## **Electronic Supplementary Information**

for

SSynthesis of Cyclic Olefin Copolymers (COCs) by Ethylene Copolymerisations with

Cyclooctene, Cycloheptene, and with Tricyclo[6.2.1.0(2,7)]undeca-4-ene: Effect of Cyclic

## Monomer Structures on Thermal Properties

Hitoshi Harakawa, Masaki Okabe, and Kotohiro Nomura\*

Department of Chemistry, Tokyo Metropolitan University, 1-1 Minami Osawa, Hachioji, Tokyo 192-0397, Japan.

### Contents

1. Additional results for ethylene copolymerisation with cyclooctene, cycloheptene, and with tricyclo[6.2.1.0(2,7)]undeca-4-ene.

2. Selected NMR spectra in the copolymers including assignment of resonances and estimation of comonomer contents

3. DSC thermograms in the copolymers.

4. Plots of glass transition temperature  $(T_g)$  vs comonomer content (mol%) in ethylene copolymers with norbornene (NBE), tetracyclododecene (TCD), and with tricyclo[6.2.1.0(2,7)]undeca-4-ene (TCUE).

# 1. Additional results for ethylene copolymerisation with cyclooctene, cycloheptene, and with tricyclo[6.2.1.0(2,7)]undeca-4-ene.

**Table S1.** Copolymerization of ethylene (E) with cyclooctene (COE) by **1-9** - MAO catalysts.<sup>*a*</sup> Additional results (1).

run	cat.	COE	Е	temp.	yield	activity <sup>c</sup>	$M_{ m n}{}^d$	$M_{ m w}/$	cont. <sup>e</sup>	$T_{\rm g} \left(T_{\rm m}\right)^f$
	(µmol)	conc. <sup>b</sup> / M	/ atm	/ °C	/ mg			$M_{\mathrm{n}}{}^{d}$	/ mol%	/ °C
1	1 (1.5)	2.5	2	25	78	312	59000	1.35	-	38 (129)
2	<b>1</b> (1.0)	5.0	2	25	42.5	255	35200	1.40	-	38 (127)
<b>S</b> 1	<b>1</b> (1.0)	5.0	2	25	48.4	251	37100	1.53		28 (126)
3	<b>2</b> (1.5)	2.5	2	25	74.4	298	32600	1.36	-	$40^{i}$
4	<b>2</b> (1.0)	2.5	4	25	124	742	48300	1.29	-	$32^{j}$
<b>S</b> 2	<b>2</b> (1.0)	2.5	4	25	123	736	52500	1.35	-	31 <sup><i>j</i></sup>
5	<b>2</b> (1.5)	5.0	2	25	57.7	231	20700	1.62	$43.7^{h}$	$44^{j}$
6 <sup><i>g</i></sup>	<b>2</b> (1.5)	5.0	2	25	81.5	326	10000	1.77	-	46 (115)
$S3^{g}$	<b>2</b> (1.5)	5.0	2	25	82.0	328	9830	2.03		42 (115)
7	<b>3</b> (0.2)	5.0	4	25	146	4380	318000	1.47	22.5	6
26	<b>3</b> (0.05)	1.0	4	25	89.2	10700	1260000	2.05	7.0	(70)
27	<b>3</b> (0.2)	2.5	4	25	228	6850	863000	1.89	16.1	-15 (32)
<b>S</b> 4	<b>3</b> (0.2)	2.5	4	25	245	7360	759000	1.37	$16.7^{i}$	-17 (32)
28	<b>3</b> (0.2)	2.5	4	50	122	3660	423000	1.49	$17.4^{i}$	-14 (32)
8	<b>3</b> (0.5)	5.0	2	25	138	1660	239000	1.57	28.0	32
9 <sup>g</sup>	<b>3</b> (0.5)	5.0	2	25	202	2420	218000	2.21	-	13 <sup>j</sup>
<b>S</b> 5	<b>3</b> (0.5)	5.0	2	25	241	2890	374000	1.50	-	$12^{j}$
10	<b>3</b> (0.5)	7.5	2	25	114	1360	141000	1.57	28.1	36
<b>S</b> 6	<b>3</b> (0.5)	7.5	2	25	112	1340	131000	1.34	29.3 <sup><i>i</i></sup>	38
11	<b>4</b> (1.0)	2.5	2	25	83.5	501	5600	1.24	-	$14^{j}$
<b>S</b> 7	<b>4</b> (1.0)	2.5	2	25	72.0	432	6310	1.30		$17^{j}$
12	<b>4</b> (1.0)	5.0	2	25	100	602	5700	1.21	-	$18^{j}$
<b>S</b> 8	<b>4</b> (1.0)	5.0	2	25	90.3	542	4240	1.19		20 <sup><i>i</i></sup>
13	5 (0.01)	5.0	4	25	250	150000	1670000	1.72	-	(69)
<b>S</b> 9	5 (0.01)	5.0	4	25	215	129000	1360000	1.78		(70)
14	5 (0.01)	5.0	2	25	127	76400	1650000	2.52	$16.0^{i}$	-20 (48)
S10	5 (0.01)	5.0	2	25	103	61900	2430000	2.40	$16.0^{i}$	-20 (46)
15	5 (0.03)	7.5	2	25	148	29700	1210000	2.30	20.2	-5
<b>S</b> 11	5 (0.03)	7.5	2	25	142	28400	1330000	2.22	20.2	-5.2
16	<b>6</b> (0.01)	5.0	4	25	290	174000	1470000	1.48	-	(71)
S12	<b>6</b> (0.01)	5.0	4	25	290	174000	1130000	2.06		(71)
17	<b>6</b> (0.01)	5.0	2	25	201	121000	3050000	2.15	16.3 <sup><i>i</i></sup>	-19 (56)
18	6 (0.02)	7.5	2	25	119	35800	2540000	2.15	20.4	-5.5
S13	6 (0.02)	7.5	2	25	109	32600	1570000	2.08	20.4	-2.9

