

Supporting Information

Copper-promoted cyanoalkylation/ring-expansion of vinylcyclopropanes with α -C–H bonds in alkylnitriles toward 3,4-dihydronaphthalenes

Zan Chen,^{†a} Quan Zhou,^{†a} Qing-Nan Chen,^a Pu Chen,^a Bi-Quan Xiong,^a Yun Liang,^b

Ke-Wen Tang,^a Jun Xie,^{*a} Yu Liu^{*ab}

^a Department of Chemistry and Chemical Engineering, Hunan Institute of Science and Technology, Yueyang 414006, China

^b Key Laboratory of the Assembly and Application of Organic Functional Molecules of Hunan Province, Hunan Normal University, Changsha, Hunan 410081, China.

[phone and fax number: +86073-0864-0122 and E-mail addresses:

lyxtmj_613@163.com (Yu Liu) and 12108014@hnist.edu.cn (Jun Xie)]

[†] These authors contributed equally to this work

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1. General Information	

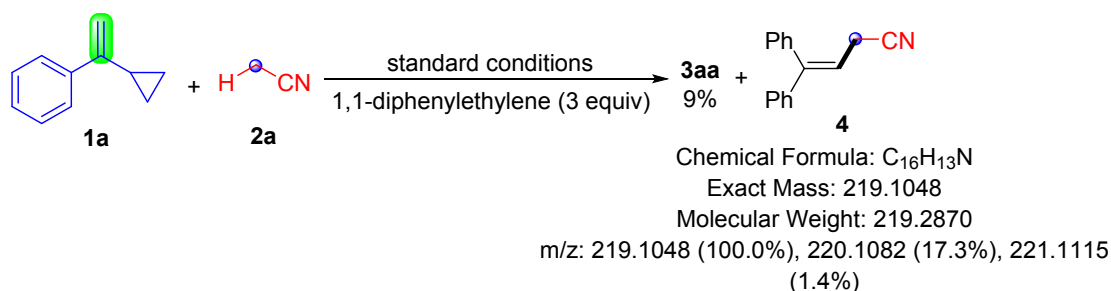
Unless otherwise stated, all commercial reagents were used as received. Cupric acetate (MACKLIN, $\geq 99\%$), Silver carbonate (BK, 99%), aldehydes (Innochem, $>98\%$) and ketones (Innochem, $>98\%$) were used without further treatment. All reagents and solvents were commercially available and used without any further purification unless specified. All solvents were dried and distilled according to standard procedures. Flash column chromatography was performed using silica gel (0.25mm, 300-400 mesh). Analytical thin-layer chromatography was performed using glass plates pre-coated with 0.25mm 300-400 mesh silica gel impregnated with a fluorescent indicator (254 nm). All reactions were carried out with magnetic stirring and in dried glassware. Nuclear magnetic resonance (NMR) spectra are recorded in parts per million from internal tetramethylsilane on the δ scale. ^1H NMR, ^{19}F NMR and ^{13}C NMR spectra were recorded in CDCl_3 on a Bruker DRX-400 spectrometer operating at 400 MHz, 376 MHz and 100 MHz, respectively. All chemical shift values are quoted in ppm and coupling constants quoted in Hz. The solvent peak was used as a reference value, for ^1H NMR: TMS = 0.00 ppm, for ^{13}C NMR: CDCl_3 = 77.00 ppm. The following abbreviations were used to explain multiplicities: s = singlet, d = doublet, dd = doublet of doublet, t = triplet, td = triplet of doublet, q = quartet, m = multiplet, and br = broad. High-resolution mass spectra (HRMS) were obtained on an Agilent mass spectrometer using ESI-TOF (electrospray ionization-time of flight).

2. Experiment Section

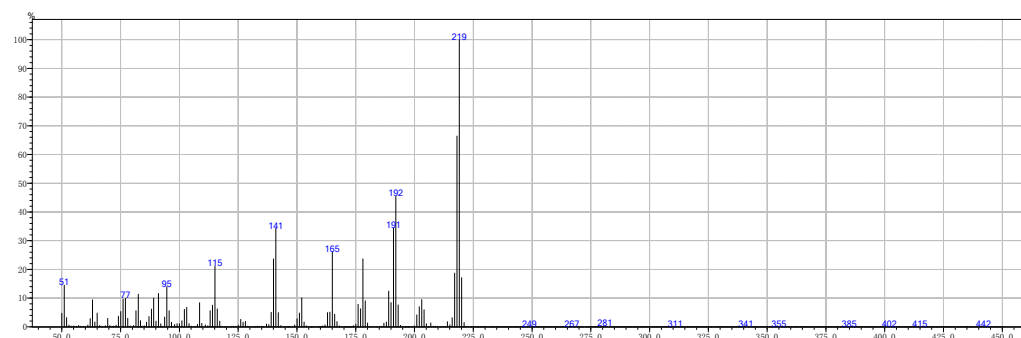
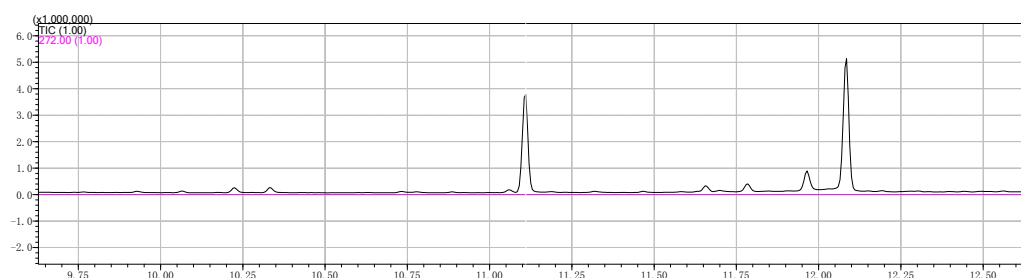
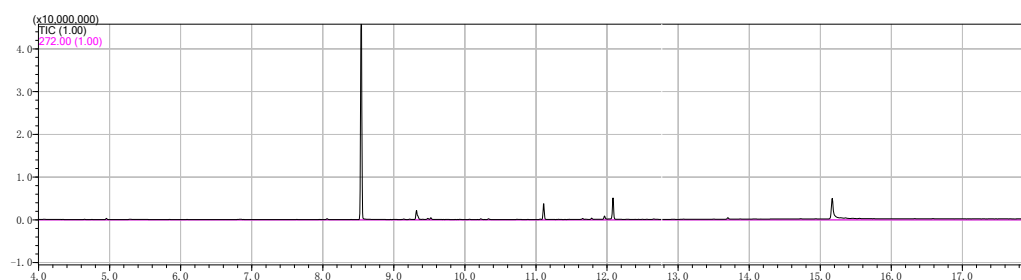
2.1 General Procedure for the Synthesis of Vinylcyclopropanes (1):

All Vinylcyclopropane **1**^[1] were synthesized according to the known methods.

2.2 Radical Trapping Experiments.



The GC-MS analysis results of raw reaction mixture:

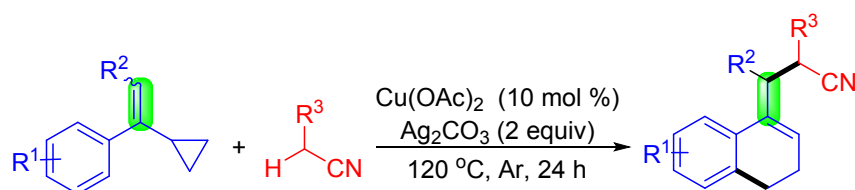


[MS Spectrum]	58.05	975	0.20	77.05	47592	9.86
# of Peaks400	59.15	743	0.15	78.05	14444	2.99
Raw Spectrum 11.110 (scan :	60.05	411	0.09	79.05	16750.35	
1423)	61.05	30470.63		80.55	23590.49	
Background No	62.05	13955	2.89	81.55	27129	5.62
Background Spectrum	63.05	45473	9.42	82.50	54774	11.35
Base Peakm/z 219.05 (Inten :	64.05	80541.67		83.45	10454	2.17
482,772)	65.05	23242	4.81	84.10	12280.25	
Event# 1	66.10	23800.49		85.05	18360.38	
m/z Absolute Intensity	67.10	665	0.14	86.05	77981.62	
Relative Intensity	68.55	18660.39		87.10	17471	3.62
50.00 22687 4.70	69.50	14231	2.95	88.10	29825	6.18
51.05 69942 14.49	70.45	21820.45		89.05	48106	9.96
52.05 15439 3.20	71.45	857	0.18	90.05	9441	1.96
53.00 29930.62	72.05	11920.25		91.10	56239	11.65
54.00 11180.23	73.05	27140.56		92.10	50781.05	
55.10 13420.28	74.05	17745	3.68	92.70	991	0.21
56.15 10090.21	75.05	25766	5.34	93.75	16341	3.38
57.15 22770.47	76.05	46222	9.57	94.65	67443	13.97

95.65	27097	5.61	141.05	165152	34.21	185.00	593	0.12
96.65	7584	1.57	142.05	23883	4.95	186.05	886	0.18
98.00	3935	0.82	143.10	2182	0.45	187.00	6095	1.26
99.05	5257	1.09	144.10	150	0.03	188.05	8072	1.67
100.05	5002	1.04	145.10	150	0.03	189.00	59847	12.40
101.10	10098	2.09	146.00	102	0.02	190.00	40743	8.44
102.10	29324	6.07	147.00	594	0.12	191.05	165850	34.35
103.10	32668	6.77	148.00	188	0.04	192.05	219272	45.42
104.10	5242	1.09	149.00	2937	0.61	193.05	36882	7.64
105.10	1121	0.23	150.05	13335	2.76	194.05	2645	0.55
106.65	605	0.13	151.05	23097	4.78	195.15	367	0.08
107.65	4063	0.84	152.05	48566	10.06	196.10	102	0.02
108.60	40481	8.39	153.05	8145	1.69	197.10	94	0.02
109.55	5774	1.20	154.05	1591	0.33	198.00	180	0.04
111.05	3226	0.67	155.00	230	0.05	198.95	308	0.06
112.10	1828	0.38	156.00	87	0.02	199.95	1723	0.36
113.10	27033	5.60	157.00	167	0.03	201.00	20089	4.16
114.10	36295	7.52	158.00	185	0.04	202.05	33701	6.98
115.10	101881	21.10	159.00	66	0.01	203.05	45737	9.47
116.10	30108	6.24	160.00	106	0.02	204.05	28782	5.96
117.15	9584	1.99	161.05	1630	0.34	205.05	5140	1.06
118.10	894	0.19	161.95	3210	0.66	205.95	521	0.11
119.10	359	0.07	163.00	23935	4.96	206.95	6373	1.32
120.10	191	0.04	164.00	24841	5.15	208.00	1252	0.26
121.00	236	0.05	165.05	125614	26.02	208.95	1005	0.21
122.00	1006	0.21	166.05	21171	4.39	209.90	276	0.06
123.10	1031	0.21	167.05	8906	1.84	211.00	199	0.04
124.05	1020	0.21	168.05	1310	0.27	212.00	716	0.15
125.05	2696	0.56	169.00	116	0.02	213.05	480	0.10
126.10	12452	2.58	170.00	47	0.01	214.00	8465	1.75
127.10	7554	1.56	171.00	57	0.01	215.05	3555	0.74
128.10	9086	1.88	172.00	70	0.01	216.00	15254	3.16
129.10	1177	0.24	173.00	108	0.02	217.05	90363	18.72
129.90	137	0.03	174.00	1764	0.37	218.05	320654	66.42
130.90	345	0.07	175.05	4435	0.92	219.05	482772	100.00
131.90	111	0.02	176.00	37662	7.80	220.05	82650	17.12
133.10	1145	0.24	177.05	30254	6.27	221.10	71991.49	
134.00	489	0.10	178.05	114315	23.68	221.90	554	0.11
135.10	969	0.20	179.05	43849	9.08	222.90	183	0.04
135.85	469	0.10	180.05	6456	1.34	223.90	108	0.02
137.05	4530	0.94	181.00	628	0.13	224.90	49	0.01
138.05	3677	0.76	182.00	130	0.03	225.90	89	0.02
139.05	24246	5.02	183.00	185	0.04	226.90	78	0.02
140.05	114076	23.63	184.00	129	0.03	227.90	57	0.01

228.90	46	0.01	236.90	36	0.01	244.90	86	0.02
229.90	52	0.01	237.90	108	0.02	245.90	58	0.01
230.90	162	0.03	238.90	55	0.01	246.90	30	0.01
231.90	81	0.02	239.90	73	0.02	247.90	58	0.01
232.90	33	0.01	240.90	29	0.01	248.90	282	0.06
233.90	73	0.02	241.90	47	0.01	249.90	106	0.02
234.90	65	0.01	242.90	60	0.01	250.90	231	0.05
235.90	103	0.02	243.90	58	0.01	251.90	68	0.01

2.3 Typical Experimental Procedure for the Synthesis of 1-cyanoethylated 3,4-dihydronaphthalenes.



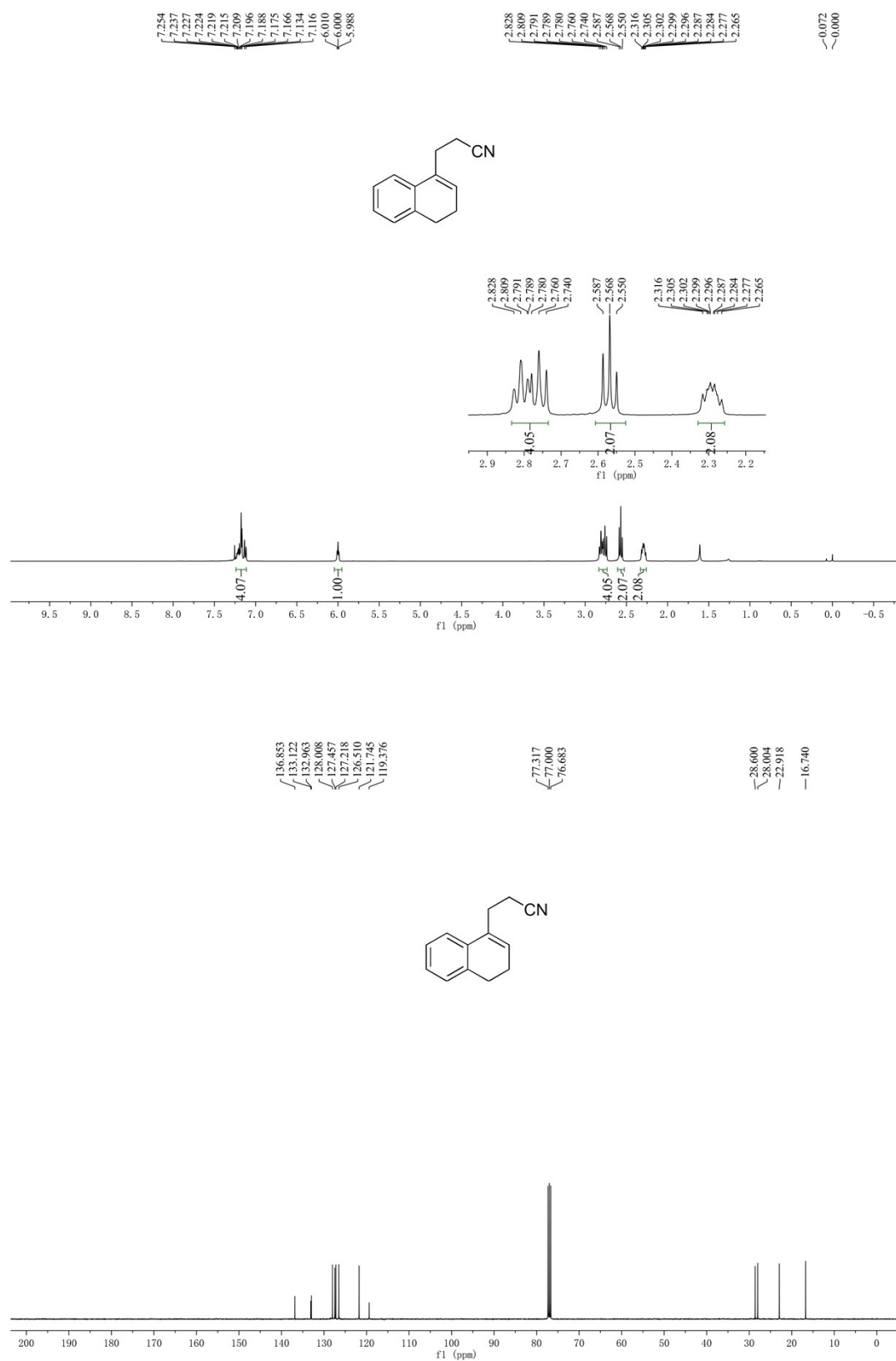
To a Schlenk tube were added VCP **1** (0.2 mmol, 0.1 M), alkyl nitriles **2a** (2 mL), Cu(OAc)₂ (4 mg, 10 mol %) and Ag₂CO₃ (110.3 mg, 2 equiv). Then the tube was stirred at 120 °C (oil bath temperature) in argon atmosphere for 24 h until complete consumption of starting material as monitored by TLC and/or GC-MS analysis. After the reaction was finished, the reaction mixture removal of the solvent, the crude product was purified by column chromatography (petroleum ether/ethyl acetate, 20 : 1) to provide the desired products **3**. An amplified experiment conducted in the presence of VCP **1a** (1g, 6.94 mmol), Cu(OAc)₂ (126.1 mg, 10 mol %), Ag₂CO₃ (3.83 g, 2 equiv) and CH₃CN (50 mL) at 120 °C under argon atmosphere for 72 h could successfully afford the desired product **3aa** in 52% yield (660.8 mg).

3. References

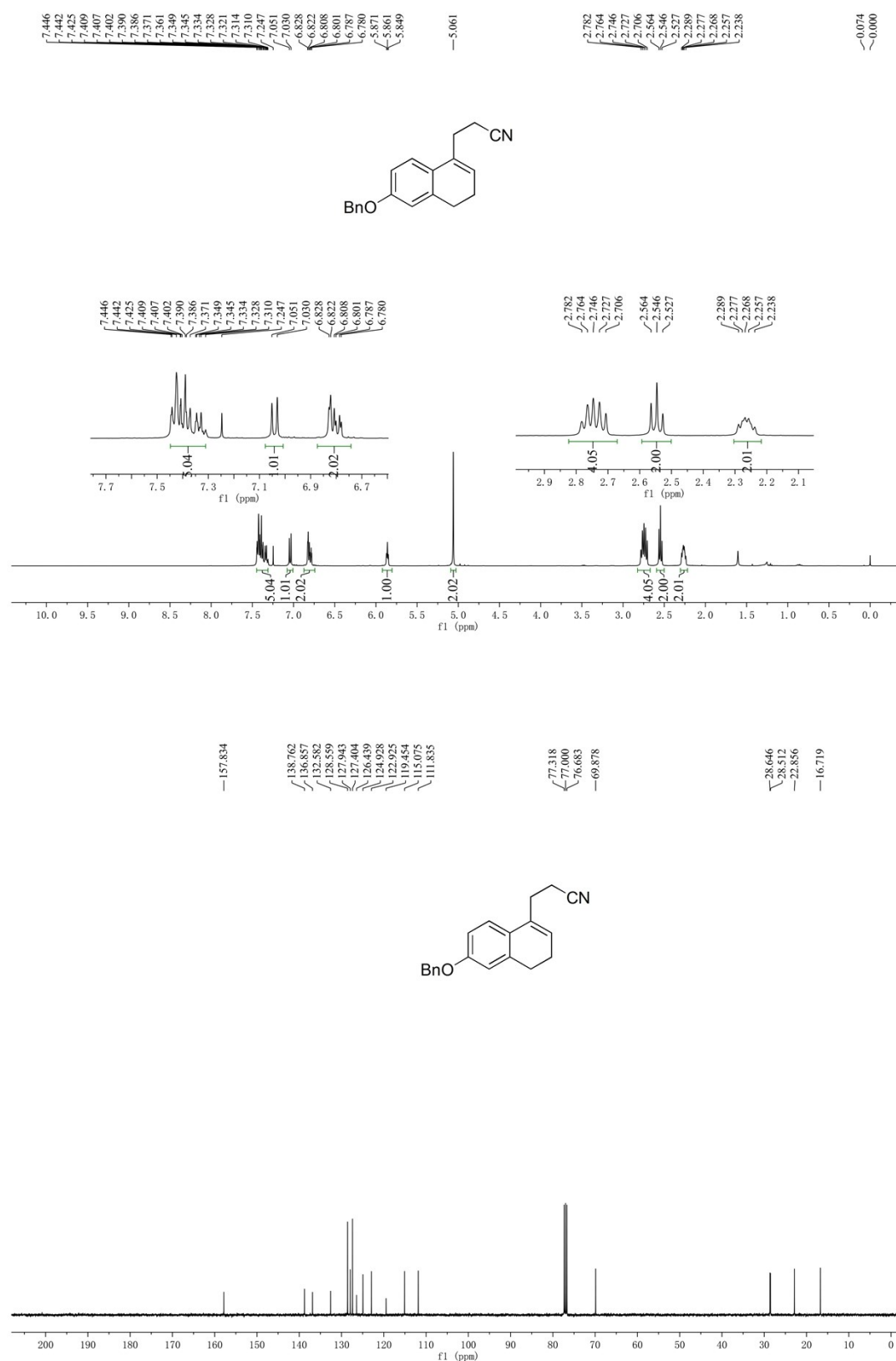
- [1] Li, J.; Chen, J.-Z.; Jiao, W.; Wang, G.-Q.; Li, Y.; Cheng, X.; Li, G.-G. Difluoroalkylation/C–H Annulation Cascade Reaction Induced by Visible-Light Photoredox Catalysis. *J. Org. Chem.* **2016**, *81*, 9992–10001.

