

Supporting Information

Synthesis and free radical photopolymerization of triphenylamine-based oxime ester photoinitiators

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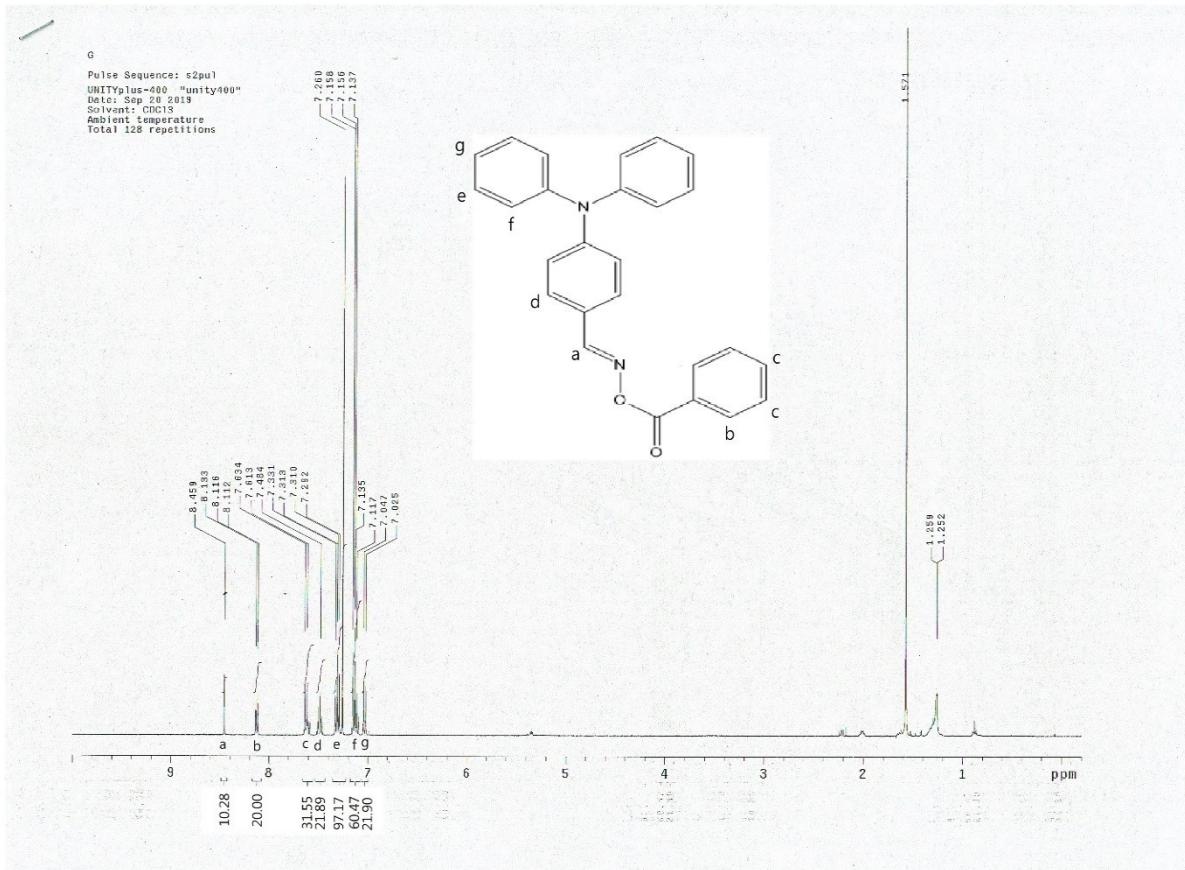


Fig. S1 ¹H NMR of the TP-1.

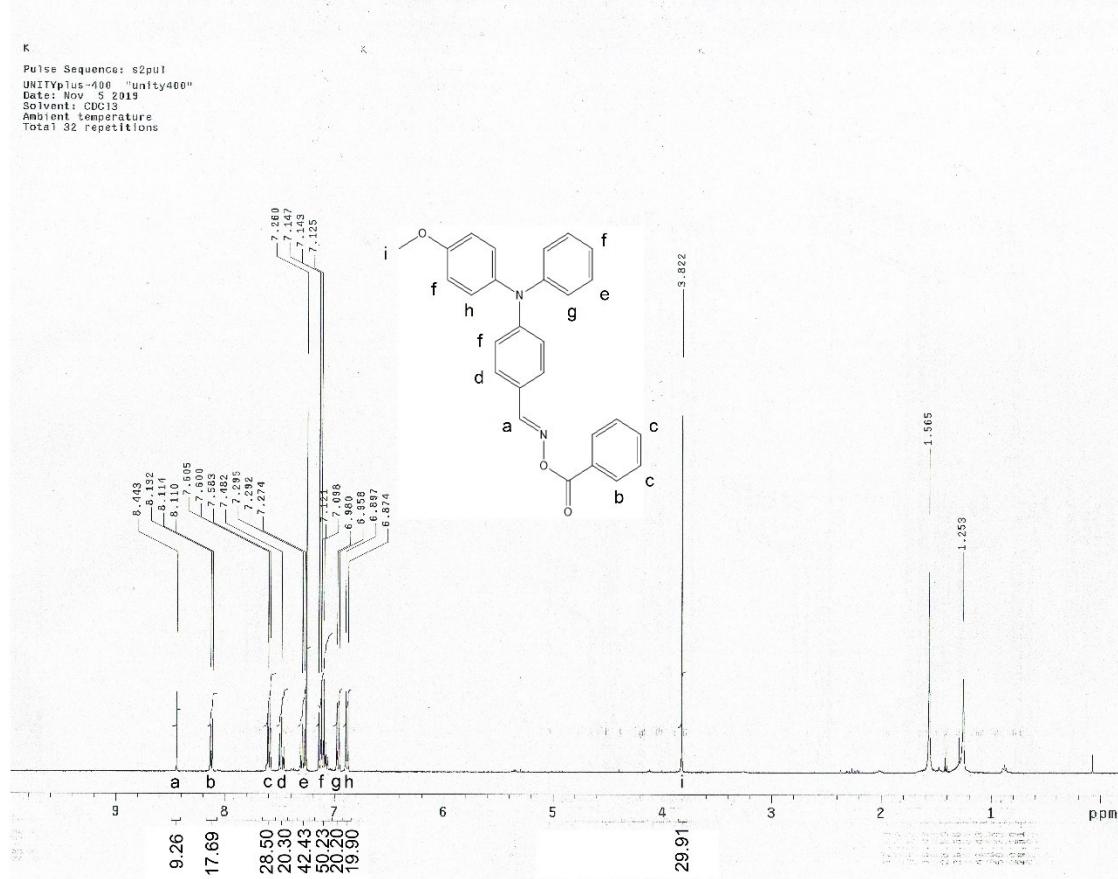


Fig. S2 ^1H NMR of the TP-2.

