Supplementary Information

Well-defined hydrogen and organofunctional polysiloxanes with spiro-

fused siloxane backbones

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General Information

All experimental manipulations involving air-sensitive compounds were performed in a nitrogen atmosphere of MBRAUN LABmaster Pro SP glove box.

All chemicals were reagent grade and used as received without further purification. Tris(pentafluorophenyl)borane (B(C₆F₅)₃), tetrakis(dimethylsilyloxy)silane, phenylsilane, Karstedt's catalyst, 1-octene, allyl glycidyl ether, ethylene glycol monoallyl ether, and 4-chlorostyrene were purchased from Tokyo Chemical Industry Co., Ltd. Toluene, *n*-hexane, acetone, and CDCl₃ were purchased from Fujifilm Wako Pure Chemical Corporation. *n*-Hexylsilane was purchased from Aldrich. Vinylpentamethyldisiloxane was purchased from Gelest, Inc.

Column chromatography was performed with silica gel (Kanto Silica gel 60 N, 100-210 µm) or activated alumina (Fujifilm Wako Pure Chemical Corporation). Gel Permeation Chromatography (GPC) was performed with a YMC Multiple Preparative HPLC LC-Forte/R using YMC-GPC T4000-40 & YMC-GPC T2000-40 columns.

NMR spectra were recorded on Bruker AVANCE III HD (¹H NMR at 600 MHz; ¹³C{¹H} NMR at 150 MHz; ²⁹Si{¹H} NMR at 119MHz) NMR spectrometer. The High-resolution ESI mass spectra were obtained on Bruker micrOTOF II. The matrix-assisted laser desorption ionisation time-of-flight (MALDI-TOF) mass spectra were collected on Bruker Autoflex Speed. Differential Scanning Calorimetry (DSC) analyses were performed using HITACHI DSC7020.

Size-exclusion chromatography (SEC) was performed at 45 °C using a Waters ACQUITY Advanced Polymer Chromatography (APC) System consisting of a p-Isocratic Solvent Manager (Model AIS), Sample Manager pFTN (Model ASM), Column Manager-S (Model AZC), PDA TS Detector (Model ADT), and Refractive Index (RI) Detector (Model URI) equipped with a Waters APCTM XT45 column (linear, 4.6 mm × 150 mm; pore size, 4.5 nm; bead size, 1.7 µm; exclusion limit, 5000), a Waters APCTM XT200 column (linear, 4.6 mm × 150 mm; pore size, 20.0 nm; bead size, 2.5 µm; exclusion limit, 70 000), and a Waters APCTM XT450 column (linear, 4.6 mm × 150 mm; pore size, 45.0 nm; bead size, 2.5 µm; exclusion limit, 400 000) in toluene at a flow rate of 0.70 mL min⁻¹. The numberaverage molecular weight $(M_{n,SEC})$ and the molecular weight dispersity (D_M) were determined based on a calibration curve prepared using polystyrene (PS) samples from a TSKgel[®] standard polystyrene oligomer kit (Tosoh) with weight-average molecular mass (M_w) and (D_M) values of 19.0×10⁵ kDa (1.04), 9.64×10^4 g mol⁻¹ (1.01), 3.79×10^4 g mol⁻¹ (1.01), 1.74×10^4 g mol⁻¹ (1.01), 1.02×10^4 g mol⁻¹ (1.02), 5.06×10^3 g mol⁻¹ (1.02), 2.63×10^3 g mol⁻¹ (1.05), 1.01×10^3 g mol⁻¹ (1.16), and 5.9×10^2 g mol⁻¹ ¹ (1.19), along with PS samples from Chemco Co. with M_w (D_M) values of 17.0×10⁵ g mol⁻¹ (<1.06), 4.75×10^4 g mol⁻¹ (1.06), 9.00×10^3 g mol⁻¹ (1.04), and 4.00×10^3 g mol⁻¹ (1.03). The values of refractive index increment, dn/dc, were found to be positive for 2a and negative for 2b and 2c.