4. Spectra

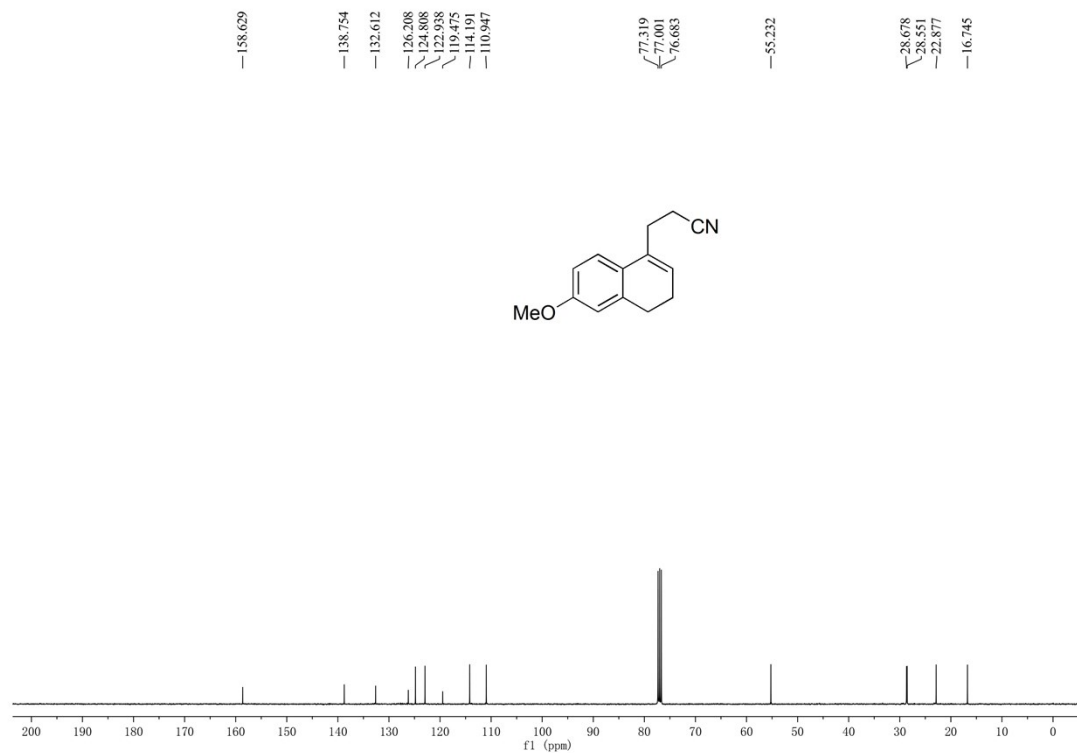
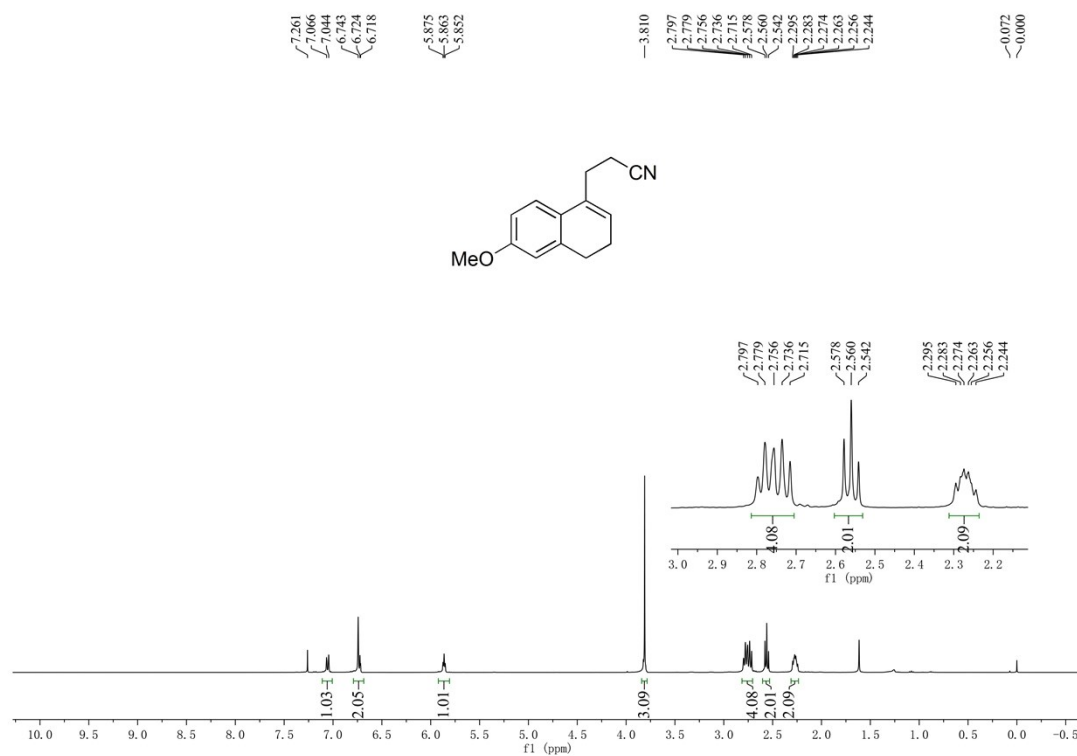
3-(3,4-Dihydronaphthalen-1-yl)propanenitrile (3aa)



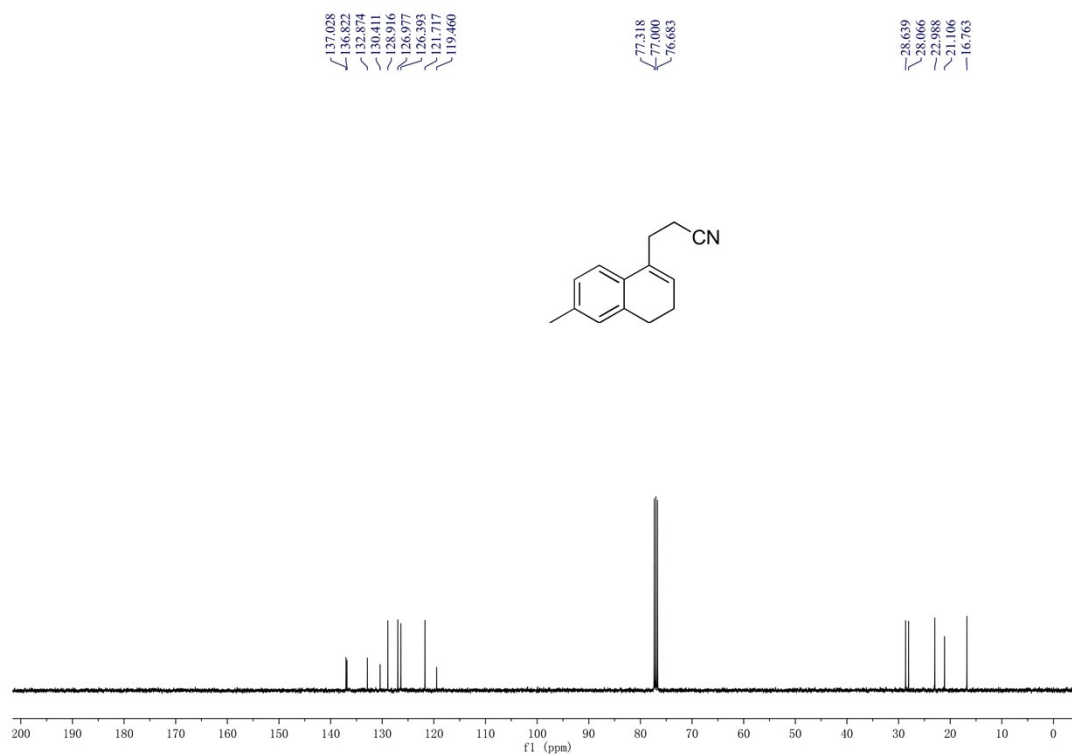
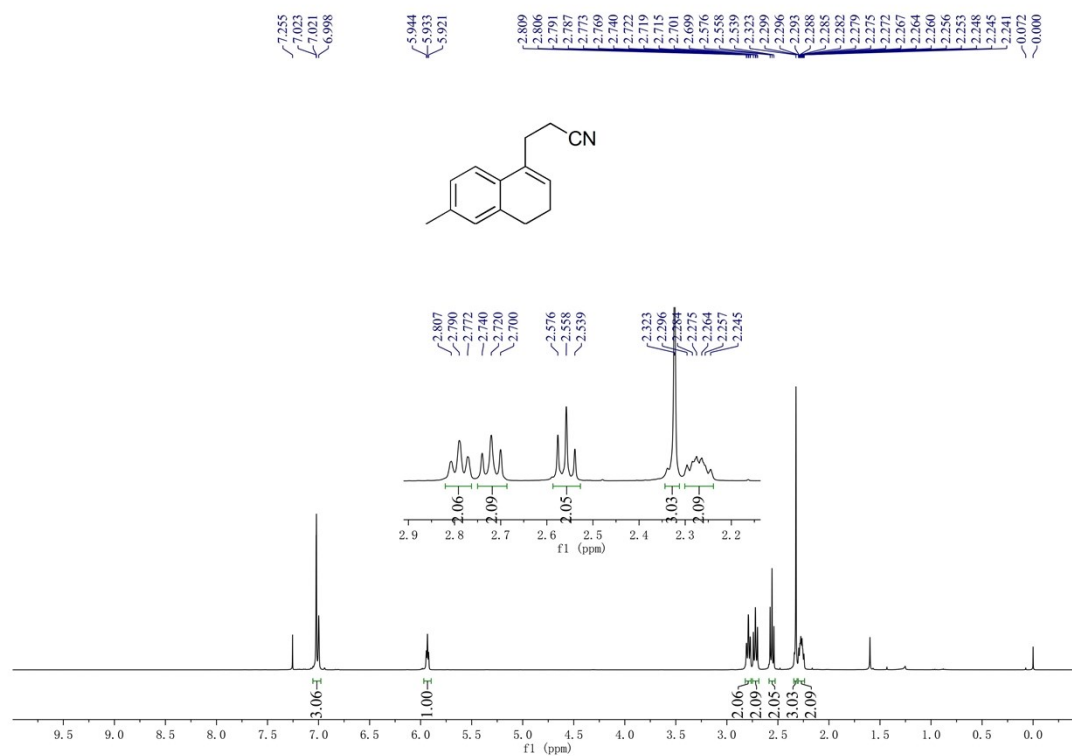
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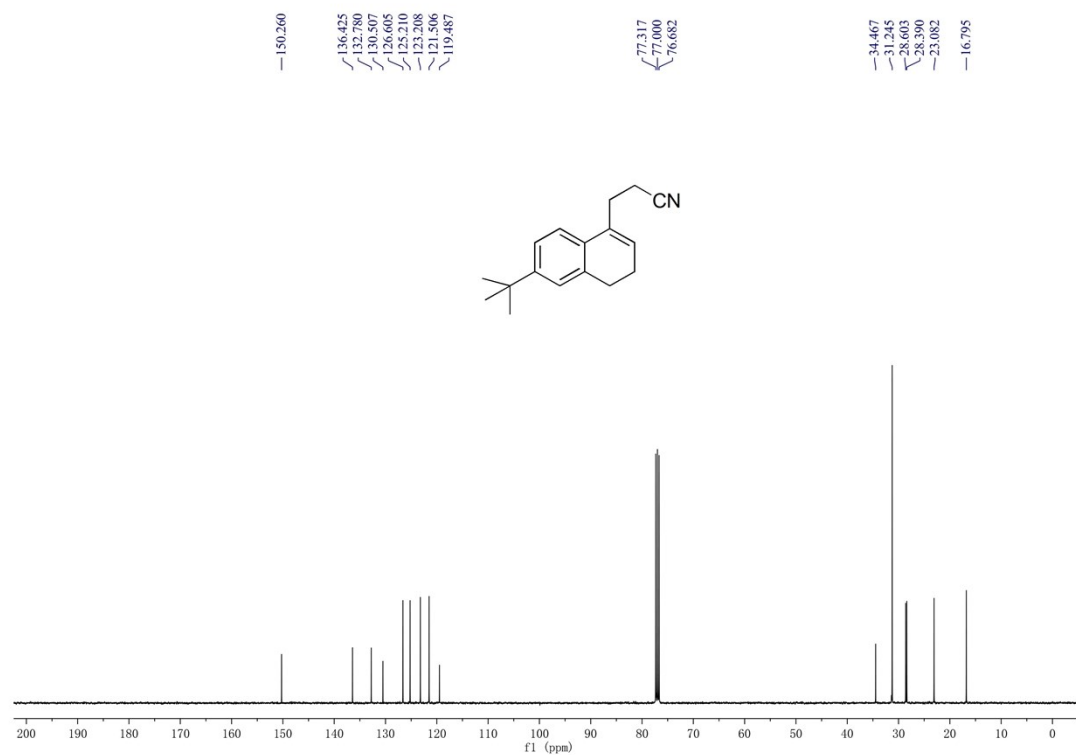
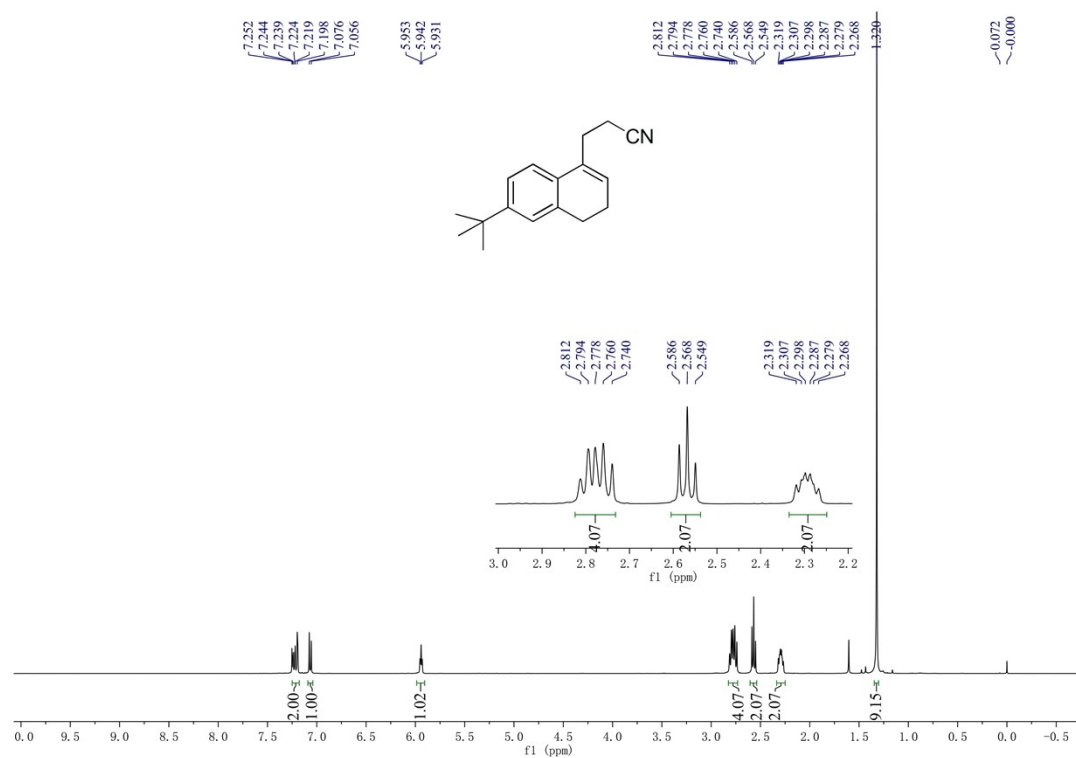
3-(6-Methoxy-3,4-dihydronaphthalen-1-yl)propanenitrile (3ca)



