

## A TCBD-based AB<sub>2</sub>-type Second-order Nonlinear Optical Hyperbranched Polymer Prepared by a Facile Click-type Postfunctionalization

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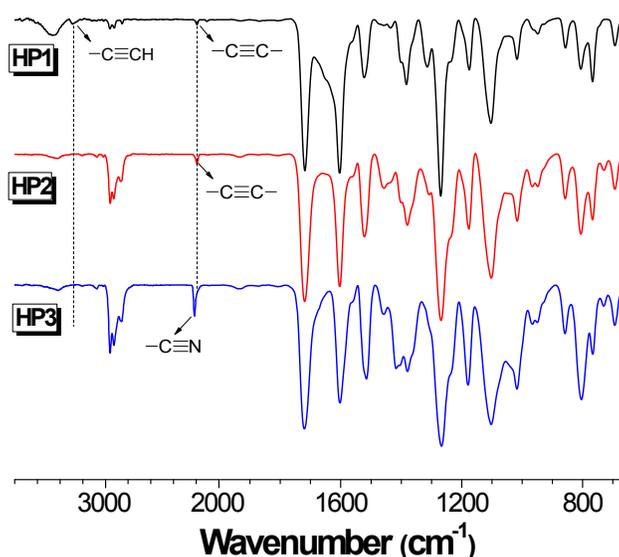
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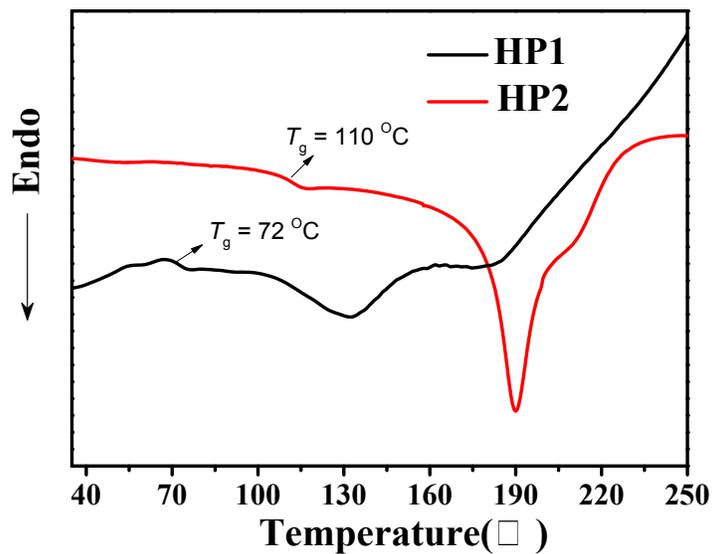
**Table S1.** Elemental analysis results.<sup>a</sup>

No.	N[%]	C[%]	H[%]
<b>HP1</b>	2.27 (2.57)	65.58 (74.84)	4.27 (4.99)
<b>HP2</b>	5.90 (7.99)	63.56 (71.97)	5.23 (6.33)
<b>HP3</b>	6.24 (/)	61.21 (/)	5.30 (/)

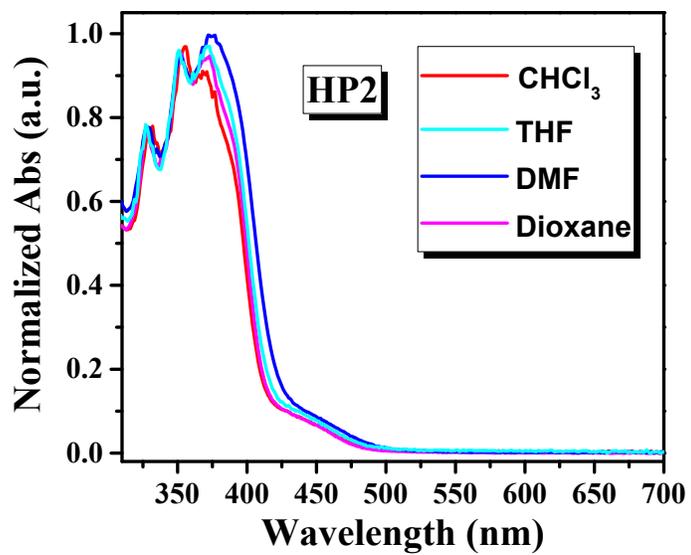
<sup>a</sup> The theoretical EA values of **HP1** and **HP2** are listed in brackets for comparison, except **HP3** due to the uncertainty of the exact ratio of TCBD groups. Note: A large deviation between experimental and theoretical data has been obtained. Bases on the data of DSC measurements, we attribute the deviation to the occurring of the cross-linking for **HP1** at 132 °C, while at 190 °C for **HP2** and **HP3**, which would make the HPs more thermally stable and thus lead to a large deviation for EA results.



