

Combined DFT and Geometrical-Topological Analysis of Li-ion conductivity in Complex Hydrides

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Supporting information

Crystal structure of investigated complex hydrides

Sample	Space Group	Lattice constants (Å)	Ref.
LiBH₄	<i>Pnma</i>	$a = 7.178 \text{ \AA}$ $b = 6.803 \text{ \AA}$ $c = 4.436 \text{ \AA}$	1
Li₂NH	<i>Fm3m</i>	$a = 5.074$	2
LiNH₂	<i>I-4</i>	$a = 5.079$ $c = 10.113$	3
Li₂BH₄NH₂	<i>R-3</i>	$a = 14.498$ $c = 9.248$	4
Li₄BH₄(NH₂)₃	<i>I2₁3</i>	$a = 10.673$	4
Li₅(BH₄)₃NH	<i>Pnma</i>	$a = 10.203,$ $b = 11.501$ $c = 7.047$	5

Table S1. Structural details of all considered samples.

Topological analysis

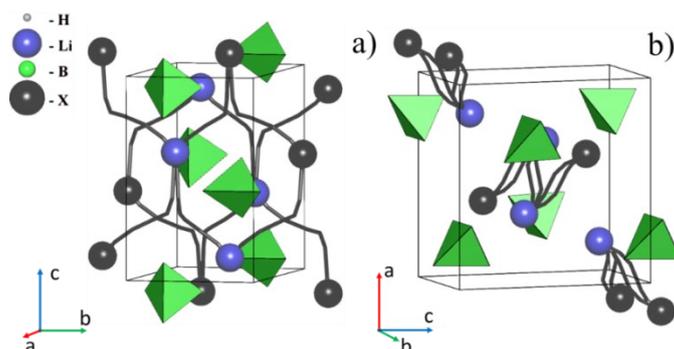


Figure S1. The lowest energy Li^+ migration pathways in the LiBH_4 structure. The migration energies of the pathways: a) 0.242; b) 0.298 eV. Hereafter the points labeled as X correspond to equilibrium Li positions and are linked by the NEB trajectories.

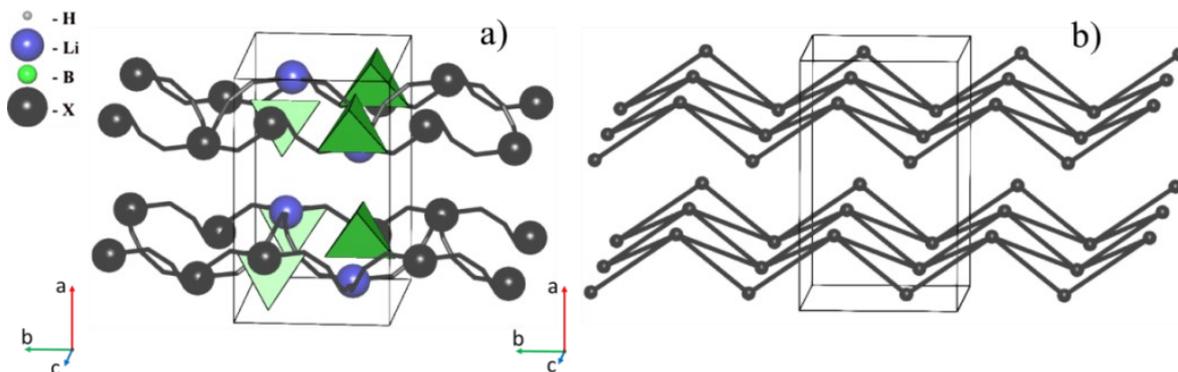


Figure S2. a) The lowest energy two-periodic Li^+ migration map in the LiBH_4 structure, b) the simplified migration map of the sql topology.

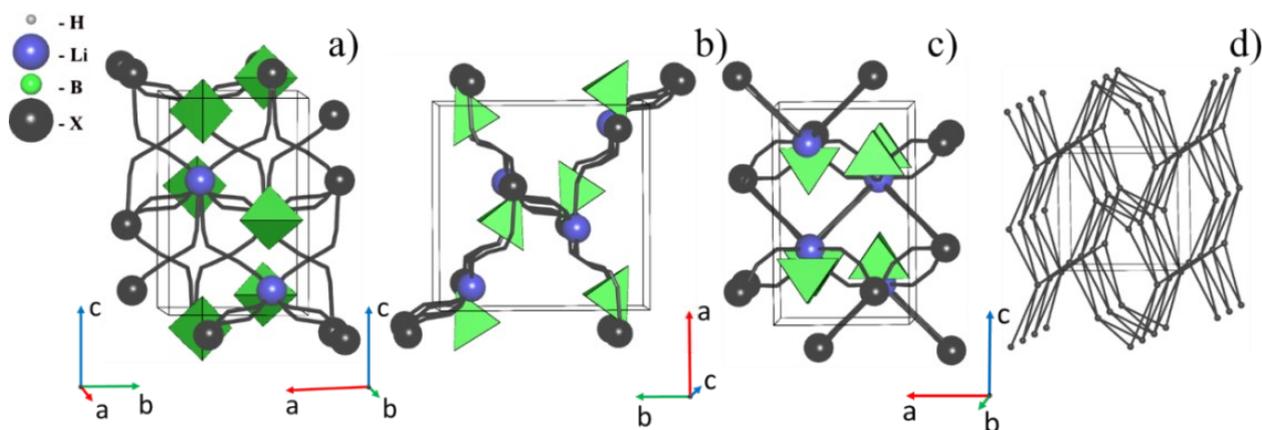


Figure S3. a)-c) The lowest energy three-periodic Li^+ migration map in the LiBH_4 structure, d) the simplified migration map of the acs topology.

