

Supporting Information (SI)

Photochromism and photocatalysis of organic-inorganic hybrid iodoargentates modulated by argentophilic interactions

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1. Chemical analytical experiment for photoproducts

Dissolve 0.20 mmol powder samples of **3**, **3P**, **4** and **4P** in 5 ml dimethyl sulfoxide containing 1 mmol and 1 mmol NaI and Na₂S₂O₃, respectively. After several minutes, a small number of black precipitated particles are obtained in the solution of **3P** and **4P**. And then black precipitated particles are washed, dried and treated with concentrated HNO₃, solution of NaCl and concentrated NH₃·H₂O in turn, exhibiting the generation of metal silver particles. The above processes are implemented in the dark at room temperature.

2. Figures

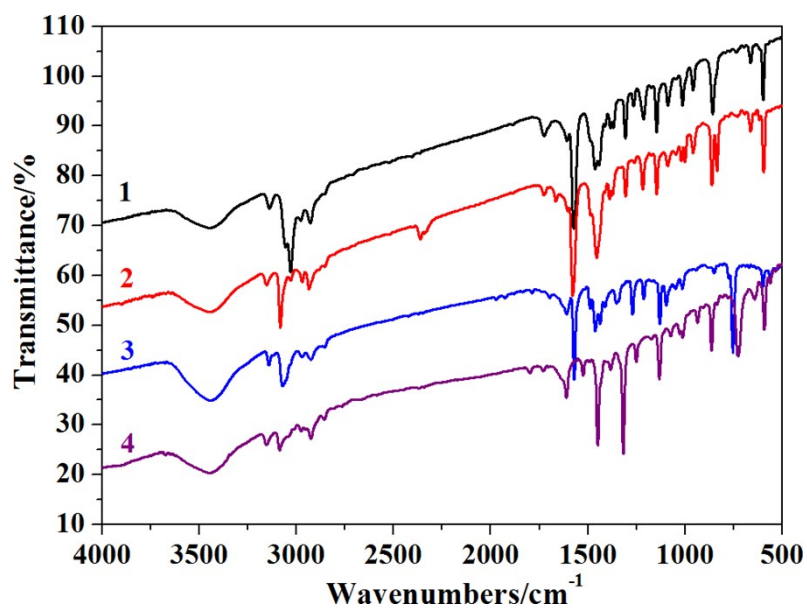


Fig. S1. IR spectra of 1-4.

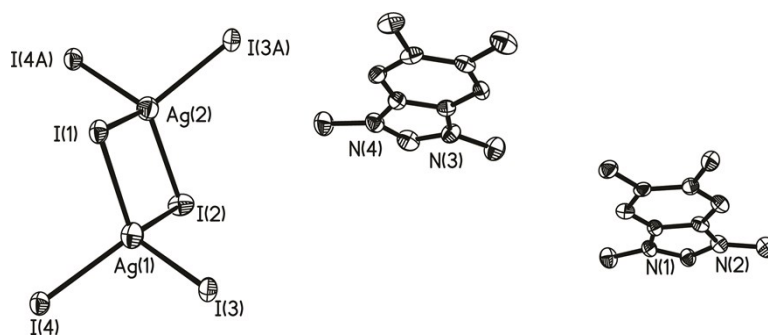


Fig. S2. The asymmetric unit diagram of 1.

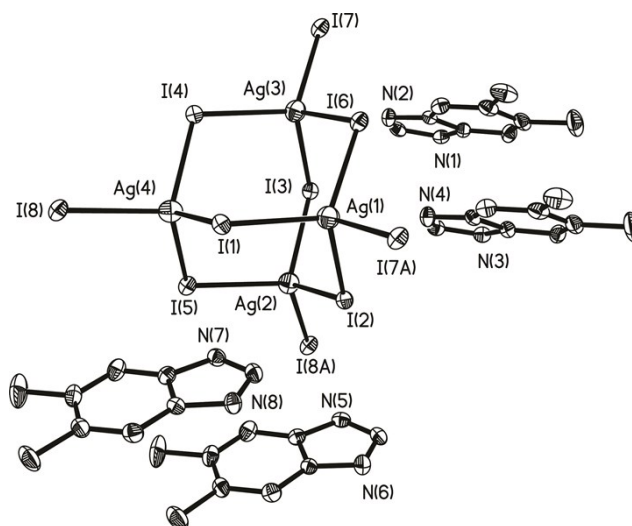


Fig. S3. The asymmetric unit diagram of 2.

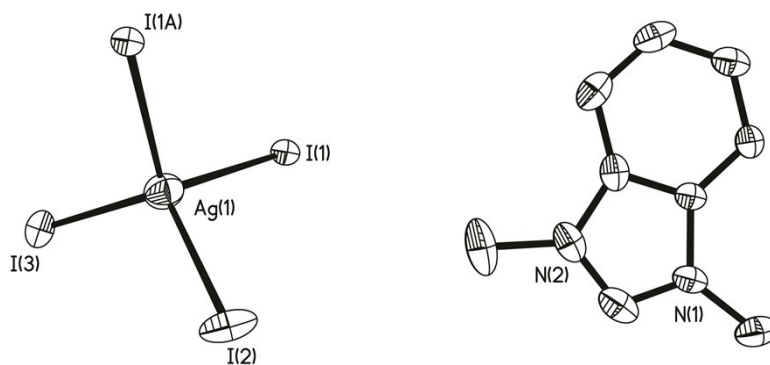


Fig. S4. The asymmetric unit diagram of 3.

