

Electronic Supporting Information

Metal-Free Benzoylation of Imidazoheterocycles by Oxidative Decarboxylation of Aryl glyoxylic Acids

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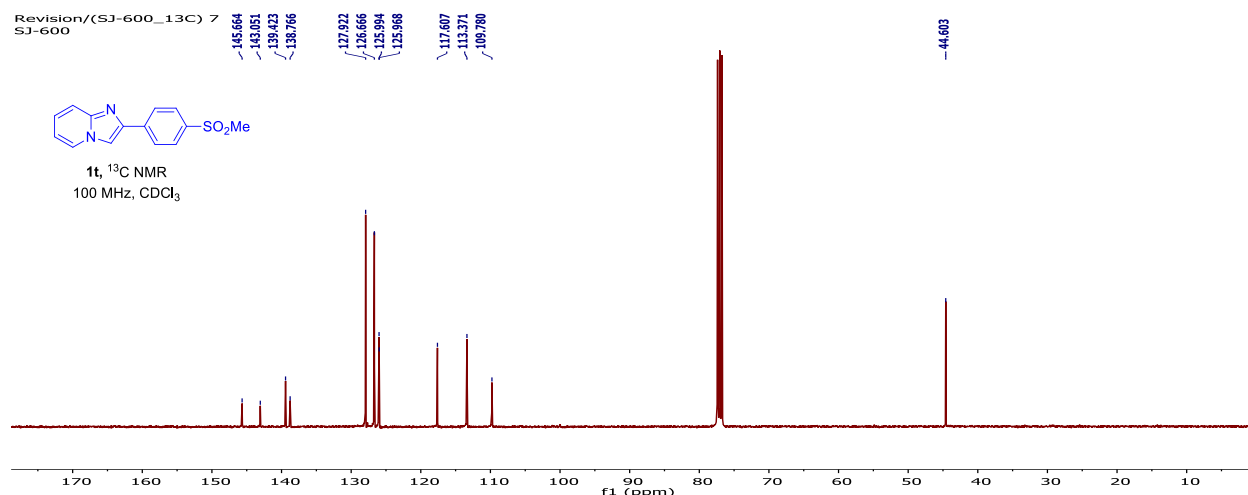
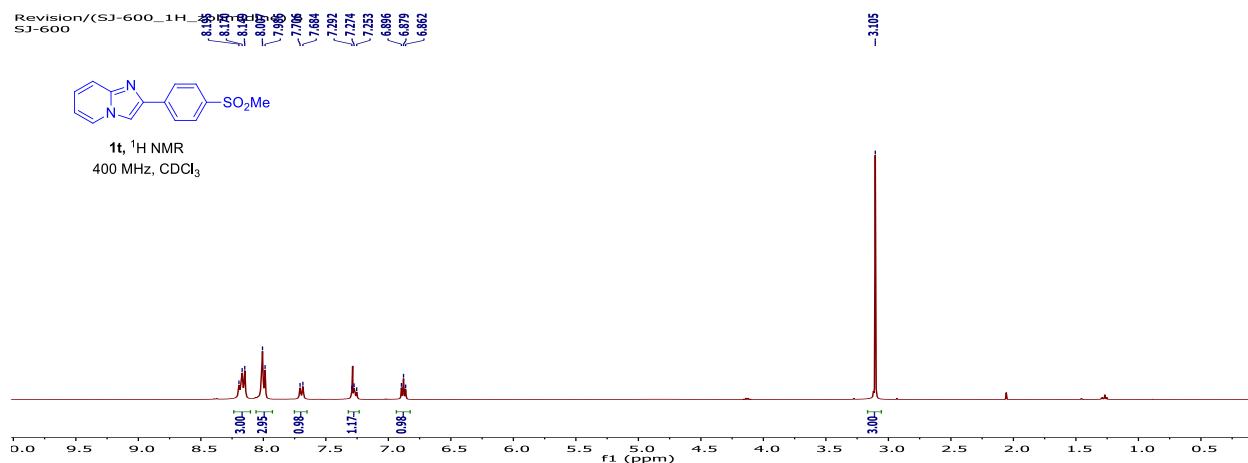
^bDepartment of Chemistry, Birla Institute of Technology and Science Pilani, Hyderabad Campus, Telangana, 500078, India.

E-mail: anilkumar@pilani.bits-pilani.ac.in (Anil Kumar)

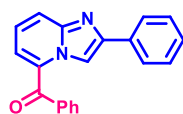
1. Copy of ¹H and ¹³C NMR for compound **1t**, **3**, **4**, **5** and **8**

(2-(4-(Methylsulfonyl)phenyl)imidazo[1,2-a]pyridin-5-yl)(phenyl)methanone (Zolimidine, **1t**)

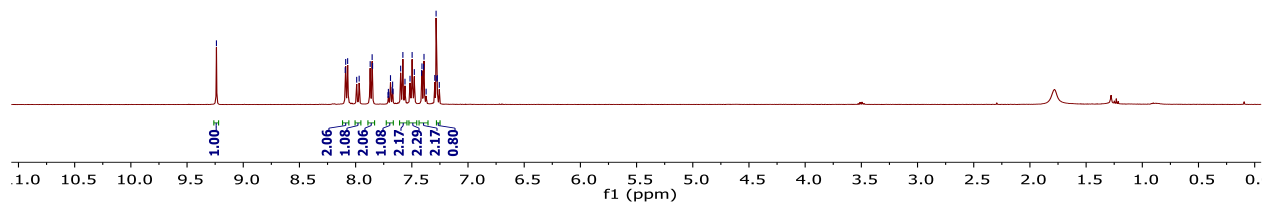
Eluent: *n*-Hexane; brown solid; yield: 200 mg (74 %); M. P. = 225-227 °C; FT-IR (neat) 2848, 2254, 1720, 1641, 1521, 1284, 727 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.20 – 8.15 (m, 3H), 8.00 (d, *J* = 8.4 Hz, 3H), 7.69 (d, *J* = 8.8 Hz, 1H), 7.29 – 7.25 (m, 1H) (with merged CDCl₃ peak), 6.88 (t, *J* = 6.8 Hz, 1H), 3.10 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 145.7, 143.0, 139.4, 138.8, 127.9, 126.7, 126.0, 126.0, 117.6, 113.4, 109.8, 44.6; HRMS (ESI) calcd for C₁₄H₁₃N₂O₂S⁺ [M + H]⁺ 273.0692, found 273.0690.



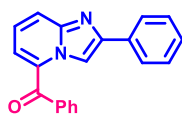
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SJ-280



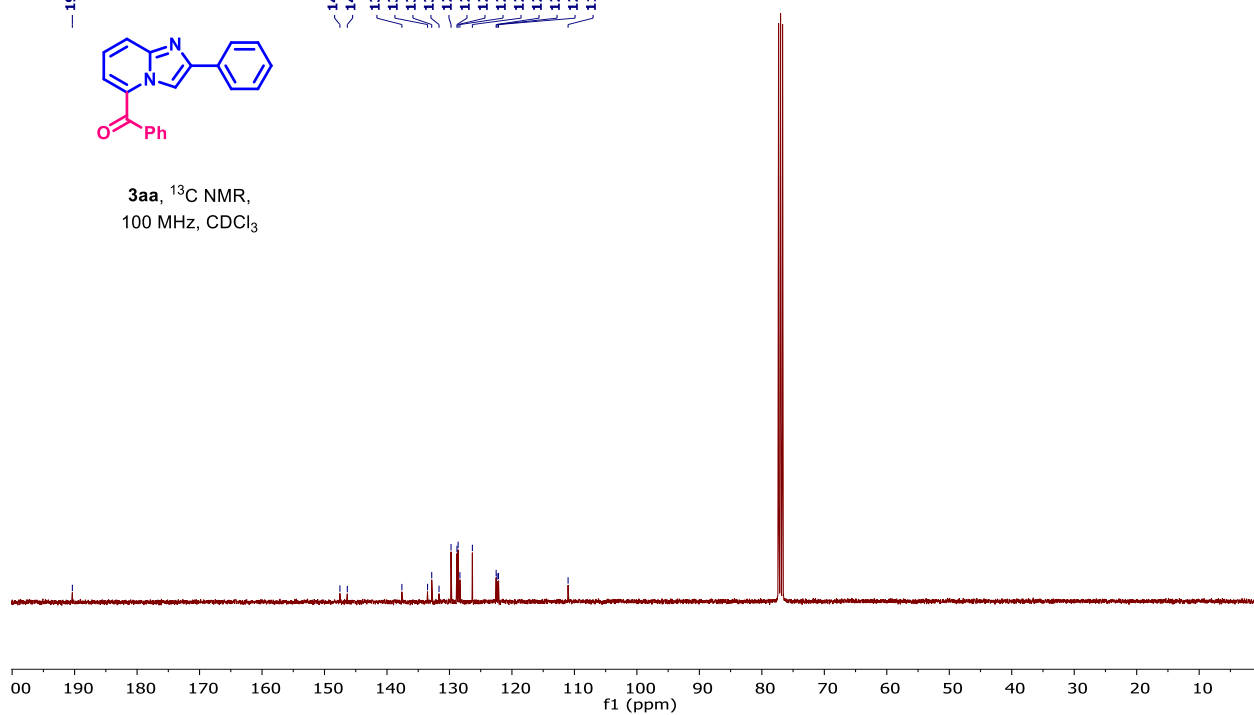
3aa, ^1H NMR,
400 MHz, CDCl_3



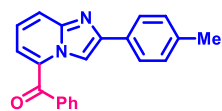
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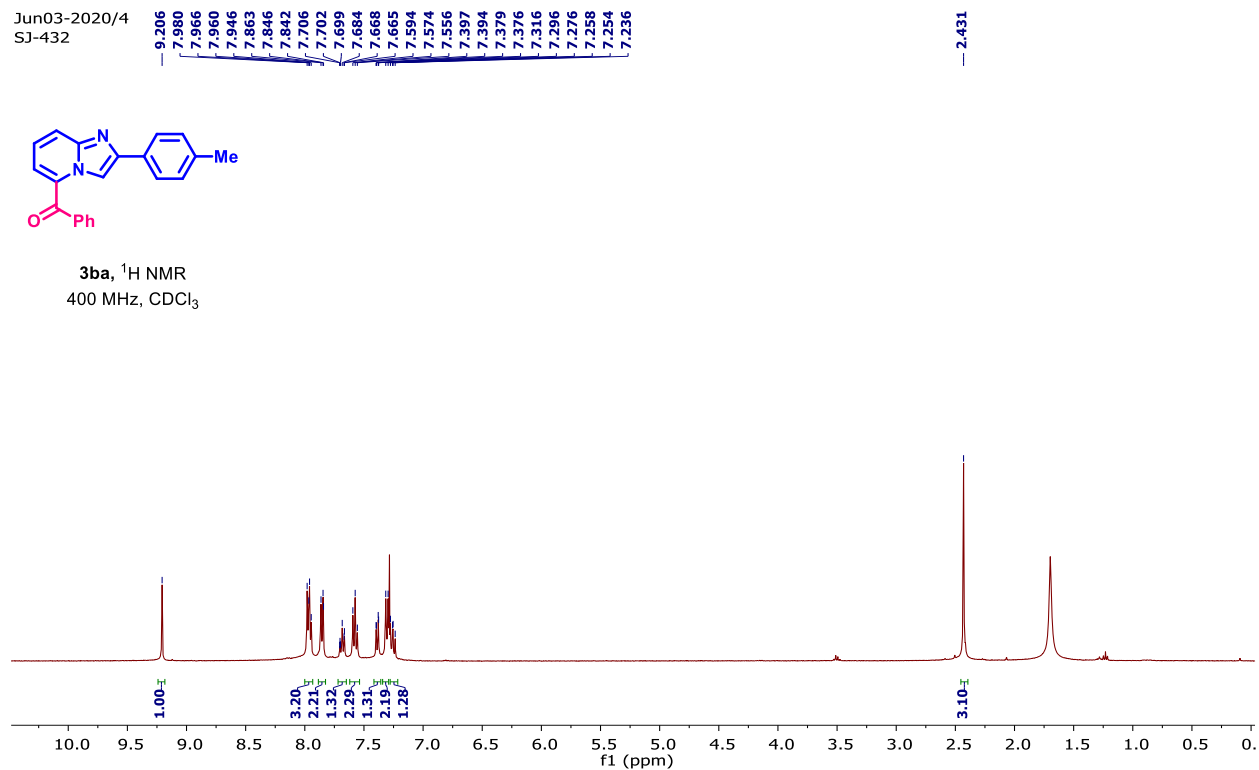
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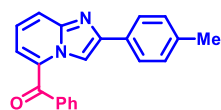
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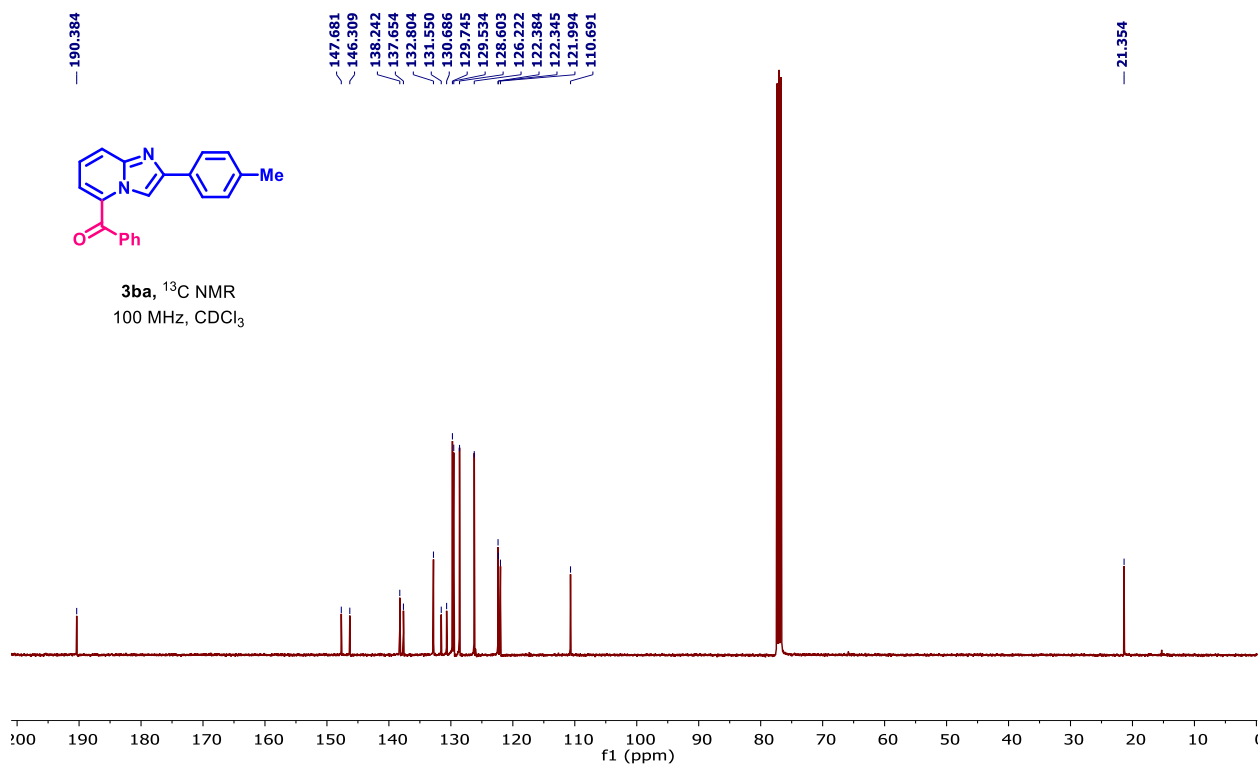
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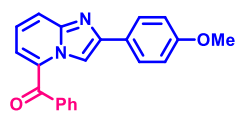
190.384



