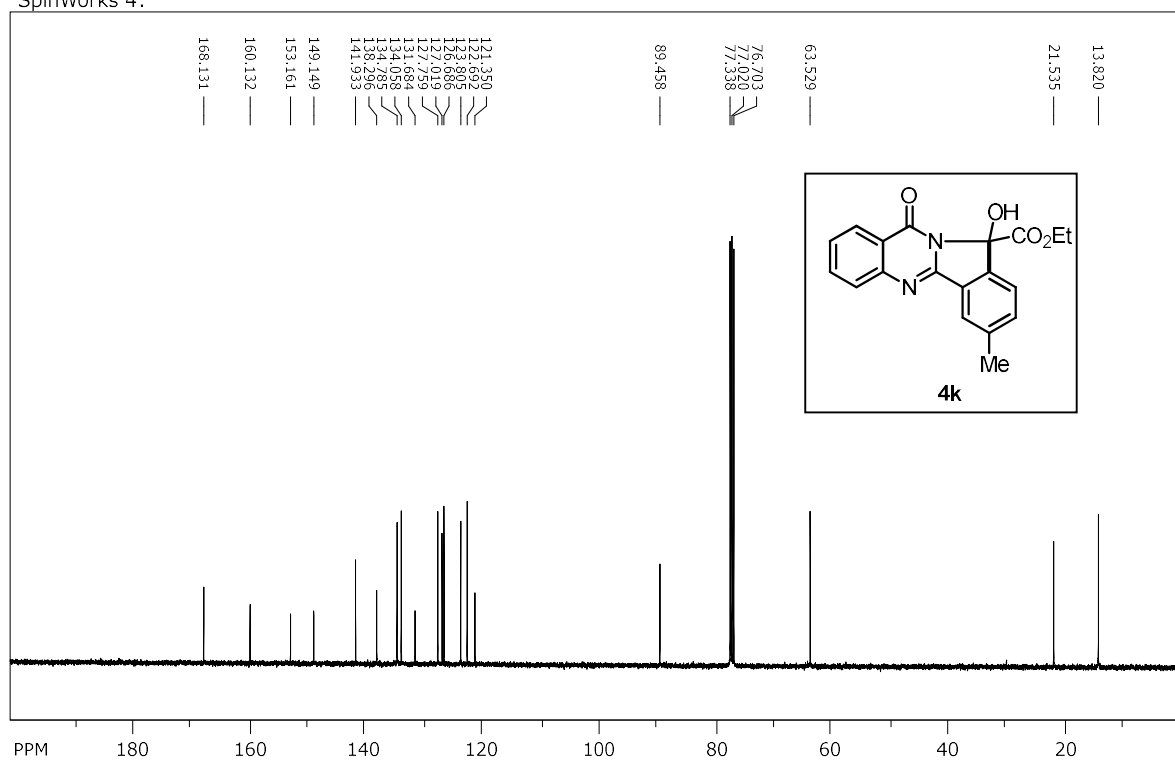
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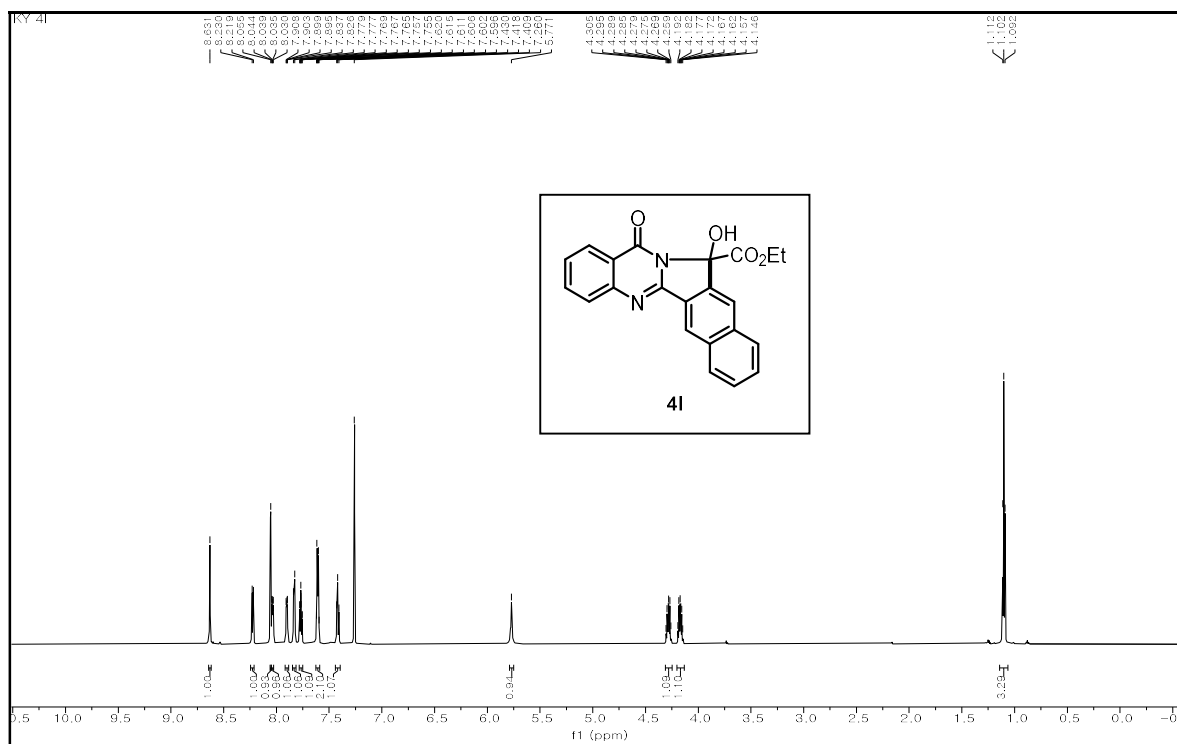


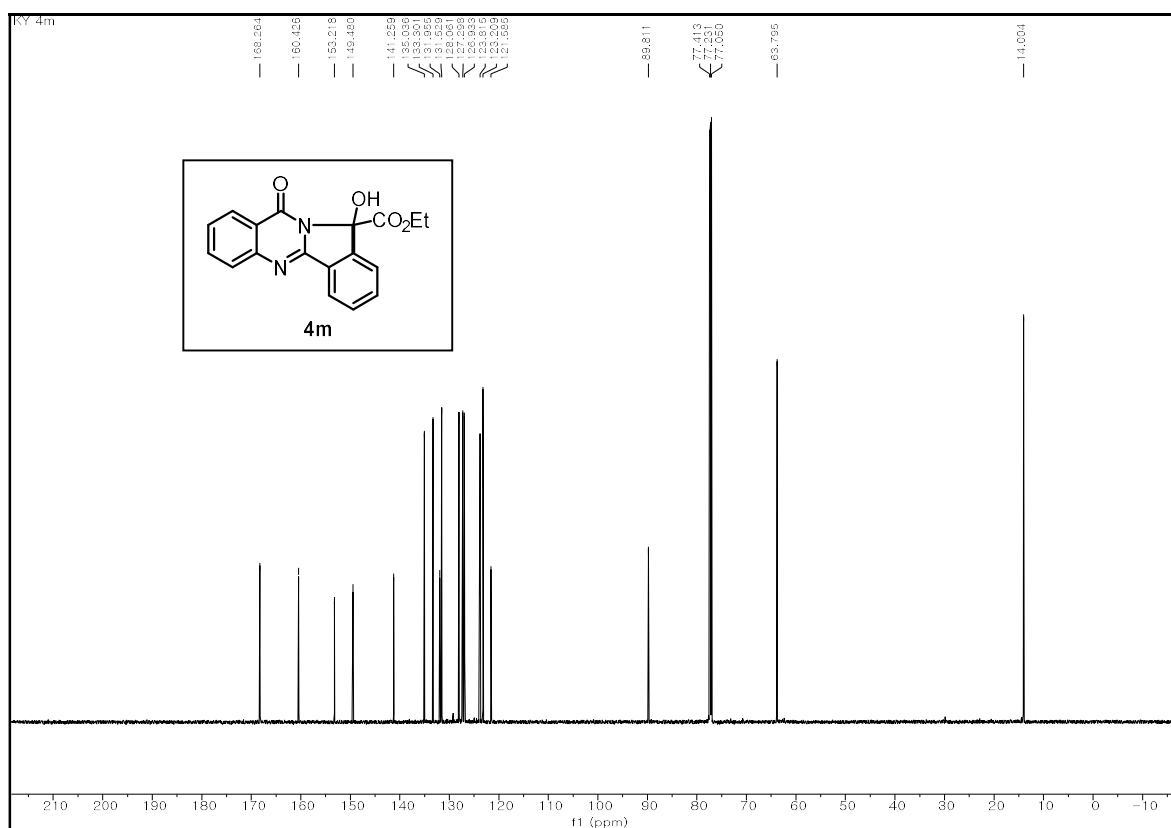
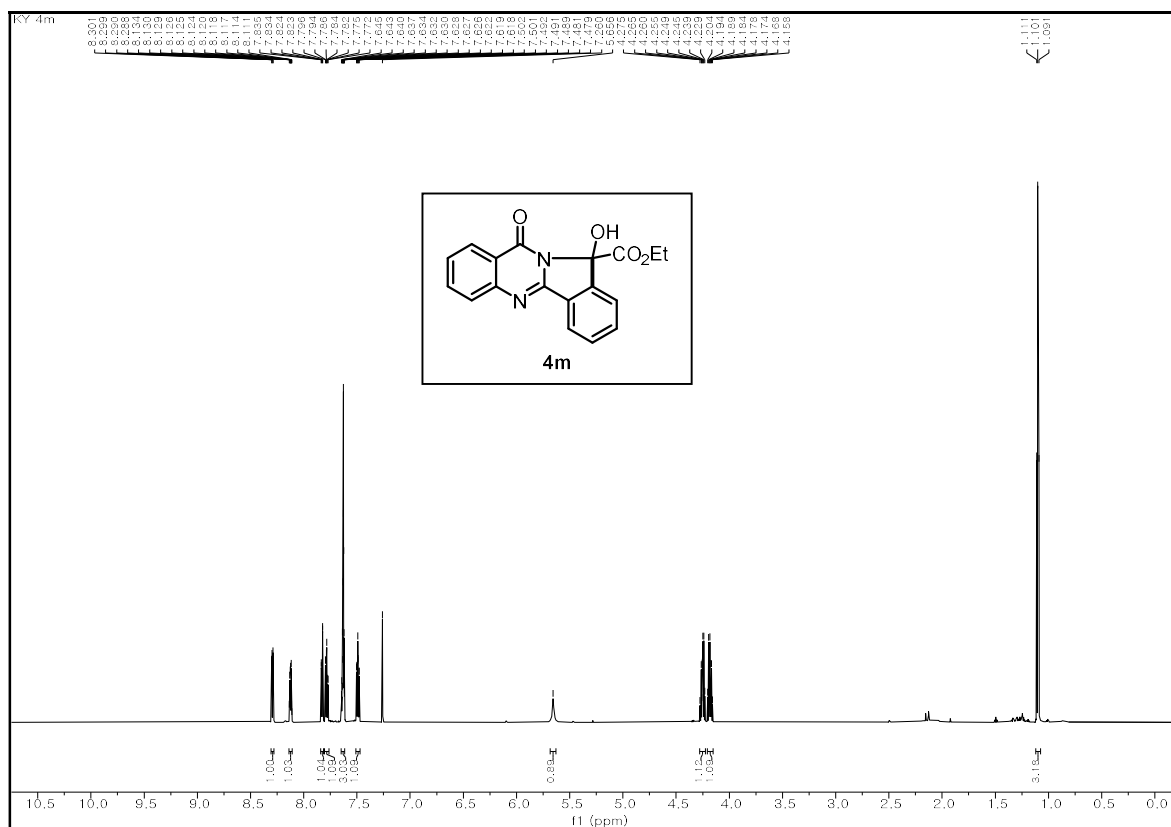
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