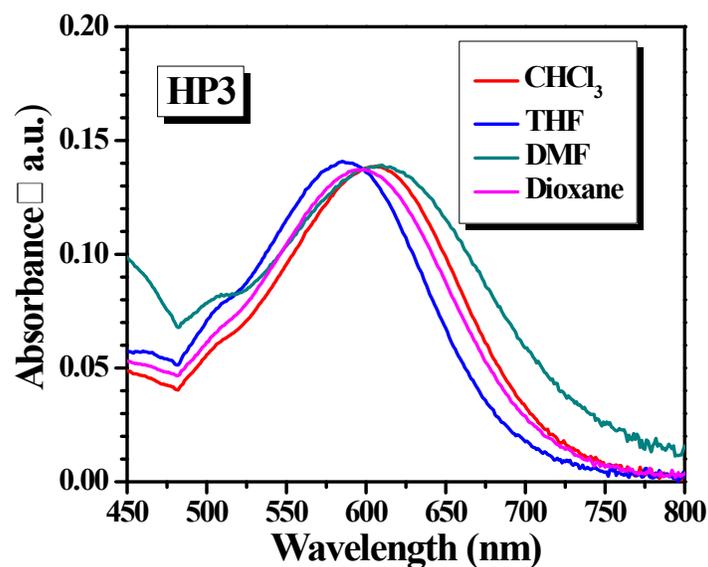
**Fig. S1** FT-IR spectra of **HP1-HP3**.