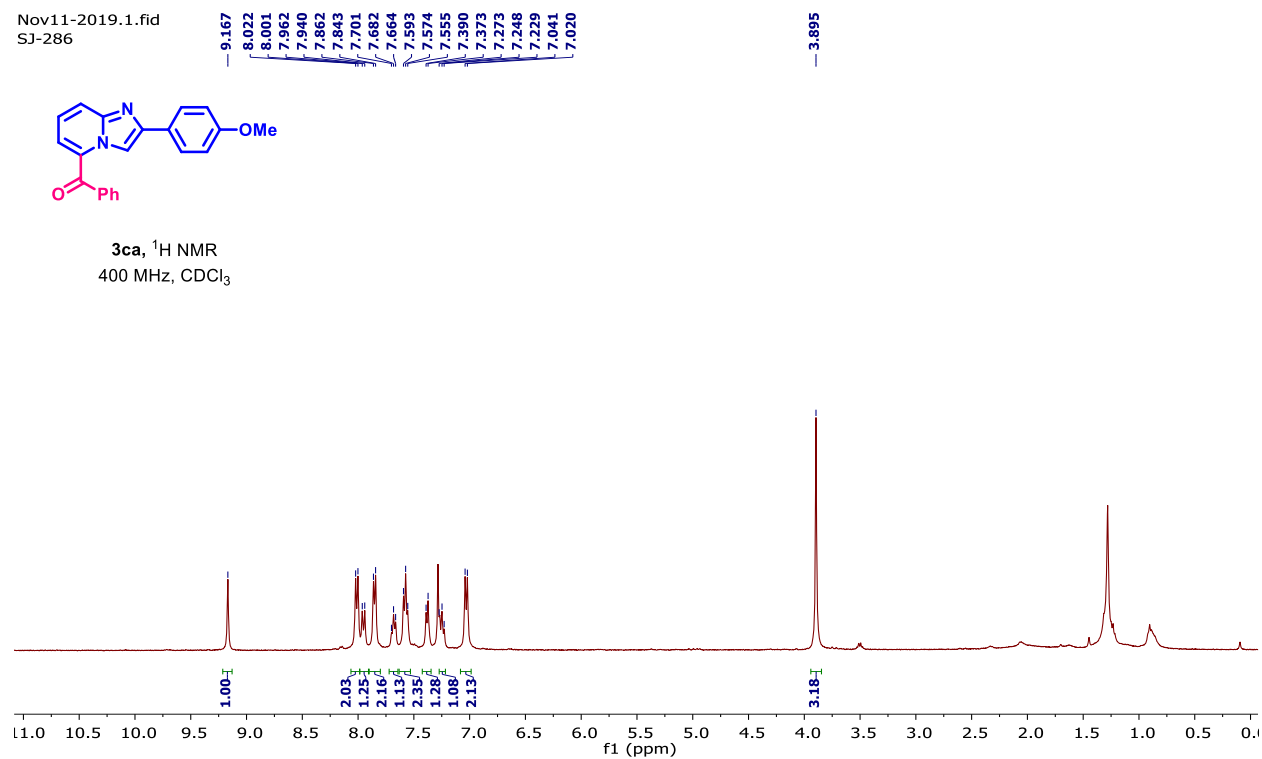
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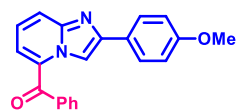
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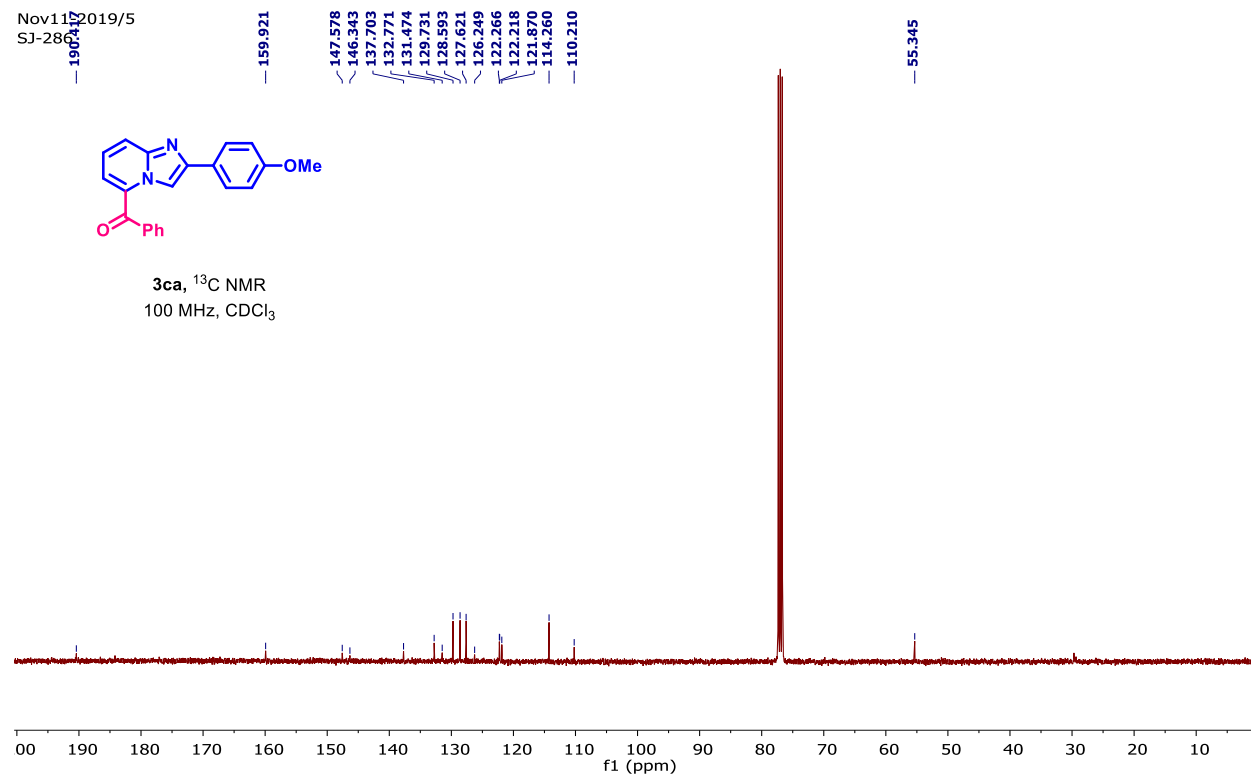
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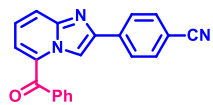
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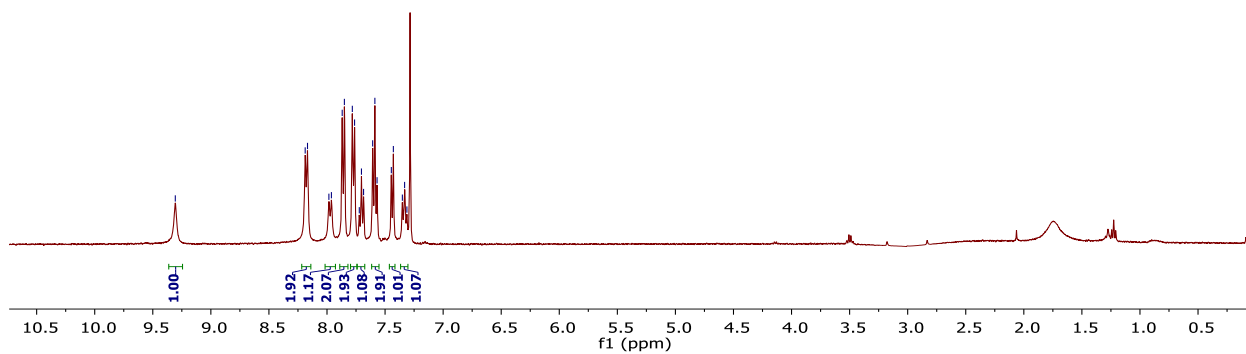
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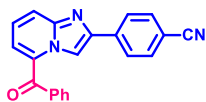
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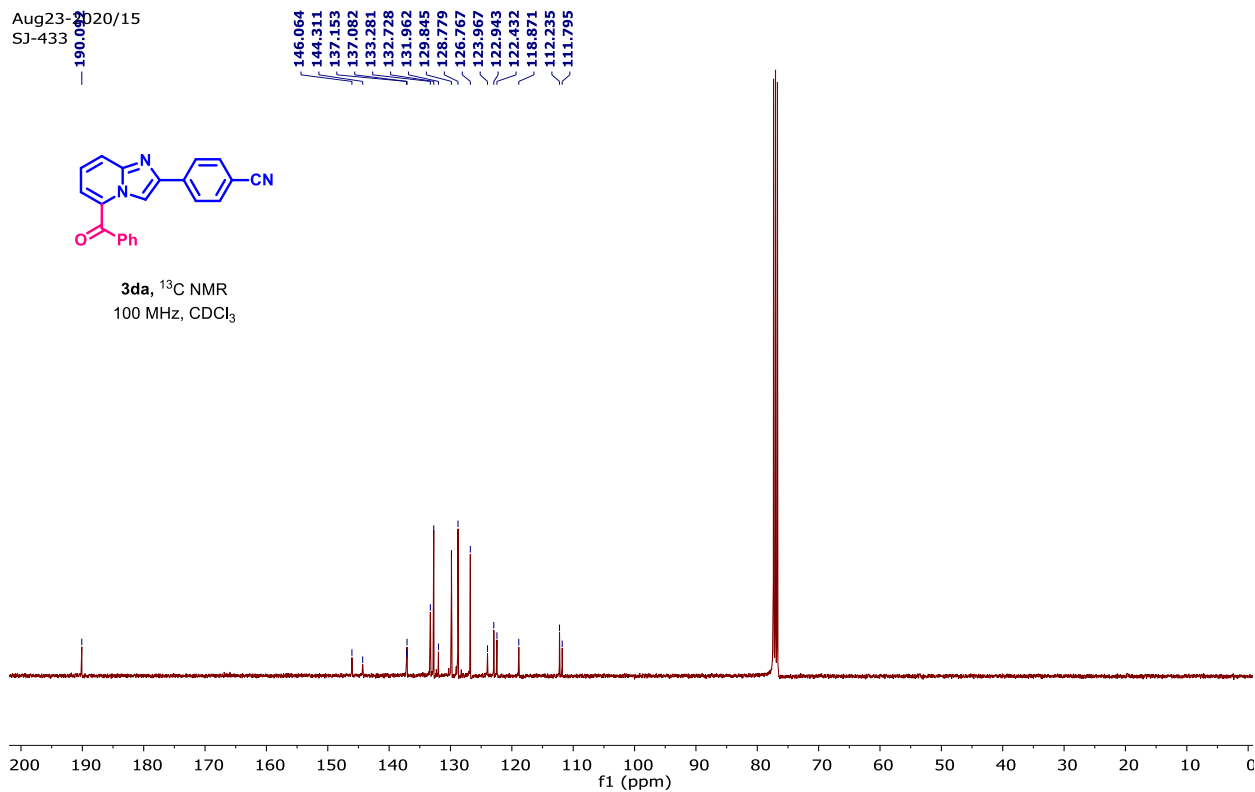
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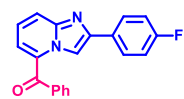
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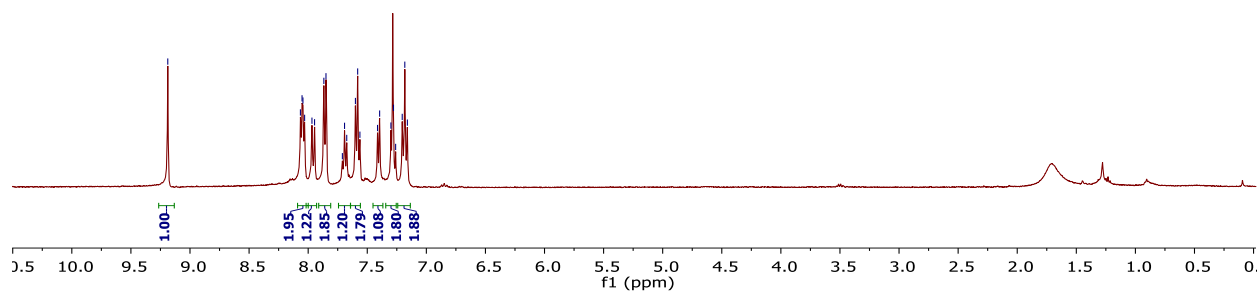
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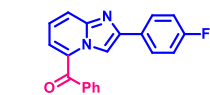
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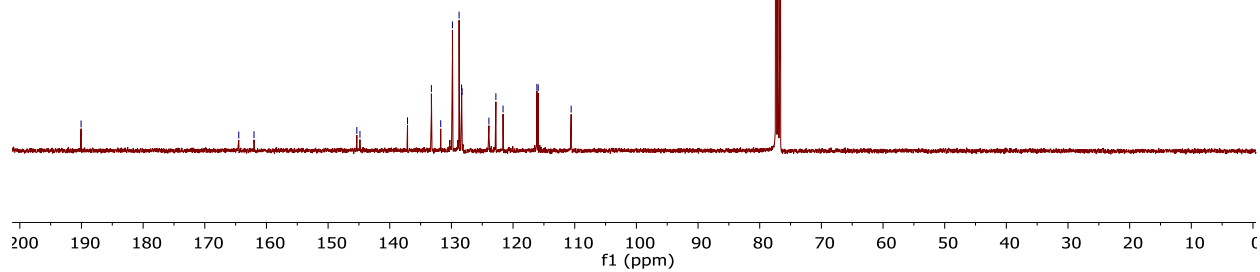
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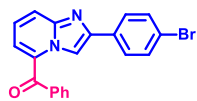
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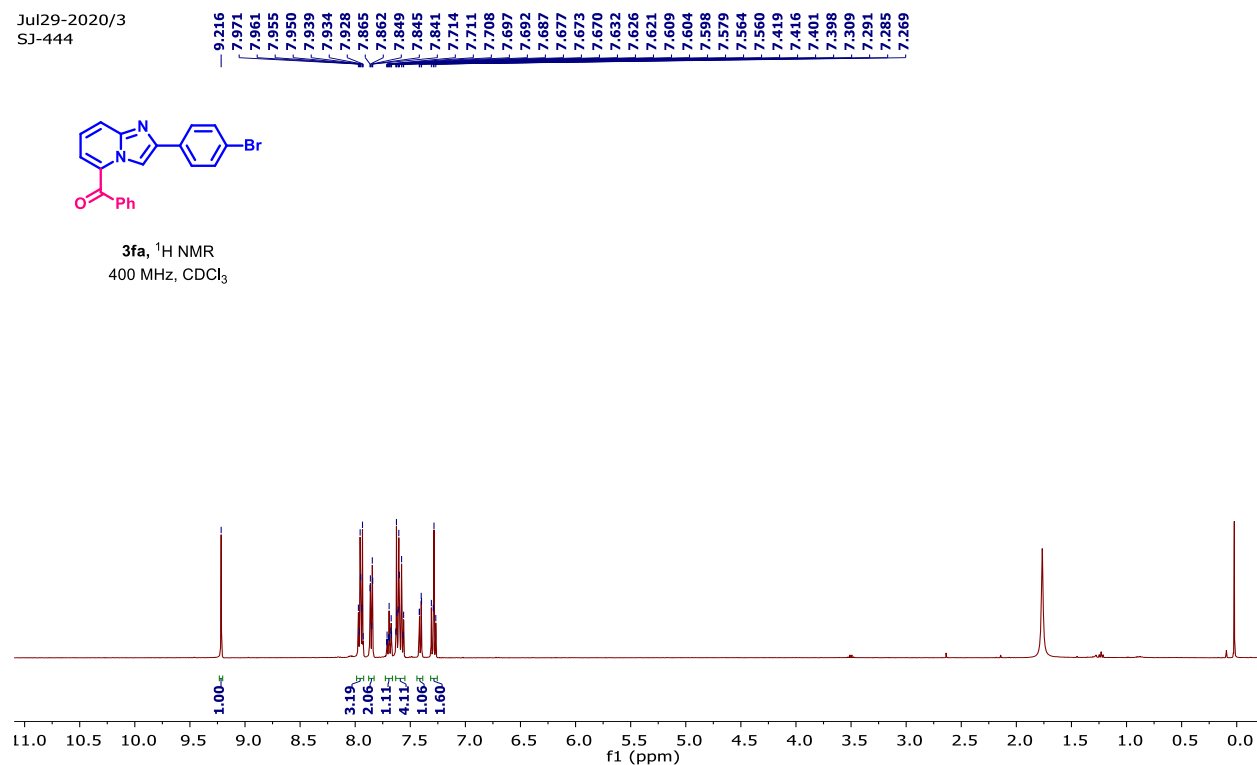
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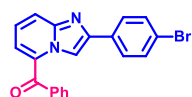
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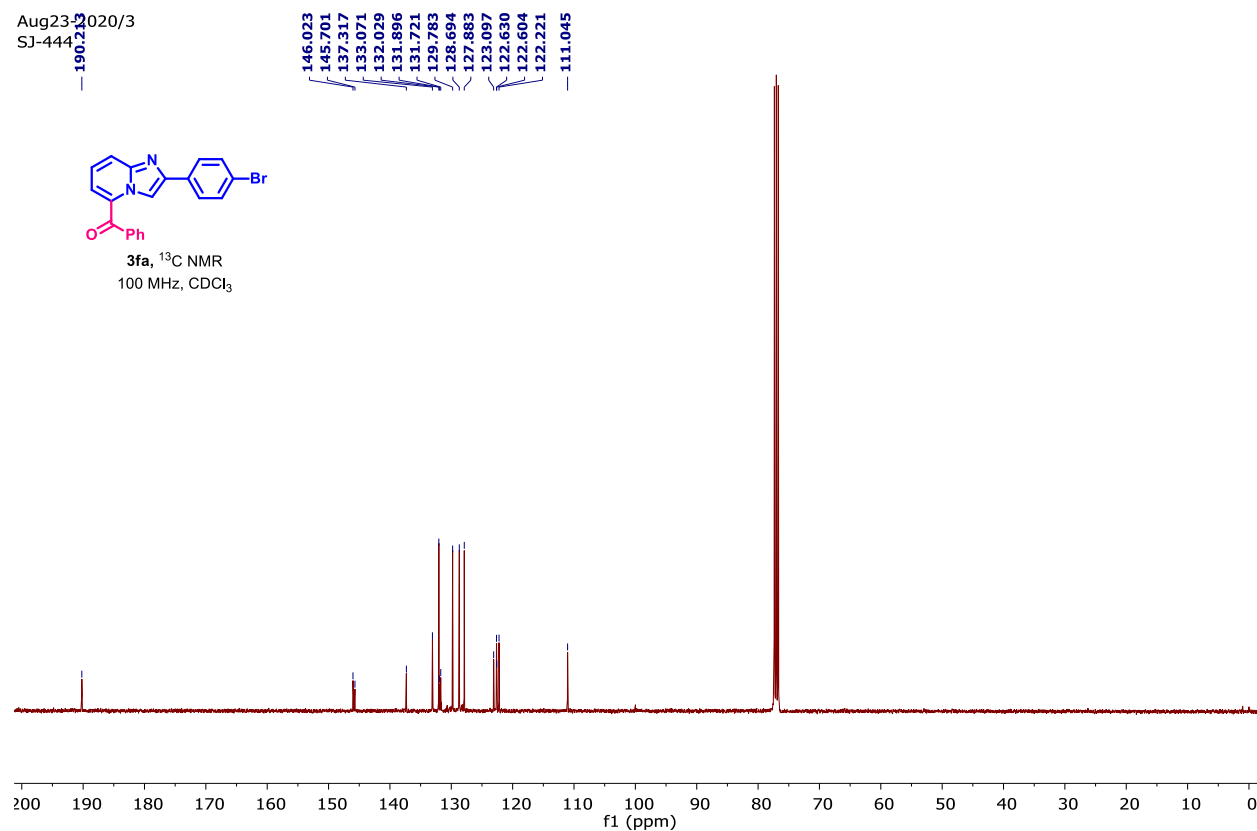
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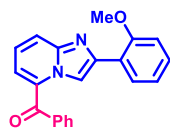
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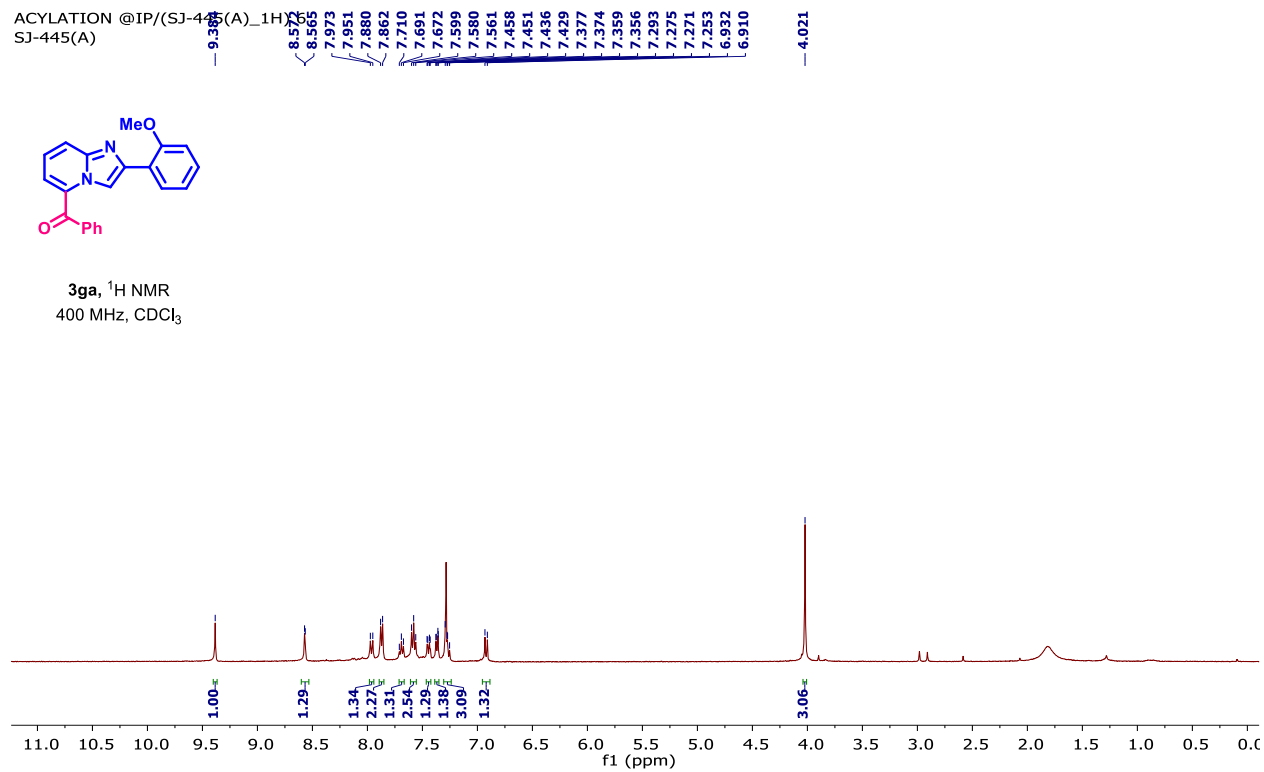
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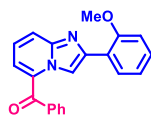
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SJ-445(A)



