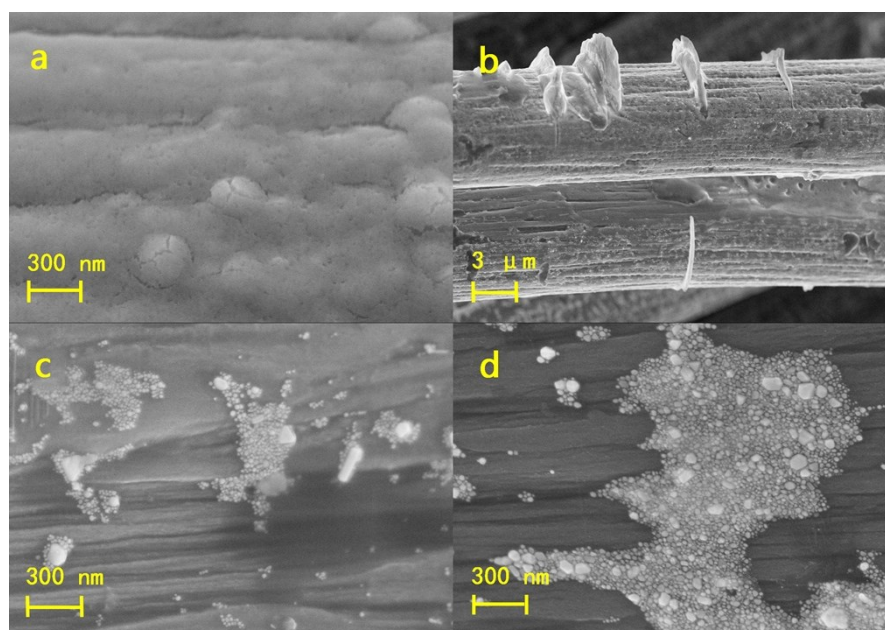


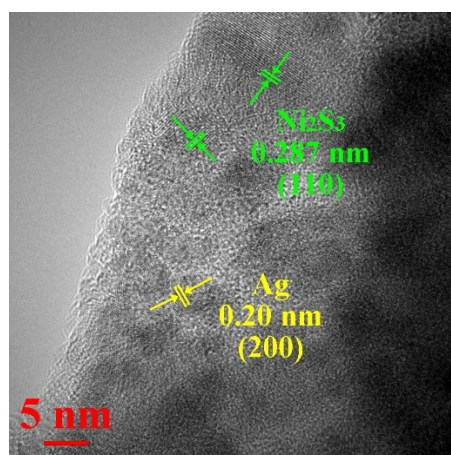
Supplementary information:

## Uncover the role of Ag in layer-alternating $\text{Ni}_3\text{S}_2/\text{Ag}/\text{Ni}_3\text{S}_2$ as electrocatalyst with enhanced OER performance

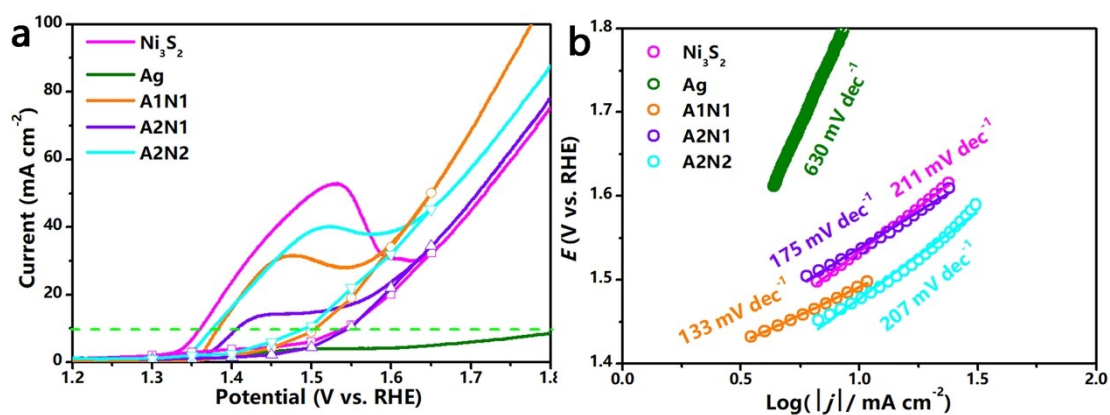
Rui Guo<sup>1,2,3</sup>, Yan He<sup>1,3,\*</sup>, Renchao Wang<sup>1,3</sup>, Junhua You<sup>4</sup>, Hongji Lin<sup>5</sup>, Chiente Chen<sup>5</sup>, Tingshan Chan<sup>5</sup>, Xuanwen Liu<sup>1,3,\*</sup>, Zhiwei Hu<sup>6,\*</sup>