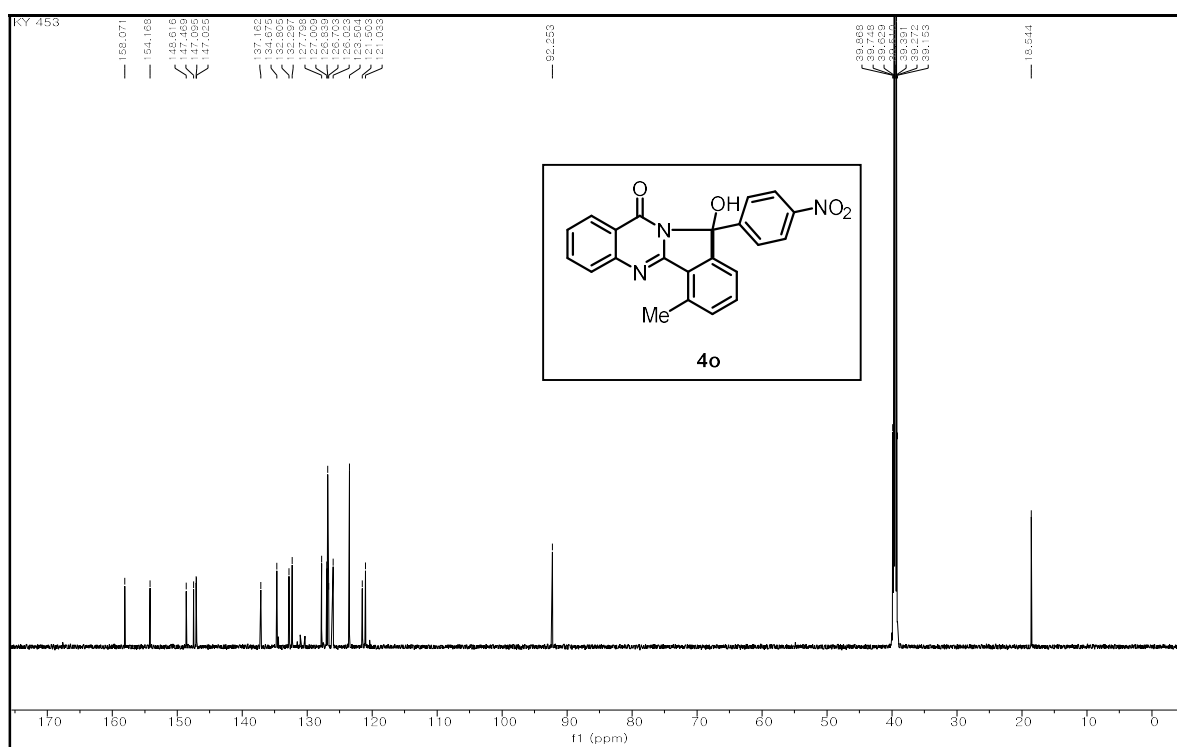
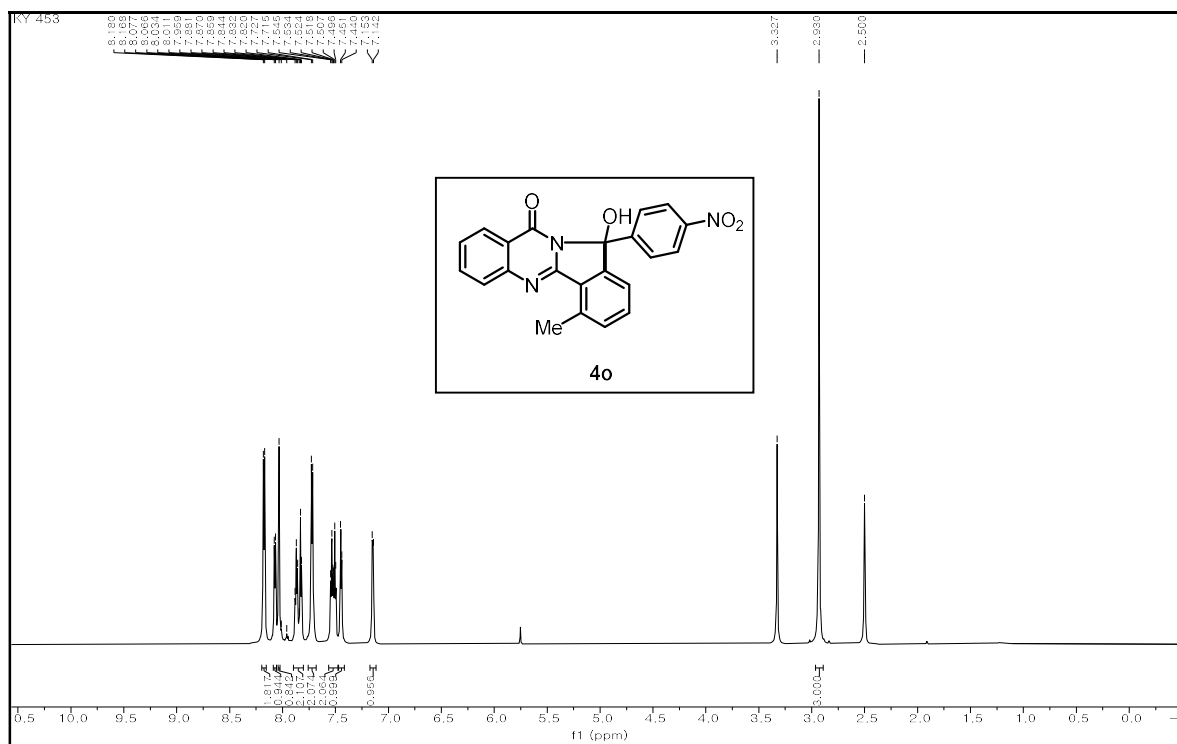


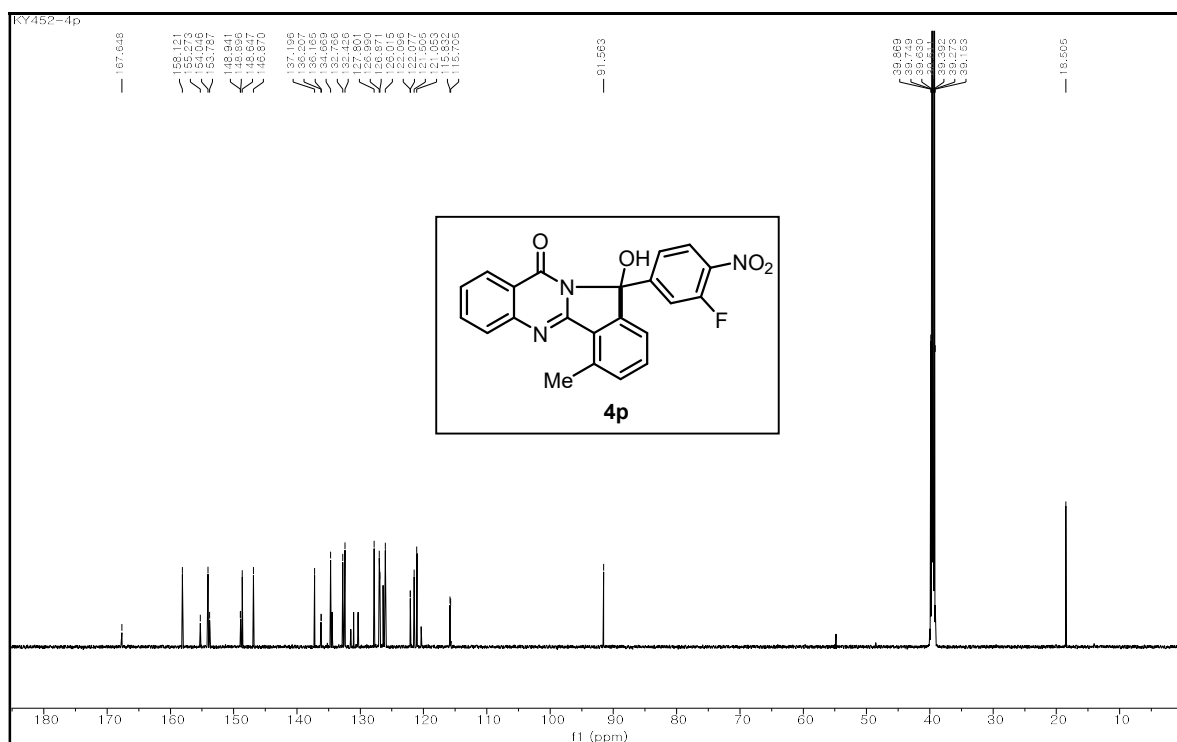
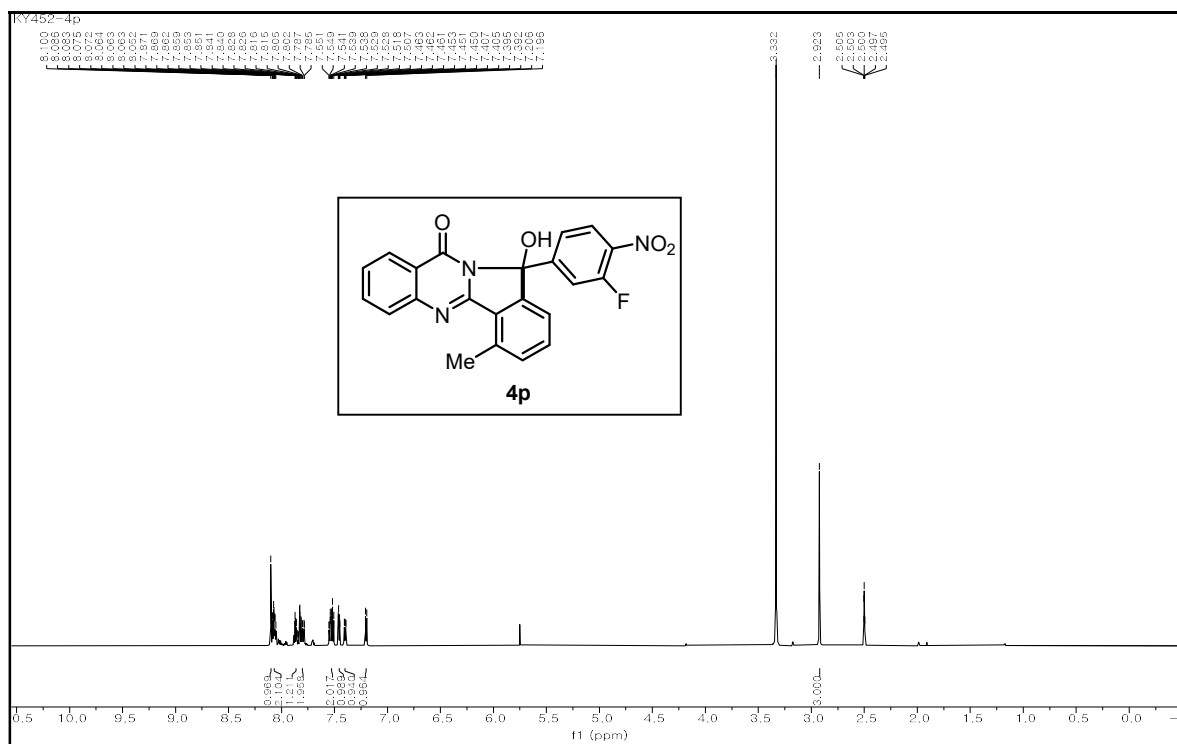
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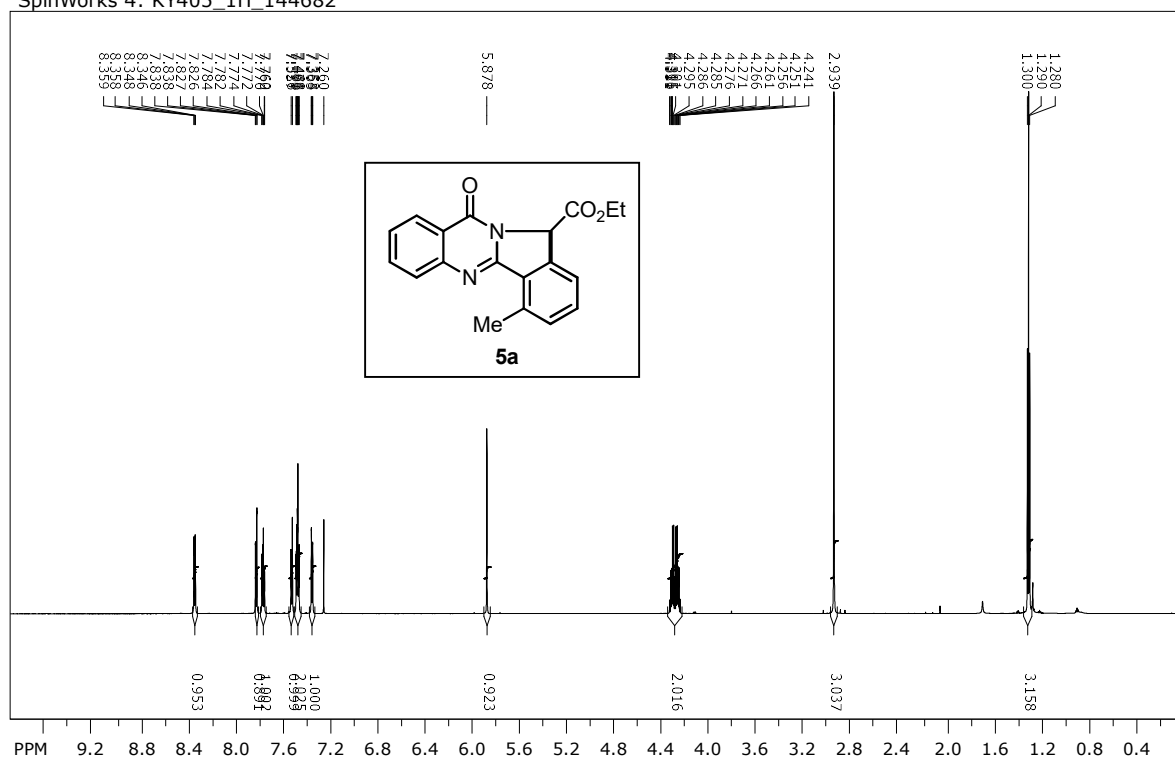