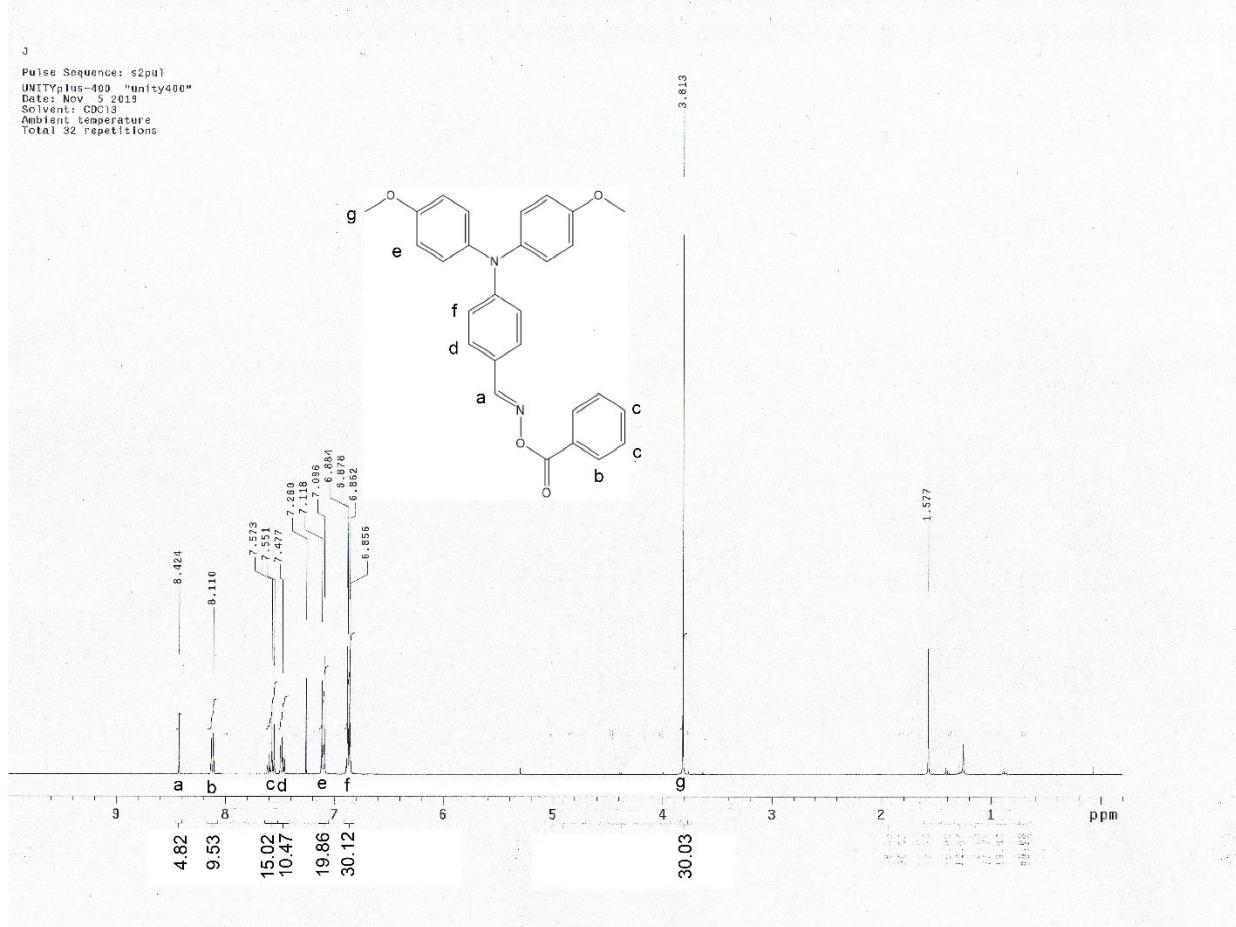


Fig. S3 ¹H NMR of the TP-3.