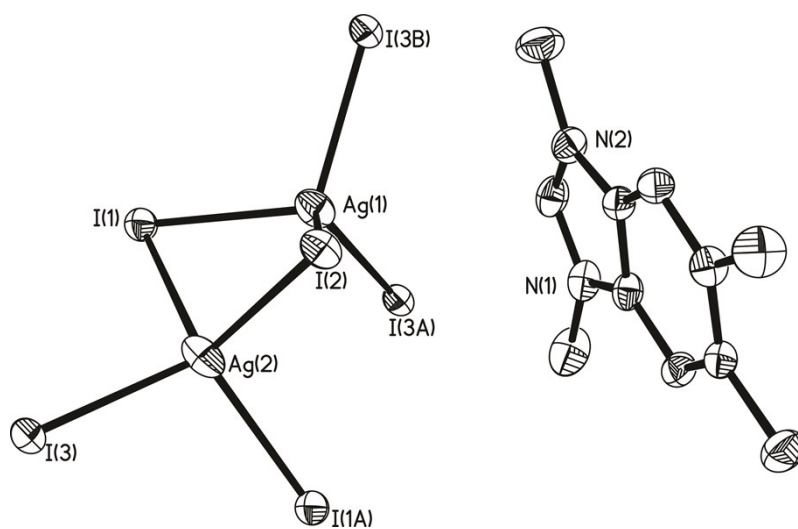


Fig. S5. The asymmetric unit diagram of 4.

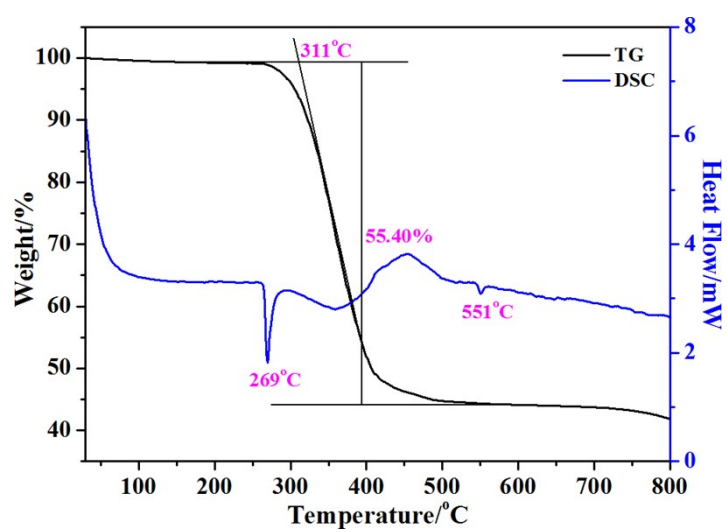


Fig. S6. Thermo-gravimetric (TG) and differential scanning calorimetry (DSC) curves of 1.

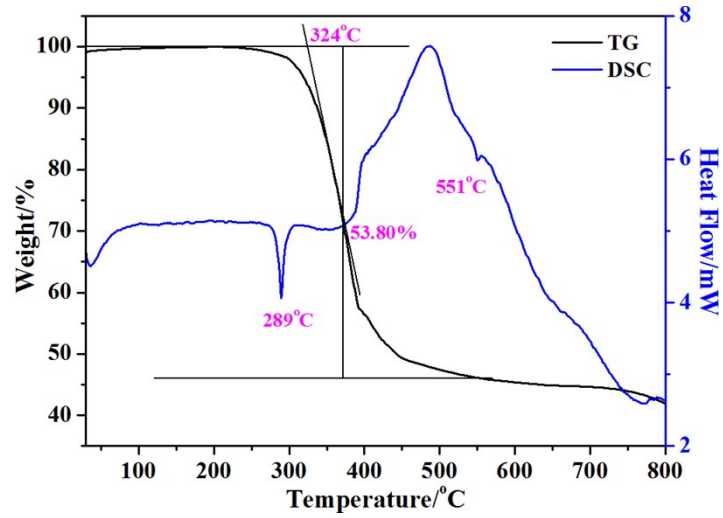


Fig. S7. Thermo-gravimetric (TG) and differential scanning calorimetry (DSC) curves of **2**.

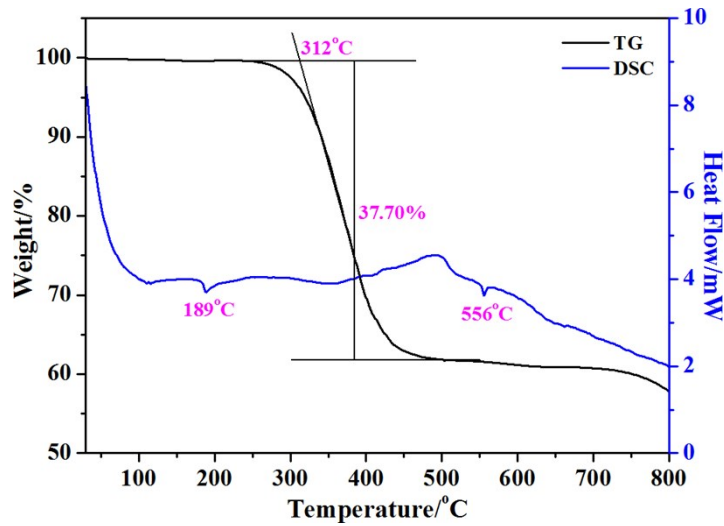


Fig. S8. Thermo-gravimetric (TG) and differential scanning calorimetry (DSC) curves of **3**.

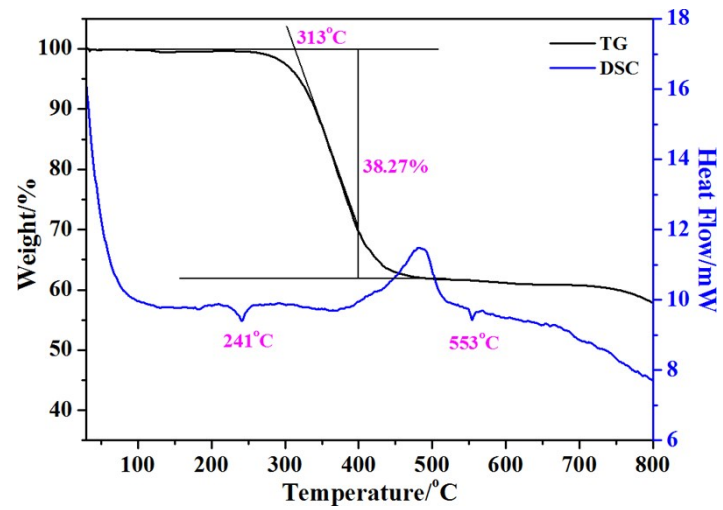


Fig. S9. Thermo-gravimetric (TG) and differential scanning calorimetry (DSC) curves of **4**.

