Supporting information

High performance phenyl-free LED photoinitiator for cationic or hybrid photopolymerization and its application in LED cationic 3D printing

Liqun Tang, Jun Nie, Xiaoqun Zhu\*

State Key Laboratory of Chemical Resource Engineering, Beijing University of Chemical Technology, Beijing 100029, P. R. China.

E-mail: <u>zhuxq@mail.buct.edu.cn</u>.

## Synthesis of 1,3-bis(1-methyl-1H-pyrrol-2-yl)prop-2-en-1-one

2mL N-Methylpyrrole-2-carboxaldehyde, 2mL 2-Acetyl-1-methylpyrrole, 20mL ethanol and 0.1g NaOH, 10g deionized water were added into a 100mL three-necked round bottom flask. The solution was stirred by magnetic stirrer at room temperature for 8 h under the nitrogen protection. The crude product was obtained by filtration under reduced pressure and was purified by recrystallization from Ethanol. The pure product was obtained as a Light yellow crystal. <sup>1</sup>H NMR (400 MHz, CDCl3,\delta): 7.856-7.837 (t,1H,NMPy-CH) ,7.818 (s,1H,NMPy-CH) ,7.665 (d, 1H,C-CH=C) ,7.286 (s,1H,NMPy-CH) ,7.215-7.178 (m,1H, C=CH-C=O) 6.878-6.868 (d, 1H,NMPy-CH) ,6.845 (s,1H,NMPy-CH) ,6.262-6.246 (t,1H,NMPy-CH) 3.794 (s,6H,N-CH3)

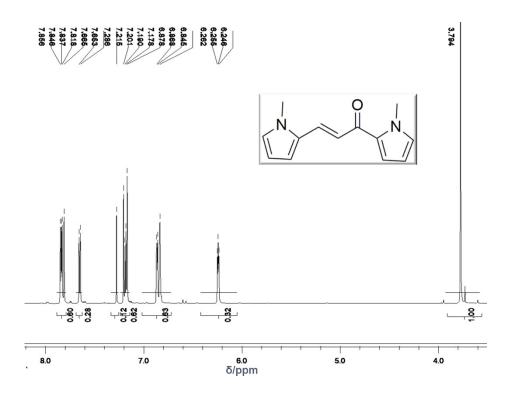


Figure S1 The <sup>1</sup>H NMR spectrum of BMO

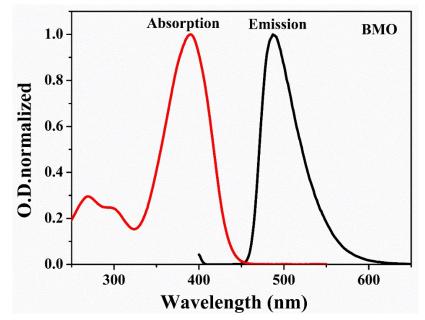


Figure S2 Normalized absorption and emission spectra of BMO in acetonitrile

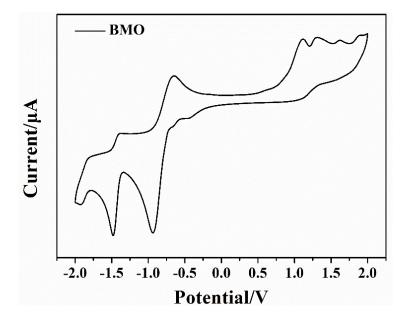


Figure S3 Cyclic voltammograms of BMO ( $4 \times 10^{-4}$ mol/L) in anhydrousacetonitrile solution purged with N<sub>2</sub>. A platinum electrode at a scan rate of 100 mV s<sup>-1</sup> with 0.1 M tetrabutylammonium hexafluorophosphate as the supporting electrolyte, and the reference electrode was a saturated calomel electrode (SCE). Ferrocene was used as a standard, and the potentials determined from the half-peak potential were referred to the reversible formal potential of this compound (+0.38 V/SCE).