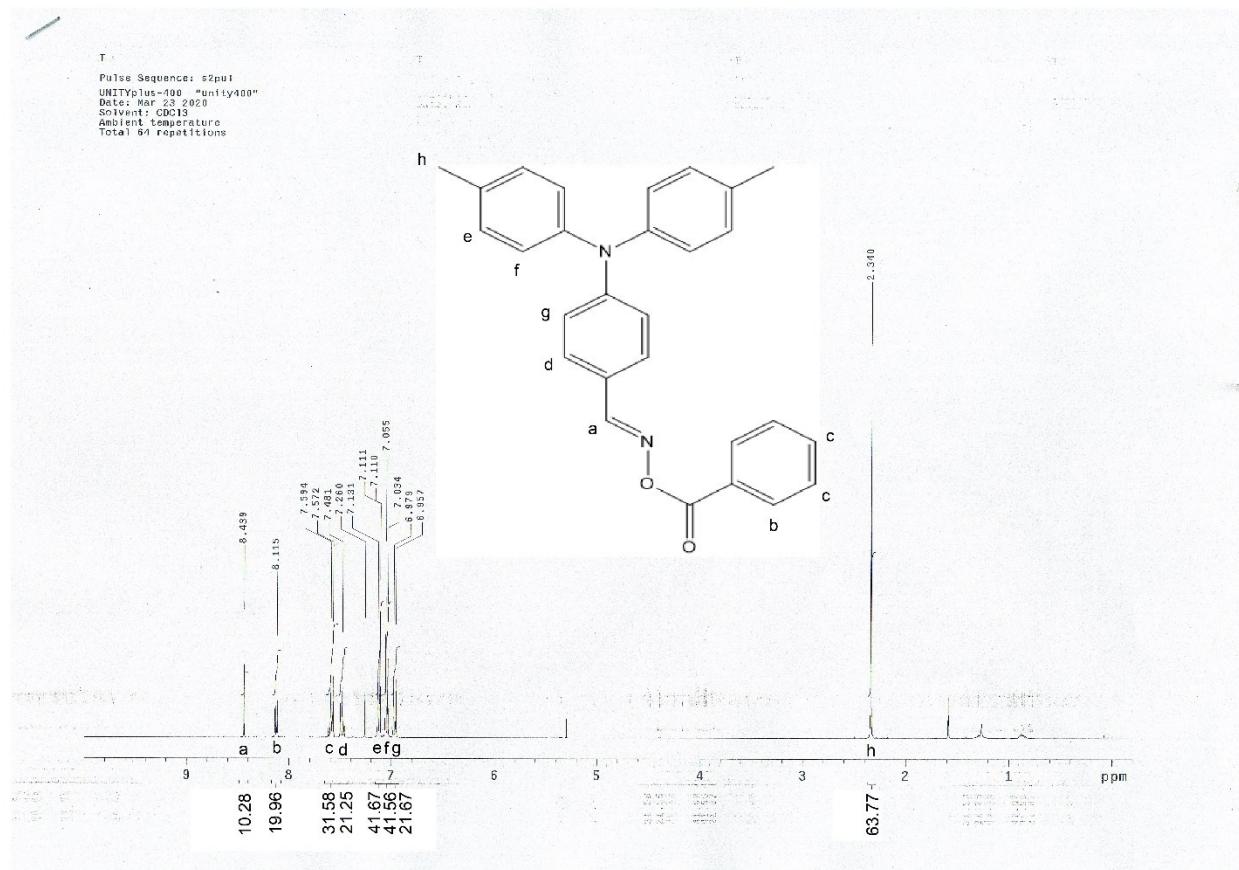


Fig. S4 ¹H NMR of the TP-4.

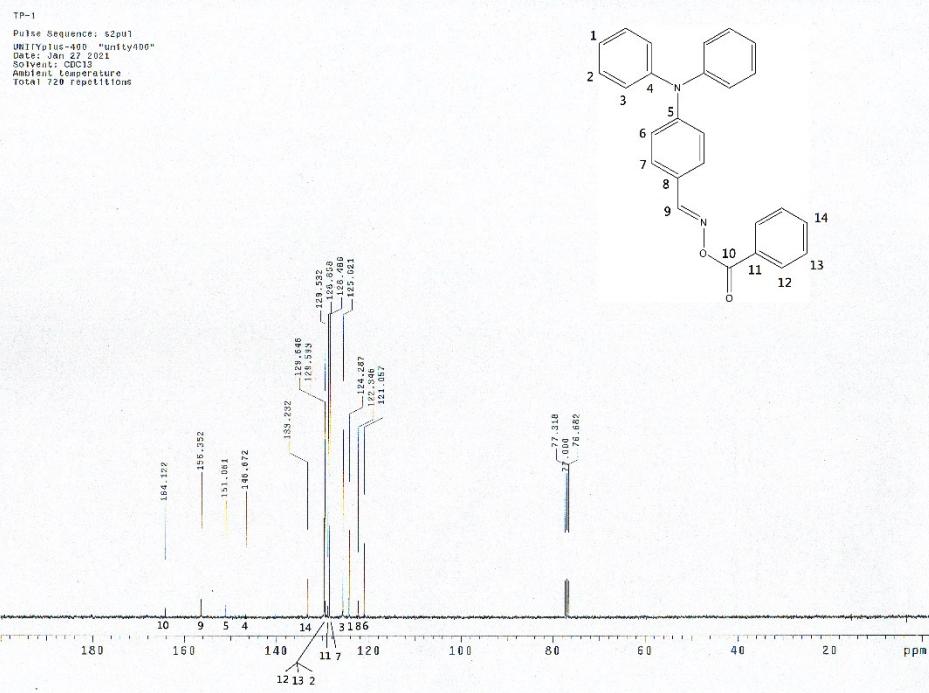


Fig. S5 ¹³C NMR of the TP-1.

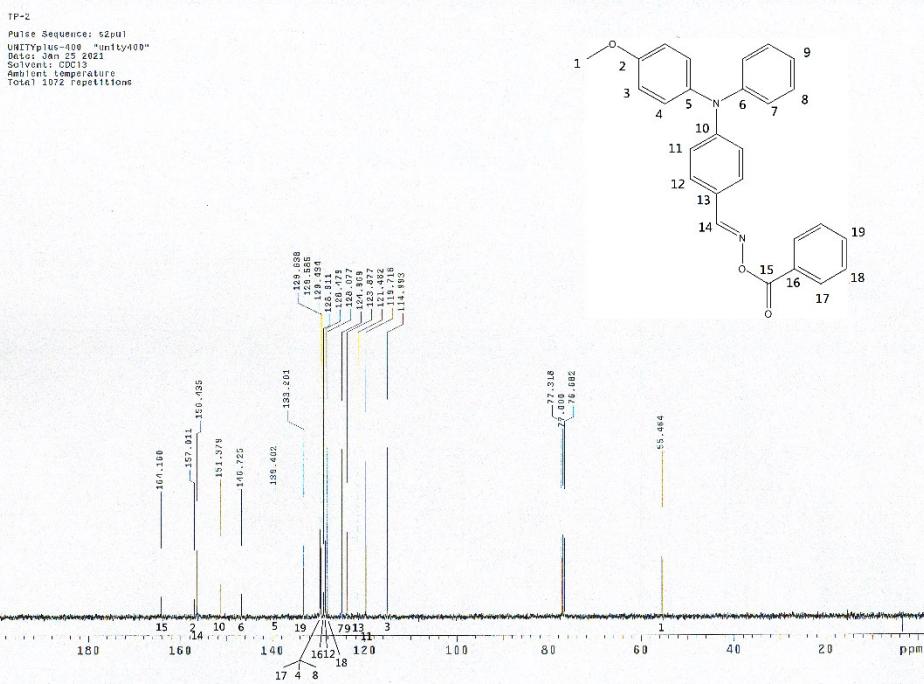


Fig. S6 ¹³C NMR of the TP-2.