Table S1. Con	tinued.
---------------	---------

19 <b>7</b> (0.01)	5.0	4	25	139	83100	2270000	2.56	2.6	(105)
S14 7 (0.01)	5.0	4	25	125	75200	2660000	2.48	2.6	(106)
20 7 (0.01)	5.0	2	25	110	66100	1700000	3.09	-	(92)
S15 7 (0.01)	5.0	2	25	124	74300	2010000	2.51		(92)
21 7 (0.01)	7.5	2	25	98.9	59300	1050000	2.30	7.2	-17 (76)
22 <b>8</b> (1.0)	5.0	4	25	121	726	143000	1.73		(66)
S16 8 (1.0)	5.0	4	25	135	810	148000	1.82		(67)
23 <b>8</b> (1.0)	2.5	2	25	94.4	566	161000	1.54		(77) <sup>j</sup>
24 <b>8</b> (1.0)	5.0	2	25	55.1	331	155000	1.36	9.9	(59) <sup><i>j</i></sup>
S17 8 (1.0)	2.5	2	25	74.7	448	150000	1.56		(76)
25 <b>9</b> (0.5)	2.5	2	25	52.1	625	372000	1.24	-	(41) <sup><i>j</i></sup>
S18 9 (0.5)	2.5	2	25	39.3	472	443000	1.25		$44^j$
26 <b>9</b> (0.5)	5.0	2	25	133	1590	314000	1.39	-	(55) <sup><i>j</i></sup>
S19 9 (0.5)	5.0	2	25	111	1340	371000	1.67		46 <sup><i>j</i></sup>
27 <b>10</b> (0.05)	5.0	2	25	137	16500	290000	2.18	-	(133)
28 <b>11</b> (0.01)	2.5	2	25	105	63000	2500	4.12		(103,124)
29 11 (0.02)	5.0	2	25	122	36600	1700	3.95		(108)
S20 11 (0.02)	5.0	2	25	119	35700	1600	4.39		(110)
30 11 (0.05)	7.5	2	25	96.2	11500	820	5.18		(98)
S21 11 (0.05)	7.5	2	25	115	13800	1100	4.08		(99)