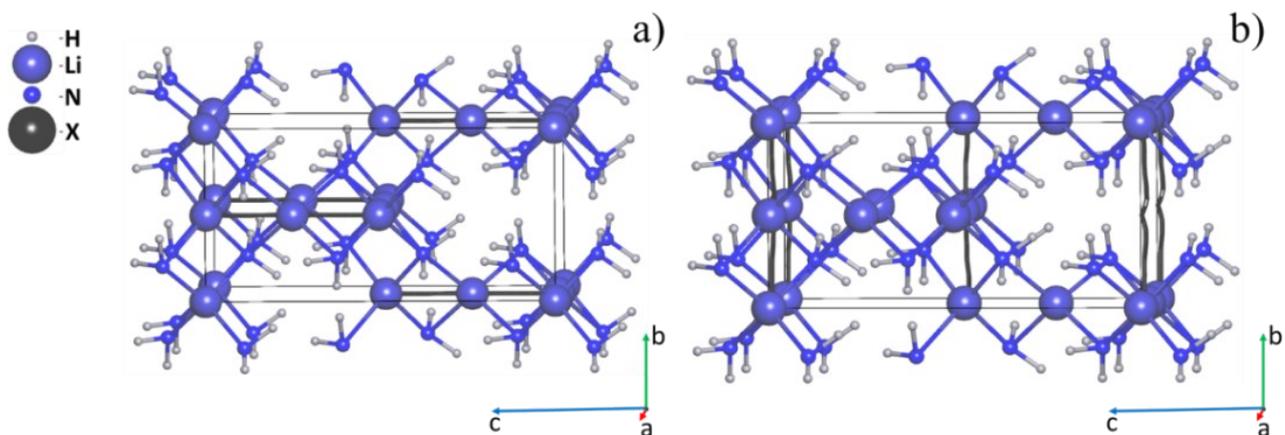


Figure S4. The lowest energy Li^+ migration pathways in the structure LiNH_2 . The migration energies of the pathways: a) 0.179; b) 0.461 eV.

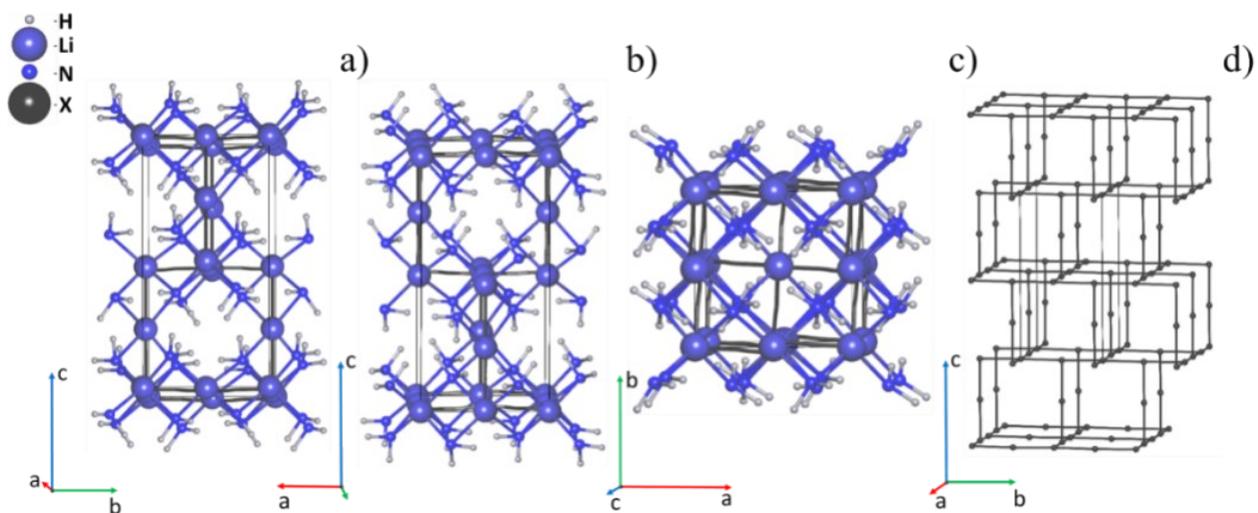


Figure S5. a)-c) The lowest energy three-periodic Li^+ migration map in the LiNH_2 structure, d) the simplified migration map of the tfa topology.

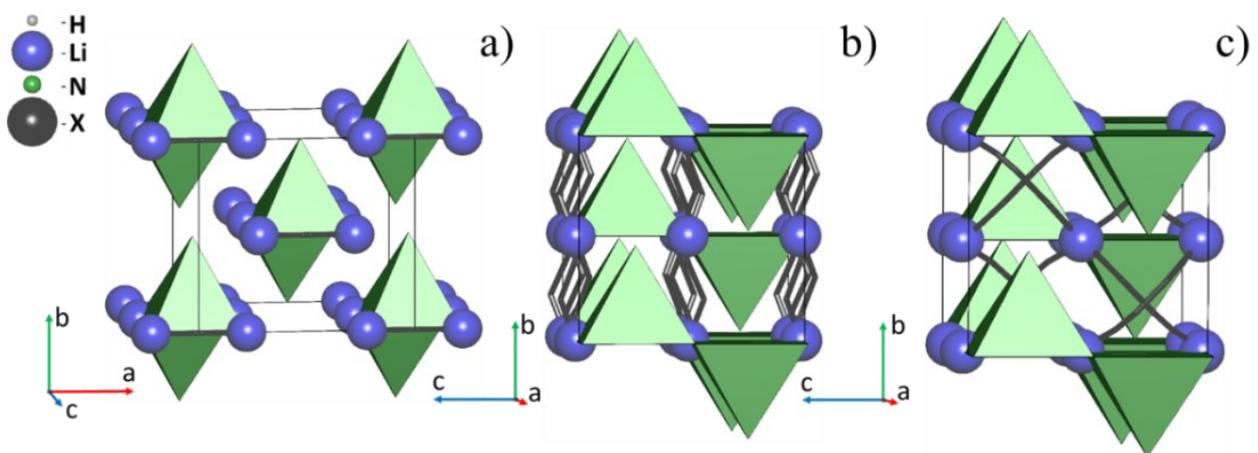


Figure S6. The lowest energy Li^+ migration pathways in the Li_2NH structure. The migration energies of the pathways: a) 0.007; b) 0.308; c) 0.321 eV.