TP-3
Pulse Sequence: zgppr1
UNITYPlus-400 "unify400"
Date: Jan 25 2021
Solvent: CDCl₃
Ambient temperature
Total 800 repetitions

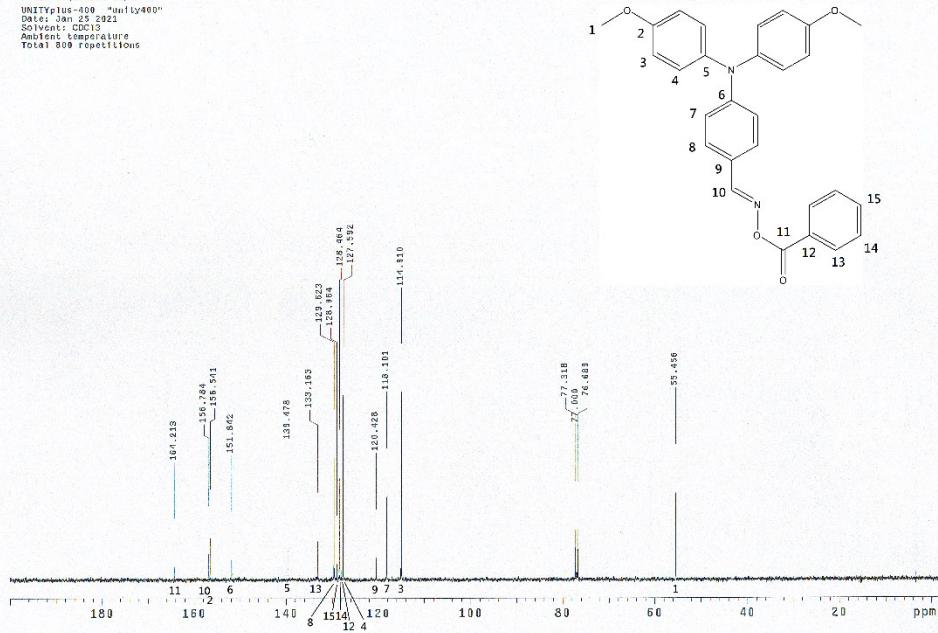


Fig. S7 ¹³C NMR of the TP-3.