**Fig. S2** DSC curves of **HP1-HP2** measured under nitrogen at a heating rate of 10 °C/min.

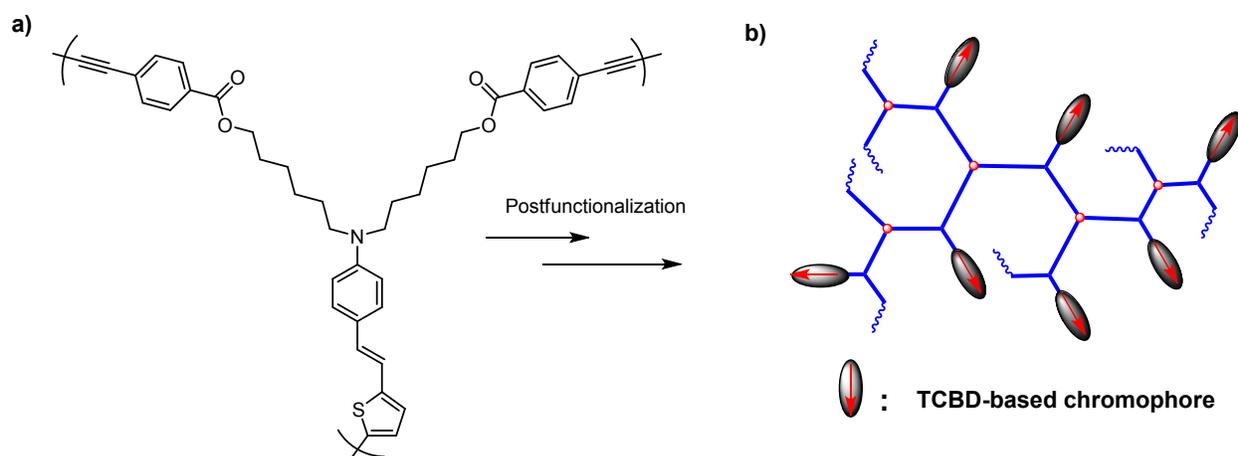


**Fig. S3** Solution absorption spectra of **HP2** in different solvents.

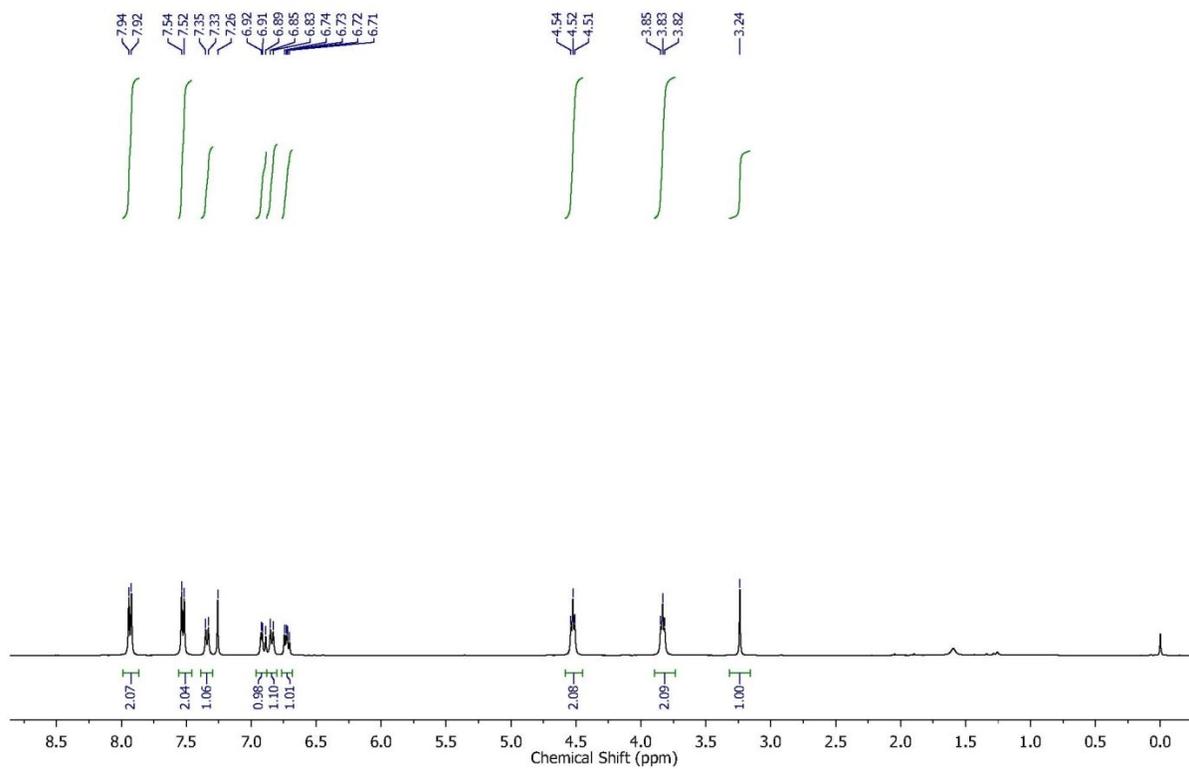


Solvents	THF	Dioxane	CHCl <sub>3</sub>	DMF
$\lambda_{\max}$	587 nm	598 nm	608 nm	610 nm

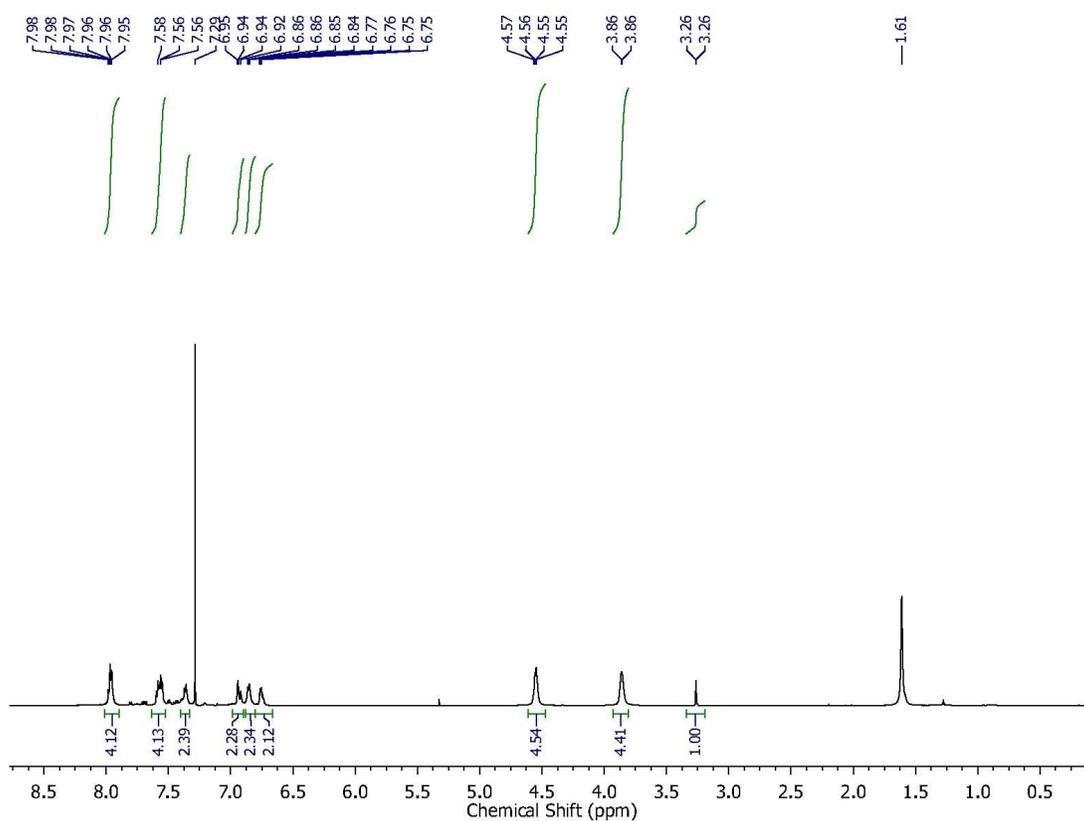
**Fig. S4** Solution absorption spectra of **HP3** in different solvents, with the absorption maxima listed in Table.