**Fig. S1.** SEM images of Ag samples by electrodeposition for (a) 2 s, (b) 4 s, (c) 6 s, (d) 8 s.



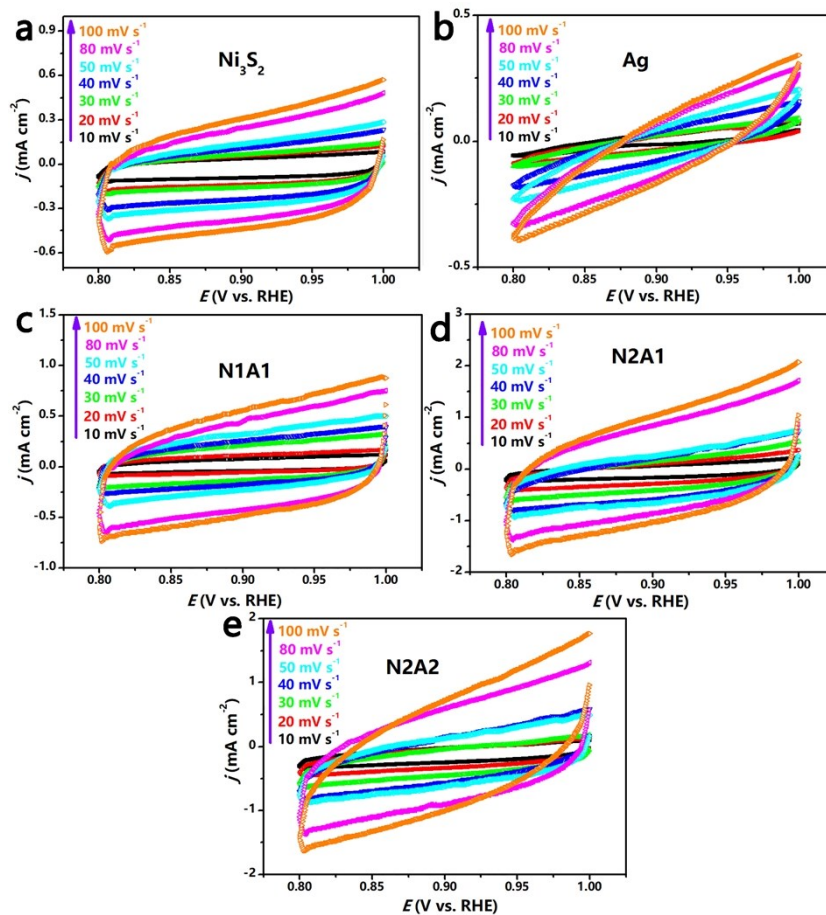
**Fig. S2.** The HRTEM images of N1A1.