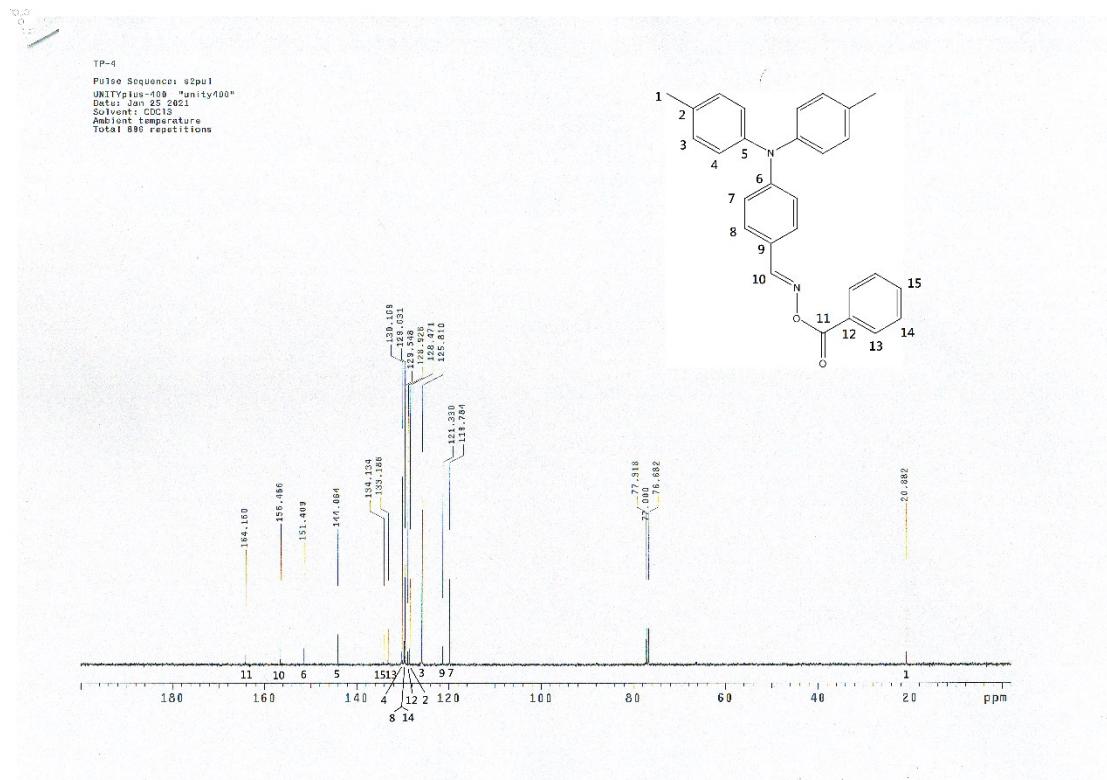
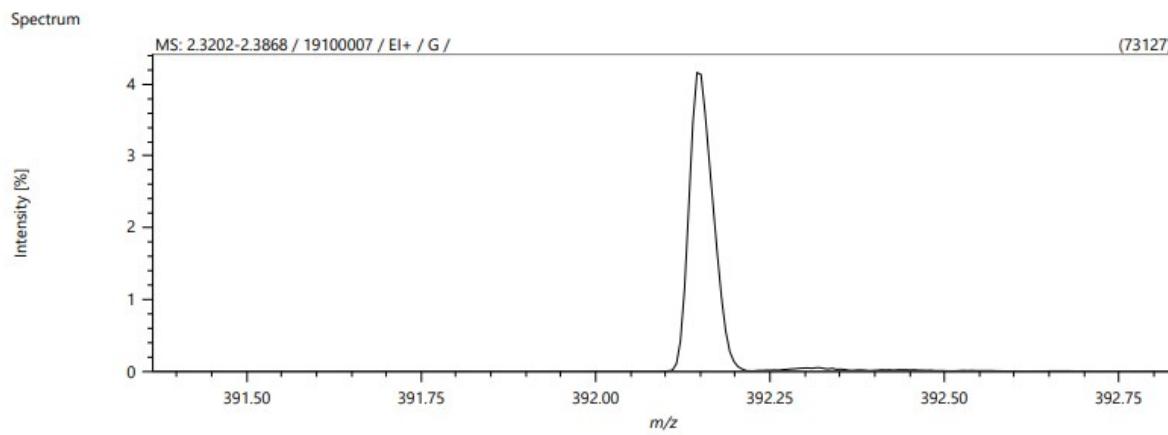
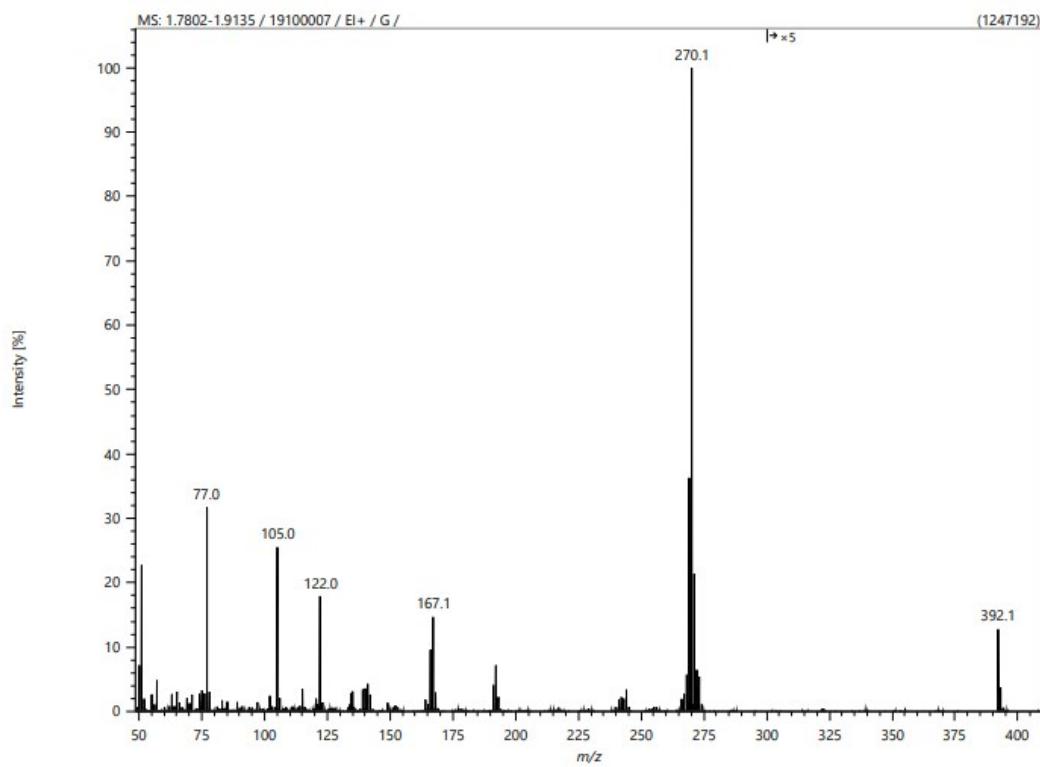


Fig. S8 ¹³C NMR of the TP-4.