<sup>a</sup>Conditions: toluene + COE total 10 mL, d-MAO (prepared by removing toluene and AlMe<sub>3</sub> from the commercially available TMAO-S) 3.0 mmol. <sup>b</sup> Initial COE concentration (mol/L). <sup>c</sup>Activity = kg-polymer/mol-M·h (M = Ti or Zr). <sup>d</sup>GPC data in *o*-dichlorobenzene vs polystyrene standards. <sup>e</sup>COE content (mol%) estimated by <sup>13</sup>C NMR spectra. <sup>f</sup>By DSC thermograms. <sup>g</sup>Al<sup>i</sup>Bu<sub>3</sub> (500 equiv) and [Ph<sub>3</sub>C][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] (1.5 equiv) were used instead of d-MAO. <sup>h</sup>COE content in the whole polymer estimated by the <sup>13</sup>C NMR spectrum. <sup>i</sup>Estimated on the basis of the plots of  $T_g$  and COE content. <sup>j</sup>Small  $T_m$  shoulder at ca.120 °C was also observed on the DSC thermogram.



run	3	MAO	time	yield	activity <sup>b</sup>	$M_{\rm n}{}^c$	$M_{ m w}/$	cont. <sup>d</sup>	$T_{\rm g} \left(T_{\rm m}\right)^e$
	/ µmol	/ mmol	/ min	/ mg		×10 <sup>-4</sup>	$M_n^c$	/ mol%	/ °C
35	0.5	1.0	10	87.4	1050	23.4	1.31	28.9 <sup>f</sup>	36
S22	0.5	1.0	10	84.8	1020	22.5	1.32	28.7 <sup>f</sup>	35
8	0.5	3.0	10	138	1660	23.9	1.57	28.0	32
36	0.5	5.0	10	171	2050	37.9	1.53	27.5 <sup>f</sup>	30
S23	0.5	5.0	10	195	2340	19.9	1.63	24.5 <sup>f</sup>	17
36	0.5	3.0	5	111	2670	31.9	1.34	27.7 <sup>f</sup>	31
S24	0.5	3.0	5	114	2730	30.5	1.38	28.7 <sup>f</sup>	35
8	0.5	3.0	10	138	1660	23.9	1.57	28.0	32
38	0.5	3.0	15	193	1550	31.9	1.41	$27.5^{f}$	30
S25	0.5	3.0	15	222	1770	26.1	1.55	27.5 <sup>f</sup>	30

**Table S2.** Copolymerization of ethylene (E) with cyclooctene (COE) by  $Cp*TiCl_2(O-2,6-^iPr_2C_6H_3)$ (3) - MAO catalyst (ethylene 4 atm, COE 5.0 M, 25 °C).<sup>*a*</sup> Additional results (2).

<sup>*a*</sup>Conditions: toluene + COE total 10 mL, COE 5.0 M, d-MAO (prepared by removing toluene and AlMe<sub>3</sub> from the commercially available TMAO-S). <sup>*b*</sup>Activity = kg-polymer/mol-Ti·h. <sup>*c*</sup>GPC data in *o*-dichlorobenzene vs polystyrene standards. <sup>*e*</sup>COE content (mol%) estimated by <sup>13</sup>C NMR spectra. <sup>*e*</sup>By DSC thermograms. <sup>*f*</sup>Estimated on the basis of the plots of  $T_g$  and COE content.