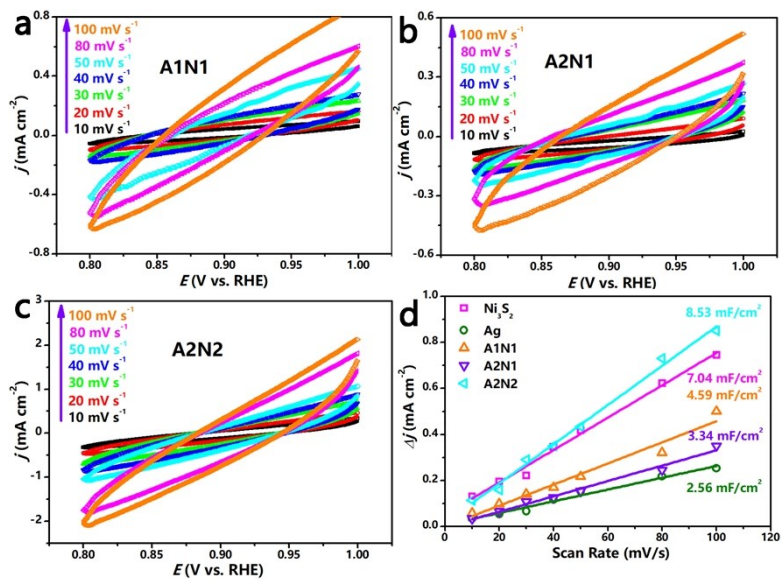
**Fig. S3.** (a) The OER polarization curves and (b) Tafel plots of Ni<sub>3</sub>S<sub>2</sub>, Ag, A1N1, A2N1, and A2N2

**Table S1.** Comparison of the OER performance of N2A1 with Ni<sub>3</sub>S<sub>2</sub> electrocatalysts in alkaline electrolyte.

| Catalysts                                                          | $\eta_{10}$ (mV)              | Tafel Slope (mV dec <sup>-1</sup> ) | Electrolyte | Ref.      |
|--------------------------------------------------------------------|-------------------------------|-------------------------------------|-------------|-----------|
| Ni <sub>3</sub> S <sub>2</sub> nanoporous thin films               | 319 @ 20 mA cm <sup>-2</sup>  | 101.2                               | 1.0 M KOH   | S1        |
| Ni <sub>3</sub> S <sub>2</sub> /VS <sub>2</sub>                    | 227                           | 59.9                                | 1.0 M NaOH  | S2        |
| Fe-doped Ni <sub>3</sub> S <sub>2</sub> particles film             | 253 @ 100 mA cm <sup>-2</sup> | 65.5                                | 1.0 M KOH   | S3        |
| Ni <sub>3</sub> S <sub>2</sub> nanorods/NF                         | 157                           | 153                                 | 1.0 M KOH   | S4        |
| Fe <sub>17.5%</sub> -Ni <sub>3</sub> S <sub>2</sub> /NF            | 214                           | 42                                  | 1.0 M KOH   | S5        |
| Ni <sub>3</sub> S <sub>2</sub> /NF-2                               | 425 @ 100 mA cm <sup>-2</sup> | -                                   | 1.0 M KOH   | S6        |
| Co <sub>3</sub> O <sub>4</sub> @Ni <sub>3</sub> S <sub>2</sub> /NF | 260 @ 20 mA cm <sup>-2</sup>  | 121.7                               | 1.0 M KOH   | S7        |
| CdS/Ni <sub>3</sub> S <sub>2</sub> /PNF                            | 151                           | 174                                 | 1.0 M KOH   | S8        |
| Ag/Ni <sub>3</sub> S <sub>2</sub> (N2A1)                           | 187                           | 118                                 | 1.0 M KOH   | This work |



**Fig. S4.** Cyclic voltammogram curves of  $\text{Ni}_3\text{S}_2$  (a), Ag (b), N1A1 (c), N2A1 (d), and N2A2 (e) measured in the range of 0.8 ~ 1.0 V vs. RHE at different scan rates: 10, 20, 30, 40, 50, 80, and 100  $\text{mV s}^{-1}$ .