Elemental Composition

Parameters

Tolerance: ± 10.00 ppm
 Electron: Odd/Even
 Charge: +1
 DBE: -1.5 - 200.0

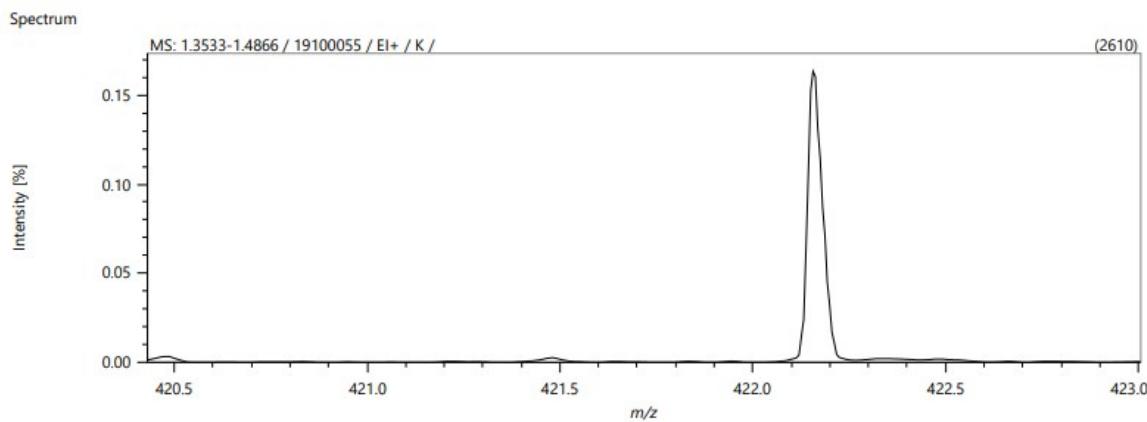
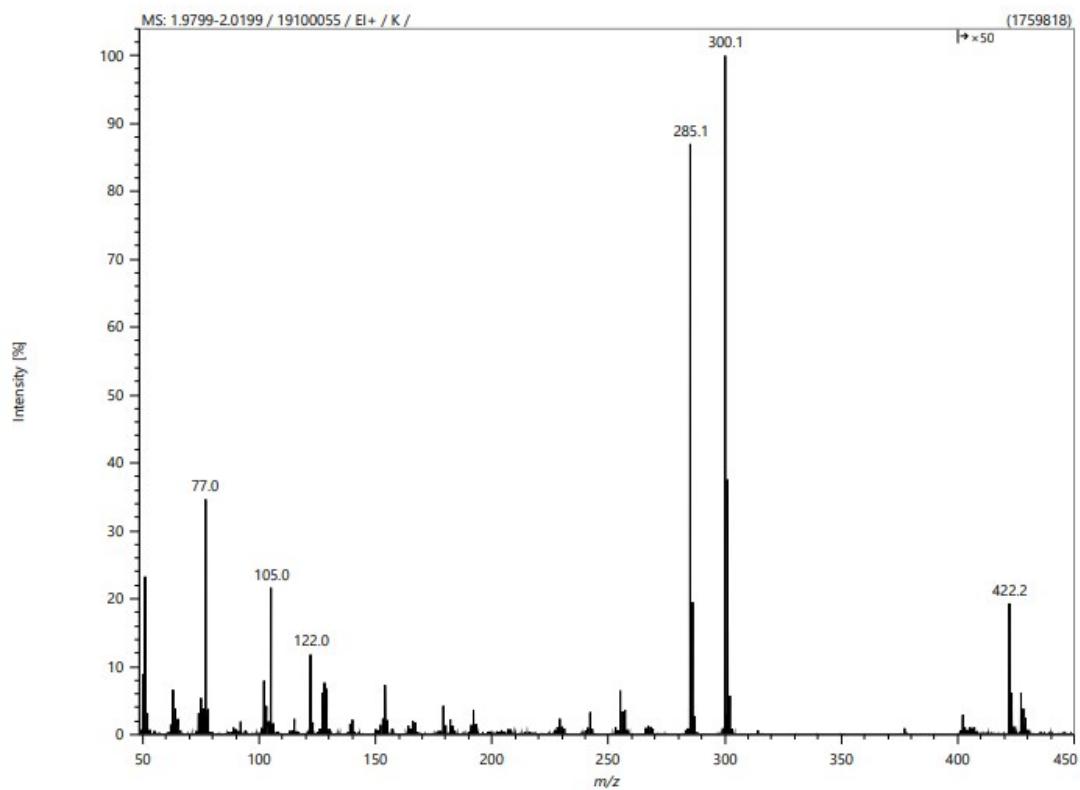
Elements Set 1:

Symbol	C	H	S	F	N	O	P
Min	0	0	0	0	0	0	0
Max	26	20	0	0	2	4	1

Results

Mass	Intensity	Formula	Calculated Mass	Mass Difference [mDa]	Mass Difference [ppm]	DBE
392.15178	73127.44	C ₂₆ H ₂₀ N ₂ O ₂	392.15193	-0.15	-0.37	18.0

Fig. S9 Electron impact mass spectrometry (EI-MS) of the TP-1.



Elemental Composition

Parameters
 Tolerance: ± 10.00 ppm
 Electron: Odd/Even
 Charge: +1
 DBE: -1.5 - 200.0

Elements Set 1:

Symbol	C	H	Br	P	N	O	Si
Min	0	0	0	0	0	0	0
Max	27	22	0	0	2	3	1

Results

Mass	Intensity	Formula	Calculated Mass	Mass Difference [mDa]	Mass Difference [ppm]	DBE
422.16226	2609.85	C ₂₇ H ₂₂ N ₂ O ₃	422.16249	-0.24	-0.56	18.0

Fig. S10 Electron impact mass spectrometry (EI-MS) of the **TP-2**.

