

Supporting Information

In-situ “grafting from” for the synthesis of polymer brushes on upconversion nanoparticles via NIR-mediated RAFT polymerization

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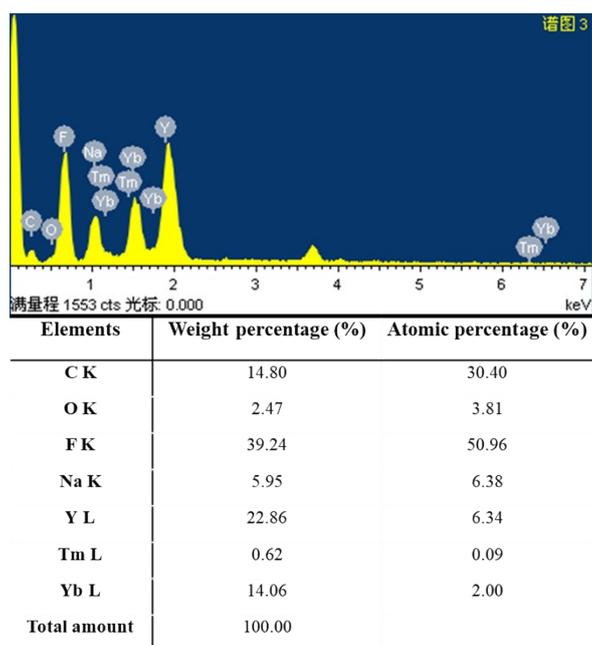


Figure S1. EDS spectrum of upconversion nanoparticles with OA molecular capped.

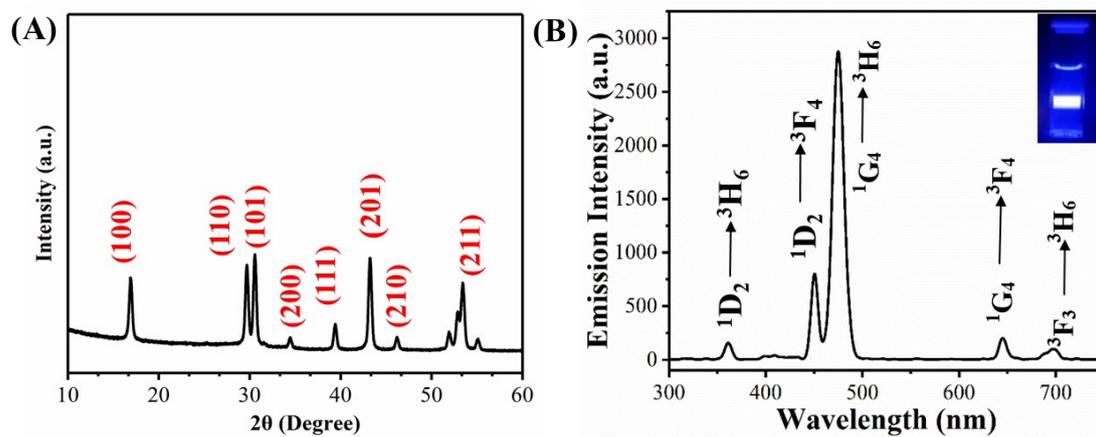


Figure S2. (A) Wide-angle XRD pattern of synthesized UCNP@OA; (B) Upconversion luminescence (UCL) spectrum of UCNP@OA (excited with 980 nm laser).

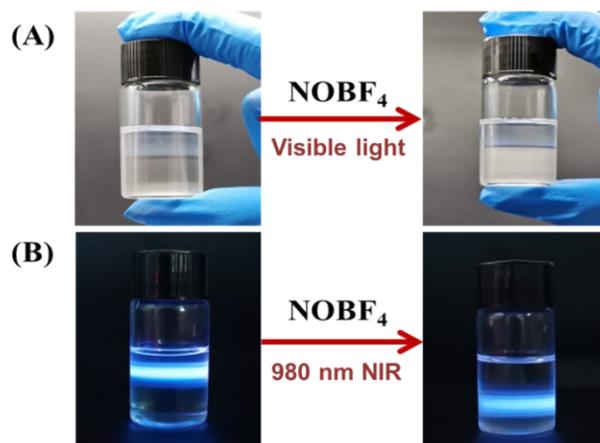


Figure S3. Digital photographs of upconversion nanoparticles dispersions (A) and upconversion luminescence (B) of NaYF₄:Yb/Tm nanoparticles before and after NOBF₄ modification, respectively. (The upper layer is cyclohexane and the bottom layer is DMF)

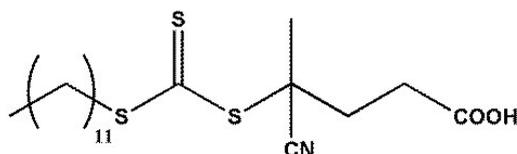


Figure S4. Chemical Structure of CDTPA as RAFT agent used in this study.