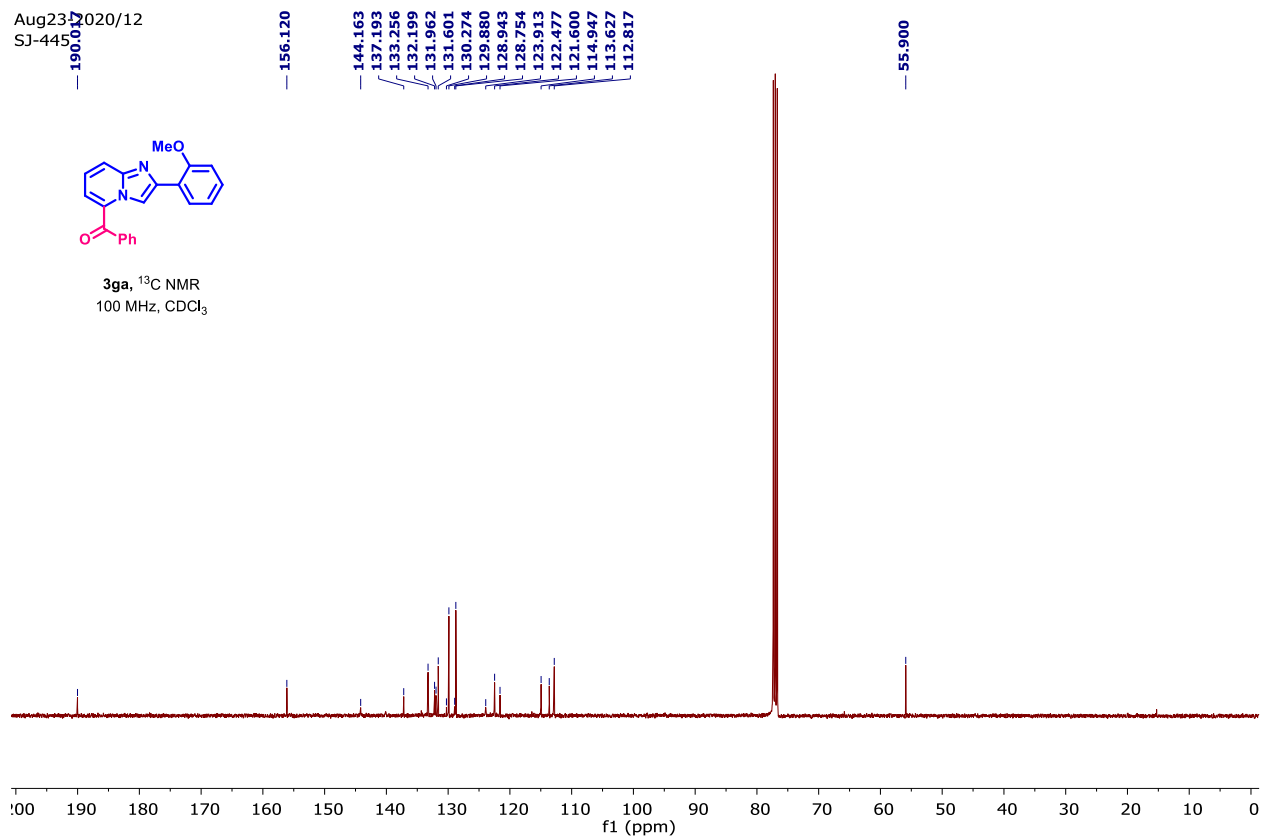
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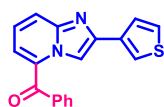
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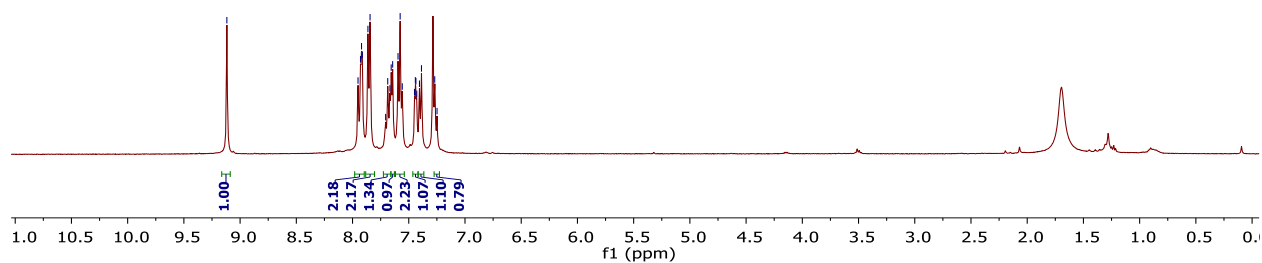
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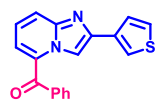
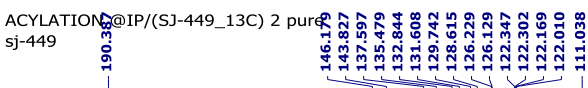
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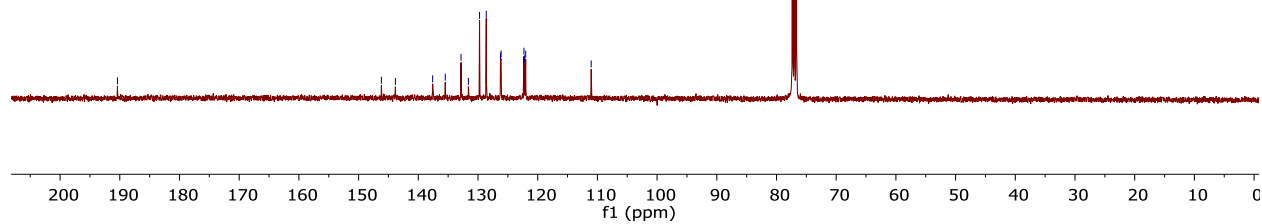
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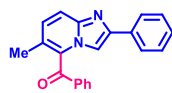
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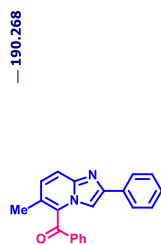
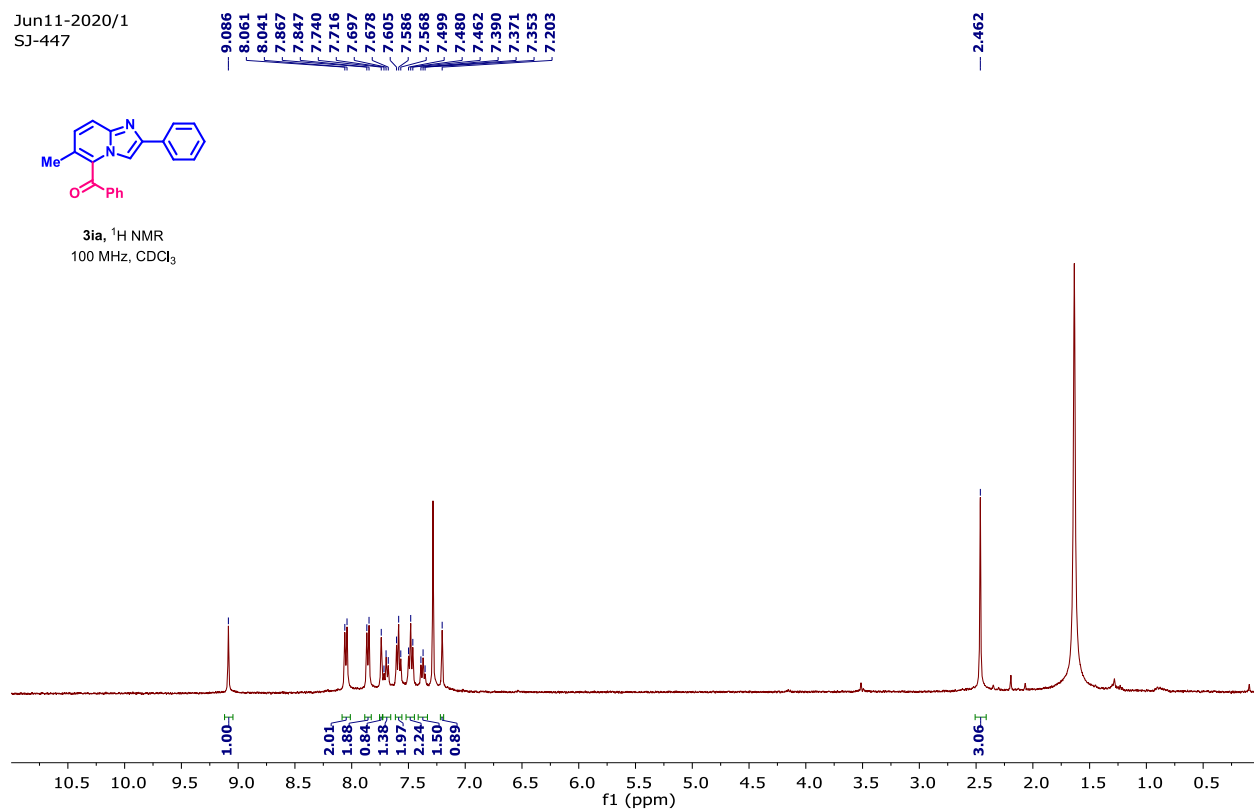
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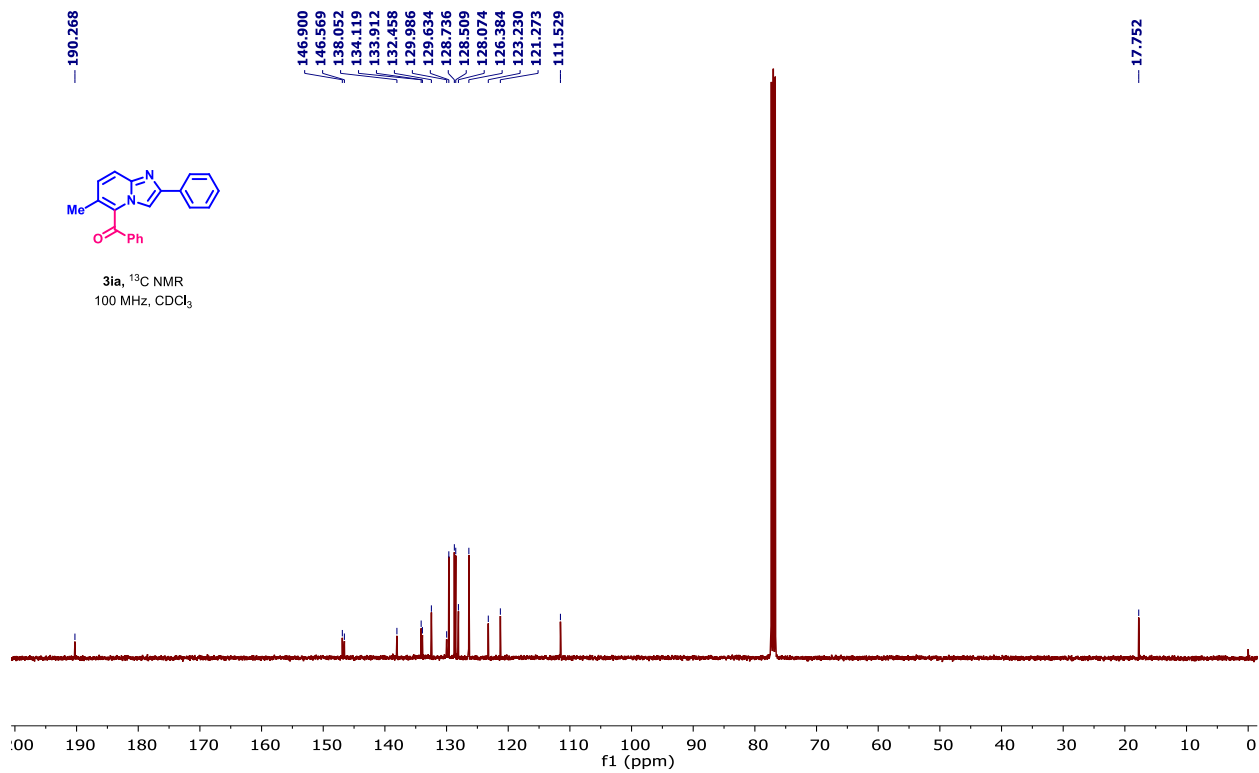
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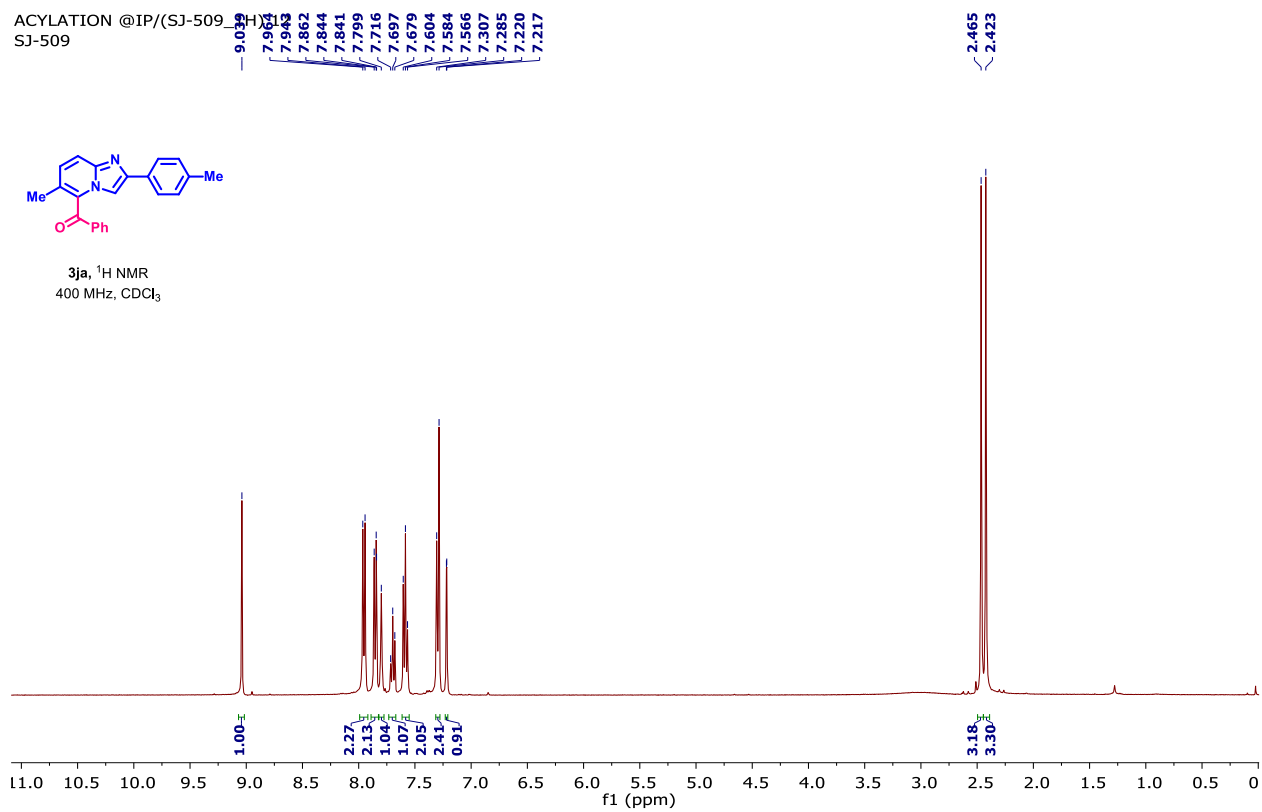
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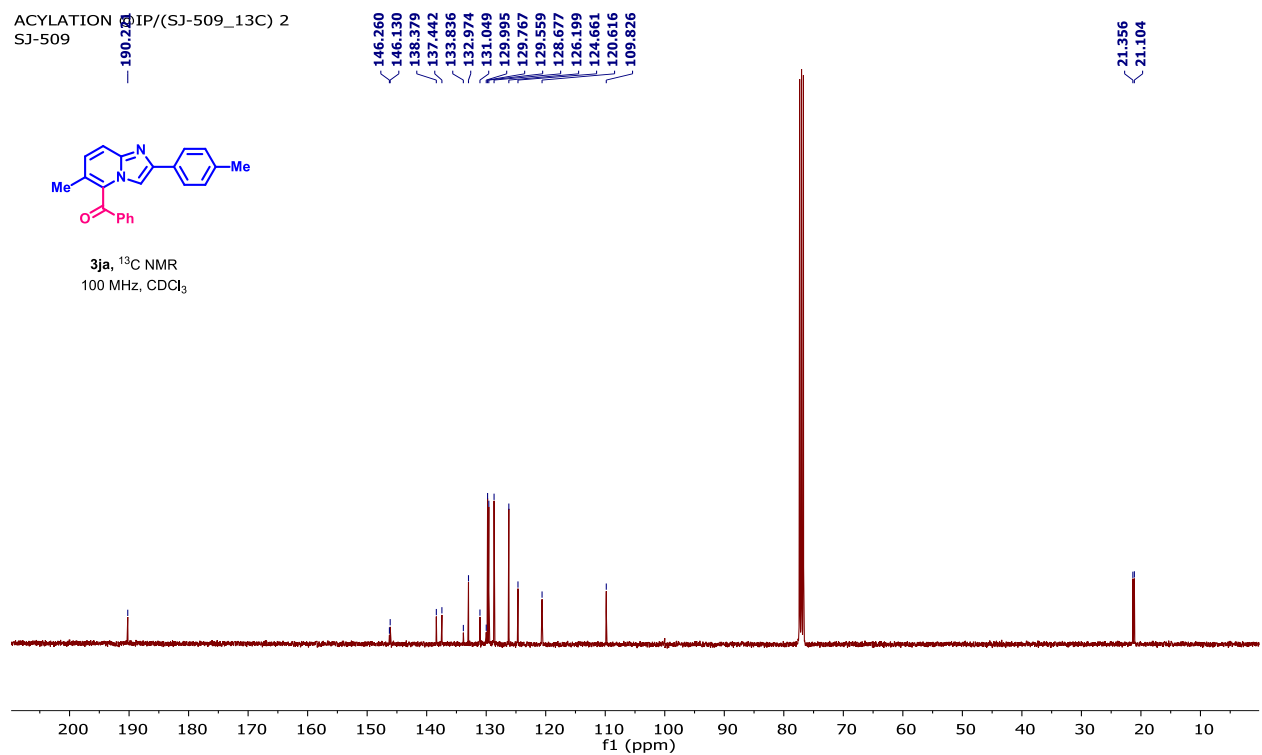
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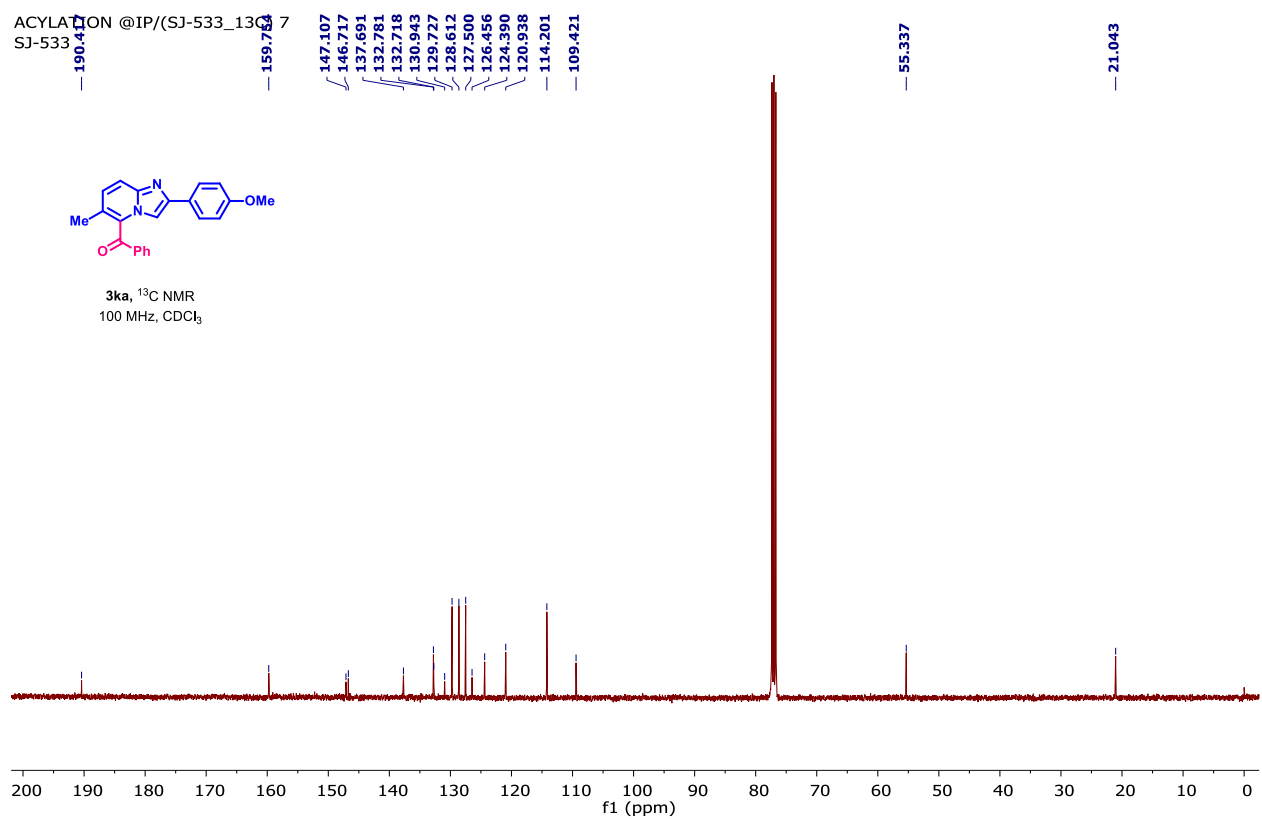
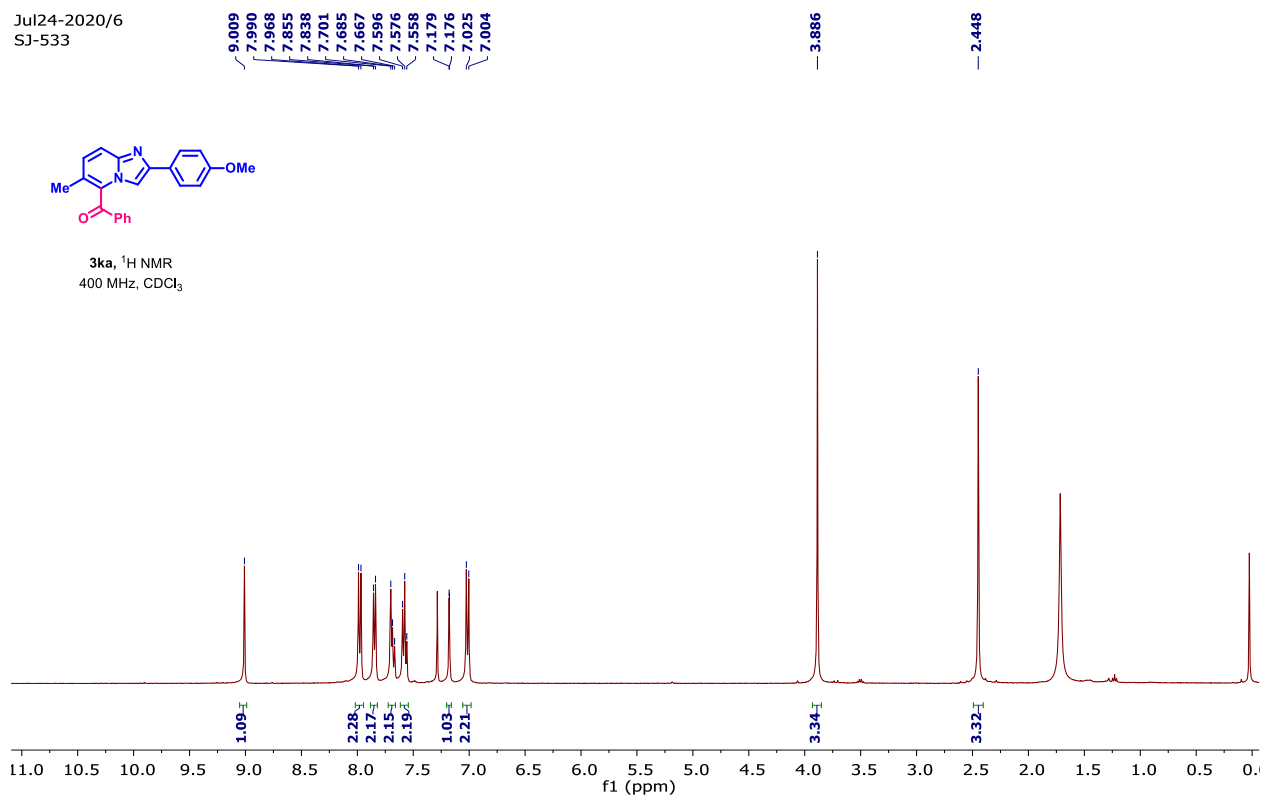
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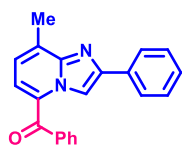
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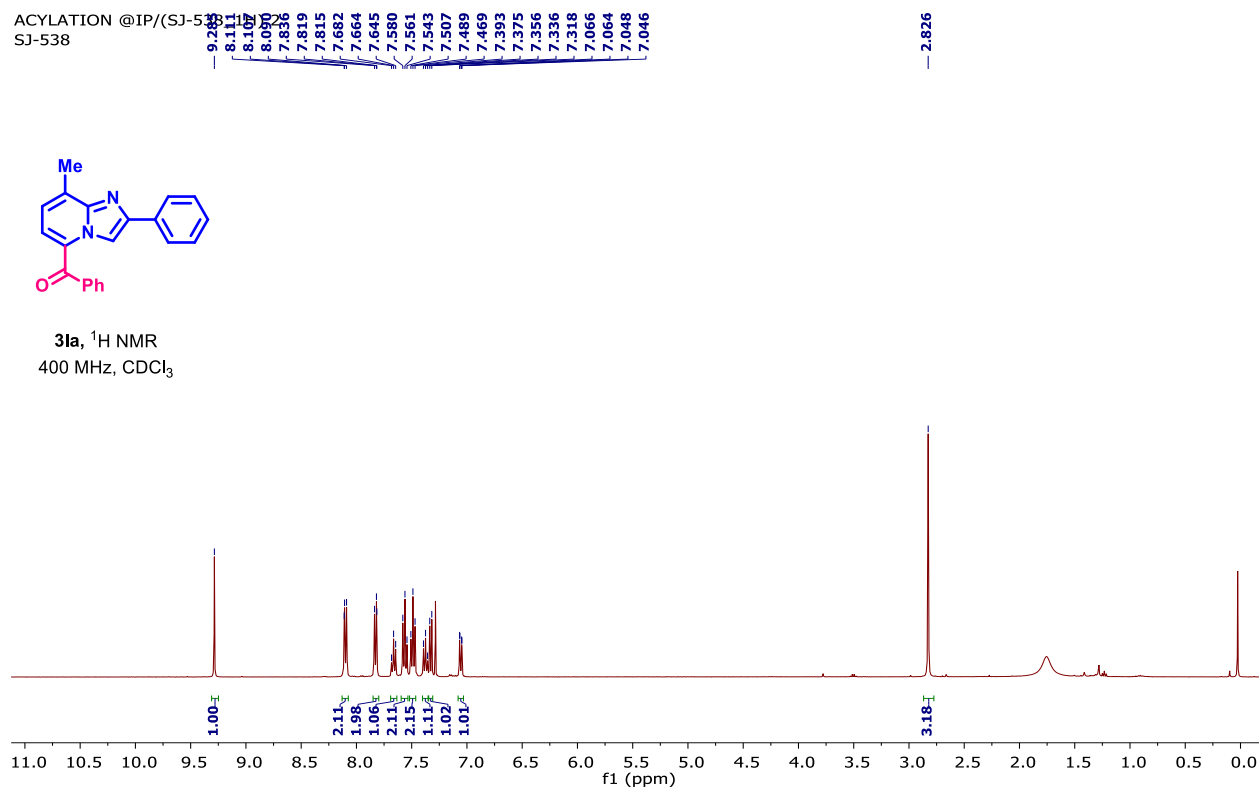
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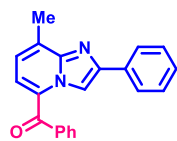
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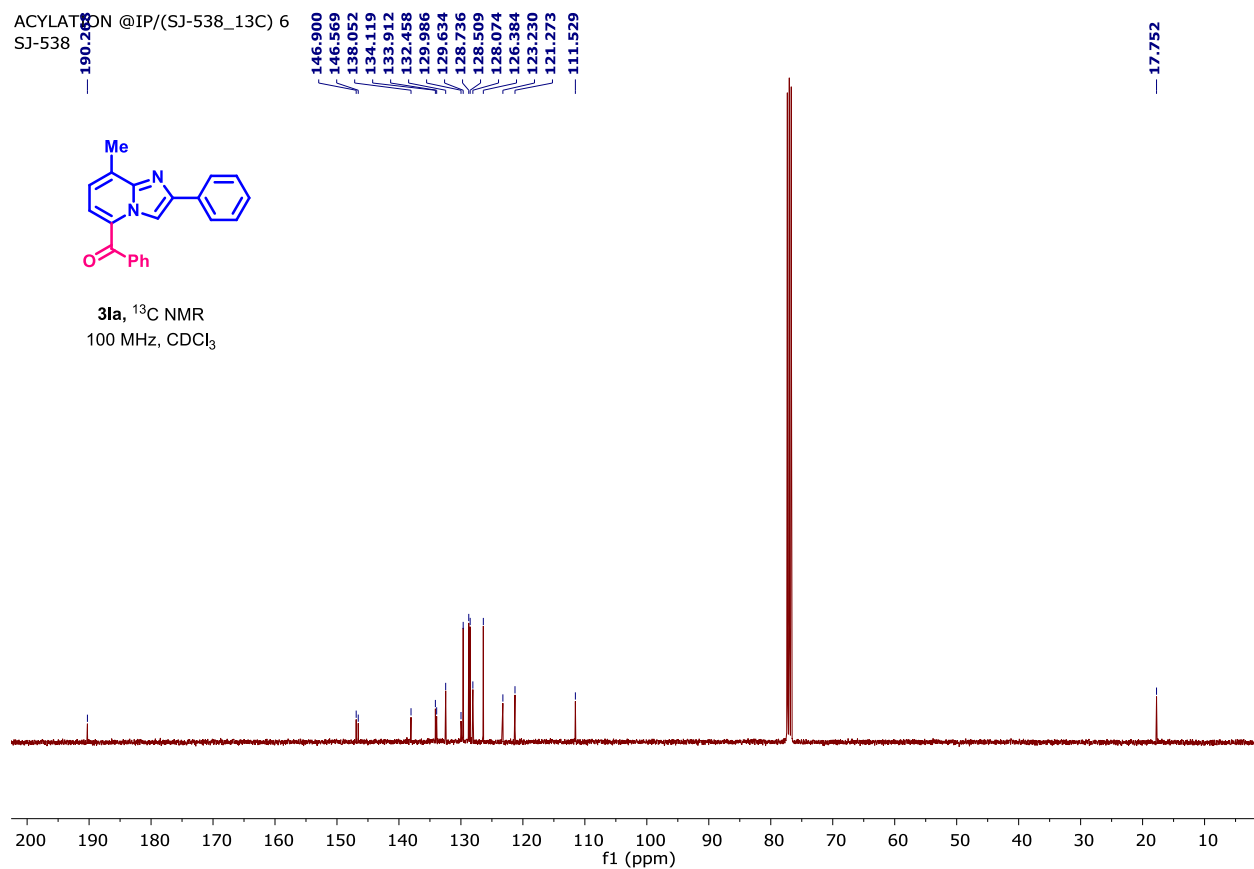
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400 MHz, CDCl_3