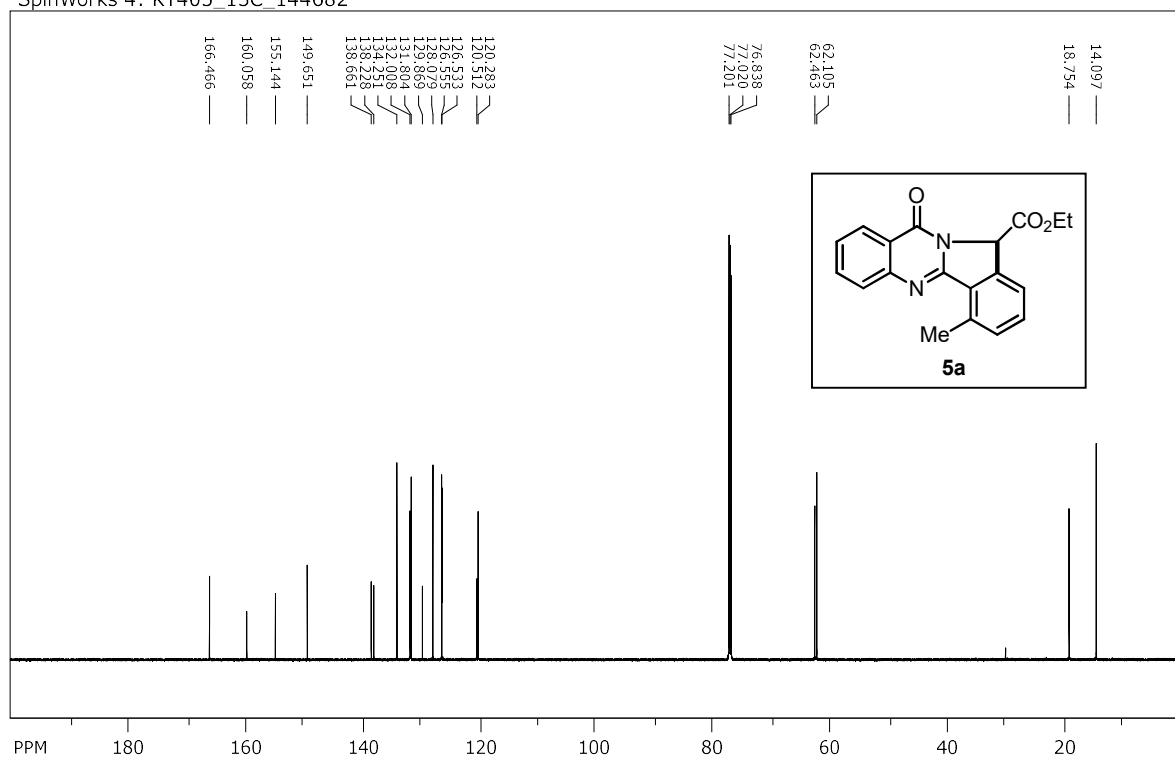




SpinWorks 4: KY405_1H_144682



SpinWorks 4: KY405_13C_144682



Chemical Shift (ppm)	Integration
1.153, 1.140, 1.126, 1.120, 1.100, 1.082, 1.062, 1.042, 1.022, 1.002, 0.982, 0.962, 0.942, 0.922, 