nalvailavana		M.W. of separated peak/Da					
porysnoxane		<i>n</i> = 3	<i>n</i> = 4	<i>n</i> = 5	<i>n</i> = 6		
2.	PS-equivalent M.W.	1230	1630	2070	2540		
2a	actual M.W.	2208	2944	3680	4416		
21	PS-equivalent M.W.	1130	1490	1890	2330		
20	actual M.W.	2232	2977	3721	4465		
2c	PS-equivalent M.W.	1020	1330	1730	2160		
	actual M.W.	2257	3009	3761	4513		

Table S1. Calculated molecular weights relative to poly(styrene) standards and actual molecularweights of separated peaks in approximate number distribution of polysiloxanes 2



Figure S1. Comparing the ¹H NMR spectra of spirosiloxane 1a, *n*-hexylsilane, and polysiloxane 2b. (a) spirosiloxane 1a, (b) *n*-hexylsilane, and (c) polysiloxane 2b ($M_{n,SEC} = 3.86$ kDa, $D_M = 2.96$), in CDCl₃.



n = 3



Figure S3. Positive-ion MALDI-TOF mass spectrum of macrocyclic polysiloxanes **2b** ($M_{n,SEC} = 3.86$ kDa, $D_M = 2.96$) acquired in reflector mode using DCTB as the matrix and sodium trifluoroacetate as the cationising agent.



Figure S4. Comparing the ¹H NMR spectra of spirosiloxane 1b, phenylsilane, and polysiloxane 2c. (a) spirosiloxane 1b, (b) phenylsilane, and (c) polysiloxane 2c ($M_{n,SEC} = 2.51$ kDa, $D_M = 2.40$), in CDCl₃.



n = 3



Figure S6. Positive-ion MALDI-TOF mass spectrum of macrocyclic polysiloxanes **2c** ($M_{n,SEC} = 2.51$ kDa, $D_M = 2.40$) acquired in reflector mode using DCTB as the matrix and sodium trifluoroacetate as the cationising agent.



Figure S7. ¹H NMR spectra of (a) polysiloxane 2a' prepared under more concentrated conditions ($[1a]_0 = 25 \text{ mM}$) and (b) polysiloxane 2a" prepared under more diluted conditions ($[1a]_0 = 6.25 \text{ mM}$).



Figure S8. MALDI-TOF MS spectrum of (a) polysiloxane 2a' prepared under more concentrated conditions ($[1a]_0 = 25 \text{ mM}$) and (b) polysiloxane 2a" prepared under more diluted conditions ($[1a]_0 = 6.25 \text{ mM}$).



Figure S9. Molecular weight distributions of (a) polysiloxane 2a' prepared under more concentrated conditions ($[1a]_0 = 25 \text{ mM}$) and (b) polysiloxane 2a" prepared under more diluted conditions ($[1a]_0 = 6.25 \text{ mM}$), determined by SEC using toluene as the eluent, narrowly dispersed poly(styrene)s as standards, and a differential refractometer detector.



Figure S10. A zoomed-in version of the molecular weight distributions of polysiloxanes 2a-2c determined by SEC



Figure S11. SEC curves of polysiloxanes **2a-2c** using toluene as the eluent and a differential refractometer detector.



Figure S12. Comparing ¹H NMR spectra of (a) polysiloxane 2a, (b) polysiloxane 3a, (c) polysiloxane 3b, (d) polysiloxane 3c, (e) polysiloxane 3d, and (f) polysiloxane 3e, in CDCl₃.



Figure S13. Molecular weight distributions of polysiloxanes **3a-3e** determined by SEC using toluene as the eluent, narrowly dispersed poly(styrene)s as standards, and a differential refractometer detector.

