## **Supplementary Information**

## *In-situ* growth of urchin-like cobalt-chromium phosphide on 3D graphene foam for efficient overall water splitting

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Fig. S1. SEM images of CoP@3DGF.



**Fig. S2.** SEM images of CoCr-P@3DGF with different Co/Cr ratios of (a-b) 1:1, (c-d) 3:1 and (e-f) 4:1.



Fig. S3. TEM images of (a)  $Co_2Cr_1$ -LDH@3DGF and (b)  $Co_2Cr_1$ -P@3DGF.



Fig. S4. EDS spectrum of  $Co_2Cr_1$ -P@3DGF composite.



Fig. S5. Element mapping images of Co<sub>2</sub>Cr<sub>1</sub>-P@3DGF (red: Co; green: Cr; yellow: P).



**Fig. S6.** Tafel curves of CoCr-P@3DGF with different Co/Cr ratios towards HER in 1.0 M KOH.



Fig. S7. CV curves of (a) Co-P@3DGF, (b) Co<sub>1</sub>Cr<sub>1</sub>-P@3DGF, (c) Co<sub>2</sub>Cr<sub>1</sub>-P@3DGF, (d)  $Co_3Cr_1$ -P@3DGF, (e) Co<sub>4</sub>Cr<sub>1</sub>-P@3DGF with various scan rates in 1.0 M KOH. (f) Double layer capacitances of CoCr-P@3DGF with different Co/Cr ratios.



Fig. S8. (a) The polarization, (b) Tafel curves and (c) CV curves of  $Co_2Cr_1$ -P@3DGF after 50 h stability tests for HER reaction in 1.0 M KOH. (d) Double layer capacitances of  $Co_2Cr_1$ -P@3DGF before and after HER reaction.



Fig. S9. (a) SEM images of  $Co_2Cr_1$ -P@3DGF after the stability test of HER. Highresolution XPS spectra of  $Co_2Cr_1$ -P@3DGF for (b) Co 2p, (c) P 2p, (d) O 1s, (e) Cr 2p before and after HER test.



Fig. S10. (a) Polarization curves and (b) corresponding Tafel curves of  $Co_2Cr_1$ -P@3DGF, Co-P@3DGF, Pt/C@3DGF and bare 3DGF in 0.5 M H<sub>2</sub>SO<sub>4</sub>.



Fig. S11. (a) The polarization, (b) Tafel curves and (c) CV curves of  $Co_2Cr_1$ -P@3DGF after 50 h stability tests for OER reaction in 1.0 M KOH. (d) Double layer capacitances of  $Co_2Cr_1$ -P@3DGF before and after OER reaction.



Fig. S12. (a) SEM images of  $Co_2Cr_1$ -P@3DGF after the stability test of OER. Highresolution XPS spectra of  $Co_2Cr_1$ -P@3DGF for (b) Co 2p, (c) P 2p, (d) O 1s, (e) Cr 2p before and after OER test.

Co/Cr feeding ratio –	EDS analysis [at%]			Estimated shaming formula
	Со	Cr	Р	- Estimated chemical formula
1:1	50.75	7.53	41.71	Co <sub>0.87</sub> Cr <sub>0.13</sub> -P
2:1	50.91	5.77	43.32	Co <sub>0.89</sub> Cr <sub>0.11</sub> -P
3:1	45.21	4.62	50.17	Co <sub>0.91</sub> Cr <sub>0.09</sub> -P
4:1	46.10	2.65	51.25	Co <sub>0.95</sub> Cr <sub>0.05</sub> -P

 Table S1. Estimated chemical formula of CoCr-P with different Co/Cr feeding ratios.