0.902, 0.882, 0.862, 0.842, 0.822, 0.802, 0.782, 0.762, 0.742, 0.722, 0.702, 0.682, 0.662, 0.642, 0.622, 0.602, 0.582, 0.562, 0.542, 0.522, 0.502, 0.482, 0.462, 0.442, 0.422, 0.402, 0.382, 0.362, 0.342, 0.322, 0.302, 0.282, 0.262, 0.242, 0.222, 0.202, 0.182, 0.162, 0.142, 0.122, 0.102, 0.082, 0.062, 0.042, 0.022, 0.002	3.147, 3.129
2.144, 2.043	2.144, 2.043
5.996	1.016
7.260	0.971

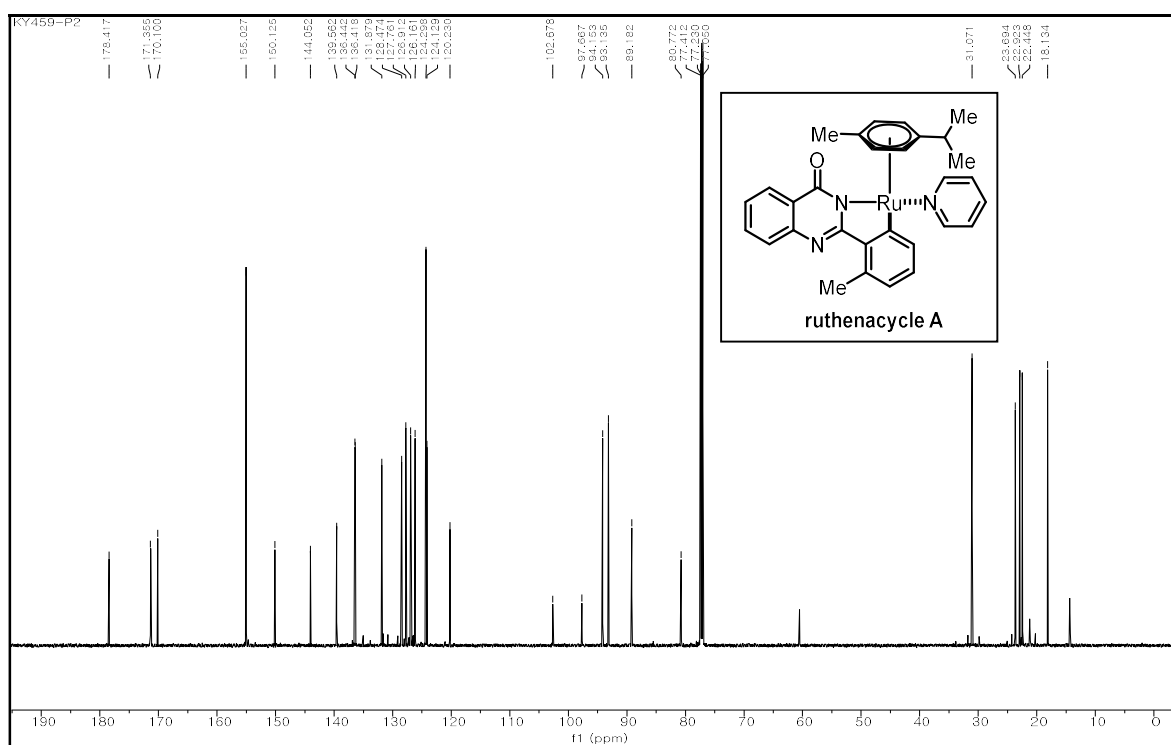
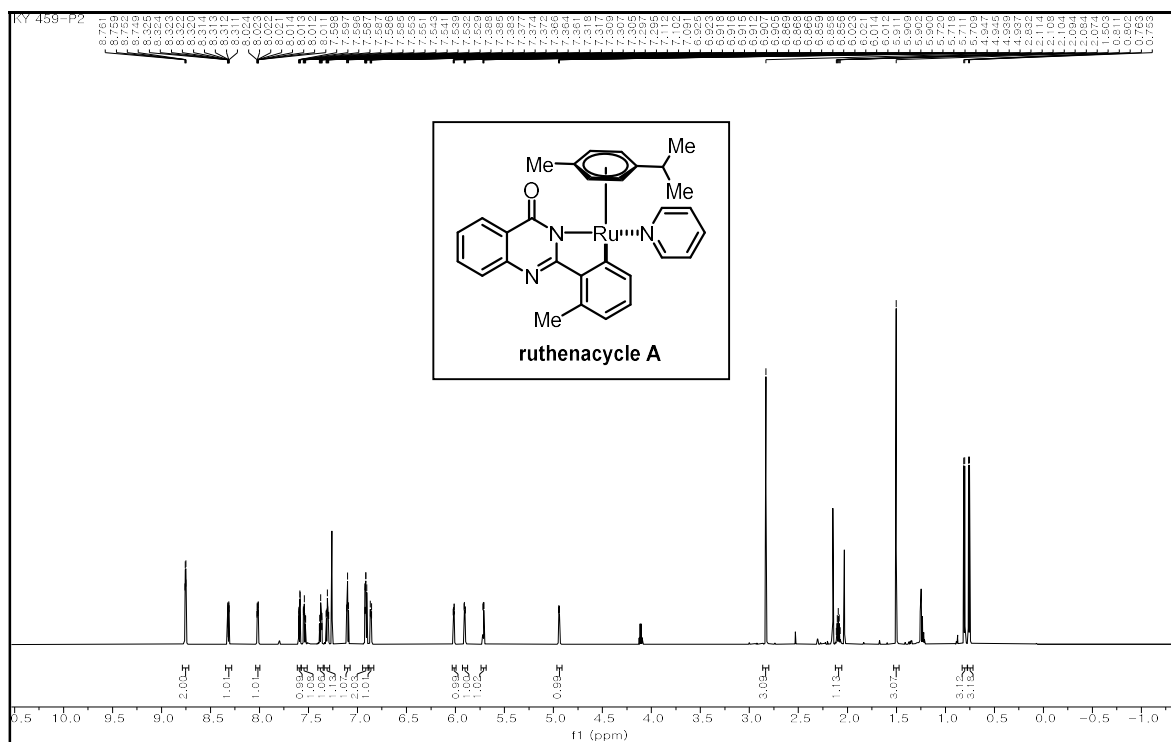
Spinworks 4.

Chemical structure of **5b** is shown in the top right corner. The structure is a tricyclic compound consisting of a benzene ring fused to a pyridine ring, which is further fused to a benzene ring. The pyridine ring has a carbonyl group (C=O) at position 2 and a hydroxyl group (OH) at position 4. The benzene ring fused to the pyridine ring has an ethyl ester group (CO₂Et) at position 1 and a hydroxyl group (OH) at position 3.

13C NMR spectrum (CDCl₃) of compound **5b**. The x-axis is labeled PPM and ranges from 180 to 20. The spectrum shows several peaks in the aromatic/unsaturated region (110-145 ppm) and two aliphatic peaks (61.668 and 62.809 ppm). A chemical structure of **5b** is shown in the top right corner.

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¹⁹F NMR spectra of compounds 3p and 4p