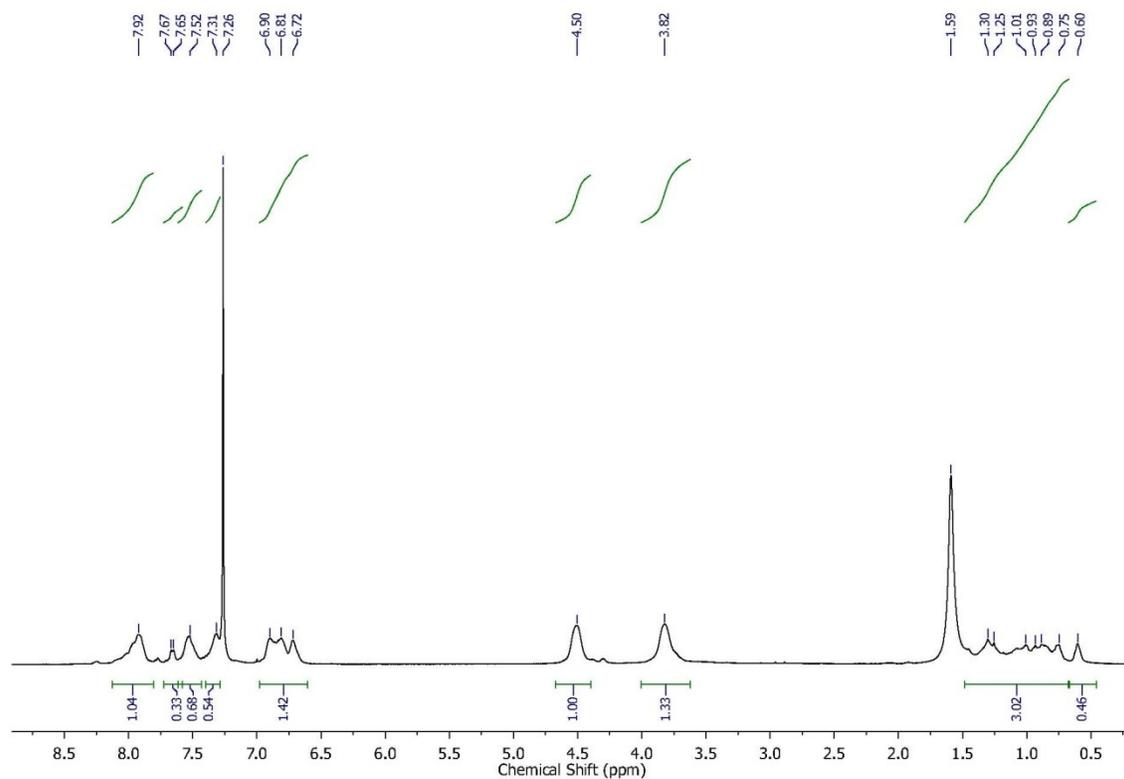
**Fig. S5.** Proposed two design strategies for improving the NLO effects of TCBD-based hyperbranched polymers.



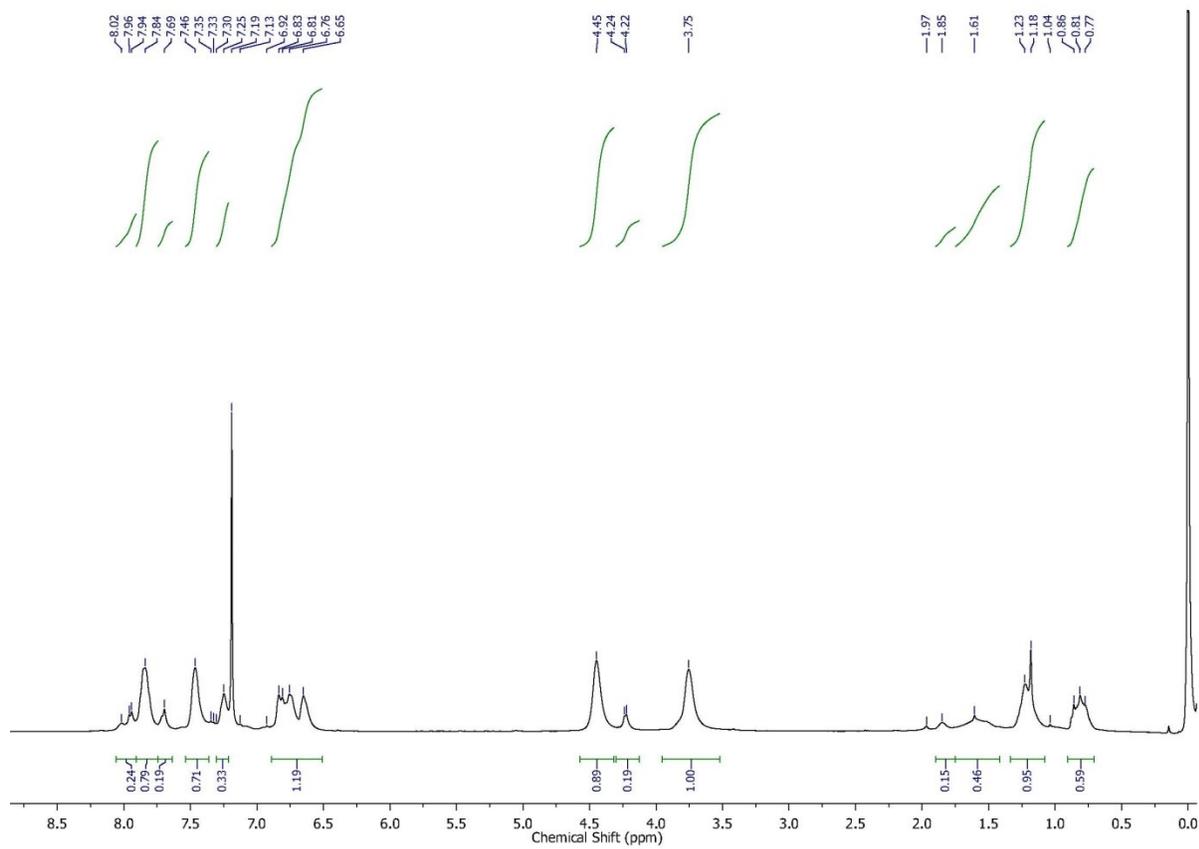
**Fig. S6**  $^1\text{H}$  NMR spectrum of monomer **3** conducted in  $\text{CDCl}_3$ .