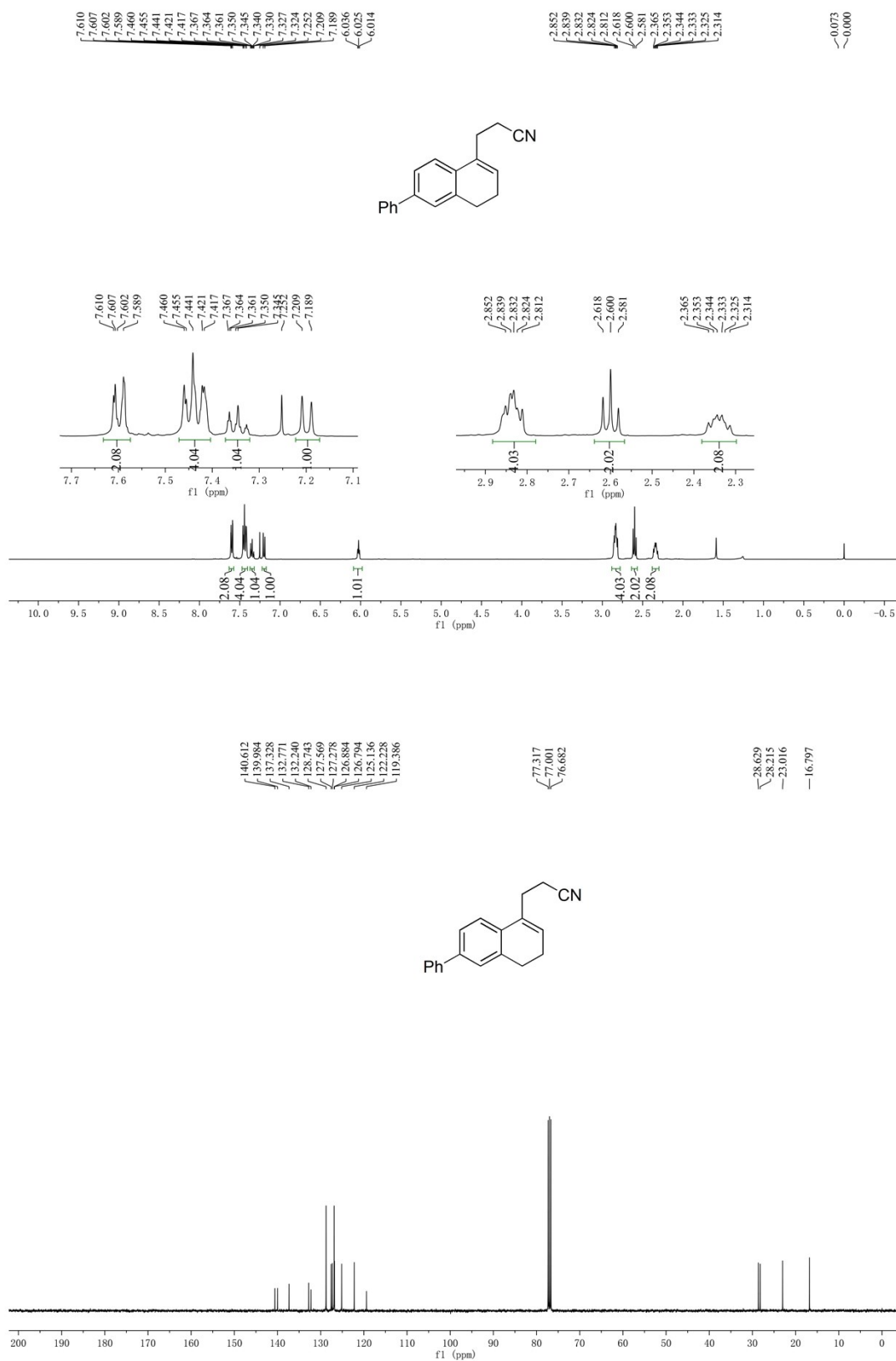
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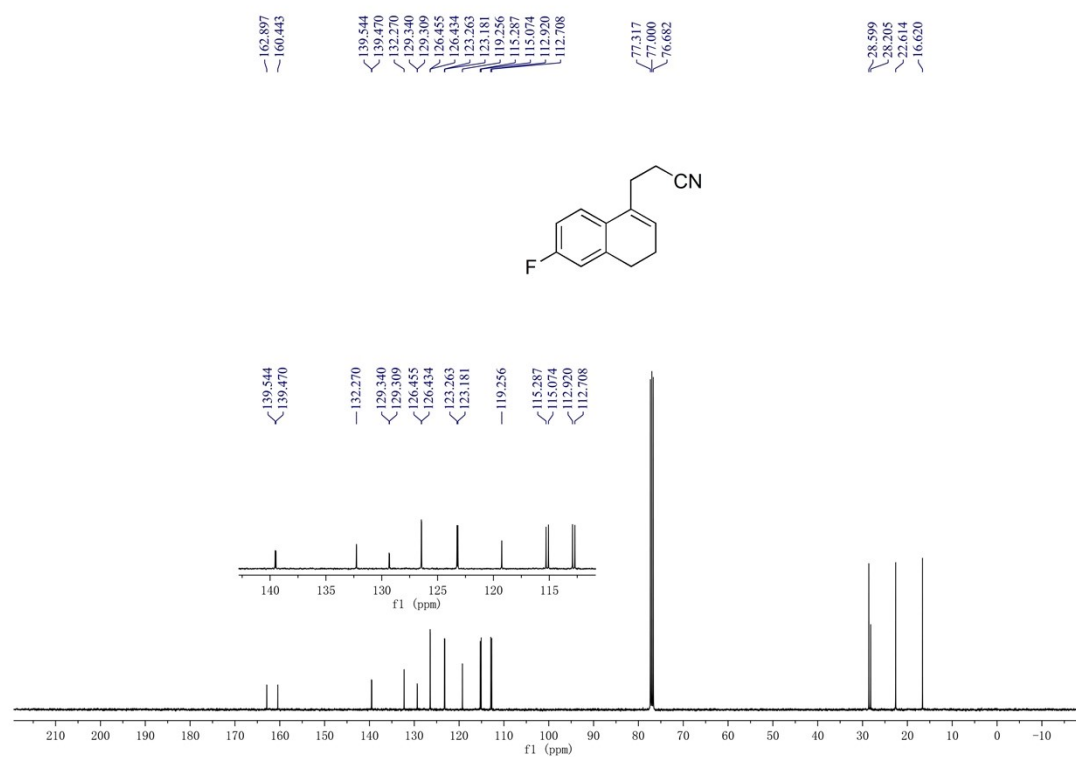
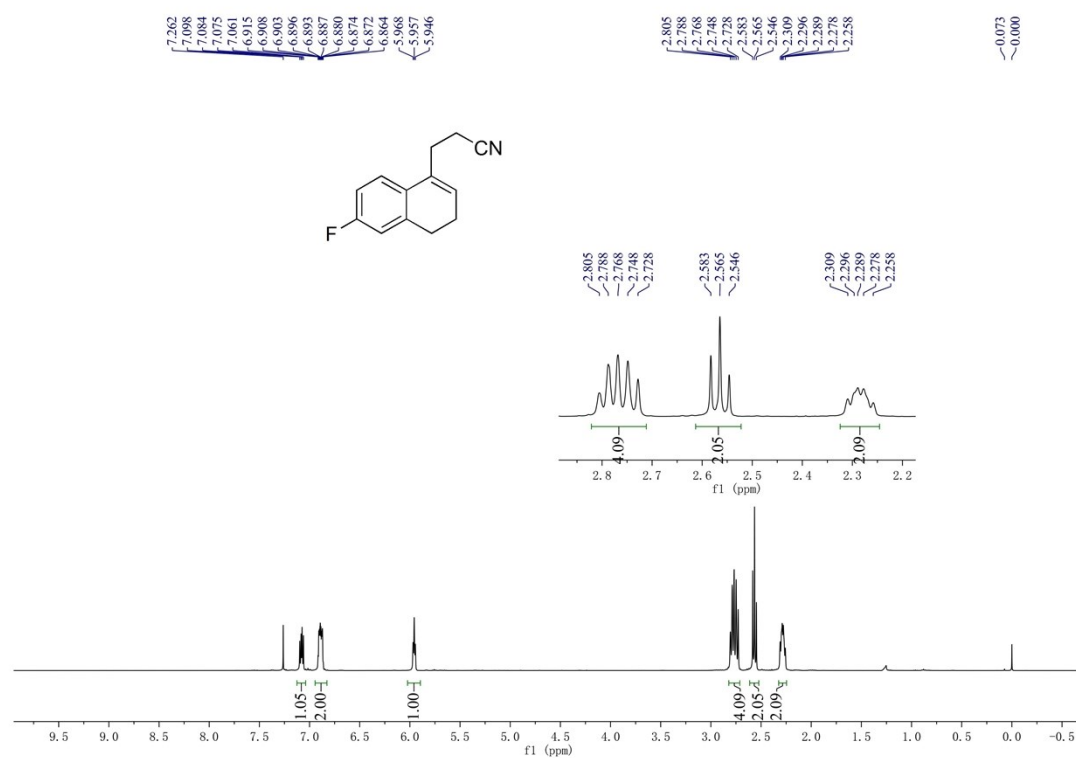
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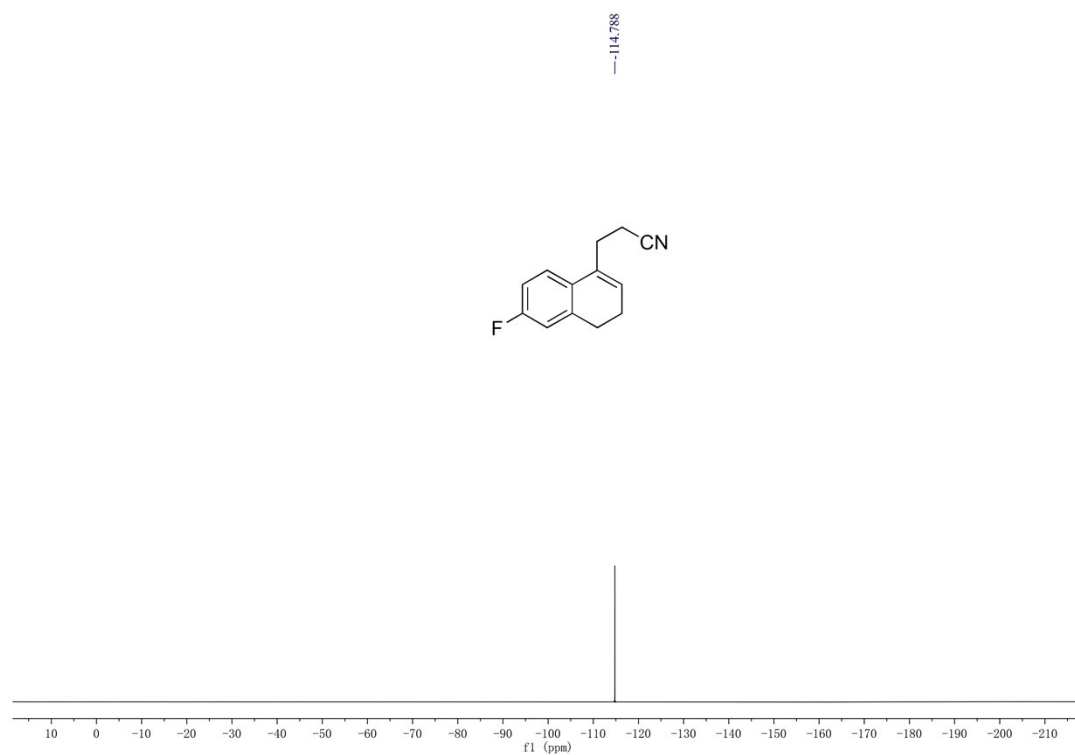


3-(6-Phenyl-3,4-dihydronaphthalen-1-yl)propanenitrile (3fa)

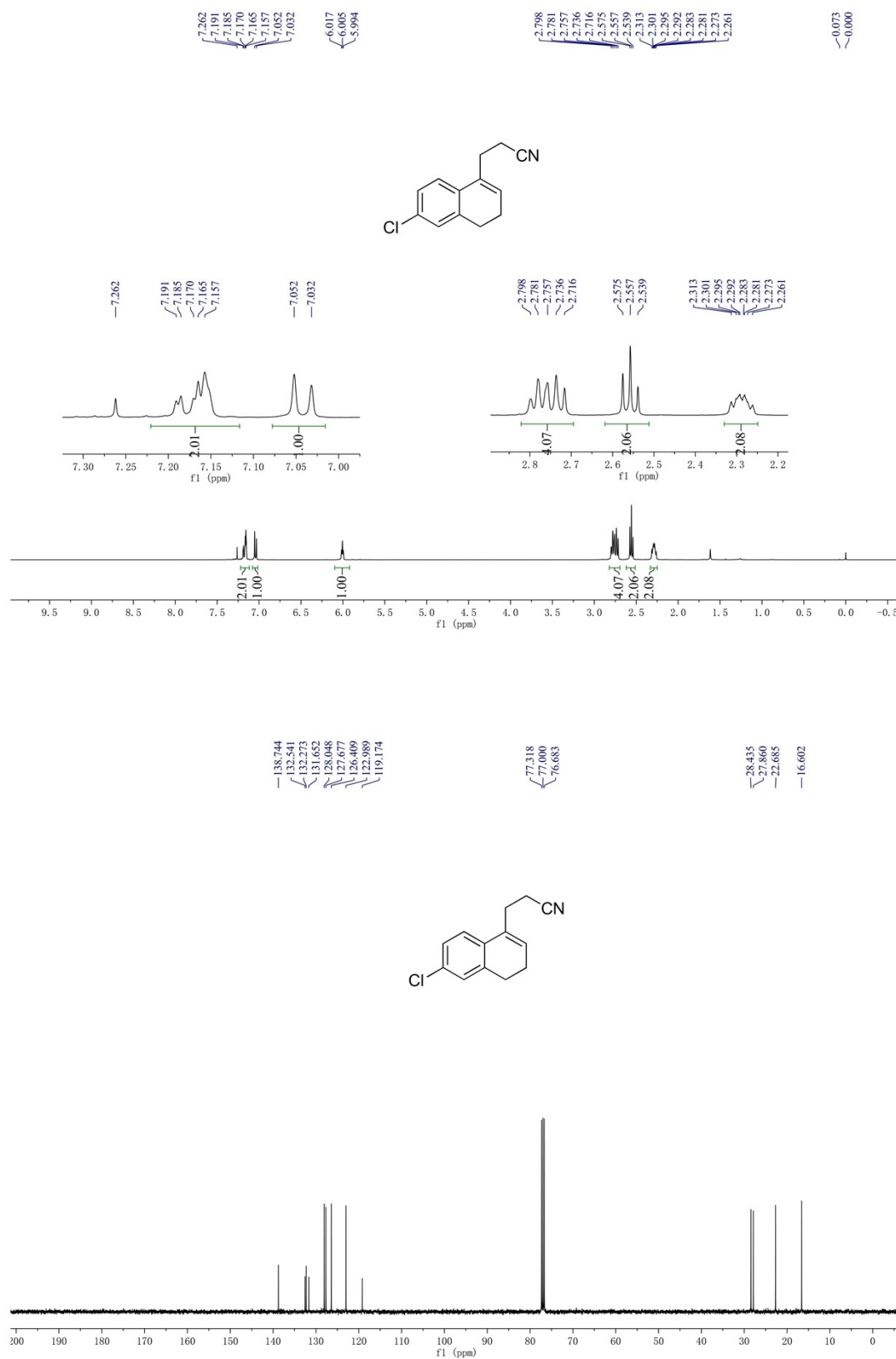


3-(6-Fluoro-3,4-dihydronaphthalen-1-yl)propanenitrile (3ga)

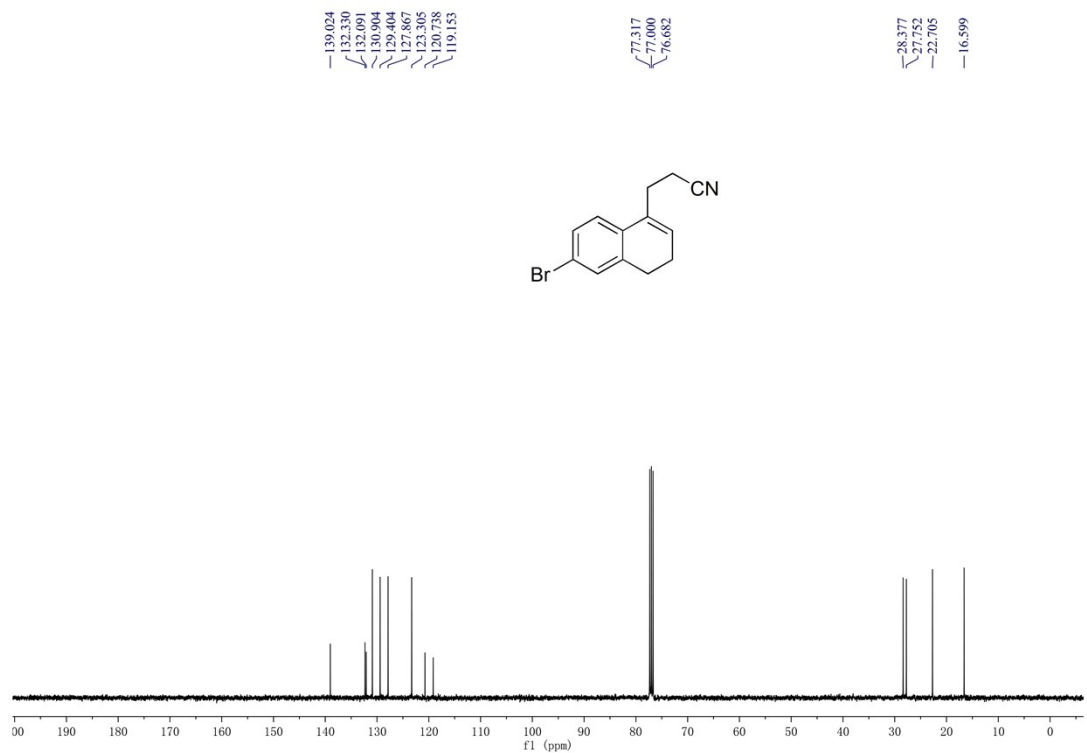
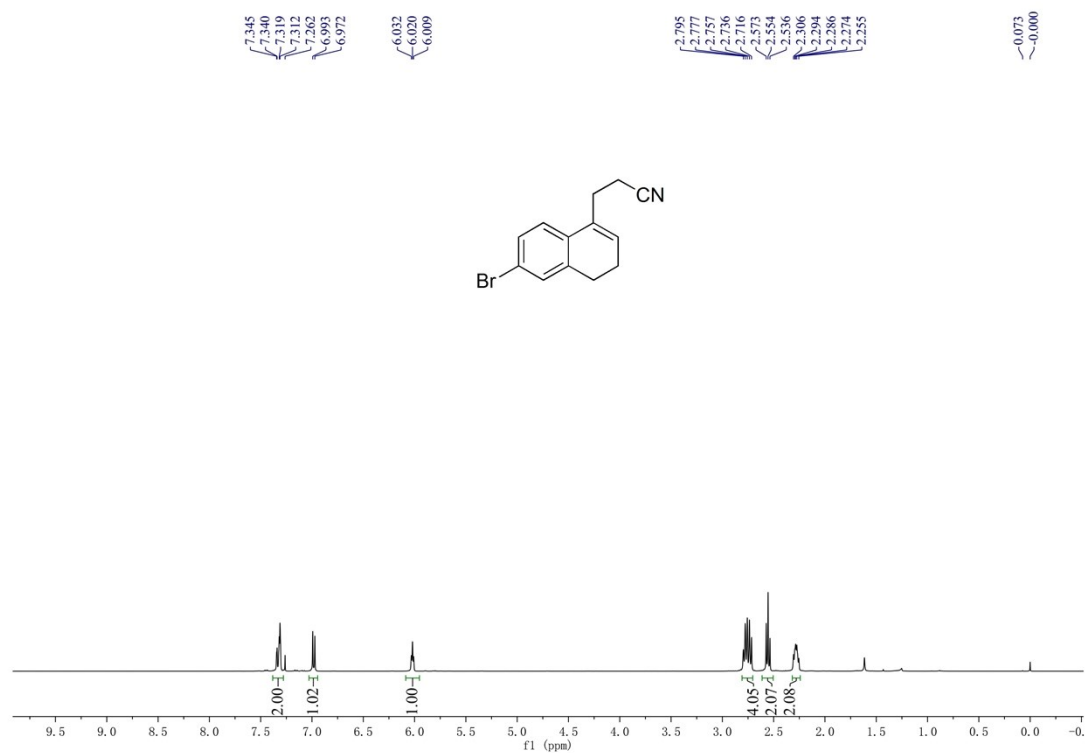




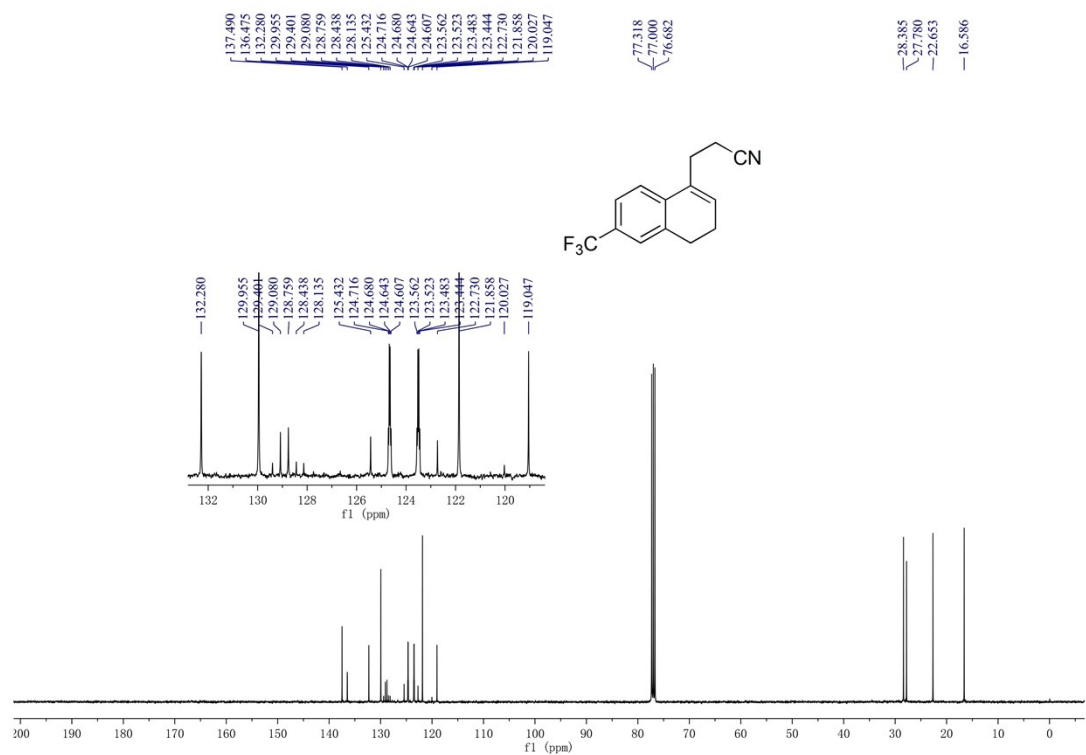
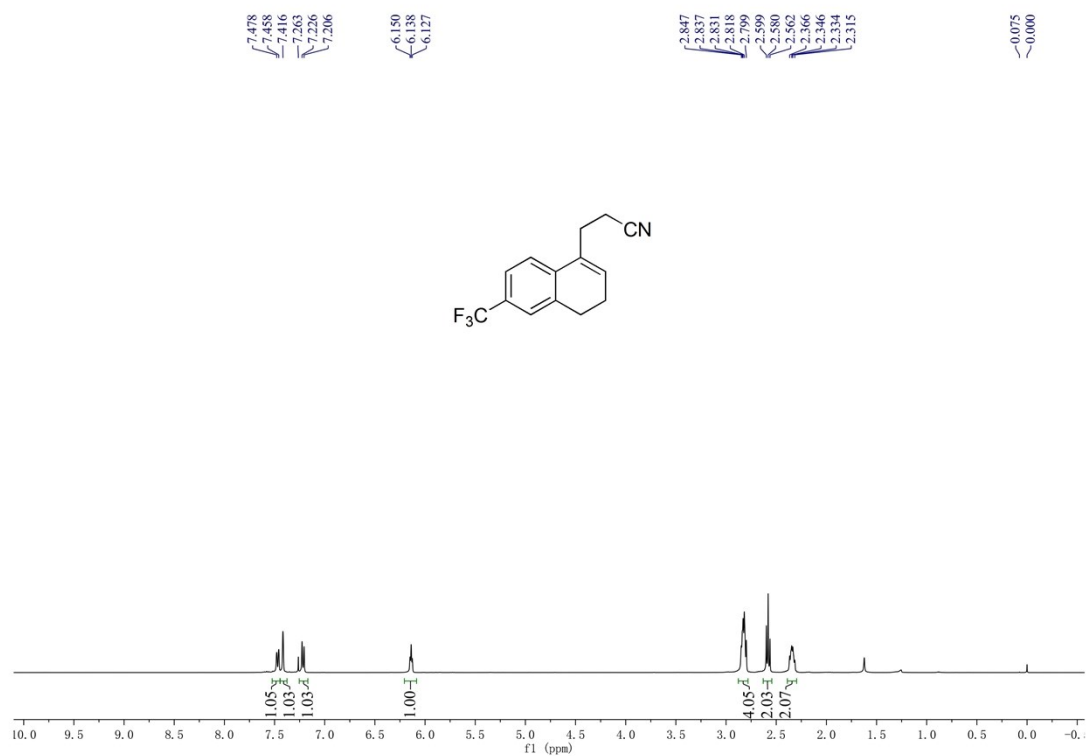
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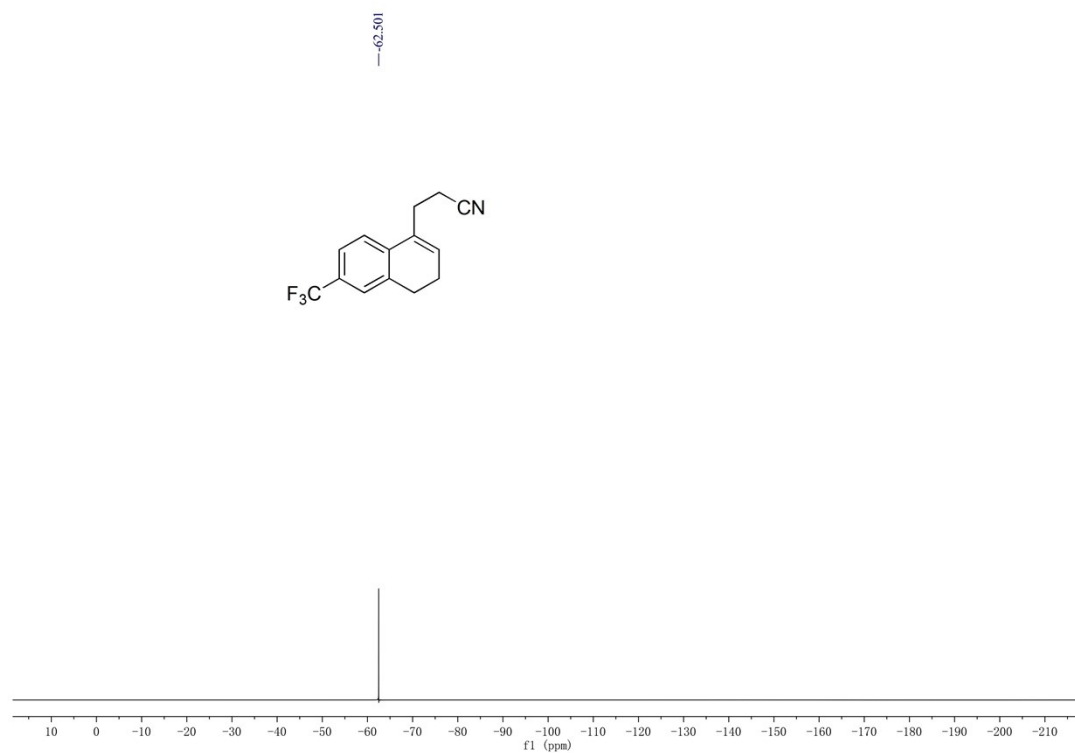