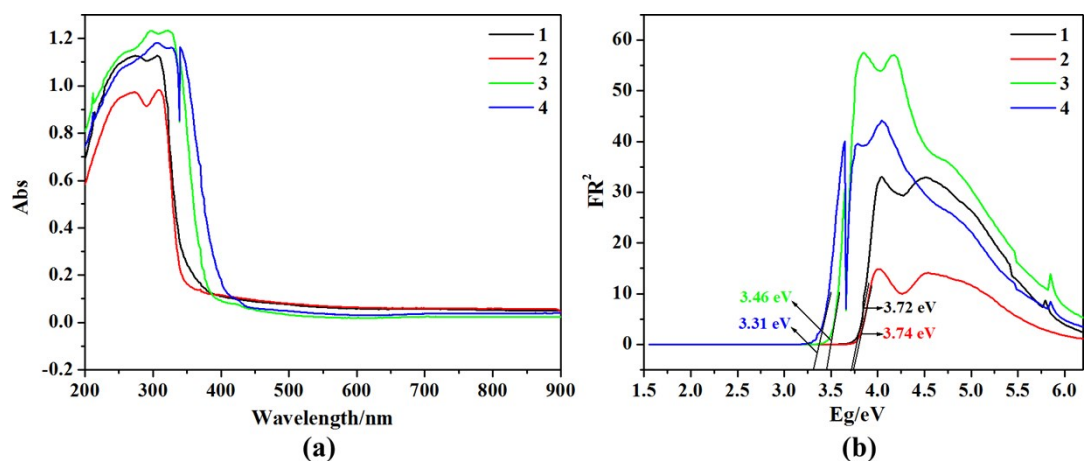


Fig. S10. The UV-vis absorption spectra (a) and band gaps (b) for **1**, **2**, **3** and **4**.

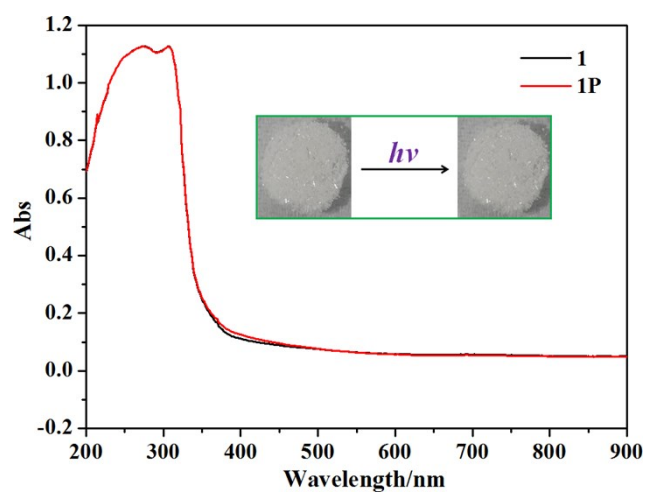


Fig. S11. UV-vis absorption spectra of **1** before and after UV light irradiation. The inset showing the photochromic behavior of **1**.

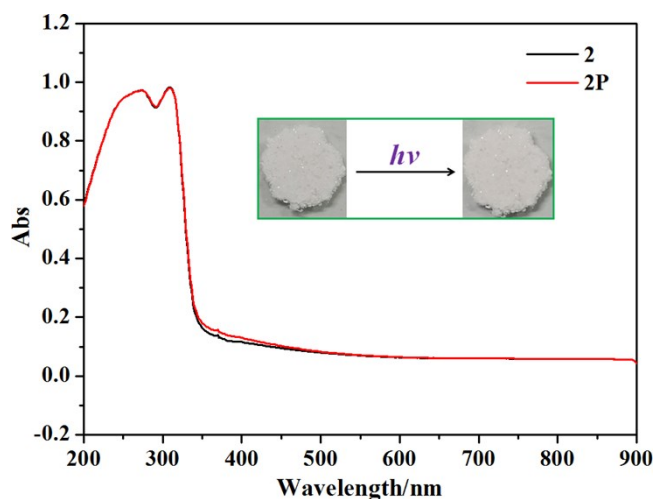


Fig. S12. UV-vis absorption spectra of **2** before and after UV light irradiation. The inset showing the photochromic behavior of **2**.

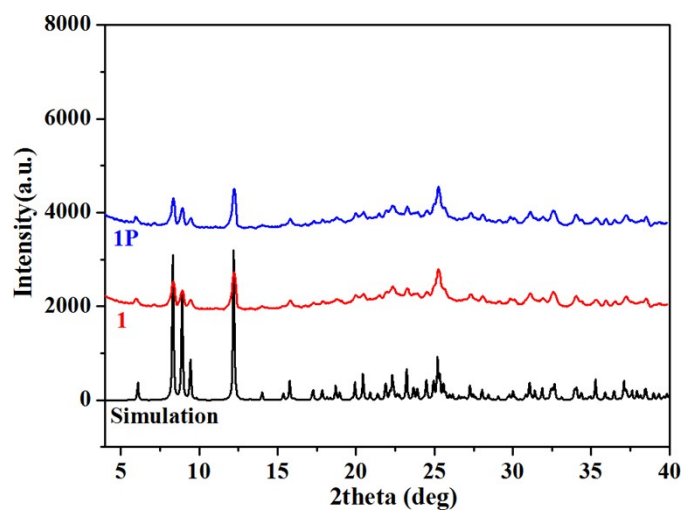


Fig. S13. Powder X-ray diffraction (PXRD) patterns of **1** and **1P** at room temperature.