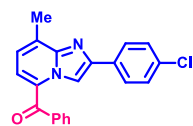
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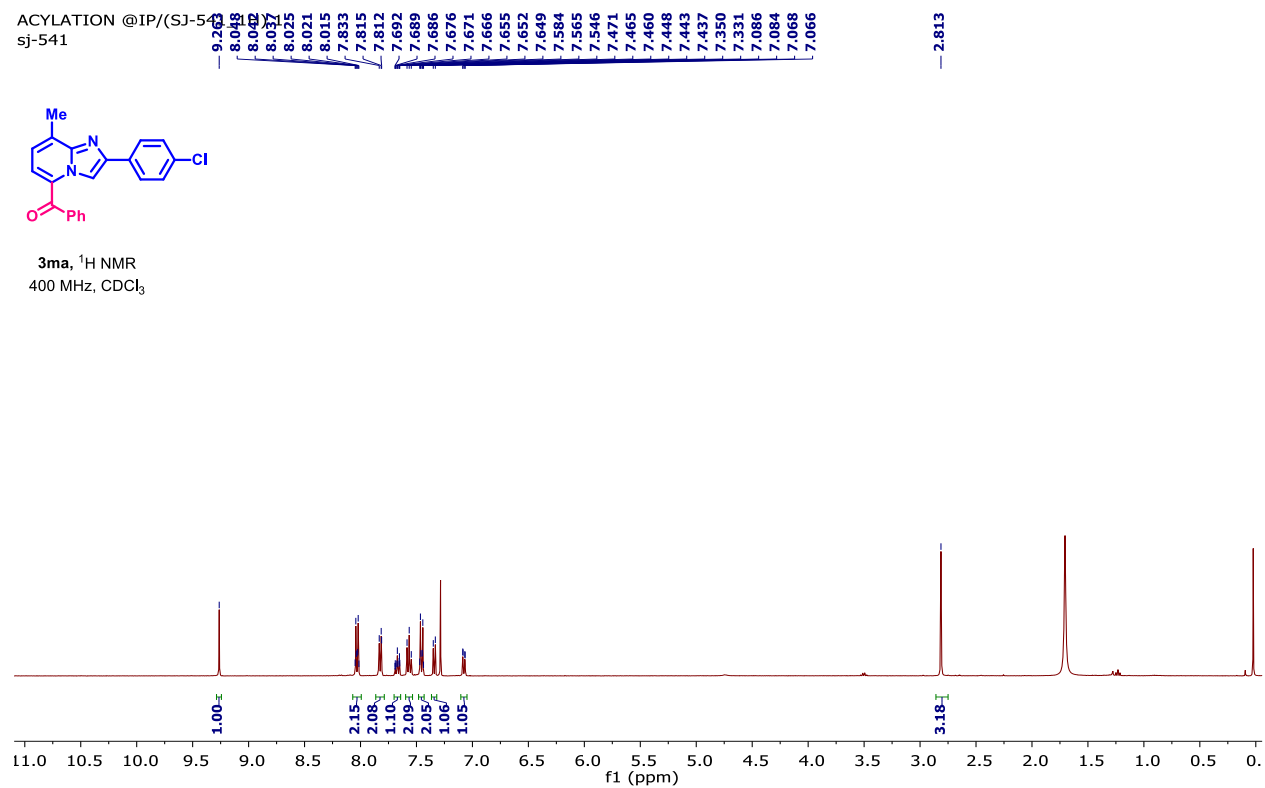
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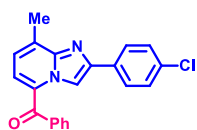
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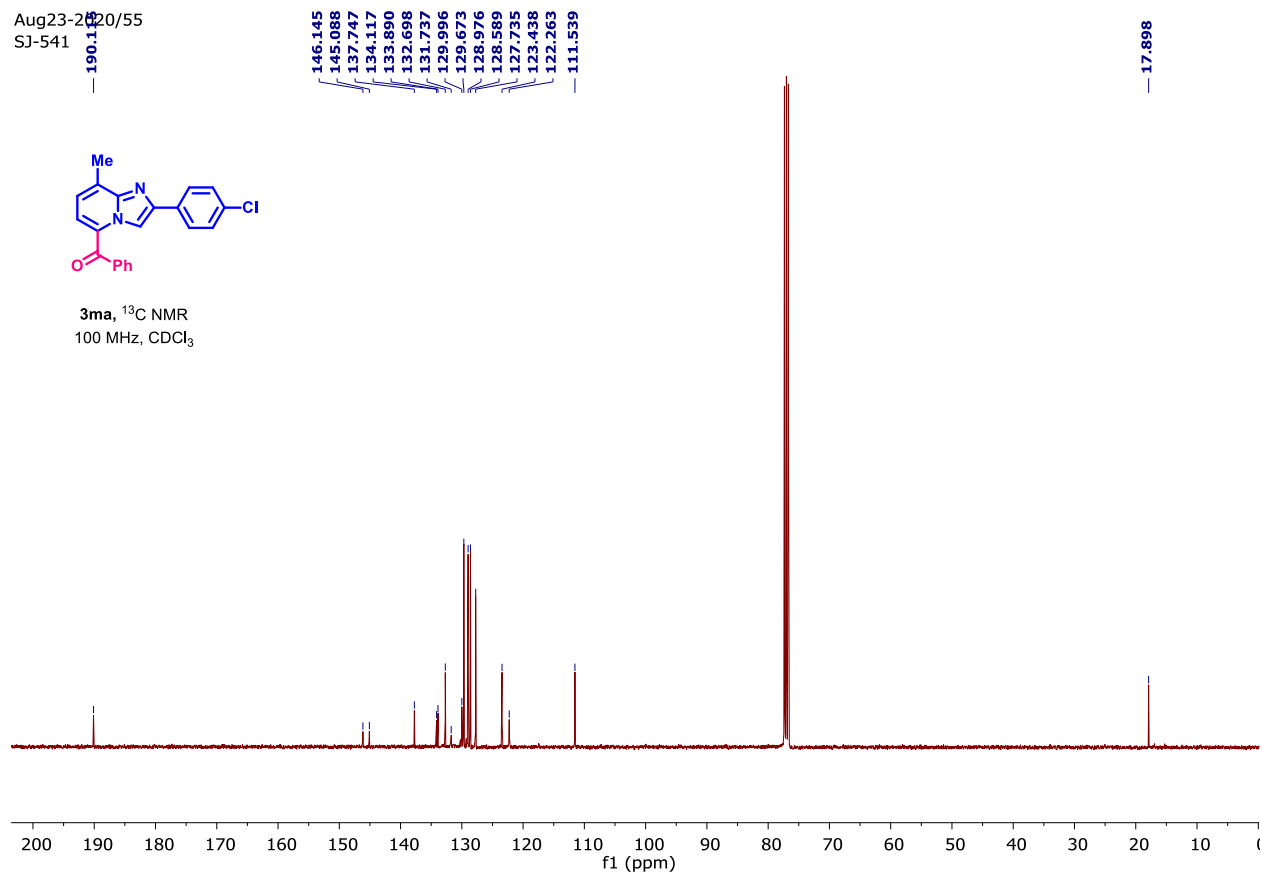
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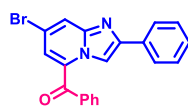
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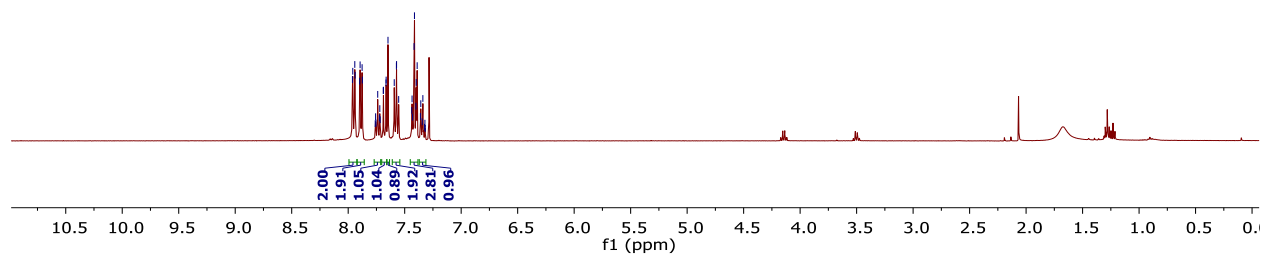
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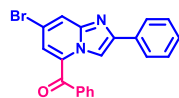
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SJ-478



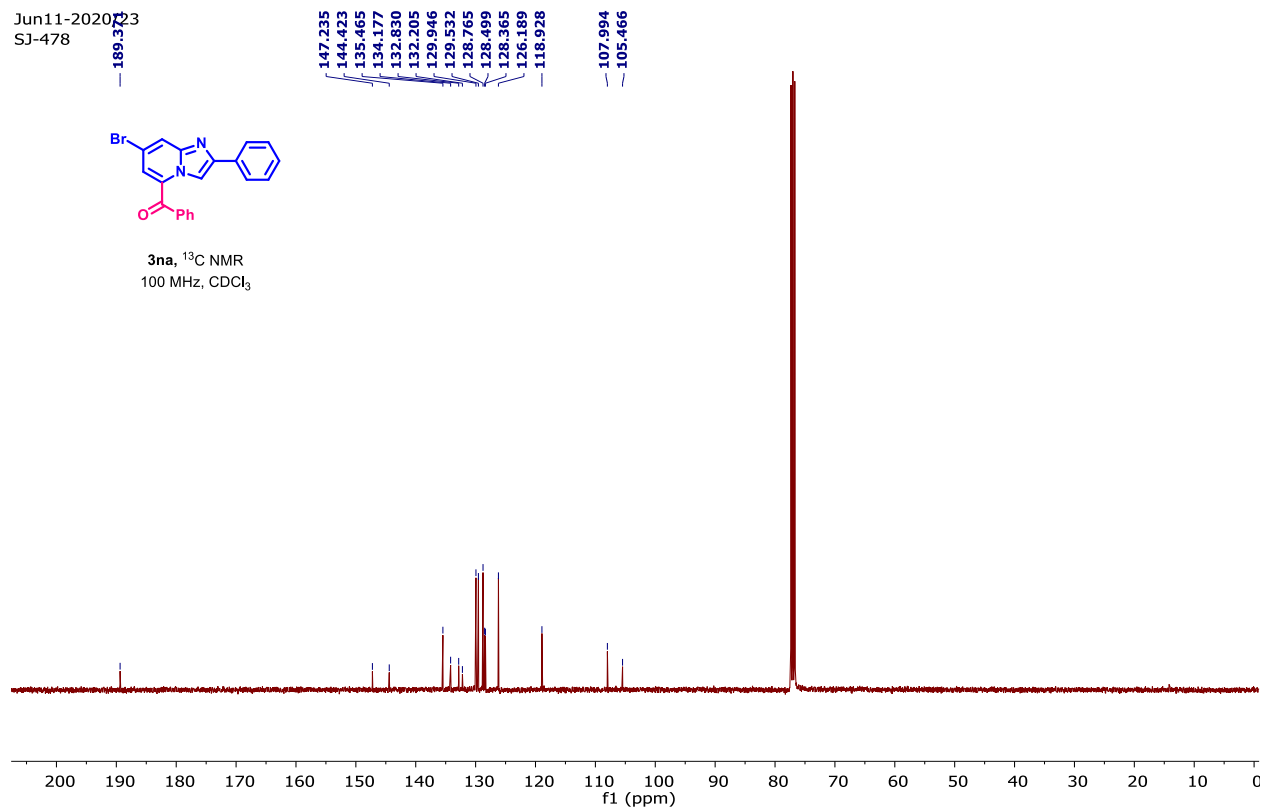
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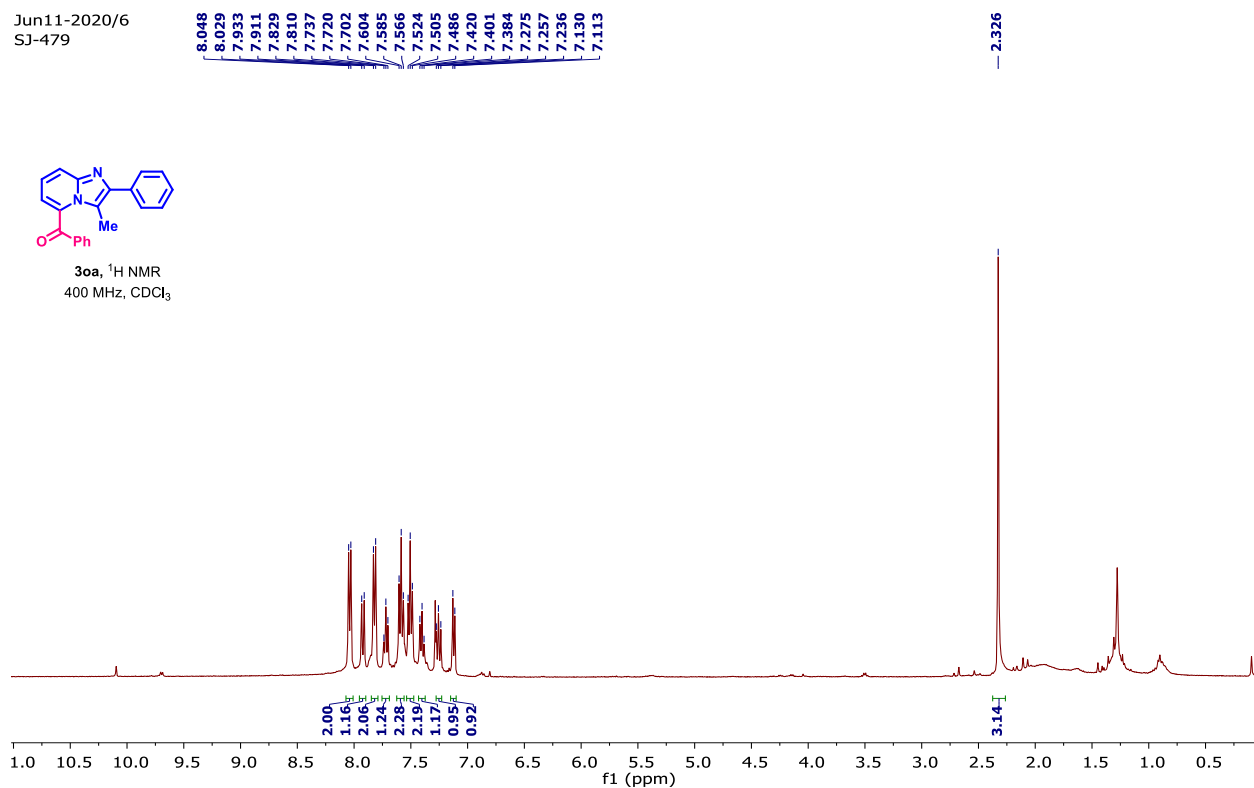
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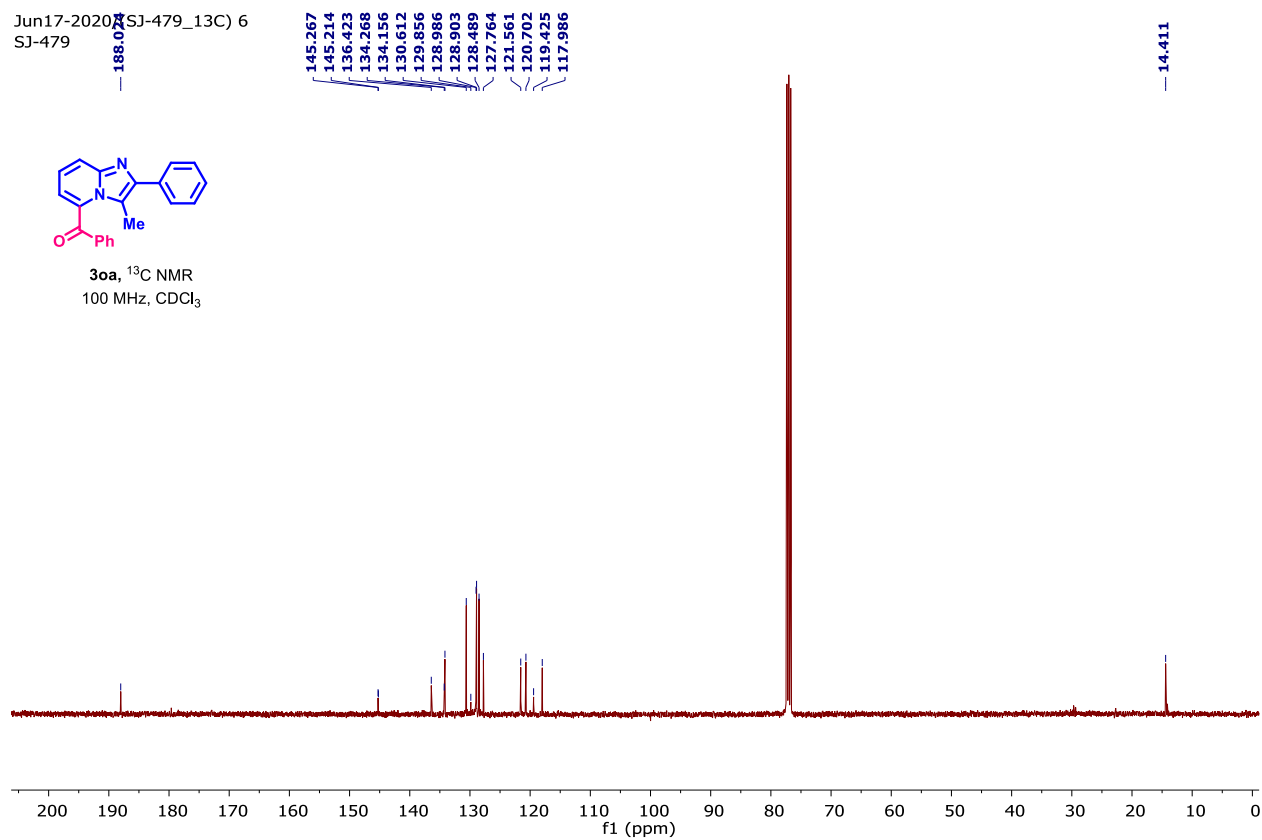
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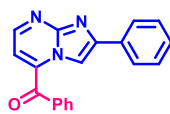
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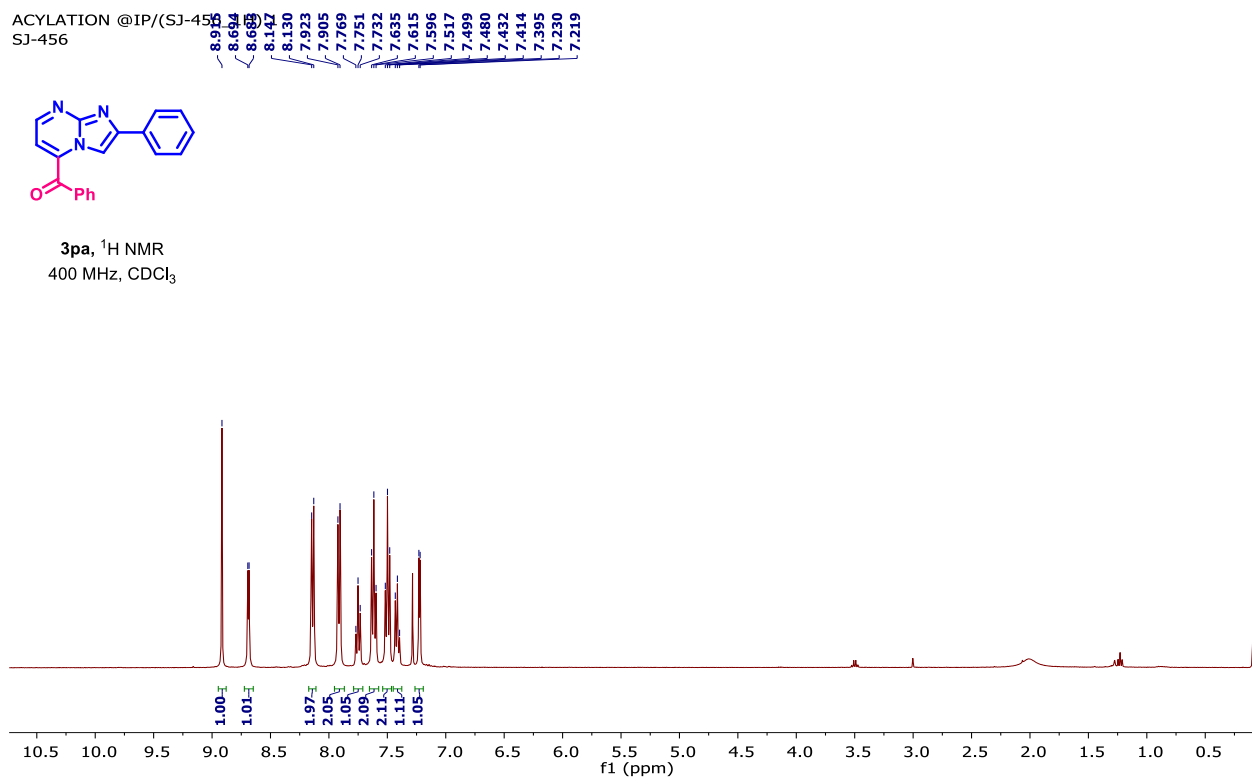
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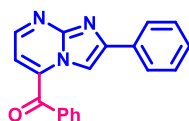
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SJ-456