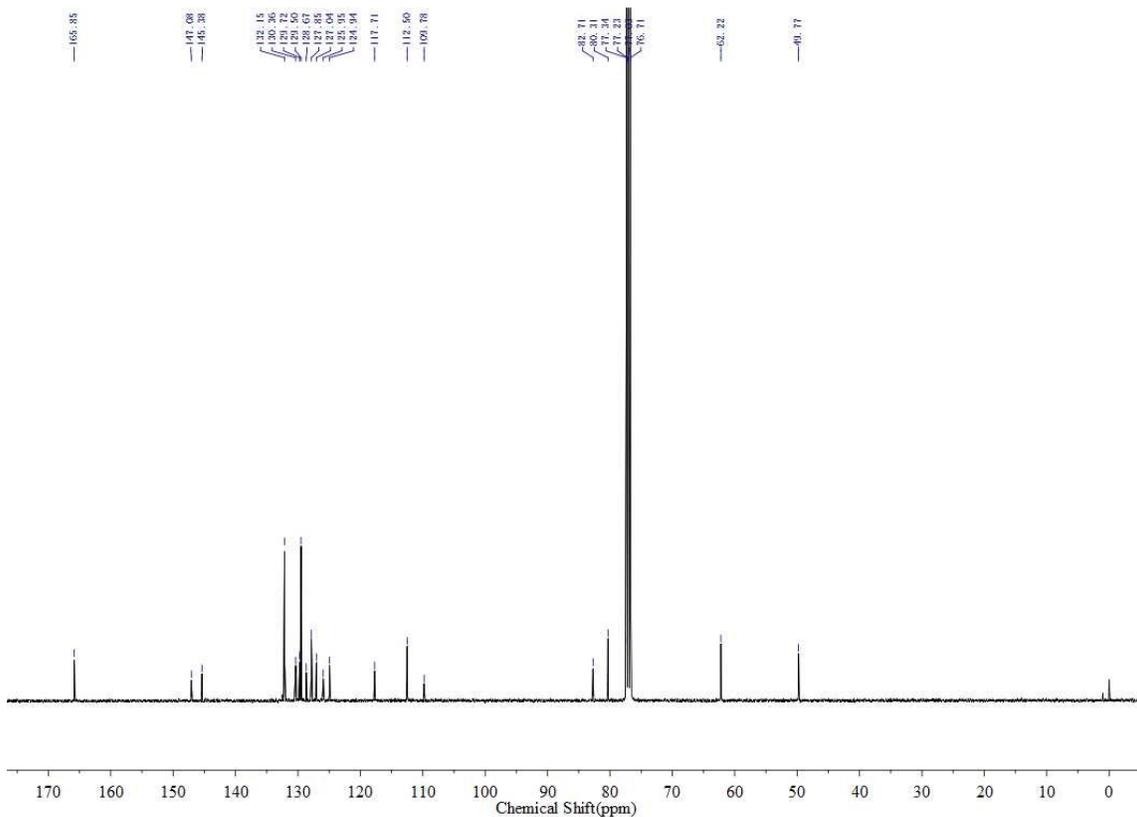
**Fig. S7**  $^1\text{H}$  NMR spectrum of **HP1** conducted in  $\text{CDCl}_3$ .