3-(6-Bromo-3,4-dihydronaphthalen-1-yl)propanenitrile (3ia)

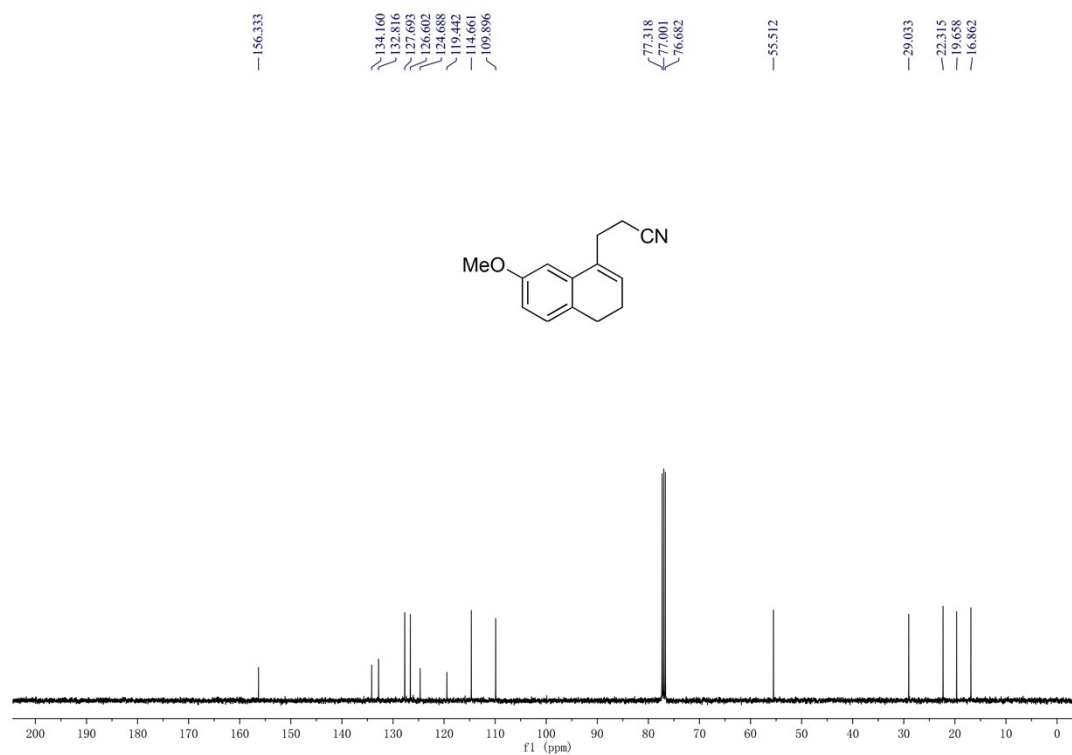
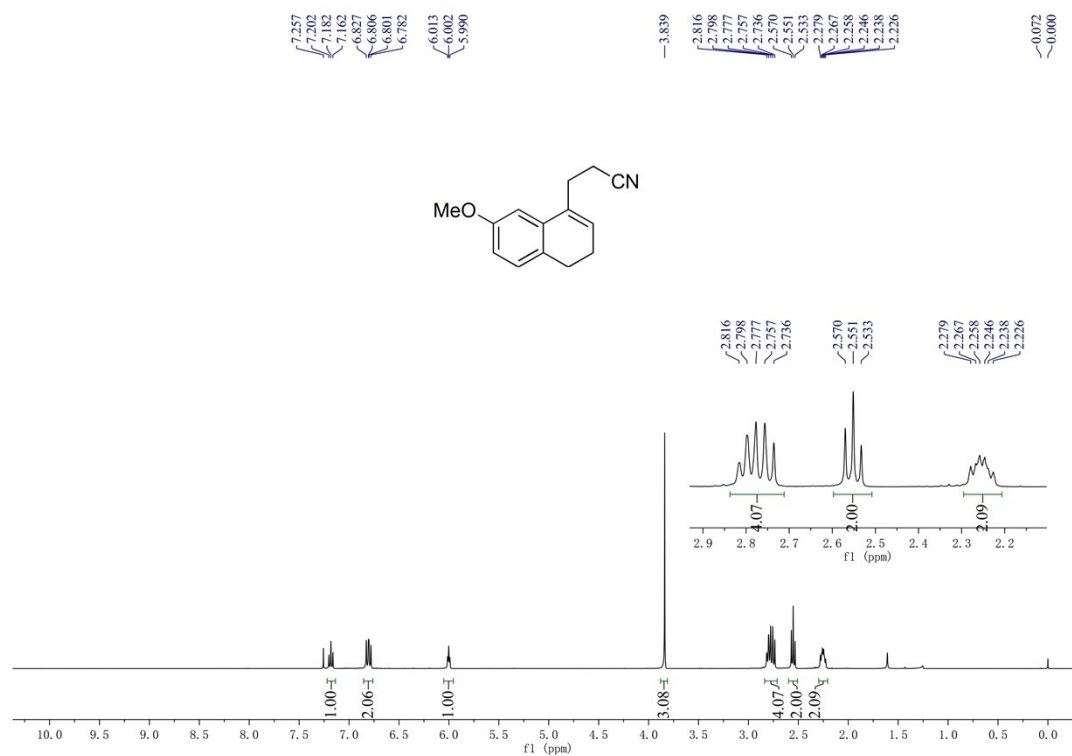


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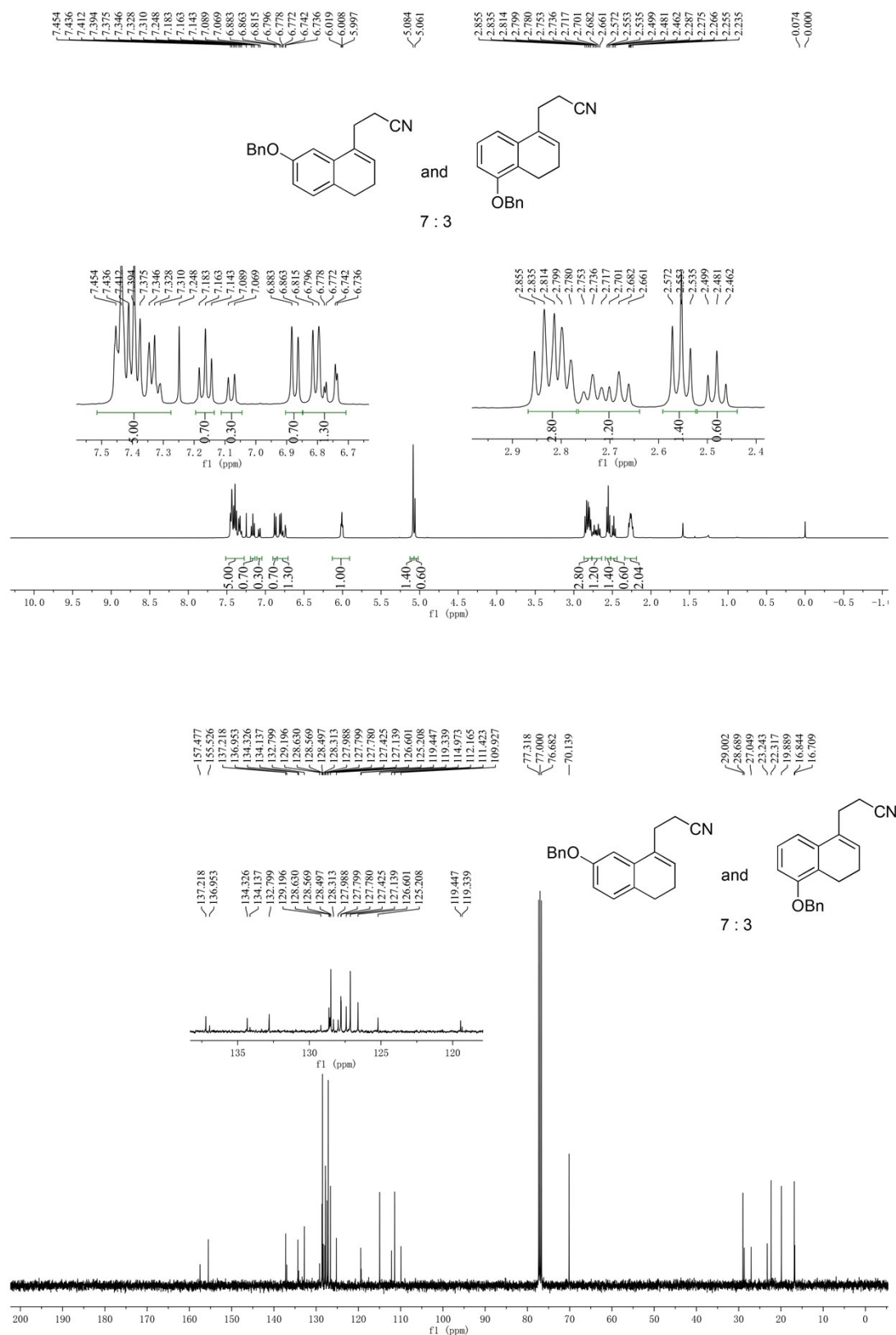




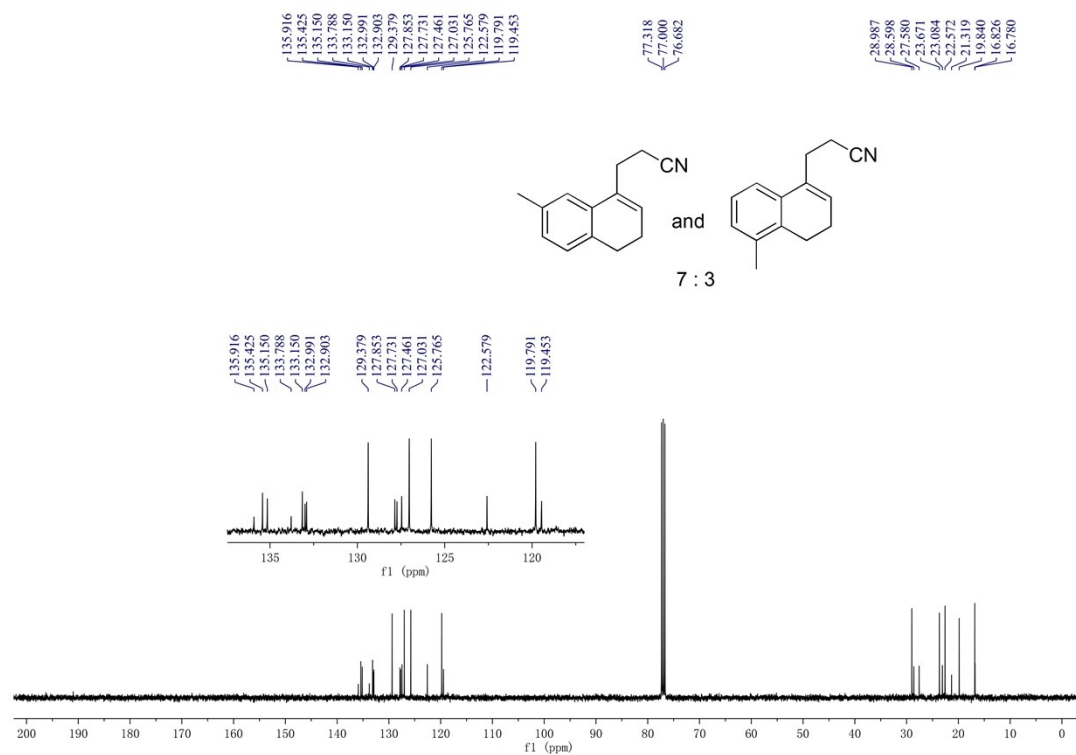
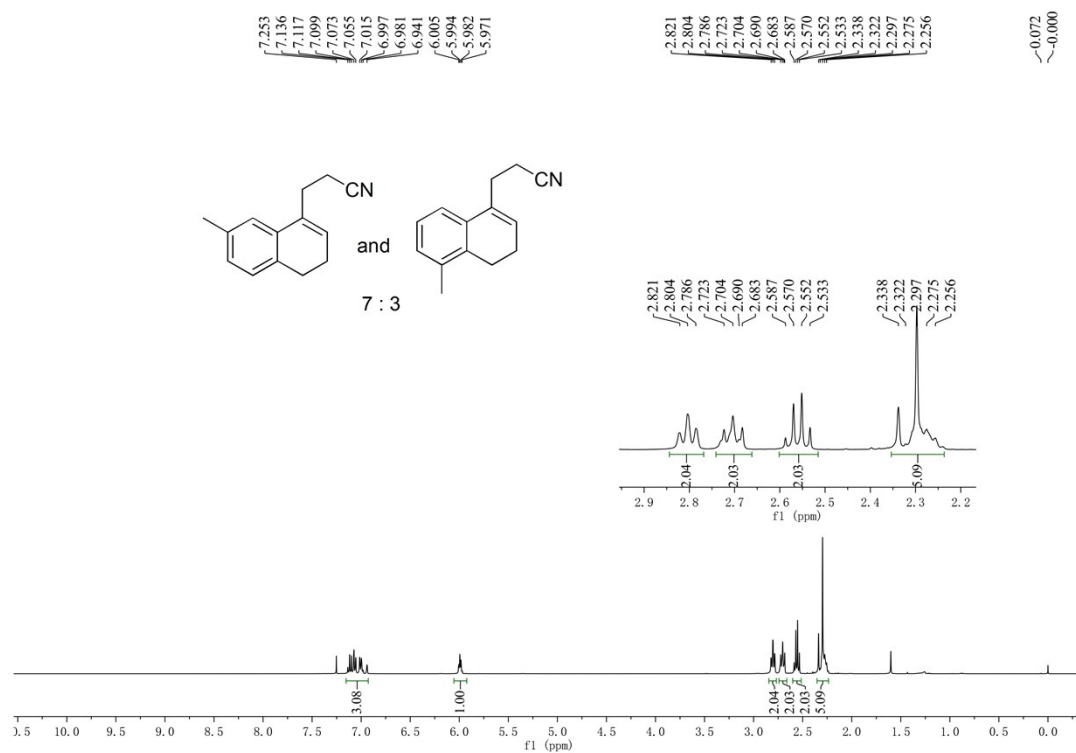
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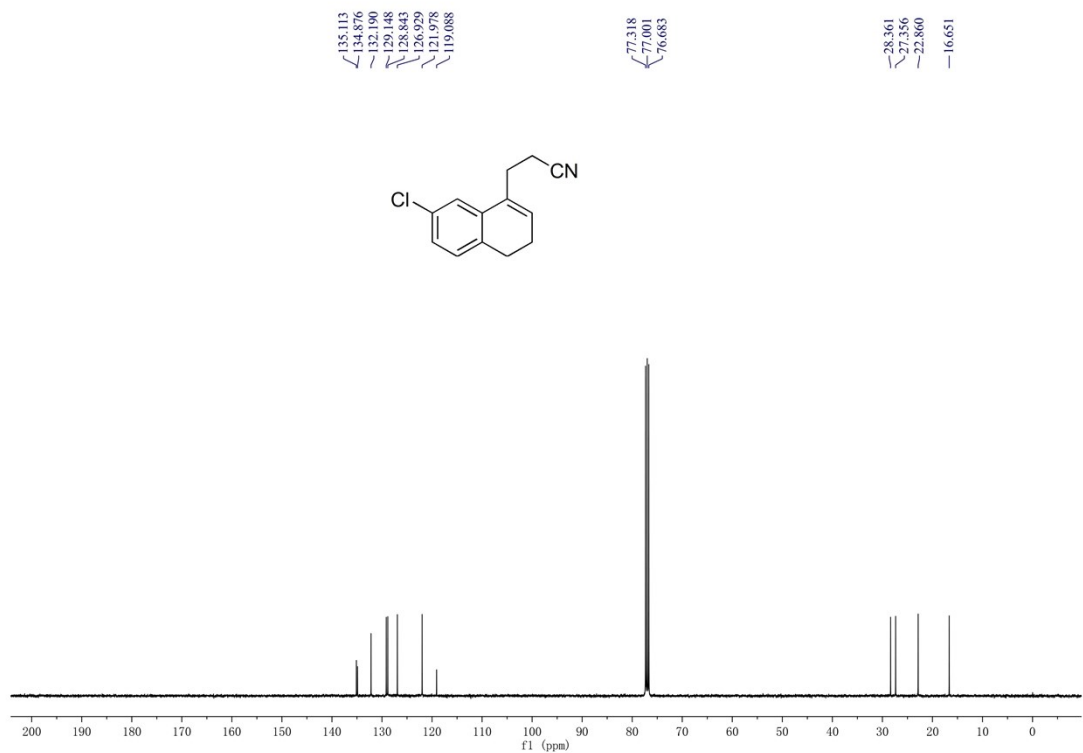
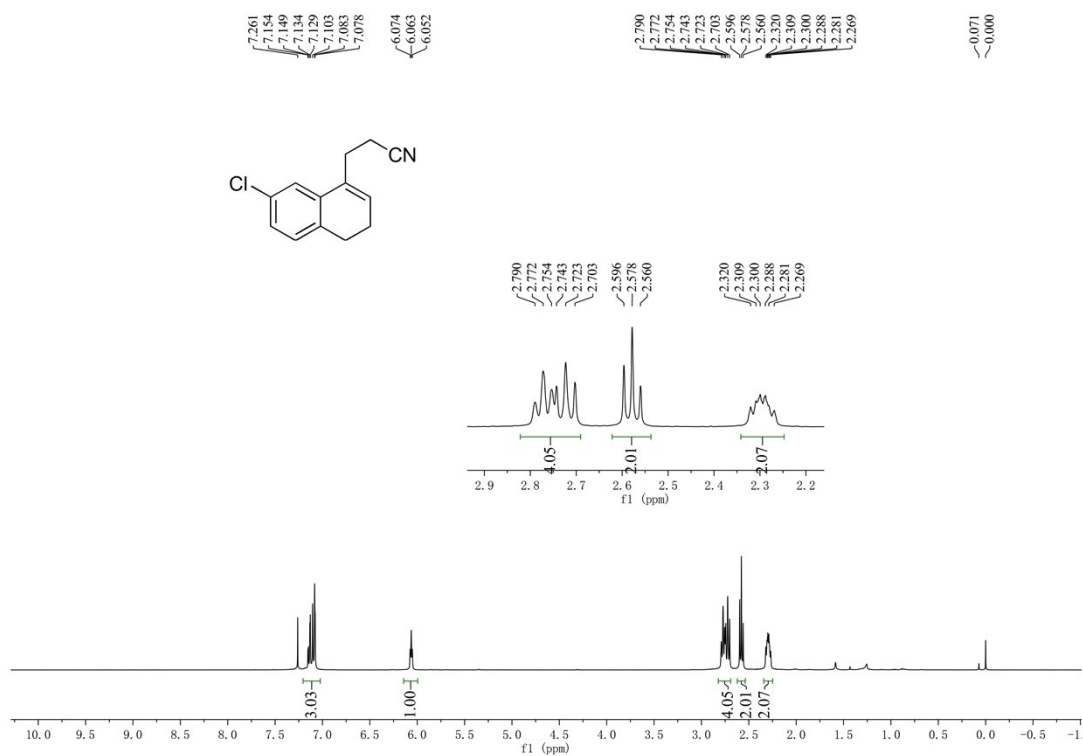
3-(7-Bbenzyloxy)-3,4-dihydronaphthalen-1-yl)propanenitrile (3la) and 3-(5-(benzyloxy)-3,4-dihydronaphthalen-1-yl)propanenitrile (3la')



