Supplementary Information for

# I<sub>2</sub>-promoted [4 + 2] Cycloaddition of in Situ Generated Azoalkenes

# with Enaminones: Facile and Efficient Synthesis of 1,4-

## **Dihydropyridazines and Pyridazines**

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#### 1. General information

The solvents were dried and distilled prior to use by the literature methods. Analytical TLC plates, Sigma-Aldrich silica gel 60F200 were viewed by UV light (254 nm). Column chromatographic purifications were performed on SDZF silica gel 160. <sup>1</sup>H and <sup>13</sup>C NMR spectra were obtained on a Bruker NMR spectrometer at 400 MHz and 100 MHz, respectively, and referenced internally based on the residual solvent signal. The data reporting for <sup>1</sup>H NMR spectra is as follows: chemical shift ( $\delta$ , ppm), multiplicity (s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet), coupling constant in hertz, and number of protons. The data reporting for <sup>13</sup>C spectra is given as chemical shift ( $\delta$ , ppm). High-resolution mass spectra (HRMS) were obtained on an Agilent mass spectrometer using electrospray ionization-time of flight (ESI-TOF). Melting points were measured with a melting point instrument without correction. All the chemical reagents were purchased from commercial sources and used as received unless otherwise indicated.

#### 2. General procedure for the synthesis of N-sulfonylhydrazones (1)



A typical procedure for the synthesis of *N*-sulfonylhydrazones (1a-1u) – *Synthesis of 1a:* A mixture of acetophenone compounds (5.0 mmol) and *p*-toluenesulfonhydrazide (5.0 mmol) in 7.5 mL of MeOH was stirred at 60 °C for 0.5-3 h to afford the *N*-tosylhydrazone **1a** as a white precipitate. After that, the precipitate was washed and filtered with petroleum ether twice and dried under a vacuum to provide the pure compound **1a** (white solid, 1.37 g, 95%).<sup>1-4</sup>

#### **3.** General procedure for the synthesis of enaminones (2)



A typical procedure for the synthesis of enaminones (2a-2m, 2o) – Synthesis of 2a: To a stirred solution of acetophenone (5.0 mmol, 1.0 eq.) in toluene (5.0 mL), 1,1-dimethoxy-*N*, *N*-dimethylmethanamine (7.0 mmol, 1.4 eq.) was added and stirred at 110 °C. After completion of the reaction (monitored by TLC), it was quenched with water, extracted with ethyl acetate and dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>. Then the reaction mixture was concentrated under reduced pressure and purified by silica gel column chromatography (eluent: petroleum ether (60-90 °C)/EtOAc = 1:1, v/v) to give the desired product **2a** (yellow solid, 856 mg, 98%).<sup>2,5</sup>



**Procedure for the synthesis of enaminones 2n**: To a stirred solution of enaminones **2a** (2 mmol, 1.0 eq.) in EtOH (5 mL), dibenzylamine (8 mmol, 4.0 eq.) was added and then heated at reflux for 12h. After completion of the reaction (monitored by TLC), the reaction mixture was cooled to room temperature and concentrated under reduced pressure to get the crude compound. The crude compound was purified by silica gel column chromatography (eluent: petroleum ether (60-90 °C)/EtOAc = 1:1, v/v) to give the desired product **2n** (yellow solid, 345 mg, 53%).<sup>6</sup>

### 4. Supplementary Reference

- S. P. Hu, S. Y. Du, Z. G. Yang, L. F. Ni and Z. K. Chen, *Adv. Synth. Catal.* 2019, 361, 3124.
- (2) M. Y. Ni, J. G. Zhang, X. Y. Liang, Y. J. Jiang and T. P. Loh, *Chem. Commun.* 2017, **53**, 12286.
- (3) Z. Chen, Q. Yan, Z. Liu, Y. Xu and Y. Zhang, Angew. Chem., Int. Ed. 2013, 52, 13324.
- (4) X. L. Yang, X. X. Peng, F. Chen and B. Han. Org. Lett. 2016, 18, 2070.
- (5) C. Song, C. Yang, H. Zeng, W. Zhang, S. Guo and J. Zhu, Org. Lett. 2018, 20, 3819.
- (6) Y. Gao, Y. Liu, L. Wei, and J. Wan, Res. Chem. Intermed. 2017, 43, 5547.

