

Synthesis of $\text{CoS}_2/\text{SnO}_2@\text{MoS}_2$ nanocubes heterostructures for achieving enhanced electrocatalytic hydrogen evolution in acidic media

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Materials

Tin chloride pentahydrate ($\text{SnCl}_4 \cdot 5\text{H}_2\text{O}$, AR), Cobalt chloride hexahydrate ($\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$, AR), sodium citrate dihydrate ($\text{C}_6\text{H}_{15}\text{Na}_3\text{O}_7 \cdot 2\text{H}_2\text{O}$, ≥99%), ammonium molybdate tetrahydrate ($(\text{NH}_4)_6\text{Mo}_7\text{O}_{24} \cdot 4\text{H}_2\text{O}$, AR) and ammonium sulfide ($(\text{NH}_4)_2\text{S}$, AR) were purchased from Aladdin Chemistry Co., Ltd. Ethanol ($\text{CH}_3\text{CH}_2\text{OH}$, ≥99.7%), sodium hydroxide (NaOH, AR), aqueous ammonia solution ($\text{NH}_3 \cdot \text{H}_2\text{O}$, 25%~28%) and sulfuric acid (H_2SO_4 , 95%~98%) were acquired from Tianjin Zhiyuan Chemical Reagent Co., Ltd. Nafion solution (5 wt%) and commercial Pt/C (20 wt%) was acquired from Sigma-Aldrich and Shanghai Hesen Electric Co., Ltd, respectively. All reagents were used directly without any further purification. Water deionized with a Millipore system was used in this work.

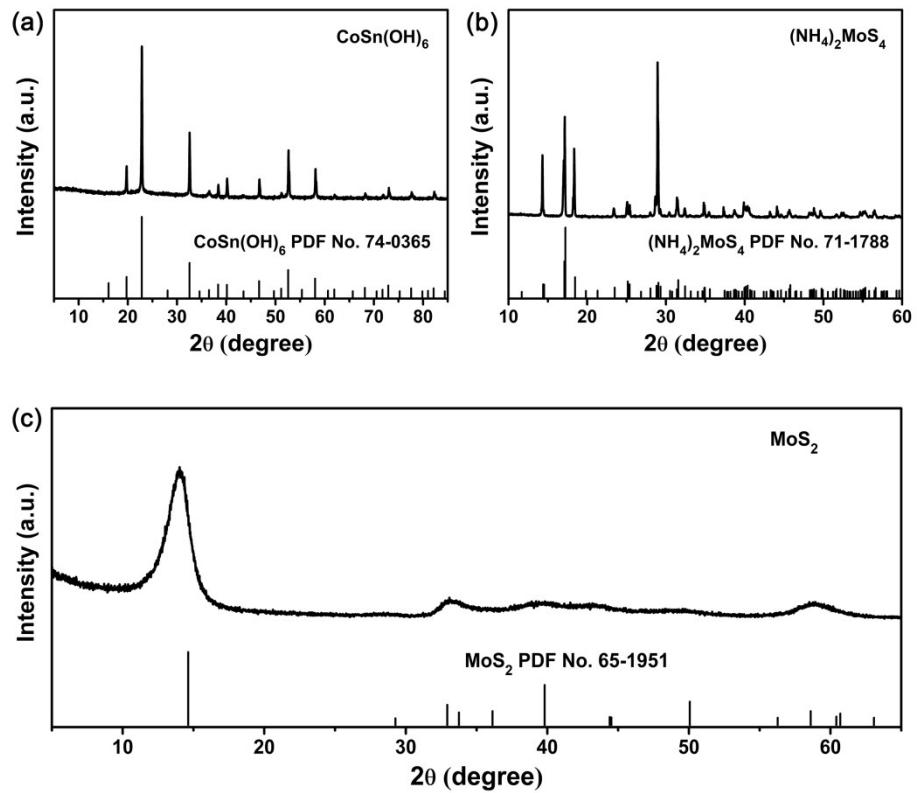


Fig. S1 XRD patterns of (a) $\text{CoSn}(\text{OH})_6$ nanocubes; (b) $(\text{NH}_4)_2\text{MoS}_4$ particles and (c) MoS_2 particles.