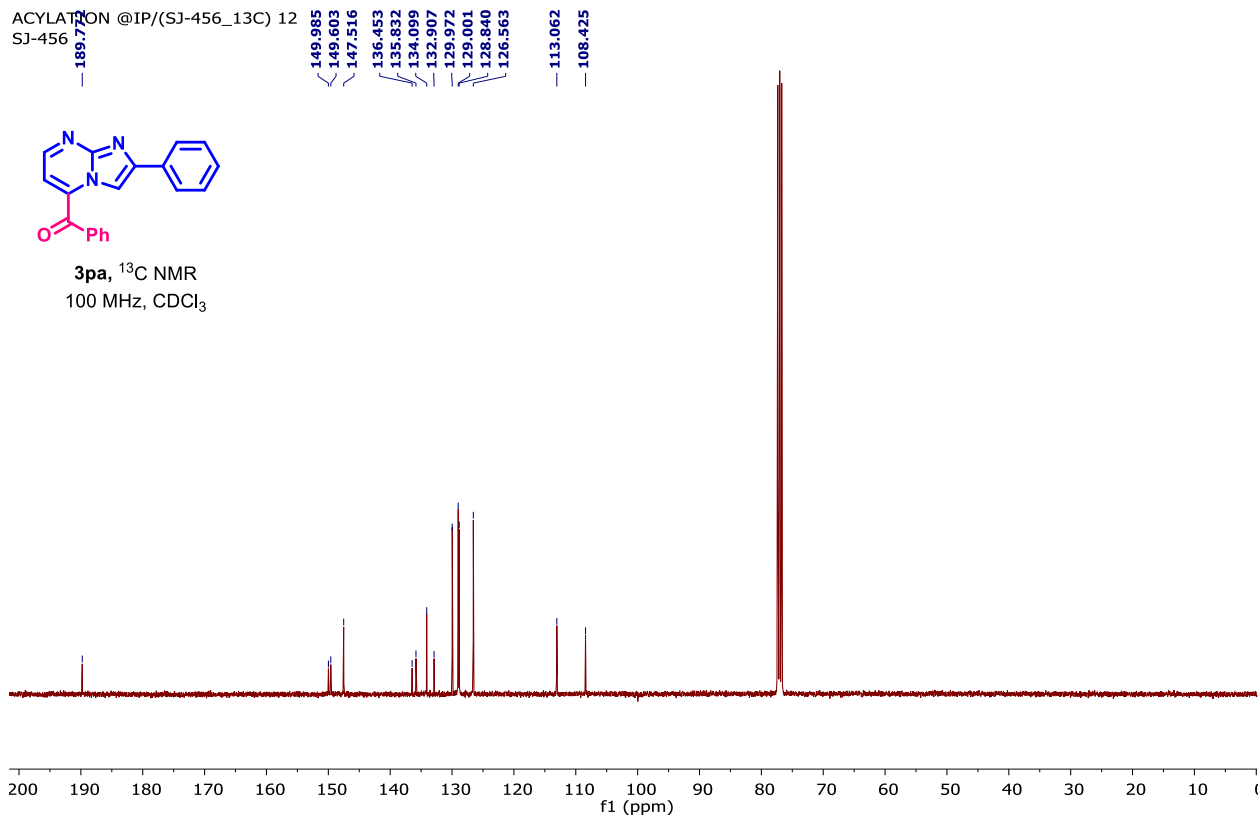
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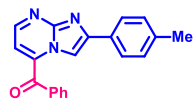
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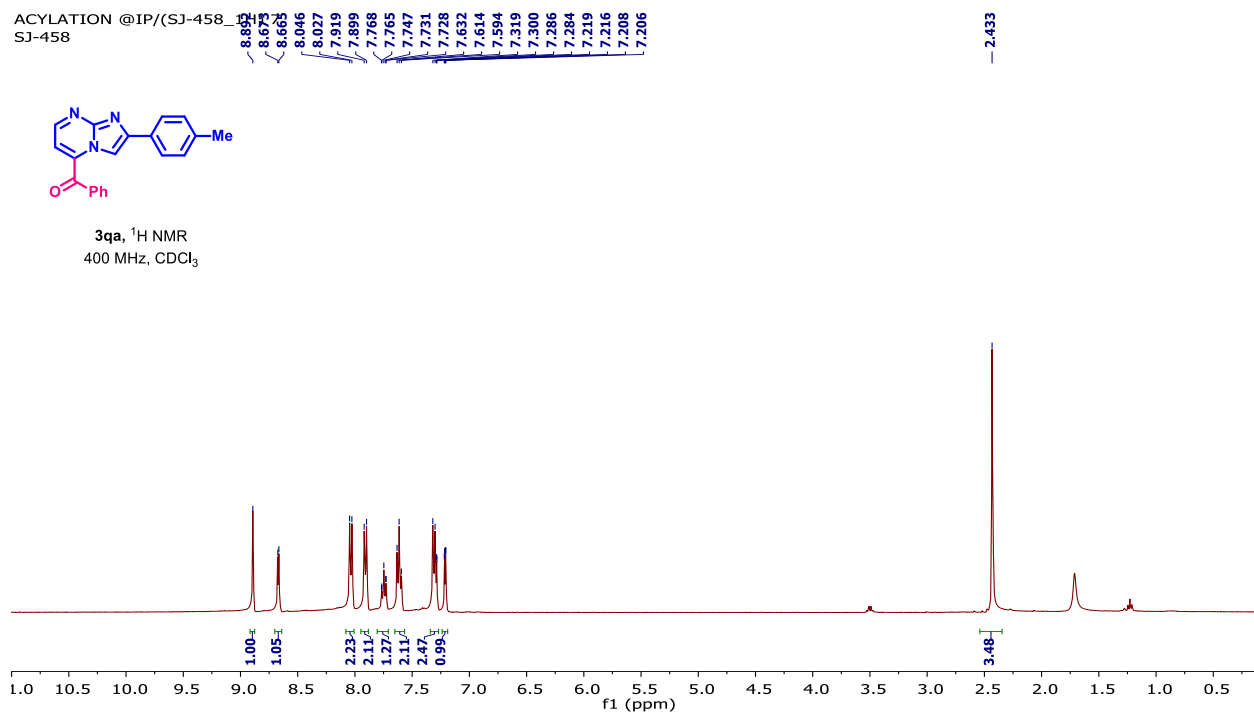
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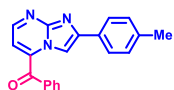
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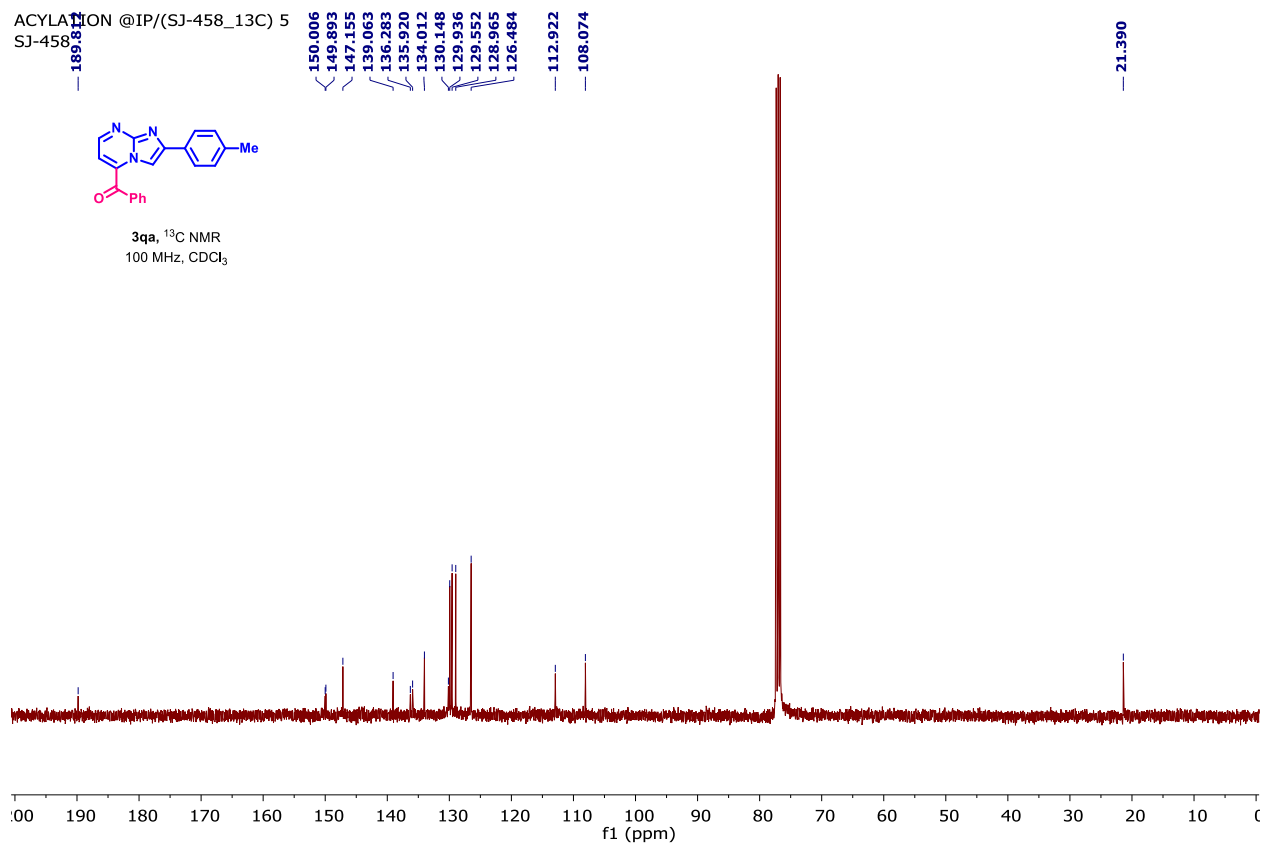
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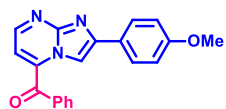
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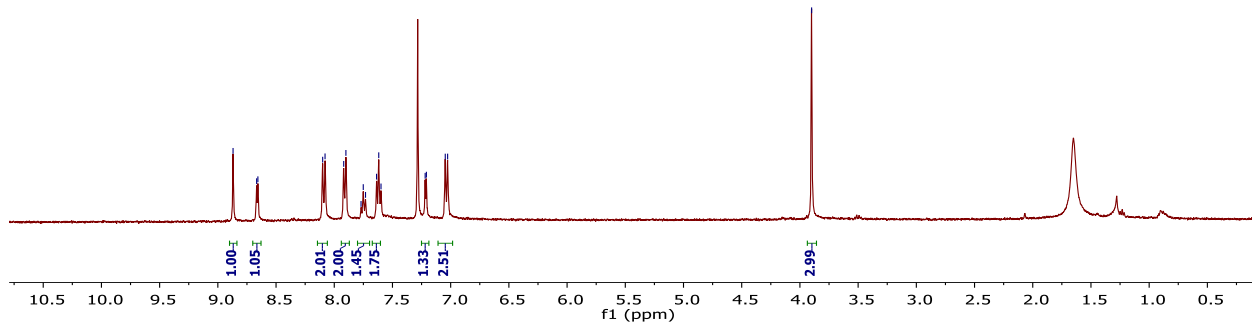
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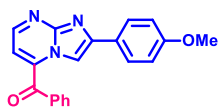
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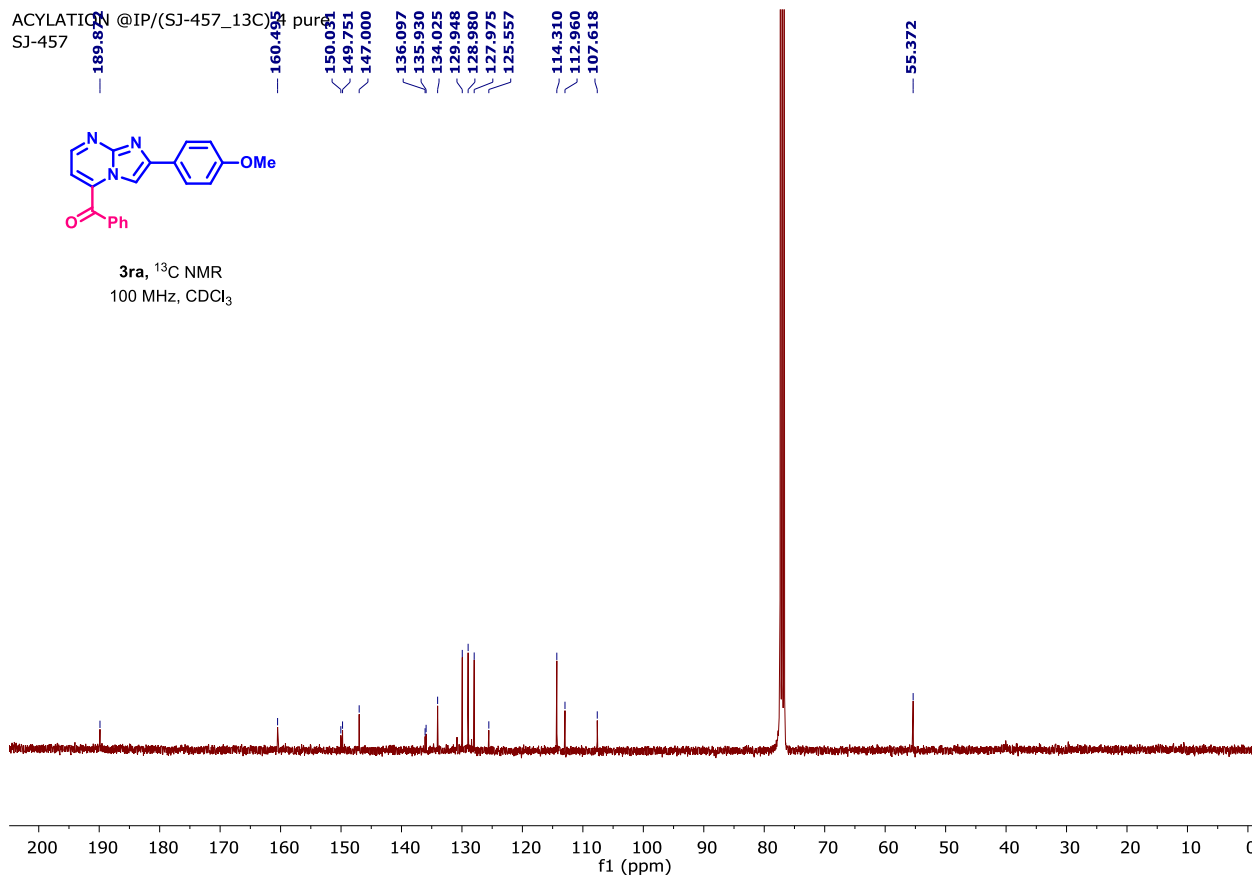
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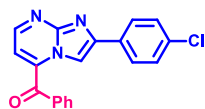
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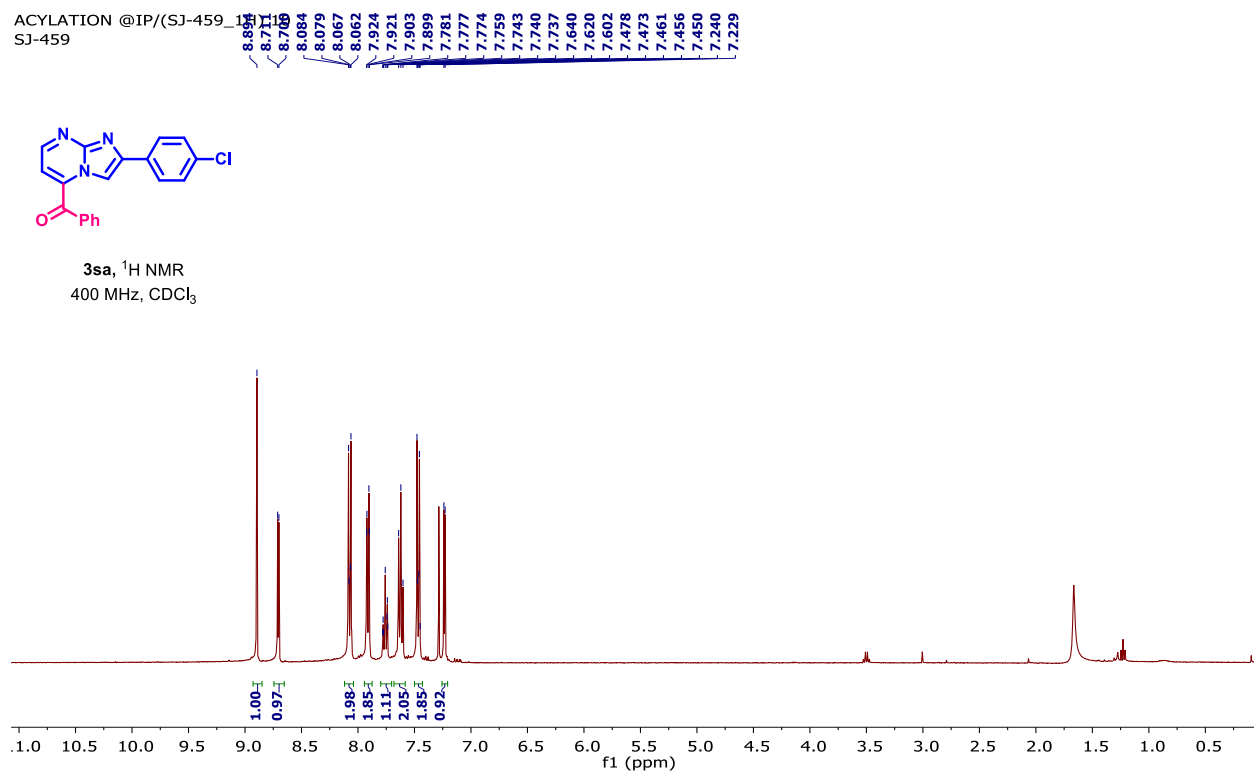
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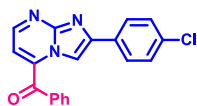
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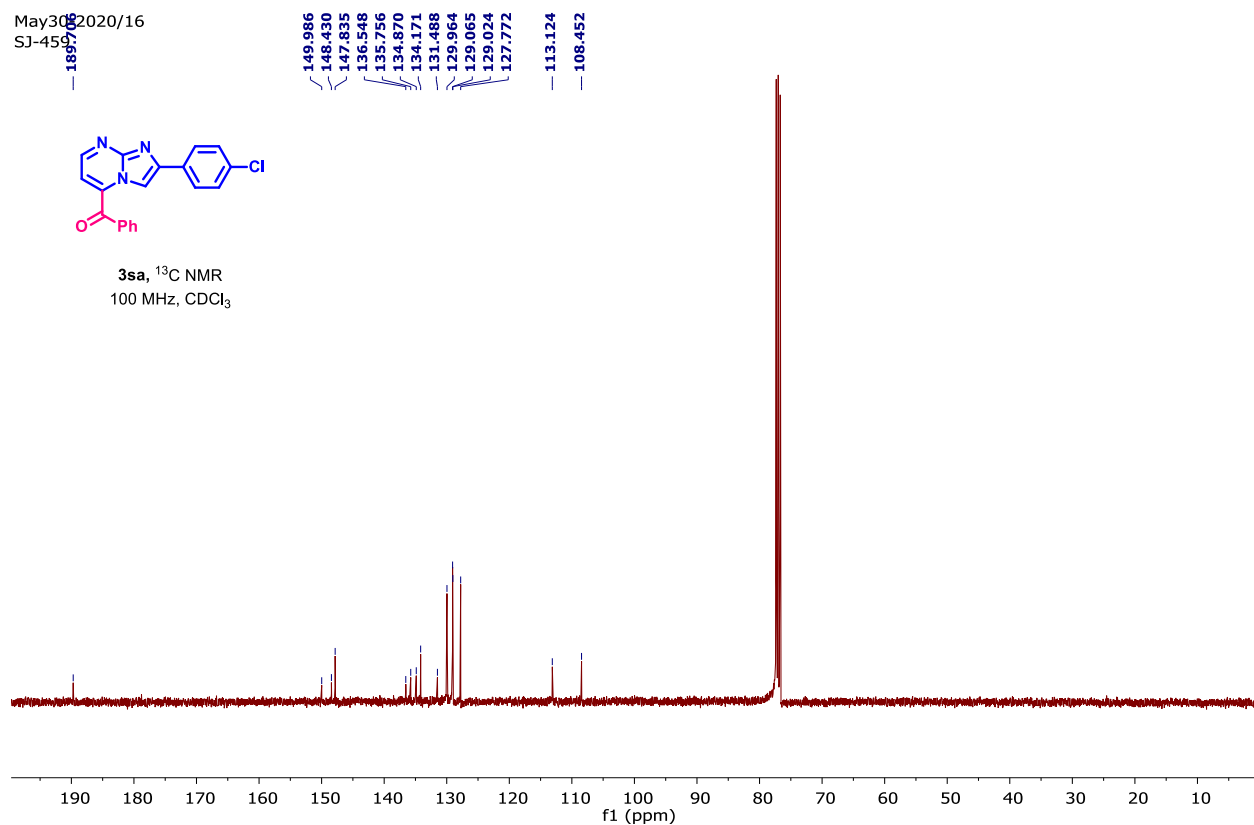
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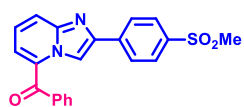
May30, 2020/16
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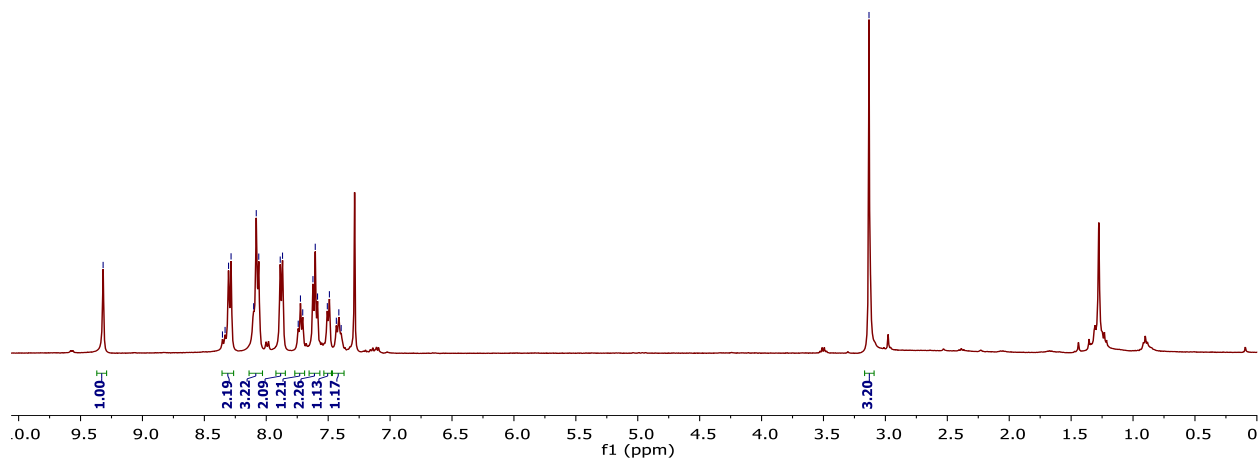
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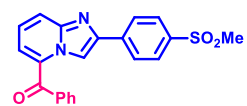
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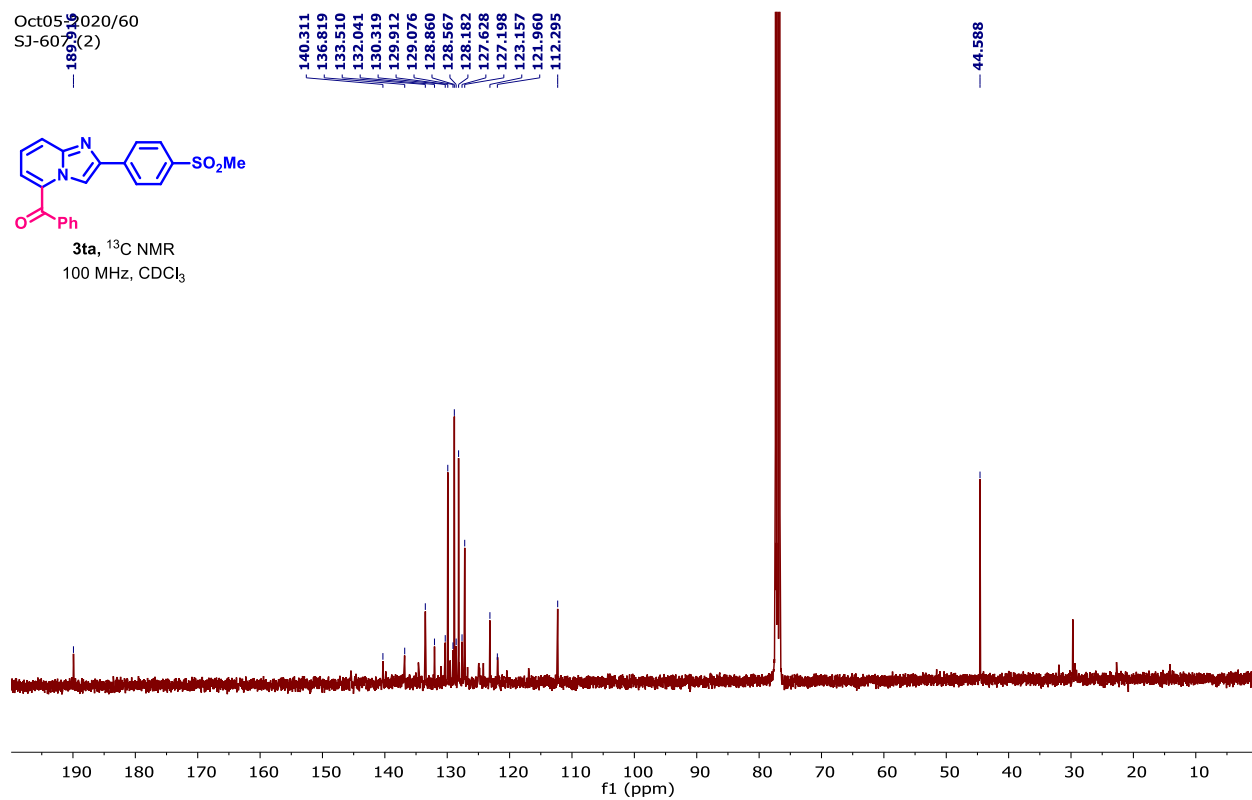
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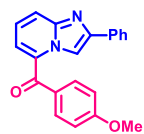
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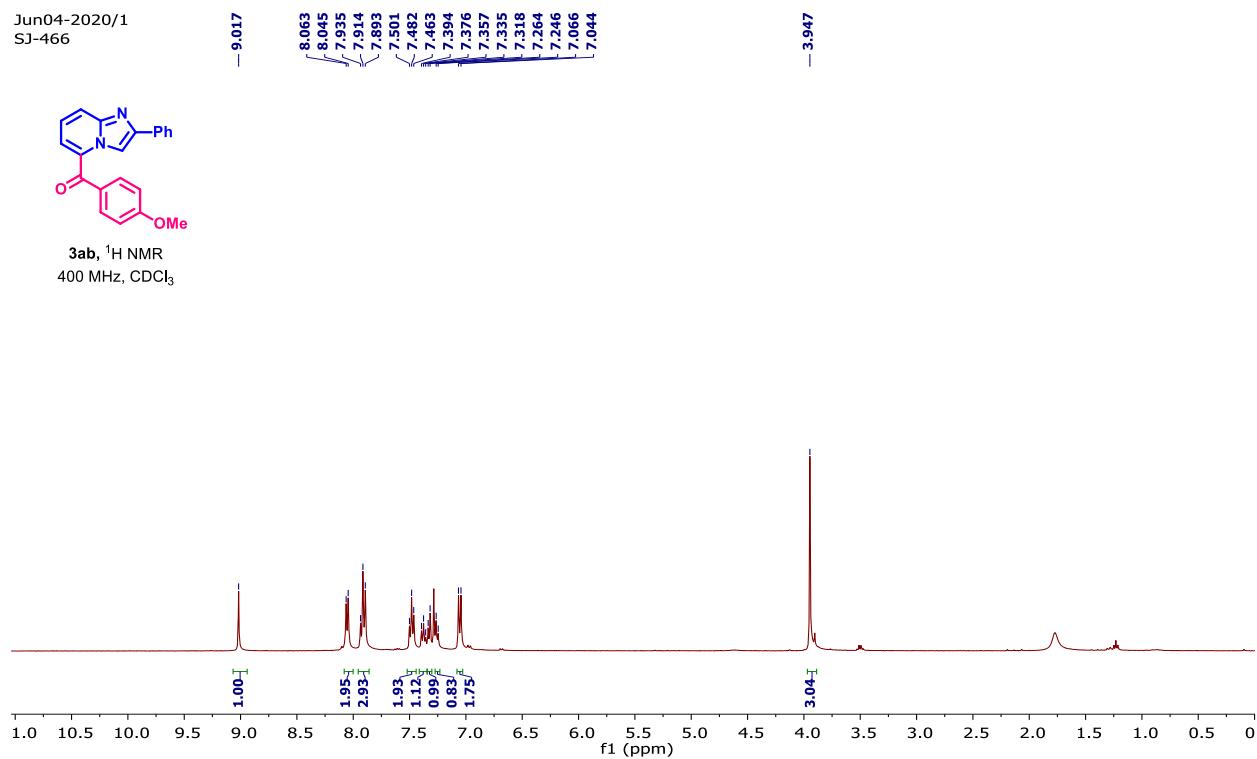
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Jun04-2020/1
SJ-466