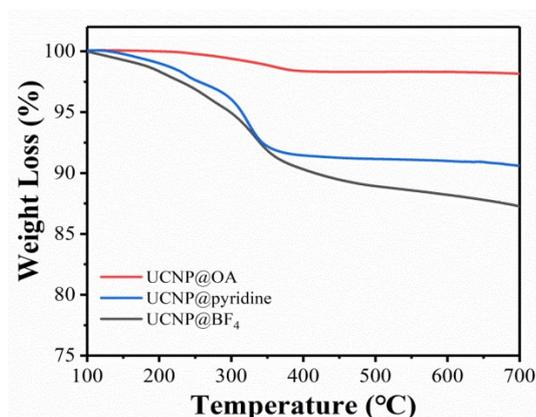


Figure S5. TGA analysis of UCNPs@ligand after ligand exchange with CDTPA. The same amounts (10 mg) of UCNPs@OA, UCNPs@pyridine, UCNPs@BF₄ were added to the 2 mL DMF solution of CDTPA (0.05 mmol/mL), and centrifuged to collect the powders after stirring for 3 h, separately.

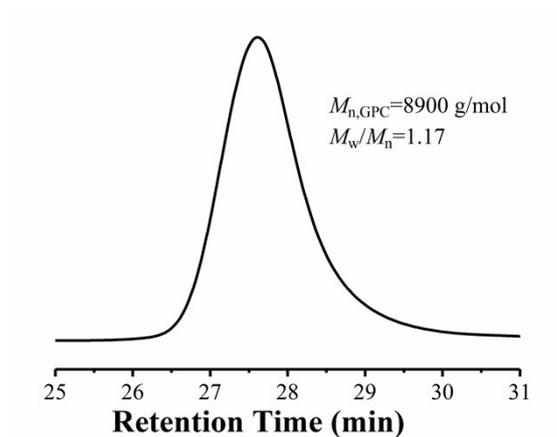


Figure S6. SEC characterization of purified free PMMA corresponding to the same sample of ^1H NMR characterization.

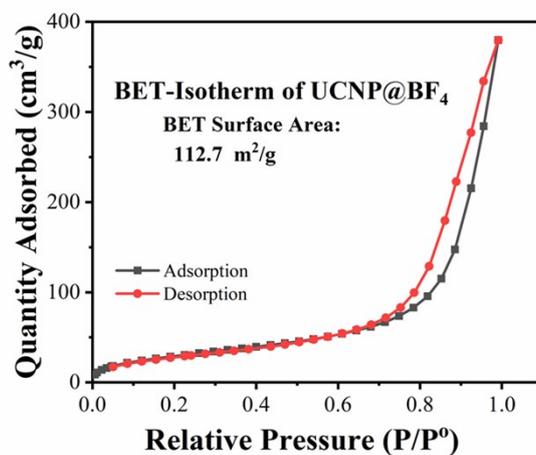


Figure S7. Nitrogen adsorption–desorption isotherms of UCNP@BF $_4^-$ sample.

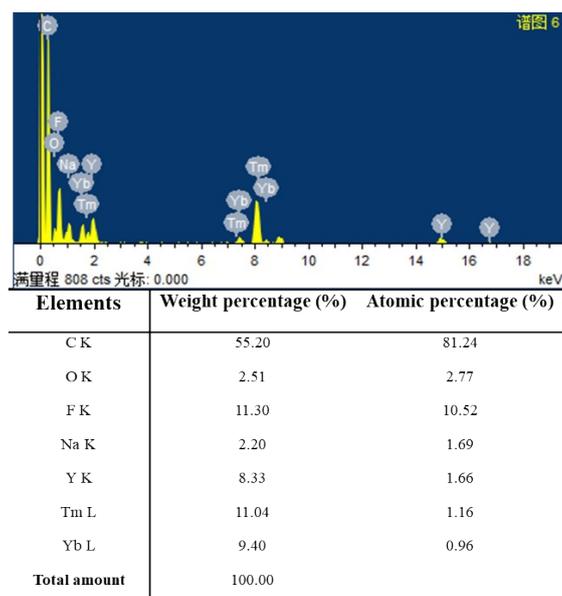


Figure S8. EDS spectrum of upconversion nanoparticles with PMMA capped.