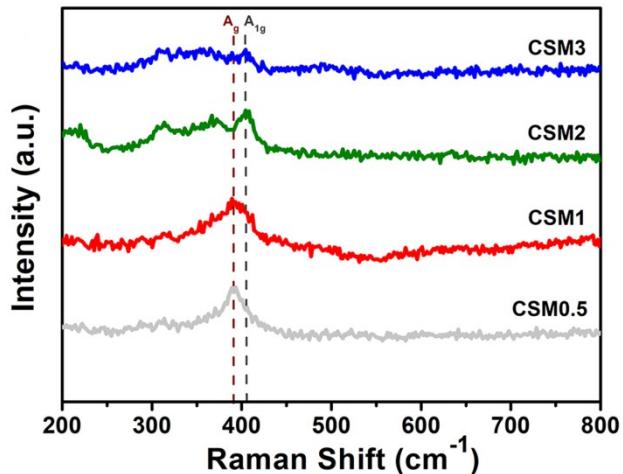


Fig. S2 Raman spectra of corresponding CSM_y ($y = 0.5, 1, 2$ and 3) nanocubes.

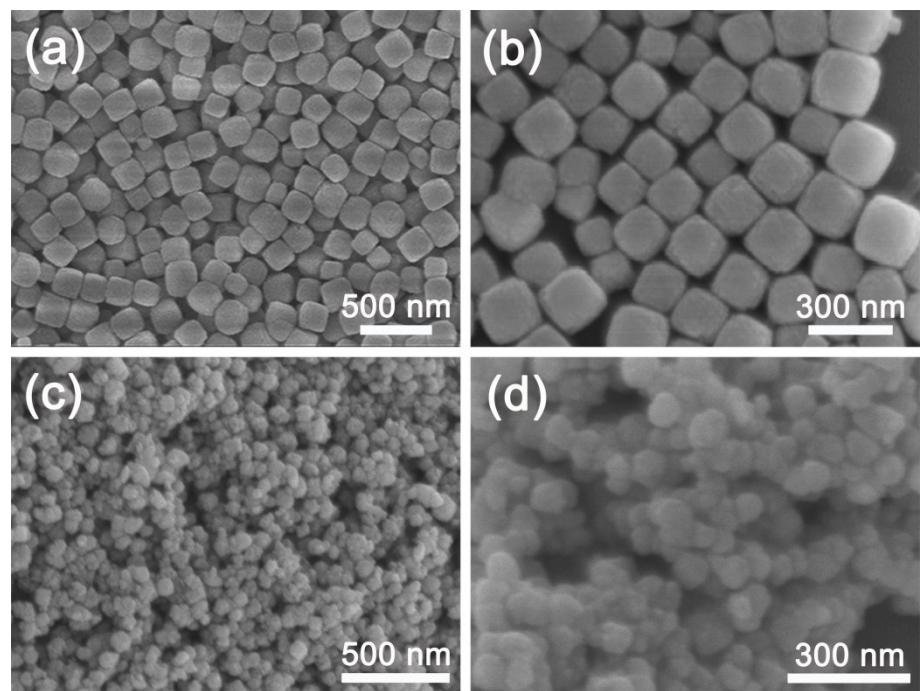


Fig. S3 SEM images of (a) and (b) $\text{CoSn}(\text{OH})_6$ nanocubes; (c) and (d) CSM0.5 nanocubes.