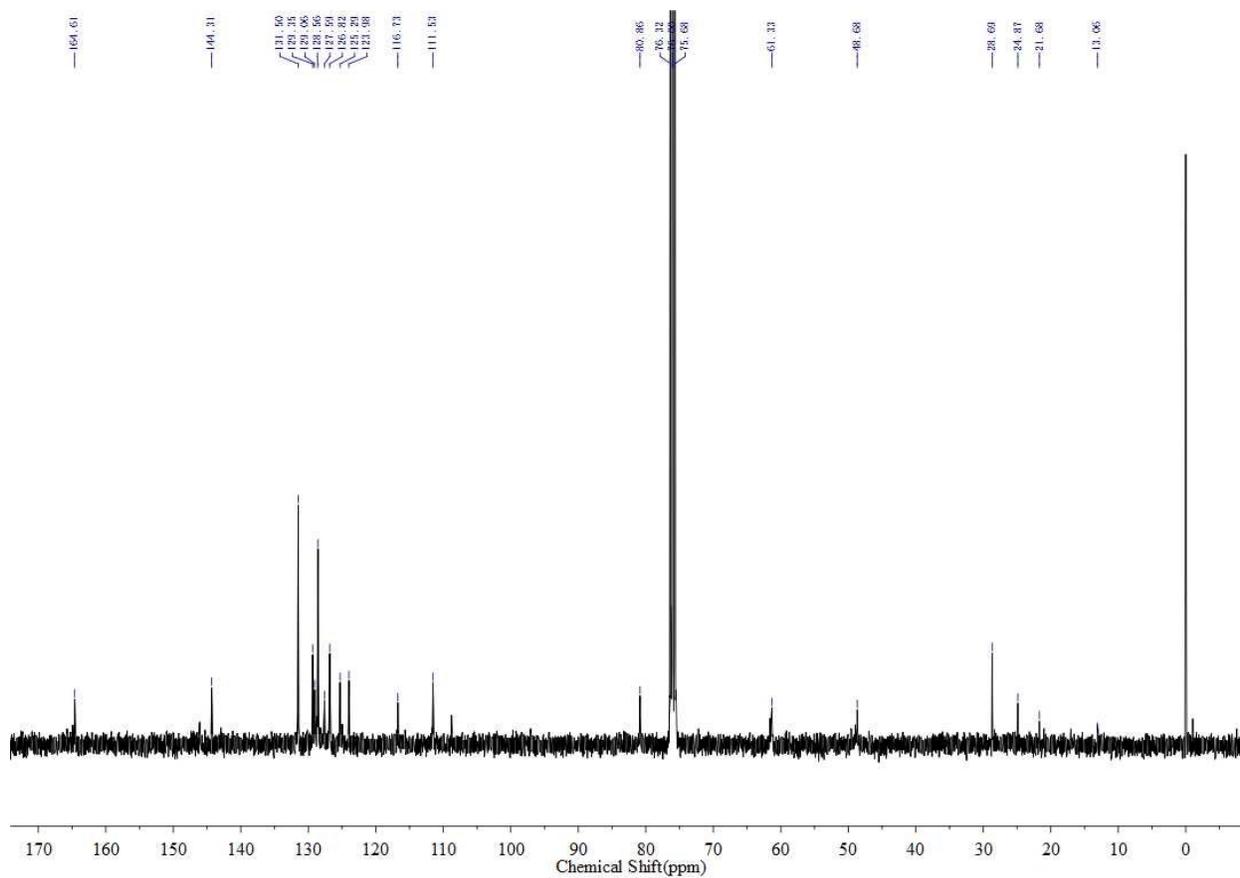
**Fig. S8**  $^1\text{H}$  NMR spectrum of **HP2** conducted in  $\text{CDCl}_3$ .