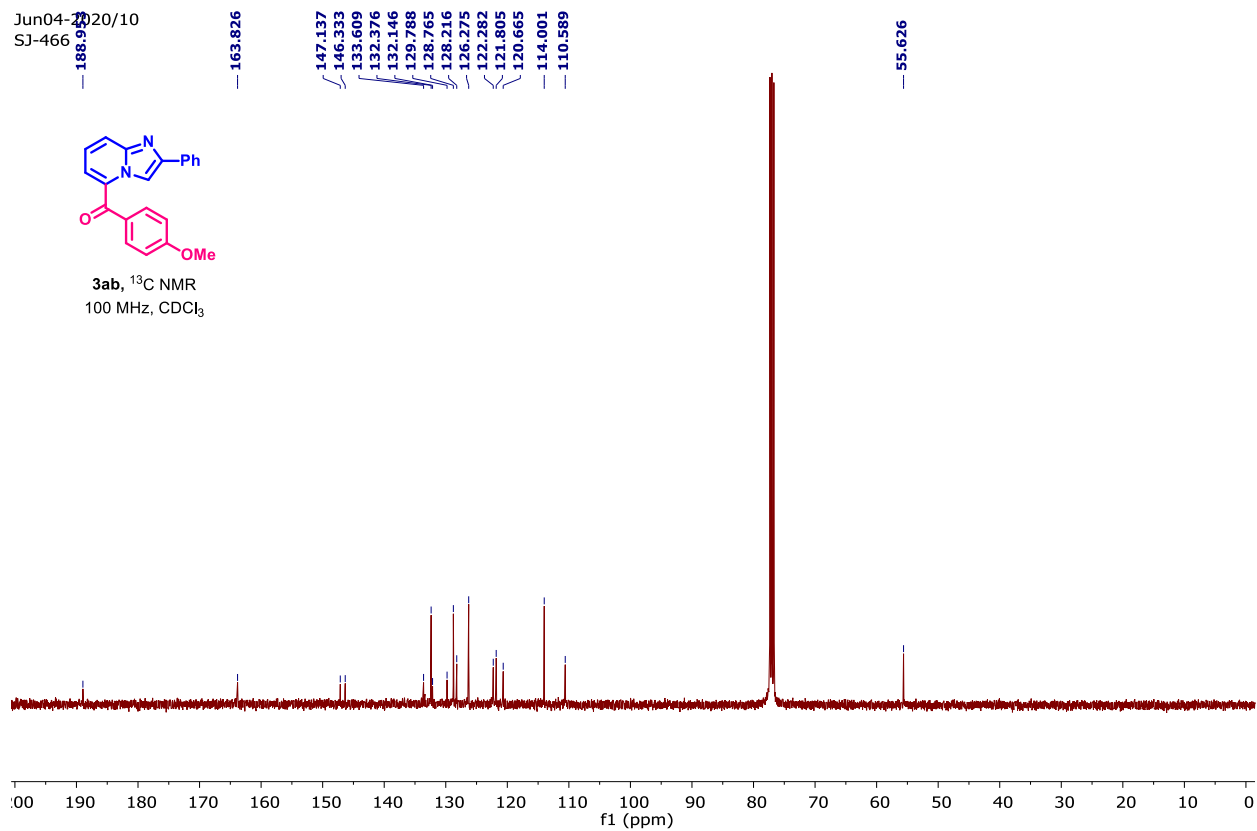
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400 MHz, CDCl_3



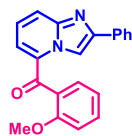
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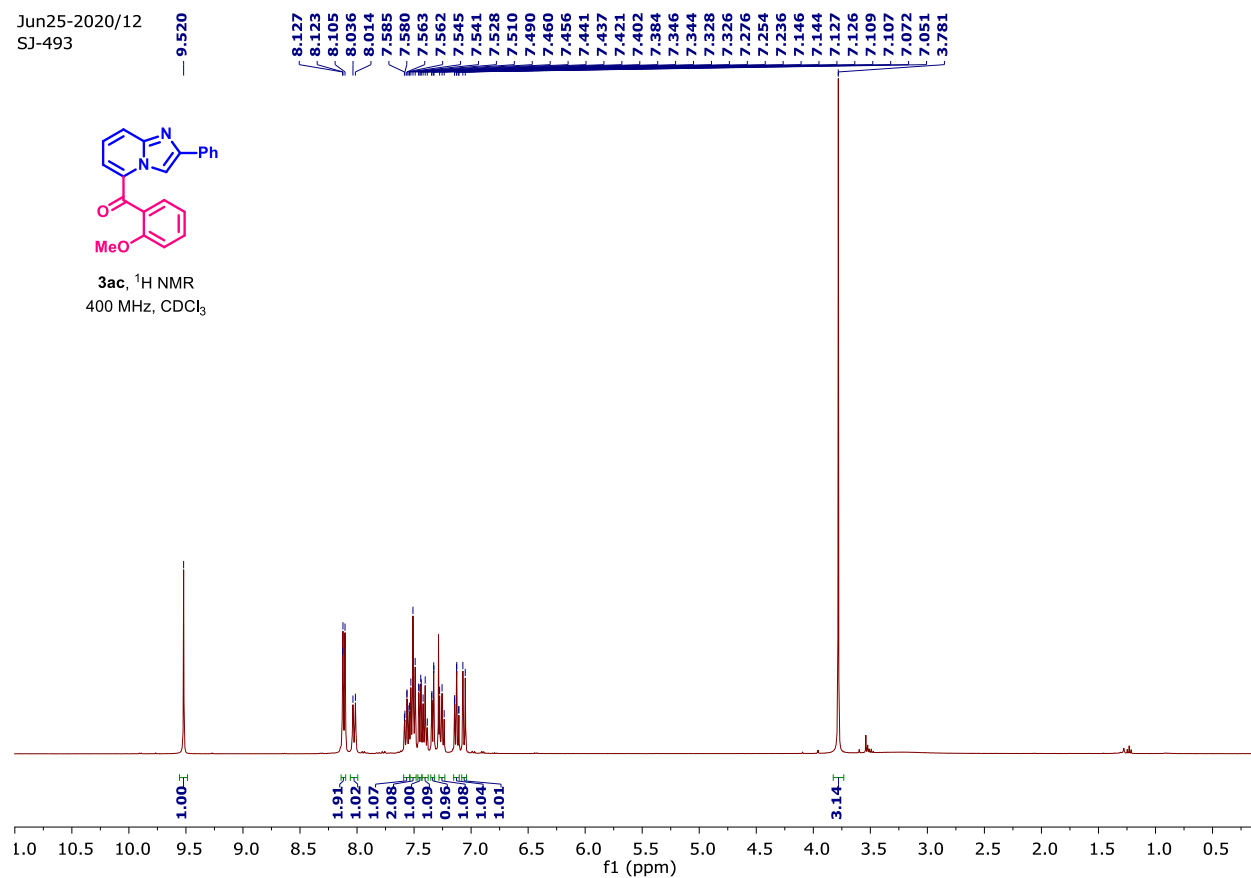
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Jun25-2020/12
SJ-493



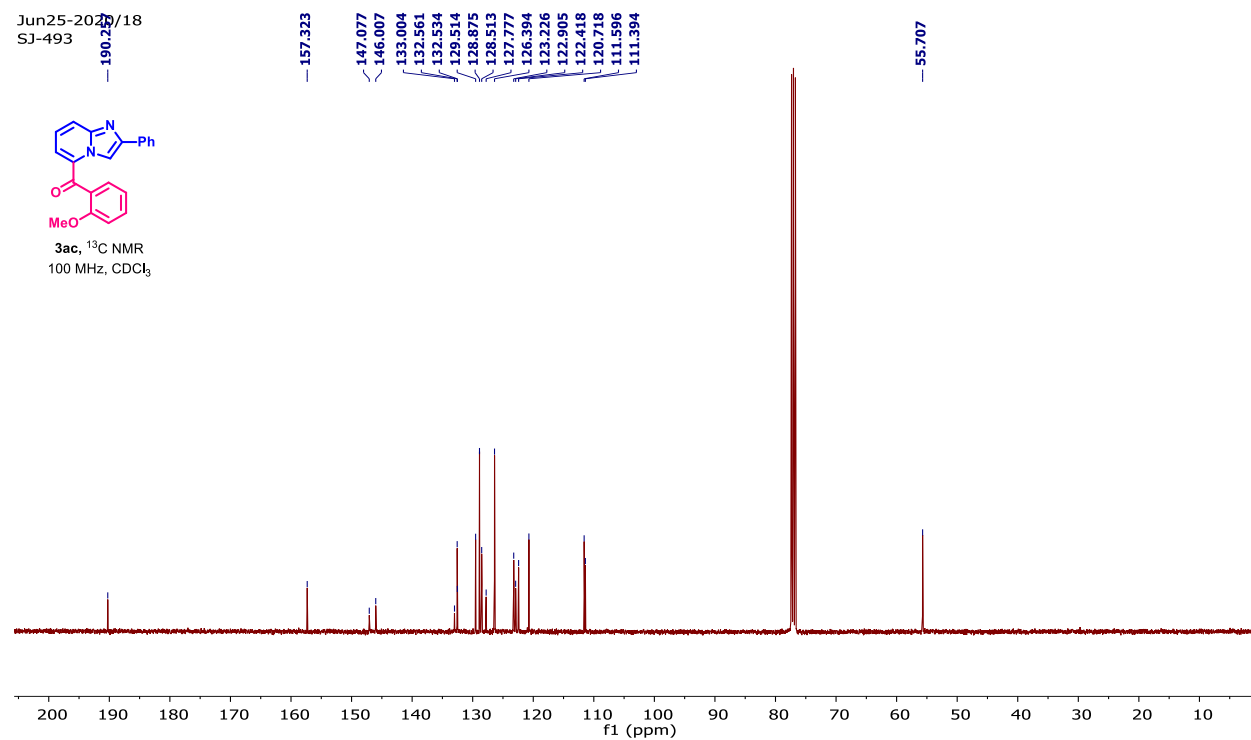
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400 MHz, CDCl_3



Jun25-2020/18
SJ-493



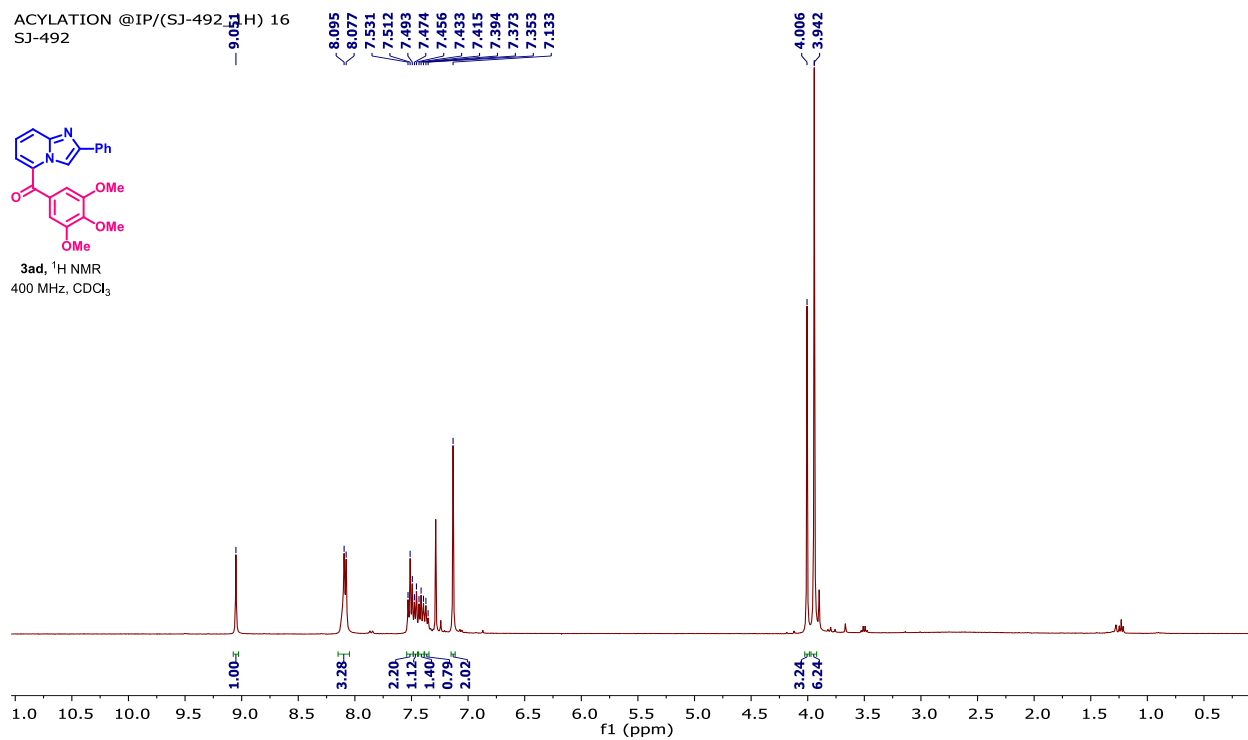
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100 MHz, CDCl_3



ACYLATION @IP/(SJ-492_13C) 16
SJ-492



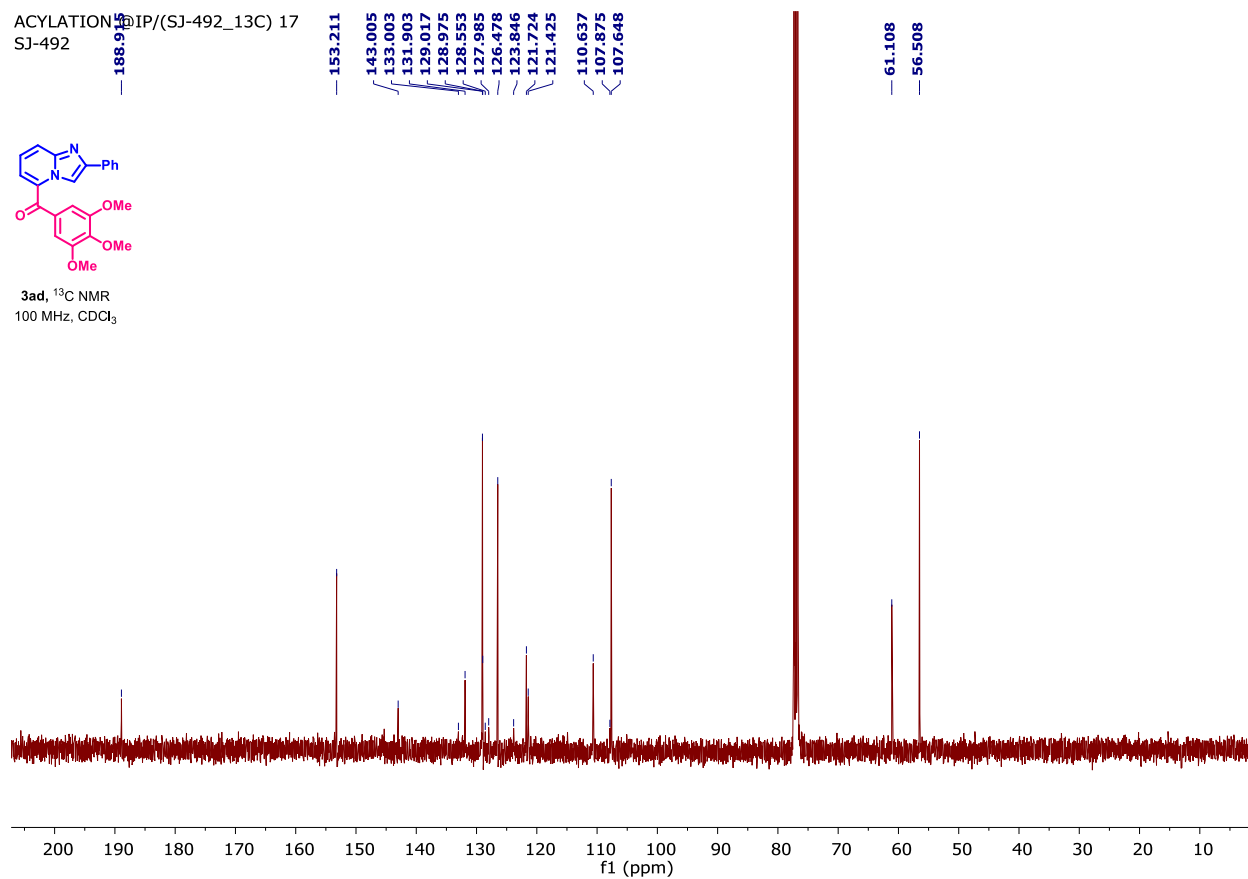
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400 MHz, CDCl_3



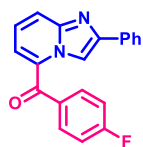
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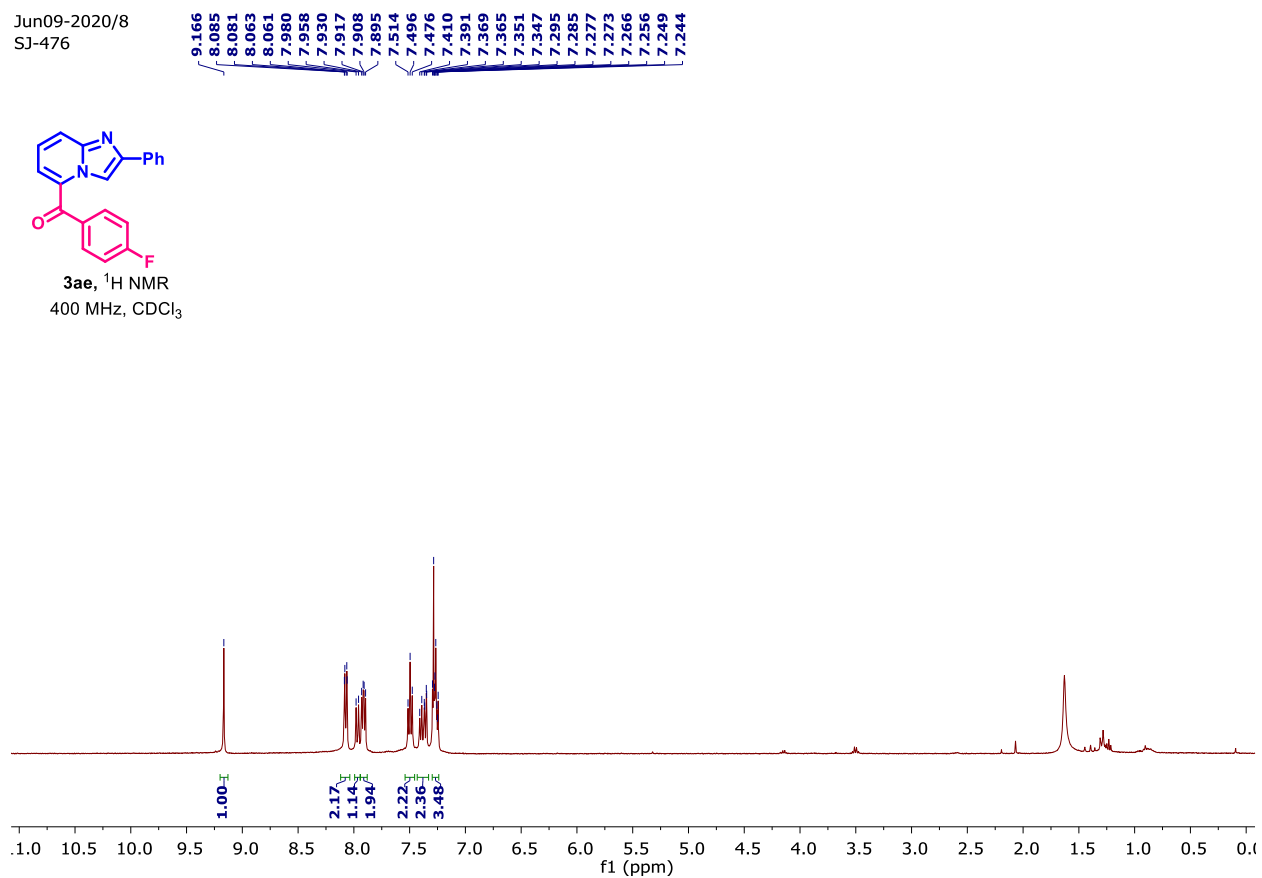
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100 MHz, CDCl_3