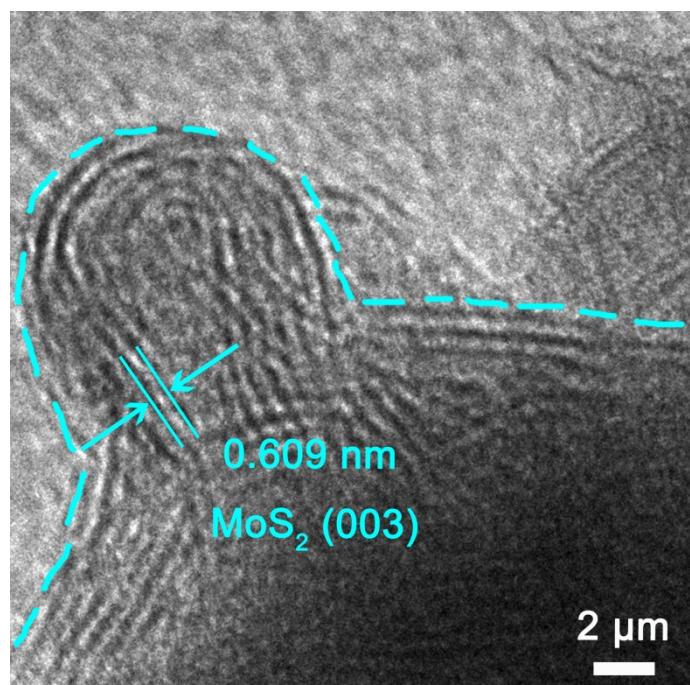


Fig. S4 HRTEM image of CSM1 nanocubes.

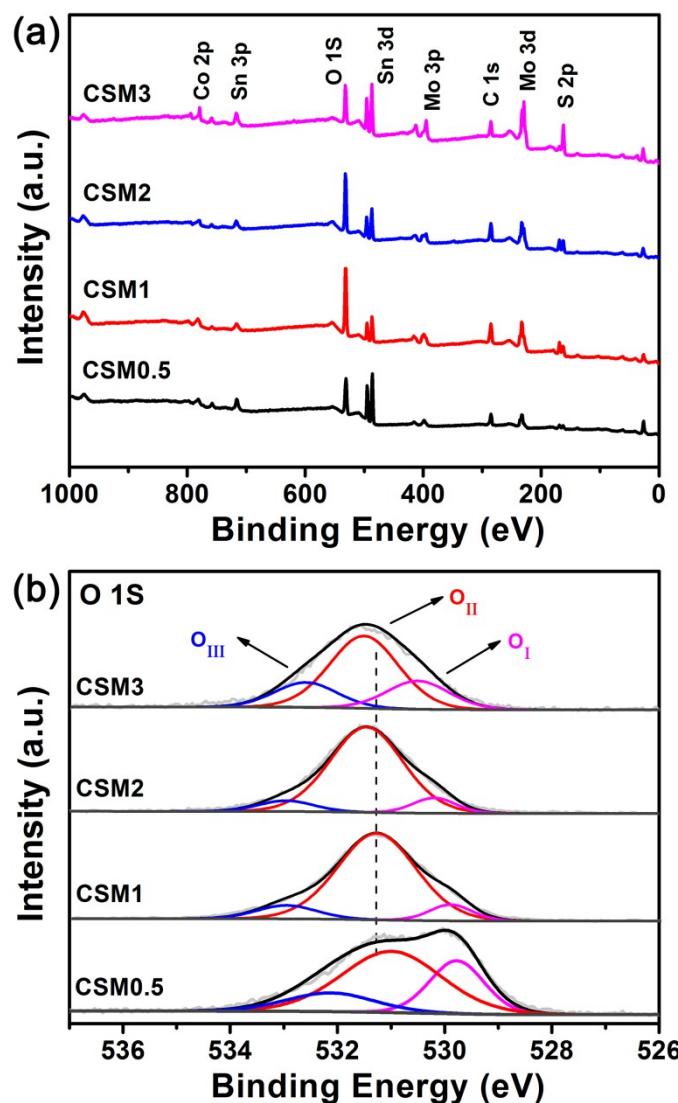


Fig. S5 XPS spectra of the CSM y ($y = 0.5, 1, 2$ and 3) nanocubes (a) survey and (b) O 1s.