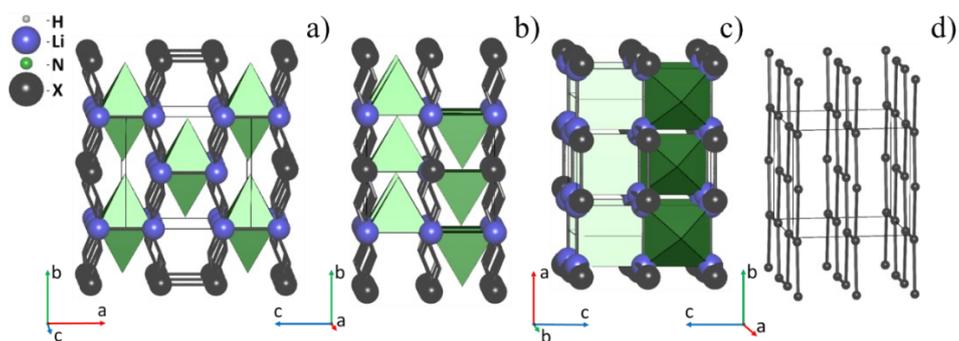


Figure S7. a)-c) The lowest energy two-periodic Li^+ migration map in the Li_2NH structure, d) the simplified migration map of the hcb topology.

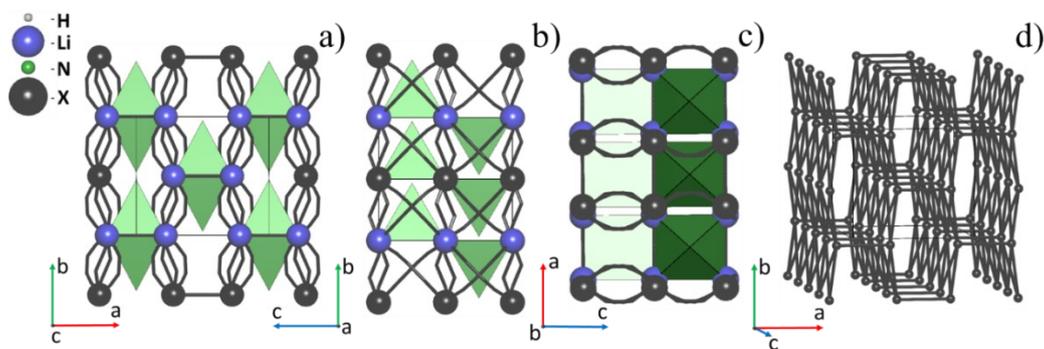


Figure S8. a)-c) The lowest energy three-periodic Li^+ migration map in the Li_2NH structure, d) the simplified migration map of the sqc2-7-Cmmm topology.

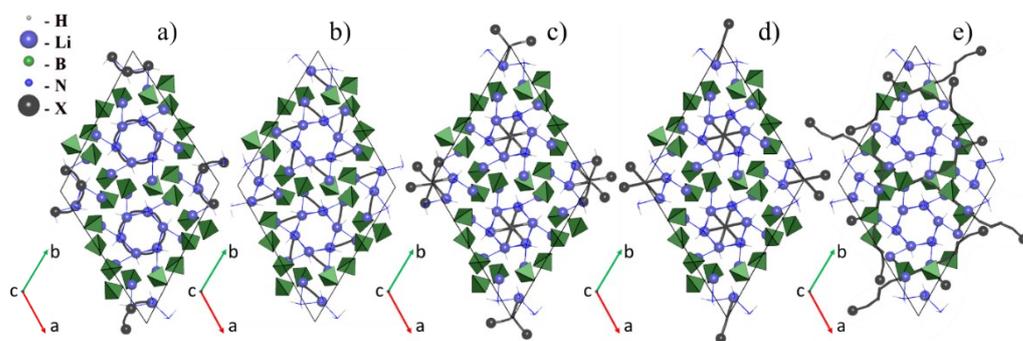


Figure S9. The lowest energy Li^+ migration pathways in the $\text{Li}_2\text{BH}_4\text{NH}_2$ structure. The migration energies of the pathways: a) 0.176; b) 0.365; c) 0.445; d) 0.481; e) 0.525 eV.

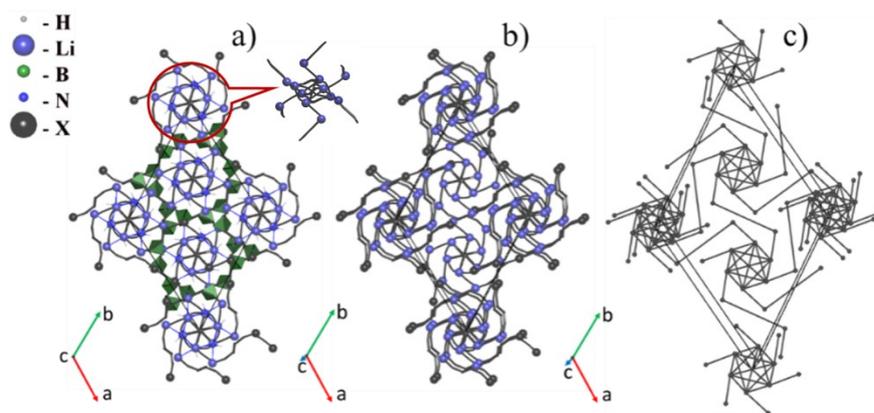


Figure S10. a), b) The lowest energy three-periodic 2-fold interpenetrated Li^+ migration map in the $\text{Li}_2\text{BH}_4\text{NH}_2$ structure, c) the simplified migration map.