**Fig. S9**  $^1\text{H}$  NMR spectrum of **HP3** conducted in  $\text{CDCl}_3$ .



**Fig. S10.**  $^{13}\text{C}$  NMR spectrum of **HP1** conducted in  $\text{CDCl}_3$ .



**Fig. S11.**  $^{13}\text{C}$  NMR spectrum of **HP2** conducted in  $\text{CDCl}_3$ .

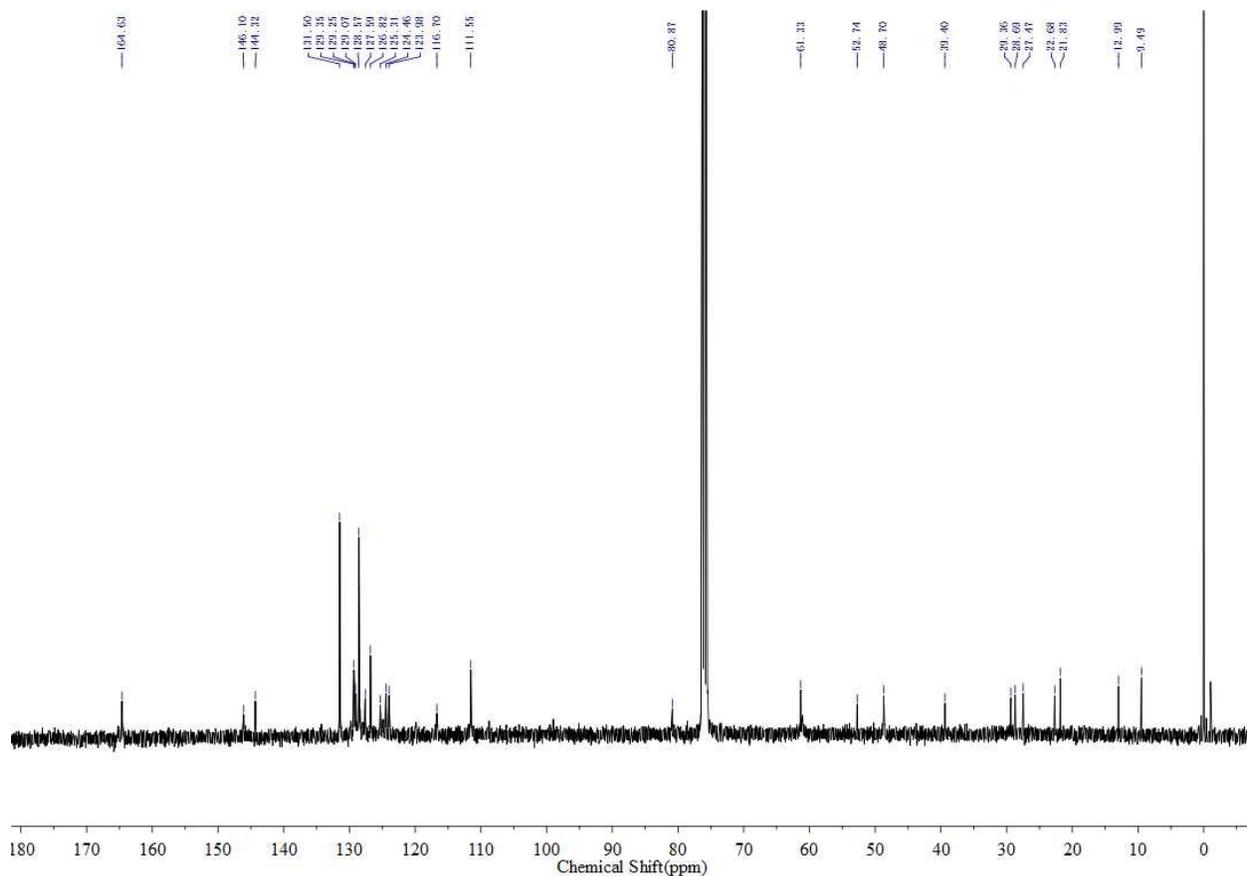
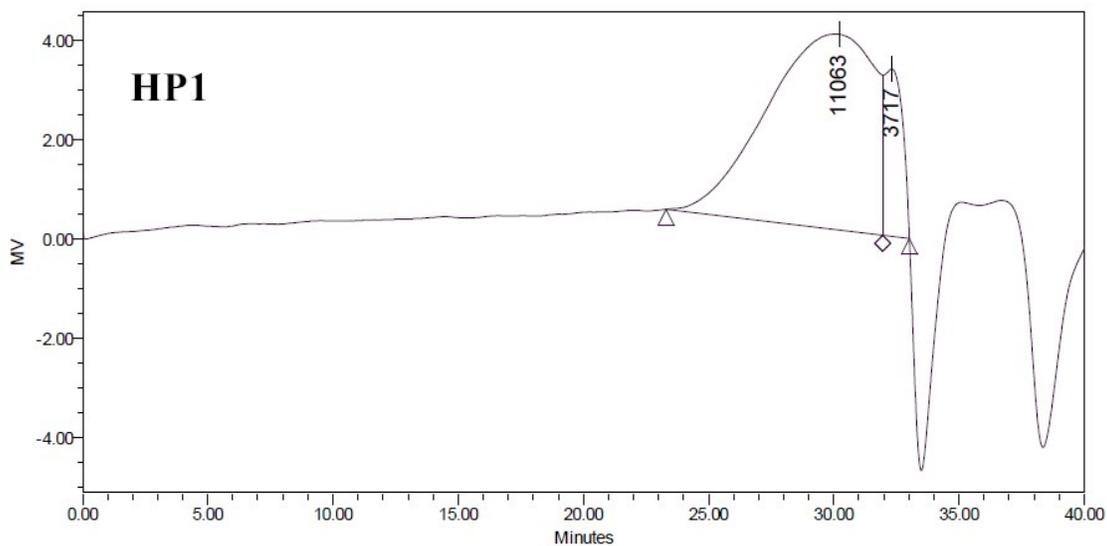


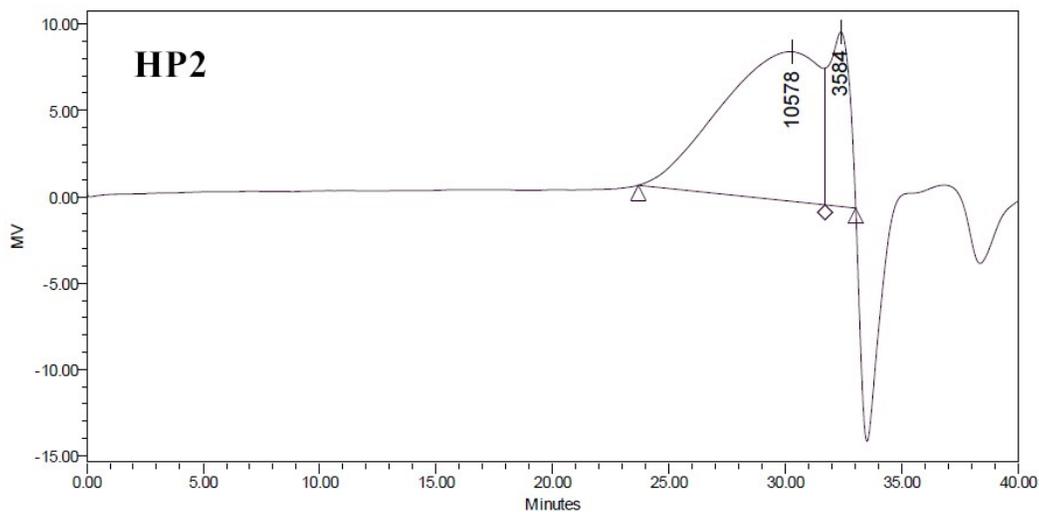
Fig. S12.  $^{13}\text{C}$  NMR spectrum of **HP3** conducted in  $\text{CDCl}_3$ .