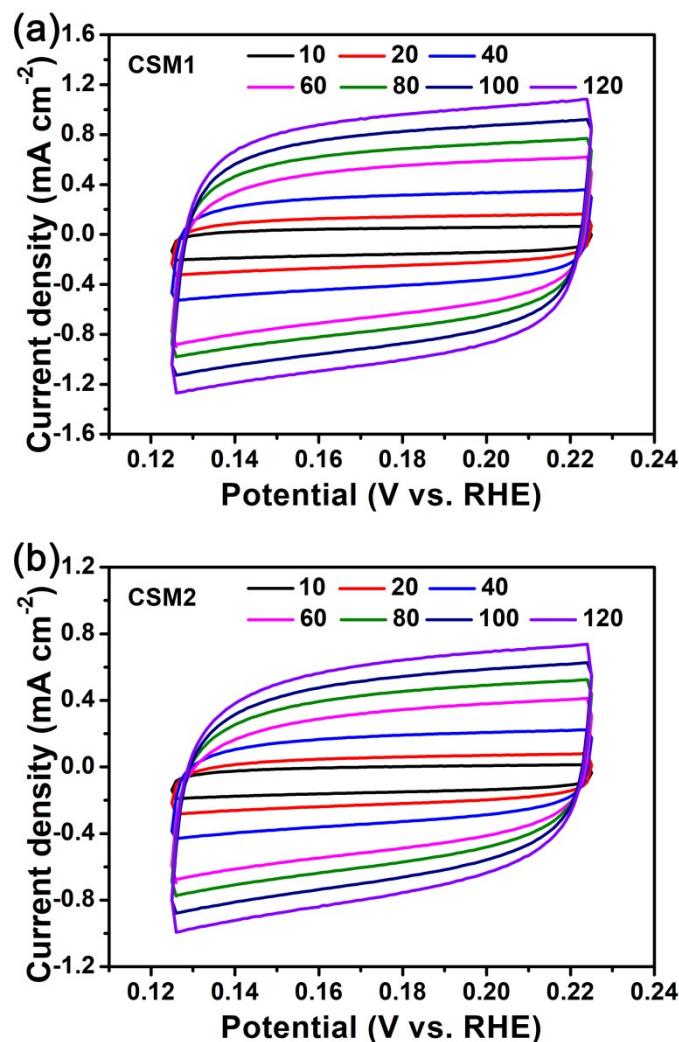


Fig. S6 CV curves of (a) CSM1 and (b) CSM2 modified electrodes in the double layer region at scan rate of 10, 20, 40, 60, 80, 100 and 120 mV s^{-1} in 0.5 M H_2SO_4 .

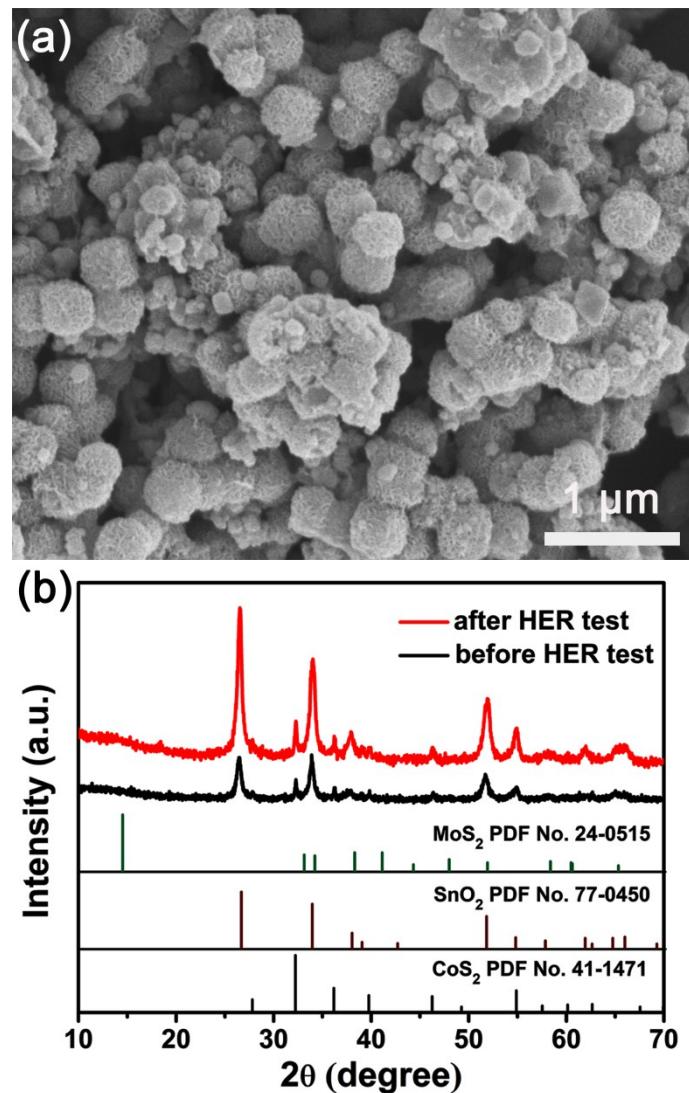


Fig. S7 (a) SEM images of CMS1 nanocubes after HER stability test. (b) The XRD spectra of CMS1 nanocubes before and after OER stability test in 0.5 M H₂SO₄.