### 5. X-Ray crystallographic studies

Single crystals for the X–ray diffraction studies for compounds **3e** was carried out on a SMART APEX diffractometer with graphite-monochromated Mo radiation  $(\lambda = 0.71073 \text{ Å})$ . Cell parameters were obtained by global refinement of the positions of all collected reflections. Intensities were corrected for Lorentz and polarization effects and empirical absorption. The structures were solved by direct methods and refined by full-matrix least squares on  $F^2$ . All non-hydrogen atoms were refined anisotropically. All hydrogen atoms were placed in calculated positions. Structure solution and refinement were performed by using the SHELXL-97 package. The Xray crystallographic files, in CIF format, are available from the Cambridge Crystallographic Data Centre on quoting the deposition numbers CCDC 2014085 for **3e**. Copies of this information may be obtained free of charge from The Director, CCDC, 12 Union Road, Cambridge CB2 IEZ, UK (Fax: +44-1223-336033; e-mail: deposit@ccdc.cam.ac.uk or www: http://www.ccdc.cam.ac.uk).



Figure 1. Molecular structure of compound 3e.

<b>Table S1.</b> Crystal data and structure refinement for	r 3	e.
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Empirical formula	$C_{24}H_{19}BrN_2O_3S$	
Formula weight	495.38	
Temperature	293(2) K	
Wavelength	0.71073 Å	
Crystal system, space group	Triclinic, P-1	
Unit cell dimensions	a = 8.1722(4) Å	$\Box = 80.9050(10)^{\circ}$

	b = 10.5252(5) Å	$\Box$ b= 79.4010(10)°	
	c = 13.3654(6) Å	$\Box$ g = 76.2740(10)°	
Volume	1089.82(9) Å <sup>3</sup>		
Z, Calculated density	2, 1.510 Mg/m <sup>3</sup>		
Absorption coefficient	2.011 mm <sup>-1</sup>		
F(000)	504		
Crystal size	0.170 x 0.140 x 0.120 mm <sup>3</sup>		
Theta range for data collection	2.390 to 25.999°		
Index ranges	-10<=h<=10, -12<=k<=12, -16<=l<=16		
Reflections collected	27113		
Independent reflections	4245 [R(int) = 0.0458]		
Completeness to theta = $25.242$	99.2 %		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	0.7456 and 0.4312		
Refinement method	Full-matrix least-squares on F <sup>2</sup>		
Data/restraints/parameters	4245 / 0 / 282		
Goodness-of-fit on F <sup>2</sup>	1.018		
Final R indices [I > 2 sigma(I)]	R1 = 0.0300, wR2 = 0.0	784	
R indices (all data)	R1 = 0.0352, wR2 = 0.0	819	
Extinction coefficient	0.062(4)		
Largest diff. peak and hole	0.241 and -0.242 e.Å $^{\text{-3}}$		

# 6. Copies of NMR spectra for compounds

FJJ-151P



FJJ-151PC



FJJ-254-H



FJJ-R254













FJJ-R249-13C







FJJ-R260 hq-4-114



hq-4-114























zkd-20-1







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FJJ-R257-2
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FJJ-R256

#### 7,9403 7,945 7,9175 7,9175 7,9175 7,9175 7,9143 7,9143 7,9144 7,909 7,7590 7,5909 7,5909 7,5905 7,59



FJJ-R256



FJJ-R255

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FJJ-R255















FJJ-R291













FJJ-R264-C







FJJ-R266-C







FJJ-R276



FJJ-R275











FJJ-R267-



FJJ-R267-







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FJJ-R287
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FJJ-R274













-9.3886 -9.3886 -9.3849 -9.3849 -8.1352 -8.1352 -8.1122 -8.1122 -8.1122 -8.1122 -8.1122 -8.1122 -8.1122 -8.1122 -7.7854 -7.77854 -7.77854 -7.77854 -7.75558 -7.75558 -7.75578 -7.5558 -7.75558 -



FJJ-R214





FJJ-0BC-1H, C



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FJJ-OBC-1H, C
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FJJ-R271-2
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FJJ-0BC-2



FJJ-0BC-2







FJJ-R271







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FJJ-R279
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FJJ-0BC-3H, C



FJJ-0BC-3C









FJJ-0BC-4



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FJJ-0BC-4
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FJJ-R290





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FJJ-R285-CF3
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FJJ-N0-1



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FJJ-N0-1
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