Preparation of Spirosiloxane Monomers 1

4,12-diisopropoxy-2,2,6,6,10,10,14,14-octamethyl-4,12-diphenyl-1,3,5,7,9,11,13,15-octaoxa-2,4,6,8,10,12,14-heptasilaspiro[7.7]pentadecane (1a)



B(C₆F₅)₃ (34.2 mg, 5 mol%) and acetone (390 μL, 5.2 mmol) were dissolved in toluene (20 mL). To the mixture was added Si(OSiMe₂H)₄ (482 μL, 1.3 mmol) with stirring at room temperature. After 30 min, phenylsilane (321 μL, 2.6 mmol) was added. After 1 h, acetone (195 μL, 2.6 mmol) was added. After 2 h, the reaction mixture was passed through a short silica gel pad (eluent: *n*-hexane). The crude product was purified by GPC (eluent: *n*-hexane) to give **1a** as a colourless oil (551.6 mg, 59%). ¹H NMR (600 MHz, CDCl₃): δ 7.68-7.62 (m, 4H), 7.44-7.40 (m, 2H), 7.39-7.33 (m, 4H), 4.26 (sep, *J* = 6.1 Hz, 2H), 1.22 (d, *J* = 6.1Hz, 12H), 0.26 (s, 6H), 0.20 (s, 6H), 0.13 (s, 6 H), 0.07 ppm (s, 6H). ¹³C {¹H} NMR (150 MHz, CDCl₃): δ 134.3, 133.5, 130.1, 127.8, 65.6, 25.5, 0.62, 0.54, 0.45, 0.37. ²⁹Si {¹H} NMR (119 Hz, CDCl₃): δ -16.72, -16.73, -72.7, -105.6. HRMS (ESI): *m/z* calculated for [C₂₆H₅₂NO₁₀Si₇] (M+NH₄): 734.1971; found 734.1963.

4,12-dihexyl-4,12-diisopropoxy-2,2,6,6,10,10,14,14-octamethyl-1,3,5,7,9,11,13,15-octaoxa-2,4,6,8,10,12,14-heptasilaspiro[7.7]pentadecane (1b)



B(C₆F₅)₃ (76.8 mg, 5 mol%) and acetone (901 µL, 12.0 mmol) were dissolved in toluene (45 mL). To the mixture was added Si(OSiMe₂H)₄ (1116 µL, 3.0 mmol) with stirring at room temperature. After 30 min, *n*-hexylsilane (987 µL, 6.0 mmol) was added at 0 °C. After 20 h, acetone (540 µL, 7.2 mmol) was added. After 8 h, the reaction mixture was passed through a short silica gel pad (eluent: *n*-hexane). The crude product was purified by GPC (eluent: *n*-hexane) to give **1b** as a colourless oil (1.35 g, 61%). ¹H NMR (600 MHz, CDCl₃): δ 4.19 (sep, *J* = 6.1 Hz, 2H), 1.44-1.21 (m, 16H), 1.18 (d, *J* = 6.1 Hz, 12 H), 0.88 (t, *J* = 7.0 Hz, 6H), 0.60-0.52 (m, 4H), 0.16 (s, 6H), 0.15 (s, 6H), 0.131 (s, 6H) 0.126 (s, 6H). ¹³C{¹H} NMR (150 MHz, CDCl₃): δ 65.0, 32.8, 31.7, 25.6, 23.1, 22.7, 14.3, 13.2, 0.52, 0.51 (two signals are missing due to overlap).

²⁹Si{¹H} NMR (119 Hz, CDCl₃): δ –18.015, –18.024, –59.8, –105.5.

HRMS (ESI): *m/z* calculated for [C₂₆H₆₈NO₁₀Si₇] (M+NH₄): 750.3223; found 750.3214.