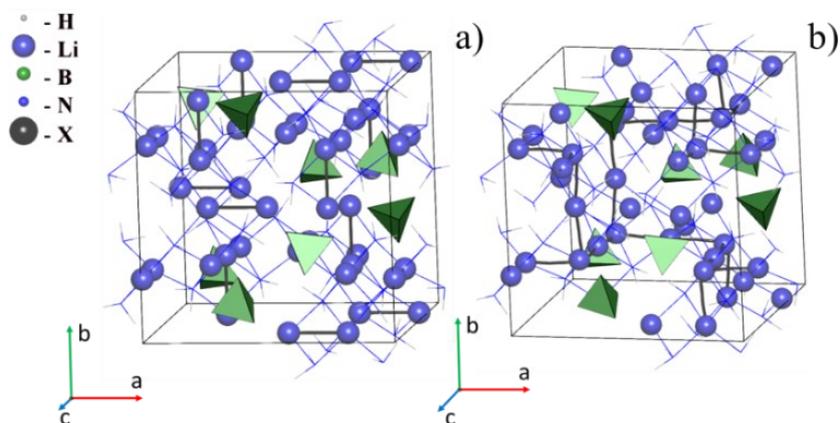


Figure S11. The lowest energy Li^+ migration pathways in the $\text{Li}_4\text{BH}_4(\text{NH}_2)_3$ structure. The migration energy of the pathways: a) 0.212; b) 0.296 eV.

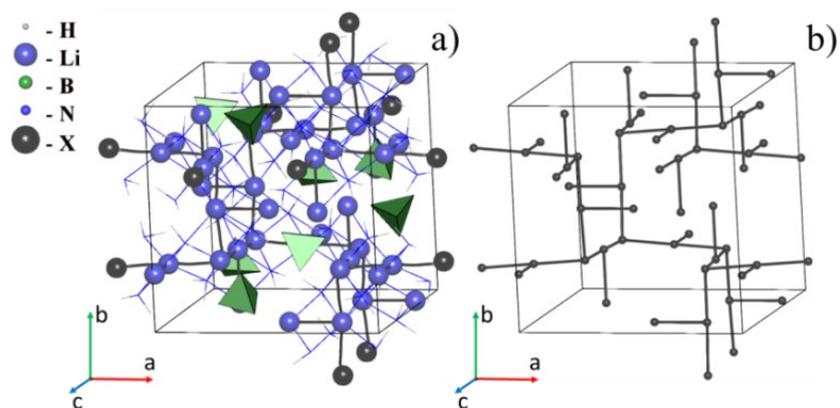


Figure S12. a) The lowest energy three-periodic Li^+ migration map in the structure $\text{Li}_4\text{BH}_4(\text{NH}_2)_3$, b) the simplified migration map of the srs topology.

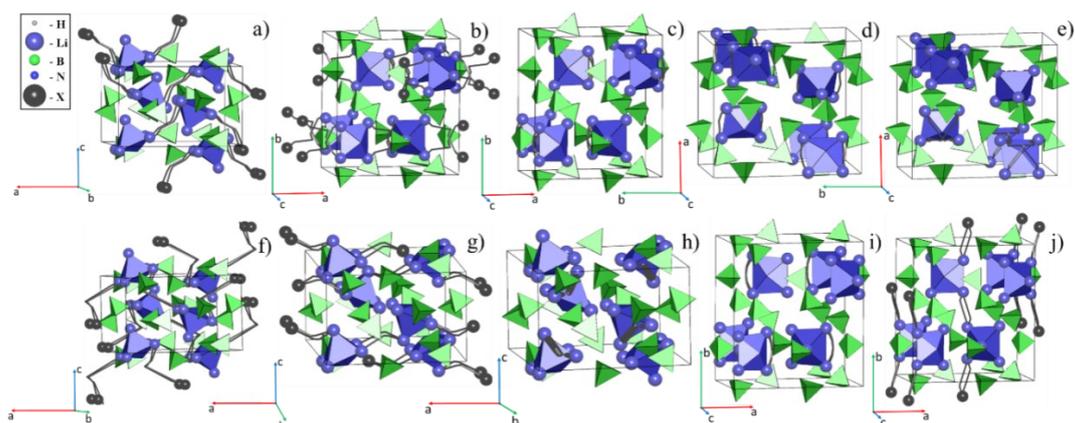


Figure S13. The lowest energy Li^+ migration pathways in the $\text{Li}_5(\text{BH}_4)_3\text{NH}$ structure. The migration energies of the pathways: a) 0.237; b) 0.282; c) 0.340; d) 0.397; e) 0.401; f) 0.479; g) 0.536; h) 0.541; i) 0.547; j) 0.559 eV.