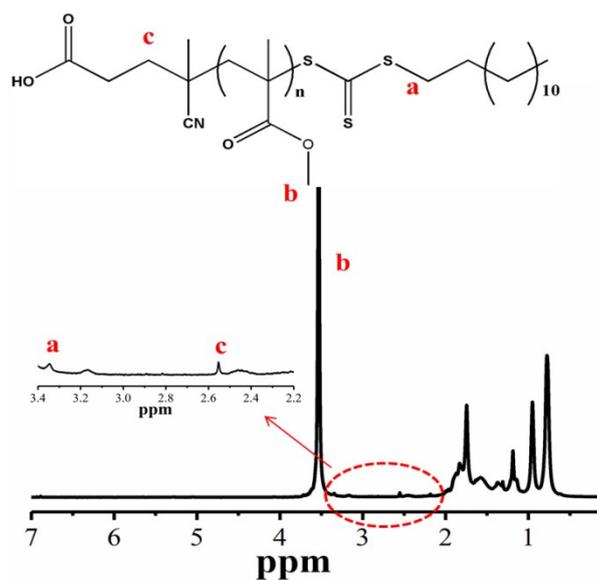


Figure S9. ^1H NMR spectra of UCNP grafted PMMA ended with RAFT agent synthesized by NIR-mediated RAFT polymerization.

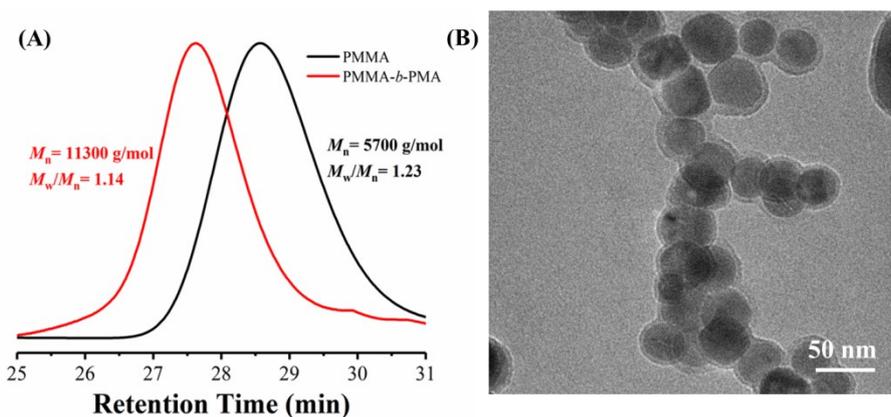


Figure S10. Molecular weight distributions of corresponding free polymers of UCNP@PMMA and UCNP@PMMA-*b*-PMA synthesized by grafting from the approach using NIR-mediated RAFT followed by chain extension with MA. (b) TEM image of the UCNP@PMMA-*b*-PMA after chain extension.

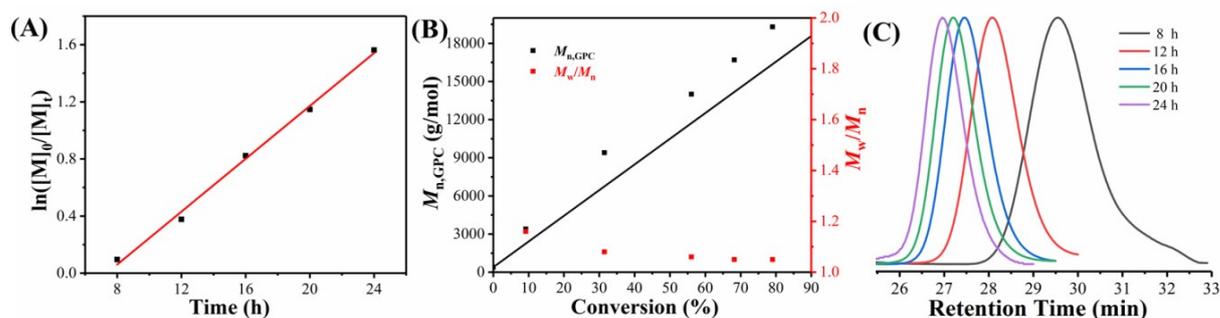


Figure S11. Kinetic analysis of the NIR-initiated RAFT polymerization of *t*BA under the conditions $[tBA]_0:[CDTPA]_0=200:1$ with 5 mg/mL UCNPs in 50% (v/v) DMSO at room temperature. (A) Plot of $\ln([M]_0/[M]_t)$ versus exposure time. (B) Evolution of number-average molecular weight and molecular weight distribution (M_w/M_n). (C) Molecular weight distributions at different time points.