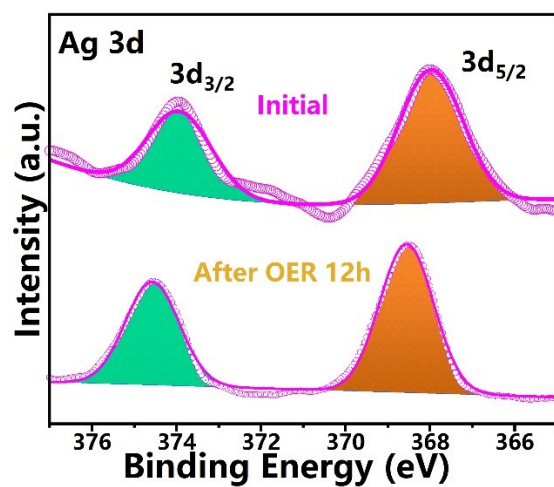


**Fig. S5.** Cyclic voltammogram curves of A1N1 (a), A2N1 (b), and A2N2 (c) measured in the range of 0.8 ~ 1.0 V vs. RHE at different scan rates: 10, 20, 30, 40, 50, 80, and 100  $\text{mV s}^{-1}$ . (d)

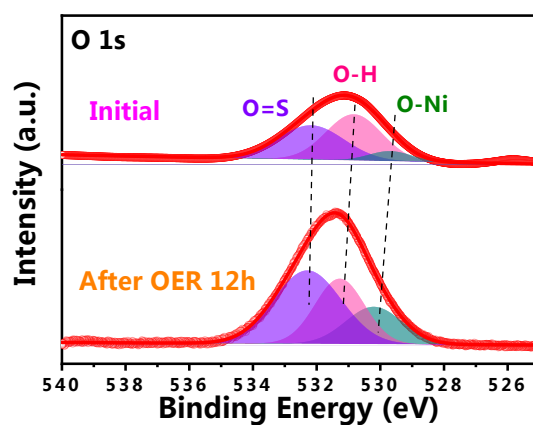
Double layer capacitance (Cdl) of as-prepared electrocatalysts.

**Table S2.** Electrocatalytic performances of different samples

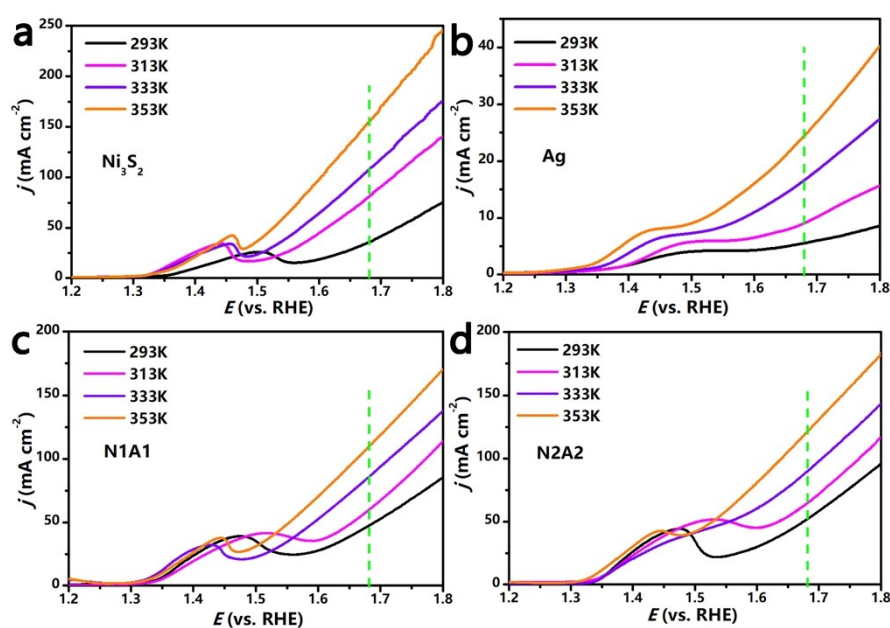
| Sample | OER                               |                                     |                 |            |
|--------|-----------------------------------|-------------------------------------|-----------------|------------|
|        | $\eta_j$ @ 10 mA cm <sup>-2</sup> | Tafel slope (mV dec <sup>-1</sup> ) | $R_{ct}/\Omega$ | $W/\Omega$ |
| Ag     | -                                 | 630                                 | 6.25            | 53.31      |
| Ni3S2  | 308                               | 211                                 | 0.56            | 3.42       |
| N1A1   | 289                               | 190                                 | 0.41            | 2.14       |
| N2A1   | 187                               | 118                                 | 0.19            | 1.62       |
| N2A2   | 233                               | 154                                 | 0.50            | 1.61       |