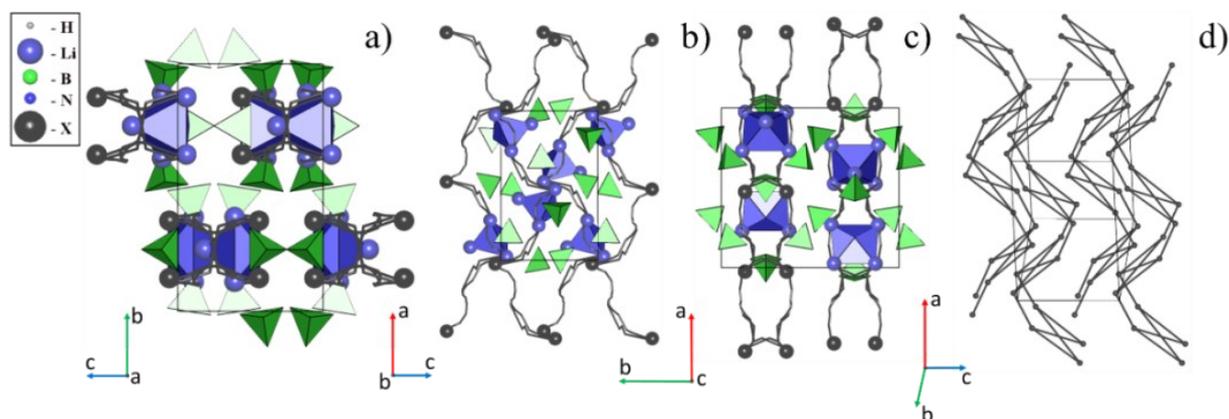


Figure S14. a)-c) The lowest energy one-periodic Li^+ migration map in the $\text{Li}_5(\text{BH}_4)_3\text{NH}$ structure, d) the simplified migration map of the $6^3(0,2)$ topology.

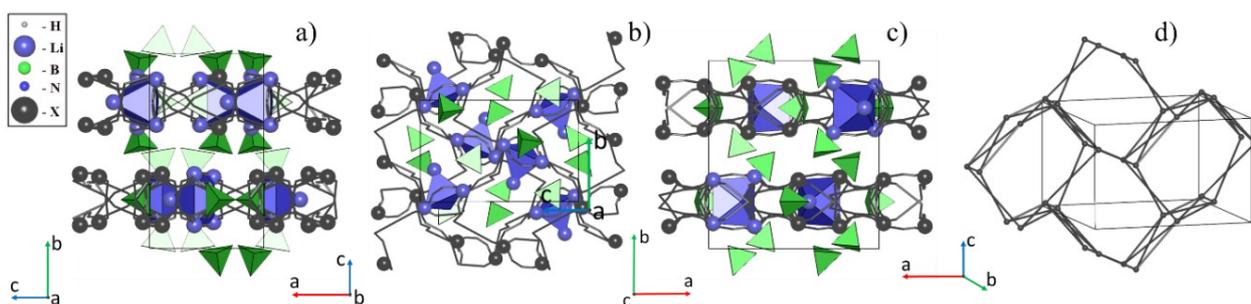


Figure S15. a)-c) The lowest energy two-periodic Li^+ migration map in the $\text{Li}_5(\text{BH}_4)_3\text{NH}$ structure, d) the simplified migration map with an unknown topology.

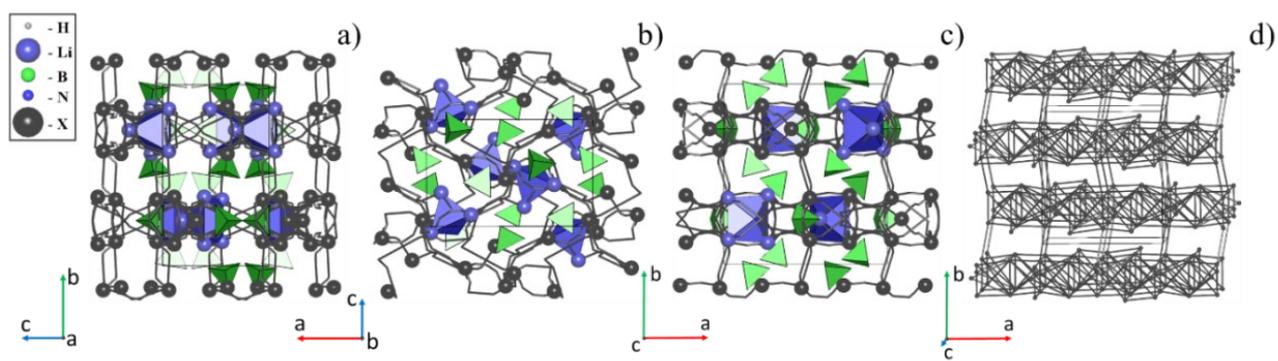


Figure S16. a)-c) The lowest energy three-periodic Li^+ migration map in the $\text{Li}_5(\text{BH}_4)_3\text{NH}$ structure, d) the simplified migration map with an unknown topology.

Statistical analysis of literature data of Li-ion conductivity in LiBH₄

A statistical analysis was performed in order to obtain the average value of activation energy (E_A), $\ln\sigma_0$ and Li-ion conductivity at room temperature for LiBH₄, using literature data.⁶⁻¹⁴ Only data related to the orthorhombic phase stable at room temperature have been considered.