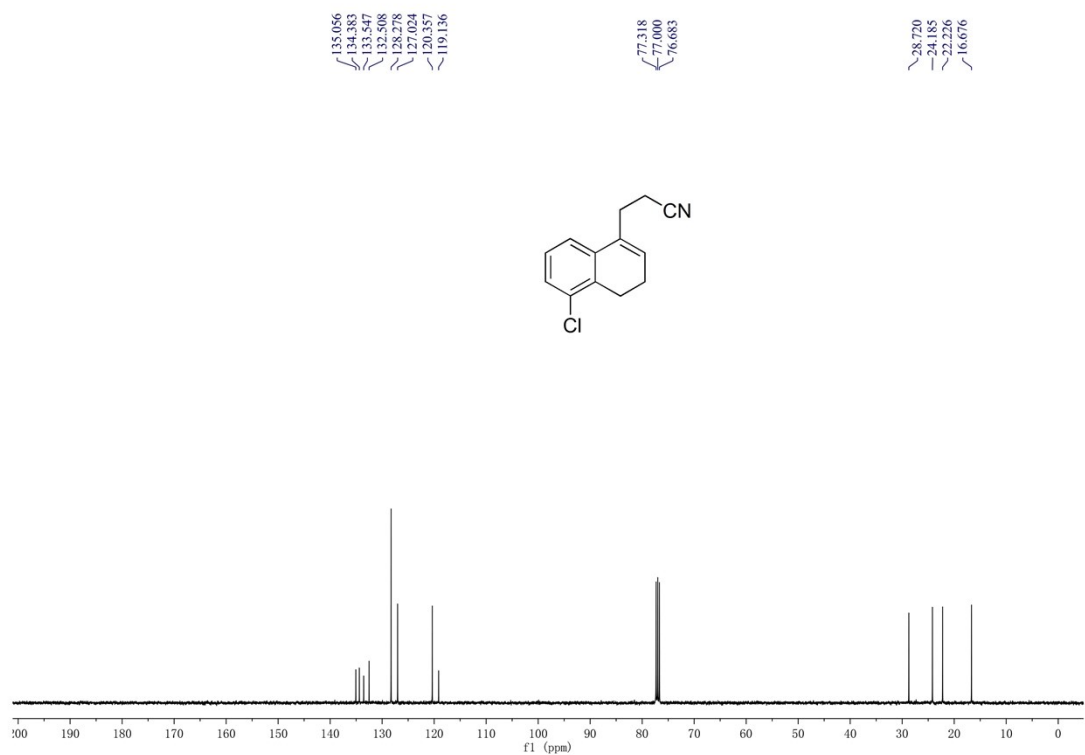
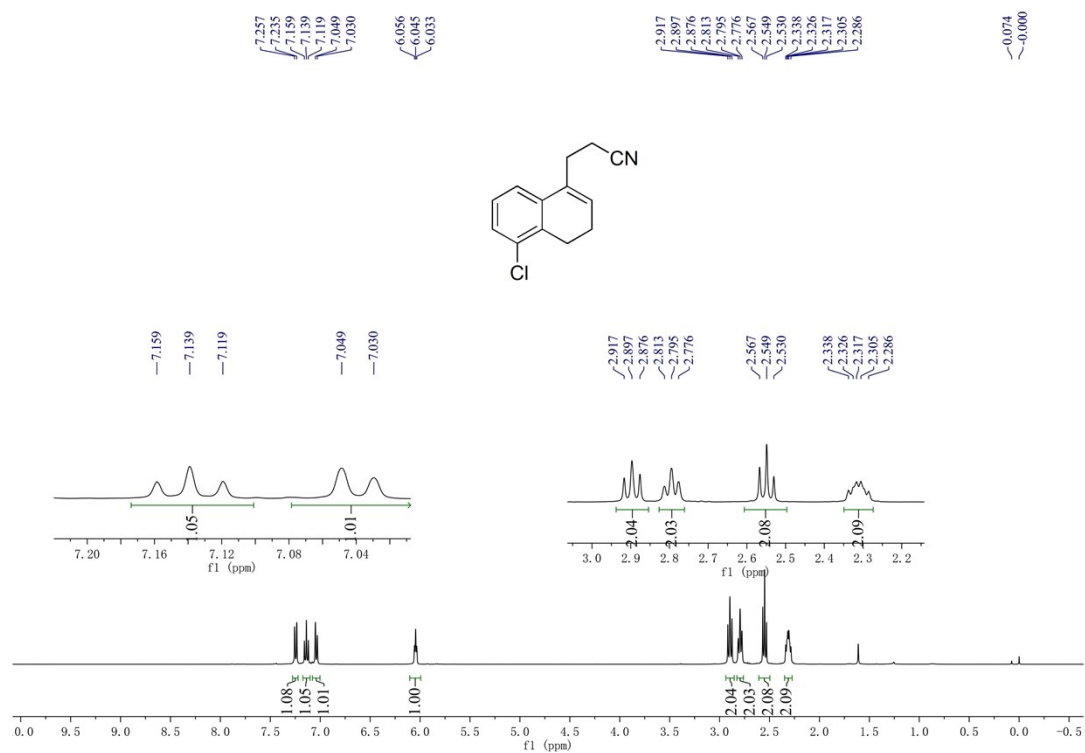
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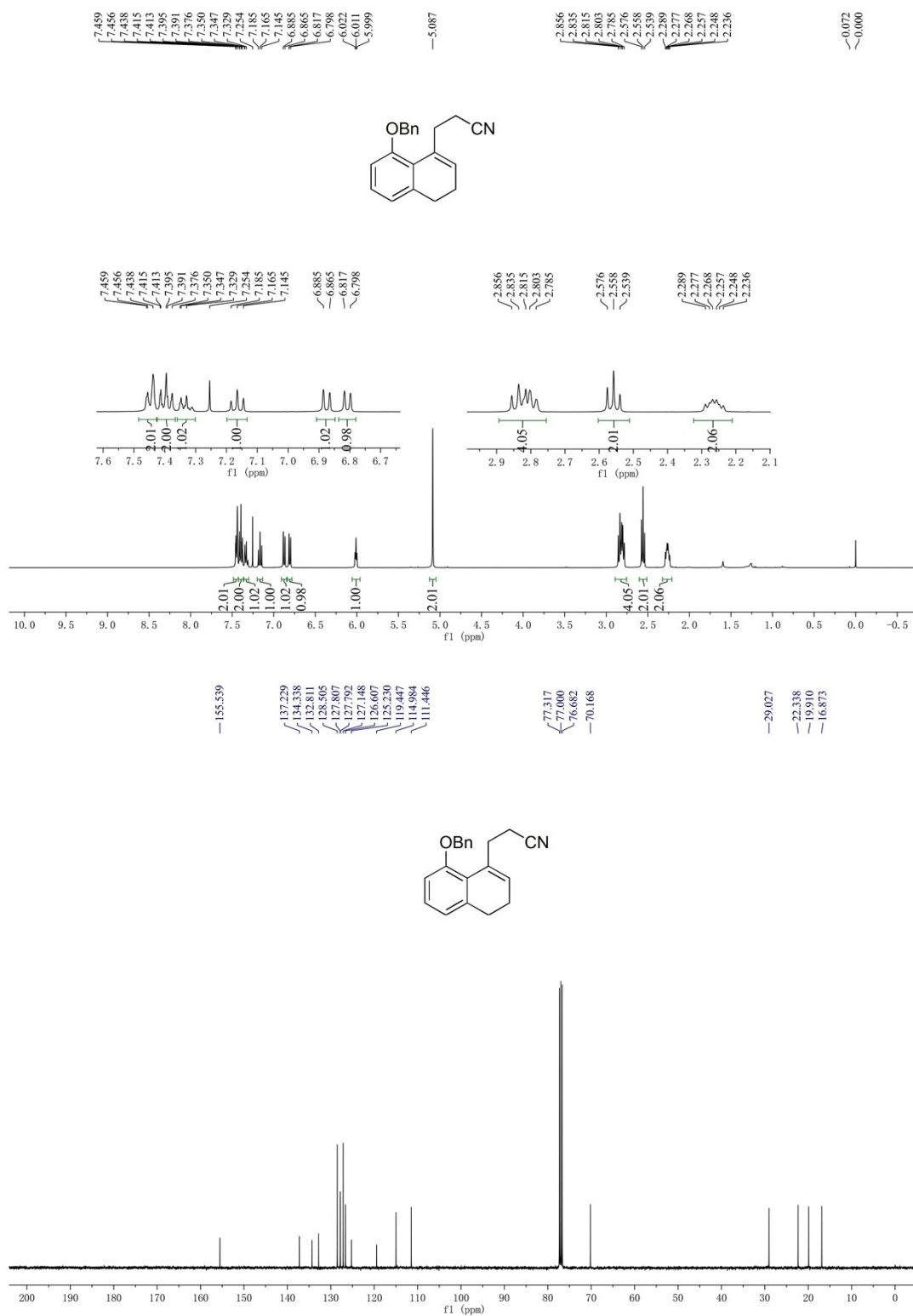
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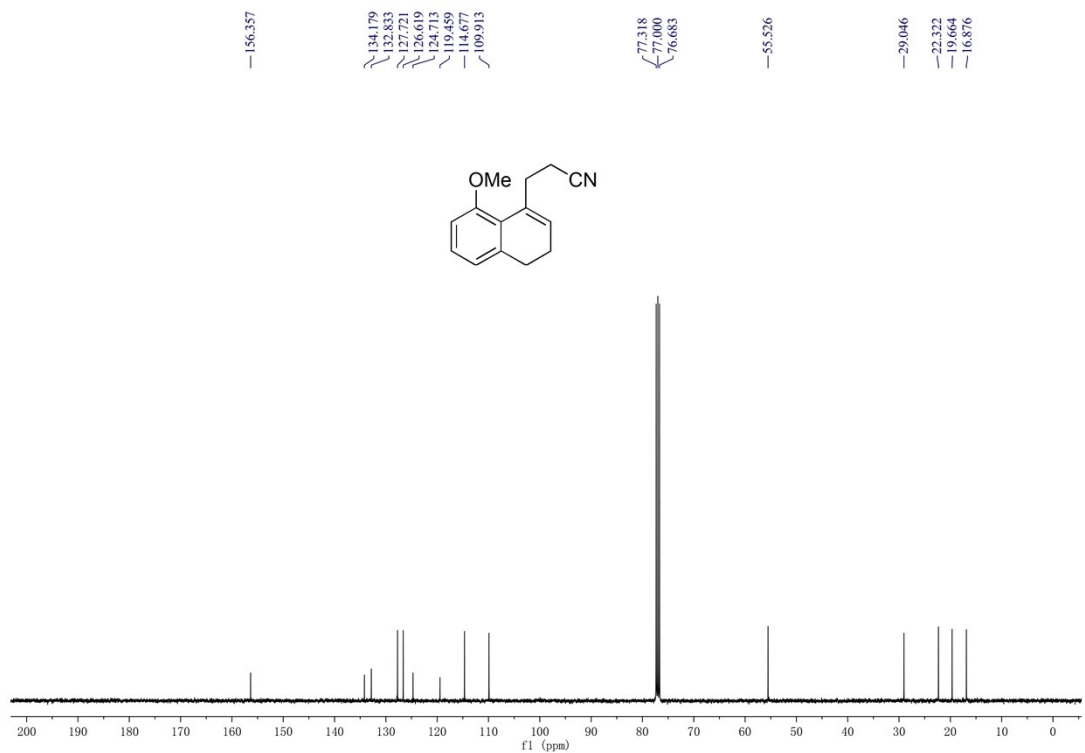
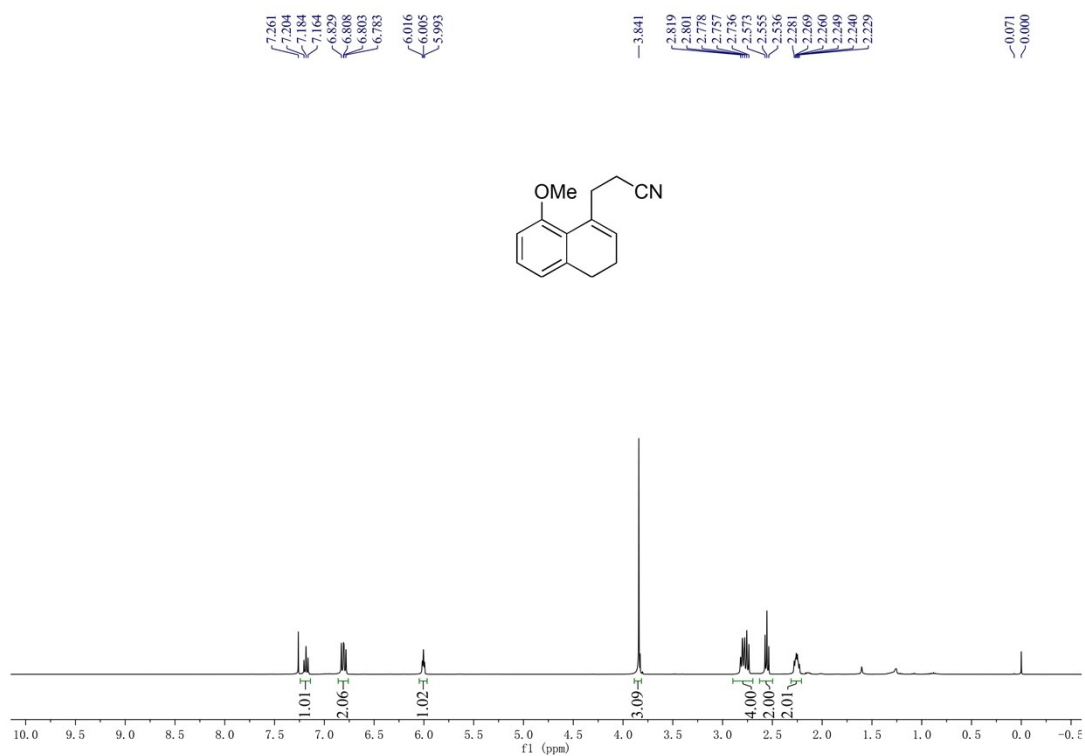
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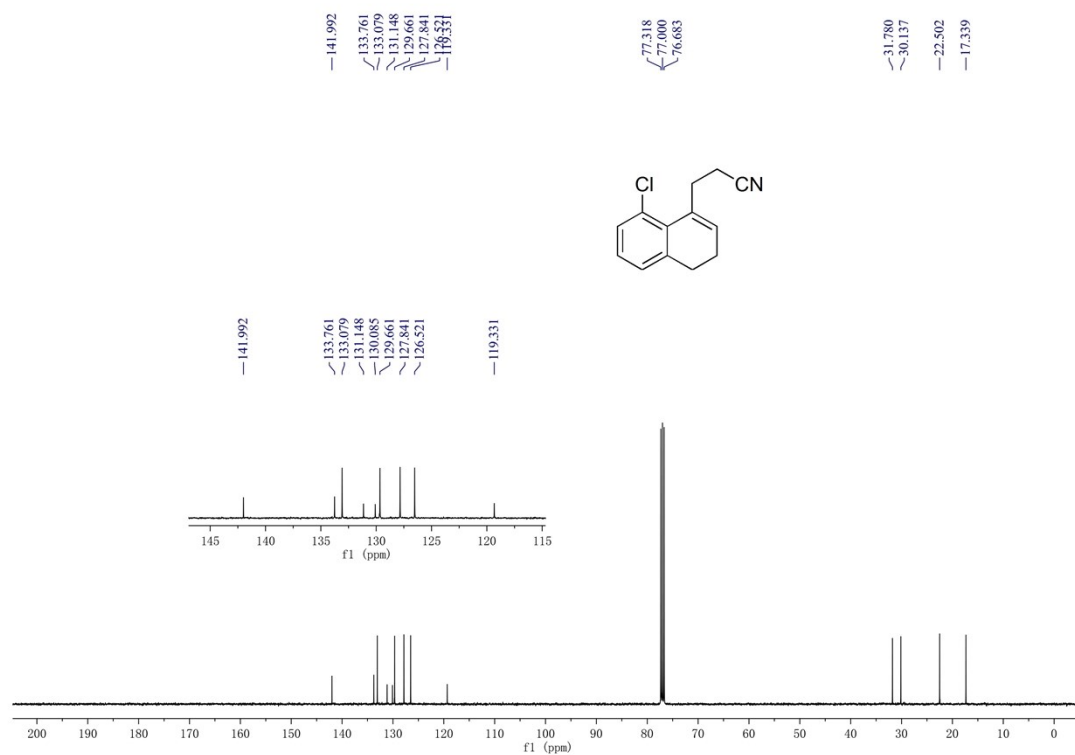
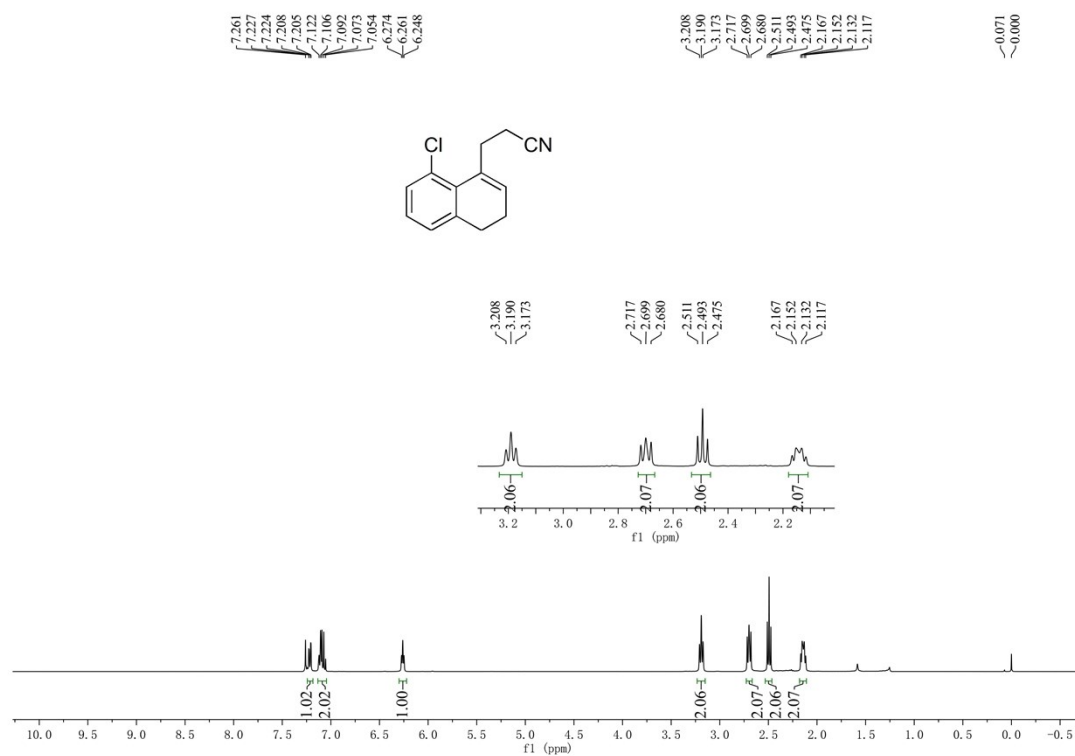
3-(8-(Benzyloxy)-3,4-dihydronaphthalen-1-yl)propanenitrile (3oa)