Broad Unknown Modified Universal Peak Table

	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						12962	22112	11063	38146	58197	1.705968	1.725086
2								3717				

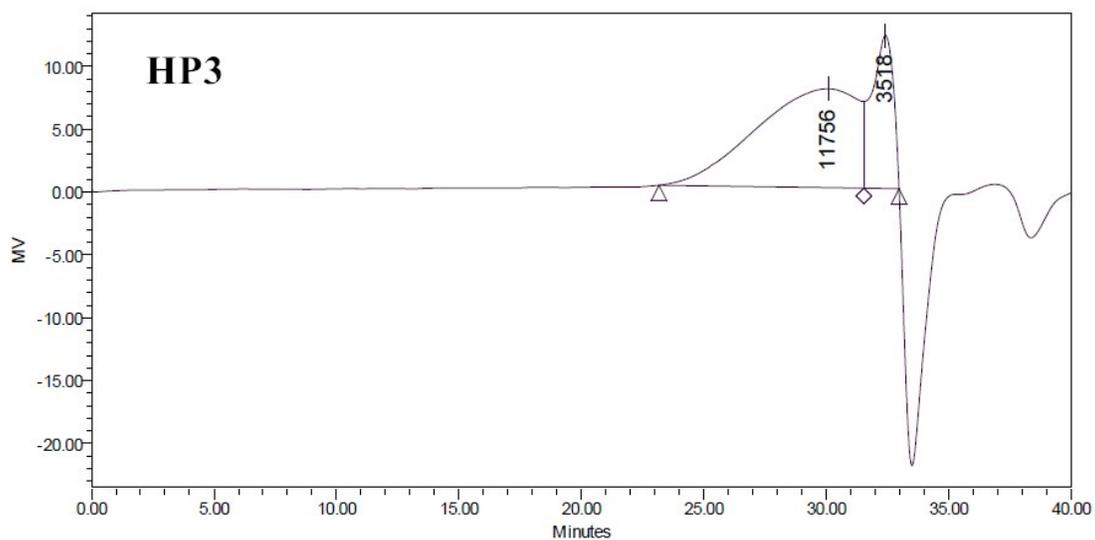
Fig. S13. GPC curve and data of **HP1**.



**Broad Unknown Modified Universal Peak Table**

	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1								3584				
2						14128	23746	10578	40050	58621	1.680735	1.686634

**Fig. S14. GPC curve and data of HP2.**



**Broad Unknown Modified Universal Peak Table**

	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						15240	25330	11756	43440	66150	1.662149	1.714945
2								3518				

**Fig. S15. GPC curve and data of HP3.**

# Mass Spectrum List Report

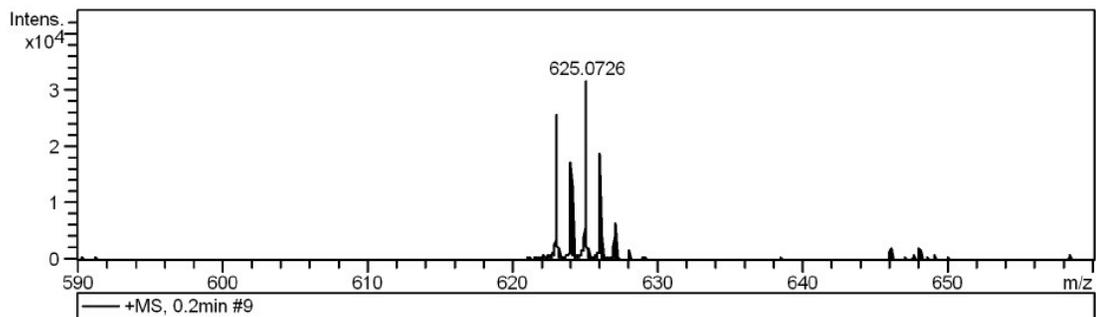
## Analysis Info

Analysis Name D:\Data\Lizhongan\li-zangxiaobo20190710-1.d  
Method tune\_low\_20170906\_50-1200.m  
Sample Name li-zangxiaobo20190710-1  
Comment

Acquisition Date 7/10/2019 10:52:20 AM  
Operator BDAL@DE  
Instrument / Ser# micrOTOF 10401

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Active			Set Dry Heater	200 °C
Scan Begin	50 m/z	Set Capillary	4000 V	Set Dry Gas	4.0 l/min
Scan End	1200 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste



#	m/z	Res.	S/N	I	FWHM
1	623.0726	10188	204.0	25590	0.0612
2	624.0777	9805	135.7	17036	0.0636
3	625.0726	10566	252.2	31742	0.0592
4	626.0754	10675	148.0	18636	0.0586
5	627.0781	10532	51.3	6462	0.0595
6	628.0778	10744	12.7	1605	0.0585

**Fig. S16.** HR-mass spectrum of monomer **3**.