Electronic Supplementary Information

Suppression of menthyl anthranilate (UV-A sunscreen)-sensitized singlet oxygen generation by Trolox and α -tocopherol

Shogo Kitasaka ^a, Mikio Yagi ^a*, Azusa Kikuchi ^a*

^a Department of Chemistry, Graduate School of Engineering Science, Yokohama National University, Tokiwadai, Hodogaya-ku, Yokohama 240-8501, Japan.



Fig. S1 UV absorption spectra of (a) MA (0.4 mmol dm⁻³, red solid line), (b) Trolox (20 mmol dm⁻³, dash-dot-dash line), (c) α -tocopherol (16 mmol dm⁻³, dotted line) and (d) 3-EtAA(20 mmol dm⁻³, broken line) in ethanol at 25 °C.



Fig. S2 Kinetic traces of the phosphorescence intensity of singlet oxygen taken following 355 nm laser excitation of MA (0.4 mmol dm⁻³) in air-saturated ethanol at 25 °C in the absence (red) and presence of 3-EtAA (16 mmol dm⁻³). The phosphorescence intensity was monitored at 1274 nm.



Fig. S3 UV absorption spectra of (a) MA (0.4 mmol dm⁻³, red solid line) in the presence of (b) Trolox (20 mmol dm⁻³, dash-dot-dash line), (c) α -tocopherol (16 mmol dm⁻³, dotted line) and (d) 3-EtAA(20 mmol dm⁻³, broken line) in ethanol at 25 °C.



Fig. S4 Time-resolved phosphorescence spectrum of singlet oxygen generated by excitation of MA in air-saturated isododecane. The sampling times were set at 2–17 μ s after the 355 nm YAG laser pulse.

$$MA + h^{v} \longrightarrow MA^{*}$$

$$^{1}MA^{*} \longrightarrow K_{F} MA + h^{v}_{F}$$

$$^{1}MA^{*} \longrightarrow K_{IC} MA$$

$$^{1}MA^{*} \longrightarrow MA^{*}$$

$$^{1}MA^{*} + ^{3}O_{2} \longrightarrow MA + ^{3}O_{2}$$

$$^{1}MA^{*} + Q \longrightarrow MA + Q$$

$$^{3}MA^{*} \longrightarrow MA$$

$$^{3}MA^{*} \longrightarrow MA$$

$$^{3}MA^{*} + ^{3}O_{2} \longrightarrow MA + ^{1}O_{2}$$

$$^{3}MA^{*} + Q \longrightarrow MA + Q$$

Scheme S1 kinetics of the excited singlet and triplet states of MA.