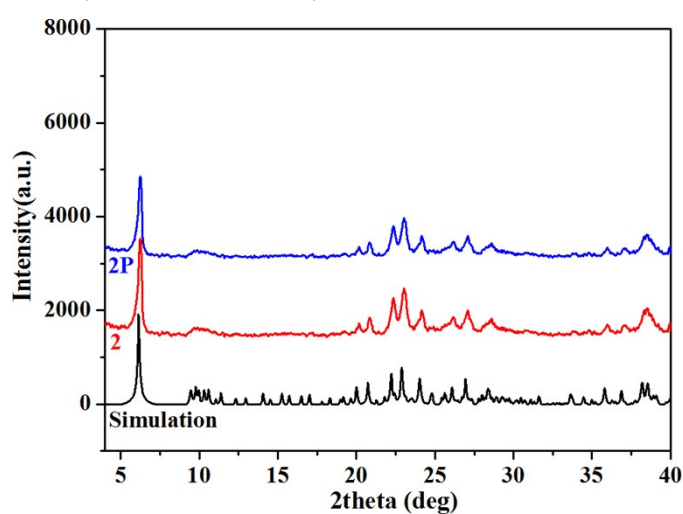


Fig. S14. Powder X-ray diffraction (PXRD) patterns of **2** and **2P** at room temperature.

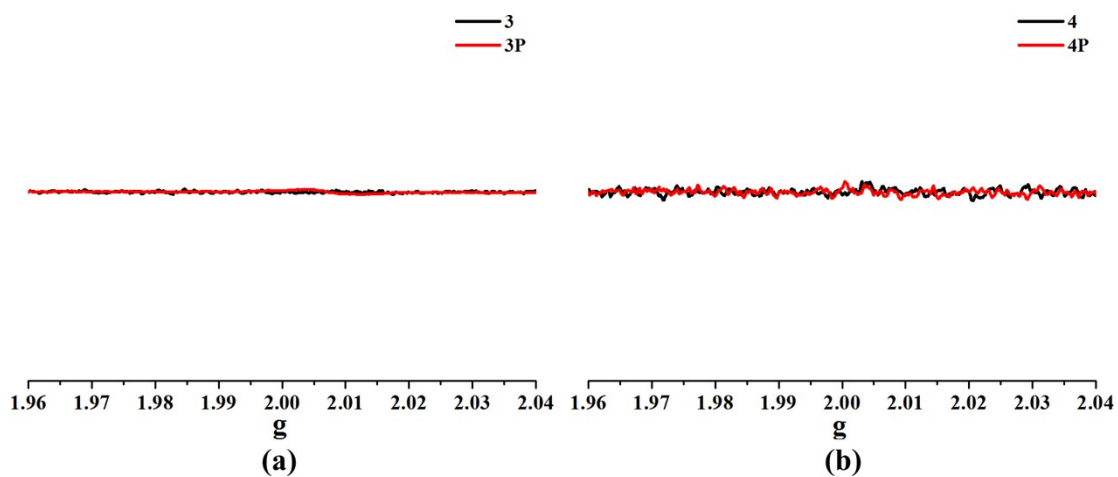


Fig. S15. The EPR spectra of **3** and **3P** (a), and **4** and **4P** (b)

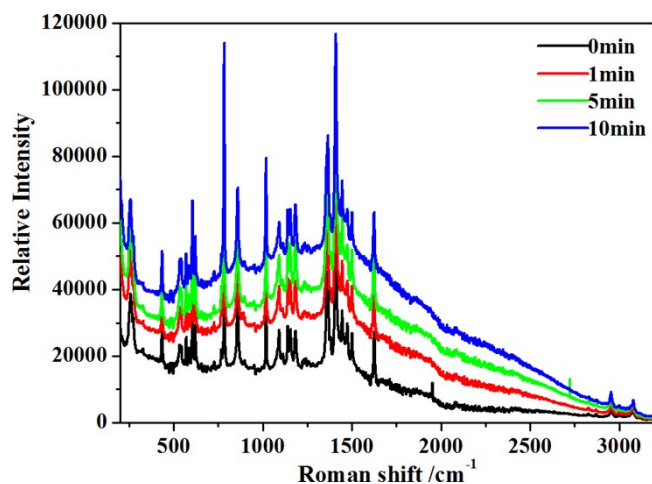


Fig. S16. The surface enhanced raman spectra (SERS) of **4** upon UV light irradiation.

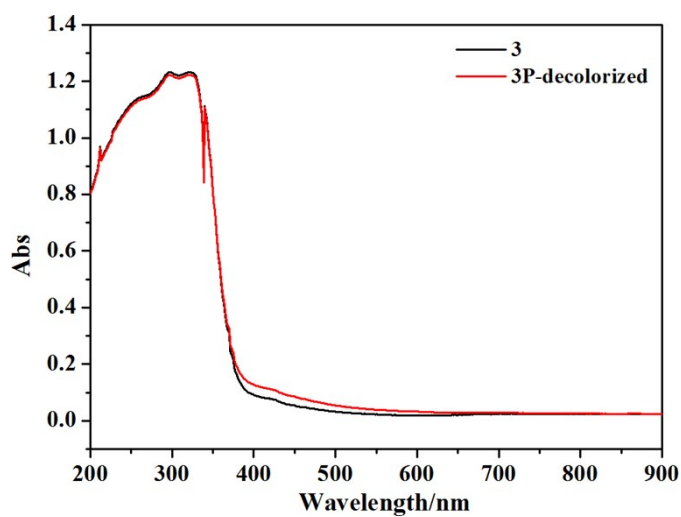


Fig. S17. UV-vis absorption spectra of **3** and **3P-decolorized**.