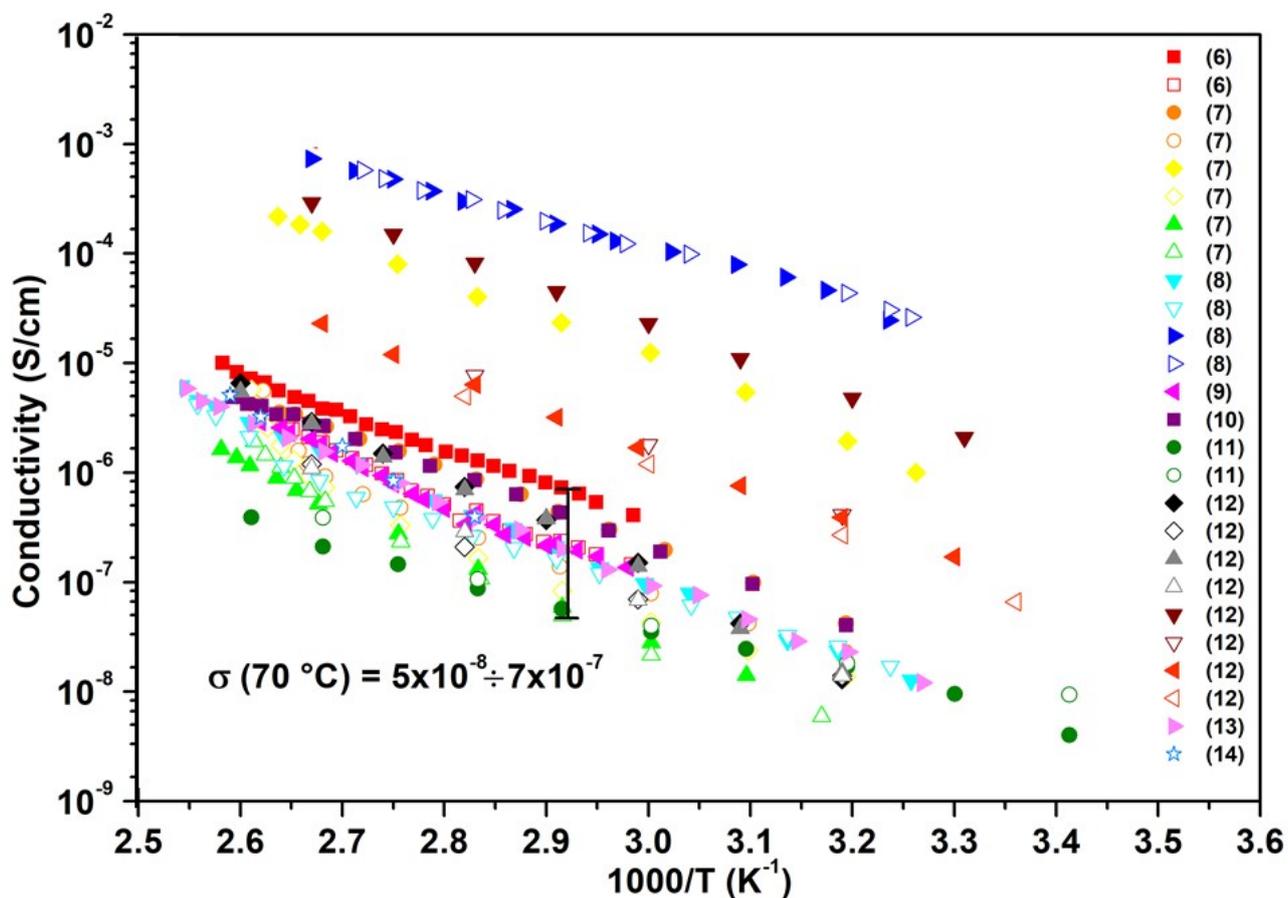


Figure S17. Li-ion conductivity data for orthorhombic LiBH₄ reported in literature. Full and empty symbols refer to the heating and cooling temperature-dependent EIS ramp, respectively. The error bar for the Li-ion conductivity at 70 °C is indicated as an example.

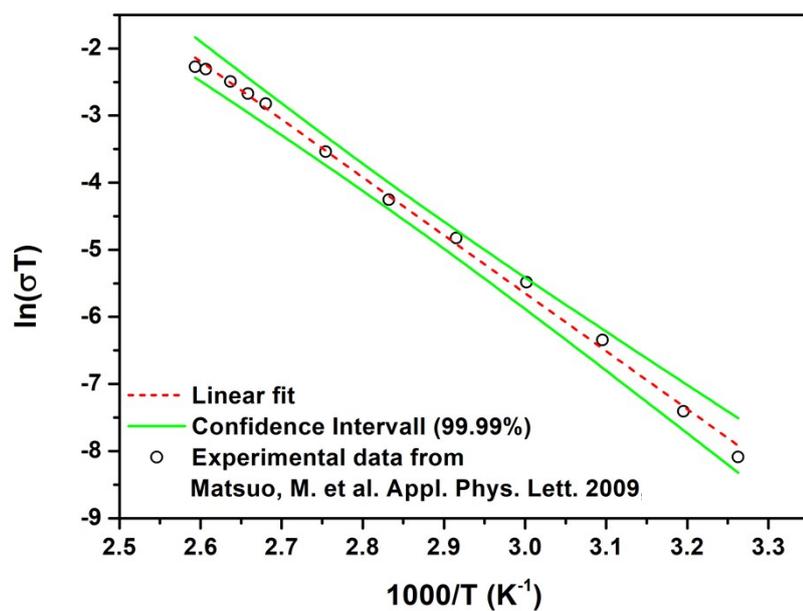


Figure S18. Example of linear fit on Li-ion conductivity for the orthorhombic phase of LiBH_4 as a function of inverse temperature. Data from Matsuo, M. et al. *Appl. Phys. Lett.* **2009**, 94, 224103 (Ref. 6).