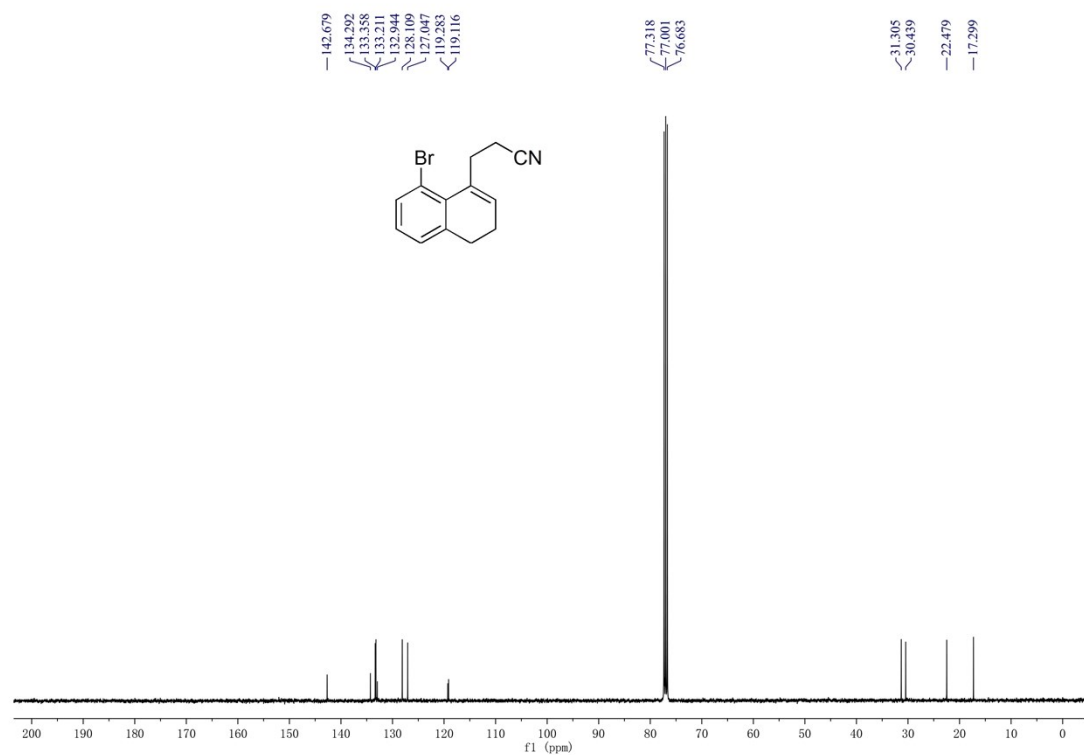
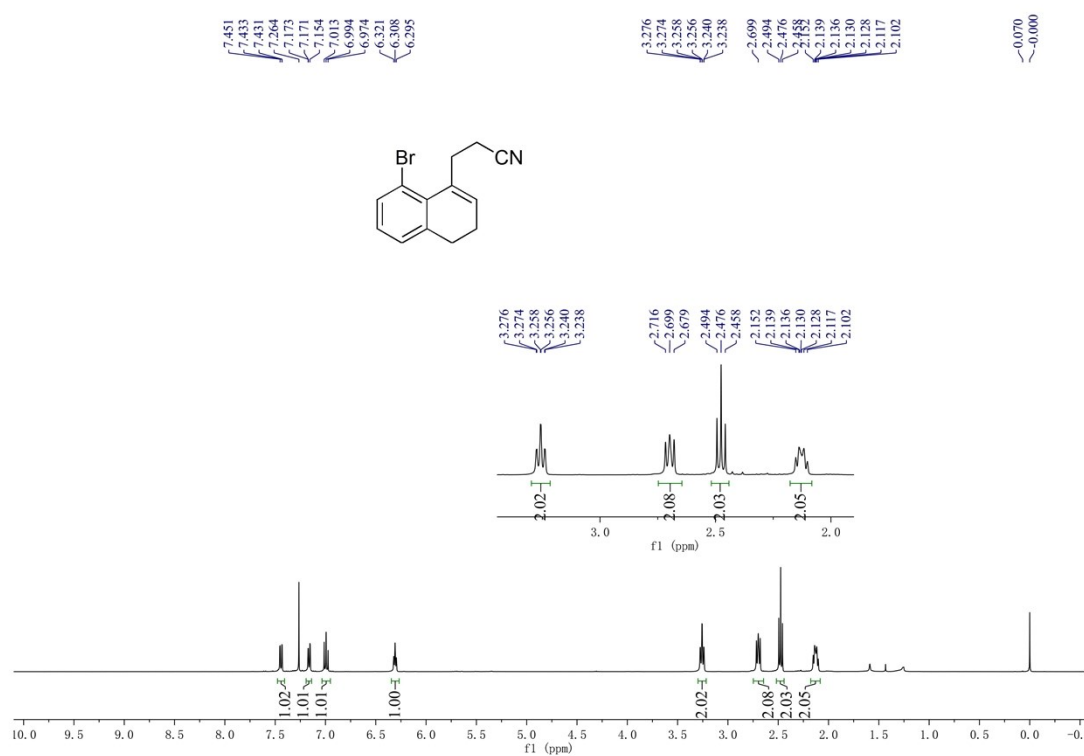
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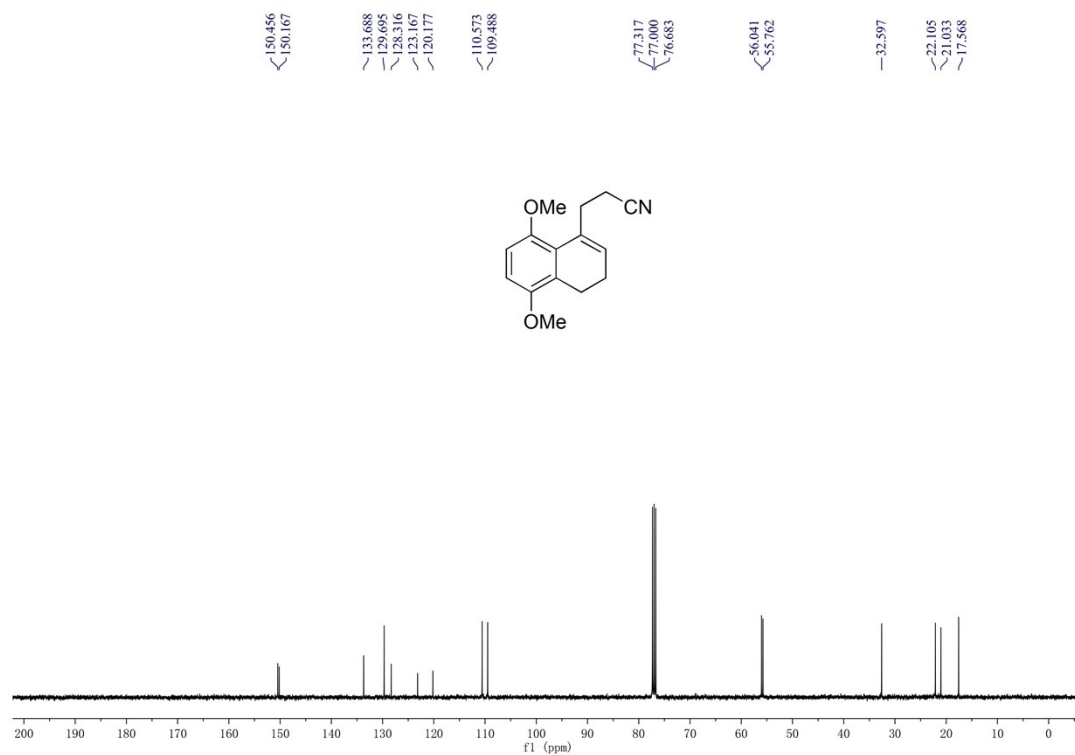
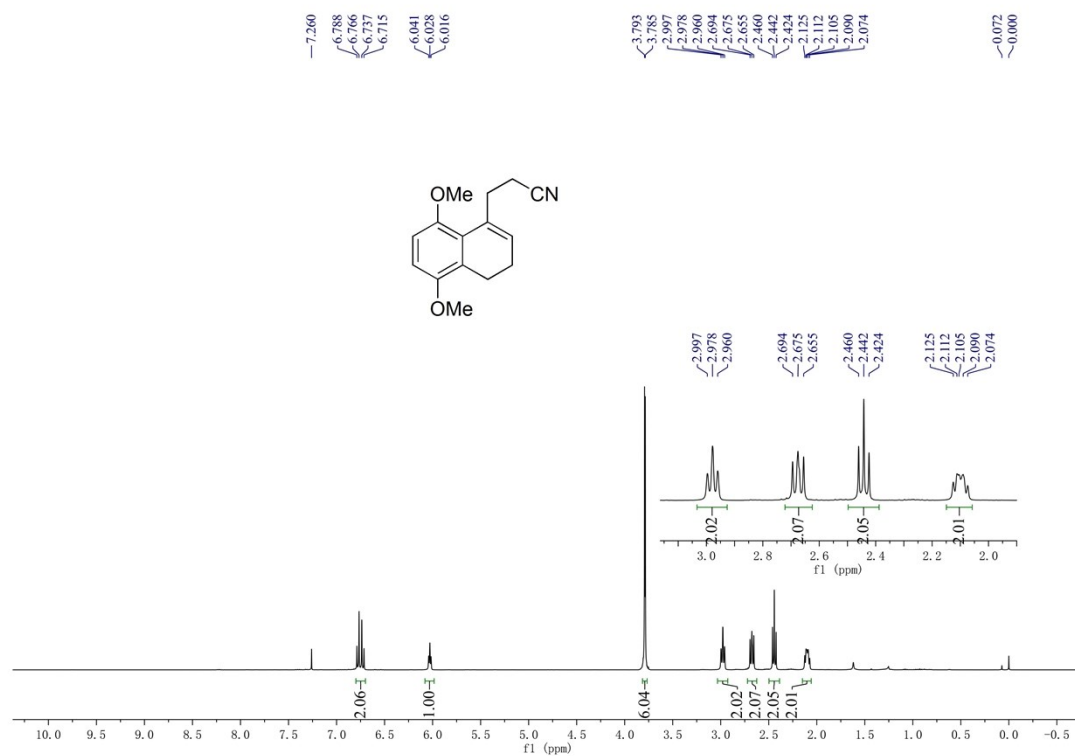
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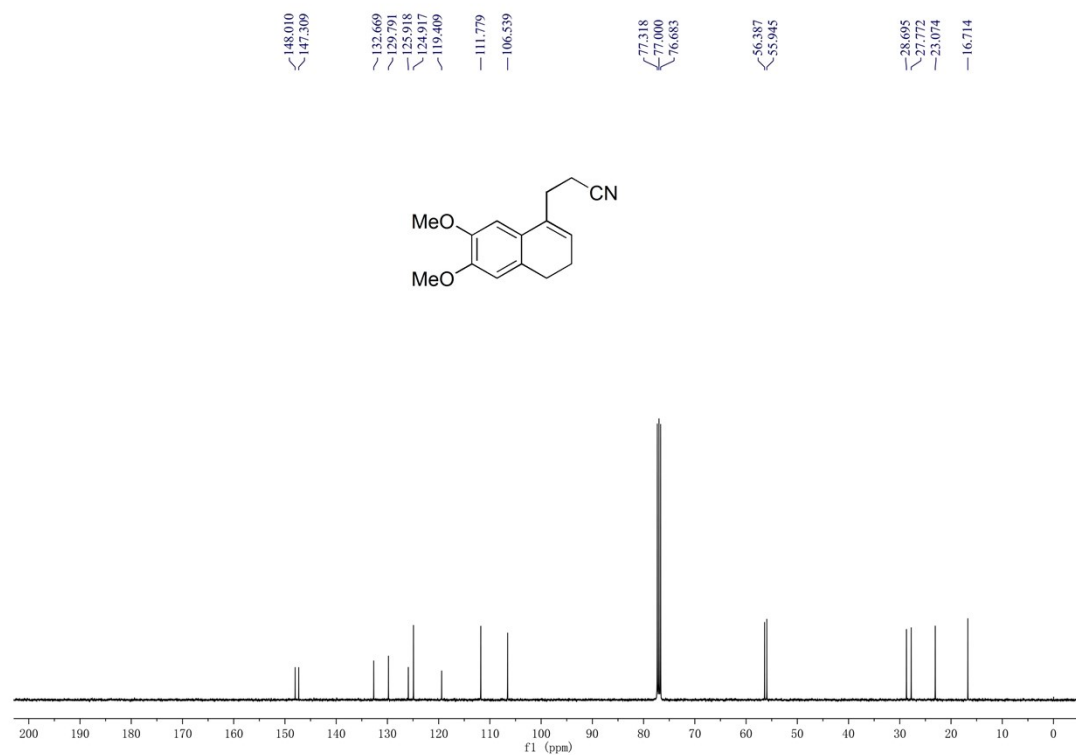
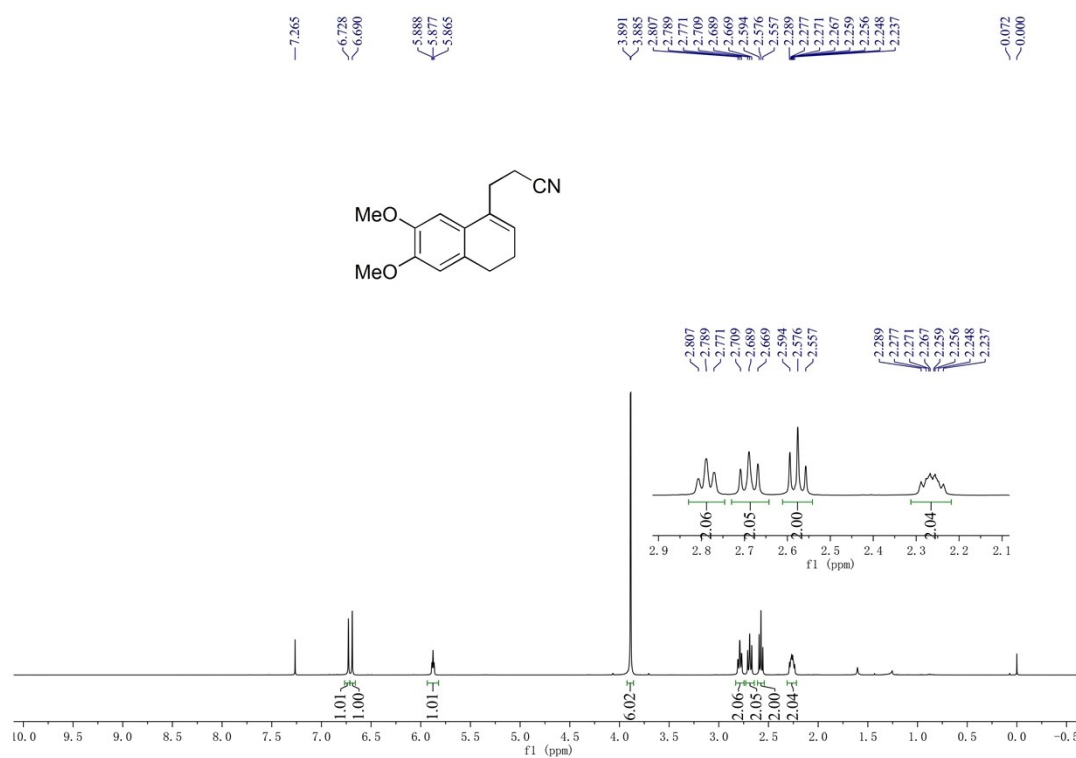
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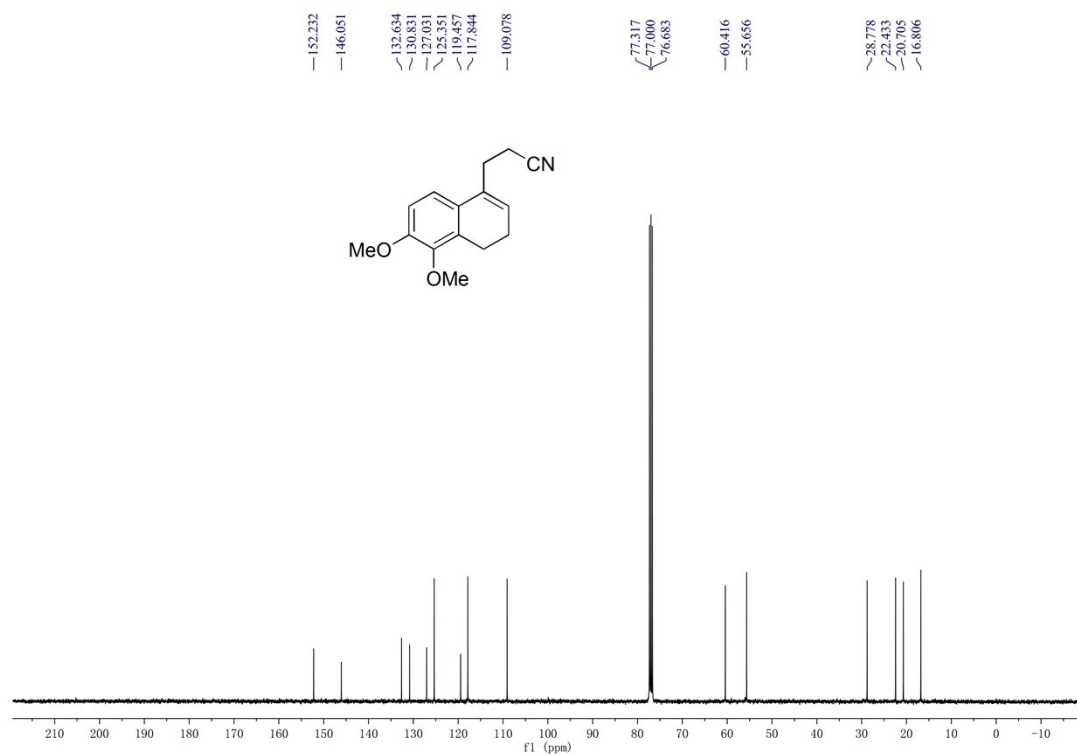
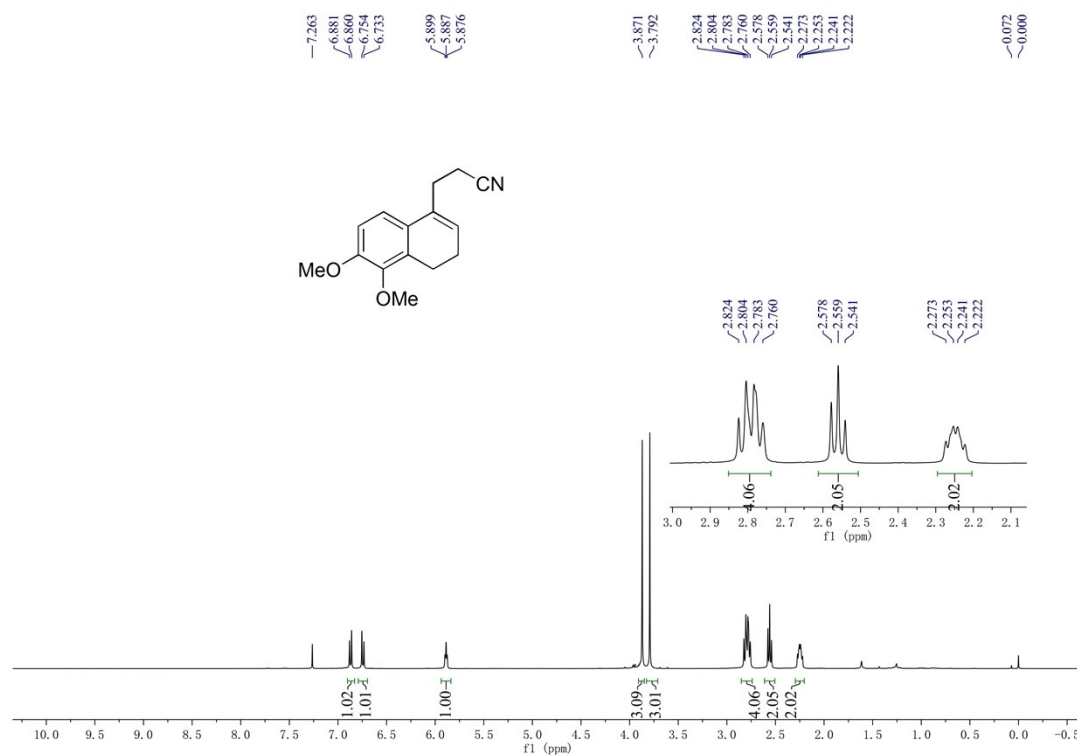
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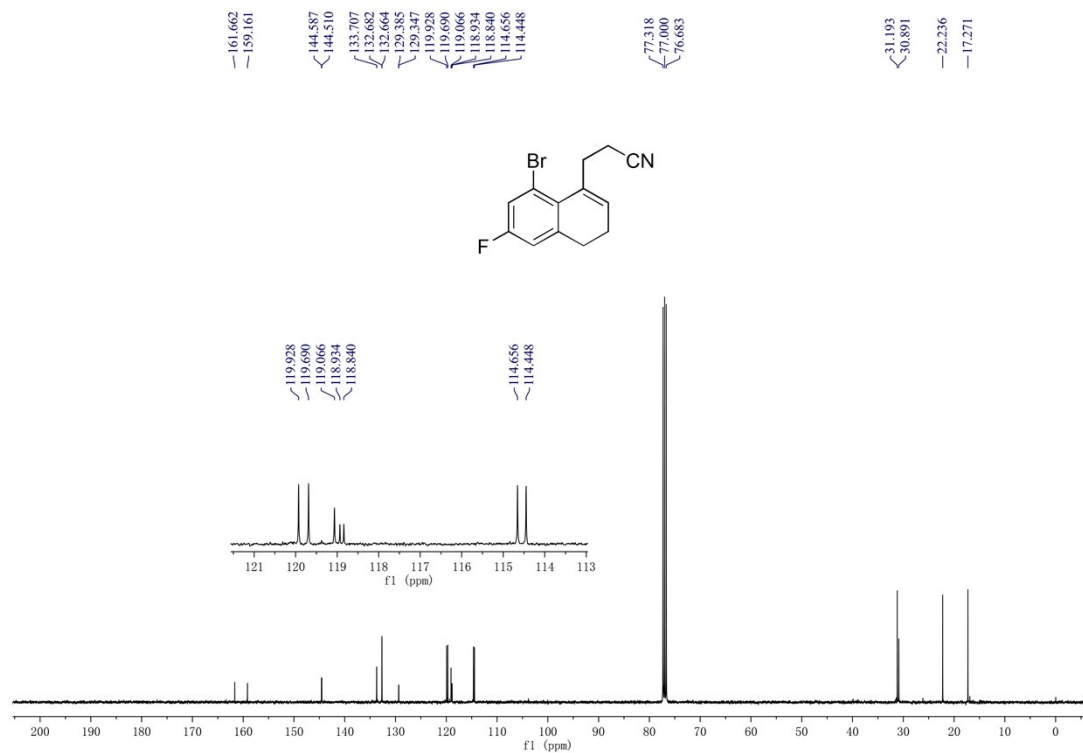
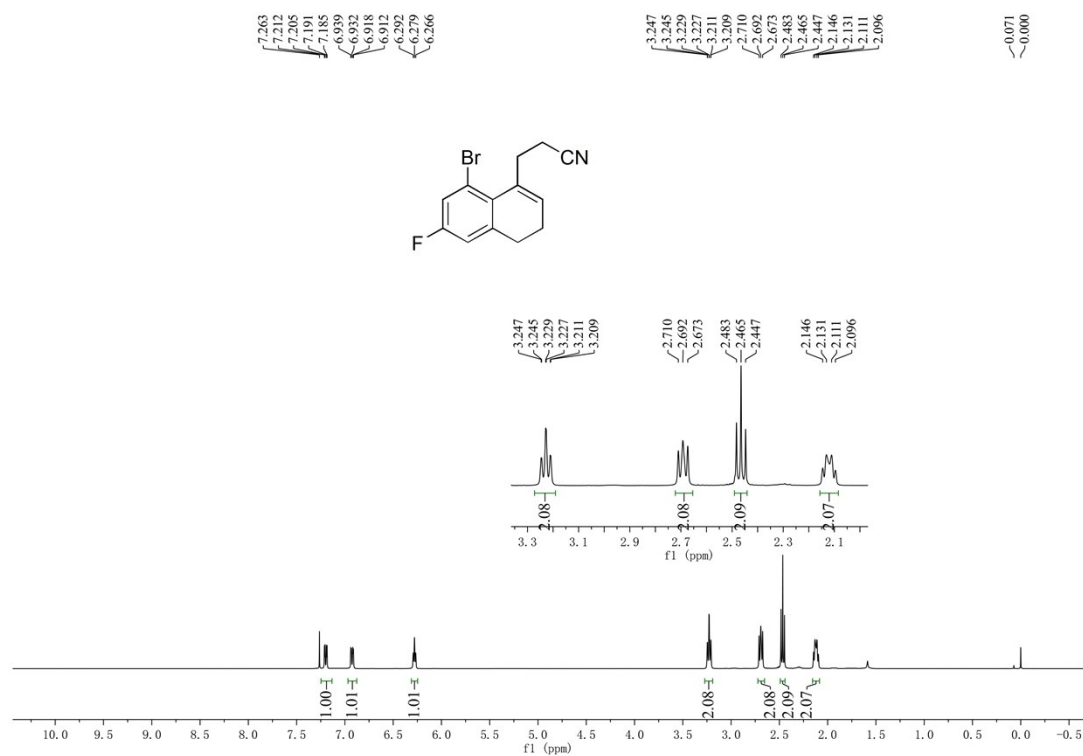
3-(5,6-Dimethoxy-3,4-dihydronaphthalen-1-yl)propanenitrile (3ta)