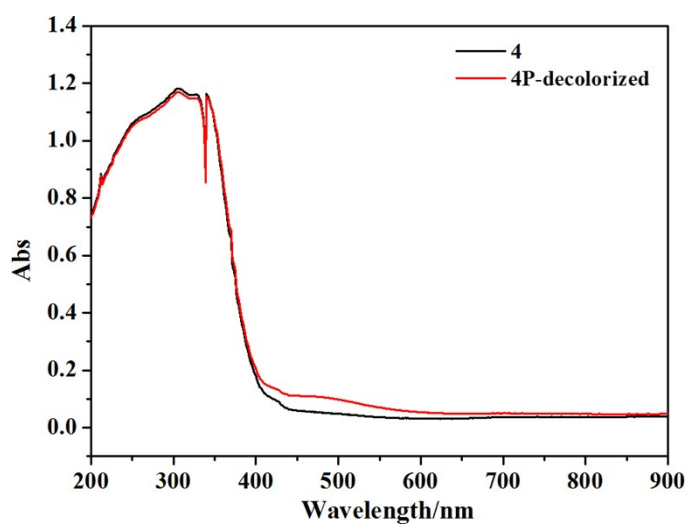


Fig. S18. UV-vis absorption spectra of **4** and **4P-decolorized**.

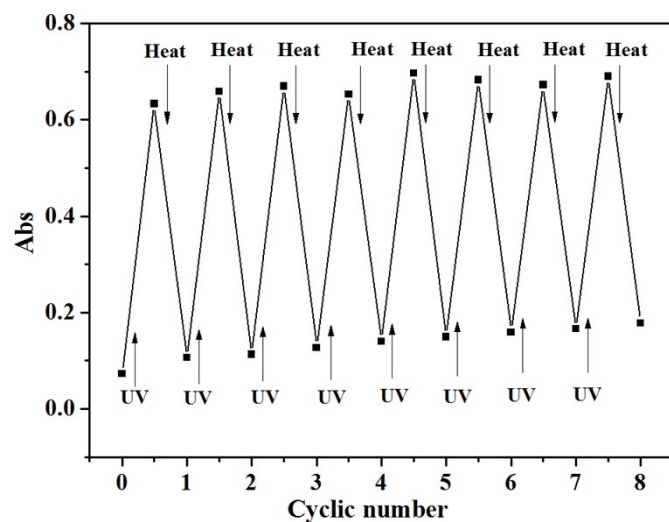


Fig. S19. The coloration-decoloration processes with repeated UV irradiation/heating of **3**.

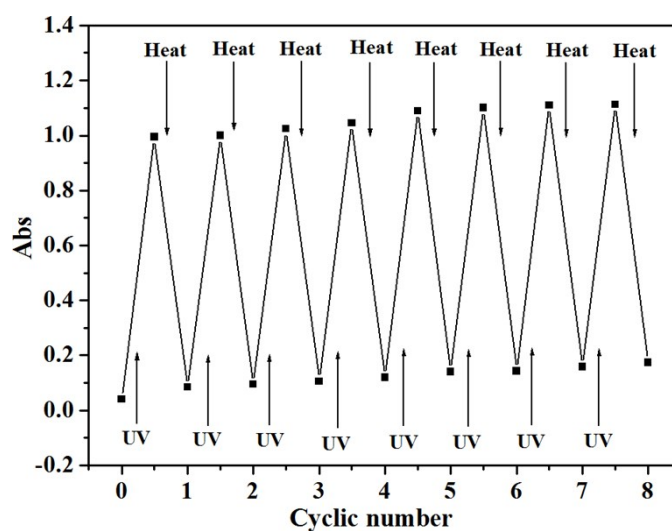


Fig. S20. The coloration-decoloration processes with repeated UV irradiation/heating of **3**.

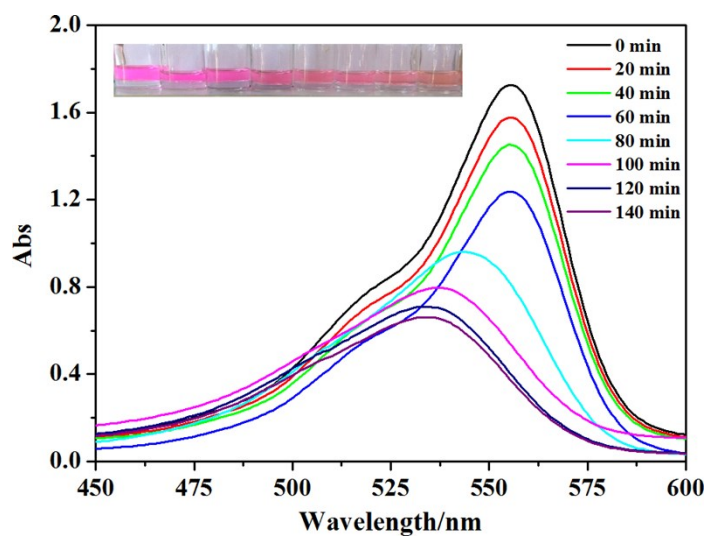


Fig. S21. UV-vis absorption spectra of the RhB solution in the presence of **3** upon light irradiation.