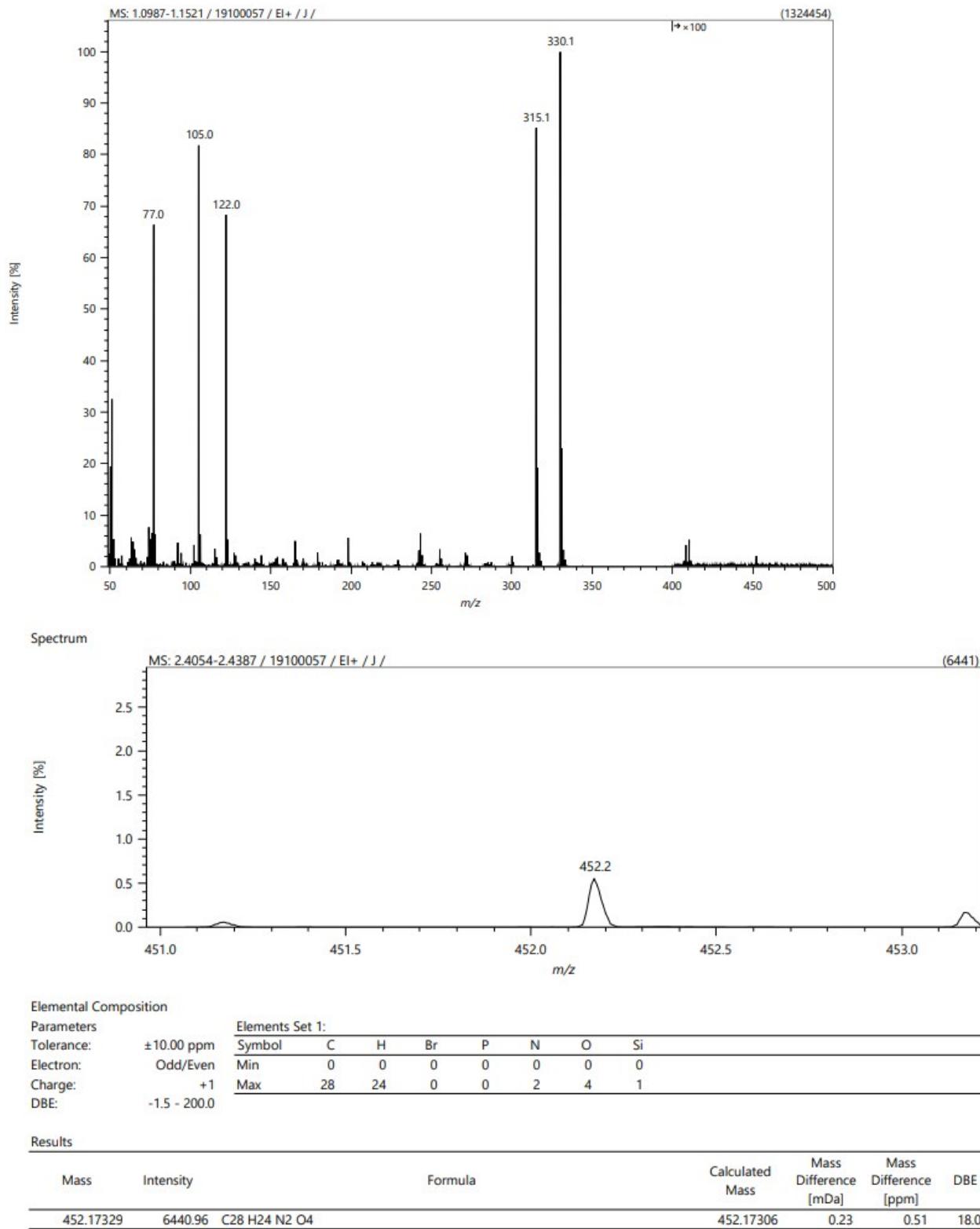


Fig. S11 Electron impact mass spectrometry (EI-MS) of the TP-3.

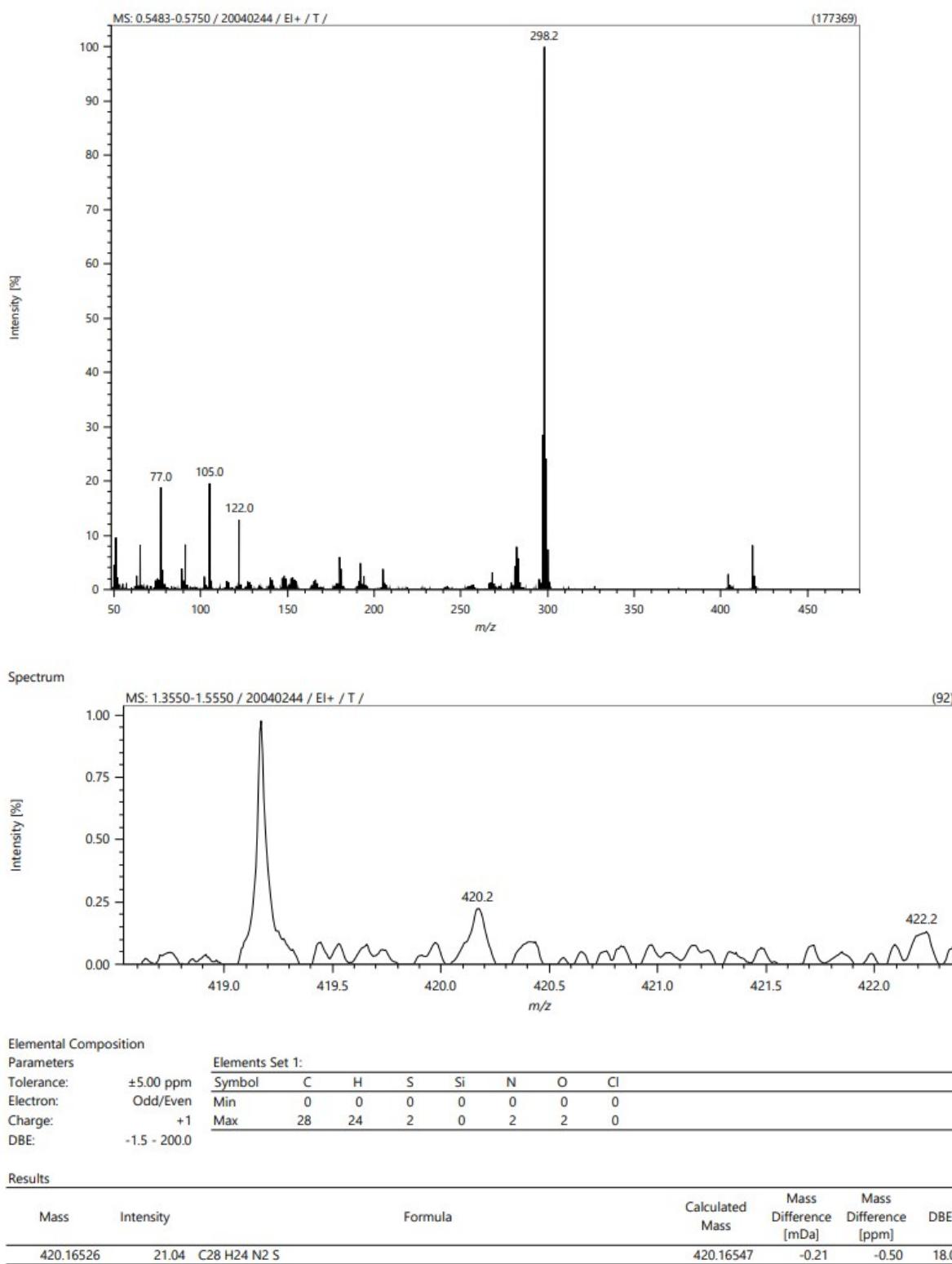


Fig. S12 Electron impact mass spectrometry (EI-MS) of the TP-4.