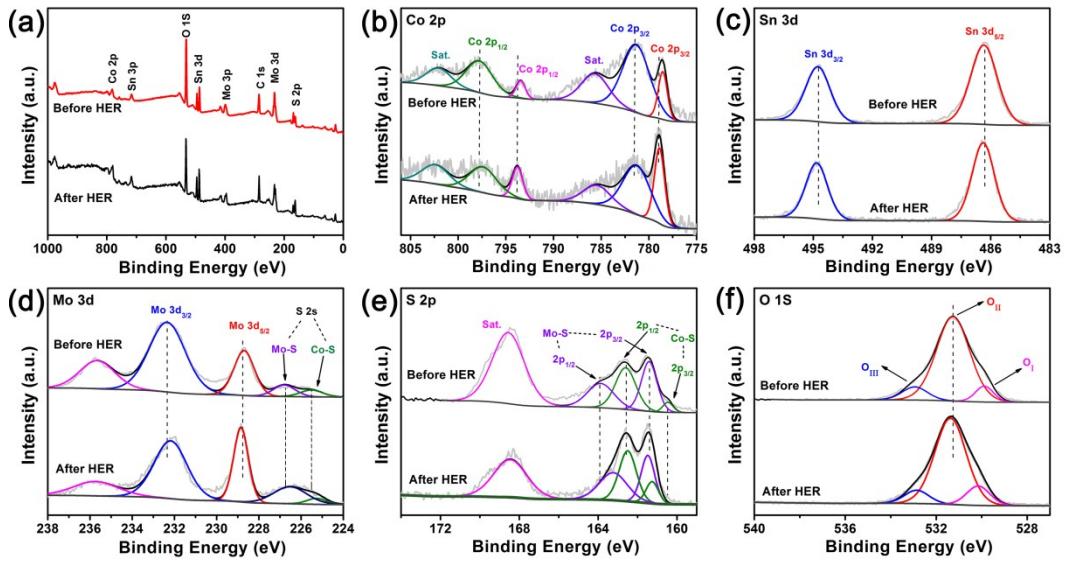


Fig. S8 XPS spectra of CMS1 nanocubes before and after OER stability test in 0.5 M H₂SO₄ (a) survey, (b) Co 2p, (c) Sn 3d, (d) Mo 3d, (e) S 2p and (f) O 1s.

Table S1 Electrochemical performance of molybdenum sulfide-based electrocatalysts in 0.5 M H₂SO₄

| Catalyst | Current density (j , mA cm ⁻²) | η at corresponding j (mV) | Tafel slope (mV dec ⁻¹) | References (Year) |
|--|--|-------------------------------------|--|----------------------|
| CSM1 | 10 | 196 | 69 | This work |
| m-CNMS | 10 | 215 | 50.2 | [1] (2019) |
| VMS2 | 10 | 194 | 59 | [2] (2019) |
| CoS ₂ @MoS ₂ | 10 | 290 | 85.9 | [3] (2017) |
| Cu-MoS ₂ | 10 | 211 | 86 | [4] (2017) |
| Ni-MoS ₂ | 11 | 300 | 89 | [5] (2017) |
| Co ₃ S ₄ @MoS ₂ | 10 | 210 | 88 | [6] (2017) |
| Co _x Mo _{1-x} S ₂ | 10 | 357 | 120 | [7] (2017) |
| 1.0Ni-MoS ₂ | 10 | 173 | 69 | [8] (2019) |
| mPF-MoS ₂ | 10 | 210 | 90 | [9] (2017) |
| MCM@MoS ₂ –Ni | 10 | 161 | 81 | [10] (2018) |
| VS ₂ @MoS ₂ | 10 | 177 | 54.9 | [11] (2017) |

References

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