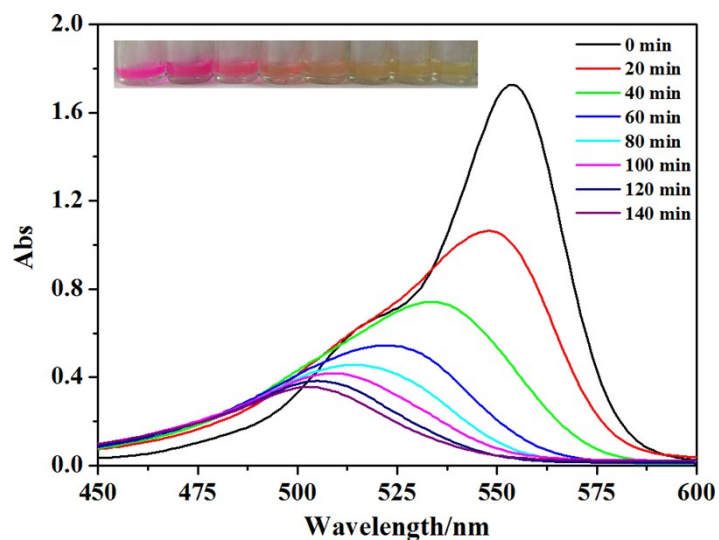


Fig. S22. UV-vis absorption spectra of the RhB solution in the presence of **4** upon light irradiation.

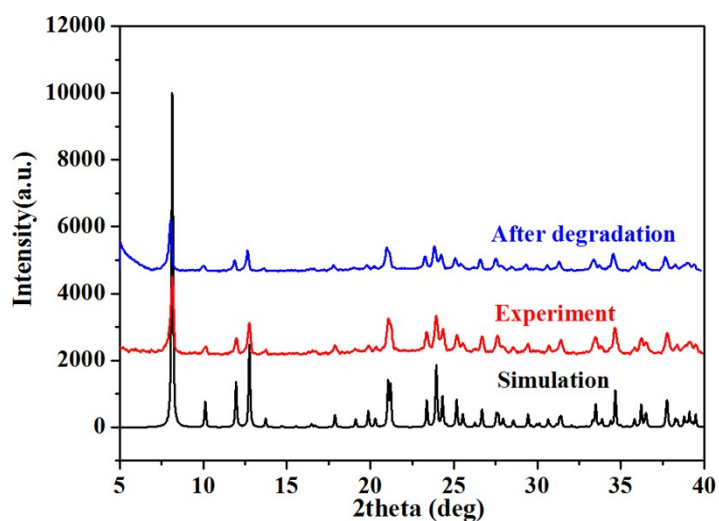


Fig. S23. The PXRD patterns of **3** before and after photodegradation.

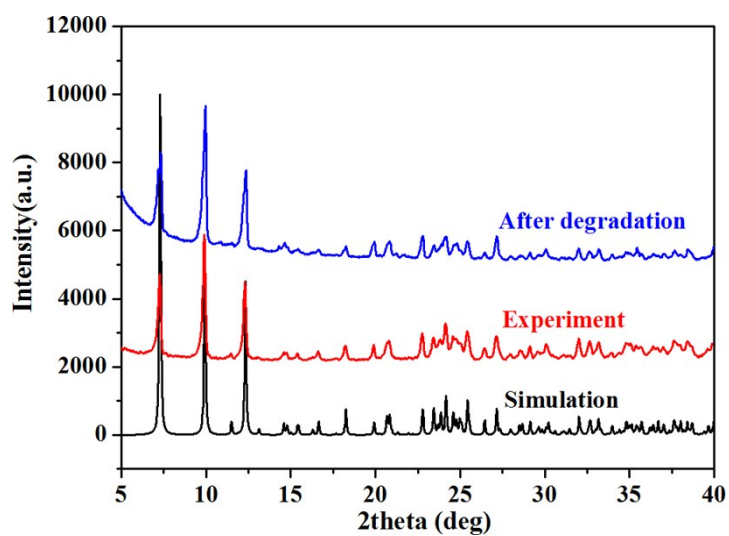


Fig. S24. The PXRD patterns of **4** before and after photodegradation.

3. Tables

Table S1 Crystal data and structure refinement for compounds **1-4**.

Compounds	1	2	3	4
CCDC code	2003491	2003492	2003493	2003494
Empirical formula	C ₂₂ H ₃₀ N ₄ Ag ₂ I ₄	C ₃₆ H ₄₄ N ₈ Ag ₄ I ₈	C ₉ H ₁₁ N ₂ Ag ₂ I ₃	C ₁₁ H ₁₅ N ₂ Ag ₂ I ₃
Formula weight	1073.84	2035.47	743.64	771.69
Crystal size (mm)	0.05×0.05×0.4	0.35×0.22×0.0	0.07×0.03×0.0	0.29×0.12×0.1
Crystal system	Monoclinic	Triclinic	Orthorhombic	Monoclinic
Space group	<i>P2₁/n</i>	<i>P-1</i>	<i>Pnma</i>	<i>P2₁/c</i>
<i>a</i> (Å)	7.0674(4)	9.4556(14)	17.4877(15)	8.9078(7)
<i>b</i> (Å)	19.8103(10)	9.4596(14)	6.6760(6)	24.2649(18)
<i>c</i> (Å)	21.2365(12)	29.009(4)	13.8701(12)	8.1079(6)
α (°)	90	87.486(3)	90	90
β (°)	92.976	87.279(3)	90	91.485(2)
γ (°)	90	84.662(3)	90	90
<i>V</i> (Å ³)	2969.3(3)	2558.8(7)	1619.3(2)	1751.9(2)
<i>Z</i>	4	2	4	4
<i>D_c</i> (g cm ⁻³)	2.402	2.642	3.050	2.926
<i>F</i> (000)	1984	1856	1328	1392
μ (mm ⁻¹)	5.495	6.369	8.117	7.508
Reflections collected	43180	37987	20906	23273
Unique reflections	7394	12463	2169	4289
<i>R</i> _{int}	0.0404	0.0340	0.0229	0.0340
Goodness-of-fit on <i>F</i> ²	1.194	1.156	1.175	1.258
<i>R</i> ₁ / <i>wR</i> ₂ , [<i>I</i> ≥2σ(<i>I</i>)] ^{a,b}	0.0433, 0.1273	0.0442, 0.1350	0.0426, 0.1278	0.0405, 0.1139
<i>R</i> ₁ / <i>wR</i> ₂ , (all data)	0.0566, 0.1420	0.0563, 0.1537	0.0521, 0.1518	0.0462, 0.1232
$\Delta\rho_{\max}$, $\Delta\rho_{\min}$ (e Å ⁻³)	1.164, -1.966	1.916, -2.403	2.714, -2.561	2.820, -2.595