**Table S3**. Copolymerization of ethylene (E) with cycloheptene (CHP) by  $Cp*TiCl_2(O-2,6-^{i}Pr_2C_6H_3)$ (3)–d-MAO catalyst.<sup>*a*</sup> Additional results.

run	cat. 3	CHP	Е	temp	yield	activity <sup>c</sup>	$M_{\rm n}{}^d$	$M_{ m w}/$	cont. <sup>e</sup>	$T_{\rm g} (T_{\rm m})^f$
	/ µmol	conc. <sup>b</sup> / M	/ atm	/ °C	/ mg		×10 <sup>-4</sup>	$M_{\rm n}{}^d$	/ mol%	/ °C
43	0.001	1.0	4	25	77.7	466000	244	1.36	10.3	-67
44	0.01	2.5	2	25	69.6	41800	132	1.77	32.3	-5
S26	0.01	2.5	2	25	68.4	41000	164	1.58	31.7 <sup>g</sup>	-10
45	0.01	5.0	2	25	63.3	38000	174	1.54	35.7	9
S27	0.01	5.0	2	25	62.4	37400	201	1.58	35.2 <sup>g</sup>	9
46	0.01	5.0	4	50	161	96500	308	1.34	32.8 <sup>g</sup>	-4
47	0.02	7.5	2	25	92.6	27800	178	1.54	37.1	17

<sup>a</sup>Conditions: toluene + CHP total 10 mL, d-MAO 3.0 mmol, 10 min. <sup>b</sup> Initial CHP concentration (mol/L). <sup>c</sup>Activity= kg-polymer/mol-Ti·h. <sup>d</sup>GPC data in *o*-dichlorobenzene vs polystyrene standards. <sup>e</sup>CHP content (mol%) estimated by <sup>13</sup>C NMR spectra. <sup>f</sup>By DSC thermograms. <sup>g</sup>Estimated on the basis of the plots of  $T_g$ and CHP content.

run	cat. 1	TCUE	Е	yield	activity <sup>c</sup>	${M_{ m n}}^d$	$M_{ m w}/$	cont. <sup>e</sup>	$T_{\rm g}(T_{\rm m})^f$
	/ µmol	conc. <sup>b</sup> / M	/ atm	/ mg		×10 <sup>-4</sup>	$M_{\rm n}{}^d$	/ mol %	/ °C
48	0.02	1.0	4	74.8	22400	20.4	1.53	19.5	43
49	0.1	1.0	2	98.7	5920	6.16	1.52	26.5	64
50	0.5	2.5	2	155	1860	2.34	1.68	35.1	116
S28	0.5	2.5	2	140	1670	2.08	2.04	34.6 <sup>g</sup>	111
51	0.8	5.0	2	204	1530	1.38	2.10	38.8	130
S29	0.8	5.0	2	231	1730	1.65	1.81	39.4 <sup>g</sup>	134
52	0.5	5.0	2	222	1640	1.59	2.02	40.0 <sup>g</sup>	137

**Table S4.** Copolymerization of ethylene (E) and tricyclo[6.2.1.0(2,7)]undeca-4-ene (TCUE) by  $(1,2,4-Me_3C_5H_2)$ TiCl<sub>2</sub>(O-2, $6^{-i}$ Pr<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (1)–d-MAO catalyst.<sup>*a*</sup> Additional results.

<sup>*a*</sup>Conditions: toluene + TCUE total 10 mL, 25 °C, 10 min, d-MAO (prepared by removing toluene and AlMe<sub>3</sub> from the commercially available TMAO-212) 3.0 mmol. <sup>*b*</sup>Initial TCUE concentration (mol/L). <sup>*c*</sup>Activity= kg-polymer/mol-Ti·h. <sup>*d*</sup>GPC data in *o*-dichlorobenzene vs polystyrene standards. <sup>*e*</sup>TCUE content (mol%) estimated by <sup>13</sup>C NMR spectra. <sup>*f*</sup>By DSC thermograms. <sup>*g*</sup>Estimated on the basis of the plots of  $T_g$  and TCUE content. <sup>*h*</sup>d-MAO 2.0 mmol. <sup>*i*</sup>d-MAO 1.0 mmol.