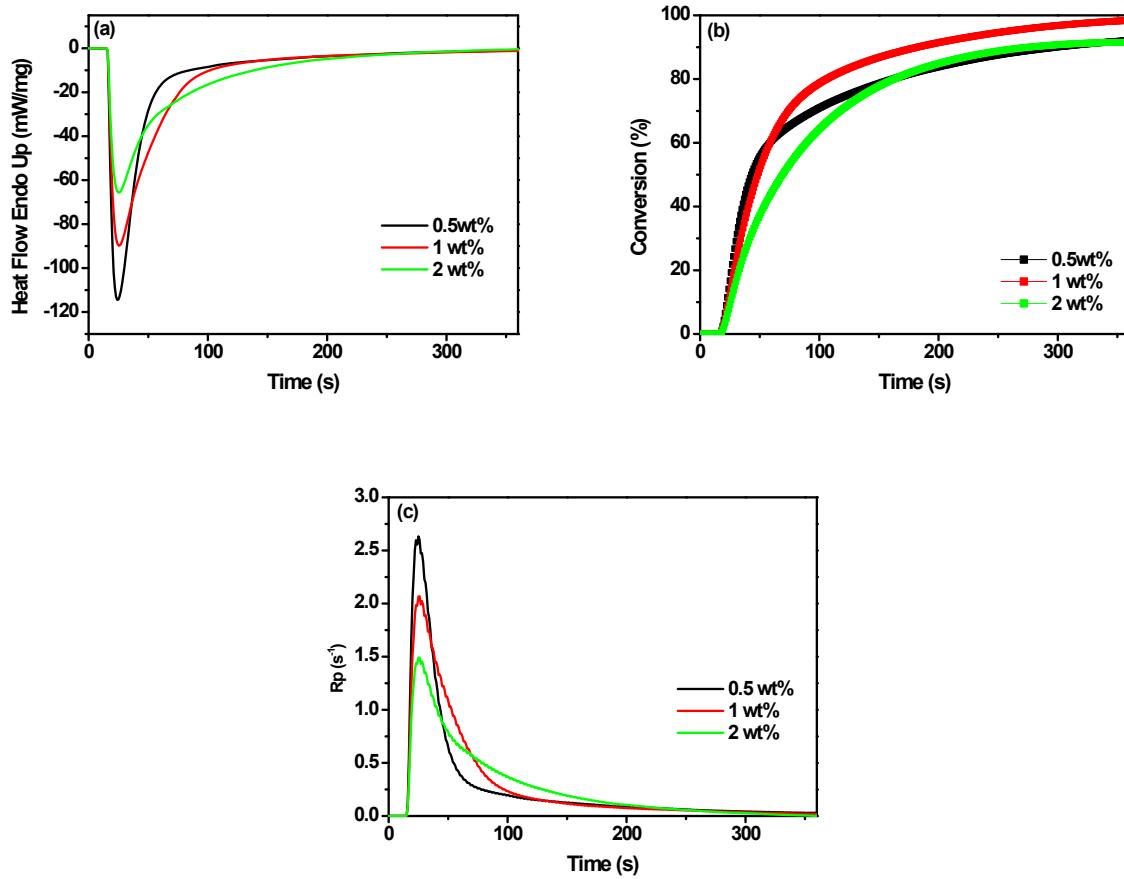


Fig. S13 (a) Heat flow versus time, (b) conversion versus time and (c) R_p versus time of TMPTA photopolymerization initiated by various weight ratios of **TP-1** (0.5, 1 wt% and 2 wt%) under UV light irradiation. The irradiation starts for $t = 24$ s.

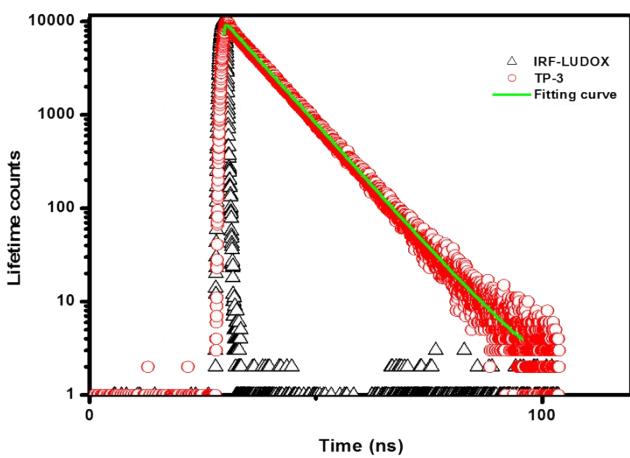


Fig. S14 Time correlated single photon counting of **TP-3** in CH_2Cl_2 , $\lambda_{\text{ex}} = 367 \text{ nm}$, $\lambda_{\text{em}} = 550 \text{ nm}$, mono-exponential curve fitting.

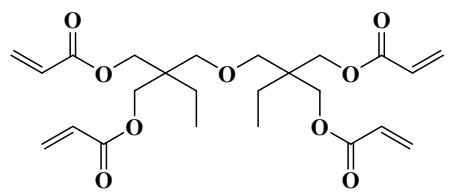


Fig. S15 Structure of the ditrimethylolpropane tetraacrylate (TA) monomer.

Table S1 Photo-DSC results derived from various weight ratios of **TP-1** (0.5, 1 wt% and 2 wt%)^a

TP-1 (wt%)	Final Conversion (%)	ΔH _t	H _{max}	R _{pmax}	T _{max}
		(kJ/mol) ^b	(mW/mg) ^c	(s ⁻¹) ^d	(s) ^e
0.5	94	81	114	2.63	24
1	99	85	90	2.07	26
2	92	79	66	1.49	25

^a. Measured with 180 mW cm⁻² of UV light (λ : 250-450 nm) for 6 min.

^b. ΔH_t is the totally reaction heat enthalpy within 6 min.

^c. H_{max}: maximum heat flow values.

^d. R_{pmax}: maximum rate of polymerization.

^e. T_{max}: time at maximum heat flow.

Table S2 Fluorescence lifetime of **TP-1–4** in CH₂Cl₂.

First singlet state lifetime (ns)			
TP-1	TP-2	TP-3	TP-4
0.85	0.8	8	1.6