$$^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$$

$$^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$$

Table S2 Selected bond lengths (Å) and angles (°) for **1-4**

Compound 1			
Ag(1)-I(2)	2.8201(8)	Ag(2)-I(3)#1	2.8478(8)
Ag(1)-I(3)	2.8482(8)	Ag(2)-I(4)#1	2.8839(9)
Ag(1)-I(4)	2.8770(8)	Ag(2)-I(1)	2.9176(8)
Ag(1)-I(1)	2.9175(8)	Ag(2)-I(2)	2.8271(9)
I(2)-Ag(1)-I(3)	110.19(3)	I(3)#1-Ag(2)-I(4)#1	106.45(3)
I(2)-Ag(1)-I(4)	118.13(3)	I(2)-Ag(2)-I(1)	100.89(3)
I(3)-Ag(1)-I(4)	106.63(3)	I(3)#1-Ag(2)-I(1)	111.43(3)
I(2)-Ag(1)-I(1)	101.06(3)	I(4)#1-Ag(2)-I(1)	108.28(2)
I(3)-Ag(1)-I(1)	111.78(3)	I(2)-Ag(2)-I(3)#1	111.14(3)
I(4)-Ag(1)-I(1)	109.08(2)	I(2)-Ag(2)-I(4)#1	118.58(3)
Symmetry code: #1 x-1,y,z; 2 x+1,y,z; #1 x,y,z-1; #2 -x+1,-y+1,-z+1; #3 x,y,z+1			
Compound 2			
Ag(1)-I(1)	2.8297(8)	Ag(1)-I(2)	2.8487(9)
Ag(1)-I(6)	2.8577(9)	Ag(1)-I(7)#1	2.9249(9)
Ag(2)-I(3)	2.8488(9)	Ag(2)-I(2)	2.8548(9)
Ag(2)-I(5)	2.8608(9)	Ag(2)-I(8)#2	2.8794(9)
Ag(3)-I(3)	2.8483(9)	Ag(3)-I(4)	2.8545(8)
Ag(3)-I(6)	2.8597(9)	Ag(3)-I(7)	2.8753(9)
Ag(4)-I(1)	2.8298(9)	Ag(4)-I(4)	2.8500(9)
Ag(4)-I(5)	2.8586(9)	Ag(4)-I(8)	2.9251(9)
I(1)-Ag(1)-I(2)	109.18(3)	I(1)-Ag(1)-I(6)	117.71(3)
I(2)-Ag(1)-I(6)	110.57(3)	I(1)-Ag(1)-I(7)#1	102.79(2)
I(2)-Ag(1)-I(7)#1	107.21(3)	I(6)-Ag(1)-I(7)#1	108.66(3)
I(3)-Ag(2)-I(2)	108.97(3)	I(3)-Ag(2)-I(5)	108.75(3)
I(2)-Ag(2)-I(5)	105.51(2)	I(3)-Ag(2)-I(8)#2	105.41(3)
I(2)-Ag(2)-I(8)#2	115.78(3)	I(5)-Ag(2)-I(8)#2	112.28(3)
I(3)-Ag(3)-I(4)	108.47(3)	I(3)-Ag(3)-I(6)	108.90(3)
I(4)-Ag(3)-I(6)	105.39(2)	I(3)-Ag(3)-I(7)	105.81(2)
I(4)-Ag(3)-I(7)	115.87(3)	I(6)-Ag(3)-I(7)	112.25(3)
I(1)-Ag(4)-I(4)	109.31(3)	I(1)-Ag(4)-I(5)	117.82(3)

I(4)-Ag(4)-I(5)	110.44(3)	I(1)-Ag(4)-I(8)	102.81(2)
I(4)-Ag(4)-I(8)	107.12(3)	I(5)-Ag(4)-I(8)	108.62(3)

Symmetry code: #1 x+1,y,z; #2 x,y-1,z; #3 x-1,y,z; #4 x,y+1,z

Compound 3

Ag(1)-I(3)	2.7990(9)	Ag(1)-I(2)	2.8011(9)
Ag(1)-I(1)#1	2.8979(8)	Ag(1)-I(1)	2.9508(9)
Ag(1)-Ag(1)#2	3.1268(15)		
I(3)-Ag(1)-I(2)	114.86(3)	I(3)-Ag(1)-I(1)#1	111.73(3)
I(2)-Ag(1)-I(1)#1	112.95(3)	I(3)-Ag(1)-I(1)	103.68(3)
I(2)-Ag(1)-I(1)	104.93(3)	I(1)#1-Ag(1)-I(1)	107.74(2)

Symmetry code: 1 -x+1,-y,-z+1; x,-y+1/2,z; -x+1,y-1/2,-z+1; x,-y-1/2,z

Compound 4

Ag(1)-I(1)	2.8544(8)	Ag(2)-I(3)	2.8155(8)
Ag(1)-I(3)#1	2.8577(8)	Ag(2)-I(1)#2	2.9263(9)
Ag(1)-I(2)	2.8620(8)	Ag(2)-I(1)	3.0727(9)
Ag(1)-I(3)#2	2.9375(8)	Ag(2)-I(2)	2.7737(7)
Ag(1)-Ag(2)#2	3.1846(8)	Ag(2)-Ag(1)#2	3.1846(8)
Ag(1)-Ag(2)	3.0664(9)	Ag(2)-Ag(2)#2	2.8909(13)
I(1)-Ag(1)-I(3)#1	109.61(3)	I(2)-Ag(2)-I(1)	102.28(2)
I(1)-Ag(1)-I(2)	105.71(3)	I(3)-Ag(2)-I(1)	95.62(3)
I(3)#1-Ag(1)-I(2)	104.01(2)	I(1)#2-Ag(2)-I(1)	122.44(2)
I(1)-Ag(1)-I(3)#2	102.13(2)	I(2)-Ag(2)-I(3)	131.36(3)
I(3)#1-Ag(1)-I(3)#2	108.58(2)	I(2)-Ag(2)-I(1)#2	104.12(3)
I(2)-Ag(1)-I(3)#2	126.19(3)	I(3)-Ag(2)-I(1)#2	103.37(2)