2. Selected <sup>13</sup>C NMR spectra for resultant copolymers including assignment of resonances and estimation of comonomer contents.



**Figure S1**. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-COE) prepared by ('BuC<sub>5</sub>H<sub>4</sub>)TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**2**) - MAO catalyst [run 5, Table 1, COE 43.7 mol% (COE content in the whole polymer)].



**Figure S2.** <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 130 °C) for poly(ethylene-*co*-COE) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 7, Table 1, COE 22.5 mol%).



**Figure S3**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 130 °C) for poly(ethylene-*co*-COE) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 8, Table 1, COE 28.0 mol%).



**Figure S4**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 130 °C) for poly(ethylene-*co*-COE) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 10, Table 1, COE 28.1 mol%).



**Figure S5.** <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$ /bromobenzene- $d_5$  at 150 °C) for poly(ethylene*co*-COE) prepared by CpTiCl<sub>2</sub>(N=C'Bu<sub>2</sub>) (**5**) - MAO catalyst (run 15, Table 1, COE 20.2 mol%).



**Figure S6**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$ /bromobenzene- $d_5$  at 150 °C) for poly(ethylene*co*-COE) prepared by ('BuC<sub>5</sub>H<sub>4</sub>)TiCl<sub>2</sub>(N=C'Bu<sub>2</sub>) (**6**) - MAO catalyst (run 18, Table 1, COE 20.4 mol%).



**Figure S7**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$ /bromobenzene- $d_5$  at 150 °C) for poly(ethylene*co*-COE) prepared by (indenyl)TiCl<sub>2</sub>(N=C'Bu<sub>2</sub>) (7) - MAO catalyst (run 19, Table 1, COE 2.6 mol%).



**Figure S8**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$ /bromobenzene- $d_5$  at 150 °C) for poly(ethylene*co*-COE) prepared by (indenyl)TiCl<sub>2</sub>(N=C<sup>t</sup>Bu<sub>2</sub>) (7) - MAO catalyst (run 21, Table 1, COE 7.2 mol%).



**Figure S9.** <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 130 °C) for poly(ethylene-*co*-COE) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 31, Table 2, COE 7.0 mol%).



**Figure S10**. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-COE) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 32, Table 2, COE 16.1 mol%).



**Figure S11**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 130 °C) for poly(ethylene-*co*-COE) prepared by [Me<sub>2</sub>Si(C<sub>5</sub>Me<sub>4</sub>)(N'Bu)]TiCl<sub>2</sub> (**8**) - MAO catalyst (run 24, Table 1, COE 9.9 mol%).



**Figure S12**. <sup>13</sup>C NMR and the dept spectrum (in *o*-dichlorobenzene- $d_4$ /bromobenzene- $d_5$  at 150 °C) for poly(ethylene-*co*-COE) prepared by CpTiCl<sub>2</sub>(N=C'Bu<sub>2</sub>) (**5**) - MAO catalyst (run 15, Table 1, COE 20.2 mol%).



(spectrum, run 19, Table 1)

COE (mol%) = 
$$\frac{(C_{1,2} + C_{3,8} + C_{4,7} + C_{5,6})/8}{(C_{1,2} + C_{3,8} + C_{4,7} + C_{5,6})/8 + (C_{\alpha} + C_{\beta} + C_{\gamma} + C_{PE})/2} \times 100$$



**Figure S13**. <sup>13</sup>C NMR and the dept spectrum (in *o*-dichlorobenzene- $d_4$  at 110 °C) for poly(ethylene-*co*-COE) prepared by [Me<sub>2</sub>Si(Ind)<sub>2</sub>]ZrCl<sub>2</sub> (**11**) - MAO catalyst (run 28, Table 1).



**Figure S14**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 110 °C) for poly(ethylene-*co*-COE) prepared by [Me<sub>2</sub>Si(Ind)<sub>2</sub>]ZrCl<sub>2</sub> (**11**) - MAO catalyst (run 29, Table 1).