Preparation of Hydrogen Polysiloxanes 2

hydrogen polysiloxane 2a



B(C₆F₅)₃ (10.2 mg, 1 mol%) and **1a** (1.44 g, 2.0 mmol) were dissolved in toluene (160 mL). To the mixture was added phenylsilane (247 μ L, 2.0 mmol) with stirring at room temperature. After 20 h, the reaction mixture was passed through a short alumina pad (eluent: *n*-hexane), and the volatiles was removed under reduced pressure to give hydrogen polysiloxane **2a** as a colourless oil (1.46 g, >95%). The crude product was analysed by NMR, SEC, DSC, and MALDI-TOF MS without further purification and used for the next reactions.

¹H, ¹³C{¹H}, and ²⁹Si{¹H} NMR spectra of the product are shown in Figure S9-S11.



hydrogen polysiloxane 2b

B(C₆F₅)₃ (2.6 mg, 1 mol%) and **1a** (358.6 mg, 0.5 mmol) were dissolved in toluene (40 mL). To the mixture was added *n*-hexylsilane (80.6 μ L, 0.5 mmol) with stirring at room temperature. After 20 h, the reaction mixture was passed through a short alumina pad (eluent: *n*-hexane), and the volatiles was removed under reduced pressure to give hydrogen polysiloxane **2b** as a colourless oil (354.0 mg, 95%). The crude product was analysed by NMR, SEC, DSC, and MALDI-TOF MS without further purification.

¹H, ¹³C{¹H}, and ²⁹Si{¹H} NMR spectra of the product are shown in Figure S12-S14.

hydrogen polysiloxane 2c



B(C₆F₅)₃ (1.2 mg, 1 mol%) and **2a** (178.1 mg, 0.24 mmol) were dissolved in toluene (19 mL). To the mixture was added phenylsilane (29.6 μ L, 0.24 mmol) with stirring at room temperature. After 20 h, the reaction mixture was passed through a short alumina pad (eluent: *n*-hexane), and the volatiles was removed under reduced pressure to give hydrogen polysiloxane **2c** as a colourless oil (173.7 mg, >95%). The crude product was analysed by NMR, SEC, DSC, and MALDI-TOF MS without further purification.

¹H, ¹³C{¹H}, and ²⁹Si{¹H} NMR spectra of the product are shown in Figure S15-S17.

Typical Procedure for the Synthesis of Polysiloxane 3a

Karstedt's catalyst (0.1 mg, 0.1 μ mol) and **2a** (73.7 mg, 0.10 mmol) were dissolved in toluene (0.4 mL). To the mixture was added 1-octene (16 μ L, 0.10 mmol) with stirring at room temperature. After 10 h, the reaction mixture was passed through a short silica gel pad (eluent: *n*-hexane) to give polysiloxane **3a** as a pale yellow oil (84.0 mg, >95%). The crude product was used for NMR and MALDI-TOF MS analyses without further purification.

Other polysiloxanes **3b-3e** were also synthesised in the same procedure by using different olefins.

polysiloxane 3a



pale yellow oil (84.0 mg, >95%) 1 H, 13 C{ 1 H}, and 29 Si{ 1 H} NMR spectra of the product are shown in Figure S18-S20.

polysiloxane 3b



pale yellow oil (80.1 mg, 94%) ^{1}H , $^{13}C{^{1}H}$, and $^{29}Si{^{1}H}$ NMR spectra of the product are shown in Figure S21-S23.

polysiloxane 3c



pale yellow oil (78.8 mg, 94%) ^{1}H , $^{13}C\{^{1}H\}$, and $^{29}Si\{^{1}H\}$ NMR spectra of the product are shown in Figure S24-S26.

polysiloxane 3d



yellow oil (85.8 mg, >95%) ^{1}H , $^{13}C\{^{1}H\}$, and $^{29}Si\{^{1}H\}$ NMR spectra of the product are shown in Figure S27-S29.

polysiloxane 3e



pale yellow oil (90.3 mg, >95%) 1 H, 13 C{ 1 H}, and 29 Si{ 1 H} NMR spectra of the product are shown in Figure S30-S32.