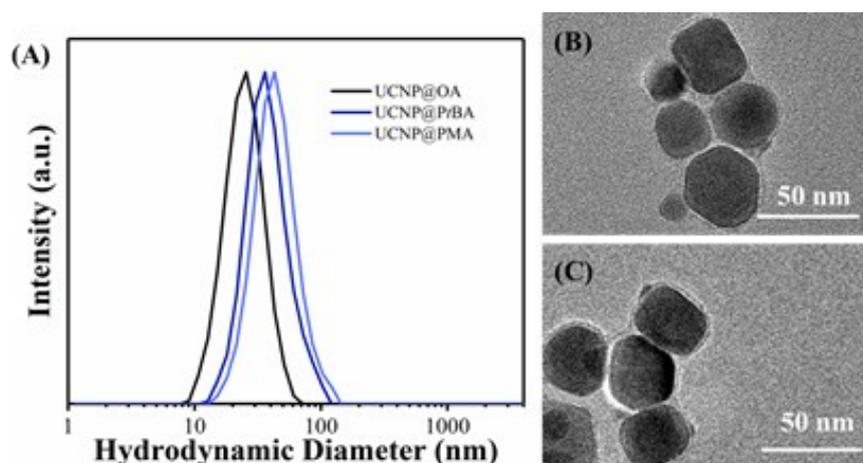


Figure S12. (A) DLS analysis of different nanoparticles of as-synthesized UCNP@OA, and PMA and PtBA coated particles. (B, C) TEM images of UCNP@PtBA and UCNP@PMA, respectively.

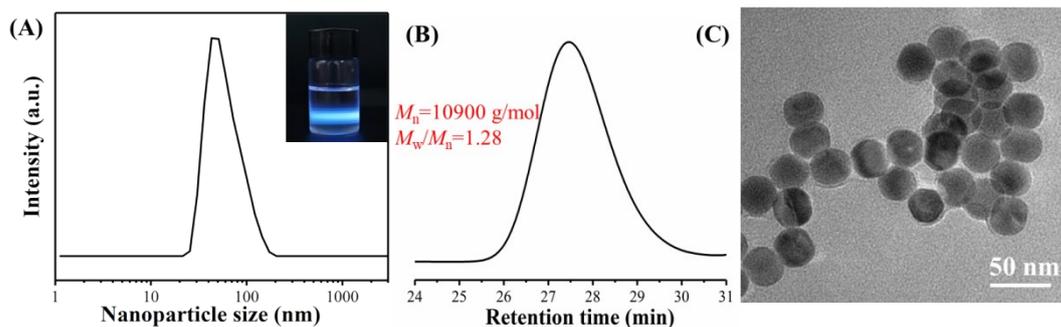


Figure S13. (A) DLS analysis, SEC characterization (B) and TEM image (C) of as-synthesized UCNP@PAA nanoparticles.

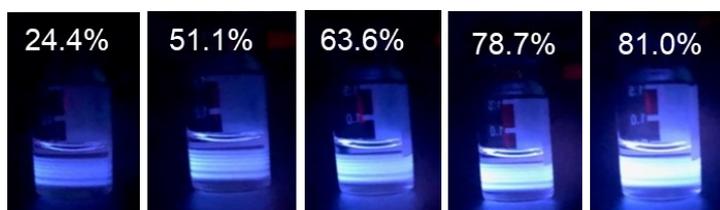


Figure S14. Digital photographs of upconversion nanoparticles during the polymerization process at different monomer conversions. (Reaction conditions: $[MMA]_0:[CDTPA]_0=200:1$ with 5 mg/mL UCNPs in 50% v/v DMSO at room temperature under total 980 nm laser irradiation).

The amount of CDTPA and polymers grafted onto the surface of the upconversion nanoparticles was calculated using the weight loss (*loss-wt*) based on the residual weight at 650°C, as determined by TGA (**Figure 6**) and the surface area of UCNPs.