Symmetry code: #1 x,y,z-1; #2 -x+1,-y+1,-z+1; #3 x,y,z+1

Table S3. Hydrogen bonds of for compounds **1-4** (Å and °).

Compound 1				
D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
C(1)-H(1)...I(3)#1	0.930	3.094	3.946	153.17
C(2)-H(2C)...I(4)#1	0.960	3.231	4.147	160.20
C(4)-H(4)...I(2)#2	0.930	3.067	3.992	173.26
C(10)-H(10A)...I(3)#1	0.960	3.209	4.123	159.83
C(10)-H(10B)...I(3)#3	0.960	3.252	4.185	164.53
C(12)-H(12)...I(1)#2	0.930	2.877	3.760	159.01
C(13)-H(13A)...I(1)#2	0.960	3.271	4.146	152.32
C(15)-H(15)...I(3)#1	0.930	3.078	4.000	171.57
C(20)-H(20)...I(4)#4	0.930	3.288	4.202	168.19
#1 x-1, y, z-1; #2 x-1/2, -y+1/2, z-1/2; #3 -x+2, -y+1, -z+1; #4 -x+3/2, y+1/2, -z+3/2				
Compound 2				
D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
C(1)-H(1)...I(1)#1	0.930	3.083	3.771	132.17
C(1)-H(1)...I(2)#2	0.930	3.110	3.724	125.15
C(10)-H(10)...I(3)	0.930	3.106	3.703	123.62
C(10)-H(10)...I(4)#3	0.930	3.004	3.684	131.20
C(19)-H(19)...I(1)#3	0.930	3.064	3.758	132.79
C(19)-H(19)...I(4)#4	0.930	3.126	3.736	124.76
C(28)-H(28)...I(2)	0.930	2.991	3.682	132.38
C(28)-H(28)...I(3)#5	0.930	3.127	3.720	123.34
N(1)-H(1A)...I(7)#3	0.860	2.845	3.643	155.11
N(2)-H(2A)...I(2)#2	0.860	3.292	3.796	119.96
N(2)-H(2A)...I(3)	0.860	3.231	3.691	116.09
N(2)-H(2A)...I(7)	0.860	3.212	3.865	134.69
N(3)-H(3)...I(6)#3	0.860	2.894	3.678	152.47
N(4)-H(4)...I(2)	0.860	3.141	3.599	115.68
N(4)-H(4)...I(3)	0.860	3.245	3.753	120.33
N(4)-H(4)...I(6)	0.860	3.253	3.889	132.81
N(5)-H(5A)...I(8)#3	0.860	2.847	3.643	154.57
N(6)-H(6)...I(3)#5	0.860	3.225	3.686	116.17
N(6)-H(6)...I(4)#4	0.860	3.289	3.793	120.06

N(6)-H(6)···I(8)#4	0.860	3.216	3.866	134.33
N(7)-H(7)···I5	0.860	3.216	3.676	152.37
N(8)-H(8)···I(3)#5	0.860	3.249	3.757	120.38
N(8)-H(8)···I(4)#5	0.860	3.127	3.593	116.39
N(8)-H(8)···I(5)#5	0.860	3.261	3.893	132.39

#1 x-1, y, z-1; #2 x-1/2, -y+1/2, z-1/2; #3 -x+2, -y+1, -z+1; #4 -x+3/2, y+1/2, -z+3/2

Compound 3

D-H···A	d(D-H)	d(H···A)	d(D···A)	<(DHA)
C(2)-H(2)···I(1)#1	0.930	3.190	3.839	128.53
C(5)-H(5)···I(3)#2	0.930	3.008	3.899	160.95
C(8)-H(8)···I(3)#3	0.930	3.202	4.043	151.40
C(9)-H(9A)···I(1)#3	0.960	3.175	4.088	159.36
C(9)-H(9B)···I(3)	0.960	3.311	4.236	162.32

#1 -x+1, -y+1, -z+1; #2 -x+1, -y+1, -z+2; #3 -x+3/2, -y+1, z+1/2

Compound 4

D-H···A	d(D-H)	d(H···A)	d(D···A)	<(DHA)
C(1)-H(1C)···I(3)#1	0.960	2.998	3.915	160.18
C(1)-H(1A)···I(3)#2	0.960	3.215	4.079	150.54
C(2)-H(2)···I(1)#3	0.930	3.297	3.909	125.33
C(3)-H(3A)···I(3)#3	0.960	3.178	4.101	161.72
C(6)-H(6)···I(2)	0.930	3.192	4.111	170.03

#1 x-1, y, z-1; #2 -x+1, -y+1, -z+1; #3 -x+1, -y+1, -z; #4 x, y, z-1; #5 x, -y+1/2, z-1/2

Table S4. Synthesis conditions for compounds **1-4**.

Entry	DMBI/BI (mmol)	AgI (mmol)	NaI·2H ₂ O (mmol)	HI (mL)	Methanol (mL)	Acetonitrile (mL)	Product
1	0.6	0.6	2	0.2	2	3	1
2	0.6	0.6	2	0.2	5	0	2
3	0.6	1.2	2	0.2	2	3	3
4	0.6	1.2	2	0.2	2	3	4