**Figure S15**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 110 °C) for poly(ethylene-*co*-COE) prepared by [Me<sub>2</sub>Si(Ind)<sub>2</sub>]ZrCl<sub>2</sub> (11) - MAO catalyst (run S21).



**Figure S16**. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-CHP) prepared by ( ${}^{t}BuC_{5}H_{4}$ )TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**2**) - MAO catalyst (run 40, Table 3, CHP 38.6 mol%).



**Figure S17.** <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-CHP) prepared by ( $^{t}BuC_{5}H_{4}$ )TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**2**) - MAO catalyst (run 42, Table 3, CHP 40.8 mol%).



**Figure S18**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 130 °C) for poly(ethylene-*co*-CHP) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 43, Table 3, CHP 10.3 mol%).



**Figure S19**. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-CHP) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 44, Table 3, CHP 32.3 mol%).



**Figure S20**. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-CHP) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 45, Table 3, CHP 35.7 mol%).



**Figure S21**. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-CHP) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 47, Table 3, CHP 37.1 mol%).



**Figure S22**. <sup>13</sup>C NMR and the dept spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene*co*-CHP) prepared by Cp\*TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**3**) - MAO catalyst (run 45, Table 3, CHP 35.7 mol%).

CHP (mol%) = 
$$\frac{(C_{1,2} + C_{7,3} + C_{4,6} + C_5)/7}{(C_{1,2} + C_{7,3} + C_{4,6} + C_5)/7 + (C_{\alpha,\beta} + C_{\gamma} + C_{PE})/2} \times 100$$



**Figure S23**. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) and the dept spectrum for poly(ethylene-*co*-CHP) prepared by ('BuC<sub>5</sub>H<sub>4</sub>)TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (**2**) - MAO catalyst (run 42, Table 3, CHP 40.8 mol%).



Figure S24. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-TCUE) prepared by (1,2,4-Me<sub>3</sub>C<sub>5</sub>H<sub>2</sub>)TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (1) - MAO catalyst (run 48, Table 4, TCUE 19.5 mol%).



Figure S25. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-TCUE) prepared by (1,2,4-Me<sub>3</sub>C<sub>5</sub>H<sub>2</sub>)TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (1) - MAO catalyst (run 49, Table 4, TCUE 26.5 mol%).



Figure S26. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-TCUE) prepared by (1,2,4-Me<sub>3</sub>C<sub>5</sub>H<sub>2</sub>)TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (1) - MAO catalyst (run 50, Table 4, TCUE 35.1 mol%).



Figure S27. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-TCUE) prepared by (1,2,4-Me<sub>3</sub>C<sub>5</sub>H<sub>2</sub>)TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (1) - MAO catalyst (run 51, Table 4, TCUE 38.8 mol%).



**Figure S28**. <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 130 °C) for poly(ethylene-*co*-TCUE) prepared by CpTiCl<sub>2</sub>(N=C'Bu<sub>2</sub>) (**5**) - MAO catalyst (run 56, Table 4, TCUE 9.4 mol%).



**Figure S29.** <sup>13</sup>C NMR spectrum (in *o*-dichlorobenzene- $d_4$  at 130 °C) for poly(ethylene-*co*-TCUE) prepared by CpTiCl<sub>2</sub>(N=C'Bu<sub>2</sub>) (**5**) - MAO catalyst (run 57, Table 4, TCUE 20.7 mol%).



**Figure S30**. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-*co*-TCUE) prepared by CpTiCl<sub>2</sub>(N=C'Bu<sub>2</sub>) (**5**) - MAO catalyst (run 58, Table 4, TCUE 31.7 mol%).



Figure S31. <sup>13</sup>C NMR spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene-co-TCUE) prepared by CpTiCl<sub>2</sub>(N=C<sup>t</sup>Bu<sub>2</sub>) (5) - MAO catalyst (run 62, Table 4, TCUE 40.7 mol%).