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SJ-476



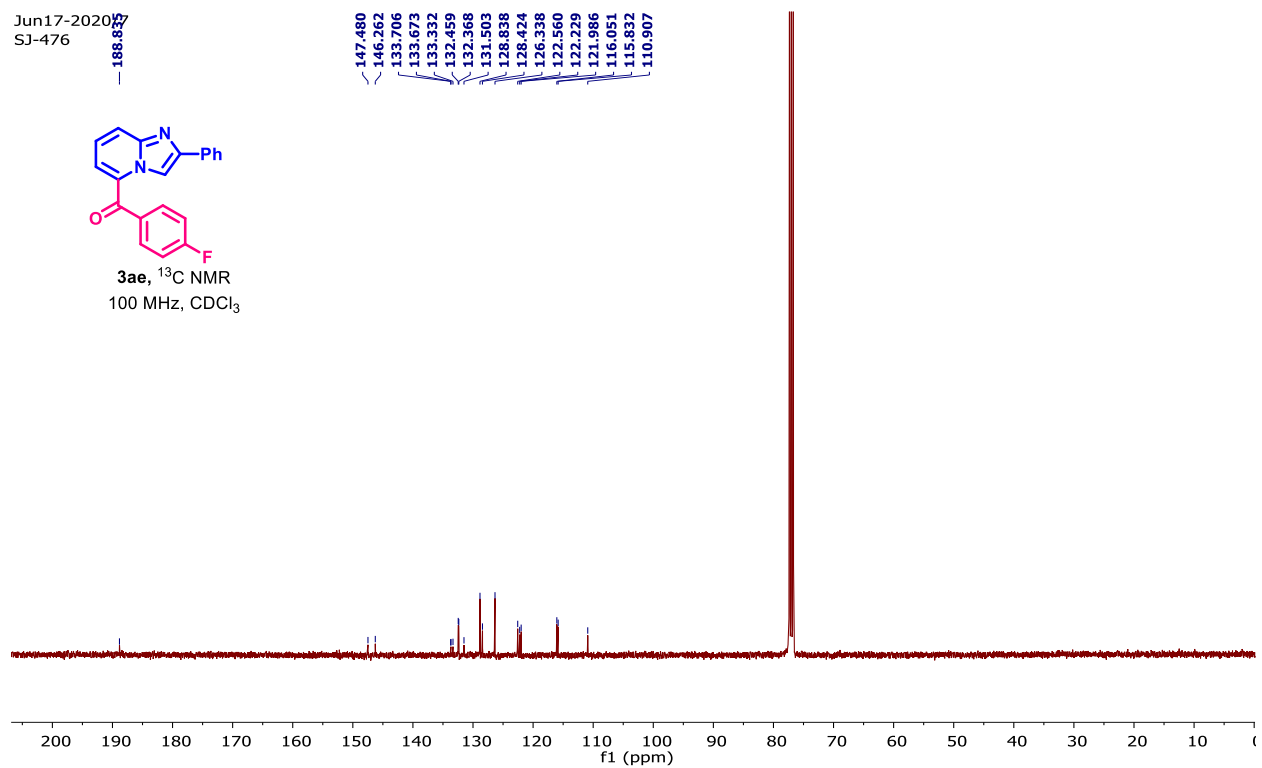
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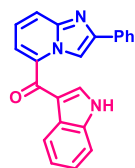
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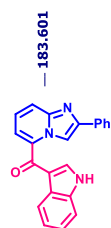
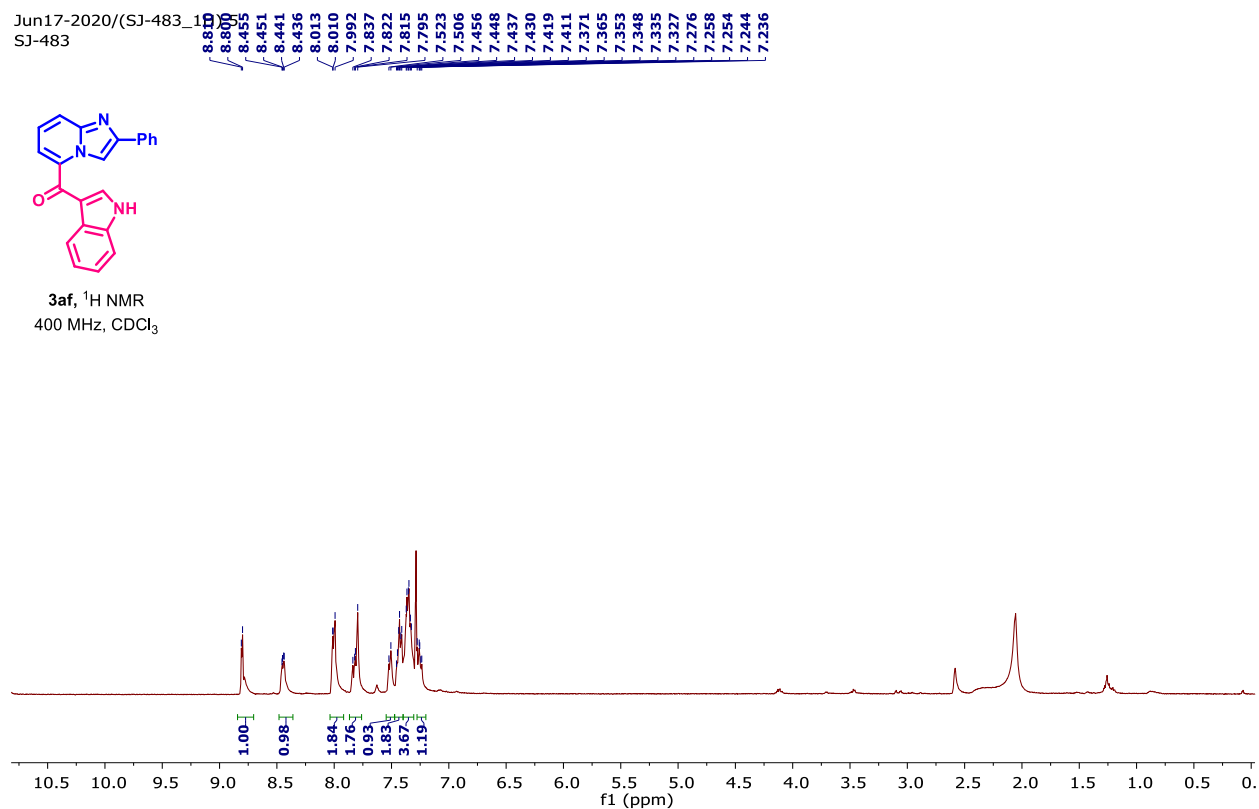
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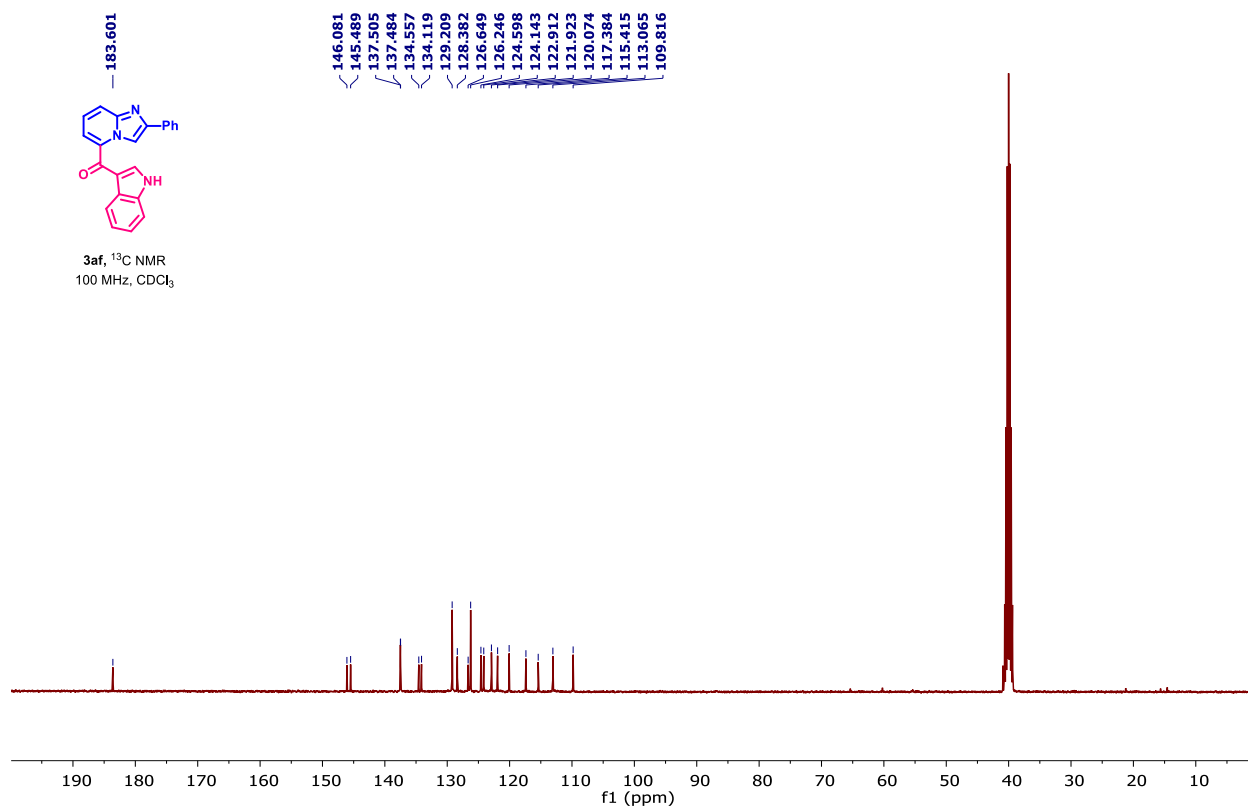
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SJ-483



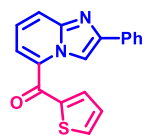
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400 MHz, CDCl_3



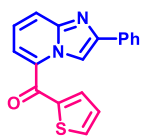
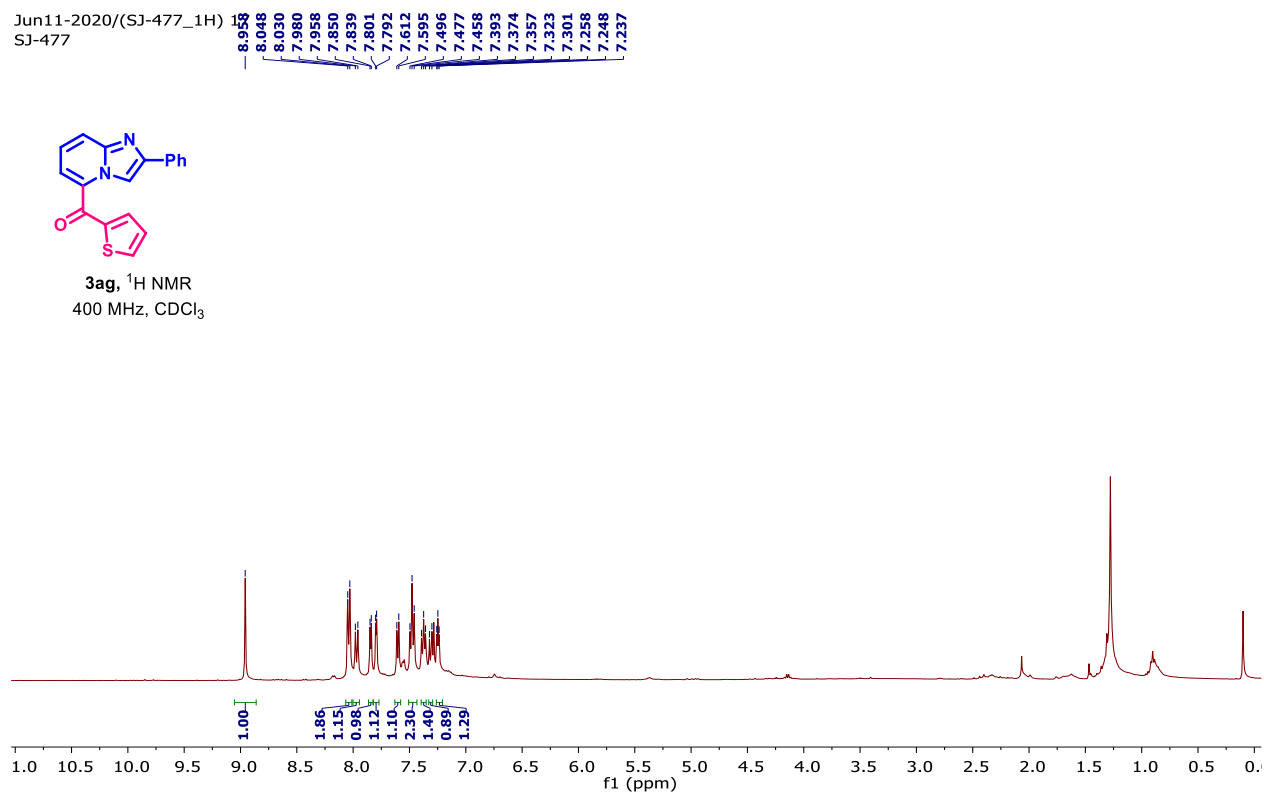
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100 MHz, CDCl_3



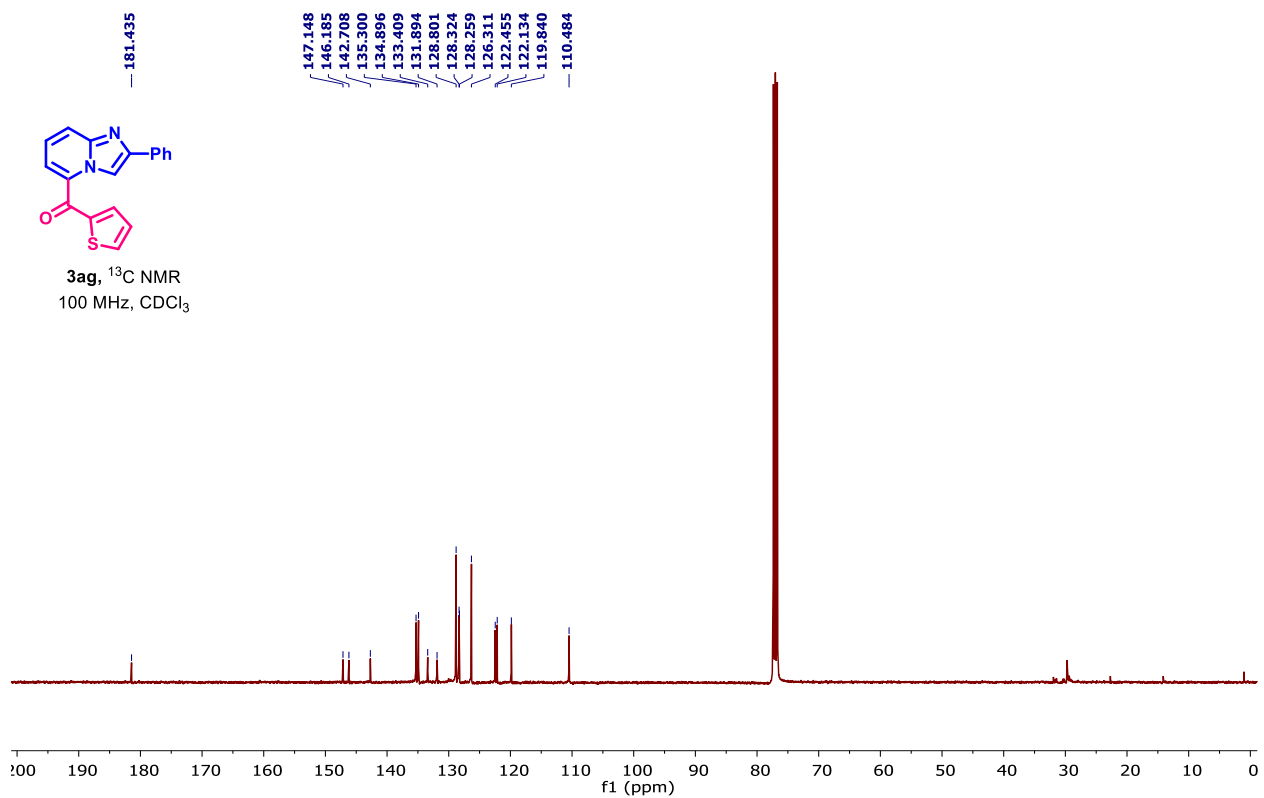
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SJ-477



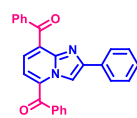
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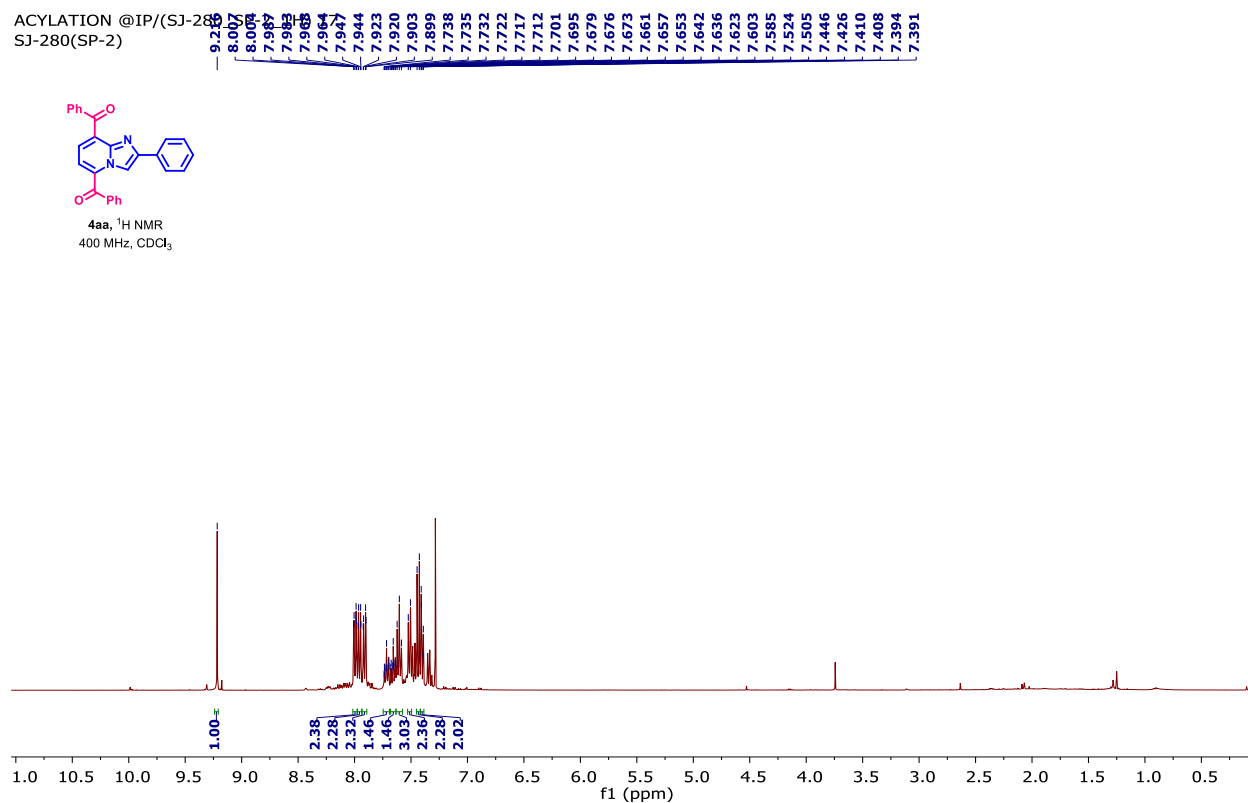
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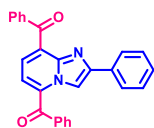
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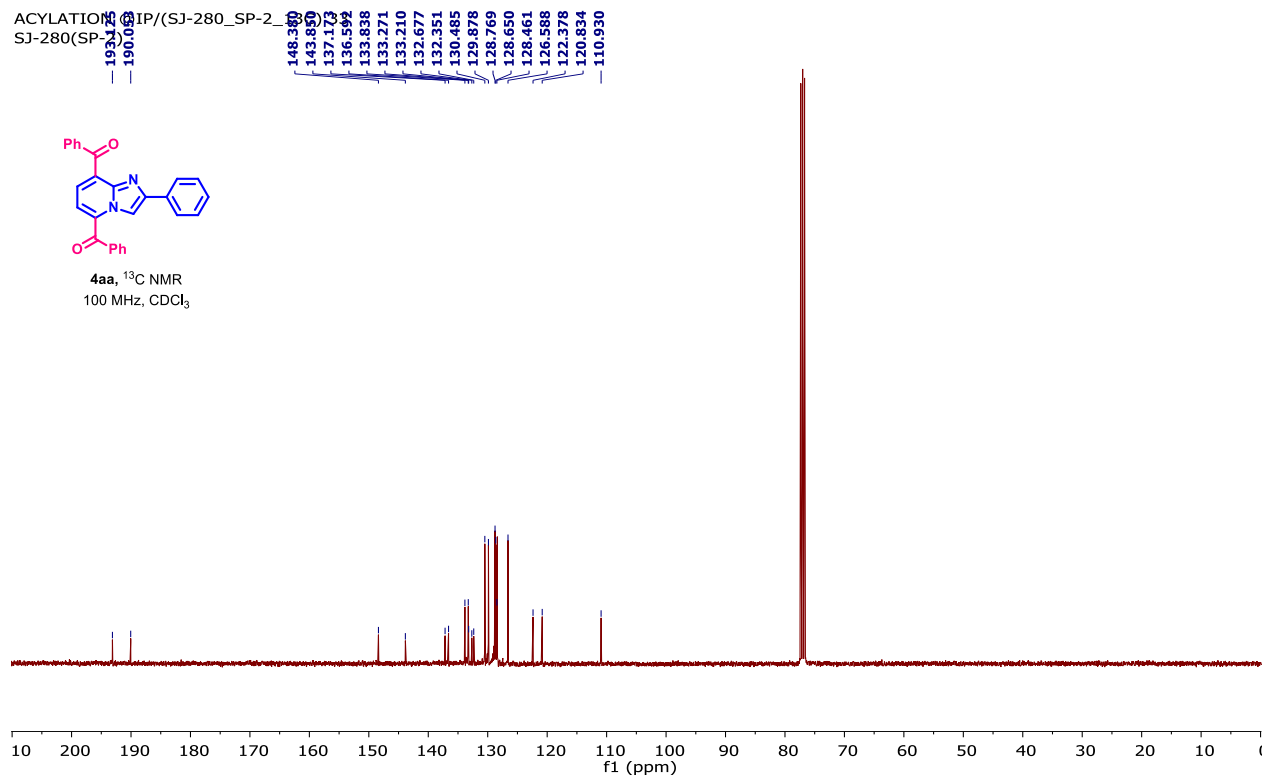
4aa, ¹H NMR
400 MHz, CDCl₃



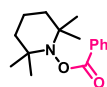
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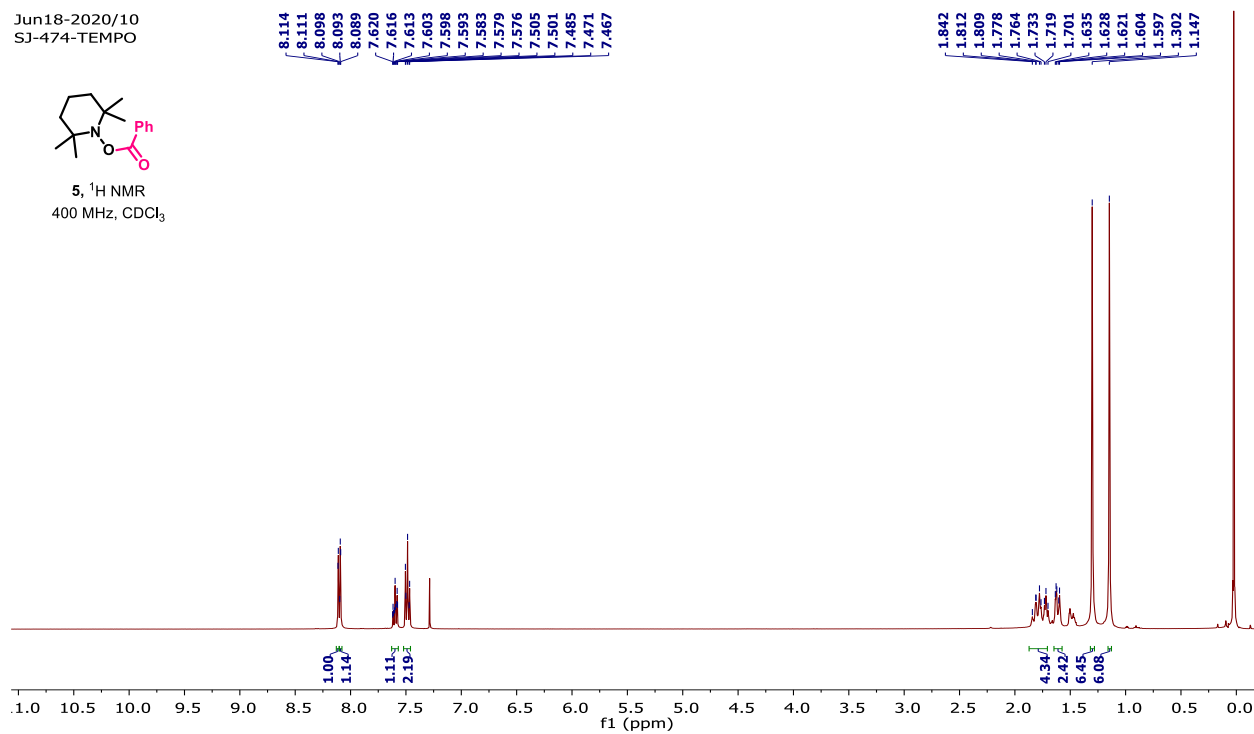
4aa, ¹³C NMR
100 MHz, CDCl₃