The grafting density (*GD*) was estimated according to the following equation:

$$\text{Grafting Density} \left(\frac{\text{chain}}{\text{nm}^2} \right) = \frac{(\text{loss-wt}/M_{n,CDTPA}) \times N_a}{m_{UCNPs} \times S_{UCNPs}} \quad (S1)$$

$$\text{Grafting Density} \left(\frac{\text{chain}}{\text{nm}^2} \right) = \frac{(\text{loss-wt}/M_{n,polymer}) \times N_a}{m_{UCNPs} \times S_{UCNPs}} \quad (S2)$$

Where $M_{n,CDTPA}$, $M_{n,polymer}$ corresponds to the molecular weight of CDTPA and polymer grafted, respectively. N_a is Avogadro constant and m_{UCNPs} is the molar mass

of polymer grafted UCNPs used for the TGA analysis (e.g. mass of modified nanoparticles = initial mass before TGA analysis – loss of weight). The weight loss was calculated through the difference between the weights at 100°C temperature and at 650°C. S_{UCNPs} is the specific surface area of UCNPs. For example, for UCNP@CDTPA and UCNP@PMMA-14 h:

$$\text{Grafting Density of CDTPA} \left(\frac{\text{chain}}{\text{nm}^2} \right) = \frac{(0.116/403.67 \text{ g/mol}) \times 6.02 \times 10^{23}}{(1 - 0.116) \times 112.7 \text{ m}^2/\text{g}} = 1.736 \left(\frac{\text{chain}}{\text{nm}^2} \right) \quad (S3)$$

$$\begin{aligned} \text{Grafting Density of PMMA} \left(\frac{\text{chain}}{\text{nm}^2} \right) \\ = \frac{(0.295/13900 \text{ g/mol}) \times 6.02 \times 10^{23}}{(1 - 0.295) \times 112.7 \text{ m}^2/\text{g}} = 0.161 \left(\frac{\text{chain}}{\text{nm}^2} \right) \quad (S4) \end{aligned}$$

Table S1. Results of polymers synthesized by the “grafting from” approach using NIR-initiated RAFT polymerization with [MMA]:[CDTPA] = [200]: [1] in DMSO solvent at room temperature

Entry	Time (h)	$M_{n,GPC}^a$ (g/mol)	M_w/M_n^a	Shell Thickness ^b (nm)	Grafting Density ^c (chains/nm ²)
1	6	7500	1.21	~ 3.4	0.094
2	10	10800	1.29	~ 5.8	0.131
3	14	13900	1.22	~ 8.5	0.162
4	18	15500	1.14	~ 11.9	0.201

^a Number-average molecular weight $M_{n,GPC}$ and dispersity (M_w/M_n) of corresponding free PMMA (unbound) after different polymerization time determined by GPC using polystyrene calibration. The reaction were performed at room temperature under 980

nm NIR laser 6 W/cm²) in DMSO. ^b The thickness of the core-shell structure was measured by TEM. ^c The grafting density of the polymer on the UCNPs was calculated by TGA analysis.

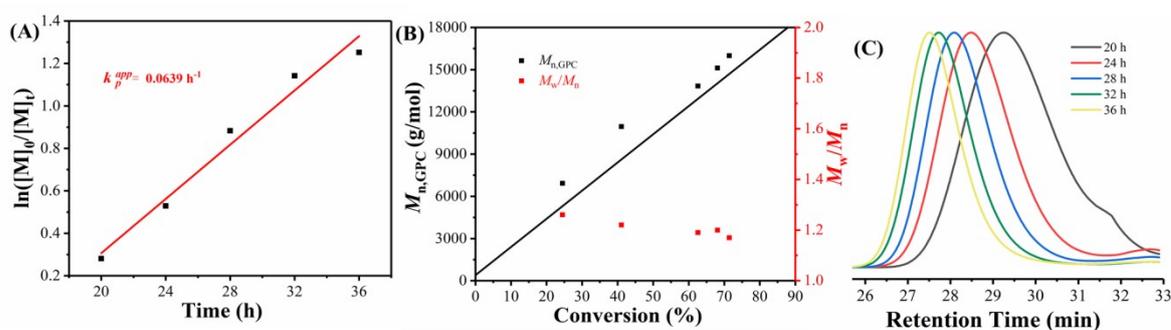


Figure S15. Kinetic analysis of the NIR-initiated RAFT polymerization of MMA when the polymerization vessel was shielded by chicken skin with a thickness of 1.2 mm under the conditions $[MMA]_0:[CDTPA]_0=200:1$ with 5 mg/mL UCNPs in 50% (v/v) DMSO at room temperature. (A) Plot of $\ln([M]_0/[M]_t)$ versus exposure time. (B) Evolution of number-average molecular weight and molecular weight distribution (M_w/M_n). (C) Molecular weight distributions at different time points.

Compared with nonscreened reaction ($k_p^{app} = 0.0849 \text{ h}^{-1}$), the apparent growth rate constant have a slight decrease ($k_p^{app} = 0.0639 \text{ h}^{-1}$).