Temperature-dependent EIS ramp	Temperature-dependent EIS cycle	Milled	E_A (eV)	C.I.	Ref.
Heating	1 st	No	0.69	0.04	(1)
Cooling	1 st	No	0.81	0.09	(1)
Heating	1 st	No	0.71	0.04	(2)
Cooling	1 st	No	0.7	0.1	(2)
Heating	1 st	Yes	0.74	0.07	(2)
Cooling	1 st	Yes	0.7	0.3	(2)
Heating	2 nd	Yes	0.8	0.1	(2)
Cooling	2 nd	Yes	0.9	0.2	(3)
Heating	1 st	No	0.77	0.04	(3)
Cooling	1 st	No	0.7	0.1	(3)
Heating	1 st	Yes	0.51	0.07	(3)
Cooling	1 st	Yes	0.51	0.05	(3)
Heating	1 st	No	0.8	0.1	(4)
Heating	1 st	No	0.71	0.04	(5)
Heating	1 st	No	0.50	0.1	(6)
Heating	1 st	No	0.91	0.08	(7)
Heating	2 nd	No	0.89	0.06	(7)
Heating	1 st	Yes	0.70	0.03	(7)
Heating	2 nd	Yes	0.75	0.04	(7)
Heating	1 st	No	0.76	0.04	(8)
Average			0.73	0.08	
Average exclude bold data			0.75	0.07	

Table S2. Activation energy (eV) calculated by linear plot of $\ln(\sigma T)$ and $1000/T$ of literature data, for LiBH_4 (99.99 % of confidence). CI is the confidence interval half width. The bold (red) data were excluded in the second calculation of the average activation energy data. 1st and 2nd refers to the temperature-dependent EIS cycle data.

Temperature-dependent EIS ramp	Temperature-dependent EIS cycle	Milled	$\ln\sigma_0$	C.I.	Ref.
Heating	1 st	No	15	1	(1)
Cooling	1 st	No	18	3	(1)
Heating	1 st	No	15	1	(2)
Cooling	1 st	No	13	5	(2)
Heating	1 st	Yes	20	2	(2)
Cooling	1 st	Yes	14	10	(2)
Heating	2 nd	Yes	17	3	(2)
Cooling	2 nd	Yes	20	8	(3)
Heating	1 st	No	16	1	(3)
Cooling	1 st	No	14	4	(3)
Heating	1 st	Yes	15	2	(3)
Cooling	1 st	Yes	14	2	(3)
Heating	1 st	No	18	3	(4)
Heating	1 st	No	15	1	(5)
Heating	1 st	No	6	4	(6)
Heating	1 st	No	21	2	(7)
Heating	2 nd	No	22	2	(7)
Heating	1 st	Yes	19.3	0.3	(7)
Heating	2 nd	Yes	17	1	(7)
Heating	1 st	No	16	1	(8)
Average			16	3	
Average exclude bold data			16	2	

Table S3. $\ln\sigma_0$ calculated by linear plot of $\ln(\sigma T)$ and $1000/T$ of literature data, for LiBH_4 (99.99 % of confidence). CI is the confidence interval half width. The bold (red) data were excluded in the second calculation of the average $\ln\sigma_0$. 1st and 2nd refers to the temperature-dependent EIS cycle data.

Temperature-dependent EIS ramp	Temperature-dependent EIS cycle	Milled	$\ln(\sigma T)_{FIT}$	$\ln(\sigma T)_{MAX}$	$\ln(\sigma T)_{MIN}$	Ref.
Heating	1 st	No	-11.40	-11.18	-11.63	(1)
Cooling	1 st	No	-13.12	-12.59	-13.66	(1)
Heating	1 st	No	-11.98	-11.78	-12.19	(2)
Cooling	1 st	No	-12.78	-13.93	-11.63	(2)
Heating	1 st	Yes	-8.23	-7.85	-8.61	(2)
Cooling	1 st	Yes	-13.48	-12.12	-14.84	(2)
Heating	2 nd	Yes	-14.41	-13.80	-15.02	(2)
Cooling	2 nd	Yes	-14.90	-13.49	-16.30	(3)
Heating	1 st	No	-12.95	-12.74	-13.16	(3)
Cooling	1 st	No	-12.83	-12.23	-13.43	(3)
Heating	1 st	Yes	-5.04	-4.77	-5.31	(3)
Cooling	1 st	Yes	-5.01	-4.82	-5.20	(3)
Heating	1 st	No	-13.17	-12.53	-13.81	(4)
Heating	1 st	No	-12.01	-11.81	-12.21	(5)
Heating	1 st	No	-12.90	-12.53	-13.26	(6)
Heating	1 st	No	-13.31	-13.04	-13.58	(7)
Heating	2 nd	No	-13.72	-13.44	-13.99	(7)
Heating	1 st	Yes	-7.25	-6.36	-8.13	(7)
Heating	2 nd	Yes	-11.81	-11.24	-12.37	(7)
Heating	1 st	No	-12.90	-12.68	-13.12	(8)
Average $\ln(\sigma T)$			-11.66	-11.25	-12.07	C.I.
Average σ at 30 °C			2.85x10⁻⁸	4.30x10⁻⁸	1.89x10⁻⁸	1.21x10⁻⁸
Average $\ln(\sigma T)$ exclude bold data			-12.76	-12.54	-12.97	C.I.
Average σ at 30 °C exclude bold data			9.5x10⁻⁹	1.18x10⁻⁸	7.66x10⁻⁹	2.07x10⁻⁹

Table S4. $\ln(\sigma T)$ and Li-ion conductivity at 30 °C (303.14 K) calculated using the data ($1000 E_A/k_B$ and $\ln\sigma_0$) obtained by the linear plot of $\ln(\sigma T)$ and $1000/T$ of literature data (see Table S1 and Table S2) for LiBH_4 . $\ln(\sigma T)_{FIT}$ has been calculated using the slope and intercept obtained by the linear fit. $\ln(\sigma T)_{MAX}$ and $\ln(\sigma T)_{MIN}$ were calculated considering the maximum and the minimum values, of the slope and intercept, of confidence interval obtained by the linear fit. The bold (red) data were excluded in the second calculation of the average $\ln(\sigma T)$ data. 1st and 2nd refers to the temperature-dependent EIS cycle data. CI is the confidence interval half width, calculated averaging the $\ln(\sigma T)_{MAX}$ and the $\ln(\sigma T)_{MIN}$.