Figure S14. 1H NMR Spectra of Spirosiloxane 1a

Figure S15. 13C NMR Spectra of Spirosiloxane 1a



Figure S16. 29Si NMR Spectra of Spirosiloxane 1a



Figure S17. 1H NMR Spectra of Spirosiloxane 1b







~	· · · · ·	· · · · · · · · · · · · · · · · · · ·			 J		 	WDW SSB LB GB PC	EM 0 1.00 Hz 0 1.40
	Hex, \sim O	Me₂ O ^{_Si} .O Si Si O _{Si} .O Me₂ 1	Me₂ O´ ^{Si} .O Si Si O`SíO Me₂ b	OHex				NAME S EXPNO PROCNO F2 - Acqu: Date_ Time INSTRUM PROBHD PULPROG TD SOLVENT NS SWH FIDRES AQ RG DW DE TE D1 D11 TD0 ===== C SFO1 NUC1 P1 PLW1 ==== C SFO2 NUC2 CPDPRG[2 PCPD2 PLW2 PLW12 PLW13 F2 - Proc SI SF	spirosiloxanelb 11 1 isition Parameter 20200916 7.43 spect 5 mm CPBBO BB- zgpg30 65536 CDC13 64 2 33333.332 Hz 0.508626 Hz 0.9830400 sec 193.87 15.000 usec 19.62 usec 298.0 K 2.00000000 sec 0.03000000 sec 1 HANNEL f1 ===================================

Figure S19. 29Si NMR Spectra of Spirosiloxane 1b



Figure S20. 1H NMR Spectra of Polysiloxane 2a

5.15





	$Me_2 \qquad Me_2 \qquad Ph \qquad H \\ Ph \qquad O^{-Si} O O^{-Si} O^$	Current Data Parameters NAME polysiloxane2a EXPNO 10 PROCNO 1 F2 - Acquisition Parameter Date_ 20190827 Time 12.22 INSTRUM spect PROBHD 5 mm PABBO BB/ PULPROG 2g30 TD 65536 SOLVENT CDC13 NS 8 DS 2 SWH 12019.230 Hz FIDRES 0.183399 Hz AQ 2.7262976 sec RG 32 DW 41.600 usec DE 10.66 usec TE 298.0 K D1 1.0000000 sec TD0 1
9 8 7 86R		0 ppm