**Fig. S6.** High resolution XPS spectra of Ag 3d before and after OER for 12 h.



**Fig. S7.** High resolution XPS spectrums of O 1S of N2A1 before and after OER for 12 h.



**Fig. S8.** LSV curves of  $\text{Ni}_3\text{S}_2$  (a), Ag (b), N1A1 (c), and N2A2 (d) at a scan rate of  $5 \text{ mV s}^{-1}$  in 1.0 M KOH towards OER as the temperature rose from 293 to 353 K.

## Reference

- S1 J. Dong, F. Zhang, Y. Yang, Y. Zhang, H. He, X. Huang, X. Fan and X. Zhang, (003)-Facet-exposed  $\text{Ni}_3\text{S}_2$  nanoporous thin films on nickel foil for efficient water splitting, *Appl. Catal. B*, 2019, **243**, 693-702.
- S2 X. Zhong, J. Tang, J. Wang, M. Shao, J. Chai, S. Wang, M. Yang, Y. Yang, N. Wang, S. Wang, B. Xu and H. Pan, 3D heterostructured pure and N-Doped  $\text{Ni}_3\text{S}_2/\text{VS}_2$  nanosheets for high efficient overall water splitting, *Electrochim. Acta*, 2018, **269**, 55-61.
- S3 N. Cheng, Q. Liu, A. M. Asiri, W. Xing and X. Sun, A Fe-doped  $\text{Ni}_3\text{S}_2$  particle film as a high-efficiency robust oxygen evolution electrode with very high current density, *J. Mater. Chem. A*, 2015, **3**, 23207-23212.
- S4 W. Zhou, X. Wu, X. Cao, X. Huang, C. Tan, J. Tian, H. Liu, J. Wang and H. Zhang,  $\text{Ni}_3\text{S}_2$  nanorods/Ni foam composite electrode with low overpotential for electrocatalytic oxygen evolution, *Energy Environ. Sci.*, 2013, **6**, 2921-2924.
- S5 G. Zhang, Y. Feng, W. Lu, D. He, C. Wang, Y. Li, X. Wang and F. Cao, Enhanced Catalysis of Electrochemical Overall Water Splitting in Alkaline Media by Fe Doping in  $\text{Ni}_3\text{S}_2$  Nanosheet Arrays, *ACS Catal.*, 2018, **8**, 5431-5441.
- S6 G. Liu, Z. Sun, X. Zhang, H. Wang, G. Wang, X. Wu, H. Zhang and H. Zhao, Vapor-phase hydrothermal transformation of a nanosheet array structure  $\text{Ni}(\text{OH})_2$  into ultrathin  $\text{Ni}_3\text{S}_2$  nanosheets on nickel foam for high-efficiency overall water splitting, *J. Mater. Chem. A*, 2018, **6**, 19201-19209.
- S7 Y. Gong, Z. Xu, H. Pan, Y. Lin, Z. Yang and X. Du, Hierarchical  $\text{Ni}_3\text{S}_2$  nanosheets coated on  $\text{Co}_3\text{O}_4$  nanoneedle arrays on 3D nickel foam as an efficient electrocatalyst for the oxygen evolution reaction, *J. Mater. Chem. A*, 2018, **6**, 5098-5106.
- S8 S. Qu, J. Huang, J. Yu, G. Chen, W. Hu, M. Yin, R. Zhang, S. Chu and C. R. Li,  $\text{Ni}_3\text{S}_2$  Nanosheet Flowers Decorated with CdS Quantum Dots as a Highly Active Electrocatalysis Electrode for Synergistic Water Splitting, *ACS Appl. Mater. Interfaces*, 2017, **9**, 29660-29668.