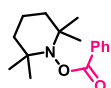
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SJ-474-TEMPO



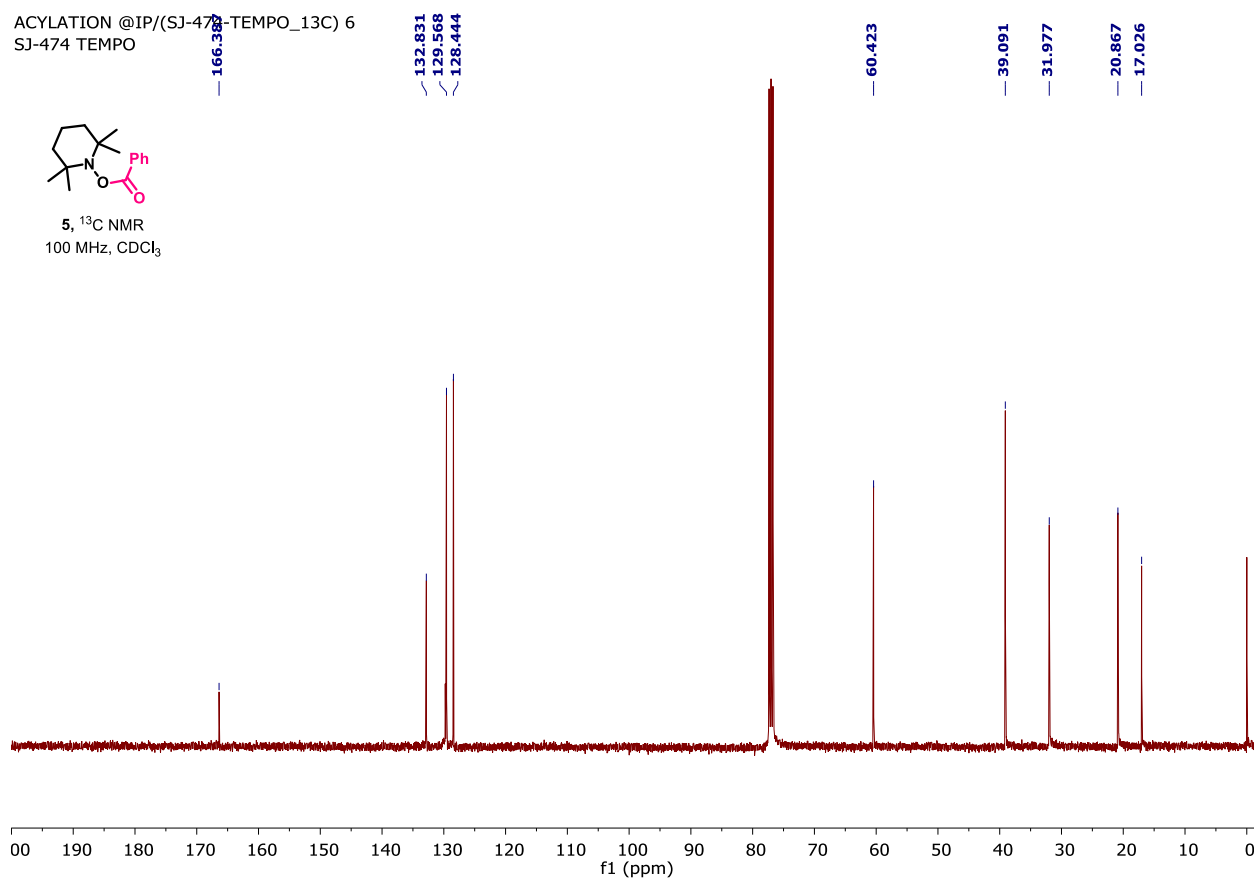
5, ^1H NMR
400 MHz, CDCl_3



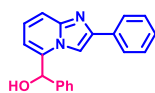
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SJ-474 TEMPO



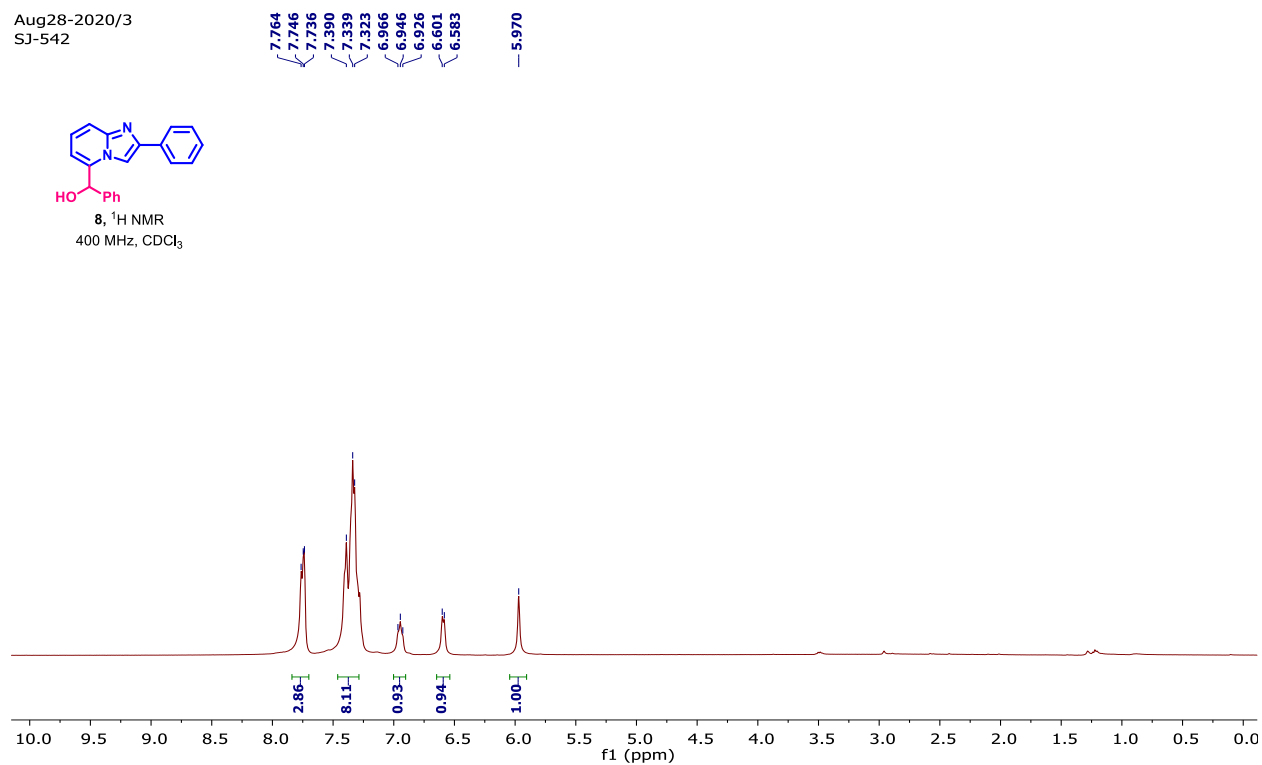
5, ^{13}C NMR
100 MHz, CDCl_3



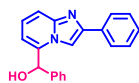
Aug28-2020/3
SJ-542



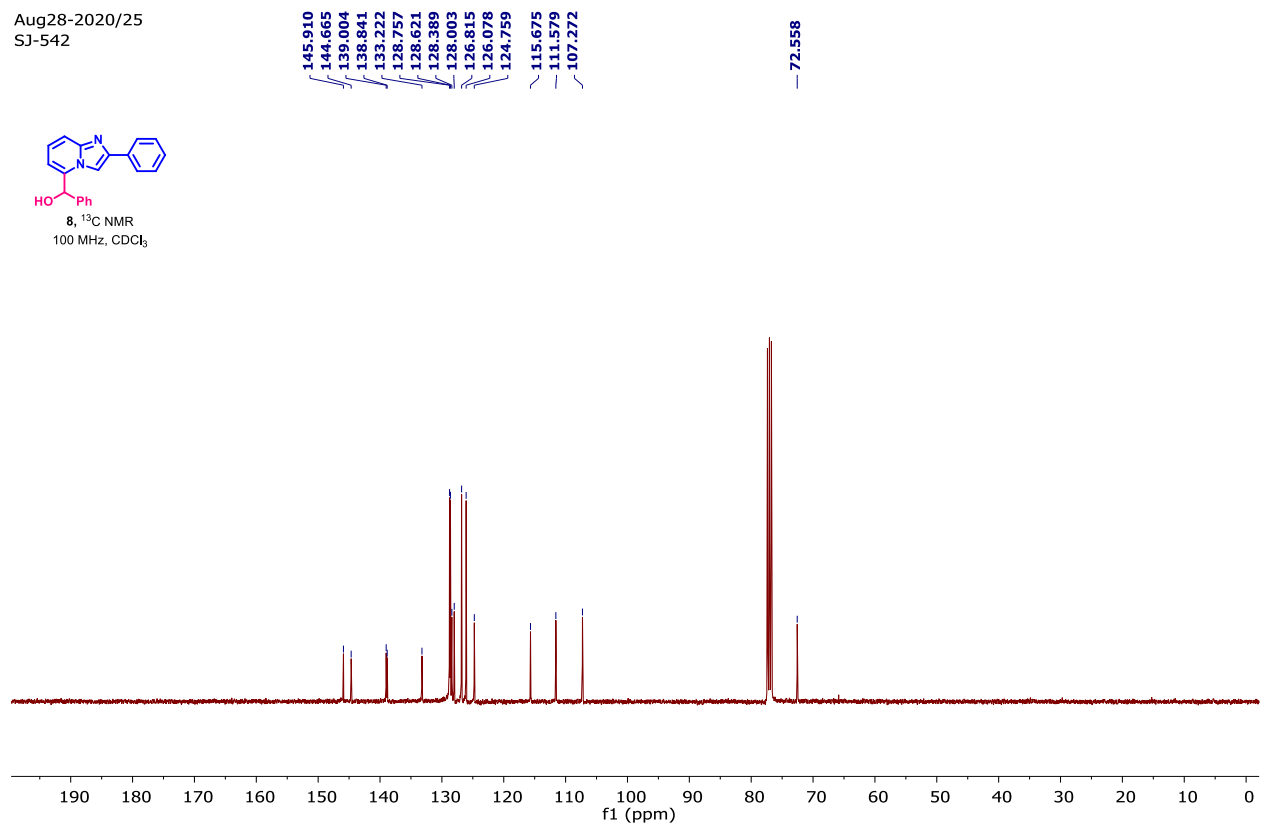
8, ^1H NMR
400 MHz, CDCl_3



Aug28-2020/25
SJ-542

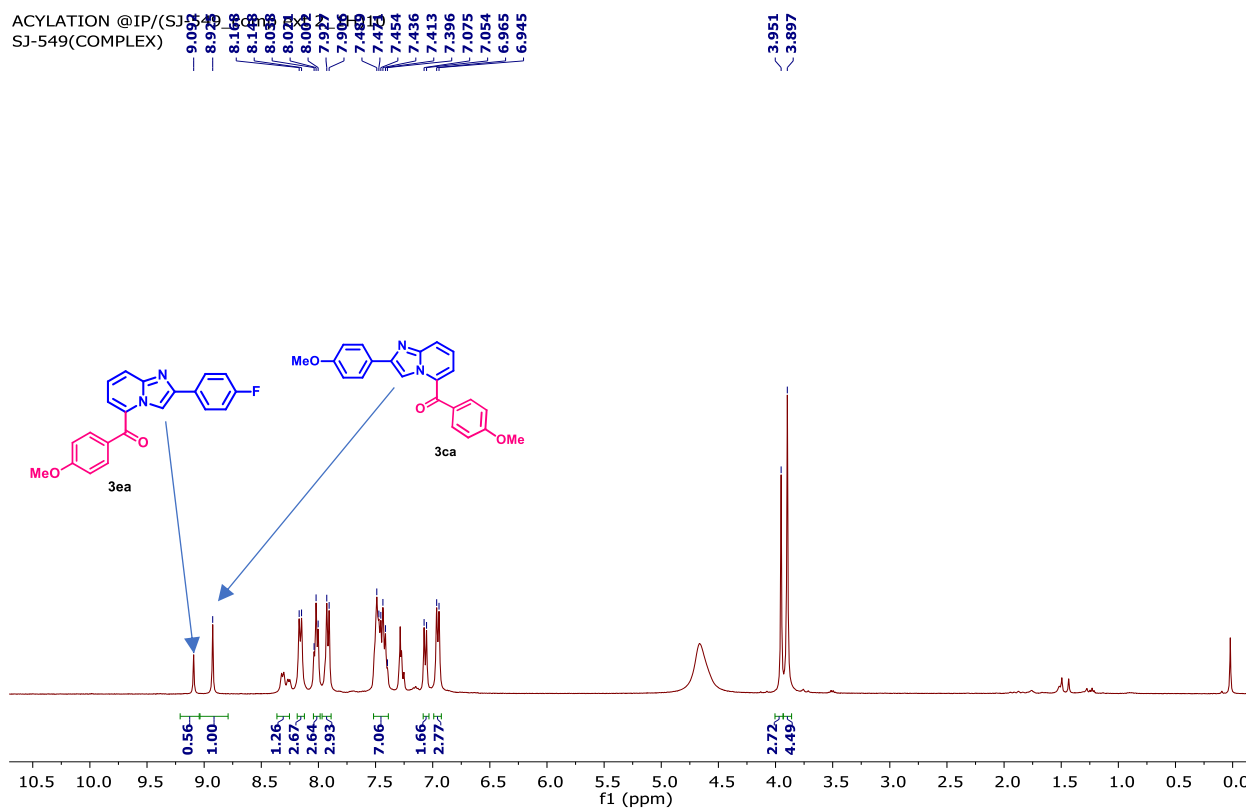


8, ^{13}C NMR
100 MHz, CDCl_3



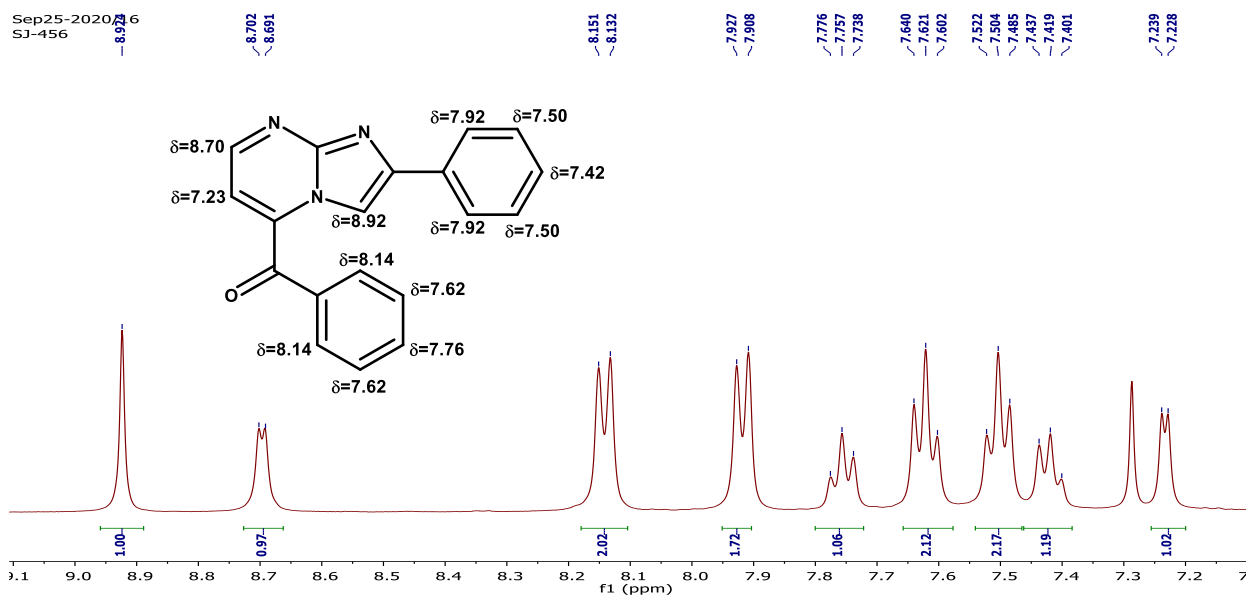
2. Competitive experiment

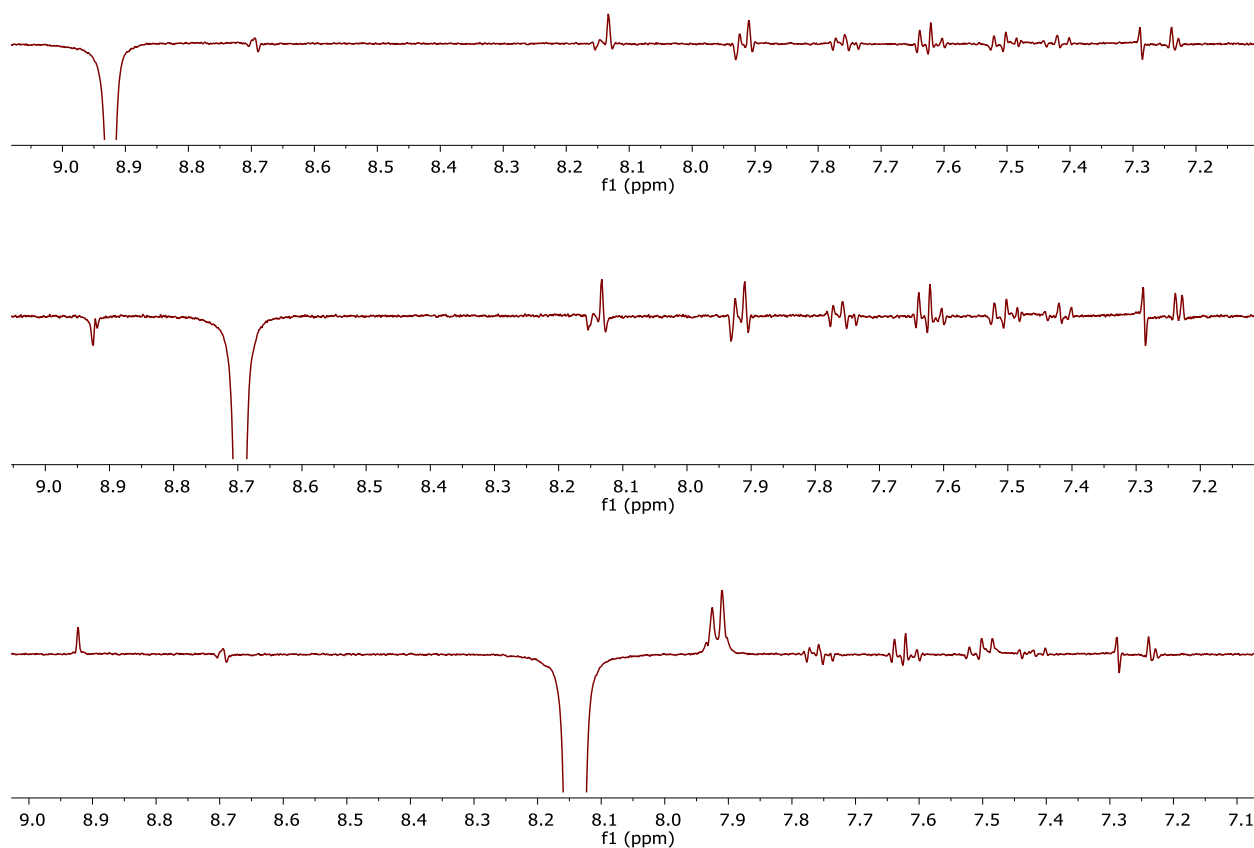
ACYLATION @IP/(SJ)
SJ-549(COMPLEX)



1D NOE Experiment of 3pa

Sep25-2020/16
SJ-456





3. Single crystal X-ray analysis of **3aa**

The compound **3aa** (Exp 565_AK-SJ-280) (C₂₀H₁₄N₂O) crystallized in orthorhombic Pbca space group. The ORTEP diagram of compound is given in Figure S1, CCDC 1978834. The bond lengths N1-C1, N1-C5, N1-C6, N2-C7 and N2-C1 bond lengths respectively are 1.4016(15), 1.3925(15), 1.3770(16), 1.3693(16) and 1.3271(16) Å, respectively and the corresponding bond angles C1-N1-C5, C1-N1-C6, C5-N1-C6, C1-N2-C7 are 121.55(10), 106.07(10), 132.29(11), and 105.34(10)°, respectively. The torsional angles C6-C7-C8-C9 and C5-C14-C15-C16 are -7.05 and 43.10°, respectively.

Experimental

Single crystals of **3aa** [C₂₀H₁₄N₂O] were grown from slow evaporation of dichloromethane solution. A suitable crystal was selected and mounted on a XtaLAB Pro: Kappa dual home/near diffractometer. The crystal was kept at 93(2) K during data collection. Using Olex2², the structure was solved with the ShelXT³ structure solution program using Intrinsic Phasing and refined with the ShelXL⁴ refinement package using Least Squares minimization.

Table S1 Crystal data and structure refinement for 3aa (Exp 565_AK-SJ-280)

Identification code	Exp 565_AK-SJ-280
Empirical formula	C ₂₀ H ₁₄ N ₂ O
Formula weight	298.33
Temperature/K	93(2)
Crystal system	orthorhombic
Space group	Pbca
a/Å	8.3127(2)
b/Å	17.1849(3)
c/Å	20.9916(4)
α/°	90
β/°	90
γ/°	90
Volume/Å ³	2998.71(11)
Z	8
ρ _{calc} /cm ³	1.322
μ/mm ⁻¹	0.656
F(000)	1248.0
Crystal size/mm ³	0.1 × 0.1 × 0.1
Radiation	CuKα (λ = 1.54184)
2θ range for data collection/°	8.424 to 159.084
Index ranges	-10 ≤ h ≤ 8, -21 ≤ k ≤ 17, -25 ≤ l ≤ 26
Reflections collected	10545
Independent reflections	3204 [R _{int} = 0.0331, R _{sigma} = 0.0312]
Data/restraints/parameters	3204/0/208
Goodness-of-fit on F ²	1.096
Final R indexes [I >= 2σ (I)]	R ₁ = 0.0442, wR ₂ = 0.1217
Final R indexes [all data]	R ₁ = 0.0464, wR ₂ = 0.1239
Largest diff. peak/hole / e Å ⁻³	0.24/-0.27

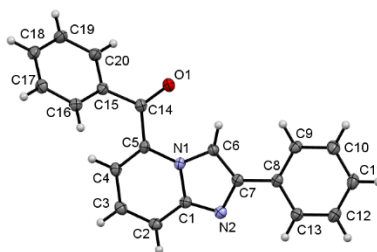


Figure S1. The ORTEP crystal structure diagram of compound **3aa**. The thermal ellipsoids are drawn at the 50% probability level.

4. References

- (1) Wadhwa, K.; Yang, C; Paul, R. W.; Kris, C. D.; Chemrbukar, S. R.; Reddy, R. E.; Synthesis of arylglyoxylic acids and their collision-induced dissociation. *Synth. Commun.* **2008**, 38, 4434-4444.
- (2) Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschman, H. Olex2: A complete structure solution, refinement and analysis program. *J. Appl. Cryst.* **2009**, 42, 339-341.
- (3) Sheldrick, G. Shelxet – intergated space group and crystal-structure determination. *Acta Cryst. A* **2015**, 71, 3-8.
- (4) Sheldrick, G. Crystal structure refinement with shelxet *Acta Cryst. C* **2015**, 71, 3-8.