Figure S21. 13C NMR Spectra of Polysiloxane 2a



Figure S22. 29Si NMR Spectra of Polysiloxane 2a



Figure S23. 1H NMR Spectra of Polysiloxane 2b



Figure S24. 13C NMR Spectra of Polysiloxane 2b



Figure S25. 29Si NMR Spectra of Polysiloxane 2b



Figure S26. 1H NMR Spectra of Polysiloxane 2c



Figure S27. 13C NMR Spectra of Polysiloxane 2c



Figure S28. 29Si NMR Spectra of Polysiloxane 2c



Figure S29. 1H NMR Spectra of Polysiloxane 3a







Figure S30. 13C NMR Spectra of Polysiloxane 3a



Figure S31. 29Si NMR Spectra of Polysiloxane 3a



Figure S32. 1H NMR Spectra of Polysiloxane 3b





SF01 600.1330006 MHz NUC1 1H P1 11.00 usec PLW1 25.00000000 MHz WDW EM SSB 0 LB 0.20 Hz GB 0 PC 1.00		$\begin{array}{cccc} & Me_2 & Me_2 \\ Ph_{O}^{Si_{O}} & O^{Si_{O}} \\ & Si_{Si}^{O} & Si_{O}^{O} \\ & Si_{O}^{O} & O^{Si_{O}} \\ & Me_2 & Me_2 \end{array}$	Ph,			$\begin{array}{cccc} \mbox{Current Data Parameters} \\ \mbox{NAME} & \mbox{polysiloxane3b} \\ \mbox{EXPNO} & 10 \\ \mbox{PROCNO} & 1 \\ \mbox{F2} - \mbox{Acquisition Parameter} \\ \mbox{Date} & 20191023 \\ \mbox{Time} & 11.15 \\ \mbox{Instrum} & \mbox{spect} \\ \mbox{PrOBHD} & 5 \mbox{mm PABBO BB} \\ \mbox{PuLPROG} & \mbox{zg30} \\ \mbox{TD} & 65536 \\ \mbox{SOLVENT} & \mbox{CDC13} \\ \mbox{NS} & 8 \\ \mbox{DS} & 2 \\ \mbox{SWH} & 12019.230 \mbox{Hz} \\ \mbox{FIDRES} & 0.183399 \mbox{Hz} \\ \mbox{AQ} & 2.7262976 \mbox{sec} \\ \mbox{RG} & 25.4 \\ \mbox{DW} & 41.600 \mbox{usec} \\ \mbox{DE} & 10.66 \mbox{usec} \\ \mbox{DE} & 298.0 \mbox{K} \\ \mbox{D1} & 1.00000000 \mbox{sec} \\ \mbox{TD0} & 1 \\ \end{array}$
9 8 7 6 5 4 3 2 1 0 ppm	9 8 7	6 5 4	3 2	1	0 ppm	CHANNEL f1 SF01 600.1330006 MHz NUC1 1H P1 11.00 usec PLW1 25.00000000 W F2 - Processing parameters SI 65536 SF SF 600.1300000 MHz WDW EM SSB 0 LB 0.20 Hz GB 0 PC 1.00

Figure S33. 13C NMR Spectra of Polysiloxane 3b



Figure S34. 29Si NMR Spectra of Polysiloxane 3b



Figure S35. 1H NMR Spectra of Polysiloxane 3c





Figure S36. 13C NMR Spectra of Polysiloxane 3c



Figure S37. 29Si NMR Spectra of Polysiloxane 3c





Figure S39. 13C NMR Spectra of Polysiloxane 3d



Figure S40. 29Si NMR Spectra of Polysiloxane 3d



Figure S41. 1H NMR Spectra of Polysiloxane 3e





SiMe	Current Data Parameters NAME polysiloxane3e EXPNO 10 PROCNO 1
$\begin{array}{c} Me_2 & Me_2 & Ph, \\ Me_2 & Me_2 & Ph, \\ Ph, & O^{-Si}, & O, & O^{-Si}, \\ O, & Si, & Si, \\ O^{-Si}, & O, & O^{-Si}, \\ O^{-Si}, & O^{-Si}, & O^{-Ph}, \\ Me_2 & Me_2 \end{array}$	$\begin{array}{ccccc} F2 & - \ Acquisition \ Parameter \\ Date_ & 20200716 \\ Time & 8.27 \\ INSTRUM & spect \\ PROBHD & 5 \ mm \ CP2 \ BB-1H \\ PULPROG & zg30 \\ TD & 65536 \\ SOLVENT & CDC13 \\ NS & 8 \\ DS & 2 \\ SWH & 12019.230 \ Hz \\ FIDRES & 0.183399 \ Hz \\ AQ & 2.7262976 \ sec \\ RG & 16 \\ DW & 41.600 \ usec \\ DE & 15.65 \ usec \\ TE & 298.0 \ K \\ D1 & 1.0000000 \ sec \\ TD0 & 1 \\ \end{array}$
	ID0 I SF01 600.1330006 MHz NUC1 1H P1 12.00 usec PLW1 15.00000000 W F2 - Processing parameters 65536 SF 600.1300128 MHz WDW EM SSB 0 LB 0.20 Hz GB 0 PC 1.00
9 8 7 6 5 4 3 2 1 0 ppm	

Figure S42. 13C NMR Spectra of Polysiloxane 3e



Figure S43. 29Si NMR Spectra of Polysiloxane 3e