Catalysts	j (mA cm <sup>-2</sup> )	Overpotential (mV)	Tafel slope (mV dec <sup>-1</sup> )	Ref.
Co <sub>2</sub> Cr <sub>1</sub> -P@3DGF	10	118	71.2	This work
Er-doped CoP NMs	10	66	61	S1
Hollow Mo-CoP arrays	10	40	65	S2
Co <sub>5.47</sub> N NP@N-PC	10	149	86	S3
Co <sub>4</sub> Ni <sub>1</sub> P NTs	10	129	52	S4
CoP/C	10	160	20	85
Cu <sub>0.075</sub> Co <sub>0.0925</sub> P/CP	10	47	47.2	S6
(Ni,Co)Se <sub>2</sub> -GA	10	127	79	S7
CoP-400	10	151	72	S8
S-CoP	10	109	79	<b>S</b> 9
$Co_3S_4$ (a) $MoS_2$	10	136	74	S10

**Table S2.** Comparison of HER performances of  $Co_2Cr_1$ -P@3DGF with recently reportedcatalysts in 1.0 M KOH.

Catalysts	j	Overpotential	Tafel slope	Ref.
	(mA cm <sup>-2</sup> )	(mV)	(mV dec <sup>-1</sup> )	
Co <sub>2</sub> Cr <sub>1</sub> -P@3DGF	10	88	68	This work
Er-doped CoP NMs	10	52	32	S1
Cu <sub>0.075</sub> Co <sub>0.0925</sub> P/CP	10	70	55.1	S6
Urchin-like CoP	10	105	46	S11
CoP-400	10	113	67	S8
CoP <sub>3</sub> /Ni <sub>2</sub> P	10	115	49	S12
W-CoP NAs/CC	10	89	58	S13
CoMoP-5	10	95	61.1	S14
CoP@CC	10	131	64	S15
CoP/NCNHP	10	140	53	S16
HNDCM-Co/CoP	10	138	64	S17

**Table S3.** Comparison of HER performances of  $Co_2Cr_1$ -P@3DGF with recently reportedcatalysts in 0.5 M H<sub>2</sub>SO<sub>4</sub>.

Catalurta	j	Overpotential	Tafel slope	Def
Catalysts	(mA cm <sup>-2</sup> )	(mV)	(mV dec <sup>-1</sup> )	Kel.
Co <sub>2</sub> Cr <sub>1</sub> -P@3DGF	10	270	79.4	This work
Er-doped CoP NMs	10	256	70	<b>S</b> 1
Hollow Mo-CoP nanoarrays	10	305	56	S2
Co <sub>5.47</sub> N NP@N-PC	10	248	72	S3
Co <sub>4</sub> Ni <sub>1</sub> P NTs	10	245	87	S4
CoP/C	10	430	115	S5
Cu <sub>0.075</sub> Co <sub>0.0925</sub> P/CP	10	221	70.4	S6
(Ni,Co)Se <sub>2</sub> -GA	10	250	70	S7
CoCr-LDH	10	340	81	S18
Co@Co <sub>9</sub> S <sub>8</sub> nanochains	10	285	86.5	S19
S-CoP	10	270	82	S9

**Table S4.** Comparison of OER performances of  $Co_2Cr_1$ -P@3DGF with recently reportedcatalysts in 1.0 M KOH.

Catalysts	Cell voltage (V) at j = 10 mA cm <sup>-2</sup>	Ref.	
Co <sub>2</sub> Cr <sub>1</sub> -P@3DGF	1.56	This work	
Er-doped CoP NMs	1.58	S1	
Hollow Mo-CoP nanoarrays	1.56	S2	
Co <sub>5.47</sub> N NP@N-PC	1.62	S3	
Co <sub>4</sub> Ni <sub>1</sub> P NTs	1.59	S4	
Cu <sub>0.075</sub> Co <sub>0.0925</sub> P/CP	1.55	S6	
(Ni,Co)Se <sub>2</sub> -GA	1.6	S7	
Co <sub>3</sub> S <sub>4</sub> @MoS <sub>2</sub>	1.58	S10	
CoP@CC	1.68	S15	
CoP/NCNHP	1.64	S16	
CoP-Co <sub>2</sub> P@PC/PG	1.567	S20	

**Table S5.** Comparison of overall water splitting performances of  $Co_2Cr_1$ -P@3DGF withrecently reported catalysts in 1.0 M KOH.

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