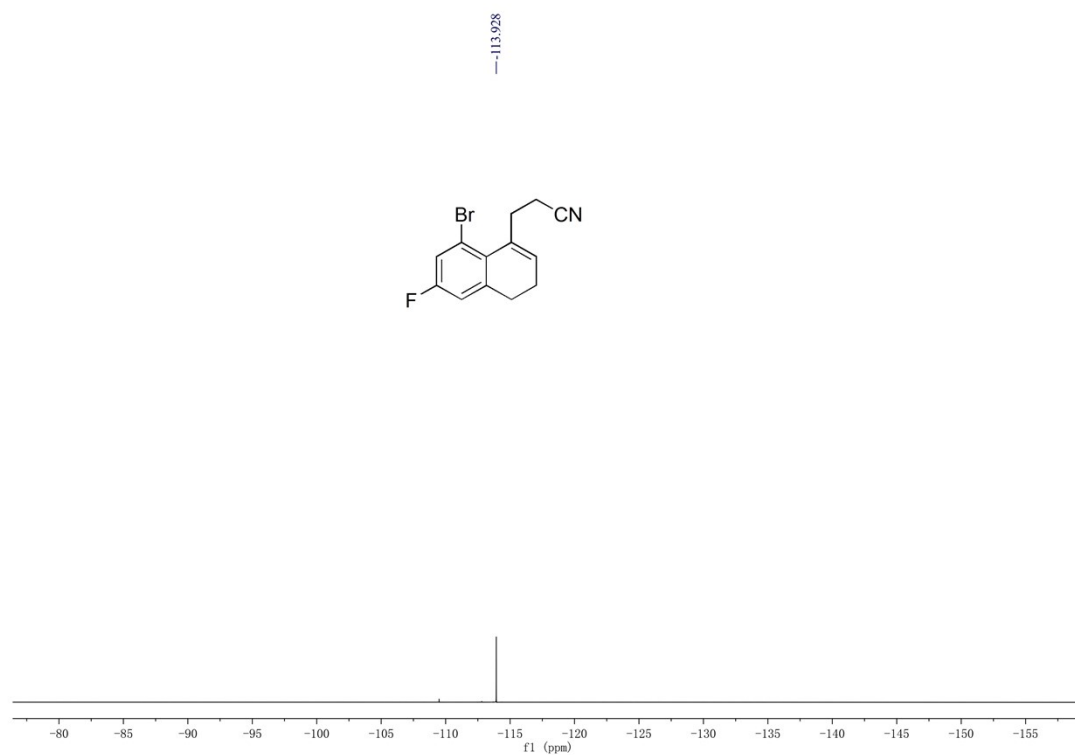


3-(6,7-Dimethoxy-3,4-dihydronaphthalen-1-yl)propanenitrile (3ta')

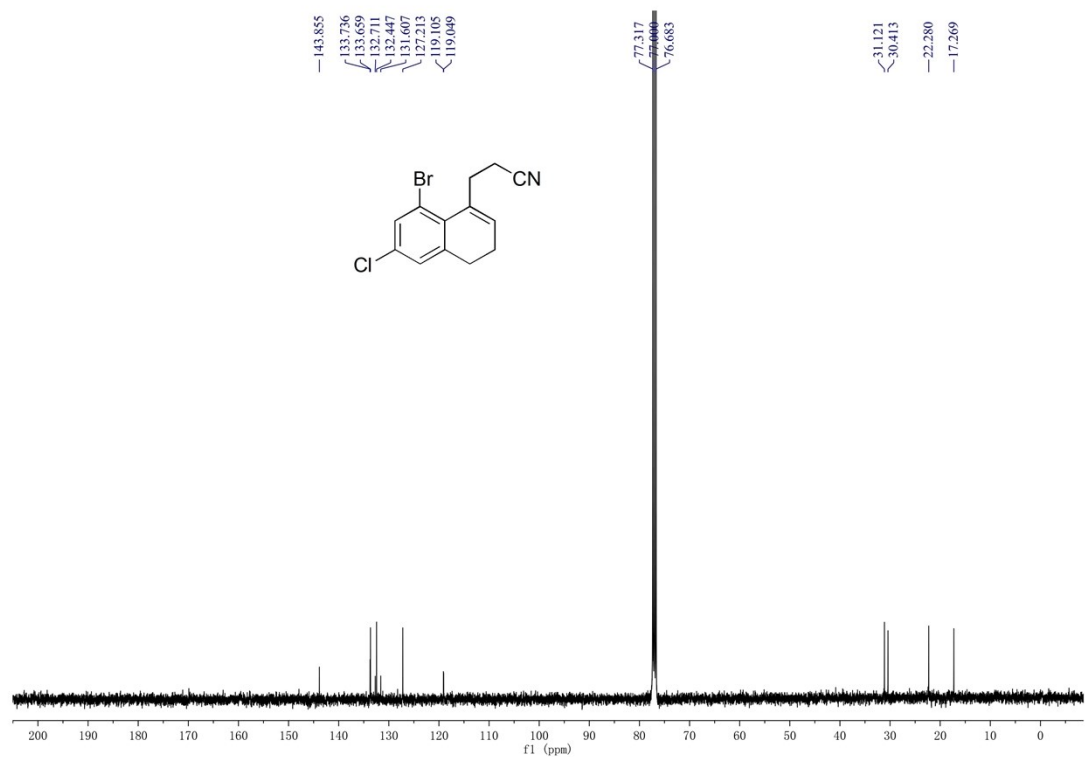
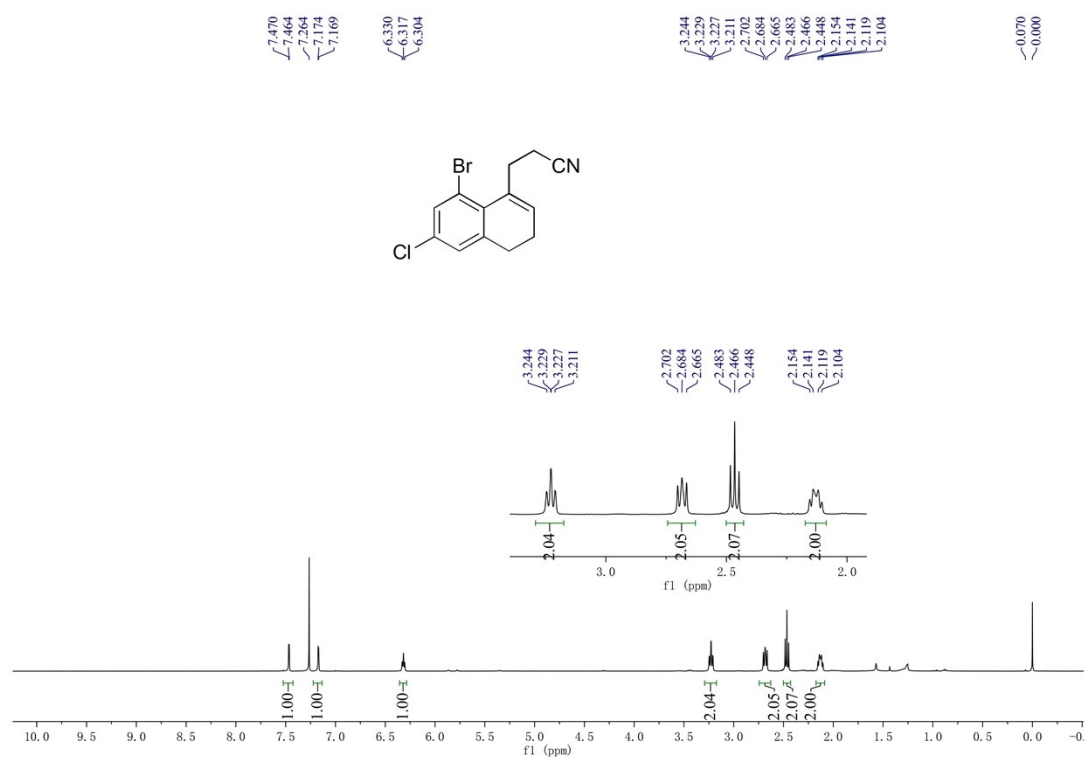


3-(8-Bromo-6-fluoro-3,4-dihydronaphthalen-1-yl)propanenitrile (3ua)

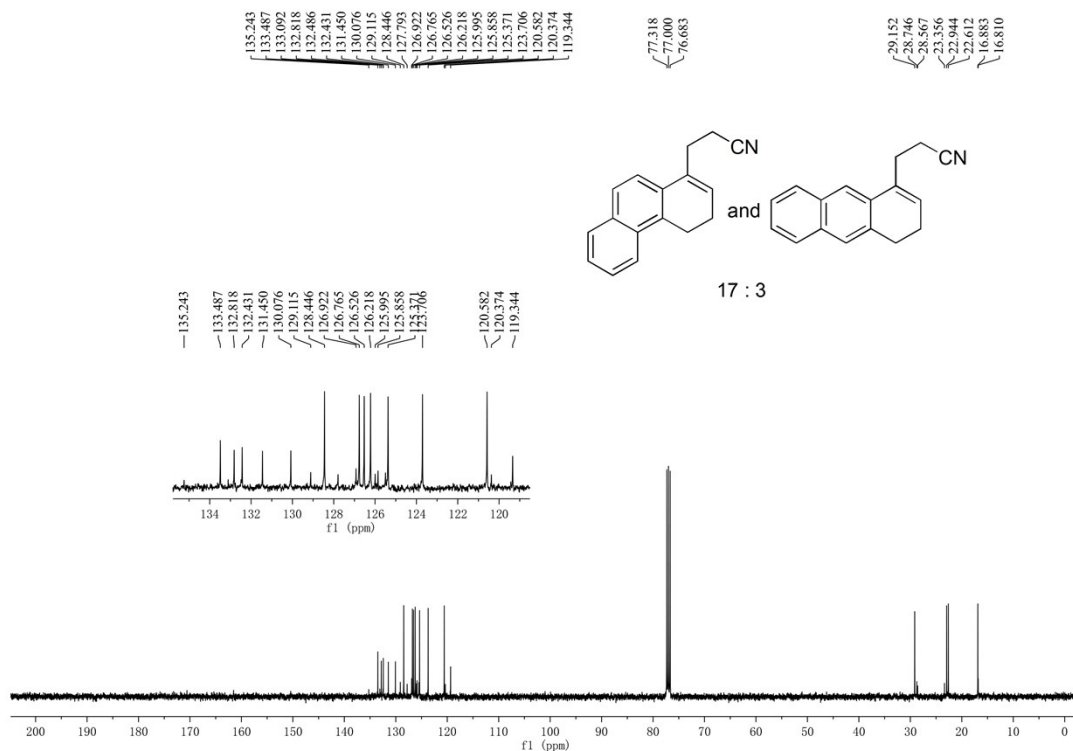
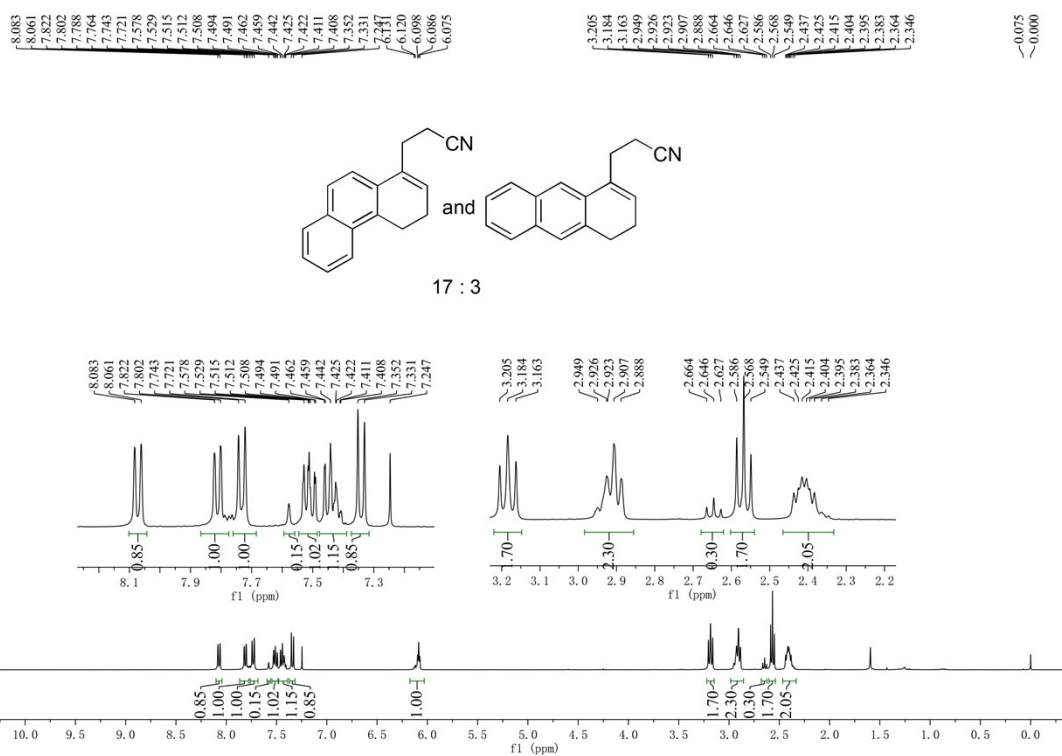




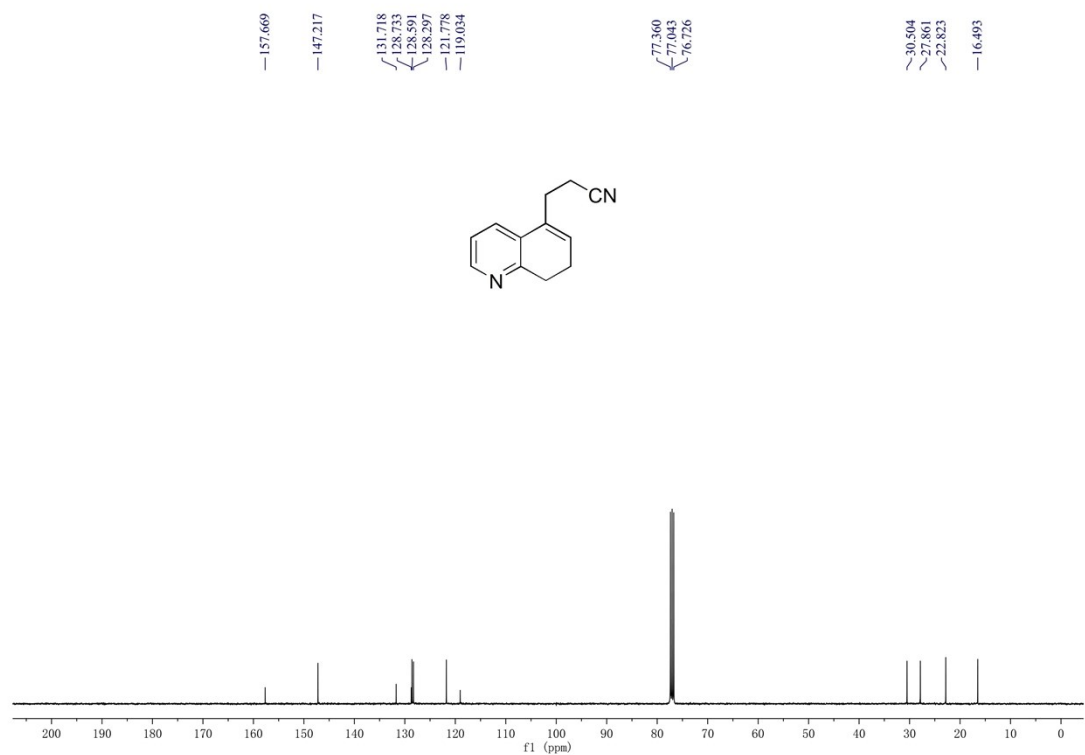
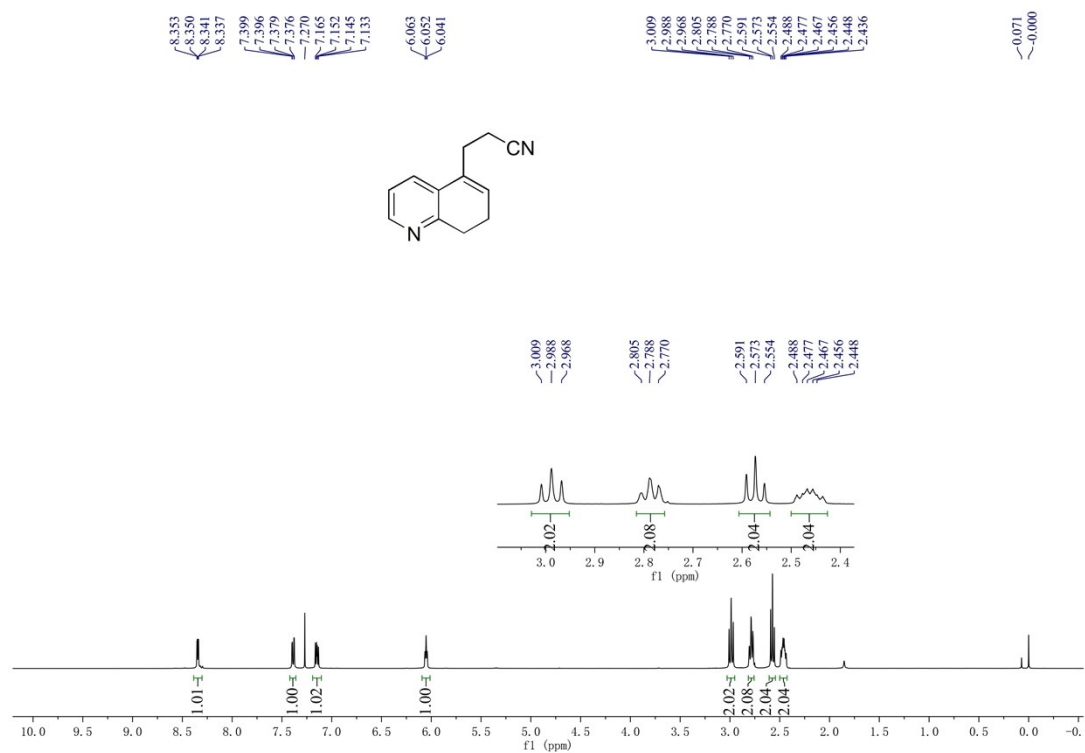
3-(8-Bromo-6-chloro-3,4-dihydronaphthalen-1-yl)propanenitrile (3va)