**Figure S32**. <sup>13</sup>C NMR and the dept spectrum (in 1,1,2,2-tetrachloroethane- $d_2$  at 110 °C) for poly(ethylene*co*-TCUE) prepared by (1,2,4-Me<sub>3</sub>C<sub>5</sub>H<sub>2</sub>)TiCl<sub>2</sub>(O-2,6-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) (1) - MAO catalyst (run 48, Table 4, TCUE 19.5 mol%).



### **3.** DSC thermograms in the copolymers.



**Figure S33**. DSC thermograms of polymers prepared by 1, 2, 4 - MAO catalysts in ethylene polymerization in the presence of COE. Detailed results are shown in Table 1 (runs 2, 5, 12).



**Figure S34**. DSC thermograms of poly(ethylene-*co*-COE)s prepared by  $Cp*TiCl_2(O-2,6-Cl_2C_6H_3)$  (3) - MAO or borate catalysts. Detailed results are shown in Table 1 (runs 8, 9).



Figure S35. Detailed results are shown in Table 1 (runs 13-21).



**Figure S36**. DSC thermograms of polymers prepared by **8** - MAO catalysts in ethylene polymerization in the presence of COE. Detailed results are shown in Table 1 (runs 22-24).



**Figure S37**. DSC thermograms of polymers prepared by **9** - MAO catalysts in ethylene polymerization in the presence of COE. Detailed results are shown in Table 1 (run 26).



**Figure S38**. DSC thermograms of polymers prepared by **10** - MAO catalyst in ethylene polymerization in the presence of COE. Detailed results are shown in Table 1 (run 27).



**Figure S39**. DSC thermograms of polymers prepared by **11** - MAO catalyst in ethylene polymerization in the presence of COE. Detailed results are shown in Table 1 (runs 28-30).



**Figure S40**. DSC thermograms for poly(ethylene-*co*-CHP)s prepared by Cp'TiCl<sub>2</sub>(O-2,6-<sup>*i*</sup>Pr<sub>2</sub>C<sub>6</sub>H<sub>3</sub>) [Cp' =  ${}^{t}$ BuC<sub>5</sub>H<sub>4</sub> (**2**), Cp\* (**3**)] – d-MAO catalysts. CHP content: 10.3 mol% (run 43, Table 3), 32.3 mol% (run 44), 35.7 mol% (run 45), 37.1 mol% (run 47), 38.6 mol% (run 40), and 40.8 mol% (run 42).



**Figure S41**. DSC thermograms of polymers prepared by **1-4** - MAO catalysts in ethylene polymerization in the presence of TCUE. Detailed results are shown in Table 4 (runs S28,53-55).



**Figure S42**. DSC thermograms of polymers prepared by **5-7** - MAO catalysts in ethylene polymerization in the presence of TCUE. Detailed results are shown in Table 4 (runs 62-64).



**Figure S43**. DSC thermograms of polymers prepared by **8** - MAO catalysts in ethylene polymerization in the presence of TCUE. Detailed results are shown in Table 4 (runs 65).

4. Plots of glass transition temperature  $(T_g)$  vs comonomer content (mol%) in ethylene copolymers with norbornene (NBE), tetracyclododecene (TCD), and with tricyclo[6.2.1.0(2,7)]undeca-4-ene (TCUE).



**Figure S44**. Plots of glass transition temperature ( $T_g$ ) vs comonomer content (mol%) in ethylene copolymers with norbornene (NBE, cited from: K. Nomura, *Chin. J. Polym. Sci.*, 2008, **26**, 513-523.), tetracyclododecene (TCD, cited from: W. Apisuk, H. Ito, and K. Nomura, *J. Polym. Sci. Part A: Polym. Chem.*, 2016, **54**, 2662-2667.), and with tricyclo[6.2.1.0(2,7)]-undeca-4-ene (TCUE).