Statistical analysis for LiNH₂, Li₂NH, Li₂(NH₂)(BH₄) and Li₄(NH₂)₃(BH₄)

Statistical analysis was performed in order to obtain the average value of activation energy for LiNH₂, Li₂NH, Li₂(NH₂)(BH₄) and Li₄(NH₂)₃(BH₄), using the data reported in literature.^{10,15–18}

Compound	Temperature-dependent EIS ramp	Temperature-dependent EIS cycle	E_A	C.I.	$\ln\sigma_0$	C.I.	Ref.
LiNH ₂	Heating	1 st	0.9	0.04	19	1	(10)
	Heating	1 st	1.05	0.08	19	2	(11)
	Average		0.98	0.06	19	2	
Li ₂ NH	Heating	1 st	0.63	0.01	21.8	0.2	(11)
	Heating	1 st	0.58	0.02	20,1	0,9	(12)
	Heating	1 st	0.6	0.1	20	5	(12)
Average		0.60	0.04	21	2		
Li ₂ NH ₂ BH ₄	Heating	1 st	0.7	0.04	23	1	(10)
	Cooling	1 st	0.67	0.06	22	2	(10)
	Average		0.69	0.06	23	2	
Li ₄ (NH ₂) ₃ BH ₄	Heating	1 st	0.26	0.01	7,0	0,4	(5)
	Heating	1 st	0.33	0.01	8.7	0.4	(10)
	Cooling	1 st	0.55	0.03	13	1	(10)
	Heating	1 st	0.34	0.01	11	1	(13)
Average		0.37	0.02	10.1	0.6		

Table S5. Activation energy (eV) and $\ln\sigma_0$ calculated by linear plot of $\ln(\sigma T)$ and $1000/T$ of literature data, for the LiNH₂, Li₂NH, Li₂(NH₂)(BH₄) and Li₄(NH₂)₃(BH₄) (95 % of confidence). CI is the confidence interval half width.

Compound	Temperature-dependent EIS ramp	Temperature-dependent EIS cycle	$\ln(\sigma T)_{FIT}$	$\ln(\sigma T)_{MAX}$	$\ln(\sigma T)_{MIN}$	Ref.
LiNH ₂	Heating	1 st	-14.97	-14.71	-15.23	(5)(10)
	Heating	1 st	-20.90	-19.75	-22.05	(11)
	Average $\ln(\sigma T)$		-17.94	-17.23	-18.64	C.I.
	Average σ at 30 °C		5.36x10⁻¹¹	1.09x10⁻¹⁰	2.64x10⁻¹¹	4.11x10⁻¹¹
Li ₂ NH	Heating	1 st	-2.41	-2,38	-2,44	11
	Heating	1 st	-2.22	-2,17	-2,27	12
	Heating	1 st	-1.96	-1,32	-2,61	12
	Average $\ln(\sigma T)$		-2.20	-1,96	-2,44	C.I.
Average σ at 30 °C		3.66x10⁻⁴	4.66x10⁻⁴	2.88x10⁻⁴	8.92x10⁻⁵	
Li ₂ NH ₂ BH ₄	Heating	1 st	-3.36	-3.22	-3.50	(5)(10)
	Cooling	1 st	-3.62	-3.43	-3.80	(10)
	Average $\ln(\sigma T)$		-3.49	-3.33	-3.65	C.I.
	Average σ at 30 °C		1.01x10⁻⁴	1.18x10⁻⁴	8.57x10⁻⁵	1.63x10⁻⁵
Li ₄ (NH ₂) ₃ BH ₄	Heating	1 st	-2.93	-2.86	-3.00	(5)
	Heating	1 st	-3.53	-3.44	-3.62	(10)
	Cooling	1st	-7.59	-7.39	-7.79	(10)
	Heating	1 st	-2.74	-2.11	-3.37	(13)
	Average $\ln(\sigma T)$		-4.19	-3.95	-4.44	C.I.
	Average σ at 30 °C		4.98x10⁻⁵	6.37x10⁻⁵	3.89x10⁻⁵	1.24x10⁻⁵
	Average $\ln(\sigma T)$ exclude bold data		-3,06	-2.80	-3.33	C.I.
Average σ at 30 °C exclude bold data		1.54x10⁻⁴	2.00x10⁻⁴	1.14x10⁻⁴	4.09x10⁻⁵	

Table S6. $\ln(\sigma T)$ and Li-ion conductivity at 30 °C (303.14 K) calculated using the data ($1000 E_A/k_B$ and $\ln\sigma_0$) obtained by the linear plot of $\ln(\sigma T)$ and $1000/T$ of literature data (see Table S1 and Table S2) for the LiNH₂, Li₂NH, Li₂(NH₂)(BH₄) and Li₄(NH₂)₃(BH₄). $\ln(\sigma T)_{FIT}$ has been calculated using the slope and intercept obtained by the linear fit. $\ln(\sigma T)_{MAX}$ and $\ln(\sigma T)_{MIN}$ were calculated considering the maximum and the minimum values, of the slope and intercept, of confidence interval obtained by the linear fit. The bold (red) data were excluded in the second calculation of the average $\ln(\sigma T)$ data. 1st and 2nd refers to the temperature-dependent EIS cycle data. CI is the confidence interval half width, calculated averaging the maximum and the $\ln(\sigma T)_{MAX}$ and the $\ln(\sigma T)_{MIN}$.

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