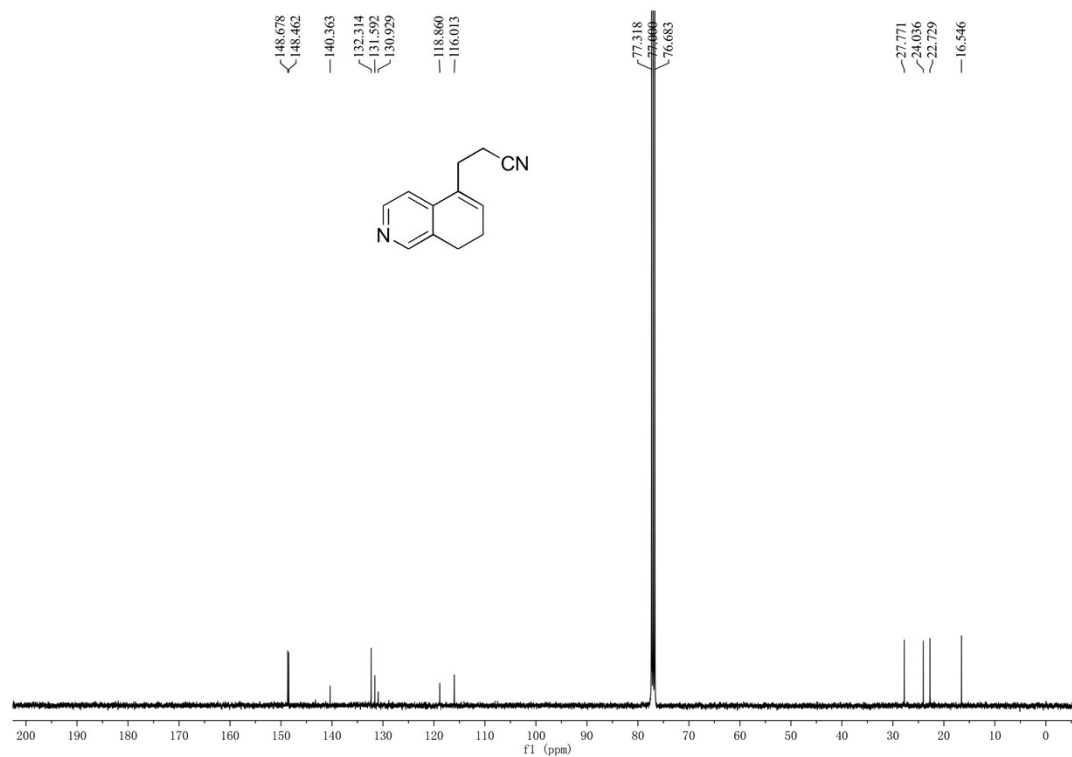
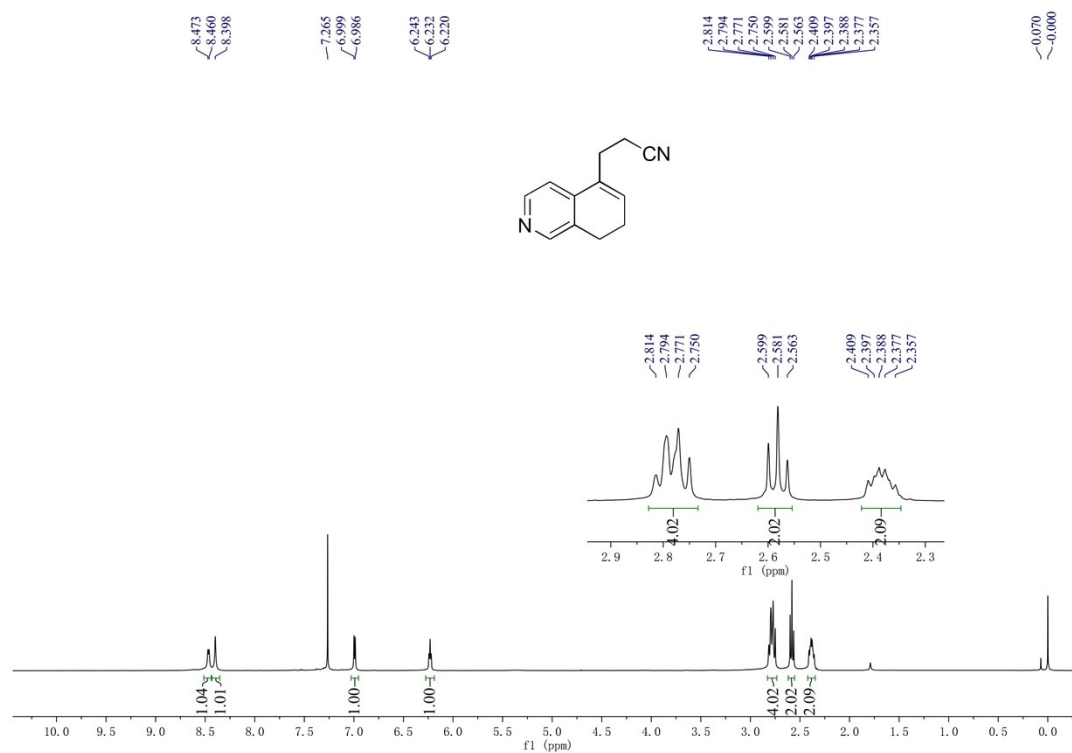
3-(3,4-dihydrophenanthren-1-yl)propanenitrile (3wa) and 3-(3,4-dihydroanthracen-1-yl)propanenitrile (3wa')



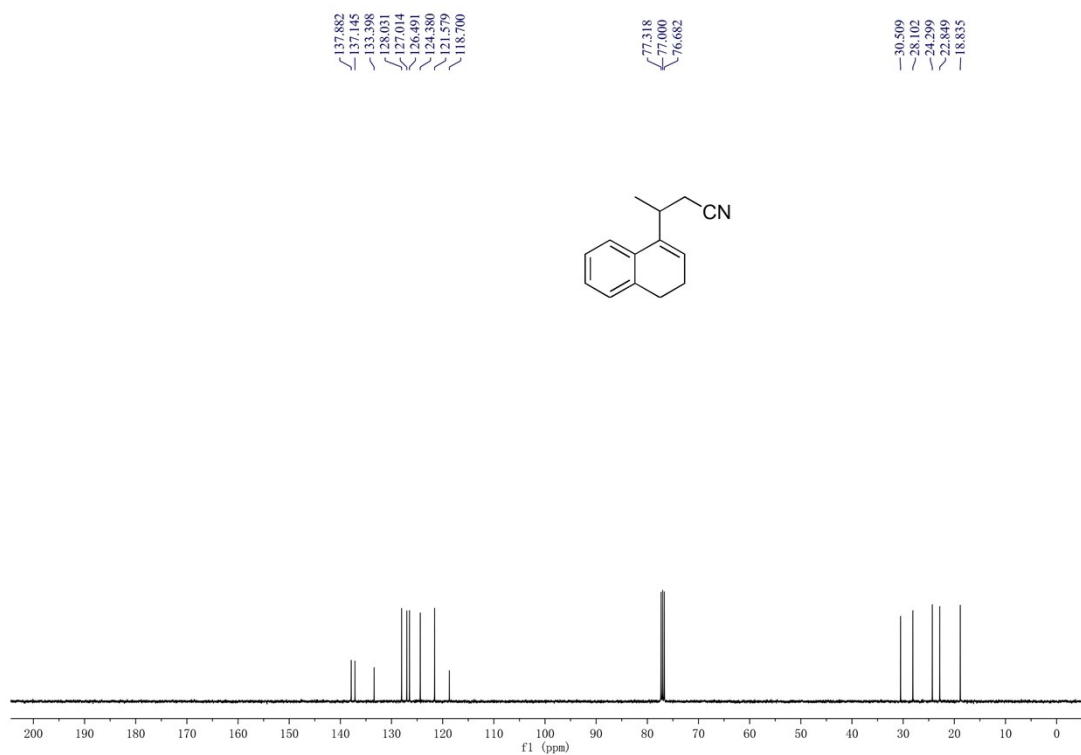
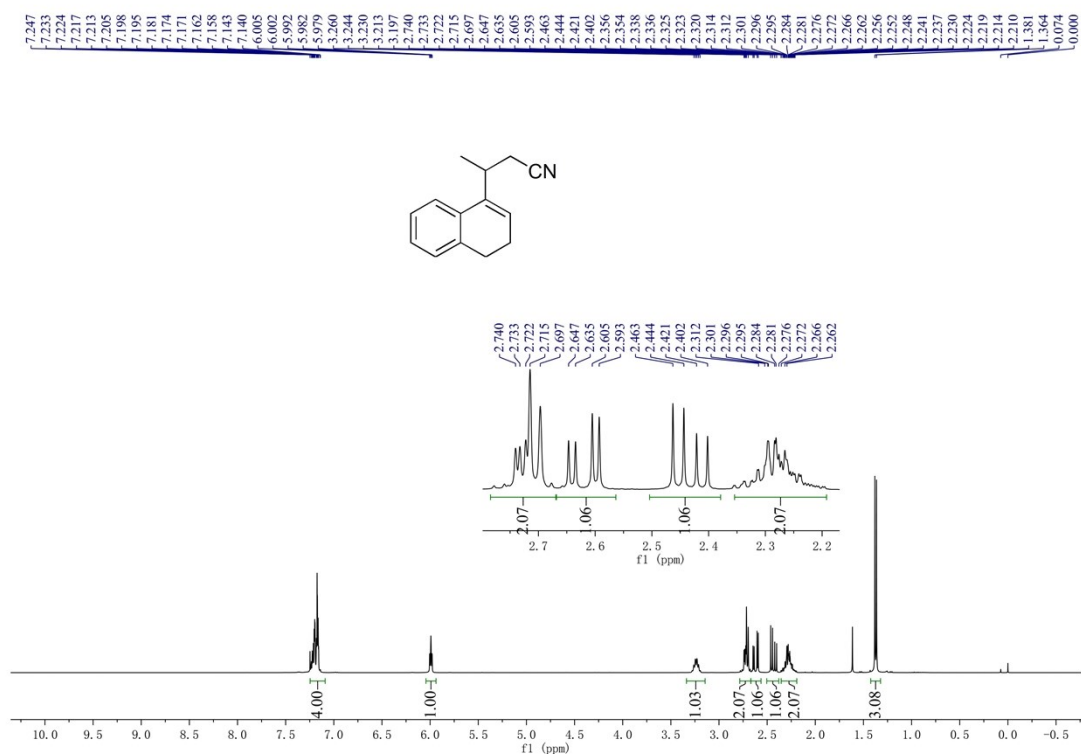
3-(5,6-Dihydroisoquinolin-8-yl)propanenitrile (3a)



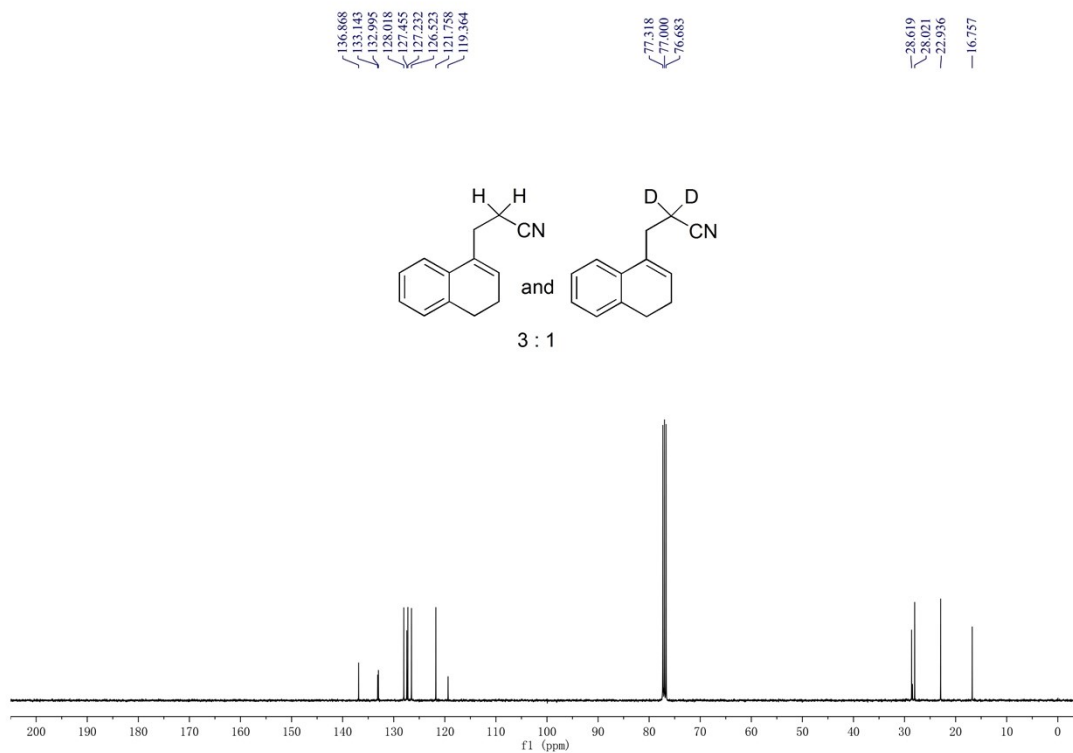
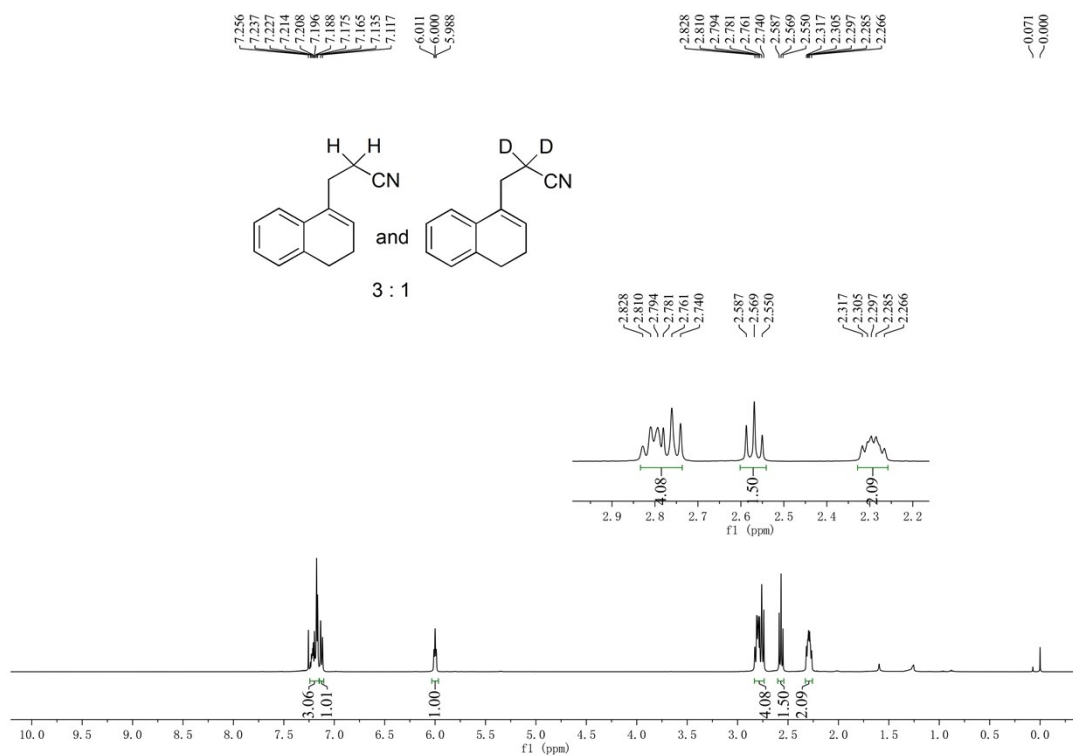
3-(7,8-Dihydroisoquinolin-5-yl)propanenitrile (3ya)



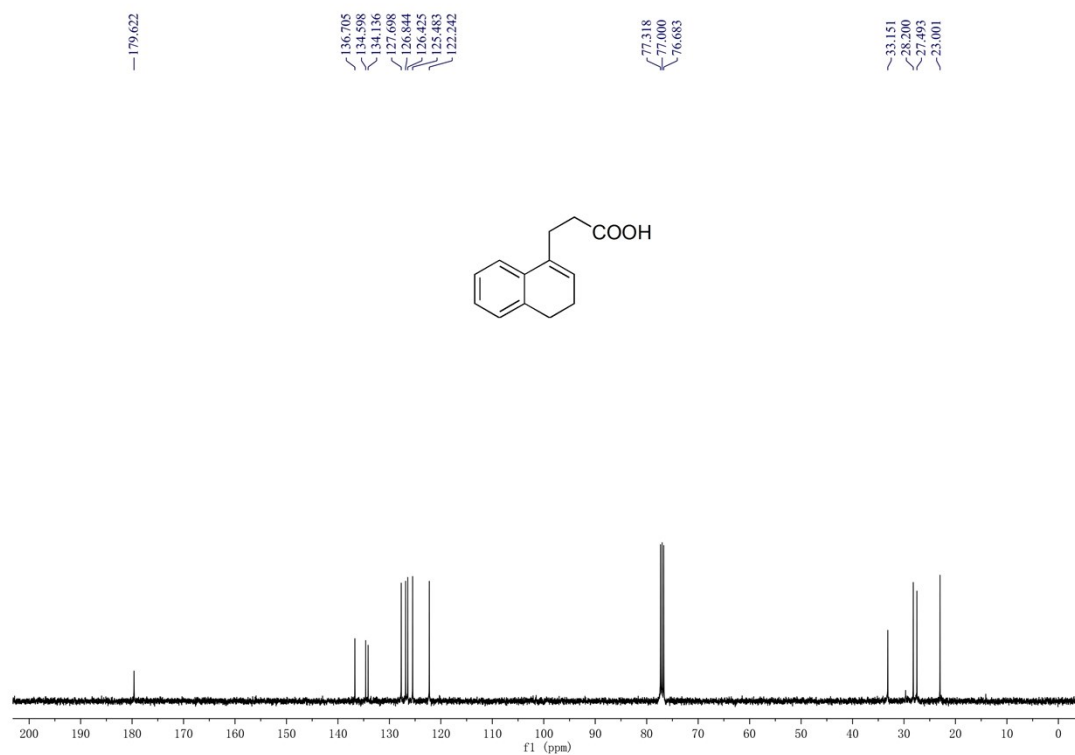
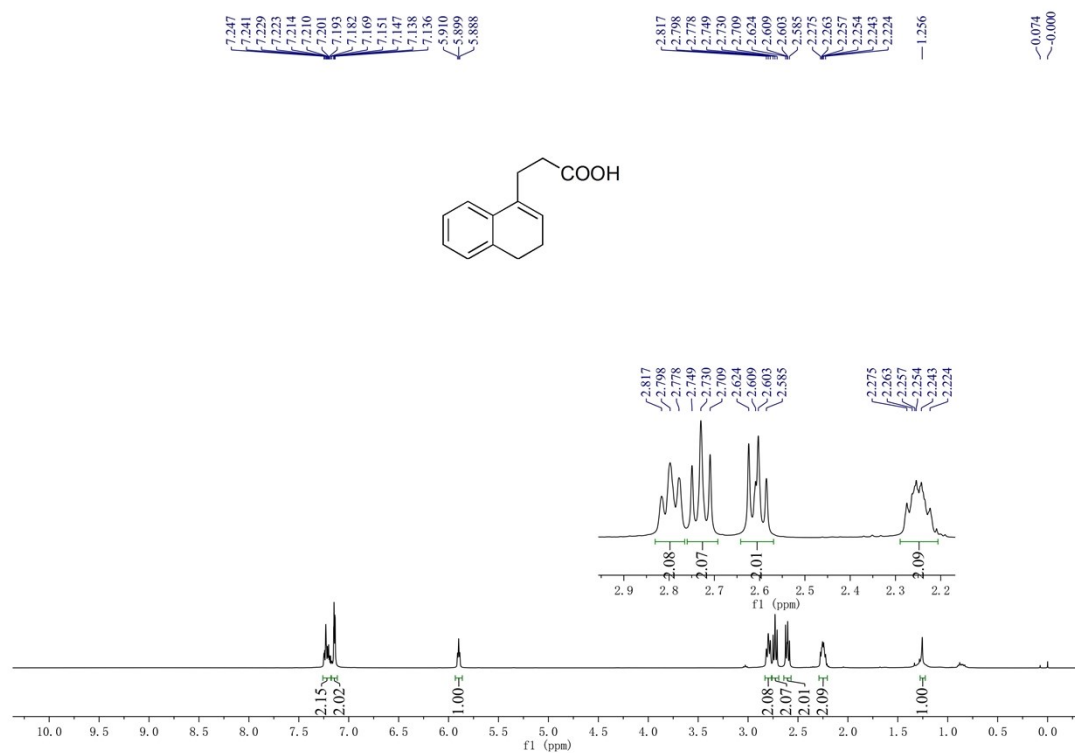
3-(3,4-Dihydronaphthalen-1-yl)butanenitrile (3aaa)



3aa+3aa-D2



3-(3,4-Dihydronaphthalen-1-yl)propanoic acid (5)



3-(3,4-Dihydronaphthalen-1-yl)propanamide (6)

